## Specific Heat Report Sheet (30 Points)

Name	Date		
Lab Instructor	Lab Day		
Part 1: Specific Heat of Copper Metal		Q.	7
	Your Data	Neighbor 1	Neighbor 2
Weight of copper taken, grams		/	
Weight of water taken, grams			
Temperature of heated copper, °C			er.
Temperature of cool water, °C			?
Final temperature reached, °C		74777	
Heat GAINED by water, Joules			<u>, , , , , , , , , , , , , , , , , , , </u>
Heat LOST by copper, Joules			
Specific heat of copper, Joules/g°C			
Average value of specific heat of copper meta	al		

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Part 2: Specific Heat of Glass Beads			
	Your Data	Neighbor 1	Neighbor 2
Weight of glass taken, grams			-
Weight of water taken, grams			1
Temperature of heated glass, °C			+
Temperature of cool water, °C			
Final temperature reached, °C	_		
Heat GAINED by water, Joules		_	7.60.
Heat LOST by glass, Joules	-		- er
Specific heat of glass, Joules/g°C	)		e dis
Average value of specific heat of glass bead	s		<b>X</b>
Y			