## JQ.4.6.Setup

## September 23, 2014

(4.6) Given the lower flammability limit of n-butane  $(n - C_4H_10)$  in air is 1.8% by volume, calculate the adiabatic flame temperature at the limit. Assume that the initial temperature to be  $25^{\circ}C$ . Use Table 4.5. Setup.

You know how to calculate flame temperature. I find it easier to work in mass fractions. Be careful with the mole fraction and mass fraction definitions.