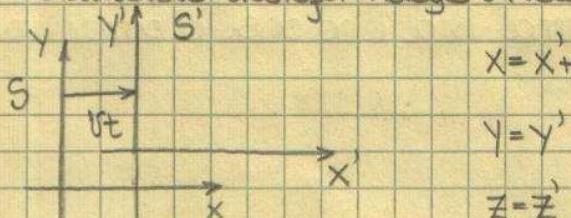


Specijalna teorija relativnosti (A. Einstein, 1905)

① Neovisnost zakona fizike o izboru koordinatnog sustava

- Galilejeve transformacije i Newtonova jednadžba gibanja



$$x = x' + vt$$

$$y = y'$$

$$z = z'$$

$$t = t'$$

$$\frac{d^2x'}{dt'^2} = \frac{d^2x}{dt^2}$$
 (akceleracije su jednake u sustavu S i S')

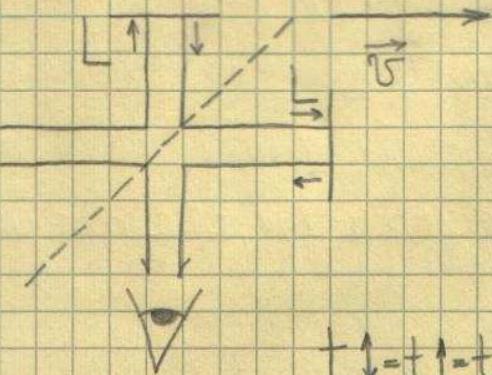
- Galilejeve transformacije i elektromagnetizam, "Blesak" u $\vec{r}=0, t=0$

$$S: x^2 + y^2 + z^2 - c^2 t^2 \text{ (sfera)}$$

$$S': (x+vt)^2 + y^2 + z^2 = c^2 t^2 \text{ (nije sfera, nije OK)}$$

Rješenja: eter ili odustajanje Galilejevih transformacija

② Michelson-Morleyev eksperiment



$$t_{\rightarrow}: L + vt_{\rightarrow} = ct_{\rightarrow}$$

$$t_{\rightarrow} = \frac{L}{c-v}$$

$$t_{\leftarrow} = \frac{L}{c+v}$$

$$t_{\uparrow} = t_{\downarrow} = t_{\perp}: \sqrt{L^2 + (vt_{\perp})^2} = ct_{\perp}, t_{\perp} = \frac{L}{\sqrt{c^2 - v^2}}$$

$$\text{kašnjenje: } \Delta t = (t_{\rightarrow} + t_{\leftarrow}) - (t_{\uparrow} + t_{\downarrow})$$

$$\Delta t = \frac{2L}{c} \cdot \frac{1}{1 - v^2/c^2} \left[1 - \sqrt{1 - v^2/c^2} \right]$$

Michelson i Morley dobiju $\Delta t_{\text{exp}} = 0$ (ne mogu izmjeriti), ubijera ideja etera, čeka se nova teorija koja će pokazati pogrešku Michelsona i Morleya.

(Njihov se korak u gibanju skratio, što nisu znali, dokazala spec.teorija relat.)

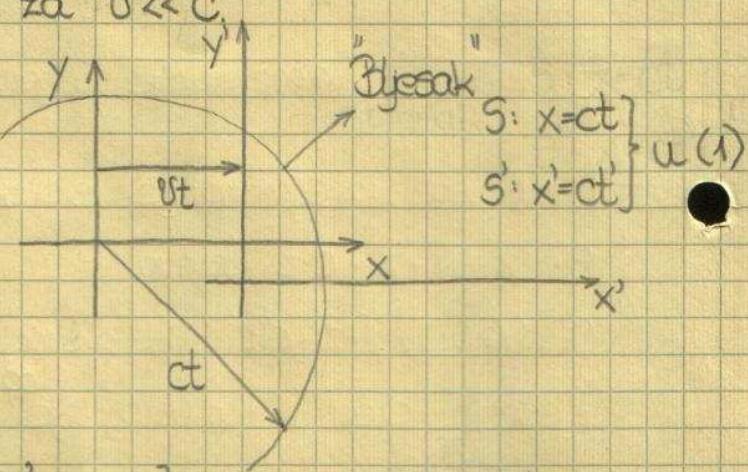
③ Einsteinovi postulati

1EP: Svi fizički zakoni imaju isti oblik u svim koordinatnim sustavima koji se pomicaju u odnosu na drugi gibaju jednoliko po pravcu.

2EP: Brzina svjetlosti je u svim sustavima ista, i ne ovisi o brzini izvora ili primatrača.

④ Lorentzove transformacije

- 1.) Prelaze u Galilejeve transformacije za $v \ll c$,
- 2.) Simetrične u odnosu na $\pm v$.
- 3.) Linearne u x, y, z, t .
- 4.) Sutladne sa 2EP.



$$\left. \begin{aligned} x' &= \gamma(x - vt) \\ x &= \gamma(x' + vt) \end{aligned} \right\} (1)$$

$$\text{eliminacijom } x \text{ i } x' \text{ slijedi: } \left. \begin{aligned} t' &= \gamma t + \frac{\gamma^2 - 1}{\gamma v} x \\ t &= \gamma t' - \frac{\gamma^2 - 1}{\gamma v} x' \end{aligned} \right\} (2)$$

$$t' = \gamma(t - \frac{vx}{c^2})$$

$$t = \gamma(t' + \frac{vx}{c^2})$$

$$\left. \begin{aligned} t' &= \frac{\gamma t}{c} (c - v) \\ t &= \frac{\gamma t'}{c} (c + v) \end{aligned} \right\} \Rightarrow \gamma = \frac{1}{\sqrt{1 - v^2/c^2}} \quad (\gamma \text{ uvratimo u } x, x', t \text{ i } t' \text{ gore u (1) i (2)})$$

Lorentzove transformacije i Bijesak

$$S: x'^2 + y'^2 + z'^2 = c^2 t'^2$$

$$S: x'^2 + y'^2 + z'^2 = c^2 t'^2 \quad (\text{kada uvratimo gornje izraze za } x \text{ i } t) \checkmark$$

⑤ Transformacije brzina

$$S: \vec{u} = \frac{d\vec{r}}{dt}$$

$$S: \vec{u}' = \frac{d\vec{r}'}{dt'}$$

$$x' = \gamma(x - vt)$$

$$t' = \gamma(t - \frac{vx}{c^2})$$

$$u_x' = \frac{dx'}{dt'} = \gamma \left(\frac{dx}{dt} \cdot \frac{dt}{dt'} - \frac{v dt}{dt'} \right) = \gamma(u_x - v) \frac{dt}{dt'}$$

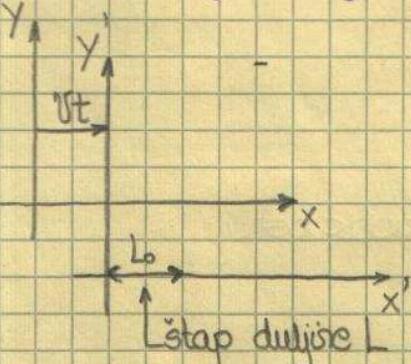
$$\frac{dt'}{dt} = \gamma(1 - \frac{v u_x}{c^2})$$

$$u_x' = \gamma(u_x - v) \cdot \frac{1}{\gamma(1 - \frac{v u_x}{c^2})} = \frac{u_x - v}{1 - \frac{v u_x}{c^2}}$$

$$u_y' = \frac{dy'}{dt'} = \frac{dy}{dt} \cdot \frac{dt}{dt'} = u_y \frac{1}{\gamma(1 - \frac{v u_x}{c^2})} = \frac{u_y \sqrt{1 - v^2/c^2}}{1 - \frac{v u_x}{c^2}}$$

$$u_z' = \frac{u_z \sqrt{1 - v^2/c^2}}{1 - \frac{v u_x}{c^2}}$$

⑥ Kontrakcija duljine



$$S: L_0 = x_2' - x_1' \quad @ t'$$

$$S: L = x_2 - x_1 \quad @ t$$

$$L_0 = x_2' - x_1' = \gamma(x_2 - vt) - \gamma(x_1 - vt) = \gamma(x_2 - x_1) = \gamma L, \quad L \neq L_0, \quad \gamma = \frac{1}{\sqrt{1 - v^2/c^2}} > 1$$

⑦ Dilatacija vremena

$$S: \Delta t = t_2 - t_1 \quad @ x$$

$$S: \Delta t' = t_2' - t_1' \quad @ x'$$

$$\Delta t' = \gamma(t_2 - \frac{vx}{c^2}) - \gamma(t_1 - \frac{vx}{c^2}) = \gamma(t_2 - t_1) = \gamma \Delta t$$

⑧ Relativistička dinamika

Količina gibanja: $\vec{p} = \gamma m \vec{v}$

Energijski kvant: $E = \gamma m c^2$

Newtonov aksiom: $\frac{d\vec{p}}{dt} = \frac{d}{dt} \frac{m\vec{v}}{\sqrt{1 - v^2/c^2}} = \vec{F}$

Rad i energija: $W = \int_{\text{kon.}}^k \vec{F} \cdot d\vec{s} = \int_{\text{kon.}}^k \frac{d\vec{p}}{dt} \cdot d\vec{s} = \int_{\text{kon.}}^k \vec{F} d\vec{s}$

$d(\vec{F} \cdot \vec{p}) = \vec{F} d\vec{p} + \vec{p} d\vec{F}$... parcijalna integracija

$$\int_{\text{kon.}}^k \vec{F} d\vec{p} - \int_{\text{kon.}}^k \vec{p} d\vec{F} = \vec{F} \cdot \vec{p} \Big|_{\text{kon.}}^k - m \int_{\text{kon.}}^k \vec{v} d\vec{s}$$

Zaog jednostavnosti: 1D, $\vec{v}_p = 0$

$$W = \Sigma_{\text{kon.}} m \cdot v_{\text{kon.}}^2 - m \int_0^{v_{\text{kon.}}} \frac{\vec{v} d\vec{v}}{\gamma}$$

$$= \Sigma_{\text{kon.}} m v_{\text{kon.}}^2 + m c^2 \gamma \Big|_0^{v_{\text{kon.}}} = \Sigma_{\text{kon.}} m v_{\text{kon.}}^2 + m c^2 \left(\frac{1}{\gamma} - 1 \right)$$

$$- \gamma m v^2 + \frac{m c^2}{\gamma} - m c^2 = \gamma m c^2 \left(\frac{v^2}{c^2} + \frac{1}{\gamma} \right) - m c^2 = \boxed{\gamma m c^2 - m c^2 \equiv E_{\text{kin.}}}$$

\downarrow Energija kinematička

$$E = \gamma m c^2 - m c^2 + E_{\text{kin.}}$$

Mehanika fluida

1. Fluid

Tecućina $B = \frac{1}{V} = V \frac{\partial P}{\partial V} \Rightarrow$ volumenski modul elastičnosti

Plin "idealni plin"

$$\gamma = \frac{1}{B} = -\frac{1}{V} \frac{\Delta V}{\Delta P} = \frac{1}{P} \Rightarrow$$
 koeficijent izotermne kompresije (adijuktatska brzina)

2. Statika fluida

$$\text{Tlak: } p = \frac{|\vec{F}|}{S} - \frac{|\vec{dF}|}{ds}$$

Jednadžba hidrostatske ravnoteže:

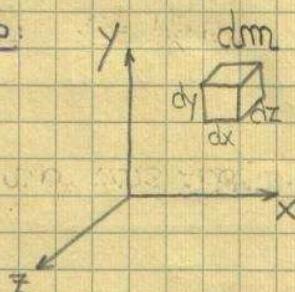
Priopštawke:

- dm u statičkoj ravni

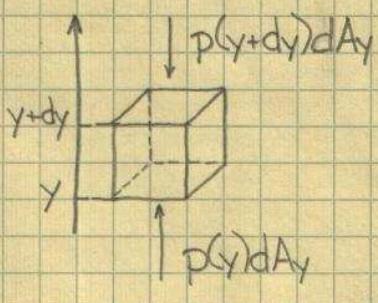
- tlak ovisi samo o y

- fluid se nalazi u polugravitacijske sile $\vec{g} = -g \hat{j}$

Sile u x i z smjeru se dokidaju.



Sile u y smjeru:



$$dF_y = -gdm + p(y)dx dy - p(y+dy)dx dy$$

$$= -g\rho dx dy dz + p(y)dx dz - p(y+dy)dx dz$$

$$= 0$$

$$g\rho dy = p(y) - p(y+dy)$$

$$-g\rho = \frac{p(y+dy) - p(y)}{dy}, \text{ kada } dy \rightarrow 0, \text{ kockica je mala}$$

$$-\rho g = \frac{dp}{dy}$$

Hidrostatski tlak

Prijetpostavke:

- ispunjena jednolikost hidrostatske ravnoteže

- fluid je nestlačiv

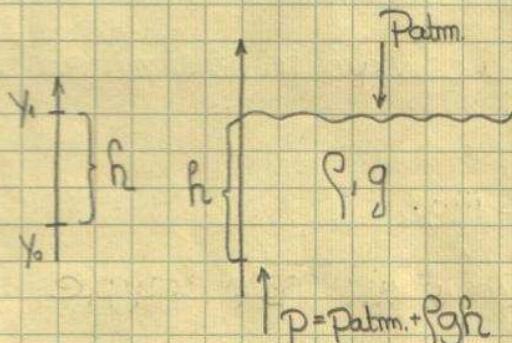
- g je konstanta

$$\frac{dp}{dy} = -\rho g / dy, \checkmark$$

$$\int dp = -\rho g \int dy$$

$$p(y_1) - p(y_0) = -\rho g (y_1 - y_0)$$

$$p(y_0) = p_0 + \rho gh$$



Pascalov princip - u nestlačivom, mitemom fluidu Δp se prenosi na cijelu fluid

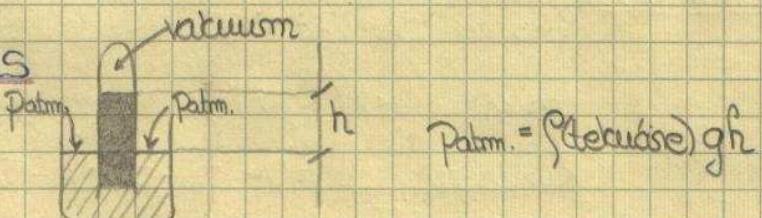
Hidraulički tipčak (hidraulička prečka)



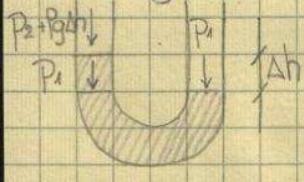
$$\Delta p = \frac{F_1}{S_1} = \frac{F_2}{S_2} \rightarrow \text{Pascalov princip}$$

Izbi rezultat dobijemo i principom virtuelnog rada, $\Delta V = -F_1 \Delta y_1 = F_2 \Delta y_2$ te očuvanjem količine fluida, $\Delta V = -S_1 \Delta y_1 = S_2 \Delta y_2$.

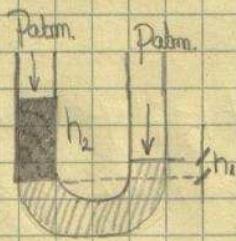
Torricellijev pokus



U cijev



$$\rho g \Delta h = \Delta p$$



$$\rho_1 h_1 = \rho_2 h_2$$

Bazometarska formula

Prijetpostavke:

- T=konstanta

- idealni plin, $pV=nRT$, $n=\frac{m}{M}$

- hidrostatska ravnoteža

- g=konstanta

Jednadžba hidrostatske ravnoteže:

$$\frac{dp}{dh} = -\frac{\rho g}{\rho_0} \frac{P}{P_0} / . dh, \int_{h=0}^H$$

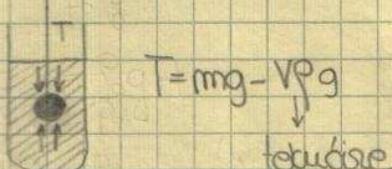
$$\int \frac{dp}{P} = -\frac{\rho g}{\rho_0} \int dH$$

$$\ln \frac{p}{p_0} = -\frac{\rho g H}{\rho_0} / c'$$

$$p_H = p_0 e^{-\frac{\rho g H}{\rho_0}}$$

Anfimedesov zakon (uzgon)

Tijelo umrojeno u tekućinu gubi na težini za težinu istisnute tekućine.



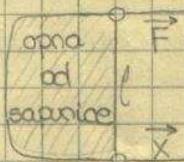
$$T = mg - V \rho g$$

tekućina

③ Površinska napetost

$$\sigma = \frac{\Delta W}{\Delta S} = \frac{\Delta F}{\Delta l}$$

napetost površine (sigma)

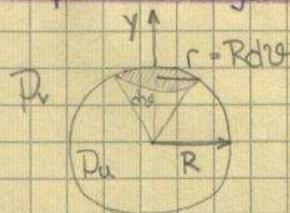


Princip virtuelnog rada: povećanje površine $\Delta S = 2l \Delta x$

obnovljeni rad $\Delta W = F \Delta x$

$$\sigma = \frac{\Delta W}{\Delta S} = \frac{F}{2l} \Rightarrow F = \sigma \cdot 2l \quad \text{jed ova imaju dnuje strane}$$

Laplaceova formula



$$\text{Površina: } dA = (R dr)^2 \pi$$

$$\text{Opseg: } dl = 2(R dr) \pi$$

$$\text{Ravnoteža: } dF_y = 0 = (P_u - P_v) dA - \sigma dl dr$$

$$P_u - P_v = \frac{2\sigma}{R} \quad \text{voda u zraku, zrak u vodi}$$

$$P_u - P_v = \frac{4\sigma}{R} \quad \text{zrak u zraku}$$

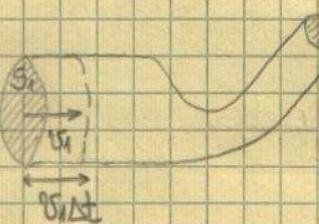
4.) Dinamika fluida

Pretpostavke:

- fluid je nestlačiv, izotermos, nije viskozan
- tok je stacionarni i laminarisan

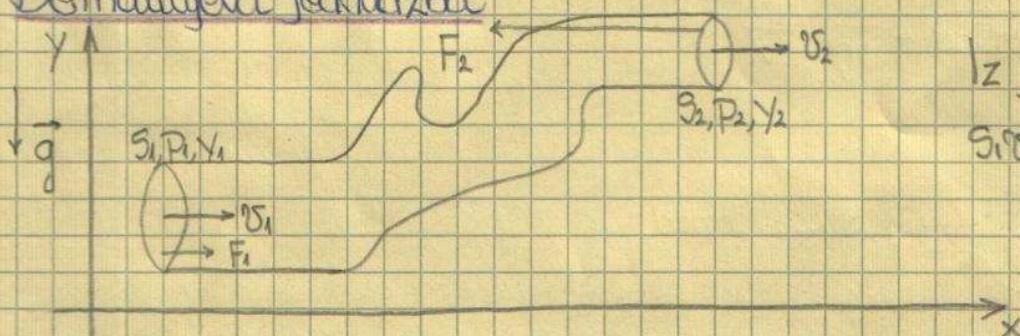
Jednadžba kontinuiteta:

- količina fluida u volumenu je stacionarna



$$\text{uz: } \Delta V_1 = S_1 v_1 \Delta t \quad \left. \begin{array}{l} \text{izlaz: } \Delta V_2 = S_2 v_2 \Delta t \\ \Delta V_1 + \Delta V_2 = 0 \end{array} \right\} \Rightarrow S_1 v_1 = S_2 v_2$$

Bernoullijeva jednadžba



Iz jed. kontinuiteta:

$$S_1 v_1 = S_2 v_2 = S v$$

RAD SILA: $\Delta W_1 = F_1 \Delta x_1 = S_1 p_1 \Delta t = S p_1 \Delta t$

$$\Delta W_2 = -F_2 \Delta x_2 = -S_2 p_2 \Delta t = -S p_2 \Delta t$$

UKUPNI RAD: $\Delta W = \Delta W_1 + \Delta W_2 = S \Delta t (p_1 - p_2)$

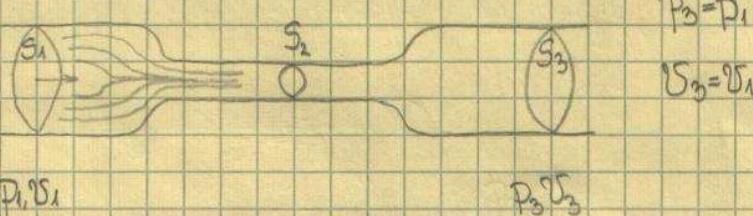
OČUVANJE ENERGIJE: $\Delta W = \Delta E_{kin} + \Delta E_{pot}$

$$\Delta E_{kin} = \frac{1}{2} \Delta m (v_2^2 - v_1^2) = \frac{1}{2} S v \Delta t (v_2^2 - v_1^2)$$

$$\Delta E_{pot} = \Delta m g (y_2 - y_1) = S v \Delta t g (y_2 - y_1)$$

$$p_1 - p_2 = \frac{1}{2} \rho (v_2^2 - v_1^2) + \rho g (y_2 - y_1) \quad \text{ili} \quad p + \frac{1}{2} \rho v^2 + \rho gy = \text{konst.}$$

5.) Viskoznost



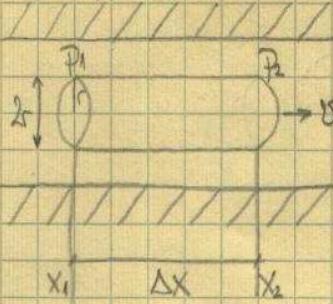
$$p_3 = p_1 \quad p_3 < p_1$$

$$v_3 = v_1$$

SILA: $F_v = \eta S \frac{v}{t} = \eta S \frac{dv}{dt}$

η ... koef. viskoznosti

Poisuelleov zakon



SILA NA VALJAK:

$$F_m = \eta 2\pi r \frac{d\sigma}{dr} = \eta \pi (P_1 - P_2)$$

$$2\eta (x_2 - x_1) \frac{d\sigma}{dr} = \eta (P_1 - P_2)$$

$$\frac{d\sigma}{dr} = -\frac{\Delta P}{2\eta \Delta x} \quad / \int_{R}^{r}$$

$$\sigma(r) - \sigma(R) = -\frac{1}{2\eta \Delta x} \Delta P \cdot \frac{r^2}{R}$$

$$\sigma(r) = \frac{R^2}{4\eta} \cdot \frac{\Delta P}{\Delta x} \left(1 - \frac{r^2}{R^2} \right)$$

\hookrightarrow brezna bojama tekućina teče u centru cijevi

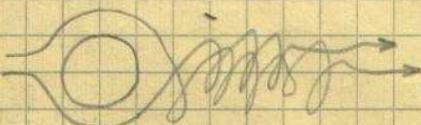
Tok:

$$q = \frac{\Delta V}{\Delta t}, \quad \Delta V = A \Delta t \quad \text{da}$$

$$q = \int \sigma da = \int \sigma 2\pi r dr = \frac{\pi}{8\eta} \cdot \frac{\Delta P}{\Delta x} \cdot R^4$$

Turbulencija:

Laminarna:  mala brzina

Turbulenta:  velika brzina

Prelaz laminarnog u turbulentno strujanje

Reynoldsov broj

$$Re = \frac{\rho U L}{\eta} = \frac{U L}{\nu}$$

\hookrightarrow kinetička viskoznost

Stokesov zakon

-kugla, spono gibaće: $F_m = 6\pi\eta R \dot{\sigma}$

-opadanje: $F_m = \frac{1}{2} C_d \rho S v^2$
 \hookrightarrow bez otpora

$$C_d = \frac{24}{Re} + \frac{6}{1+Re} + \frac{2}{5}$$

Toplina

① Termodynamička temperatura je mjerba kinetičke energije.
"apsolutna nula" (kada se čestice u plinu prestanu gbiti)

② Topinski razmjer

DULJINA: $\Delta l = \alpha l \Delta t \rightarrow$ promjena temperature
 ↓ koef. linearog rastezanja

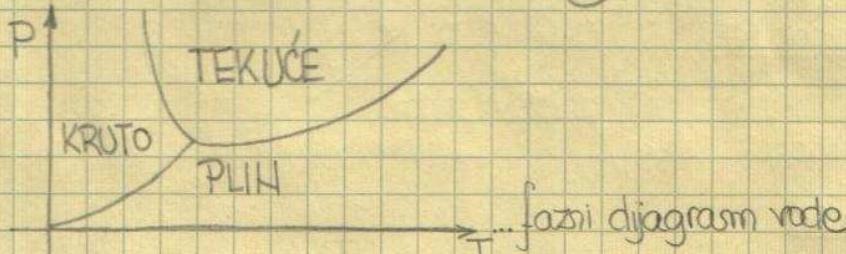
POVRSINA: $a = l^2$ $\frac{\Delta a}{a} = \frac{a' - a}{a} = \frac{(l + \Delta l)^2 - l^2}{l^2} = \frac{1}{l^2} (2l\Delta l + \Delta l^2) = \frac{2\Delta l}{l} = 2\alpha \Delta t \Rightarrow \alpha = 2\alpha$

VOLUMEN: $V = l^3$ $\frac{\Delta V}{V} = \frac{V' - V}{V} = \frac{(l + \Delta l)^3 - l^3}{l^3} = \frac{1}{l^3} (l^3 + 3l^2\Delta l + 3l\Delta l^2 + \Delta l^3 - l^3) =$
 $= \frac{3\Delta l}{l} + \frac{3\Delta l^2}{l^2} + \frac{\Delta l^3}{l^3} = \frac{3\Delta l}{l} = 3\alpha \Delta t \Rightarrow \alpha = 3\alpha$
 ↓ koef. volumenskog rastezanja

③ Plinski zakoni

Jednadžba stanja idealnog plina

$$pV = nRT \quad 8.314 \text{ J/mol}\cdot\text{K} \quad \text{ili} \quad pV = NkT \quad 1.381 \cdot 10^{-23} \text{ J/K, Boltzmannova konstanta}$$



Avogadroov zakon: Pm jednakom tlaku i temperaturi, isti volumeni različitih plinova sadrže jednak broj čestica.

3 empirijska zakona: (oni su međusobno ovjeni)

1) Boyle-Mariotteov zakon $T = \text{konst.}, pV = \text{konst.}$ (izotermno)

2) Gay-Lussacov $p = \text{konst.}, \frac{V}{T} = \text{konst.}$ (izobarsko)

3) Charles-Gay-Lussac $V = \text{konst.}, \frac{P}{T} = \text{konst.}$ (izohorno)

Izvod za izohorni proces (analogni i za sve ostale procese): \rightarrow

$$(p_0, V_0, T_0) \xrightarrow{\text{izobarmo}} (p_1, V_1, T_1) \xrightarrow{\text{izotermno}} (p_1, V_0, T_1)$$

$$\frac{V_0}{T_0} = \frac{V_1}{T_1}$$

$$p_0 V_1 = p_1 V_0$$

$$T_0 = \frac{V_0}{V_1} T_1$$

$$p_0 = \frac{p_1}{V_1} V_0$$

$$\frac{p_0}{T_0} = \frac{p_1}{T_1}$$

Standardni uvjeti: $p_0 = 1 \text{ atm.} = 101325 \text{ Pa}$, $T_0 = 0^\circ\text{C} = 273.15 \text{ K}$, $V_0 = 22.41383 \text{ L}$

Izvod jednadžbe stanja:

$$(p_0, V_0, T_0) \xrightarrow{\text{izobarmo}} (p_0, V, T) \xrightarrow{\text{izotermno}} (p, V, T)$$

$$\frac{V_0}{T_0} = \frac{V}{T}$$

$$p_0 V = p V$$

$$V' = V_0 \frac{T}{T_0}$$

$$p V = \frac{p_0 V_0 T}{T_0} = RT \Rightarrow \text{općenito: } p V = n R T = \frac{N R T}{N_A} k$$

Daltonov zakon parcijalnih tlakova

$$p = p_1 + p_2 + \dots \quad \downarrow \quad \begin{matrix} p V = n R T \\ n = n_1 + n_2 + \dots \end{matrix} \quad \Rightarrow \quad \begin{matrix} p V = n_1 R T \\ p_1 V = n_1 R T \\ \vdots \end{matrix}$$

4. Kalorimetrija

Specifični toplinski kapacitet

$$c_{H2O} = 4190 \text{ J K}^{-1} \text{ kg}^{-1}$$

$$\Delta Q = c m \Delta t$$

\hookrightarrow spec. toplinski kapacitet $[\text{J K}^{-1} \text{ kg}^{-1}]$

$$\Delta Q = C_n \Delta t$$

\hookrightarrow molarni topl. kapac. $[\text{J K}^{-1} \text{ mol}^{-1}]$

Dulong Petibovo pravilo

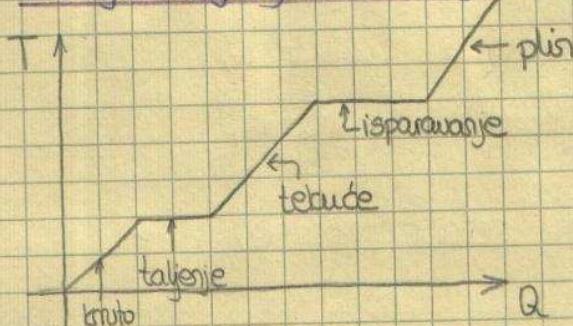
Prawilo tvrdi da je za krude tvari $C_p \approx C_v \approx 3R$ tj. $C_p \approx C_v \approx 3R$.

Meyersova relacija (vrijedi samo za plinove)

$$C_p - C_v = R$$

Adijabatski koeficijent plina $\gamma_k = \frac{C_p}{C_v}$

Promjene agregatnih stanja, latente topline



$$Q_{\text{tajenja}} = m l_t \rightarrow \text{spec. latent. toplina tajenja, } [\text{J kg}^{-1}]$$

$$Q_{\text{isparavanja}} = m l_i \rightarrow \text{spec. latent. toplina isparavanja, } [\text{J kg}^{-1}]$$

5. Vodenje topline

KONVEKCIJA: medij, gibanje

RADIJACIJA: elektro-magnetsko zračenje

KONDUKCIJA: medij, mikrovanje

$$\text{Toplinski tok: } \Phi = \frac{dQ}{dt} = \lambda S \frac{\Delta T}{\Delta x} \quad \begin{matrix} \text{koef. toplinske vodljivosti} \\ \uparrow \end{matrix}$$

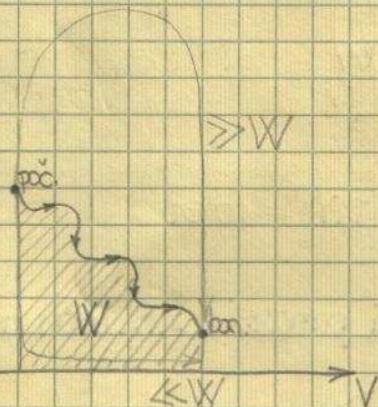
"Ohmova zakon" $\Phi = \frac{\Delta T}{R}, R = \frac{\Delta x}{\lambda S}$

6. Prvi zakon termomodulastike

$$d'Q = dU + d'W \rightarrow \begin{matrix} \downarrow \\ \text{funkcija stanja} \end{matrix} \quad \rightarrow \begin{matrix} \text{funkcija procesa} \\ P \uparrow \end{matrix}$$

$$dW = pdV$$

$$W = \int pdV$$



Izvod Meyemove relacije:

$$d'Q = nC_V dT = dU \rightarrow d'Q = dU + d'W \quad \leftarrow dW = pdV = nRdT$$

$$d'Q = nC_P dT \quad nC_P dT = nC_V dT + nRdT$$

$$C_P - C_V = R$$

$$1\text{-atomni plinovi: } C_P = \frac{5}{2}R, C_V = \frac{3}{2}R, \gamma = \frac{5}{3}$$

$$2\text{-atomni plinovi: } C_P = \frac{7}{2}R, C_V = \frac{5}{2}R, \gamma = \frac{7}{5}$$

$$3\text{-atomni plinovi: } C_P = 4R, C_V = 3R, \gamma = \frac{4}{3}$$

Adijabata $d'Q=0$ (sustav ne može ili ne stigne razmijesiti toplinu s okolinom)

$$d'Q = dU + d'W = 0$$

$$pV = nRT$$

$$= nC_V dT + pdV$$

$$pdV + Vdp = nRdT$$

$$= \frac{nC_V}{nR} (pdV + Vdp) + pdV$$

$$dT = \frac{(pdV + Vdp)}{nR}$$

$$= \left(\frac{C_V}{R} + 1 \right) pdV + \frac{C_V}{R} Vdp / \cdot \frac{R}{pV}$$

$$\frac{C_V}{R} \frac{dP}{P} = - \left(\frac{C_V + R}{R} \right) \frac{dV}{V}$$

$$\frac{dP}{P} = - \left(\frac{C_V + R}{C_V} \right) \frac{dV}{V} = \frac{C_P}{C_V} \frac{dV}{V} = - \frac{\gamma}{\gamma - 1} \frac{dV}{V}$$

$$\frac{dP}{P} = - \frac{\gamma}{\gamma - 1} \frac{dV}{V}$$

$$\ln \frac{P_2}{P_1} = - \frac{\gamma}{\gamma - 1} \ln \frac{V_2}{V_1}$$

$$\frac{P_2}{P_1} = \left(\frac{V_2}{V_1} \right)^{-\frac{\gamma}{\gamma - 1}}$$

$$\frac{P_2}{P_1} \cdot \frac{V_1^{\gamma}}{V_2^{\gamma}} = 1$$

$$P_1 V_1^{\gamma} = P_2 V_2^{\gamma} = \text{konst.}$$

Rad pm revenzibilnog procesa:

$$\text{Po izotermi: } W_t = nRT \ln \frac{V_{\text{kon.}}}{V_{\text{poč.}}}$$

$$\text{Po izobari: } W_p = p(V_{\text{kon.}} - V_{\text{poč.}})$$

$$\text{Po izohori: } W_v = 0$$

$$\text{Po adijabati: } W_{\text{ad.}} = \frac{nR}{\gamma - 1} (T_{\text{poč.}} - T_{\text{kon.}})$$

$$\ln \frac{P_2}{P_1} = - \frac{\gamma}{\gamma - 1} \ln \frac{V_2}{V_1}$$

$$\frac{P_2}{P_1} = \left(\frac{V_2}{V_1} \right)^{-\frac{\gamma}{\gamma - 1}}$$

$$\frac{P_2}{P_1} \cdot \frac{V_1^{\gamma}}{V_2^{\gamma}} = 1$$

$$P_1 V_1^{\gamma} = P_2 V_2^{\gamma} = \text{konst.}$$

7. Kružni procesi, 2. zakon termodynamike

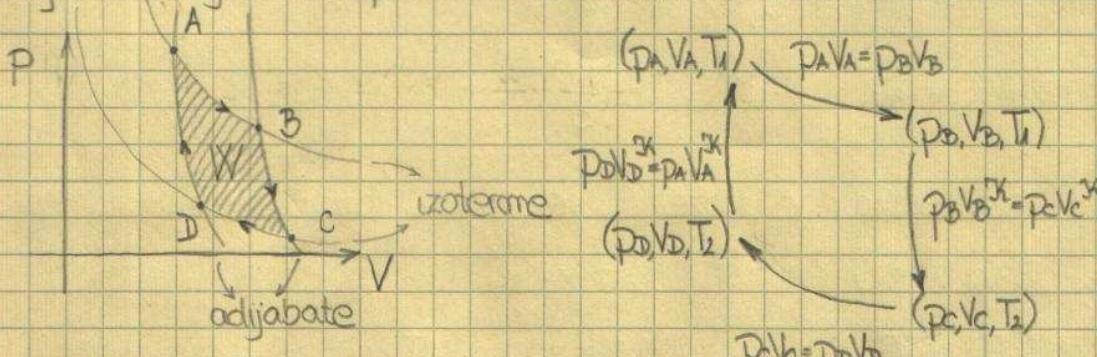


$$W = \int dW = \int pdV = Q$$

2. zakon termodynamike: Nemoguće je napraviti toplinski stroj koji bi smu toplinu iz jednog spremnika pretvorio u rad.

Carnotov kružni proces

- najekonistiviji kružni proces u termodynamici



Ekonistivost:

$$\eta = \frac{\text{Dobiveno}}{\text{Uloženo}} = \frac{\text{Rad}}{\text{Toplina iz } T_1}$$

$$\text{Rad} = \int dW = W_{AB} + W_{BC} + W_{CD} + W_{DA}$$

$$W_{AB} = nRT_1 \ln \frac{V_B}{V_A} = Q_1 > 0$$

$$W_{BC} = \frac{nR}{\gamma - 1} (T_1 - T_2) = -W_{DA}$$

$$W_{CD} = nRT_2 \ln \frac{V_D}{V_C} = Q_2 < 0$$

$$\eta = \frac{W_{AB} + W_{BC} + W_{CD} + W_{DA}}{Q_1} = 1 + \frac{W_{CD}}{W_{AB}} = 1 + \frac{T_2 \ln \frac{V_D}{V_C}}{T_1 \ln \frac{V_B}{V_A}} = 1 - \frac{T_2}{T_1} = \eta_{Carnot}$$

• Kod inverzog Carnotovog procesa (dormute strelice) rad je negativne jer ga mi obavljamo.

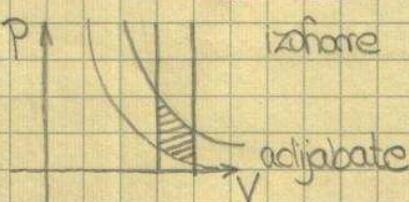
Toplinska pumpa (T...efikasnost)

$$\Gamma_{TP} = \frac{|Q_1|}{|W|}$$

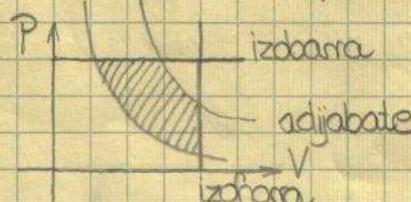
Hladnjak

$$\Gamma_H = \frac{|Q_2|}{|W|}$$

Otto kružni proces



Dieselov kružni proces



8) Entropija - fuzikacija stanja $dS = \frac{d'Q}{T}$

Kako se mijenja entropija u jednom ciklusu Carnotovog kružnog procesa:

$$\frac{|Q_2|}{Q_1} = \frac{T_2}{T_1}$$

$$\Delta S_1 = \int_A^B dS = \int_A^B \frac{dQ}{T_1} = \frac{Q_1}{T_1}, \quad \Delta S_2 = \int_D^C dS = \int_D^C \frac{dQ}{T_2} = \frac{|Q_2|}{T_2}, \quad \Delta S = \Delta S_1 + \Delta S_2$$

Clausiusov teorem $\rightarrow \oint dS = 0$

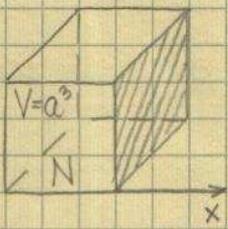
Entropija idealnog plina

$$\begin{aligned} \Delta S &= S_2 - S_1 = \int_1^2 \frac{dQ}{T} = \int_1^2 \frac{dU + pdV}{T} = \int_1^2 \frac{pdV}{T} + \int_1^2 \frac{nCdT}{T} = \int_1^2 \frac{nR}{V} dV - nC \ln T \Big|_1^2 + nR \ln V \Big|_1^2 \\ &= nC \ln \frac{T_{kon.}}{T_{poč.}} + nR \ln \frac{V_{kon.}}{V_{poč.}} \end{aligned}$$

Molekularno-kinetička teorija plinova

1. Model idealnog plina

② Tlak plina



- pri udaru i-te čestice o desni zid na zid se prenosi: $\Delta P_{\text{pl}} = 2 \Delta p_{\text{pl}} = 2 m_i \Delta v_{xi}$
- vjerojatnost da čestica s $\Delta v_{xi} > 0$ udari u desni zid u Δt : $q_i = \frac{\Delta v_{xi} \Delta t}{a}$

$$\text{Tlak} = \frac{\text{Sila}}{\text{Površina}} = \frac{\Delta p / \Delta t}{\text{Površina}}$$

$$p = \frac{1}{\Delta t} \sum \frac{\Delta P_{\text{pl}} \cdot q_i}{a^2} = \frac{1}{a^2 \Delta t} \sum 2 m_i \Delta v_{xi} \frac{\Delta v_{xi} \Delta t}{a} = \frac{2}{a^2} \sum m_i \frac{\Delta v_{xi}^2}{\Delta t} = \frac{2}{3a^2} \frac{\sum m_i \Delta v_{xi}^2}{2} = \frac{2}{3V} \sum_i E_{\text{kin},i} = \frac{2N}{3V} E_{\text{kin}}$$

③ Ekviparticijski teoremi

- 1-atomna mol. imala 3 stupnja slobode, 2-atomska 5, 3-i više atomska 6

i... stupnji slobode

$$U = N \frac{i k T}{2} = n i \frac{RT}{2}, \quad \overline{E}_{\text{kin.}} = \frac{3}{2} k T$$

④ Toplinski kapacitet plina

$$C_v = \frac{1}{2} R$$

$$C_p = \left(1 + \frac{1}{2}\right) R$$

$$\gamma = 1 + \frac{1}{2}$$

⑤ Toplinski kapacitet knutih tvrđi

