

Chapter 5

Introduction to Proteins:

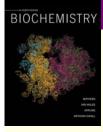
The Primary Level of Protein Structure

Amino acid sequence of a protein

生化分生科 游佳融 2014/09

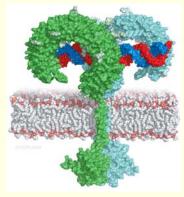
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BIOCHEMISTRY Living cells all have a similar basic chemistry Central Dogma 中心法則 -1957, Francis Crick 克里克 DNA->RNA->protein t-RNA t-RNA **mRNA** Ribosome Amino acid (a.a.) t-RNA **Protein** t-RNA Post Translational 1953 **Modifications** СНО PO_4 **Active Protein** FA



Proteins play an enormous variety of roles:

- transport and storage of small molecules
- structural framework of cells and tissues
- muscle contraction
- immune responses
- blood clotting
- enzymes-the biological catalysts



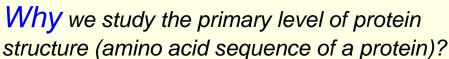
Function: Structure/conformation dependent

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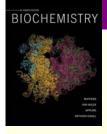
Copyright © 2013 Pearson Canada Inc. Toll-like receptor protein vs. double-stranded RNA



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- The protein structure is able to fulfill a specific biological function.
- The amino acid sequence of a protein determines its three-dimensional structure.

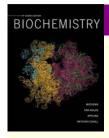


Chapter 5 Outline:

- Structures of the α-Amino Acids
- Properties of the Amino Acid Side Chains
- Peptides and the Peptide Bond
- Proteins: Polypeptides of Defined Sequence
- From Gene to Protein
- Protein Sequence Homology
- Protein Expression and Purification
- Mass, Sequence, and Amino Acid Analyses of Purified Proteins

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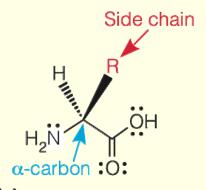
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Structures of the α -Amino Acids

To the α -carbon of every amino acid are attached:

- an amino group
- a hydrogen atom
- a side chain ("R" group)

Different α -amino acids are distinguished by their side chains.

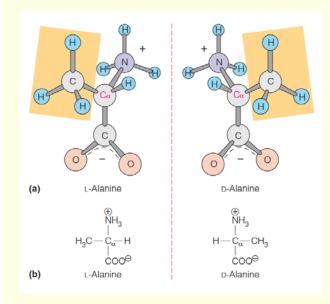


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Stereochemistry of the α -Amino Acids

立體化學



- α -Amino acids' stereochemistry is designated as D- or L-, which is best visualized from its Fischer projection
- There is a preference for L-amino acids in proteins
- Compare this to the preference for D-configured carbohydrates.

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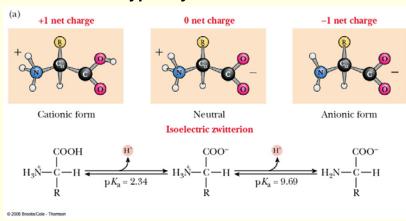
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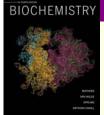
- The p K_a of the carboxylic acid is about 2
- The p K_a of the α -amino group is about 10



- Therefore, at physiological pH both the carboxylic (b) zwitterion acid group and the α -amino group will be ionized, to yield the **zwitterion form** 兩件離子的
- Amino acids are typically written in their zwitterionic form



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There are 20 different kinds of amino acids are commonly incorporated into proteins.

TABLE 5.1 Properties of the common amino acids found in proteins

Name	Abbreviations 1- and 3-letter codes	pK_a of α -COOH Group	pK_a of α -NH ₃ ⁺ Group	pK _a of Ionizing Side Chain ^a	Residue ^b Mass (daltons)	Occurrence in Proteins (mol %)
Alanine	A, Ala	2.3	9.7	_	71.08	8.7
Arginine	R, Arg	2.2	9.0	12.5	156.20	5.0
Asparagine	N, Asn	2.0	8.8	_	114.11	4.2
Aspartic acid	D, Asp	2.1	9.8	3.9	115.09	5.9
Cysteine	C, Cys	1.8	10.8	8.3	103.14	1.3
Glutamine	Q, Gln	2.2	9.1	_	128.14	3.7
Glutamic acid	E, Glu	2.2	9.7	4.2	129.12	6.6
Glycine	G, Gly	2.3	9.6	_	57.06	7.9
Histidine	H, His	1.8	9.2	6.0	137.15	2.4
Isoleucine	I, Ile	2.4	9.7	_	113.17	5.5
Leucine	L, Leu	2.4	9.6	_	113.17	8.9
Lysine	K, Lys	2.2	9.0	10.0	128.18	5.5
Methionine	M, Met	2.3	9.2	_	131.21	2.0
Phenylalanine	F, Phe	1.8	9.1	_	147.18	4.0
Proline	P, Pro	2.0	10.6	_	97.12	4.7
Serine	S, Ser	2.2	9.2	_	87.08	5.8
Threonine	T, Thr	2.6	10.4	_	101.11	5.6
Tryptophan	W, Trp	2.4	9.4	_	186.21	1.5
Tyrosine	Y, Tyr	2.2	9.1	10.1	163.18	3.5
Valine	V, Val	2.3	9.6		99.14	7.2

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Some biologically important amino acids not typically found in protein

Name	Formula	Biochemical Source, Function
β -Alanine	H ₃ N-CH ₂ -CH ₂ -COO-	Found in the vitamin pantothenic acid and in some important natural peptides
D-Alanine	COO- H-C-NH ₃ CH ₃	In polypeptides in some bacterial cell walls
γ-Aminobutyric acid	H ₃ N - CH ₂ - CH ₂ - CH ₂ - COO-	Brain, other animal tissues; functions as neurotransmitter
D-Glutamic acid	COO- H-C-NH ₃ - CH ₂ - CH ₂ -COO-	In polypeptides in some bacterial cell walls
L-Homocysteine	соо- H ₃ N-с-н сн ₂ -сн ₂ sн	Many tissues; precursor for methionine biosynthesis
L-Ornithine	COO- H ₃ Ň-C-H CH ₂ -CH ₂ -CH ₂ ŇH ₃	Many tissues; an intermediate in arginine synthesis
Sarcosine	CH ₃ -N-CH ₂ -COO-	Many tissues; intermediate in amino acid synthesis
L-Thyroxine	COO- CH2-C-H OH	Thyroid gland; is thyroid hormone (I = iodine)

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Properties of the Amino Acid Side Chains

The 20 common amino acids are classified by their side chains:

- Aliphatic (脂肪族的)
- Hydroxyl or sulfur-containing (含有 -OH 或 S)
- Aromatic (芳香族的)
- Basic (鹼性)
- Acidic and their amides (酸性以及其氨基化合物)

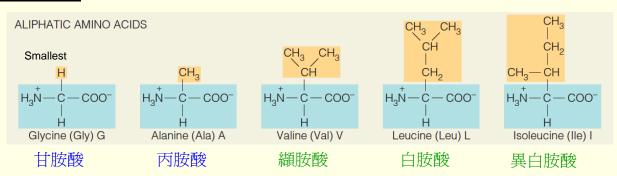
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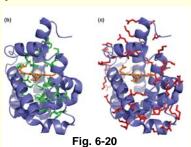
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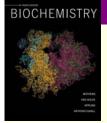
Aliphatic (脂肪族的)



 The more hydrophobic amino acids such as isoleucine are usually found within the core of a protein molecule, where they are shielded from water.

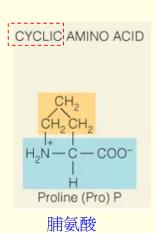
VLSEGEWQLV LHVWAKVEAD VAGHGQDILI RLFKSHPETL EKFDRFKHLK TEAEMKASED LKKHGVTVLT ALGAILKKKG HHEAELKPLA QSHATKHKIP IKYLEFISEA IIHVLHSRHP GDFGADAQGA MNKALELFRK DIAAKYKELG YQG

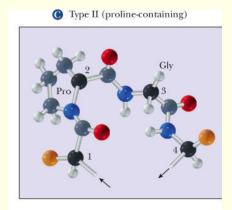




- **Proline,** is the only amino acid in this group in which the side chain forms a covalent bond with the α -amino group.
- The proline side chain has a primarily aliphatic character; however, it is frequently found on the surfaces of proteins due to its unique structural constraints.

 獨特
- ●The rigid ring of proline is well-suited to those sites in a protein structure where the protein must fold back on itself (so-called "turns").



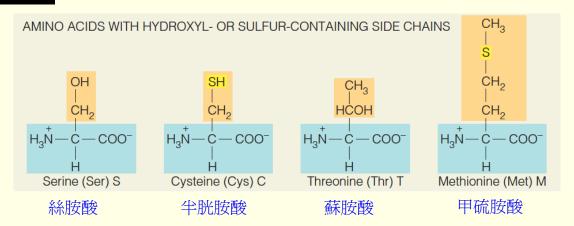


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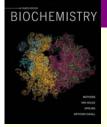
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Hydroxyl or sulfur-containing (含有 -OH 或 S)



 The –OH group of serine and the –SH group of cysteine are good nucleophiles and often play key roles in enzyme activity.
 親核性



Cysteine (半胱氨酸)

雙硫鍵

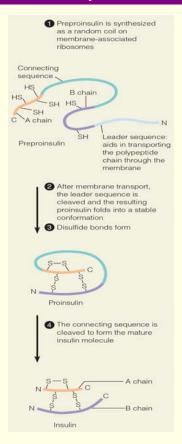
The oxidation of two cysteine side chains yields a disulfide bond. The product of this oxidation is given the name cystine. (胱氨酸, 双硫丙氨酸)

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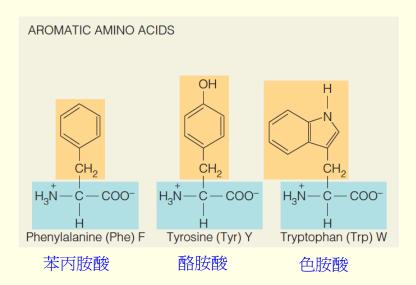
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Copyright © 2013 Pearson Canada Inc. Fig. 5.21 Structure and formation of mature insulin

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Aromatic (芳香族的)



- Phenylalanine is one of the most hydrophobic amino acids.
- Tyrosine and tryptophan have some hydrophobic character as well, but it is tempered by the polar groups in their side chains.

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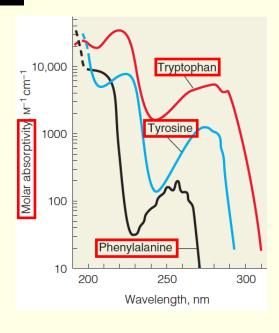
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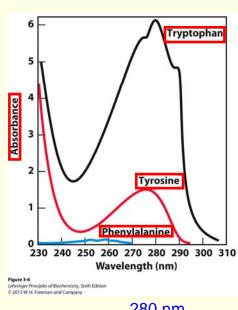
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Absorption of ultraviolet light by aromatic amino acid

應用於偵測與半定量蛋白質

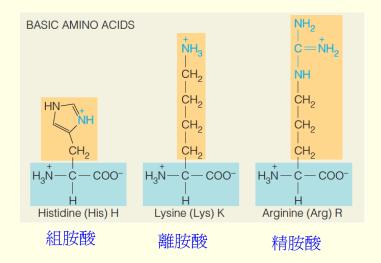




280 nm

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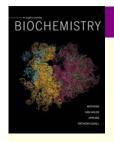
Basic (鹼性)



- Lysine ($pK_a = 10.0$) and arginine ($pK_a = 12.5$) are the more basic amino acids.
- Their side chains are almost always positively charged under physiological conditions.
- guanidino group (HNC(NH₂)₂) of arginine, strong base

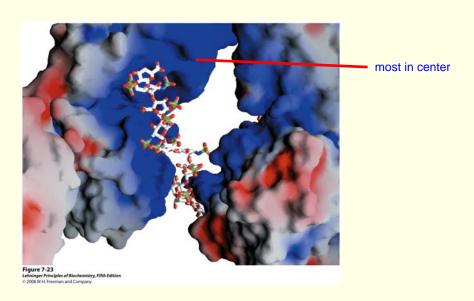
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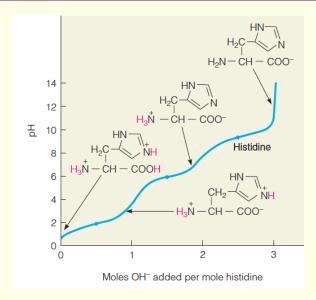
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Interaction Between A Glycosaminoglycan and Its Binding Protein Structure of fibroblast growth factor receptor (FGFR) and heparin



Blue: Basic Red: Acid

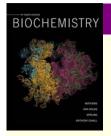
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- Histidine is the least basic of the three, and as its titration curve shows, the imidazole ring in the side chain of the free amino acid loses its proton at about pH 6.
- Because the histidine side chain has a pK_a near physiological pH, it often plays a role in enzymatic catalysis involving proton transfer.

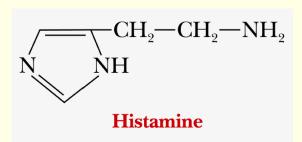
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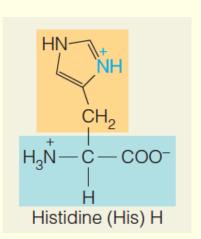
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Amino acid function other than in peptides

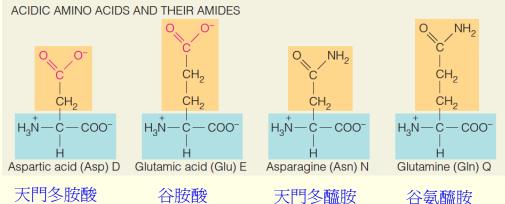




<mark>組織胺</mark> (組胺酸的衍生物)



Acidic and their amides (酸性以及其氨基化合物)



- Aspartic acid (p K_a = 3.9) and glutamic acid (p K_a = 4.2) typically carry negative charges at pH 7.
- Unlike their acidic analogs, asparagine and glutamine have uncharged polar side chains.
- Like the basic and acidic amino acids, Asn and Gln are hydrophilic and tend to be on the surface of a protein molecule, in contact with the surrounding water.

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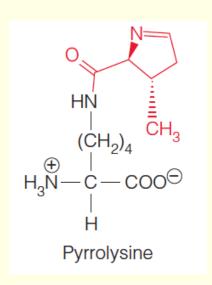
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Two amino acids, **selenocysteine** and **pyrrolysine**, are encoded genetically and incorporated into proteins; however, they are found in a relatively small number of proteins.

$$\begin{array}{c|c} & H_2C-CH_2\text{SeH} \\ \oplus & | \\ H_3N-C-COO^{\scriptsize \bigcirc} \\ & | \\ & H \\ & \text{Selenocysteine} \end{array}$$

21st aa.



22nd aa.



Modified amino acids

Amino acids can undergo **post-translational modification** (轉譯後修飾) resulting in modified amino acids with unique properties

PO₃²⁻ OH H-
$$\frac{1}{\sqrt{C}}$$
 CH₂ CH₂ $\frac{1}{\sqrt{C}}$ COO- $\frac{1}{\sqrt{C}}$ H₃N- $\frac{1}{\sqrt{C}}$ COO- $\frac{1}{\sqrt{C}}$ H₃N- $\frac{1}{\sqrt{C}}$ COO- $\frac{1}{\sqrt{C}}$ Phosphoserine 4-Hydroxyproline $\frac{1}{\sqrt{C}}$ 6-Hydroxylysine $\frac{1}{\sqrt{C}}$ Coo-coo-acid

Rich in Collagen

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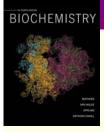
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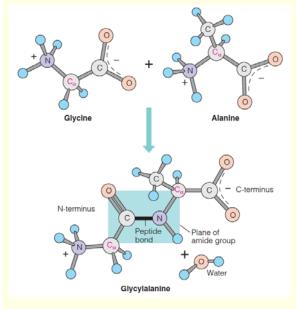
Protein phosphorylation

Phosphoserine

Phosphothreonine



Peptide Bond (胜肽鍵)



- Amino acids can be covalently linked together by formation of an amide bond between the α-carboxylic acid group on one amino acid and the α-amino group on another.
- Peptide bond
- Covalent bond
- Nearly planar
- A peptide composed of 2 amino acids is called a dipeptide.

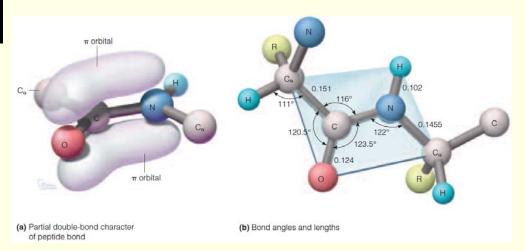
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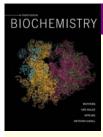
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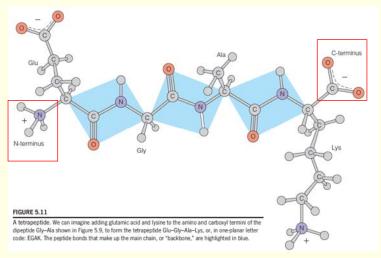
The property of peptide bond



- The amide carbonyl and amide bonds are nearly parallel.
- The six atoms shown in the blue rectangle in usually coplanar.
- There is <u>little twisting possible</u> around the peptide bond because the bond has a substantial fraction of <u>double-bond</u> <u>character</u>.



Peptides (胜肽)



Glu-Gly-Ala-Lys

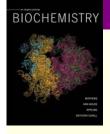
or

EGAK

- N-terminus to C-terminus.
- EGAK≠KAGE
- The portion of each amino acid remaining in the chain is called an amino acid residue.(殘基)

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Peptide chains (胜肽鏈)

- Chains containing only a few amino acid residues (like a tetrapeptide) are collectively referred to as oligopeptides.
- If the chain is longer (>15-20 residues), it is called a polypeptide.
- Polypeptides greater than ~50 residues are generally referred to as proteins.
- Note that most globular proteins contain 250–600 amino acid residues.

TABLE 5.1 P

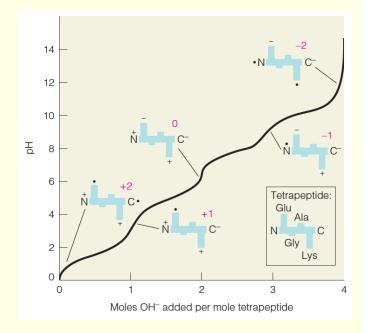
Name	Residue ^b Mass (daltons)		
Alanine	71.08		
Arginine	156.20		
Asparagine	114.11		
Aspartic acid	115.09		
Cysteine	103.14		
Glutamine	128.14		
Glutamic acid	129.12		
Glycine	57.06		
Histidine	137.15		
Isoleucine	113.17		
Leucine	113.17		
Lysine	128.18		
Methionine	131.21		
Phenylalanine	147.18		
Proline	97.12		
Serine	87.08		
Threonine	101.11		
Tryptophan	186.21		
Tyrosine	163.18		
Valine	99.14		

110~120 Da 5-30



Polyampholytic behavior of a tetrapeptide

- •This titration curve for the tetrapeptide Glu-Gly-Ala-Lys (EGAK) shows the major forms present at several pH values.
- •Note the net charges for the different ionization states in red.



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Hydrolysis of peptide bond

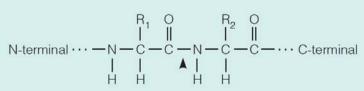
$$\begin{array}{c} :O: \\ \leftarrow \\ C\alpha_n \\ \downarrow \\ H \end{array}$$

- Peptide bond hydrolysis can be achieved by:
 - o Strong mineral acid (e.g., 6 M HCl) cleaves all peptide bonds (including the Asn and Gln amide bonds)
 - Chemicals that cleave at specific sites (e.g., CNBr cleaves at Met)
 - Proteolytic enzymes (proteases) that cleave at specific sites

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Cleavage and proteolysis of peptide chains

TABLE 5.4 The sequence specificities of some proteolytic enzymes



Enzyme Preferred Site^a Source

Trypsin $R_1 = Lys$, Arg From digestive systems of

Chymotrypsin $R_1 = Tyr$, Trp, Phe, Leu Same as trypsin

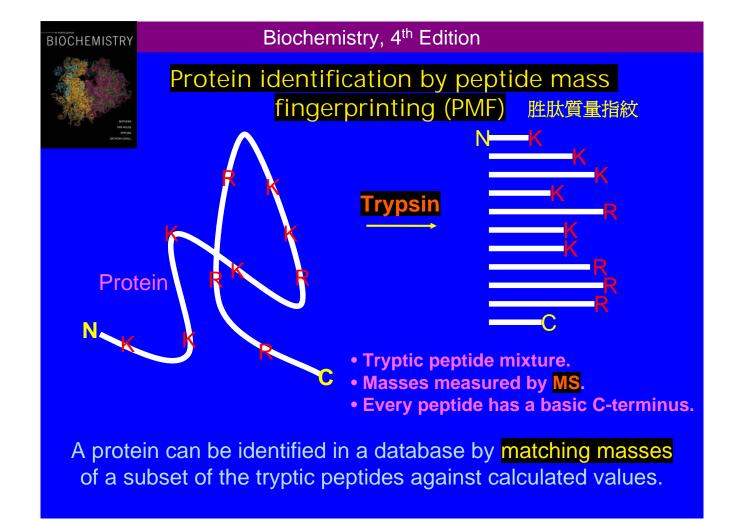
Thrombin $R_1 = Arg$ From blood; involved in coagulation

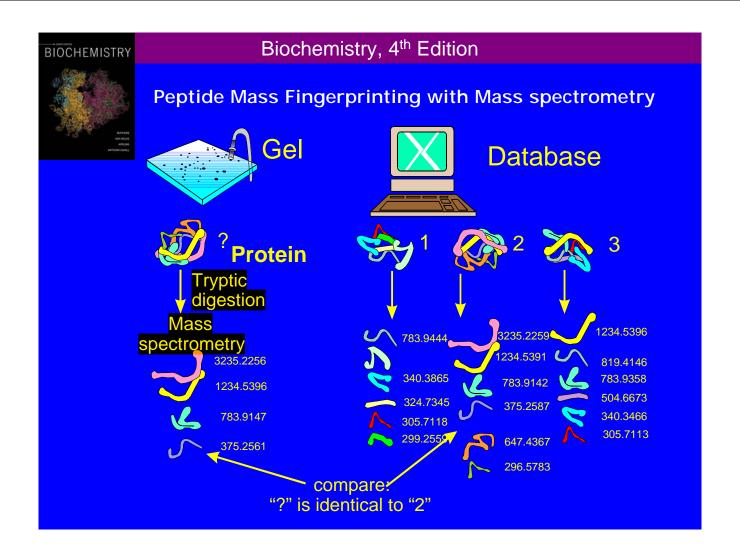
V-8 protease $R_1 = Asp$, Glu From Staphylococcus aureus Prolyl endopeptidase $R_1 = Pro$ Lamb kidney, other tissues

Subtilisin Very little specificity From various bacilli Carboxypeptidase A $R_2 = C$ -terminal amino acid From digestive systems of animals

Thermolysin R_2 = Leu, Val, Ile, Met From *Bacillus thermoproteolyticus*

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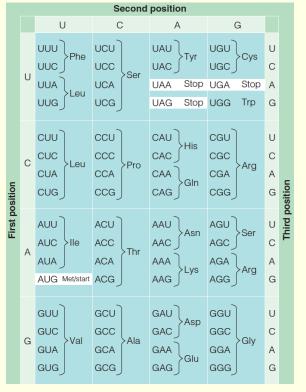




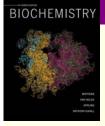


Protein sequence: from gene to protein

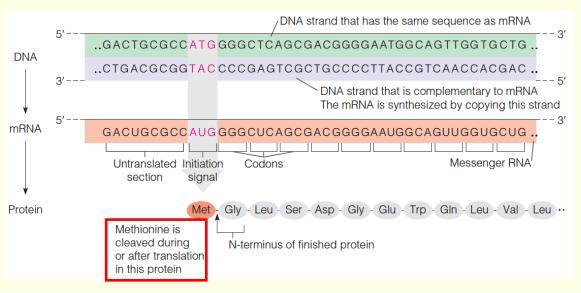
- Remember the central dogma:
 - DNA is transcribed into RNA
 - RNA is translated into protein
- Therefore, DNA codes for protein.



The genetic code



From Gene to Protein



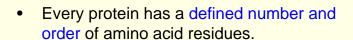
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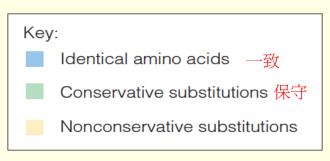
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Proteins: polypeptides of defined sequence

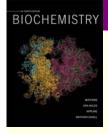


- Unique sequence (獨特序列)
- primary structure of the protein.



Whale Number 18 19 20 21 22 23 24 25 26 27 28 K V E A D I P G H G Q E V L I
K V E A D V A G H G Q D I L I Whale 33 34 35 36 37 38 39 40 41 42 43 44 45 Number Whale Number Whale 63 64 65 66 67 68 69 70 71 72 73 74 75 Number Human Whale Number Whale 91 92 93 94 95 96 97 98 99 100 101 102 103 104 105 Number Whale QSHATKHKIPIKYLE 106 107 108 109 110 111 112 113 114 115 116 117 118 119 120 Number Whale 121 122 123 124 125 126 127 128 129 130 131 132 133 134 135 Number Human Whale 136 137 138 139 140 141 142 143 144 145 146 147 148 149 150 151 152 153 Number E L F R K D M A S N Y K E L G F Q G

Sequence homology (序列列相似度) between myoglobin in humans vs. whales. 5-38



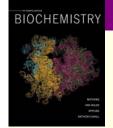
Protein Sequence Homology

- Sequence identity refers to those parts of the amino acid sequence that are an exact match. 一致度
- Sequence similarity is based on the classification of the chemical properties of the side chains such as hydrophobicity, polarity, and charge 相似度
- A widely used protocol for conducting database searches is the Basic Local Alignment Search Tool (BLAST).



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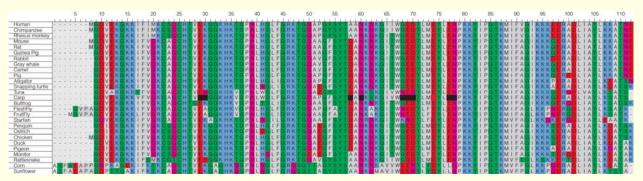
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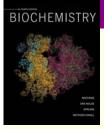
Protein Sequence Homology

- Protein sequences are classified as "homologous" in cases where any sequence similarity is thought to be the result of a common evolutionary ancestry. 同源
- **Amino acid conservation** occurs if only one amino acid is found at a given position within an alignment of homologous proteins, that amino acid is said to be absolutely "conserved".



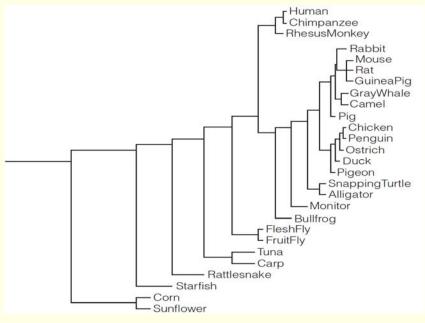
Sequence alignment and a phylogenetic tree for cytochrome *c* from different organisms.

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Sequence homology and its evolutionary relationship

A greater degree of protein sequence homology indicates a closer evolutionary relationship.



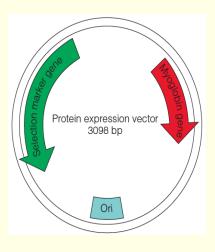
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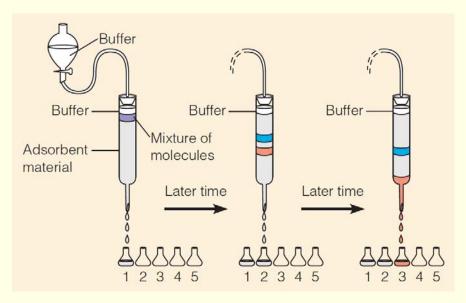
Recombinant protein expression technology 重組蛋白質表現技術



- The primary structure (sequence) of a protein corresponds to the DNA (gene) sequence.
- E. coli can be made to take up small circular DNA molecules, called "expression vectors" (表現載體).
- Plasmid (質體).



Protein Purification (I)



Proteins are generally detected by UV absorbance at 280 nm as they elute from the column.

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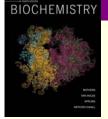
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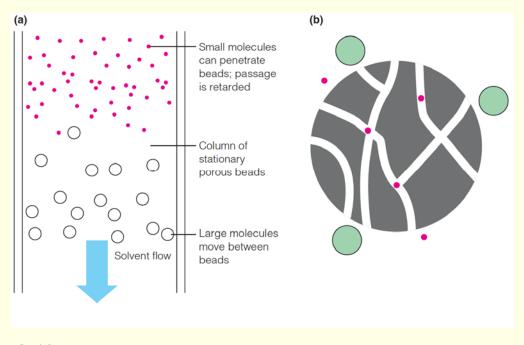
Protein Purification (II)

- Size (size-exclusion chromatography)
 分子大小,分子篩
- lonic charge (ion-exchange chromatography) 陰陽離子交換
- Affinity to ligands (affinity chromatography)
 親和層析



Size-exclusion chromatography (aka gel-filtration)

分子篩層析



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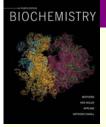
Ion-exchange chromatography

陰陽離子交換層析

lon-exchange chromatography separates molecules on the basis of their *electrical charge*.

The strength of interaction between a protein molecule and an ion exchange matrix depends on

- •the charge density on the protein
- •the *ionic strength* of the mobile phase, which is always a buffered solution.



Two main types of ion exchange matrices

Anion exchangers - carry a positive charge and bind to negatively charged proteins 陰離子交換樹酯



Cation exchangers - carry a negative charge and bind to positively charged proteins 陽離子交換樹酯



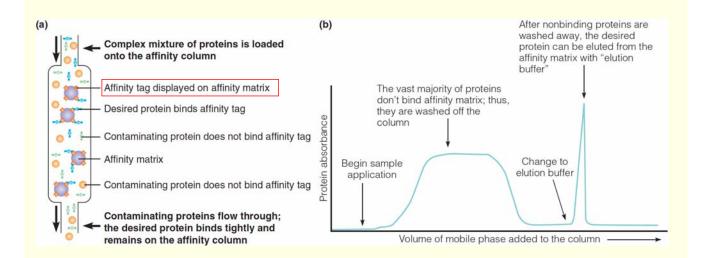
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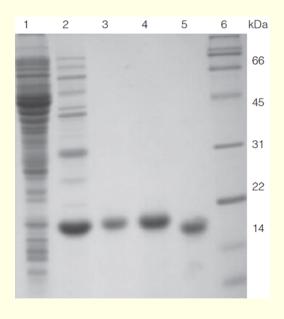
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Affinity chromatography 親和層析法





Characterization of Purified Proteins (I)



SDS- PAGE provides:

- the *purity* of the protein
- an approximate molecular weight

測定純度與分子量

Lane 1: E. coli lysate (after centrifugation)

Lane 2: IMAC purified proteins

Lane 3: 2 mg of the mutant myoglobin after SEC

Lane 4: 10 mg of the mutant myoglobin after SEC

Lane 5: SEC-purified wild-type myoglobin

Lane 6: Protein molecular weight markers

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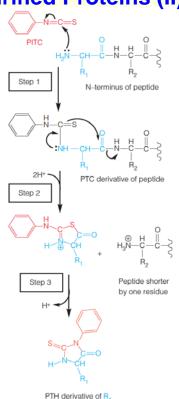
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Characterization of Purified Proteins (II)

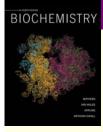
胺基酸序列分析

Unknown proteins are also characterized by their amino acid sequence.

The Edman method is based on the stepwise removal of amino acids from the *N*-terminus of a peptide by a series of chemical reactions called the "*Edman degradation*"



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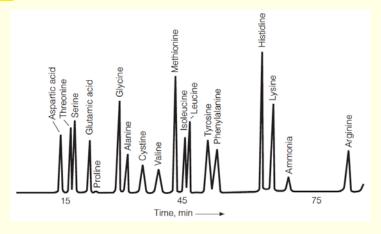
Characterization of Purified Proteins (III)

Amino acid analysis can be used to determine the amino acid composition of a protein. 胺基酸分析

Three steps are involved:

- 1. Hydrolysis of the protein to its constituent amino acids.
- 2. Separation of the amino acids in the mixture by chromatography
- 3. Quantitation of the individual amino acids.

On an anionic IEC, acidic amino acids elute first, followed by the hydrophobic, the aromatic, then the basic are last.



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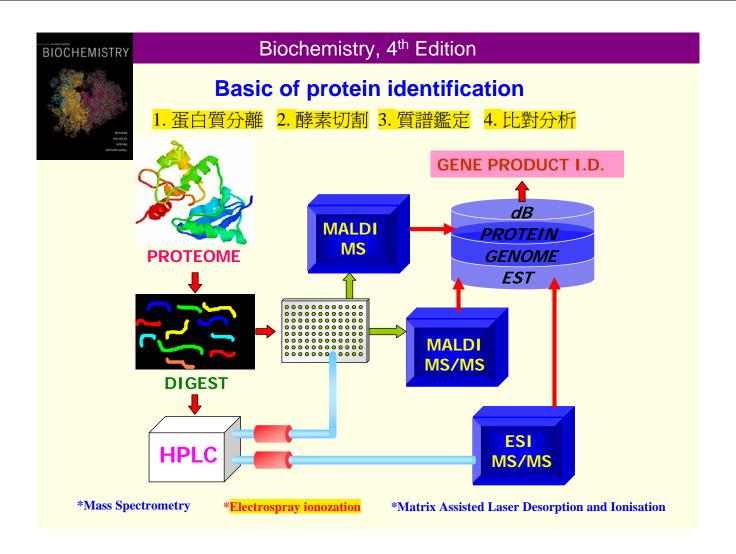
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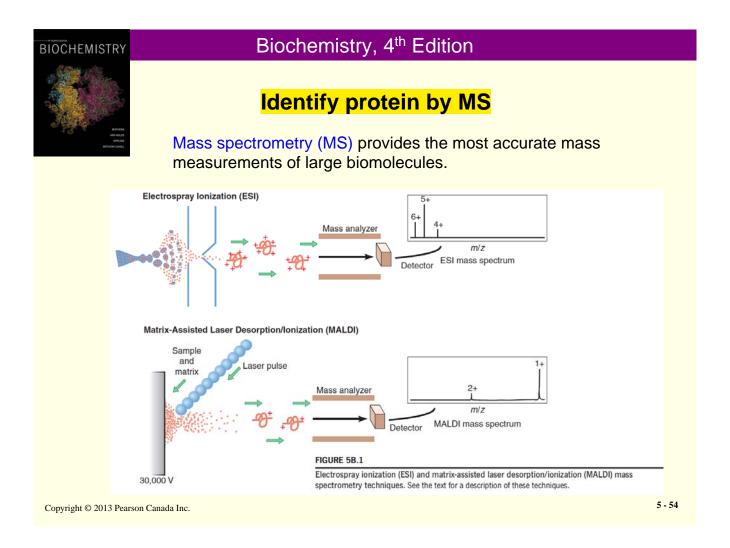
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Proteomics (蛋白質體學)

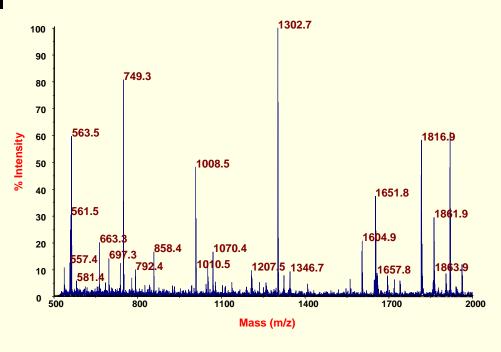
- ◆ The complement of proteins present in a given cell make up the so-called **proteome** of that cell.
- ◆ Global analysis of the proteome rather than investigating the properties of purified protein in isolation.
 - ◆ A typical proteomics experiment includes the following steps:
 - 1. Separation and isolation of proteins, or protein fragments, from cells or an organism.
 - 2. Identification by MS-MS sequencing of a particular protein within the complex mixture.
 - 3. Database searching to identify the target protein, and its putative function.

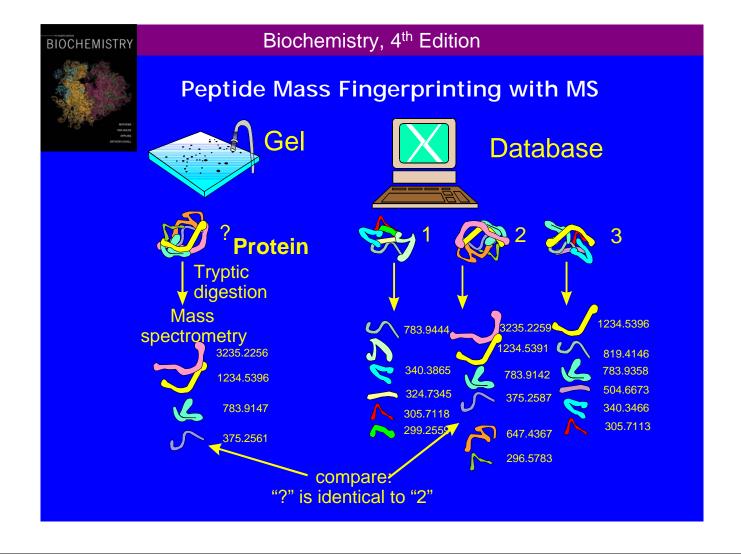




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MALDI-TOF mass spectrometry profile of tryptic digests of protein 1

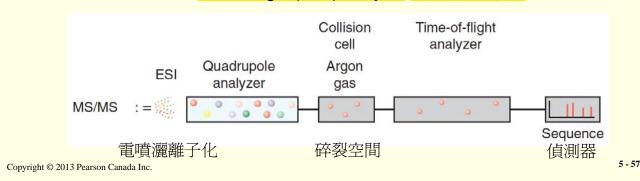


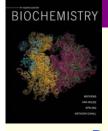


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Mass spectrometry 質譜分析

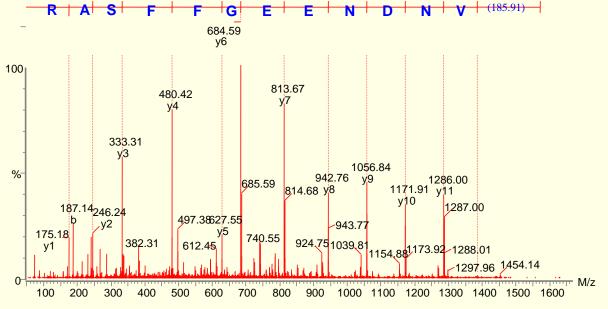
- Mass spectrometry (tandem mass spectrometry, MS-MS) can also be used to obtain peptide-sequence information. 串聯式質譜儀分析
- The mass spectrometer must have a collision cell and two mass analyzers:
 - o **quadrupole** analyzer 四極桿
 - o time-of-flight (TOF) analyzer 飛行時間分析器



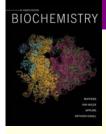


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MS/MS Spectrum of Peptide



Amino acid sequence read as differences in weight between ions in series.

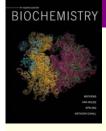


Are you understand about

- Structures of the α -Amino Acids
- Properties of the Amino Acid Side Chains
- Peptides and the Peptide Bond
- Proteins: Polypeptides of Defined Sequence
- From Gene to Protein
- Protein Sequence Homology
- Protein Expression and Purification
- Mass, Sequence, and Amino Acid Analyses of Purified Proteins

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