Scilab Textbook Companion for Modern Physics by K. S. Krane¹

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Book Description

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Scilab numbering policy used in this document and the relation to the above book.

Exa Example (Solved example)

Eqn Equation (Particular equation of the above book)

AP Appendix to Example(Scilab Code that is an Appednix to a particular Example of the above book)

For example, Exa 3.51 means solved example 3.51 of this book. Sec 2.3 means a scilab code whose theory is explained in Section 2.3 of the book.

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Chapter 1

Introduction

Scilab code Exa 1.1 Mass difference between a proton and a neutron

Scilab code Exa 1.2 Total mass of proton and electron

```
1 clear
2 clc
3 disp('Exa-1.2');
4 Mp=1.007276; Me=5.4858*10^-4; //mass of proton and electron in terms of U
5 Mt=Mp+Me; //Total mass= sum of above masses
6 printf('The combined mass of an electron and a proton was found out to be %f U.', Mt);
```

Scilab code Exa 1.3 Value of hc

Chapter 2

The Special Theroy of Relativity

Scilab code Exa 2.1 Speed of A wrt B

Scilab code Exa 2.2 Velocity of plane wrt ground

```
1 clear
2 clc
3 disp('Exa-2.2');
```

Scilab code Exa 2.3 Time taken in each case

```
1 clear
2 clc
3 disp('Exa-2.3'); // The problem is entirely
          theoretical hence following the standard
          procedure we obtain
4 printf('The time required for round trip is 2*L/(c
         *(1-(u/c)^2)). \n');
5 printf('The time required to swim across and return
          is 2*L/(c*sqrt((1-(u/c)^2)))');
```

Scilab code Exa 2.4 Minimum speed for suvival of muons

```
1 clear
2 clc
3 disp('Exa-2.3');
```

```
4 Lo=100*(10^3); c=3*(10^8); //Given values//all the
    quantities are converted to SI units
5 d=2.2*(10^-6); //time between its birth and
    decay
6 t=Lo/c //where Lo is the distance from
    top of atmosphere to the Earth. c is the velocity
    of light. t is the time taken
7 u=sqrt(1-((d/t)^2)); // using time dilaion fromula
    for finding u where u is the minimum velocity in
    terms of c;
8 printf('Hence the minimum speed required is %f c.',u
);
```

Scilab code Exa 2.5 Apparent thickness of Earth atmosphere

```
1 clear
2 clc
3 disp('Exa-2.5');
4 Lo=100*(10^3); //Lo is converted to Km
5 u=0.999978; ///u/c is taken as u since u is represented in terms of c.
6 L=Lo*(sqrt(1-u^2)); // from the length contraction formula
7 printf('Hence the apparent thickness of the Earth''s surface is %.2 f metres.',L);
```

Scilab code Exa 2.6 Solution for a b c d e

```
1 clear all
```

```
2 clc
3 disp('Exa - 2.6(a)');
4 L=65; c=3*10^8; u=0.8*c;
5 t=L/u; //The value of time taken as
     measured by the observer
6 printf ('The time for rocket to pass a point as
     measured by O is \%.2e.\n',t);
                                            //The value
      of time taken as measured by the observer
7 disp('Exa-2.6(b)');
8 \text{ Do} = 65;
                                       //given length
9 Lo= L/sqrt(1-(u/c)^2);
                                           //contracted
     length of rocket
10 printf ('Actual length according to O is \%.2 f.\n',Lo)
11 disp('Exa-2.6(c)');
                                //contracted length of
12 D=Do*(sqrt(1-(u/c)^2));
      platform.
13 printf ('Contracted length according to O'' is %.2e.\
     n',D);
14 disp('Exa-2.6(d)');
15 t1=Lo/u;
                               //time needed to pass
     according to O'.
16 printf('Time taken according to O is \%.2e.\n',t1);
17 disp('Exa-2.6(e)');
18 t2=(Lo-D)/u;
                                   //time intervals
     between the two instancs
19 printf('Time taken according to O'' is \%.2e.\n',t2);
20 disp('The value of t1 and t2 did not match');
```

Scilab code Exa 2.7 Speed of missile wrt earth

```
1 clear all 2 clc
```

```
3 disp('Exa-2.7');
4 v1=0.6; u=0.8; c=1; // all the values are measured
    in terms of c hence c=1
5 v= (v1+u)/(1+(v1*u/c^2));
6 printf('The speed of missile as measured by an
    observer on earth is %.2f c.',v);
```

Scilab code Exa 2.8 Speed of galaxy wrt earth

Scilab code Exa 2.9 Velocity of rocket2 wrt rocket1

```
7 printf('The velocity of rocket 2 wrt rocket 1 along
    x and y directions is %.2f c & %.2f c
    respectively', v21x, v21y);
```

Scilab code Exa 2.10 Time interval between the events

Scilab code Exa 2.11 momentum of proton

Scilab code Exa 2.12 Various energies of proton

```
clear all
clc
disp('Exa-2.12');
pc=1580; mc2=938; E0=938;  // all the energies in
    MeV mc2=m*c^2 and pc=p*c

E=sqrt(pc^2+mc2^2);
printf('The relativistic total energy is %.2f MeV.\n',E);  //value of Energy E

K=E-E0;  //value of possible kinetic energy
printf('The kinetic energy of the proton is %.1f MeV', K);
```

Scilab code Exa 2.13 Velocity and momentum of electron

```
1 clear all 2 clc 3 disp('Exa-2.13'); 4 E=10.51; mc2=0.511; //all the values are in MeV 5 p=sqrt(E^2-mc2^2); //momentum of the electron 6 printf('The momentum of electron is \%.1 \, f \, MeV/c \ ;
```

```
7 v=sqrt(1-(mc2/E)^2); //velocity in terms of c
8 printf('The velocity of electron is %.4f c',v);
```

Scilab code Exa 2.14 Solution for a b

```
1 clear all
2 clc
3 disp('Exa-2.14');
4 k=50; mc2=0.511*10^-3; c=3*10^8; // all the values of
     energy are in GeV and c is in SI units
5 \text{ v=sqrt}(1-(1/(1+(k/mc2))^2)); //speed of the
     electron in terms of c
6 k=c-(v*c);
                                      //difference in
     velocities
7 printf('Speed of the electron as a fraction of c is
      \%.12 \text{ f}*10^-12.\text{ n'}, v*10^-12); // v=(v*10^-12)
     *10^--12; so as to obtain desired accuracy in the
     result
8 printf('The difference in velocities is %.1f cm/s.',
     k*10^2);
```

Scilab code Exa 2.15 Rate of decrese of the mass of Sun

Scilab code Exa 2.16 Kinetic energy of pion in each case

```
1 clear all
2 clc
3 disp('Exa - 2.16');
4 K=325; mkc2=498; //kinetic energy and rest mass
     energy of kaons
5 mpic=140; //given value
6 Ek=K+mkc2;
7 pkc=sqrt(Ek^2-mkc2^2);
8 //consider the law of conservation of energy which
      yields Ek=sqrt(p1c^2+mpic^2)+sqrt(p2c^2+mpic^2)
9 // The above equations (4th degree, hence no direct
     methods) can be solved by assuming the value of
     p2c=0.
10 p1c=sqrt(Ek^2-(2*mpic*Ek));
11 //consider the law of conservation of momentum.
      which gives p1c+p2c=pkc implies
12 p2c=pkc-p1c;
13 k1=(sqrt(p1c^2+(mpic^2))-mpic); //corresponding
     kinetic energies
14 k2=(sqrt((p2c^2)+(mpic^2))-mpic);
15 printf ('The corresponding kinetic energies of the
     pions are \%.0 f MeV and \%.1 f MeV.', k1, k2);
```

Scilab code Exa 2.17 Threshold kinetic energy to produce antiprotons

```
1 clear all
2 clc
3 \text{ disp}('Ex-2.17');
                        //mpc2=mp*c^2,mp=mass of proton
4 \text{ mpc2=938; c=3*10^8;}
                        //final total energy
5 Et=4*mpc2;
6 E1=Et/2; E2=E1;
                         //applying conservation of
     momentum and energy
7 v2=c*sqrt(1-(mpc2/E1)^2);
                              //lorentz
      transformation
8 u=v2; v=(v2+u)/(1+(u*v2/c^2));
9 E=mpc2/(sqrt(1-(v/c)^2));
10 K=E-mpc2;
11 printf('The threshold kinetic energy is \%.3f Gev', K
      /10^3);
```

Chapter 3

Review of Electromagnetic waves

Scilab code Exa 3.1 Atomic Spacing of Nacl

Scilab code Exa 3.2 Time taken to release an electron

```
1 clear
2 clc
3 disp('Exa-3.2');
4 I=120;r=0.1*10^-9; Eev=2.3 //I-intensity in W/m^2 r
        in m & E in electron volt
```

Scilab code Exa 3.3 Solution for a and b

```
1 clear
2 clc
3 disp('Exa-3.3(a)');
4 w=650*10^-9; h=6.63*10^-34; c=3*10^8; //given values
      and constant taken in comfortable units
5 E=h*c/w; printf('The Energy of the electron is \%.3e
      J ',E);
6 E=E/(1.6*10^-19); printf('which is equivalent to %f
      \mathrm{eV} \backslash \mathrm{n} ',E);
7 printf('The momentum of electron is p=E/c i.e %.2f/c
       n', E);
8 disp('Exa-3.3(b)');
                                        //given energy of
9 E2=2.40;
      photon.
10 w2=h*c*10^9/(E2*1.6*10^-19);
                                       //converting the
      energy in to eV and nm
11 printf('The wavelength of the photon is %.2f nm', w2)
```

Scilab code Exa 3.4 Solution for a b and c

```
1 clear
2 clc
3 disp('Exa-3.4(a)');
                                        //both the values
4 hc=1240; phi=4.52
      are in eV
5 \text{ w1=hc/phi};
6 printf ('The cutoff wavelength of the tungsten metal
      is \%.3 \text{ fnm} \  ', w1);
7 disp('Exa-3.4(b)');
               //given value of wavelength
8 \text{ w}2=198;
9 Kmax=(hc/w2)-phi; printf('The max value of kinetic
      energy is \%.3 \, \text{f eV} \setminus \text{n',Kmax};
10 disp('Exa - 3.4(c)');
11 Vs=Kmax; printf('The numerical value of the max
      kinetic energy is same as stopping potential in
      volts. Hence %.2 f V', Vs);
```

Scilab code Exa 3.5 Solution for a b and c

```
1 clear
2 clc
3 disp('Exa-3.5(a)');
4 T1=293; Kw=2.898*10^-3;
5 w1=Kw/T1;
6 printf('The wavelength at which emits maximum radiation is %.2 f um.\n',w1*10^6);
7 disp('Exa-3.5(b)');
8 w2=650*10^-9;
9 T2=Kw/w2;
10 printf('The temperature of the object must be raised to %.0 f K.\n',T2);
11 disp('Exa-3.5(c)');
12 x=(T2/T1)^4; printf('Thus the thermal radiation at
```

```
higher temperature is %.2e times the room (lower) temperature.\n',x);
```

Scilab code Exa 3.6 Solution for a b c and d

```
1 clear
2 clc
3 disp('Exa - 3.6(a)');
4 w1=0.24; wc=0.00243; theta=60; //given values w=
      wavelength (lambeda)
5 \text{ w2=w1+(wc*(1-cosd(theta)))};
6 printf ('The wavelength of x-rays after scattering is
       \%.4 \text{ f nm} \text{ n', w2};
7 disp('Exa-3.6(b)');
8 \text{ hc} = 1240;
9 E2=hc/w2;E1=hc/w1; printf('The energy of scattered x
      -rays is \%.0 \text{ f eV} \cdot \text{n',E2};
10 disp('Exa - 3.6(c)');
11 K= E1-E2; //The kinetic energy is the difference in
      the energy before and after the collision;
12 printf ('The kinetic energy of the x-rays is \%.3 f eV\
      n',K);
13 disp('Exa - 3.6(d)');
14 phi2=atand(E2*sind(theta)/(E1-E2*cosd(theta)))
15 printf ('The direction of the scattered eletron is %
      .1f degrees', phi2);
```

Chapter 4

The Wavelike properties of particles

Scilab code Exa 4.1 Solution for a b c d and e

```
1 clear
2 clc
3 disp("Ex: 4.1 ");
                                                   // h(
4 h=6.6*10^-34;
      planck 's constant) = 6.6*10^-34
                                                      // for
5 \text{ m1} = 10^3; v1 = 100;;
      automobile
                                                  // [ 'w'-
6 \text{ w1= h/(m1*v1)};
      wavelength in metre 'm'-mass in Kg 'v'-velocity in
       metres/sec.] of the particles
7 printf("Wavelength of the automobile is %1.2e m\n",
      w1 );
                                                     // for
8 m2=10*(10^-3); v2=500;
      bullet
9 w2=h/(m2*v2);
10 printf("Wavelength of the bullet is \%1.2e \text{ m/n}", w2)
11 m3=(10^-9)*(10^-3); v3=1*10^-2;
12 \text{ w3=h/(m3*v3)};
```

```
13 printf ("Wavelength of the smoke particle is %1.2e m\
      n", w3);
14 \text{ m4} = 9.1 * 10^{-31}; k = 1 * 1.6 * 10^{-19};
      kinetic energy of the electron & using 1ev =
      1.6*10^{-19} joule
      momentum of electron ; from K=1/2*m*v^2 = h/p:
15 p = sqrt(2*m4*k);
16 \text{ w4=h/p};
17 printf ("Wavelength of the electron (1ev) is %1.2 fnm\n
      ",w4*10^9);
18 hc = 1240; pc = 100
                                                 // In the
      extreme relativist realm, K=E=pc; Given pc=100
      MeV, hc = 1240MeV
19 \text{ w5= hc/pc};
20 printf ("Wavelength of the electron (100 Mev) is %1.2 f
       fm \setminus n", w5);
```

Scilab code Exa 4.2 Minimum uncertainty in wavelength

```
clear
clc
disp('Ex-4.2');
// w=wavelength; consider k=2*(pi/w);
// differentiate k w.r.t w and replace del(k)/del(w)
= 1 for equation.4.3
// which gives del(w)= w^2 /(2*pi*del(x)), hence
w=20; delx=200; // delx=200cm and w=20cm
delw=(w^2)/(delx*2*%pi);
printf('Hence uncertainity in length is %1.2 f cm', delw);
```

Scilab code Exa 4.3 Validity of the claim

```
1 clear
2 clc
3 \text{ disp}('Ex-4.3')
                         //consider time interval of 1
4 delt=1;
      sec
5 delw=1/delt;
                        // since delw*delt =1 from
      equation 4.4
6 \text{ delf} = 0.01
                       //calculated accuracy is 0.01Hz
  delwc =2*%pi*delf // delwc-claimed accuracy from w
     =2*pi*f
8 printf ('The minimum uncertainity calculated is 1rad/
      sec. The claimed accuracy is \%.3 f rad/sec\n',
     delwc);
9 if delw==delwc then disp('Valid claim');
11 if delw~=delwc then disp('Invalid claim');
```

Scilab code Exa 4.4 Solution for a and b

```
1 clear
2 clc
3 disp('Ex-4.4(a)');
4 m=9.11*10^-31; v=3.6*10^6; //'m','v' - mass an velocity of the electron in SI units
5 h=1.05*10^-34; //planck's constant in SI
```

```
6 p=m*v; //momentum
7 delp=p*0.01; //due to 1% precision in p
8 delx = h/delp//uncertainity in position
9 printf('Uncertainity in position is %1.2f nm',delx *10^9);
10 disp('Ex-4.4((b)')
11 printf('Since the motion is strictly along X-direction, its velocity in Y direction is absolutely zero.\n So uncertainity in velocity along y is zero=> uncertainity in position along y is infinite. \nSo nothing can be said about its position/motion along Y')
```

Scilab code Exa 4.5 Solution for a and b

```
1 clear
2 clc
3 disp('Ex-4.5(a)');
4 m=0.145; v=42.5; //'m','v' - mass an velocity of the electron in SI units
5 h=1.05*10^-34; //planck's constant in SI
6 p=m*v; //momentum
7 delp=p*0.01; //due to 1% precision in p
8 delx = h/delp//uncertainity in position
9 printf('Uncertainity in position is %1.2e',delx);
10 disp('Ex-4.5(b)');
11 printf('Motion along y is unpredictable as long as the veloity along y is exactly known(as zero).');
```

Scilab code Exa 4.6 Uncertainty in x component

```
1 clc
2 clear
3 disp('Ex-4.6')
4 printf ('The uncertainity in the poisition of
    electron after it passes through the slit is
    reduced to width of the slit n = a n';
5 printf ('The uncertainity in momentum = h/a n');
6 printf('Position of landing(angle t) = sin t = tan t
     wavelenghth\n');
7 printf ('Rewriting
                   the above expression a*sint = w
    /(2*pi)\n which is similar to a*sint = w (
    neglect 2*pi) as found out by first minimum in
    diffraction by a slit of width a');
8 disp('It proves a close connection between wave
    behaviour and uncertainity principle');
```

Scilab code Exa 4.7 Range of kinetic energy of an electron

```
1 clear
2 clc
3 disp('Ex-4.7');
4 \text{ mc2}=2.15*10^-4;
                               //mc2 is the mass of the
     electron, concidered in Mev for the simplicity in
      calculations
                              // The value of h*c in Mev.
5 hc = 197
     fm for simplicity
                            // Given uncertainity in
6 \text{ delx} = 10
     position=diameter of nucleus= 10 fm
                           //Uncertainiy in momentum per
7 delp= hc/delx ;
     unit 'c' i.e (\text{Mev/c}) delp= h/delx = (\text{h*c})/(\text{c*delx})
```

Scilab code Exa 4.8 Solution for a b and c

```
1 clear
2 clc
3 \text{ disp}('Ex-4.8')
4 h=6.58*10^-16; // plack's constant
5 delt1=26*10^-9; E=140*10^6 //given values of lifetime
       and rest energy of charged pi meson
6 delE=h/delt1; k=delE/E; // k is the measure of
      uncertainity
7 printf ('Uncertainity in energy of charged pi meson
      is \%1.2 \,\mathrm{e} \,\mathrm{n}',k);
8 delt2=8.3*10^-17; E=135*10^6; //given values of
      lifetime and rest energy of uncharged pi meson
9 delE=h/delt2; k=delE/E;
10 printf ('Uncertainity in energy of uncharged pi meson
       is \%1.2 e n', k);
11 delt3=4.4*10^-24;E=765*10^6;
                                        //given values of
       lifetime and rest energy of rho meson
12 delE=h/delt3; k=delE/E;
13 printf('Uncertainity in energy of rho meson is %.1 f\
     n',k);
```

Scilab code Exa 4.9 minimum velocity of the billiard ball

```
1 clear
2 clc
3 \text{ disp}('Ex-4.9')
4 h=1.05*10^-34;
                   //value of planck's constant in J.
     sec
  delx=1;
                    // uncertainity in positon=
     dimension of the ball
                   // uncertainity in momentum
6 delp=h/delx;
7 m = 0.1;
                    //mass of the ball in kg
8 delv=delp/m;
                   // uncertainity in velocity
9 printf ('The value of minimum velocity was found out
     to be \%1.2 \,\mathrm{e} m/sec', delv);
```

Scilab code Exa 4.10 Group velocity of a wave packet in terms of phase velocity

Chapter 5

The Schrodinger Equation

Scilab code Exa 5.1 Displacement and velocity of the object

Scilab code Exa 5.2 Solution for a and b

```
1 clear
```

```
2 clc
3 disp('Exa-5.2(a)');
4 h=1.05*10^{-34}; m=9.11*10^{-31}; L=10^{-10};
                                                   // all
      the values are taken in SI units
5 E1=h^2*%pi^2/(2*m*L^2); E2=4*E1;
                                                    //
      Energies are calculated
                                                    //
6 delE=(E2-E1)/(1.6*10^-19);
      Difference in energy is converted to eV
7 printf('Energy to be supplied is %.0 f eV.\n',delE);
8 disp('Exa-5.2(b)');
9 \times 1 = 0.09 \times 10^{-10}; \times 2 = 0.11 \times 10^{-10}
                                                      //
      limits of the given region
10 probGnd=(2/L)*integrate('(sin(\%pi*x/L)^2)', 'x', x1, x2)
      );
11 printf ('The percentage probablility of finding an
      electron in the ground state is \%.2 f.\n', probGnd
      *100);
12 disp('Exa-5.2(c)');
13 x1=0, x2=0.25*10^-10;
14 probExc=(2/L)*integrate('(sin(2*\%pi*x/L)^2)', 'x', x1,
      x2);
15 printf ('The probablility of finding an electron in
      the excited state is \%.2 f.\n', probExc);
```

Scilab code Exa 5.3 Proof for average value of x

 state . $^{\prime}$);

The Rutherford Bohr model of an atom

Scilab code Exa 6.1 Average deflection angle per collision

Scilab code Exa 6.2 Solution for a and b

```
1 clear
2 clc
3 disp('Exa-6.2(a)');
4 Na=6.023*10^23; p=19.3; M=197;
5 n=Na*p/M;
             //The number of nuclei per atom
6 t=2*10^-6; Z=79; K=8*10^6; x=1.44; theta=90; //x=e
      ^2/4*pi*epsi0
7 b1=t*Z*x*cotd(theta/2)/(2*K)
                                          //impact
     parameter b
8 f1=n*%pi*b1^2*t
                                          //scattering
      angle greater than 90
9 printf ('The fraction of alpha particles scattered at
       angles greater than 90 degrees is \%.2e\n',f1);
10 disp('Exa - 6.2(b)');
11 \text{ theta}=45
12 b2=t*Z*x*cotd(theta/2)/(2*K);
                                      //scattering angle
13 f2=n*%pi*b2^2*t;
       greater than 45
14 fb=f2-f1
                                   //scattering angle
     between 45 to 90
15 printf ('The fraction of particles with scattering
      angle from 45 to 90 is \%.3e\n', fb);
```

Scilab code Exa 6.3 Distance of closest approach

Scilab code Exa 6.4 Three longest wavelengths of the Paschen series

```
1 clear
2 clc
3 disp('Exa-6.4');
4 sl=820.1;n0=3; //given values
5 n=4;w=sl*(n^2/(n^2-n0^2)); printf('The 3 longest possible wavelengths are %.0 f nm,',w);
6 n=5;w=sl*(n^2/(n^2-n0^2)); printf('%.0 f nm,',w);
7 n=6;w=sl*(n^2/(n^2-n0^2)); printf('& %.0 f nm',w);
```

Scilab code Exa 6.5 Various wavelegths in Balmer and Lymann series

```
1 clear
2 clc
3 disp('Exa - 6.5');
                            //given variables and
4 sl=364.5; n=3;
      various constants are declared in the subsequent
      steps wherever necessary
5 \text{ w1=sl*(n^2/(n^2-4))};
                             //longest wavelength of
     balmer
6 c=3*10^8;
                             //corresponding freq.
7 f1=c/(w1*10^-9);
8 n0=1; n=2;
9 w2=91.13*(n^2/(n^2-n0^2)); //first longest of lymann
10 f2=c/(w2*10^-9);
                              //correspoding freq
11 \quad n0=1; n=3
```

Scilab code Exa 6.6 wavelengths of transition

Scilab code Exa 6.7 Two longest wavelengths of triply ionized beryllium

```
1 clear
2 clc
3 disp('Exa-6.7');
4 n1=3;n2=2;Z=4;hc=1240;
5 delE=(-13.6)*(Z^2)*((1/(n1^2))-((1/n2^2)));
6 w=(hc)/delE; //for transition 1
7 printf('The wavelngth of radiation for transition (2->3) is %f nm\n',w);
8 n1=4;n2=2; // n values for transition 2
9 delE=(-13.6)*(Z^2)*((1/n1^2)-(1/n2^2));
10 w=(hc)/delE;
11 printf('The wavelngth of radiation emitted for transition(2->4) is %f nm',w);
```

The Hydrogen atom in wave mechanics

Scilab code Exa 7.1 Proof

```
1 clear
2 clc
3 disp('Exa-7.1'); //The problem is entirely
          theoretcial.
4 printf('The solution obtained is r=4*ao i.e the most
          likely distance from origin for an electron in n
          =2,l=1 state.');
```

Scilab code Exa 7.2 Probability of finding an electron closer to nucles than Bohrs orbit

```
1 clear
2 clc
3 disp('Exa-7.2');
```

```
4 // calculating radial probability P= (4/ao^3)*
    inegral(r^2 * e^(-2r/ao)) between the limits 0
    and ao for r
5 Pr=integrate('((x^2)*%e^(-x))/2','x',0,2);//
    simplifying where as x=2*r/a0; hence the limits
    change between 0 to 2
6 printf('Hence the probability of finding the
    electron nearer to nucleus is %.3f',Pr);
```

Scilab code Exa 7.3 Probability of finding an electron inisde Bohr Radius

Scilab code Exa 7.4 Length of angular momentum vectors

```
1 clear
```

Scilab code Exa 7.5 possible Z components of the vector L

```
1 clear
2 clc
3 disp('Exa-7.5'); //Thoretical question
4 disp('The possible values for m are [+2,-2] and hence any of the 5 components [-2h,2h] are possible for the L vector.');
5 printf('Length of the vector as found out previously is %.2 f*h.', sqrt(6)); // angular momentum==sqrt(1(1+1)) h
```

Scilab code Exa 7.6 Separation of beams as they leave the magnet

```
1 clear
2 clc
3 disp('Exa-7.6');
```

Scilab code Exa 7.7 Change in wavelength

Many Electron Atoms

Scilab code Exa 8.1 Energy of Ka X ray of sodium

```
1 clear
2 clc
3 disp('Exa-8.1');
4 hc=1240*10^-9;Rinfi=1.097*10^7;Z=11;  //for sodium
    atom; and other constants in MeV
5 delE=3*hc*Rinfi*(Z-1)^2/4  //change in
    energy
6 printf('The energy of the Ka x-ray of the sodium
    atom is %.3 f KeV.',delE/10^3);
```

Scilab code Exa 8.2 Solution for a and b

```
1 clear  
2 clc  
3 disp('Exa-8.2(a)');  
4 EKa=21.990; EKb=25.145; EK=25.514  
// all the
```

Scilab code Exa 8.3 Total orbital and spin quantum numbers of carbon

Scilab code Exa 8.4 Total orbital and spin quantum numbers of nitrogen

```
1 clear
2 clc
3 disp('Exa-8.4');
4 l=1; Lmax=l+1; Lmin=l-1; printf('Considering any two electrons, Value of L2e ranges from %d to %d i.e %d %d %d.\n', Lmin, Lmax, Lmin, 1, Lmax);
5 printf('Adding the angular momentum of the third electron to L2emax gives the maximum whole
```

```
angular momentum as 2+1=3; and subtracting it
from L2e=1 gives 0\n')
6 s=1/2; Smax=s+s; Smin=s-s; printf('Values of S2e are
%d &%d.\n', Smax, Smin);
7 printf('Adding and subtracting the spin of third to
S2e=1 and S2e=0 respectively gives the spins 3/2
and 1/2 for the 3 electron system.');
```

Scilab code Exa $8.5\,$ Hunds rule to find ground state quantum numbers of nitrogen

```
1 clear
2 clc
3 disp('Exa-8.5');
4 disp('The nitrogen atom has a configuration of 1s2,2 s2,2p3.');
5 disp('Let us maximize the net spin of all the 3 electrons by assigning a spin of 1/2 to each of them. Hence S=3/2.');
6 disp('To maximize Ml, the consistent values of L for the 3 electrons left are 1-1 and 0.Thus L=0 & S=3/2 are the ground state quantum numbers for nitrogen.');
```

Scilab code Exa 8.6 Ground state L and S of oxygen

```
1 clear
2 clc
3 disp('Exa-8.6');
```

- 4 disp('The Oxygen atom has a configuration of 1s2,2s2,2p4. 4 electrons in the outer most shell.');
- 5 disp('Let us maximize the net spin by assigning a spin of 1/2 to 3 of them but the fourth should have spin of -1/3. Hence S=3/2-1/2=1.');
- 6 disp('The consistent values of L for the 3 electrons are 1-1 and 0. To maximize Ml, assign a L 0f +1 to the fourth electron. Thus L=1 & S=1 are the ground state quantum numbers for Oxygen.');

Molecular Structure

Scilab code Exa 9.1 Charge on the sphere

Scilab code Exa 9.2 Solution for a and b

```
1 clear
2 clc
```

```
3 disp('Exa-9.2(a)');
4 K=1.44; Req=0.236; // K=e<sup>2</sup>/(4*pi*e0)=1.44 eV.nm
                           //coulomb energy
5 \text{ Uc}=-K/(\text{Req});
6 printf ('The coulomb energy at an equilirium
      separation distance is \%.2\,\mathrm{f} eV\n',Uc);
7 E=-4.26; delE=1.53;
                           //various standars values of
      NaCl
8 Ur=E-Uc-delE;
9 printf ('The pauli''s repulsion energy is \%.2 \text{ f eV} \setminus \text{n'},
      Ur);
10 disp('Exa-9.2(b)');
11 Req=0.1;
                       //pauli repulsion energy
12 Uc=-K/(Req);
13 E=4; delE=1.53;
14 Ur=E-Uc-delE;
15 printf ('The pauli''s repulsion energy respectively
      is is \%.2 \text{ f eV} \text{ n',Ur};
```

Scilab code Exa 9.3 vibrational frequency and photon energy of H2

```
1 clear
2 clc
3 \text{ disp}('Exa-9.3');
4 delE=0.50; delR=0.017*10^-9;
                                          // delE = E-Emin;
      delR=R-Rmin;
5 \text{ k=2*(delE)/(delR^2); c=3*10^8;}
                                          //force constant
6 m = (1.008) * (931.5*10^6) *0.5;
                                          //mass of
      molecular hydrogen
                                // vibrational
7 \text{ v= } \frac{\text{sqrt}(k*c^2/m)}{(2*\%pi)};
      frequency
8 h=4.14*(10^-15);
9 E=h*v;
10 printf ('The value of corresponding photon energy is
```

Scilab code Exa 9.4 Energies and wavelengths of 3 lowest radiations emitted by molecular H2

```
1 clear
2 clc
3 disp('Exa-9.4');
4 \text{ hc} = 1240;
                    //in eV.nm
5 m=0.5*1.008*931.5*10^6;
                                           //mass of
      hydrogen atom
                                           //equivalent
6 Req=0.074;
      radius
  a=((hc)^2)/(4*(pi^2)*m*(Req^2)); //reduced mass of
       hydrogen atom
  for L=1:3,
            delE= L*a; printf('The value of energy is %f
               eV \setminus n', delE);
            w=(hc)/delE; printf('The respective
10
               wavelength is is \%f um n', w*10^-3);
11 end
```

Scilab code Exa 9.5 Rotational Inertia of molecule

```
1 clear
2 clc
3 disp('Exa-9.5');
4 delv=6.2*(10^11);    //change in frequency
5 h=1.05*(10^-34);    //value of h in J.sec
```

Scilab code Exa 9.6 Solution for a and b

```
1 clear
2 clc
3 disp('Ex-9.6(a)');
4 delE=0.358; hc=4.14*10^-15;
                                         //hc in eV.nm
     and delE=1.44eV(given values)
5 f = (delE)/hc;
                                         //frequency
6 printf ('The frequency of the radiation is \%.3e.\n',f
     );
7 m = 0.98;
                                        //mass in terms
      of u
                                        //value of k in
8 k=4*\%pi^2*m*f^2;
     eV/m^2
9 printf('The force constant is \%.3e.\n',k);
10 disp('Ex-9.6(b)');
11 hc=1240; m=0.98*1.008*931.5*10^6; Req=0.127;
                                                        //
      various constants in terms of
12 s=((hc)^2)/(4*(\%pi^2)*m*(Req^2));
                                                       //
      expeted spacing
13 printf ('The spacing was found out to be %f which is
      very close to the graphical value of 0.0026 eV.',
     s);
```

Statistical Physics

Scilab code Exa 10.1 Various Speeds obtained from maxwell speed distribution

```
clear
clc
disp('Exa-10.1'); //Theoretical Question
//**Install and use maxim tool for symbolic
    integration. remove the '//'(comment markings)
    below and run the program.
//Vm=integrate('(v^3)*(e^(-b*v^2))','x',0,%infi);
//rest of the results follow from above
printf('The average speed is found out to be (8*k*T/m)^1/2\n');
printf('The RMS speed is (3*k*T/m)^1/2\n');
printf('The Most probable speed is found out to be (2*k*T/m)^1/2 \n where all the symbols used are conventional constants.');
```

Scilab code Exa 10.2 Frequency distribution of emitted light

Scilab code Exa 10.3 Solution for a and b

```
1 clear
2 clc
3 disp('Exa-10.3(a)');
4 kT=0.0252; E=10.2
                                                      // at
      room temperature, kT=0.0252 standard value and
      given value of E
5 \quad n2=2; n1=1; \quad g2=2*(n2^2); g1=2*(n1^2);
      values for ground and excited states
6 t=(g2/g1)*%e^{-E/kT};
                                                      //
      fraction of atoms
7 printf ('The number of hydrogen atoms required is %e
      which weighs \%e \text{ Kg}\n', 1/t, (1/t)*(1.67*10^-27));
8 disp('Ex-10.3(b)');
9 t=0.1/0.9; k=8.65*10^-5
                                                        //
      fracion of atoms in case -2 is given
10 T=-E/(\log(t/(g2/g1))*k);
                                                        //
      temperature
```

11 printf ('The value of temperature at which 1/10 atoms are in excited state is %.3 f K',T);

Scilab code Exa 10.4 Solution for a and b

```
clear
clear
clear
clear
disp('Exa-10.4(a)'); //theoretical
printf('The energy of interaction with magnetic
    field is given by uB and the degeneracy of the
    states are +-1/2 which are identical.\nThe ratio
    is therefore pE2/pE1 which gives e^(-2*u*B/k*T)')
;
disp('Ex-10.4(b)');
uB=5.79*10^-4; //for a typical atom
t=1.1;k=8.65*10^-5; //ratio and constant k
T=2*uB/(log(t)*k); //temperature
printf('The value of temperature ar which the given
    ratio exists is %.2f K',T);
```

Scilab code Exa 10.5 Fermi Energy Ef for sodium

Properties of Ionic Crystals

Scilab code Exa 11.1 Solution for a and b

```
1 clear
2 clc
3 \text{ disp}('Exa-11.1(a)');
4 c=769*10^3; Na=6.023*10^23; JeV=1.6*10^-19; //
      various constants and given values
5 Be=c/(Na*JeV);
                           //Binding energy of an ion
      pair in the lattice
6 printf ('The experimental value was found out to be %
      .4 \, \mathrm{f \ eV. \setminus n'}, Be);
7 disp('Exa-11.1(b)');
8 n=9; a=1.7476; R=0.281; k= 1.44; //Given values
      and consstants
                                            //ionic binding
9 Bc=k*a*(1-(1/n))/R;
       energy eperimentally
10 printf ('The calculated value of the binding energy
      is \%.4 \, f \, eV., Bc);
```

Scilab code Exa 11.2 Energy per neytral atom to take apart a crystal of Nacl

```
clear
clc
disp('Exa-11.2');
a=3.61;// amount of energy required to remove an
        electron from Cl- ion
b=-5.14 //amount of energy returned when an electron
        is added to Na+ ion\
c=7.98 //binding energy of NaCl atom
E=a+b+c //suom of all the energies
printf('The net energy to be supplied is %.3f eV',E)
;
```

Scilab code Exa 11.3 Solution for a and b

```
1 clear
2 clc
3 disp('Ex-11.3(a)');
4 Na=6.023*10^23; p=8.96*10^3; M=63.5*10^-3;
                                                  //Na=
     avagadro's number, p=density, M=molar mass
5 n = p*Na/M;
     density of charge carriers
6 printf ('The density of charge carriers in copper is
     \%e atoms/m3\n',n);
7 s=5.88*10^7; m=9.11*10^-31; e=1.6*10^-19;
                                                //charge
     & mass of an electron, resistance per unit length
8 t = s*m/(n*e^2);
                                                //average
      time between collisions
9 printf('The average time between collisions of
     conducting electrons is %e sec.\n',t);
10 disp('Ex-11.3(b)');
```

Nuclear Structure and Radioactivity

Scilab code Exa 12.1 Solution for a and b

```
1 clear
2 clc
3 disp('Exa-12.1(a)');
4 Z=2; A=4; N=A-Z;
                            // Given values
5 printf ('The following method of representing atoms
      is followed throughout the chapter \n t \ x, ySz \ n
      where x=atomic number y=mass number z= Neutron
      Number S=symbol of the atom\n\')
6 printf ('The helium can be reperesented as %d, %dHe%d\
     n',Z,A,N);
7 disp('Exa-12.1(b)');
8 \quad Z=50; N=66; A=Z+N;
                           //// Given values and
      standard formulae
9 printf ('The helium can be reperesented as \%d,\%d\Sn\%d\
      n',Z,A,N);
10 disp('Exa-12.1(c)');
11 A = 235; N = 143; Z = A - N;
12 printf('The helium can be reperesented as %d,%dU%d',
      Z,A,N);
```

Scilab code Exa 12.2 Approximate nuclear radii

Scilab code Exa 12.3 Density of typical nucleus and resultant mass

Scilab code Exa 12.4 Total Binding Energy

```
1 clear
2 clc
3 \text{ disp}('Exa-12.4');
4 N=30; Z=26; A=56; Mn=1.008665; Mp=1.007825; m=55.934939;
     c2=931.5; //given values and constants for case-1
5 B=((N*Mn)+(Z*Mp)-(m))*c2;
                                           //binding energy
     (per nucleon)
6 printf ('Binding nergy per nucleon for 26,56\,\mathrm{Fe}30 is \%
     .3 f \text{ MeV} \setminus n', B/A);
7 N=146; Z=92; A=238; Mn=1.008665; Mp=1.007825; m
     =238.050785; c2=931.5;
                                   //given values and
     constants for case -2
8 B=((N*Mn)+(Z*Mp)-(m))*c2;
                                     //binding energy(per
     nucleon)
9 printf('Binding nergy per nucleon for 26,56Fe30 is %
      .3 f MeV', B/A);
```

Scilab code Exa 12.5 Solution for a b c and d

```
1 clear
2 clc
3 disp('Exa -12.5(a)');
4 t12=2.7*24*3600;
                                  //converting days into
       seconds
5 w=0.693/t12;
                                  //lambeda
6 printf('The decay constant is %e\n /sec',w);
7 disp('Exa -12.5(b)');
8 printf ('The decay constant is equal to probability
      of decay in one second hence \%e \n', w);
9 disp('Exa -12.5(c)');
10 m=10^-6; Na=6.023*10^23; M=198;
                                       //given values
     and constants
                                     //number of atoms
11 N=m*Na/M;
     in the sample
                        //activity
12 Ao=w*N;
13 printf('The activity was found out to be %e Ci', Ao);
14 disp('Exa -12.5(d)');
                         //given time
15 t=7*24*3600;
16 A = Ao * %e^-(w*t);
                        //activity
17 printf ('The activity after one week was found out to
      be \%.2e decays/sec', A);
```

Scilab code Exa 12.6 Atoms at the time of solidification

```
atoms of 235U where Na is the Avagadro''s Number =6.023*10^23',r);
```

Scilab code Exa 12.7 Kinetic energy of alpha particle emitted in alpha decay

Scilab code Exa 12.8 Q value of 14C emission

```
1 clear
2 clc
3 disp('Exa-12.8');
4 m226Ra=226.025403; //mass of various elements
5 m212Pb=211.991871;
6 m14c=14.003242;
7 c2=931.5; //value of c^2
```

```
8 Q=(m226Ra-m212Pb-m14c)*c2;  //Q of the reaction
9 printf('The value of Q for 14c emission is %.3f MeV\
    n',Q);
10 printf('The probability of 14c emission is 10^-9
    times that of an alpha particle since the energy
    barrier for 14c emission is \n nearly 3 times
    higher and thicker.')
```

Scilab code Exa 12.9 Maximum Kinetic energy of emitted electron

```
1 clear
2 clc
3 disp('Ex-12.9');
4 m23Ne=22.994465; //mass of various elements
5 m23Na=22.989768;
6 c2=931.5; //value of c^2
7 Q=(m23Ne-m23Na)*c2; //Q of the reaction
8 printf('Hence the maximum kinetic energy of the emitted electrons is %.3 f MeV',Q);
```

Scilab code Exa 12.10 Q value of various decays

```
8 printf('The Q value for -VE beta emission is %.3f
      Mev \n', Qb1);
9 disp('Exa-12.10(b)');
10 \text{ m}40\text{K}=39.963999;
                                //mass of various particles
11 m40Ar=39.962384;
12 \text{ me} = 0.000549;
13 Qb2=(m40K-m40Ar-2*me)*c2;
                                           //Q value of the
      reaction
14 printf ('The Q value for +VE beta emission is %.3 f
      Mev \n', Qb2);
15 disp('Exa-12.10(c)');
16 \text{ m40K} = 39.963999;
17 m40Ar=39.962384;
18 Qec = (m40K - m40Ar) * c2;
19 printf ('The Q value for +VE beta emission is %.3 f
      Mev \setminus n', Qec);
```

Scilab code Exa 12.11 Maximum kinetic energy of emitted beta particle

```
1 clear
2 clc
3 disp('Exa-12.11');
4 Mg=12.000000; //mass of the carbon atom in amu
5 c2=931.5;
6 Eg=4.43; //given energy of gamma ray
7 Mex=Mg+(Eg/c2); //mass in excited state
8 Me=0.000549; //mass of an electron
9 Q=(12.018613-Mex-2*Me)*c2; //Q of the particle
10 printf('The maximum value of kinetic energy is %.2f MeV',Q);
```

Scilab code Exa 12.12 Rate of energy production per gram of uranium

```
1 clear
2 clc
3 disp('Exa-12.12');
4 m238U=238.050786; //mass of various quantities
5 \text{ m206Pb} = 205.974455;
6 \text{ m4He} = 4.002603;
7 c2=931.5;
              //constants
8 Na=6.023*10^23; //avagadro's number
9 Q = (m238U - m206Pb - 8*m4He)*c2;
                                  //half life years to
10 t12=(4.5)*10^9*(3.16*10^7);
     seconds conversion
11 w=0.693/t12;
                                 // lambeda
12 NoD=(Na/238)*w; //number of decays
13 E=NoD*Q*(1.6*10^-19)*10^6;
                                       //rate of
     liberation of energy, converting MeV to eV
14 printf('Rate of energy liberation is %.1e W', E);
```

Scilab code Exa 12.13 Ages of the given rocks

Scilab code Exa 12.14 Solution for a and b

```
1 clear
2 clc
3 disp('Ex-12.14(a)');
4 P=2*10^14; V=2*10^-14; R=8.314; T=295; Na
      =6.023*10^23; //varoius constants and given
      values
5 n=P*V/(R*T);
                      //ideal gas law
6 \text{ N=Na*n; f=10}^-12
                            //avagadaro's number and
      fracction of carbon molecules
7 t12=5730*3.16*(10^7);
                           //half life
8 A = (0.693/t12)*N*f;
                          //activity
9 D1w=A*7*24*60*60;
                         //decays per second
   printf('The no of decays pers second is %4.0 f \n',
10
       D1w);
11 disp('Ex-12.14(b)');
12 c1 = 1420;
                         //concentration at instant 1
13 c2 = D1w;
                         //concentration at instant 2
14 \text{ t}12y = 5730;
                        //half life
15 t=t12y*log(c2/c1)/0.693;
                                         //age of the
     sample
16 printf('Age of the sample is %.2f years',t);
```

Nuclear Reaction and Applications

Scilab code Exa 13.1 Rate of production of neutron

```
1 clear
2 clc
3 \text{ disp}('Ex-13.1')
4 v=1*1*10^-6*10^2; p=7.9; m=p*v; Na=6.023*10^23
      given values and various constants in suitable
      units
5 M=56; N=m*Na/M;
                                                      //
     number of atoms
6 i=3*10^-6;
7 q=1.6*10^-19;
8 \text{ Io=i/q};
                             //intensity
9 s=0.6*10^-24; S=1;
                                   //given values in
      suitable units
10 R=N*s*Io/S;
                                     //rate of neutrons
11 printf ('The rate of neutrons emitted from the target
       is %.2e particles per second', R);
```

Scilab code Exa 13.2 Resultant activity of 198Au

Scilab code Exa 13.3 Solution for a and b

```
1 clear
2 clc
3 disp('Exa-13.3(a)')
4 v=1.5*1.5*2.5*(10^-6)*10^2; //volume in cm3
5 p=8.9; //density in g/cm3
6 m=p*v;Na=6.023*10^23 //mass and Avagadro's number
7 M=58.9; //Given values
8 N=m*Na/M;
9 i=12*10^-6; //thickness of beam
10 q=1.6*10^-19;
```

Scilab code Exa 13.4 Solution for a and b

```
1 clear
2 clc
3 disp('Exa-13.4(a)');
4 m2H=2.014102; //mass of various particles
5 \text{ mn} = 1.008665; \text{m} 63\text{Cu} = 62.929599;
6 m64Zn=63.929145; c2=931.5;
                                     //c^2 = 931.5 \text{ MeV}
7 Q = (m2H + m63Cu - mn - m64Zn) * c2;
                                       //Q of the reaction
8 printf('The value of Q is %f MeV\n',Q);
9 disp('Exa-13.4(b)');
10 Kx = 12.00; Ky = 16.85;
11 Ky = Q + Kx - Ky
                              //kinetic energy of 64Zn
12 printf ('The value of Ky was found out to be %.2f MeV
      ', Ky);
```

Scilab code Exa 13.5 Solution for a and b

```
1 clear
2 clc
3 disp('Exa-13.5(a)');
4 mp=1.007825; m3H=3.016049; //mass of the particle 5 m2H=2.014102; c2=931.5; //constant
6 Q=(mp+m3H-(2*m2H))*c2; //Q of thereaction
7 printf('The value of q was found out to be %f MeV\n'
      ,Q);
8 disp('Exa-13.5(b)');
9 Kth1= -Q*(1+(mp/m3H)); //threshold energy of
      kinetic energy
10 printf ('The threshold kinetic energy in case-1 is %f
      MeV \setminus n', Kth1);
11 Kth2=-Q*(1+(m3H/mp));
                           //threshold kinetic
      energy in case2
12 printf('The threshold kinetic energy in case -2 is \%f
      MeV', Kth2);
```

The Four Basic Forces

Scilab code Exa 14.1 Solution for a and b

```
1 clear
2 clc
3 disp('Ex-14.1'); //theoretical question
4 printf('14.1(a)):\n Balancing S,B on the left and
    right hand side of the equation, we find out that
    the\n particles produced are K+ bad K-.\n\n');
5 printf('14.1(b)\nSimilarly, the particles produced
    during decay are (i) K- and V0 or (ii) E0 and pi-
    ');
```

Scilab code Exa 14.2 Energy of the proton and pi meson

Scilab code Exa 14.3 Maximum kinetic energy of the electron emitted in the decay

Scilab code Exa 14.4 maximum energy of the positron nad pi mesons

Scilab code Exa 14.5 Q values for reaction

Scilab code Exa 14.6 Threshold Kinetic energy to produce pi mesons

Scilab code Exa 14.7 Threshold Energy of the given reaction

Scilab code Exa 14.8 Solution for a and b

```
6 printf('U——>d+W(+). Hence the remaining processes are d+d(+)——>energy and \n W(+)——>u(+) and vu.');
```

Astrophysics and Genereal Relativity

Scilab code Exa 15.1 Change in wavelength in solar spectrum due to gravitaional shift

```
1 clear
2 clc
3 \text{ disp}('Ex-15.1');
4 \text{ w=} 121.5; // \text{lambeda}
5 G=6.67*10^-11; //Various given values and constants
6 M = 1.99 * 10^30;
7 R = 6.96 * 10^8;
8 c=3*10^8;
9 k= G*M/(R*c^2); //(delLambeda)/(lambeda)
10 delw=k*w;
                         //del(lambeda)
11 printf('The change in wavelength due to
      gravitational shift is \%.3 \text{ f pm/n'}, delw*10^3);
12 k=5.5*10^-5; //due to thermal Doppler broadening
      effect
13 delw=k*w;
14 printf ('The change in wavelength due to thermal
      Doppler broadening effect is %.1f pm', delw*10^3);
```

Scilab code Exa 15.2 Maximum energy of neutrino in the first reaction of proton proton cycle

```
1 clear
2 clc
3 \text{ disp}('Ex-15.2');
4 mp=938.280; //mass of various particles
5 \text{ me} = 0.511;
6 m2h=1875.628;
                         //mass energy on L.H.S
7 mic2=2*mp;
8 mfc2=m2h+me; //mass energy on R.H.S
9 Q=mic2-mfc2; //Q value of reation
10 pc=Q;
11 \text{ mc2} = 1875.628;
                        //kinetic threshold energy
12 K=(pc^2)/(2*mc^2);
                     //maximum energy
13 Emax = Q - K;
14 printf('The maximum neutrino energy is %.3f MeV',
      Emax);
```

The Cosmic Microwave Background Radiation

Scilab code Exa 16.1 Resultin temperature of interstellar space

```
1 clear
2 clc
3 \text{ disp}('Ex-16.1');
4 N2=0.25; N1=0.75;
      various given values
5 L2=1; L1=0;
6 E1_E2 = -4.7*(10^-4);
                                                      //
      Energy difference
7 a=(N2/N1); b=(((2*L2)+1)/((2*L1)+1)); c=E1_E2; //
      various terms involved in the formula of ratio of
       population
8 kT=(c/log(a/b));
      value of k*T
9 k=0.0000856;
      constant
10 T=kT/k;
                            //temperature of
      interstellar space
11 printf ('The temperature of interstellar space was
     found out to be %.1 f K',T);
```

Scilab code Exa 16.2 Solution for a and b

```
clear
clc
disp('Ex-16.2');
mc2=940*10^6; k=8.6*10^-5; //various constants and
    given values in suitable units

T= mc2/k; //temperature of the photons
printf('The temperature of the photons must be %.1e
    K\n',T);

t=((1.5*10^10)/T)^2; //age of universe when
    the photons have the above temperature
printf('The age of the universe for the temperature
    of the photon to be as obtained above is %.0e
    seconds',t);
```

Scilab code Exa 16.3 Relative number of neutrons and protons among the nucleus

```
1 clear
2 clc
3 disp('Ex-16.3');
4 k=8.62*10^-5; //various values and constants
5 T= 1.5*10^10;
6 delE=1.3*10^6;
7 a= delE/(k*T); //exponent in boltzmann factor
8 b=%e^-a; //ratio of neutron to protons
```

```
9 r=(1/(1+b))*100; //relative number of protons
10 printf('The percentage of protons is %.0f and
        neutrons is %.0f.',r,100-r);
```