Scilab Textbook Companion for Engineering Physics by P. V. Naik¹

Created by
Sruthi Namburi
B-tech
Computer Engineering
CVSR college of engineering
College Teacher
Pawan Kumar
Cross-Checked by

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Book Description

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Scilab numbering policy used in this document and the relation to the above book.

Exa Example (Solved example)

Eqn Equation (Particular equation of the above book)

AP Appendix to Example(Scilab Code that is an Appednix to a particular Example of the above book)

For example, Exa 3.51 means solved example 3.51 of this book. Sec 2.3 means a scilab code whose theory is explained in Section 2.3 of the book.

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Chapter 1

Acoustics

Scilab code Exa 1.1 To calculate the intensity

Scilab code Exa 1.2 distance at which sound reduces to a level

Scilab code Exa 1.3 Determination of total absorption in hall

Scilab code Exa 1.4 average absorption coefficients of surface

Scilab code Exa 1.5 determination of frequency of ultrasonic sound waves

Scilab code Exa 1.6 To determine the thickness of crystal

```
1 clc();
2 clear;
3 //To determine the thickness of a crystal to produce
    ultrasonic waves
```

Scilab code Exa 1.7 To calculate the output power

Scilab code Exa 1.8 To calculate the output power

Scilab code Exa 1.9 To determine reverberation time

Scilab code Exa 1.10 To determine the thickness of crystal

Scilab code Exa 1.11 ratio of new to old frquencies

Chapter 2

Crystal structures

Scilab code Exa 2.1 To determine lattice parameter

```
1 clc();
2 clear;
3 //To determine the lattice parameter
                                  //density of silver
4 d=6.5*10^3;
     bromide in Kg/m<sup>3</sup>
                                  //molecular weight of
5 m = 187.8;
      silver bromide
6 M = (4*m)/(6.022*10^26);
                            //mass of ion in unit
      cell in kg. There are 4 molecules per unit cell as
      it is a fcc diatomic structure
7 //d=mass of ions in unit cell/volume of unit cell
8 / (6.5*10^3 = (1.25*10^-24) / a^3
9 a = ((M/d)^{(1/3)})*10^{10}
                                  //lattice parameter
10 printf("The lattice parameter is %f A",a);
```

Scilab code Exa 2.2 free volume per unit cell

```
1 clc();
```

Scilab code Exa 2.3 To determine ratio of vacancies

Scilab code Exa 2.4 ratio of number of vacancies to number of atoms

```
6 Ev=0.95; //average
energy required to create a vacancy
7 n=exp(-Ev/(500*k)) //n500/N
8 printf("The ratio of number of vacancies to the number of atoms is %e",n);
```

Scilab code Exa 2.5 spacing of 111 planes

Scilab code Exa 2.6 To calculate density of the structure

Scilab code Exa 2.7 To calculate atomic radius

Scilab code Exa 2.8 To determine lattice parameter

Scilab code Exa 2.9 Density of free electrons

```
1 clc();
2 clear;
3 //To determine the density of free electrons
4 rho=9000;
                                          //density in kg/m
      ^3
5 \text{ w=} 65;
                                          //atomic weight
6 v = 1;
                                          //volume in m<sup>3</sup>
                                          //number of atoms
7 n=(rho*v)/(w/(6.022*10^26));
                                          //average number
8 a=1.4;
      of free electrons per atom
9 \text{ rhoe=n*a}
                                          //density of free
      electrons per atom in electrons/m<sup>3</sup>
10 printf("The density of free electrons is %e
      electrons/m^3, rhoe);
```

Scilab code Exa 2.10 To determine lattice parameter

Chapter 4

Interference

Scilab code Exa 4.1 Wavelengths absent in reflected light

```
1 clc();
2 clear;
3 //To determine the wavelengths absent in the
      reflected light
                                               //angle of
4 i = 40;
      incidence
                                              //refractive
5 \text{ mew} = 1.2;
      index
6 r=asind(sin(i)/mew);
                                             // thickness
7 t=0.23;
      of the film
8 lambda1=(2*mew*t*cosd(r))*10^3
9 lambda2=lambda1/2;
10 printf("The wavelength absent is %f nm", lambda1);
```

Scilab code Exa 4.2 To calculate thickness of air film

```
1 clc();
```

Scilab code Exa 4.3 To calculate the bandwidth

Scilab code Exa 4.4 Refractive index of liquid

Scilab code Exa 4.5 Diameter of nineth dark ring

```
1 clc();
2 clear;
3 //To determine the diameter of the ninth dark ring
4 n=9;
                                        //wavelength of
5 lambda=589*10^-9;
     light used
6 R=0.95;
                                        //radius of
     curvature of lens
7 \text{ mew}=1;
8 D=(sqrt((4*n*lambda*R)/mew))*10^2
                                        //diameter of
     the ninth dark ring
9 printf("The diameter of the ninth dark ring is %f cm
     ",D);
```

Scilab code Exa 4.6 Wavelength of light used

Scilab code Exa 4.7 To determine thickness of the plate

```
1 clc();
2 clear;
3 //To determine the thickness of the plate
4 n=50;
                                       //number of
     fringes
                                       //wavelength of
 lambda=500*10^-9;
     light used
6 \text{ mew} = 1.5;
                                       //refractive index
      of the plate
7 t=((n*lambda)/(2*(mew-1)))*10^6
                                       //thickness of the
8 printf("The thickness of the plate is %d micro meter
     ",t);
```

Scilab code Exa 4.8 Minimum thickness of film

Scilab code Exa 4.9 Thickness of the film

```
1 clc();
2 clear;
3 //To determine the thickness of the film
4 i = 35;
                                                     //
      angle of incidence
5 \text{ mew} = 1.4;
                                                     //
      refractive index
6 r=asind(sin(i)/mew);
7 //2*mew*cos(r)=n*lambda
8 n=50;
                                                    //n
     (459) = (n+1)450
9 lambda=459*10^-9;
10 t=(n*lambda/(2*mew*0.9122214))*10^6
      thickness of the film
11 printf("The thickness of the film is %f micro meters
     ",t);
```

Scilab code Exa 4.10 To calculate the wedge angle

Scilab code Exa 4.11 Refractive index of liquid

Scilab code Exa 4.12 Wavelength of light used

```
1 clc();
2 clear;
3 //To determine the wavelength of light used
4 \text{ m=8};
      eigth ring
5 n=3;
      third ring
6 \text{ dm} = 0.4;
      diameter of the eigth ring
  dn = 0.2;
      diameter of the third ring
8 R = 101;
      Radius of curvature
  lambda=(((dm^2)-(dn^2))/(4*R*(m-n)))*10^5
      wavelength of light
10 printf ("The wavelength of light used is %fcm", lambda
      )
```

Scilab code Exa 4.13 Wavelength with high transmission

Chapter 5

Diffraction

Scilab code Exa 5.1 To determine the width of the slit

Scilab code Exa 5.2 To determine central band

Scilab code Exa 5.3 To determine width of slit

Scilab code Exa 5.4 Distance between central maximum and fourth dark fringe

```
9 printf("The distance between centres of central
    maximum and the fourth dark fringe is %f mm",d);
```

Scilab code Exa 5.5 Ratio of intensities of central and secondary maximum

Scilab code Exa 5.6 Minimum number of lines per cm

Scilab code Exa 5.7 Angle at which first minimum will be observed

```
clc();
clear;
//To calculate the angle at which first minimum will
    be observed

n=1;
lambda=650*10^-9;
//wavelength
d=2*10^-6;
//widthof
hte slit
teta=asind((n*lambda)/d)
which first minimum will be observed
printf("The angle at which first minimum will be observed is %f", teta);
```

Scilab code Exa 5.8 To determine width of the slit

Scilab code Exa 5.9 Wavelength of light used

```
1 clc();
```

Scilab code Exa 5.10 Wavelength of spectral line

Scilab code Exa 5.11 Minimum grating element required

```
1 clc();
2 clear;
```

Chapter 6

Polarisation

Scilab code Exa 6.1 To determine angle of refraction

Scilab code Exa 6.2 Percentage of incident unpolarised light

```
1 clc();
2 clear;
3 //To determine the percentage of incident
            unpolarized light
4 //I=I0(cos^2(teta))
```

Scilab code Exa 6.3 Angle between planes

Scilab code Exa 6.4 Intensities of ordinary and extraordinary light

```
1 clc();
2 clear;
3 //To compare the intensities of ordinary &
        extraordinary light
4 //IE=A^2(cos^2(teta));IO=A^2(sin^2(teta))
5 //IO/IE=tan^2(teta)
```

Scilab code Exa 6.5 Thickness of quarter wave plate

```
1 clc();
2 clear;
3 //To determine the thickness of a quarter-wave plate
4 lambda=589;
     wavelength of light in nm
                                              //refractive
5 \text{ mew0} = 1.54;
      index for ordinary wave
6 \text{ mewE} = 1.55;
                                              //refractive
      index for extraordinary wave
7 t=lambda/(4*(mewE-mew0))*10^-3
                                            //thickness
     in micro meters
8 printf("The thickness of a quarter-wave plate is %f
     micro meters",t);
```

Scilab code Exa 6.6 To determine the refractive index

Scilab code Exa 6.7 To calculate the refractive index

Scilab code Exa 6.8 Angle made by the plane of vibration of the incident light with the optic axis

```
clc();
clear;
//To determine the angle made by plane of vibration
    of the incident light with optic axis

//IE=A^2*cos^2(teta);IO=A^2*sin^2(teta)
//I0/IE=tan^2(teta)=0.65
a=0.65;
//ratio of
    intensities of ordinary & extraordinary light
teta=atand(sqrt(a))
    plane of vibration of the incident light with optic axis
printf("The angle made by the plane of vibration of incident light with the optic axis is %f",teta);
```

Scilab code Exa 6.9 Phase difference between O and E rays

Scilab code Exa 6.10 Wavelength of light incident on quartz plate

```
1 clc();
2 clear;
3 //To determine the wavelength of light incident on a
        quartz plate
4 delta=50;
        //phase difference
5 mewE=1.544;
        //refractive index of extraordinary waves
6 mew0=1.553;
```

Chapter 7

Motion of charged particle

Scilab code Exa 7.1 Kinetic energy and speed of electron

```
1 clc();
2 clear;
3 //To calculate the kinetic energy and speed of the
     electron
4 e=1.6*10^-19;
                                                //charge
     of the electron
5 V = 18;
     potential difference in kV
6 \text{ K=e*V*10^3}
                                                //Kinetic
     energy in J
7 m=9.1*10^-31;
                                                //mass of
     the electron
8 v = sqrt((2*e*V*10^3)/m)*10^-7
                                                //speed of
      electron in m/s
9 printf("The kinetic energy of electron is %e J and
     speed of the electron is %f m/s", K, v);
```

Scilab code Exa 7.2 Vector displacement of electron

```
1 clc();
2 clear;
3 //To determine the vertical displacement of electron
                                               //mass of
4 m=9.1*10^-31;
      electron
5 \text{ vx} = 4 * 10^6;
                                               //velocity
      along x-axis
                                               //electric
6 E=1500;
      field strength
  1 = 0.07;
                                               //length in
      y-axis
8 q=1.6*10^-19;
                                               //charge of
       electron
9 y=(-q*E*(1^2))/(2*m*(vx^2))*10^2
                                               //vertical
      displacement of electron
10 printf("The vertical displacement of electron when
      it leaves the electric field is %f cm",y);
```

Scilab code Exa 7.3 Time required for the proton to reach maximum height

Scilab code Exa 7.4 Orbital speed of proton

Scilab code Exa 7.5 Pitch of the helix

```
1 clc();
2 clear;
3 //To calculate the pitch of the helix and radius of
      trajectory
4 v = 2 * 10^6;
                                     //speed in m/s
                                     //angle at which
5 teta=30;
      proton enters at the origin of coordinate system
                                    //magnetic field in
6 B = 0.3;
     iΤ
                                    //v(perpendicular
7 vp=v*sind(teta);
      component)
8 vpa=v*cosd(teta);
                                     //v(parallel
     component)
9 m=1.67*10^-27;
                                    //mass of proton
10 q=1.6*10^-19;
11 p = (vpa * 2 * \%pi * m) / (q * B)
                                   //pitch of the helix
      described by the proton
12 R=((m*vp)/(q*B))*10^2
                                   //radius of the
      trajectory
```

13 printf("the pitch of the helix is %f m and radius of trajectory is %f cm",p,R)

Scilab code Exa 7.6 To calculate the displacement produced

```
1 clc();
2 clear;
3 //To determine the displacement produced , the angle
     made by the beam with the axis, velocity of
      electrons
4 V = 25;
      deflecting voltage
5 1 = 0.03;
     length of deflecting planes in m
6 d=0.75;
      distance between 2 deflecting plates
                                                   //final
7 \text{ Va} = 800;
      anode voltage
8 D=0.2;
      distance between the screen and the plates
9 y=(((V*1)/(2*d*Va))*(D+(1/2)))*10^4
                                                 //
      displacement produced
10 a=((V*1)/(2*d*Va))*10^2;
                                                   //angle
11 alpha=atand(a)
     made by the beam with the axis
12 e=1.6*10^-19;
13 m=9.1*10^-31;
                                                    //mass
     of electron
14 v = ((sqrt((2*e*Va)/m))/cosd(alpha))
      velocity of electron
15 printf ("the displacement produced is %f cm, the angle
      made by the beam with the axis is %f, velocity of
       electrons is \%e \text{ m/s}, y, alpha, v);
```

Scilab code Exa 7.7 Displacement of electron beam spot on the screen

```
1 clc();
2 clear;
3 //To determine the displacement of the electron beam
       spot on the screen
4 e=1.6*10^-19;
5 B=5*10^-5;
                                            //magnetic
      field in Wb/m<sup>2</sup>
6 1 = 0.04;
                                            //length of
      magnetic field along the axis
7 m=9.1*10^-31;
                                            //mass of
      electron
8 D=0.25;
                                            //distance of
      the screen from the field
                                            //final anode
9 \text{ Va} = 600;
      voltage
10 y=(((e*B*1)/m)*sqrt(m/(2*e*Va))*(D+(1/2)))*10^2
11 printf("the displacement of the electron beam spot
      on the screen is %f cm",y);
```

Scilab code Exa 7.8 Separation on the photographic plate

Scilab code Exa 7.9 Intensity of electric field

```
1 clc();
2 clear;
3 //To determine the intensity of electric field
4 v=5.6*10^6;
                                         //speed of the
     electron
5 m=9.1*10^-31;
                                         //mass of
     electron
6 \text{ e=1.6*10^--19};
7 s=0.03;
                                        //distance
     travelled
8 E=(m*(v)^2)/(2*e*s)
                                        //intensity of
     electric field
9 printf("The intensity of electric field is %f N/C", E
     );
```

Scilab code Exa 7.10 Charge to mass ratio of the particle

```
1 clc();
2 clear;
3 //To determine the charge to mass ratio
```

Scilab code Exa 7.11 Magnetic field required to bend a beam

```
1 clc();
2 clear;
3 //To determine the magnetic field
                                       //mass of electron
4 m=9.1*10^-31;
5 v = 3 * 10^7;
                                       //speed of
     electron
                                       //radius of the
6 R = 0.05;
     circle
7 q=1.6*10^-31;
8 B=((m*v)/(q*R))*10^-9
                                       // magnetic field
9 printf("The magnetic field to bend a beam is \% f mT",
     B);
```

Scilab code Exa 7.12 To determine the magnetic field

Scilab code Exa 7.13 To determine the magnetic field

Chapter 8

Magnetic Materials and Specroscopy

Scilab code Exa 8.1 To determine temperatue

```
1 clc();
2 clear;
3 //calculating temperatue when average thermal energy
      of an atom is equal to it's magnetic energy.
4 //The given condition is (\text{mew}*B) = 3/2(kT)
                                //magnetic dipole moment
5 \text{ mew} = 0.9 * 10^{-23};
                                 //magnetic field applied
6 B=0.72;
      in T
7 k=1.38*10^-23;
8 T = (2*mew*B)/(3*k)
                               //T=temperature in Kelvin
9 printf("The temperature at which the avg.thermal
     energy of an atom is equal to its magnetic energy
      is\%f K",T);
```

Scilab code Exa 8.2 Magnetisation of paramagnetic salt

```
clc();
clear;

// Calculating the magnetisation of paramagnetic salt

// (C=mew0*M*T)/B.
// Therefore M=(C*B)/(mew0*T)

C=2*10^-3; //C is curies constant
B=0.4; //applied magnetic field T

mew0=4*%pi*10^-7;
T=300; //temperature in kelvin
M=(C*B)/(mew0*T) //M is magnetisation
printf("%f A/m",M);
```

Scilab code Exa 8.3 Zeeman shift in wave length

```
1 clc();
2 clear;
4 //To measure Zeeman shift in wave length
6 //Zeeman shift in frequency is delta V=eB/4*Pi*m.
  //v = c/lambda and dv = -cB(lambda)^2 = mod[(c*d*lambda)/
     lambda ^ 2]
  //Therefore delta lambda=(lambda^2*delta v)/c=eB(
     lambda^2)/4*pi*m*c
9 e=1.6*10^-19;
10 B=0.35;
                                //magnetic field
11 lambda=500*10^-9;
                               //wavelength in m
12 m=9.1*(10^-31);
13 c=3*(10^8)*10^-9;
                              //speed of light
14 deltalambda=(e*B*(lambda)^2)/(4*(%pi)*m*c)
15 printf("Zeeman shift in wave length is %f nm",
     deltalambda);
```

Scilab code Exa 8.4 To determine temperature

Scilab code Exa 8.5 Magnetisation of paramagnetic salt

Scilab code Exa 8.6 To calculate the magnetic field

Scilab code Exa 8.7 To calculate the magnetic field

```
1 clc();
2 clear;
3
4 //To calculate the magnetic field
5 //e/m is gyromagnetic ratio.
6 deltalambda=0.01*10^-9;
                                                     //
     Zeeman shift
7 c=3*10^8;
     speed of light in vacuum in m/s
  lambda = 600*(10)^-9;
     wavelength in m
9 e=1.6*10^-19;
10 m=9.1*10^-31;
11 B=(deltalambda*4*(\%pi)*m*c)/(e*(lambda)^2)
                                                   //
     uniform magnetic field
12 printf("Magnetic field is %f T",B);
```

Scilab code Exa 8.8 To determine e m ratio

```
1 clc();
2 clear;
3 // To determine e/m ratio
4 //e/m = (deltalambda*4*pi*c)/(B*(lambda)^2)
5 \text{ deltalambda=0.01*10}^-9;
                                                         //
      Zeeman shift
6 c=3*(10^8);
      speed of light in vacuum in m/s
  B = 0.78;
      magnetic field
                                                        //
  lambda = 550*(10^-9);
      wavelength in nm
9 Y=(deltalambda*4*(%pi)*3*(10^8))/(B*(lambda)^2)
      /m ratio
10 printf("e/m = \%e",Y);
```

Chapter 9

Quantum Theory

Scilab code Exa 9.1 Power radiated by the filament

```
1 clc();
2 clear;
3 //To calculate the power radiated by the filament
                                           //radius of
4 r=0.05;
     the wire in mm
                                           //length of
51=4;
      the wire in cm
6 A=2*\%pi*r*l*10^-5;
                                            //in m^2
7 // According to Stephen-Boltzmann law R=e*s*(T^4)
8 / P = R * A
9 e = 1;
10 T = 3000;
     temperature in K
                                              //s is
11 s=5.6703*10^-8;
      stepfan's constant
12 p=s*(T)^4*A*e
13 printf ("The power radiated by the filament is %f W',
     p);
```

Scilab code Exa 9.2 To calculate the number of photons

```
1 clc();
2 clear;
3 //To calculate the number of photons that reach the
     surface of the earth per square cm per second
4 h=6.626*10^{-34};
     plancks constant
5 c=3*10^8;
     speed of light
6 \quad lambda=550;
     wavelength in nm
 E=(h*c)/(lambda*10^-9);
     energy of photon in J
8 \text{ Es} = 0.1/E
     number of photons per square cm per second
9 printf("The number of photons that reach the surface
      of the earth per square cm per second are %e ",
     Es);
```

Scilab code Exa 9.3 Maximum kinetic energy of photoelectron

```
9 Kmax=E-phi //maximum kinetic energy
10 printf("The maximum kinetic energy is %f eV", Kmax);
```

Scilab code Exa 9.4 To determine the potential difference

```
1 clc();
2 clear;
3 //To calcuate the potential difference
4 h=6.626*10^{-34};
                                              //plancks
     constant
                                              //speed of
5 c=3*10^8;
     light
6 lambda=175*10^-9;
     wavelength of light
7 w = 5;
                                              //work
     function of nickel
8 E=(h*c)/(lambda*1.6*10^-19);
                                              //Energy of
      200 nm photon
9 //From photoelectric equation E-w is the potential
      difference
                                              //potential
10 \, p = E - w
       difference required to stop the fastest electron
11 printf("The potential difference that should be
      applied to stop fastest photoelectron emitted by
     the surface is %f eV",p);
```

Scilab code Exa 9.5 Shortest wavelength of X rays

```
4 h=6.626*10^-34;
    plancks constant
5 c=3*10^8;
    speed of light
6 e=1.6*10^-19;
7 V=50;
    accelerating voltage in kV
8 lambdamin=((h*c)/(e*V*10^3) )*10^9
9 printf("The shortest wavelength of X-rays roduced by the machine is %f nm",lambdamin);
```

Scilab code Exa 9.6 To calculate wavelength of line

Scilab code Exa 9.7 Distance between atomic planes

```
1 clc();
2 clear;
3 //To calculate the distance between atomic planes
4 //From Bragg's law 2*d*sin(teta)=n*lambda
```

Scilab code Exa 9.8 Wavelength of X rays

```
1 clc();
2 clear;
3 //To calculate the wavelength of X-rays in the
     incident beam
4 h=6.626*10^{-34};
                                            //plancks
     constant
5 teta=50;
                                            //mass of
6 m=9.1*10^-31;
     electron
                                            //speed of
7 c=3*10^8;
     light
8 deltalambda = (h/(m*c))*(1-cosd(50))*10^12
9 lambdafin=2.5;
                                            //wavelength
     of scattered X-rays
10 lambdainit=lambdafin-deltalambda
11 printf("The wavelength of X-rays in the incident
     beam is %f pm", lambdainit);
```

Scilab code Exa 9.9 To determine the power of laser

```
1 clc();
2 clear;
```

```
3 //To determine the power of the laser
4 h=6.626*10^{-34};
                                             //plancks
     constant
                                             //speed of
5 c=3*10^8;
     light
6 lambda=500;
                                             //wavelength
      of laser
                                             //Energy of
7 E=(h*c)/lambda;
     500 nm photon
                                             //number of
8 N=2.52*10^16;
     photons in a 20ms pulse
9 p=((E*N)/(20*10^-3))*10^9
10 printf("The power of the laser is %f W",p);
```

Scilab code Exa 9.10 Work function of the surface

Scilab code Exa 9.11 To calculate the accelerating voltage

```
1 clc();
2 clear;
3 //To calculate the accelerating voltage
4 h=6.626*10^{-34};
                                            //plancks
     constant
                                            //speed of
5 c=3*10^8;
     light
6 e=1.6*10^-19;
 lambdamin=0.02*10^-9;
                                            //minimum
     wavelength in nm
8 V=((h*c)/(lambdamin*e))*10^-3
     accelerating voltage
9 printf("The accelerating voltage needed to produce
     minimum wavelength of 0.02 nm is %f kV", V);
```

Scilab code Exa 9.12 Angle of second order Braggs reflections

```
1 clc();
2 clear;
3 //To determine the angle of second order bragg's
      reflections
4 // According to Bragg's eq.2*d*sin(teta)=n*lambda
5 n=2;
                                      //since second
     order Bragg's eq.
6 d=5;
                                      //since d=5(lambda
7 lambda=1;
8 a=(n*lambda)/(2*5*lambda);
                                     //angle of second
9 teta=asind(a)
     order Braggs reflections
10 printf ("The angle of second order Braggs reflection
     is %f", teta);
```

Scilab code Exa 9.13 Accelerating voltage applied

```
1 clc();
2 clear;
3 //To determine the accelerating voltage applied
4 h=6.626*10^{-34};
                                             //plancks
     constant
                                             //speed of
5 c=3*10^8;
     light
                                             //wavelength
6 lambda=0.03;
      in nm
7 E=(h*c)/(lambda*10^-9);
                                             //energy of
     photon
                                             //Total
8 TE=((E*100)/80);
     energy.E=80\% of TE in J
                                                     //
9 TE=TE/(16.0017*10^-17)
     Total energy in kV
10 printf("The electron must have been accelerated
     through a potential difference of %f kV", TE);
```

Chapter 10

Quantum Mechanics

Scilab code Exa 10.1 To calculate de Broglie wavelength

Scilab code Exa 10.2 To determine the kinetic energy

```
//de Broglie
5 \quad lambda=0.3;
      wavelength
6 p=h/(lambda*10^-9);
      uncertainity in determining momentum in kg.m/s
7 //For electron
8 me=9.1*10^-31;
                                           //mass of
      electron
                                             //kinetic
9 K1=p^2/(2*me);
      energy of electron in J
10 //For proton
11 mp=1.672*10^-27;
                                           //mass of
     proton
12 K2=p^2/(2*mp)
                                           //kinetic
      energy of proton in J
13 printf("The kinetic energy of electron is %e J and
      kinetic energy of proton is %e J", K1, K2);
```

Scilab code Exa 10.3 To calculate the kinetic energy

```
1 clc();
2 clear;
3 //To determine the kinetic energy
4 //K=p^2/(lambda^2*2*m) where K is kinetic energy
5 h=6.626*10^{-34};
                                            //plancks
     constant
                                            //de Broglie
  lambda=10^-14;
      wavelength
                                            //mass of
7 m=9.1*10^-31;
     electron
8 K=(h^2/((lambda^2)*2*m*1.6*10^-19))*10^-9
9 printf("The kinetic energy is %f Gev", K);
10 //It is not possible to confine the electron to a
     nucleus. The experimental observations show that
     even electrons assosciated with unstable atoms
     never have energy more than a fraction of MeV
```

Scilab code Exa 10.4 Uncertainty in determining the position

```
1 clc();
2 clear;
3 //To determine the uncertainity in determining the
      position of this electron
                                            //mass of
4 m=9.1*10^-31;
      electron
                                            //speed of
5 v = 6 * 10^3;
      electron
6 p=m*v;
                                            //uncertainity
      in momentum in kg.m/s
7 h=6.626*10^{-34};
                                            //plancks
      constant
8 \text{ deltap=0.00005*p};
                                            //uncertainity
       in p is 0.00005 of p
9 deltax=(h/(4*%pi*deltap))*10^3
                                          //uncertainity
      in position
10 printf ("The uncertainty in position is %f mm",
     deltax);
```

Scilab code Exa 10.5 First 2 energy levels of a particle

```
1 clc();
2 clear;
3 //To determine the first two energy levels using
    particle-in-a-box model
4 L=3*10^-5;
    //
    diameter of the sphere
```

Scilab code Exa 10.6 Ground state and first excited state energy

```
1 clc();
2 clear;
3 //To determine ground state & first excited state
     energy
                                              //plancks
4 h=6.626*10^{-34};
     constant
                                              //angular
5 a=2*10^12;
     frequency in rad/s
6 E0 = (0.5*(h/(2*\%pi*1.6*10^-19))*a)*10^3
                                             //ground
     state energy
7 E1=(1.5*(h/(2*\%pi*1.6*10^-19))*a)*10^3
                                             //first
     excited state energy
8 printf("The ground state energy is %f Mev and first
     excited state energy is %f Mev", E0, E1);
```

Scilab code Exa 10.7 To determine kinetic energy of electron

```
1 clc();
2 clear;
3 //To determine the kinetic energy of electron
4 h=6.626*10^{-34};
                                               //plancks
     constant
5 E=85;
                                               //Energy
     in keV
                                               //speed of
6 c=3*10^8;
      light
  lambda=(h*c)/(E*10^3*(1.6*10^-19));
                                               //de
     Broglie wavelength
8 m=9.1*10^-31;
                                               //mass of
     electron
9 K=((h^2)/((lambda^2)*2*m*1.6*10^-9))*10^7 //kinetic
     energy of electron
10 printf("The kinetic enery of the electron is %f keV"
     , K);
```

Scilab code Exa 10.8 To determine velocity of electron

```
1 clc();
2 clear;
3 //To determine velocity of an electron
4 lambda=0.08;
    //de Briglie wavelength
5 m=9.1*10^-31;
    //mass of electron
6 h=6.626*10^-34;
    plancks constant
7 v=h/(m*lambda*10^-9)
    velocity of the electron
8 printf("The velocity of the electron is %f m/s",v);
```

Scilab code Exa 10.9 To determine the potential difference

```
1 clc();
2 clear;
3 //To determine the potential difference
4 h=6.626*10^{-34};
                                        //plancks
     constant
5 \quad lambda = 589 * 10^{-9};
                                        //wavelength in m
6 m=9.1*10^-31;
                                        //mass of
     electron
7 e=1.6*10^-19;
8 V=((h^2)/((lambda^2)*2*m*e))*10^6 //potential
     diference
9 printf("The potential difference through which an
     electron should be accelerated to have a
     wavelength of 589 nm is %f microV ",V);
```

Scilab code Exa 10.10 Uncertainty in velocity of electron

Scilab code Exa 10.11 Length of the box for a proton

```
1 clc();
2 clear;
3 //To determine the length of a box
4 h=6.626*10^{-34};
                                                    //
     plancks constant
5 n=3;
     for second excited state
6 m=1.67*10^-27;
     mass of proton
                                                     //
7 E=0.5;
     energy in Mev
8 L=((h*n)/sqrt(8*m*E*10^6*1.6*10^-19))*10^15
                                                     //
     length of the box
9 printf("The length of the box for proton in its
     second excited state is %f fm ",L);
```

Chapter 11

Nuclear Radiations and Detectors

Scilab code Exa 11.1 To calculate the radius of Li

Scilab code Exa 11.2 Binding energy per nucleon

```
1 clc();
2 clear;
3 //To determine the binding energy per nucleon of U
    with mass no. 235
```

```
//atomic mass of uranium
4 M = 235.043945;
                            //atomic number of uranium
5 Z=92;
6 mp=1.007825;
                                                       //mass
      of proton
7 N = 143;
                                                      //no.of
      neutrons
                                                      //mass
8 \text{ mn} = 1.008665;
      of neutron
9 \quad A = 235;
                                                     //number
      of nucleons
10 B = [[(Z*mp) + (N*mn) - (M)]/A]*931.5
                                                     //Binding
       energy in MeV
11 printf ("The binding energy per nucleon is %f MeV", B)
```

Scilab code Exa 11.3 minimum energy required to remove a neutron from nucleus

```
1 clc();
2 clear;
3 //To calculate the minimum energy required to remove
       a neutron from Ca(A=43;Z=20)
4 // After removing one neutron from Ca(A=43;Z=20) it
      becomes Ca(A=42;Z=20)
                                    //\text{mass} of Ca(A=42;Z=20)
5 M = 41.958622;
                                    //mass of neutrom
6 \text{ mn} = 1.008665;
7 C=M+mn;
                                    //\text{mass} of \text{Ca}(A=43;Z=20)
8 E=42.95877;
9 D=C-E;
                                    //Binding energy of
10 B = D * 931.5
      neutron
11 printf("The binding energy of neutron is %f MeV",B);
```

Scilab code Exa 11.4 To determine the Q value of the reaction

```
1 clc();
2 clear;
3 //To determine the Q-value
4 mBe=9.012182;
                                    //Atomic mass of
     beryllium in u
5 \text{ mHe} = 4.002603;
                                   //Atomic mass of helium
                                   //mass of neutron
6 \text{ mn} = 1.008665;
7 \text{ mC} = 12.000000;
                                   //Atomic mass of carbon
8 Q = (mBe + mHe - mn - mC) * 931.5
                                   //Q is called energy
     balance of the reaction
9 printf("The Q-value is %f MeV",Q);
```

Scilab code Exa 11.5 Q value of the reaction and energy of each alpha particle

```
1 clc();
2 clear;
3 //To determine the Q-value
4 mLi=7.016004;
                                 //mass of Lithium (A=7)
                                 //mass of Hydrogen (A=1)
5 \text{ mH} = 1.007825;
                                 //mass of helium (A=4)
6 mHe=4.002603;
7 Q = [mLi + mH - 2*(mHe)]*931.5
                                  //Q is the energy
      balance of the reaction
                                 //energy of proton in
8 p=0.5;
     MeV
  //The energy of 2 alpha particles is equal to the Q-
      value + energy of proton.
                                 //energy of each alpha
10 Ealpha=(Q+p)/2
      particle
11 printf ("The Q-value of the reaction is %f MeV and
      energy of each alpha particle is %f MeV",Q,Ealpha
```

Scilab code Exa 11.6 To calculate the energy released

```
1 clc();
2 clear;
3 //To determine the amount of energy released
                                            //weight in gm
4 \text{ wt} = 1000;
5 A = 235;
                                            //mass number of
      uranium
6 N = [6.02*(10^23)/A]*wt;
                                            //no.of nuclei in
      1kg of uranium
                                            //energy-balance
7 Q = 208;
      of the reaction
                                            //Energy released
8 \quad E = N * Q;
      in MeV
9 / 1 \text{MeV} = 4.45 * 10^{-20} \text{kWh}
10 E=E*4.45*(10^-20)
11 printf("The energy released is %f kWh", E);
```

Scilab code Exa 11.7 power output of a nuclear reactor

```
1 clc();
2 clear;
3 //To determine the power output of a nuclear reactor
4 \text{ wt} = 5000;
                                       //weight in gm
                                       //mass number of
5 A = 235;
     uranium
6 N = (6.02*(10^23)/A)*wt;
                                       //number of nuclei
     in 5 Kg
                                       //Energy released
7 Ef = 208;
     per fission in MeV
8 \quad E=N*Ef;
                                       //Energy in MeV
                                        //Energy in J
9 E=E*1.6*(10^-13);
```

Scilab code Exa 11.8 Amount of U required

```
1 clc();
2 clear;
3 //To calculate the amount of Uranium (A=235) required
4 A = 235;
                                        //mass number of
     uranium
                                        //amount of
5 p=1000;
      electric power produced
6 e=0.32;
                                        //energy
      conversion efficiency of the plant
7 I=p/e;
                                        //Input nuclear
      energy in MW
8 f = 200;
                                        //fission energy
     per event in MeV
  TE=I*(10^6)*3600*24*365;
                                        //total energy in
      J
10 EF=f*(10^6)*1.6*(10^-19);
                                        //Energy released
      per fission event in J
11 N=TE/EF;
                                        //Number of
      nuclei required
12 M = (N*A)/[6.02*(10^23)]*10^-3
                                        //corresponding
     mass in kg
13 printf("The amount of uranium required is %f kg",M);
```

Scilab code Exa 11.9 Frequency of oscillator to be used

```
1 clc();
```

```
2 clear;
3 //To determine the frequency of oscillator to be
     used to accelerate protons
                                         //charge of the
4 q=1.6*(10*10^-19);
      particle
5 B=1;
                                         //magnetic field
      in T
6 m=1.67*(10^-27);
                                         //mass of proton
      in kg
  omega=(q*B)/m;
                                         //angular
     frequency in radians/s
8 v = [omega/(2*\%pi)]*10^-8;
                                         //in MHz
9 r = 0.5;
                                         //radius in m
10 s=omega*r;
                                         //speed of
     proton in m/s
11 K = [m*(s^2)]*(1/2)*6.27*10^10
                                         //kinetic energy
      of protons emerging from cyclotron
12 printf("The frequency of oscillator to accelerate
     protons is %f radians/s ,speed of proton is %f m/
     s and the kinetic energy of protons emerging from
      the cyclotron is %f MeV", omega, s, K);
```

Scilab code Exa 11.10 To calculate the radius

Scilab code Exa 11.11 To calculate the cyclotron frequency

```
1 clc();
2 clear;
3 //To calculate cyclotron frequency of of electron
4 q=1.6*(10^-19);
                                     //charge of the
     particle
5 B=5;
                                     //magnetic field in
                                     //mass of electron
6 m=9.1*(10^-31);
     in kg
7 v = (q*B)/(2*\%pi*m)
                                     //cyclotron
     frequency in Hz
8 printf("cyclotron frequency of of electron is %e Hz"
     , v);
```

Scilab code Exa 11.12 To determine the magnetic field

Scilab code Exa 11.13 To determine the magnetic field

```
1 clc();
2 clear;
3 //To calculate magnetic field
4 m=1.67*(10^-27);
    //mass of proton
5 q=1.6*10^-19;
    //charge of particle
6 v=25;
    //cyclotron frequency in MHz
7 //v=(q*B)/(2*pi*m). Therefore B=(v*2*pi*m)/q
8 B=(v*10^6*2*%pi*m)/q
9 printf("The required magnetic field is %f T",B);
```

Scilab code Exa 11.14 Charge to mass ratio for proton

8 printf("q/m=%f C/kg",d);

Chapter 13

Optical Fibre

Scilab code Exa 13.1 Numerical aperture of a fibre

Scilab code Exa 13.2 To determine the acceptance angle

Scilab code Exa 13.3 Number of guided modes

```
1 clc();
2 clear;
3 //To determine the number of guided modes
4 NA = 0.25;
                                        //numerical
     apperture
                                        //wavelength in
 lambda=0.75;
     micro metres
6 a = 25;
                                        //core radius i.e
     50/2 = 25
7 f = (2*\%pi*a*NA)/lambda;
                                         //normalised
     frequency
8 \text{ Ng}=(f^2)/2
                                         //number of
     guided modes
9 printf("The number of guided modes is %f", Ng);
```

Scilab code Exa 13.4 Signal attenuation per kilometer

Scilab code Exa 13.5 Numerical aperture for the optical fibre

Scilab code Exa 13.6 Number of guided modes

Scilab code Exa 13.7 To determine the mean optical power

```
1 clc();
2 clear;
3 //To calculate the mean optical power launched into
     a fibre
4 S=2;
                            //signal attenuation per km
                            //length in km. Since l=1 S=2
5 1 = 1;
6 //S=10*log(pi/p0)(or)pi/p0=10^(S/10)
7 p0=20;
                            //mean optical power at
     fibre output
8 \text{ pi=p0*10}^{(S/10)}
                            //mean optical power
     launched into fibre
9 printf("The mean optical power launched into a fibre
      of length of 1km is %f micri W',pi);
```

Scilab code Exa 13.8 To determine the mean optical power

Scilab code Exa 13.9 To calculate signal attenuation

```
1 clc();
2 clear;
3 //To determine signal attenuation
4 //given condition:po=pi/4where p0=mean optical power
    at fibre output, pi=mean optical power launched
    into fibre
5 //S=10*log(pi/po)
6 S=10*log10(1/(1/4)) //signal attenuation in
    dB
7 printf("Signal attenuation when po=pi/4 is %f dB",S)
    ;
```