Scilab Textbook Companion for Materials Science And Engineering: An Introduction by W. D. Callister¹

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Book Description

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Scilab numbering policy used in this document and the relation to the above book.

Exa Example (Solved example)

Eqn Equation (Particular equation of the above book)

AP Appendix to Example(Scilab Code that is an Appednix to a particular Example of the above book)

For example, Exa 3.51 means solved example 3.51 of this book. Sec 2.3 means a scilab code whose theory is explained in Section 2.3 of the book.

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Chapter 3

The Structure of Crystalline Solids

Scilab code Exa 3.1 Determination of FCC Unit Cell Volume

```
1 // Determination of FCC Unit Cell Volume
3 clear all;
4 clc;
6 printf("\tExample 3.1\n");
8 //For FCC a=2*R*sqrt(2)
9 R=poly([0], 'R');
10
11 //Edge Length
12 \ a=2*R*sqrt(2);
13
14 //Volume determination
15 V=a^3;
16
17 disp(V, "Volume is");
18
19 / End
```

Scilab code Exa 3.2 Computation of the Atomic Packing Factor for FCC

```
//Computation of the Atomic Packing Factor for FCC
2
3 clear all;
4 clc;
  printf("\tExample 3.2\n");
  //for FCC no. of atoms are 4
9 n = 4;
10
  //For FCC a=2*R*sqrt(2)
12 R=poly([0], 'R');
13
14 //Edge Length
15 \ a=2*R*sqrt(2);
16
  //Volume determination of cube
17
18
  Vc=a^3;
19
  //Volume of sphere
21
  Vs=n*4*\%pi*R^3/3;
22
23
  //Atomic packing Fraction
24 APF=Vs/Vc;
25
  disp(APF, "Atomic packing fraction is");
26
27
  //End
28
```

Scilab code Exa 3.3 Theoretical Density Computation for Copper

```
1 // Theoretical Density Computation for Copper
3 clear all;
4 clc;
6 printf("\tExample 3.3\n");
                         //Atomic radius in cm
8 R=1.28D-08;
9 \quad A_Cu = 63.5;
                         //Atomic wt of copper
                         //For FCC
10 n=4;
11
                         //Avogadro no.
12 \text{ Na=6.023D23};
13
14 \ a=2*R*sqrt(2);
15 Vc=a^3;
16
17 den=n*A_Cu/(Vc*Na);
19 printf("\nDensity is \%.2 \, f g/cm<sup>3</sup>\n",den);
20
21 / End
```

Scilab code Exa 3.5 Specification of Point Coordinates

```
//Specification of Point Coordinates

clear all;
clc;

printf("\tExample 3.5\n");

disp("Point coordinates for given positions of BCC cell are");

A=['Point_no', 'x_axis', 'y_axis', 'z_axis', '
```

```
Coordinates';
      1','0','0','0','0','000';
11 '
      2','1','0','0','100';
3','1','1','0','110';
12 ,
13 '
      4','0','1','0','010';
14 '
15 ' 5', '1/2', '1/2', '1/2', '1/2 1/2 1/2';
      6','0','0','1','001';
16 ,
17 ' 7', '1', '0', '1', '101';
      8,,,1,,,1,,,1,,,111;
18 '
19 ' 9', '0', '1', '1', '011'];
20
21 disp(A);
22
23 //End
```

Scilab code Exa 3.6 Determination of Directional Indices

```
1 // Determination of Directional Indices
3 clear all;
4 clc;
6 printf("\tExample 3.6\n");
  printf("\nThe procedure is summarised as :\n");
9
        ', 'x', 'y', 'z';
10 A = [ '
  'Projections', 'a/2', 'b', '0c';
   'In terms of a,b,c', '1/2', '1', '0';
   'Reduction','1','2','0'];
13
14
15 disp(A);
16
17 printf("\nEnclosure
                         [1 \ 2 \ 0] \setminus n");
18
```

Scilab code Exa 3.8 Determination of Directional Indices for a Hexagonal Unit Cell

```
1 // Determination of Directional Indices for a
      Hexagonal Unit Cell
3 clear all;
4 clc;
6 printf("\tExample 3.8\n");
8 //From the construction shown in the book
9 \, du = 1;
10 dv = 1;
11 dw = 1;
12
13 //The above indices are for parallelopiped
14 //To convert it for hexagonal system
15 u=(2*du-dv)/3;
16 v = (2*dv - du)/3;
17 t=-(u+v);
18 \text{ w=dw};
19
20 x = [u v t w] *3;
21 disp(x, "The indices for the given directions are");
22
23 //End
```

Scilab code Exa 3.9 Determination of Planar Indices

```
1 // Determination of Planar (Miller) Indices
```

```
2
3 clear all;
4 clc;
5
6 printf("\tExample 3.9\n");
7
8 x=[0 -1 2];
9
10 disp(x,"The intercept for the given plane is");
11
12 //End
```

Scilab code Exa 3.11 Determination of Miller Bravais Indices

```
1 // Determination of Miller Bravais Indices for a
      Plane Within a Hexagonal Unit Cell
3 clear all;
4 clc;
6 printf("\tExample 3.11\n");
8 //This plane intersects the al axis at a distance a
     from the origin of the a1-a2-a3-z coordinate axes
      system.
10 //Furthermore, its intersections with the a2 and z
      axes are -a and c.
11
12 //Therefore, in terms of the lattice parameters,
      these intersections are 1, -1 and 1.
13
14 h=1;
15 k = -1;
16 1=1;
```

```
17 i=-(h+k);
18
19 x=[h k i 1];
20 disp(x,"The indices of plane are");
21
22 //End
```

Scilab code Exa 3.12.a Interplanar Spacing

Scilab code Exa 3.12.b Diffraction Angle Computations

```
1 // Diffraction Angle Computations
2
3 clear all;
4 clc;
```

```
5
6 printf("\tExample 3.12\n");
                       //Lattice parameter in nm
8 a=0.2866;
9 h=2;
10 k=2;
11 1=0;
12
13 d_hkl=a/(sqrt(h^2+k^2+l^2));
15 printf("\n\t Part B");
16 lambda=0.1790; //Wavelength in nm
17 n=1;
18
19 theta=asind(n*lambda/(2*d_hkl));
20 printf("\nDiffraction angle is %.2f degree\n",2*
     theta);
21
22 //End
```

Chapter 4

Imperfections in Solids

Scilab code Exa 4.1 Number of Vacancies Computation

```
1 // Number of Vacancies Computation at a Specified
     temperature
3 clear all;
4 clc;
6 printf("\tExample 4.1\n");
8 \text{ Na=}6.023*10^23;
                      //Avogadro No.
9 den=8.4D+06;
                      //Density of Copper
                      //Atomic weight of Copper
10 A = 63.5;
11
12 //No. of atomic site per cubic meter
13 N=Na*den/A;
14
15 //No. of vacancies at 1000 C
                     //Activation energy in eV
16 \quad Qv = 0.9;
19
20 Nv = N * exp(-Qv/(k*T));
```

```
21 printf("\nNo.of vacancies are %.1f * 10^25 /m^3", Nv /10^25);
22
23 //End
```

Scilab code Exa 4.3 Composition Conversion From Weight Percent to Atom Percent

```
1 //Composition Conversion - From weight percent to
      Atom percent
3 clear all;
4 clc;
6 printf("\t Example 4.3\n");
8 //Conversion to Atom percent
9 function[C]=conc(C1,C2,A1,A2)
       C=C1*A2*100/((C1*A2)+(C2*A1));
10
       funcprot(0)
11
12 endfunction
13
14
                   //Aluminium wt%
15 C_A1 = 97;
                   //Copper wt%
16 C_Cu=3;
                   //Atomic wt of Aluminium
17 A_A1 = 26.98;
                   //Atomic wt of Copper
18 A_Cu = 63.55;
19
20 CAl=conc(C_Al,C_Cu,A_Al,A_Cu);
21 CCu=conc(C_Cu,C_Al,A_Cu,A_Al);
22
23 printf("\nAtomic \%\% of Al is \%.1 f \%\%", CAl);
24 printf("\nAtomic \%% of Cu is \%.1 f \%\\n", CCu);
25
26 / End
```

Scilab code Exa 4.4.a Computations of ASTM Grain Size Number

```
//Computations of ASTM Grain Size Number

clear all;
clc;

printf("\tExample 4.4\n");

printf("\n\tPart A");

N=45;  //No. of grains per square inch

//Dterminin grain size no. N=2^(n-1)
n=(log(N)/log(2))+1;
printf("\nGrain size no. is %.1f\n",n);

//End
```

Scilab code Exa 4.4.b Number of Grains Per Unit Area

```
11  M=85;
12
13  Nm=(100/M)^2*2^(n-1);
14  printf("\nAt magnification of 85x\n");
15  printf("No. of grains per inch square are %.1 f\n",Nm
         );
16
17  //End
```

Chapter 5

Diffusion

Scilab code Exa 5.1 Diffusion Flux Computation

```
1 // Diffusion Flux Computation
3 clear all;
4 clc;
6 printf("\tExample 5.1\n");
                  //Concentration at A in
8 \text{ Ca} = 1.2;
                                           kg/m^3
                                           kg/m^3
                  //Concentration at B in
9 Cb = 0.8;
10
                 // Position 1 in m
11 xa=5*10^-3;
                 //Position 2 in m
12 xb=10*10^-3;
13
14 D=3*10^-11; // Diffusion coefficient in m^2/s
15
16 J=-D*(Ca-Cb)/(xa-xb);
17 printf("\nDiffusion flux is \%.1 \, f * 10^-9 \, kg/m^2-s",J
      /10^-9);
18
19 //End
```

Scilab code Exa 5.2 Nonsteady State Diffusion Time Computation

```
//Nonsteady-State Diffusion Time Computation I
3 clear all;
4 clc;
6 printf("\tExample 5.2\n");
8 \text{ Co=0.25};
                     //Initial Conc. in wt%
                     //Surface conc. in wt%
9 \text{ Cs} = 1.2;
10 Cx = 0.8;
                     //Conc. at any x in wt%
11
                   //Position in m
12 x=5*10^-4;
13 D=1.6*10^-11;
                   //Diffusion coeff in m^2/s
14
15 C=1-((Cx-Co)/(Cs-Co));
16 z=erfinv(C);
17
18 //But C=erf(x/2 \operatorname{sqrt}(Dt))
19 t=x^2/(4*D*z^2);
20
  printf("\nTime required is %d s or %.1f h\n",t,t
21
      /3600);
22
23 //End
```

Scilab code Exa 5.3 Nonsteady State Diffusion Time Computation II

```
1 //Nonsteady-State Diffusion Time Computation II
2
3 clear all;
```

Scilab code Exa 5.4 Diffusion Coefficient Determination

Scilab code Exa 5.5 Diffusion Coefficient Activation Energy

```
1 // Diffusion Coefficient Activation Energy and
      Preexponential Calculations
3 clear all;
4 clc;
6 printf("\tExample 5.5\n");
8 //From graph log D ad 1/T are deducted
9 inv_T1=0.8*10^-3;
                                 //Reciprocal of temp.
       K^-1
10 inv_T2=1.1*10^-3;
                                 //Reciprocal of temp.
      K^{\hat{}}-1
11 \log D1 = -12.4;
12 \log D2 = -15.45;
13
                                //Gas law Constant in J/
14 R=8.31;
      mol-K
15
16 Qd=-2.3*R*(logD1-logD2)/(inv_T1-inv_T2);
17 printf("\nActivation energy is %d kJ/mol",Qd/1000);
18
19 //For calculating Peexponential factor
20 D0=10^(logD2+(Qd*inv_T2/(2.3*R)));
21 printf("\nPreexponential factor D0 is \%.1 \, \text{f} * 10^-5 \, \text{m}
      ^{2}/s n, D0/10^{-5});
22
23 / End
```

Scilab code Exa 5.6 Diffusion Temperature Time Heat Treatment Specification

```
1 // Diffusion Temperature Time Heat Treatment
      Specification
2
3 clear all;
4 clc;
6 printf("\tDesign Example 5.1\n");
                       //Initial concentration in wt%
8 \quad C0 = 0.2;
                       //Surface conc in wt%
9 Cs=1;
                       //Conc at any position X in wt%
10 Cx = 0.6;
11 x=7.5*10^-4;
                       // Position in m
                       //Preexponential factor in m<sup>2</sup>/s
12 D0=2.3*10^-5;
                       //Gas law constant in J/mol-K
13 R=8.31;
14 \quad Qd = 148000;
                       //Activation energy in J/mol
15
16 C=1-((Cx-C0)/(Cs-C0));
17 z=erfinv(C);
18 Dt=(x/(2*z))^2;
19
20 / Dt = D0 * exp(-Qd/RT) * t = value of variable Dt
21 D=Dt/D0;
22
23 T = [900 950 1000 1050];
24 \text{ for } i=1:4
       t(i)=D/\exp(-Qd/(R*(T(i)+273)))/3600;
25
26 \, \text{end}
27
28 printf("\nTemperature(in Celsius) is \n");
29 disp(T);
30 printf("\nTime is (in hours)\n");
```

```
31 disp(t);
32
33 //End
```

Chapter 6

Mechanical Properties of Metals

Scilab code Exa 6.1 Elongation Computation

```
//Elongation (Elastic) Computation

clear all;
clc;

printf("\tExample 6.1\n");

E=110*10^3; //Young's modulus of Copper in MPa
sigma=276; //Applied stress in MPa
lo=305; //Original length in mm

//Deformation
dl=sigma*lo/E;

printf("\nElongation obtained is %.2 f mm \n",dl);
//End
```

Scilab code Exa 6.2 Computation of Load to Produce Specified Diameter Change

```
1 //Computation of Load to Produce Specified Diameter
     Change
3 clear all;
4 clc;
6 printf("\tExample 6.2\n");
  del_d=-2.5*10^-3;
                      //Deformation in dia
                                              in mm
                       //Initial dia
                                      in mm
9
  d0 = 10;
10
                      //Poisson ratio for brass
11 v = 0.34;
12
13 ex=del_d/d0;
14 printf("\nStrain in x-direction is \%f", ex);
15
16 ez=-ex/v;
17 printf("\nStrain in z-direction is \%f", ez);
18
                      //Modulus of elasticity in MPa
19 E=97*10^3;
20 sigma=ez*E;
21 F=sigma*%pi*(d0^2)/4;
22
23 printf("\nApplied force is %d N",F);
24
25 //End
```

Scilab code Exa 6.3.a Modulus of elasticity

```
1 clear all;
2 clc;
4 printf("\tExample 6.3\n");
6 //From draph in the question
7 //stress and strain can be obtained
                       // in MPa
9 \text{ si2}=150;
10 \text{ si1=0};
11 e2=0.0016;
12 e1=0;
13 d0=12.8*10^-3; //Initial Diameter in m
14
15 printf("\n\tPart A");
16 //Young's Modulus = stress/strain
17 E=(si2-si1)/(e2-e1);
18 printf("\nModulus of elasticity is %.2f GPa", E/10^3)
19
20 //End
```

Scilab code Exa 6.3.c Maximum load

```
1 clear all;
2 clc;
3
4 printf("\tExample 6.3\n");
5
6 //From draph in the question
7 //stress and strain can be obtained
8
9 si2=150; // in MPa
10 si1=0;
11 e2=0.0016;
```

Scilab code Exa 6.3.d Change in length

```
1 clear all;
2 clc;
3
4 printf("\tExample 6.3\n");
6 //From draph in the question
7 //stress and strain can be obtained
9 printf("\n\tPart D");
10
11 //From stress-strain curve
12 //Strain corresponding to stress of 345 MPa is 0.06
13
14 10=250;
               //Initial lengt in mm
15 e=0.06;
               //strain
16
17 dl=e*10;
18 printf("\nChange in length is %d mm",dl);
19
20
21 / End
```

Scilab code Exa 6.4.a Ductility Computations

```
1 // Ductility
3 clear all;
4 clc;
6 printf("\tExample 6.4\n");
                 //Initial dia in mm
8 di=12.8;
  df = 10.7;
                //Final dia in mm
10
11 printf("\n\tPart A");
12
13 // Ductility in terms of Reduction Area
14 \text{ RA} = ((di^2-df^2)/di^2)*100;
15 printf("\npercent reduction in area is %d \% \n", RA);
16
17 / End
```

Scilab code Exa 6.4.b True Stress At Fracture Computations

```
//True-Stress-At-Fracture Computations
clear all;
clc;
printf("\tExample 6.4\n");
di=12.8; //Initial dia in mm
df=10.7; //Final dia in mm
```

Scilab code Exa 6.5 Calculation of Strain Hardening Exponent

Scilab code Exa 6.6.a Average Computations

```
//Average Computations
3 clear all;
4 clc;
6 printf("\tExample 6.6a\n");
  //First and Last point are arbitary to plot the
      required 4 points
9 n = [0 1 2 3 4 5];
10 TS=[510 520 512 515 522 525];
11
12 plot(n,TS, '+');
13 xtitle ('Tensile strength data', 'Sample no.', 'Tensile
       strength');
14
15 //Mean Tensile strength
16 i=2;
17 TSmean=0;
18
19 for i=2:5
20
       TSmean = TSmean + (TS(i)/4);
21 end
  printf("\nMean tensile strength is %d MPa\n", TSmean)
23
24 / End
```

Scilab code Exa 6.6.b Standard Deviation Computations

```
1 //Standard Deviation Computations
3 clear all;
4 clc;
6 printf("\tExample 6.6b\n");
  //First and Last point are arbitary to plot the
      required 4 points
9 n = [0 1 2 3 4 5];
10 TS=[510 520 512 515 522 525];
11
12 i=2;
13 TSmean=0;
14
15 \text{ for } i=2:5
        TSmean = TSmean + (TS(i)/4);
16
17 \text{ end}
18
19 //Standard Deviation
20 \quad j = 0;
21 \text{ std=0};
22
23 \text{ for } i=2:5
        std=std+((TS(i)-TSmean)^2/(4-1));
24
25 end
26
27 printf("\nStandard deviation is %.1f MPa\n", sqrt(std
      ));
28
29 //End
```

Scilab code Exa 6.7 Specification of Support Post Diameter

```
1 // Specification of Support Post Diameter
```

```
3 clear all;
4 clc;
5
6 printf("\tDesign Example 6.1\n");
                   //Minimum yield strength in MPa
8 sig_y=310;
                  // Conservative factor of safety
9 N = 5;
10
                  //Two rods must support half of the
11 F=220000/2;
      total force
12
13 sig_w=sig_y/N;
14 d=2*sqrt(F/(%pi*sig_w));
15
16 printf("\nDiameter of each of the two rods is %.1 f
     mm \ n",d);
17
18 //End
```

Chapter 7

Dislocations and Strengthening Mechanisms

Scilab code Exa 7.1.a Resolved Shear Stress

```
1 //Resolved Shear Stress Computations
3 clear all;
4 clc;
6 printf("\tExample 7.1\n");
8 x=[1 1 0]; //Indices of Plane
9 y = [0 1 0];
                 //direction of applied tensile stress
10 z = [-1 \ 1 \ 1];
                 //Direction of shear stress
11
12 function[angle] = dotproduct(a,b)
       num = (a(1)*b(1))+(a(2)*b(2))+(a(3)*b(3));
13
14
       den=sqrt((a(1)^2+a(2)^2+a(3)^2)*(b(1)^2+b(2)^2+b
          (3)^2);
15
       angle=acos(num/den);
       funcprot(0)
16
17 endfunction
```

```
19 phi=dotproduct(x,y);
20 lambda=dotproduct(y,z);
21
22 printf("\nAngles phi is %.1f degree and lambda is %
      .1 f degree n, phi*180/%pi, lambda*180/%pi);
23
24
  //When a tensile stress of 52 MPa (7500 psi) is
25
      applied
26 printf("\n\t Part A\n");
27 sigma=52;
             //in MPa
28 tr=sigma*cos(phi)*cos(lambda);
29 printf("Resolved shear stress is %.1f MPa\n", tr);
30
31 / End
```

Scilab code Exa 7.1.b Stress to Initiate Yielding Computations

```
1 // Stress-to-Initiate-Yielding Computations
2
3 clear all;
4 clc;
6 printf("\tExample 7.1\n");
                 //Indices of Plane
8 x = [1 1 0];
                 //direction of applied tensile stress
9 y = [0 1 0];
10 z = [-1 \ 1 \ 1];
                 //Direction of shear stress
11
12 function[angle] = dotproduct(a,b)
       num = (a(1)*b(1))+(a(2)*b(2))+(a(3)*b(3));
13
       den=sqrt((a(1)^2+a(2)^2+a(3)^2)*(b(1)^2+b(2)^2+b
14
          (3)^2);
       angle=acos(num/den);
15
16
       funcprot(0)
```

```
17 endfunction
18
19 phi=dotproduct(x,y);
20 lambda=dotproduct(y,z);
21
22 printf("\nAngles phi is %.1f degree and lambda is %
      .1 f degree \n", phi*180/%pi, lambda*180/%pi);
23
24 //If slip occurs on a (110) plane and in a \begin{bmatrix} -1 & 1 & 1 \end{bmatrix}
      direction, and the critical resolved shear stress
       is 30 MPa
25 printf("\n\tPart B")
26
27 trc=30; //in MPa Critical resolved shear stress
28
29 sy=trc/(cos(phi)*cos(lambda));
30 printf("\nYield strength is \%.1 f MPa\n", sy);
31
32 / End
```

Scilab code Exa 7.2 Tensile Strength and Ductility Determinations

```
printf("\nCold work is %.1 f %%\n", CW);

ts=340; //in Mpa tensile strength

duc=7; //in % Ductility

printf("\nFrom graph of Fig. 7.19b in book");

printf("\nTensile strength is %d MPa", ts);

printf("\nDuctility is %d %% EL\n", duc);

//End
```

Scilab code Exa 7.3 Description of Diameter Reduction Procedure

```
1 // Description of Diameter Reduction Procedure
2
3 clear all;
4 clc;
  printf("\tDesign Example 7.1\n");
            //Initial dia in mm
8 \text{ di} = 6.4;
  df=5.1; //Final dia in mm
10
11 //Cold Work Computation
12 CW = ((di^2-df^2)/di^2)*100;
13
14 printf("\nCold work is \%.1 \text{ f } \% \n", CW);
15
16 //From Figures 7.19a and 7.19c,
17 //A yield strength of 410 MPa
18 //And a ductility of 8% EL are attained from this
      deformation
19
20 printf("\nBut required ductility and yield strength
      is not matched at this cold work");
21 printf("\nHence required Cold work is 21.5 \%\%");
```

Failure

Scilab code Exa 8.1 Maximum Flaw Length Computation

```
1 //Maximum Flaw Length Computation
3 clear all;
4 clc;
6 printf("\tExample 8.1\n");
8 sigma=40*10^6; // in Pa Tensile stress
9 E=69*10^9; //Modulus of elaticity
10 Ys=0.3; //Specific surface energy
                                        in N/m^2
11
12 //Maximum length of a surface flaw
13 a=2*E*Ys/(%pi*sigma^2);
14
15 printf("\nMaximum lemgth of a surface falw without
      fracture : \%.4 \text{ f mm} n, a/10^-3);
16
17 / End
```

Scilab code Exa 8.2 Rupture Lifetime Prediction

```
//Rupture Lifetime Prediction
clear all;
clc;
printf("\tDesign Example 8.2\n");

T=800+273; // Temperature in K

//stress is 140 MPa
//stress is 140 MPa
//From Graph of Fig. 8.32 Larson-Miller Parameter is deduced
L_M=24*10^3;

t=10^((L_M/T)-20);

printf("\nTime to rupture is : %d hours\n",t);
//End
```

Scilab code Exa 8.3 Estimating theoretical fracture strength

```
11
12 sigm=2*sig*sqrt(a/ro);
13
14 printf("\nFracture strength is %.2e MPa\n",sigm);
15
16 //End
```

Scilab code Exa 8.4 Computation of critical shear stress

```
1 //Computation of critical shear stress
3 clear all;
4 clc;
6 printf("\tExample 8.4\n");
8 E = 393D9;
                          //Young's modulus for Al
                          //surface energy in J/m^2
9 \text{ gam} = 0.9;
                          //Crack length in m
10 a=4D-4;
11
12 sigc=sqrt(2*E*gam/(%pi*a/2));
13
14 printf("\nCritical shear stress is \%.2e\ N/m^2\n",
      sigc);
15
16 / End
```

Scilab code Exa 8.5 Determining maximum allowable surface crack length

```
1 // Determining maximum allowable surface crack length
2
3 clear all;
4 clc;
```

Phase Diagrams

Scilab code Exa 9.1 Lever Rule Derivation

```
1 //Lever Rule derivation
2
3 clear all;
4 clc;
5
6 printf("\tExample 9.1\n");
7
8 disp("Since only 2 phases are present");
9 disp("W_alpha + W_L = 1");
10 disp("W_alpha*C_alpha + W_L*C_L = C0");
11 disp("hence");
12
13 disp("W_L = (C_alpha-C0)/(C_alpha-C_L)");
14 disp("W_alpha = (C0-C_L)/(C_alpha-C_L)");
15
16
17 //End
```

Scilab code Exa 9.2 Determination of Phases Present

Scilab code Exa 9.3.a Relative Phase Amount Determinations

```
// Determination of Phases Present

clear all;
clc;

printf("\tExample 9.3\n");

printf("\n\tPart A");

Cl=40; // Overall alloy composition

Cb=98;

Ca=10;

Wa=(Cb-C1)/(Cb-Ca);

Wb=(C1-Ca)/(Cb-Ca);
```

```
16 printf("\nMass fractions for alpha and beta phases
         are : %.2 f and %.2 f respectively\n", Wa, Wb);
17
18 //End
```

Scilab code Exa 9.3.b Mass and Volume Fractions

```
1 //Computation of Phase Compositions
3 clear all;
4 clc;
6 printf("\tExample 9.3\n");
            // Overall alloy composition
8 C1 = 40;
9 Cb = 98;
10 \text{ Ca=10};
11
12 Wa = (Cb - C1) / (Cb - Ca);
13 Wb = (C1 - Ca) / (Cb - Ca);
14
15 printf("\n\tPart B");
16 d_{Sn}=7.24; // in g/cm^3 density of tin
17 d_Pb=11.23; // in g/cm^3 density of lead
18
19 Ca_Sn=10;
20 Ca_Pb=90;
21
22 Cb_Sn=98;
23 \text{ Cb_Pb=2};
24
d_a=100/((Ca_Sn/d_Sn)+(Ca_Pb/d_Pb));
26 d_b=100/((Cb_Sn/d_Sn)+(Cb_Pb/d_Pb));
27
28 printf("\nDensity of alpha phase is : \%.2 \, \mathrm{f} \, \mathrm{g/cm}^3",
```

Scilab code Exa 9.4.a fractions of total ferrite and cementite phases

```
1 // Determining ferrite and cementite phase
2
3 clear all;
4 clc;
6 printf("\t Example 9.4\n");
8 printf("\n\t Part A");
9 \quad C0 = 0.35;
10 Ca=0.022;
11 C_Fe3C=6.7;
12
13 Wa=(C_Fe3C-C0)/(C_Fe3C-Ca);
14 \text{ W_Fe3C=(C0-Ca)/(C_Fe3C-Ca)};
15
16 printf("\nMass fraction of total ferritic phase : %
      .2 f", Wa);
17 printf("\nMass fraction of Fe3C : \%.2 \text{ f} \ \text{n}", W_Fe3C);
18
19 / End
```

Scilab code Exa 9.4.b Fractions of the proeutectoid ferrite and pearlite

```
1 // Determining proeutectoid ferrite and pearlite
3 clear all;
4 clc;
6 printf("\t Example 9.4\n");
8 \quad C0 = 0.35;
9 \text{ Ca=0.022};
10 C_Fe3C=6.7;
11
12 printf("\n\tPart B");
13 C_p = 0.76;
14
15 Wp = (CO - Ca) / (C_p - Ca);
16 W_a = (C_p - C_0) / (C_p - C_a);
17
18 printf("\nMass fraction of Pearlite: %.2f", Wp);
19 printf("\nMass fraction of proeutectoid ferrite : %
      .2 \text{ f} \n", W_a);
20
21 //End
```

Scilab code Exa 9.4.c Fraction of eutectoid ferrite

```
1 // Determining eutectoid ferrite
2
3 clear all;
4 clc;
```

```
5
6 printf("\t Example 9.4\n");
8 \quad C0 = 0.35;
9 \text{ Ca=0.022};
10 C_Fe3C=6.7;
11
12 \quad C_p = 0.76;
13
14 Wp = (CO - Ca) / (C_p - Ca);
15 W_a = (C_p - C0) / (C_p - Ca);
17 Wa=(C_Fe3C-C0)/(C_Fe3C-Ca);
18 printf("\n\tPart C");
19
20 Wae=Wa-W_a;
21 printf("\nMass fraction of eutectoid ferrite : \%.3 f\
      n", Wae);
22
23 //End
```

Phase Transformations in Metals

Scilab code Exa 10.1.a Computation of Critical Nucleus Radius

```
1 //Computation of Critical Nuclear Radius
2
3 clear all;
4 clc;
5
6 printf("\tExample 10.1\n");
7
8 printf("\n\tPart A");
9 Hf=-1.16*10^9; // in J/m^3 latent heat of fusion
10 Y=0.132; // in J/m^2 Surface energy
11 Tm=1064+273; // in K Melting point of gold
12 T=1064+273-230; // in K 230 is supercooling value
13
14 r=-2*Y*Tm/(Hf*(Tm-T));
15
16 printf("\nCritical Radius is : %.2f nm\n",r/10^-9);
17
18 G=16*%pi*Y^3*Tm^2/(3*Hf^2*(Tm-T)^2);
19
```

Scilab code Exa 10.1.b Activation Free Energy Compputation

```
1 // Activation Free Energy
3 clear all;
4 clc;
6 printf("\tExample 10.1\n");
8 Hf = -1.16*10^9; // in J/m^3 latent heat of fusion
9 Y=0.132; // \text{ in } J/m^2
                          Surface energy
10 Tm=1064+273; // in K Melting point of gold
11 T=1064+273-230; // in K 230 is supercooling value
12 r = -2 * Y * Tm / (Hf * (Tm - T));
13
14 printf("\n\tPart B");
15 a=0.413*10^-9; // in m Unit Cell edge length
16
17 //unit cells per paticle
18 u_c=4*\%pi*r^3/(3*a^3);
19
20 printf("\nUnit cells per paricle are : %d",u_c);
21 printf("\nIn FCC there are 4 atoms per unit cell.\n"
     );
22
23 printf("\nTotal no. of atoms per critical nucleus
      are : %d\n", int(u_c)*4);
24
25 //End
```

Scilab code Exa 10.2 Determination the rate of recrystallization

```
1 // Determination the rate of recrystallization
3 clear all;
4 clc;
6 printf("\tExample 10.2\n");
7 n=5;
8 y = 0.3;
                    //in min
9 t = 100;
10
11
12 k=-\log(1-y)/t^n;
13
14 thalf=(-\log(1-0.5)/k)^(1/n);
15
16 rate=1/thalf;
17
18 printf("\nRate is %.2e (min)^-1^n, rate);
19
20 //End
```

Structures and Properties of Ceramics

Scilab code Exa 12.1 Computation of Minimum Cation to Anion Radius Ratio

```
//Computation of Minimum Caion-to-Anion Radius Ratio
    forCo-ordination No. of 3

clear all;
clc;

printf("\tExample 12.1\n");

printf("\nFor equilateral triangle after joining centres of the atoms \nAngle = 30\n");

a=30;

ratio=(1-cos(30*%pi/180))/cos(30*%pi/180);

printf("\nCation to anion raio is : %.3f\n",ratio);
//End
```

Scilab code Exa 12.2 Ceramic Crystal Structure Prediction

```
1 // Ceramic Crystal structure prediction
3 clear all;
4 clc;
6 printf("\tExample 12.2\n");
8 r_Fe=0.077; // in nm Radius of iron cation Fe++
10 r_0=0.140; //in nm Radius of Oxygen anion O—
11
12 ratio=r_Fe/r_0;
13
14 printf("\nRatio is : \%0.3 \, f", ratio);
16 if ratio > 0.414 & ratio < 0.732 then
17
       printf("\nCo-ordinaton no. is 6");
18
       printf("\nStructure is Rock Salt type\n");
19 end
```

Scilab code Exa 12.3 Theoretical Density Calculation

```
//Theoretical Density Determination for NaCl
clear all;
clc;
printf("\tExample12.3\n");
```

```
8 A_Na = 22.99; // in g/mol
9 A_C1=35.45; //in g/mol
10
11 r_Na=0.102*10^-7; //in cm Radius of Na+ ion
12 r_Cl=0.181*10^-7; //in cm Radius of Cl- ion
13
14 \ a=2*(r_Na+r_Cl);
15 V=a^3;
16
17 n=4; //For FCC, no. of atoms are 4 per crystal
18
19 Na=6.023*10^23; //Avogadro number
20
21 density=n*(A_Na+A_C1)/(V*Na);
22
23 printf("\nDensity is : \%0.2 \,\mathrm{f}\,\mathrm{g/cm^3}\n", density);
24
25 //End
```

Scilab code Exa 12.4 Computation of the Number of Schottky Defects

```
//Computation of the No. of Schottky Defects in KCl
clear all;
clc;
printf("\tExample 12.4\n");

Na=6.023*10^23; //Avogadro number
density=1.955; //in g/cm^3

A_K=39.1; //in g/mol
A_Cl=35.45; //in g/mol
N=Na*density*10^6/(A_K+A_Cl);
```

Scilab code Exa 12.5 Determination of Possible Point Defect Types

```
//Determination of Possible Point Defect Types

clear all;
clc;

printf("\tExample 12.5\n");

disp("Replacement of Na+ by a Ca++ ion introduces one extra positive charge");

disp("Removal of a positive charge is accomplished by the formation of one Na+ vacancy");

disp("Alternatively, a Cl- interstitial will supply an additional negative charge, negating the effect of each Ca++ ion.");

disp("The formation of this defect is highly
```

```
unlikely.");
15
16 //End
```

Polymer Structures

Scilab code Exa 14.2.a Computations of the Density

```
1 //Computations of the Density
3 clear all;
4 clc;
6 printf("\tExample 14.2\n");
8 printf("\n\tPart A");
9 Ac=12.01; //in g/mol
                          Molecular weight of Carbon
10 Ah=1.008; //in g/mol molecular weight of hydrogen
11 a=7.41*10^-8; //in cm
12 b=4.94*10^-8; //in cm
13 c=2.55*10^-8; //in cm
14 \text{ Na}=6.023*10^23;
15
16 Vc=a*b*c;
17 n=2;
18 A = (2 * Ac) + (4 * Ah);
19
20 density_c=n*A/(Vc*Na);
21
```

Scilab code Exa 14.2.b Computation of Percent Crystallinity

```
1 // Percent Crystallinity of Polyethylene
3 clear all;
4 clc;
6 printf("\tExample 14.2\n");
8 printf("\n\tPart B");
9 density_a=0.870; // in g/cm^3
10 density_s=0.925; // in g/cm^3
11
                           Molecular weight of Carbon
12 Ac=12.01;
             //in g/mol
             //in g/mol
                           molecular weight of hydrogen
13 Ah=1.008;
14 \quad a=7.41*10^-8;
                  //in cm
15 b=4.94*10^-8;
                  //in cm
                  //in cm
16 c=2.55*10^-8;
17 Na=6.023*10^23;
18
19 Vc=a*b*c;
20 n = 2;
21 A = (2 * Ac) + (4 * Ah);
22
23 density_c=n*A/(Vc*Na);
24
25 pc=density_c*(density_s-density_a)*100/(density_s*(
      density_c-density_a));
26
27 printf("\npercentage crystallinity is: %.1f %%\n",
     pc);
```

```
28
29 //End
```

Scilab code Exa 14.3.a Computations of Diffusion Flux of Carbon Dioxide

```
1 //Computations of Diffusion Flux of Carbon dioxide
      through Plasic Beverage Container
2
3 clear all;
4 clc;
5
  function[A] = approx(V,n)
7
       A = round(V*10^n)/10^n;
       funcprot(0)
8
  endfunction
9
10
11 printf("\tExample 14.3\n");
12
13 printf("\n\tPart A");
14 P1=400000; // in Pa Pressure inside the bottle
15 P2=400; // in Pa Pressure outside the bottle
16 Pm=0.23*10^-13; //Solubility Coefficient
17 dx=0.05; // in cm Thickness of wall
18
19 J=approx(-Pm*(P2-P1)/dx,8);
20 printf("\nDiffusion flux is : \%.2 \, \mathrm{f} * 10 \, \hat{} -6",J
      /10^-6);
21 printf("cm^3 STP/cm^2-s\n");
22
  //End
23
```

Scilab code Exa 14.3.b Computation of Beverage Shelf Life

```
1 //Beverage Shell Life
3 clear all;
4 clc;
6 function[A] = approx(V,n)
       A = round(V*10^n)/10^n;
       funcprot(0)
9 endfunction
10
11 printf("\tExample 14.3\tn");
13 P1=400000; // in Pa Pressure inside the bottle
14 P2=400; // in Pa Pressure outside the bottle
15 Pm=0.23*10^-13; //Solubility Coefficient
16 dx=0.05; // in cm Thickness of wall
17
18 J=approx(-Pm*(P2-P1)/dx,8);
19
20 printf("\n\tPart B");
21 A=500; //surface area of bottle in cm<sup>2</sup>
22 V_lose=750; //\text{cm}^3 STP
23
24 V = J * A;
25 t=V_lose/V;
26 \text{ time=t/(3600*24)};
27
28 printf("\nTime to escape is : %.2e sec or %d days\n"
      ,t,time);
29
30 / End
```

Composites

Scilab code Exa 16.1.a Modulus of elasticity

```
1 clear all;
2 clc;
3
4 x=poly([0], 'x');
5 printf("\tExample 16.1\n");
7 printf("\n\tPart A");
8 E_gf=69; // in GPa Elasticity of glass fibre
9 mf_gf=0.4; //Vol % of glass fibre
10 E_pr=3.4; // in GPa Elasticity of poyester resin
11 mf_pr=0.6; //Vol % of polyester resin
12
13 E_cl=(E_pr*mf_pr)+(E_gf*mf_gf);
14 printf("\nModulus of elasticity of composite is : %f
      GPa\n", E_cl);
15
16 //End
```

Scilab code Exa 16.1.b Magnitude of the load

```
1 clear all;
2 clc;
3
4 x = poly([0], 'x');
5 printf("\tExample 16.1\n");
7 E_gf=69; // in GPa Elasticity of glass fibre
8 mf_gf=0.4; //Vol % of glass fibre
9 E_pr=3.4; // in GPa Elasticity of poyester resin
10 mf_pr=0.6; //Vol % of polyester resin
11
12 printf("\n\tPart B");
13 Ac = 250; /\text{mm}^2
14 sigma=50; //MPa
15 f=(E_gf*mf_gf)/(E_pr*mf_pr);
16 Fc=Ac*sigma; //N
17 Fm = roots(f*x+x-Fc); //N
18 printf("\nFm is : \%f N\n",Fm);
19
20 Ff = Fc - Fm;
21 printf("\nLoad carried by each of fiber and matrix
      phase is : \%f N \ n", Ff);
22
23 / End
```

Scilab code Exa 16.1.c Strain determination

```
1 clear all;
2 clc;
3
4 x=poly([0],'x');
5 printf("\tExample 16.1\n");
6
7 E_gf=69; // in GPa Elasticity of glass fibre
8 mf_gf=0.4; //Vol % of glass fibre
```

```
9 E_pr=3.4; // in GPa Elasticity of poyester resin
10 mf_pr=0.6; //Vol % of polyester resin
11
12 Ac=250; /\text{mm}^2
13 sigma=50; //MPa
14 f=(E_gf*mf_gf)/(E_pr*mf_pr);
15 Fc=Ac*sigma; //N
16 Fm = roots(f*x+x-Fc); //N
17
18 Ff = Fc - Fm;
19
20 printf("\n\tPart C");
21 \quad Am = mf_pr * Ac;
22 Af = mf_gf * Ac;
23 sigma_m=Fm/Am;
24 sigma_f=Ff/Af;
25
26 e_m=sigma_m/E_pr; //Strain for matrix phase
27 e_f=sigma_f/E_gf; //Strain for fiber phase
28
29 printf("\nStrain for matrix phase is : \%f\n",e_m);
30 printf("\nStrain for fiber phase is : \%f\n", e_f);
31
32 / End
```

Scilab code Exa 16.2 Elastic Modulus Determination

```
// Elastic Modulus Determination for a Glass Fiber-
Reinforced Composite Transverse Direction

clear all;
clc;
printf("\tExample 16.2\n");
E_gf=69; // in GPa Elasticity of glass fibre
```

```
8 mf_gf=0.4; //Vol % of glass fibre
9 E_pr=3.4; // in GPa Elasticity of poyester resin
10 mf_pr=0.6; //Vol % of polyester resin
11
12 E_ct=E_pr*E_gf/((E_pr*mf_gf)+(E_gf*mf_pr)); //GPa
13
14 printf("\nIn transverse direction, modulus of elaticity is: %f GPa\n",E_ct);
15
16 //End
```

Corrosion and Degradation of Materials

Scilab code Exa 17.1 Determination of Electrochemical Cell Characteristics

```
1 // Determination of Electrochemical Cell
      Characteristics
3 clear all;
4 clc;
6 printf("\tExample 17.1\n");
8 V_Cd=-0.403; // Half Cell Potential of Cd++|Cd
9 V_Ni = -0.250;
                 //Half Cell Potential of Ni++|Ni
10 dV = V_Ni - V_Cd;
11 printf("\nStandard Cell potential is : \%f V\n", dV);
12
13 C_Ni=10^-3;
14 C_Cd=0.5;
15 n=2; //Net electron exchange in Redox reaction
16 V=-dV-(0.0592*log10(C_Ni/C_Cd)/n);
17 printf("\nNet EMF is : \%f V\n", V);
```

```
18 printf("\nHence\n");
19
20 if V<0 then
21    printf("\nNi is reduced & Cd is oxidised\n");
22 else
23    printf("\nCd is reduced & Ni is oxidised\n");
24 end
25
26 //End</pre>
```

Scilab code Exa 17.2.a Rate of Oxidation Computation

```
1 //Rate of Oxidation Computation
2
3 clear all;
4 clc;
6 printf("\tExample 17.2a\n");
8 // Activation polarisation data
9 VH_H2=0;
10 VZn_Zn2 = -0.763;
11 iZn=10^-7;
12 iH2=10^-10;
13 beta_Zn=0.09;
14 beta_H2 = -0.08;
15
16 // Part i
17 ic=10^{(VH_H2-VZn_Zn2-(beta_H2*log10(iH2))+(beta_Zn*log10(iH2))}
      log10(iZn)))/(beta_Zn-beta_H2)];
18
19 disp(ic, 'ic is ');
20
21 n=2;
                       //Exchange of 2 electrons
22 F = 96500;
                       //Faradays constant
```

Scilab code Exa 17.2.b Corrosion potential determination

```
1 clear all;
     2 clc;
     4 printf("\tExample 17.2b\n");
     6 // Activation polarisation data
     7 VH_H2=0;
    8 VZn_Zn2 = -0.763;
    9 iZn=10^-7;
10 iH2=10^-10;
11 beta_Zn=0.09;
12 beta_H2 = -0.08;
13
14 ic=10^{(VH_H2-VZn_Zn2-(beta_H2*log10(iH2))+(beta_Zn*log10(iH2))+(beta_Zn*log10(iH2))+(beta_Zn*log10(iH2))+(beta_Zn*log10(iH2))+(beta_Zn*log10(iH2))+(beta_Zn*log10(iH2))+(beta_Zn*log10(iH2))+(beta_Zn*log10(iH2))+(beta_Zn*log10(iH2))+(beta_Zn*log10(iH2))+(beta_Zn*log10(iH2))+(beta_Zn*log10(iH2))+(beta_Zn*log10(iH2))+(beta_Zn*log10(iH2))+(beta_Zn*log10(iH2))+(beta_Zn*log10(iH2))+(beta_Zn*log10(iH2))+(beta_Zn*log10(iH2))+(beta_Zn*log10(iH2))+(beta_Zn*log10(iH2))+(beta_Zn*log10(iH2))+(beta_Zn*log10(iH2))+(beta_Zn*log10(iH2))+(beta_Zn*log10(iH2))+(beta_Zn*log10(iH2))+(beta_Zn*log10(iH2))+(beta_Zn*log10(iH2))+(beta_Zn*log10(iH2))+(beta_Zn*log10(iH2))+(beta_Zn*log10(iH2))+(beta_Zn*log10(iH2))+(beta_Zn*log10(iH2))+(beta_Zn*log10(iH2))+(beta_Zn*log10(iH2))+(beta_Zn*log10(iH2))+(beta_Zn*log10(iH2))+(beta_Zn*log10(iH2))+(beta_Zn*log10(iH2))+(beta_Zn*log10(iH2))+(beta_Zn*log10(iH2))+(beta_Zn*log10(iH2))+(beta_Zn*log10(iH2))+(beta_Zn*log10(iH2))+(beta_Zn*log10(iH2))+(beta_Zn*log10(iH2))+(beta_Zn*log10(iH2))+(beta_Zn*log10(iH2))+(beta_Zn*log10(iH2))+(beta_Zn*log10(iH2))+(beta_Zn*log10(iH2))+(beta_Zn*log10(iH2))+(beta_Zn*log10(iH2))+(beta_Zn*log10(iH2))+(beta_Zn*log10(iH2))+(beta_Zn*log10(iH2))+(beta_Zn*log10(iH2))+(beta_Zn*log10(iH2))+(beta_Zn*log10(iH2))+(beta_Zn*log10(iH2))+(beta_Zn*log10(iH2))+(beta_Zn*log10(iH2))+(beta_Zn*log10(iH2))+(beta_Zn*log10(iH2))+(beta_Zn*log10(iH2))+(beta_Zn*log10(iH2))+(beta_Zn*log10(iH2))+(beta_Zn*log10(iH2))+(beta_Zn*log10(iH2))+(beta_Zn*log10(iH2))+(beta_Zn*log10(iH2))+(beta_Zn*log10(iH2))+(beta_Zn*log10(iH2))+(beta_Zn*log10(iH2))+(beta_Zn*log10(iH2))+(beta_Zn*log10(iH2))+(beta_Zn*log10(iH2))+(beta_Zn*log10(iH2))+(beta_Zn*log10(iH2))+(beta_Zn*log10(iH2))+(beta_Zn*log10(iH2))+(beta_Zn*log10(iH2))+(beta_Zn*log10(iH2))+(beta_Zn*log10(iH2))+(beta_Zn*log10(iH2))+(beta_Zn*log10(iH2))+(beta_Zn*log10(iH2))+(beta_Zn*log10(iH2))+(beta_Zn*log10(iH2))+(beta_Zn*log10(iH2))+(beta_Zn*log10(iH2))+(beta_Zn*log10(iH2))+(beta_Zn*log10(iH2))+(beta_Zn*log10(iH2))+(beta_Zn*log10(iH2))+(beta_Zn*log10(iH2))+(be
                                     log10(iZn)))/(beta_Zn-beta_H2)];
15
16 //Part ii
17 Vc=VH_H2+(beta_H2*log10(ic/iH2));
18 printf("\nii) Corrosion potential is \%.3 \, f \, V \ ", Vc);
19
20 / End
```

Scilab code Exa 17.3 Temperature Computation

```
1 //Temperature Computation
3 clear all;
4 clc;
6 printf("\tExample 17.3\n");
                                //Potential diff. b/w 2
8 \text{ dV} = 0.568;
      electrodes
9 V_Pb = -0.126;
10 V_Zn = -0.763;
11
12 \quad C_Zn = 0.01;
13 C_Pb=0.0001;
14
                             //Gas constt
15 R=8.31;
                             //Faraday's constt
16 F=96500;
17 n=2;
                             //electron exchange
18
19 T=-n*F*(dV-(V_Pb-V_Zn))/(R*log(C_Zn/C_Pb));
20
21 printf("\nFinal temp is \%.1 f K \n",T);
22
23 //End
```

Electrical Properies

Scilab code Exa 18.1 Computation of the Room Temperature Intrinsic Carrier Concentration

```
1 // Computation of the Room-Temperature Intrinsic
      Carrier Concentration for Gallium Arsenide
3 clear all;
4 clc;
6 printf("\t Example 18.1\n");
8 sigma=10^-6; // (Ohm-m)^-1 Electrical Conductivity
10 e=1.6*10^-19; //Coulomb Charge on electron
11
12 m_e=0.85; //\text{m}^2/\text{V-s} Mobility of electron
13
14 m_h=0.04; //\text{m}^2/\text{V}-s Mobility of holes
15
16 //ni is Intrinsic carrier concentration
17 ni=sigma/(e*(m_e+m_h));
18
19 printf("\n Intrinsic Carrier Concentration is: %f m
```

```
^-3\n",ni);
20
21 //End
```

Scilab code Exa 18.2 Electrical Conductivity Determination for Intrinsic Silicon

```
1 // Electrical Conductivity Determination for
      Intrinsic Silicon at 150 C
3 clear all;
4 clc;
6 printf("\t Example 18.2\n");
8 e=1.6*10^-19; //Coulomb Charge on electron
10 ni=4*10^19; //For Si at 423 K
                                    (m^{-3})
12 // Values of m_e and m_h are deduced from graphs at
     page No.689
13
14 m_e=0.06; //m^2/V-s Mobility of electron
15
16 m_h=0.022; ///m^2/V-s Mobility of holes
17
18 //sigma is electrical conductivity
19 sigma=ni*e*(m_e+m_h);
20
21 printf("\nElectrical Conductivity is : %f (Ohm-m)
      ^-1\n", sigma);
22
23 //End
```

Scilab code Exa 18.3.b Room Temperature for Extrinsic Silicon

```
//Room-Temperature for Extrinsic Silicon
3 clear all;
4 clc;
6 printf("\tExample 18.3\n");
7 printf("\n\t Part B\n");
9 n=10^23; //\text{m}^-3 Carrier Concentration
10
  e=1.6*10^-19; //Coulomb Charge on electron
11
12
13 //From graph 18.18 m_e is calculated corresponding
      to n=10^23
14
15 m_e=0.07; //\text{m}^2/\text{V-s} Mobility of electron
16
17 //For extrinsic n-type, the formula used is:
18
19 sigma=n*e*m_e;
20
21 printf("\nConductivity at n=10^23 is : %d (Ohm-m)
      -1 n, sigma);
22
23
24 / End
```

Scilab code Exa 18.3.c Elevated Temperature Electrical Conductivity Calculations

```
1 // Elevated-Temperature Electrical Conductivity
      Calculations for Extrinsic Silicon
3 clear all;
4 clc;
6 printf("\tExample 18.3\n");
8 n=10^23; //m^-3 Carrier Concentration
10 e=1.6*10^-19; //Coulomb Charge on electron
11
12 printf("\n\tPart \C\n");
13
14 //From graph 18.19a m_e2 is calculated corresponding
       to 373 K
15
16 m_e2=0.04; //m^2/V-s Mobility of electron
17
18 \text{ sigma2=n*e*m_e2};
19
20 printf("\nConductivity at T=373 K becomes: %d (Ohm-
     m)^-1\n", sigma2);
21
22 //End
```

Scilab code Exa 18.4 Hall Voltage Computation

```
// Hall Voltage Computation
clear all;
clc;
printf("\tExample 18.4\n");
```

```
8 sigma=3.8*10^7; //(Ohm-m)^-1 Electrical
      Conductivity
9
10 m_e=0.0012; //\text{m}^2/\text{V-s} Mobility of electron
11
12 Rh=-m_e/sigma; //Hall coefficient
13
14 printf("\nHall coefficient is: \%f * 10^--11 V-m/A-
      Tesla \n", Rh/10^-11);
15
16 Ix=25; //Ampere(A) Current
17
18 d=15*10^-3; //m Thickness
19
20 Bz=0.6; //Tesla Magnetic field
21
22 Vh=Rh*Ix*Bz/d;
23
24 printf("\nHall Voltage is : \%f * 10^-8 V\n", Vh
      /10^-8);
25
26 //End
```

Scilab code Exa 18.6 Acceptor Impurity Doping in Silicon

```
10 uh=[0.04 0.045 0.04]; //Mobility of holes
11
12 for i=1:3
        sigma(i)=p(i)*e*uh(i);
14 end
15
16 \operatorname{disp}('(\operatorname{Ohm-m})^-1', \operatorname{sigma}(3), '\operatorname{conductivity} is', 'm^2/V-
      s', uh(3), m-3', p(3), For hole conc and mobility '
       );
17
                                   //Avogadro no.
18 \text{ Na} = 6.023 \text{ D} 23;
                                   //Density of silicon in g/
19 den_Si=2.33D6;
      m^3
                                   //Atomic weight of silicon
20 A_Si = 28.09;
21
22 N_Si=Na*den_Si/A_Si;
23
24
25 Ca=p(3)/(p(3)+N_Si)*100;
26
27 printf("\nThus, Silicon material must contain %.1f *
        10^-5 \% B, Al, Ga or In n", Ca/10^-5);
28
29 //End
```

Chapter 19

Thermal Properties

Scilab code Exa 19.1 Thermal Stress Created Upon Heating

```
1 // Calculation of maximum temperature
3 clear all;
4 clc;
6 printf("\t Example 19.1\n");
8 To=20; // Room Temperature (degree celsius)
10 sigma = -172; //Mpa Compressive stress
12 E=100*10^3; //Mpa Young's modulus
13
14 a=20*10^-6; //Celsius^-1 Coefficient of thermal
     expansion
15
16 Tf=To-(sigma/(E*a));
17
18 printf("\nFinal Temperature is : %d C\n", Tf);
19
20 //End
```

Scilab code Exa 19.2 Final temperature Calculation

```
1 // Final temperature Calculation
3 clear all;
4 clc;
6 printf("\tExample 19.2\n");
                         // mass in lbm
8 m = 10;
                         //Heat supplied in Btu
9 dQ = 65;
                         //Initial temp in F
10 To = 77;
11
12 Cp=375*2.39*10^-4; //in Btu/lbm - F
13
14 dT=dQ/(m*Cp);
15
16 Tf = To + dT;
17
18 printf("\nFinal temp is \%.1 f F\n", Tf);
19
20 / End
```

Scilab code Exa 19.3 Computation of specific heats for Al and Fe

```
1 //Computation of specific heats for Al and Fe 2 3 clear all; 4 clc; 5 6 printf("\tExample 19.3\n");
```

```
7
                               //In J/kg-K
8 \text{ Cp\_Al} = 900;
                               //In j/kg-K
9 Cp_Fe = 448;
10
11 beta_Al=1.77D-11;
12 \text{ beta_Fe=} 2.65D-12;
13
14 T = 273;
15
16 alphal_A1=23.6D-6;
17 alphal_Fe=11.8D-6;
18
19 alphav_Al=3*alphal_Al;
20 alphav_Fe=3*alphal_Fe;
21
                                  //in kg/m^3
22 \, \text{den\_Al} = 2.71 \, \text{D3};
                                  //in kg/m^3
23 \, \text{den_Fe=7.87D3};
24
25 \text{ vo\_Al=1/den\_Al};
26 vo_Fe=1/den_Fe;
27
28 Cv_Al=Cp_Al-(alphav_Al^2*vo_Al*T/beta_Al);
29 Cv_Fe=Cp_Fe-(alphav_Fe^2*vo_Fe*T/beta_Fe);
30
31 printf("\nCv (Al) = \%d J/kg-K", Cv_Al);
32 printf("\nCv (Fe) = \%d T/kg-K\n", Cv_Fe);
33
34 / End
```

Chapter 20

Magnetic Properties

Scilab code Exa 20.1.a Saturation Magnetization

```
1 //Example 20.1 Calculation of saturation
     magnetisation and flux density for Nickel
3 clear all;
4 clc;
6 printf("Example 20.1\n");
8 b_m=9.27*10^-24; //ampere*m^2 (Bohr Magneton)
10 Na=6.023*10^23; //atoms/mol (Avogadro's No.)
11
12 d=8.9*10^6; //g/m^3 (density)
13
14 uo=4*%pi*10^-7; // Permitivity of free space
15
16 A=58.71; //g/mol (Atomic weigth of Nickel)
17
18 N=d*Na/A; //No. of atoms per cubic meter
19
20 // M is saturation magnetisation
```

Scilab code Exa 20.1.b Flux Density Computations for Nickel

```
1 //Example 20.1 Calculation of saturation
     magnetisation and flux density for Nickel
2
3 clear all;
4 clc;
6 printf("Example 20.1\n");
8 b_m=9.27*10^-24; //ampere*m^2 (Bohr Magneton)
10 Na=6.023*10^23; //atoms/mol (Avogadro's No.)
11
12 d=8.9*10^6; //g/m^3 (density)
13
14 uo=4*%pi*10^-7; // Permitivity of free space
15
16 A=58.71; //g/mol (Atomic weigth of Nickel)
17
18 N=d*Na/A; //No. of atoms per cubic meter
19
20 // M is saturation magnetisation
21 M=0.6*b_m*N; //0.6= Bohr Magneton/atom
22
23
```

```
24 //B = Saturation Flux Density
25 B=uo*M;
26
27 printf("\nSaturation Flux Density is : %f Tesla\n",B
     );
28
29 // End
```

Scilab code Exa 20.2 Saturation Magnetization Determination

```
1 //Example 20.2 Calculation of saturation
     magnetisation of Fe3O4
3 clear all;
4 clc;
5
6 printf("Example 20.2\n");
  a=0.839*10^-9; //a is edge length in m
10 b_m=9.27*10^-24; //ampere*m^2 (Bohr Magneton)
11
12 nb=8*4;
              //8 is no. of Fe++ ions per unit cell
13
              //4 is Bohr magnetons per Fe++ ion
14
15 M=nb*b_m/a^3; //M is Saturation magnetisation
16
  printf("\nSaturation Magnetisation is : %f Ampere/m\
17
     n",M);
18
19
  //End
```

Scilab code Exa 20.3 Design of a Mixed Ferrite Magnetic Material

```
1 // Design Example 20.1: Designing a cubic mixed-
      ferrite magnetic material
3 clear all;
4 clc;
6 printf(" Design Example 20.1\n");
8 x=poly([0], 'x'); // Defining X
10 Ms_Fe=5.25*10^5;
                    //Required saturation
      Magnetisation
11
12 b_m=9.27*10^-24; //ampere*m^2 (Bohr Magneton)
13
14 a=0.839*10^-9; //a is edge length in m
15
16 M=5*10^5; //From previous question result
17
18 nb=Ms_Fe*a^3/b_m;
19
  // 'x' represent fraction of Mn++ that have
20
      substituted Fe++
21
22 \text{ n=roots} (8*[5*x+4*(1-x)]-nb);
                                 //5 is Bohr magnetons
      per Fe++ ion
23
                                  //4 is Bohr magnetons
                                     per Mn++ ion
24
25 printf("\nReplacing %f percent of Fe++ with Mn++
      would produce the required saturation
      magnetisation \n", n*100);
26
27 //End
```

Chapter 21

Optical Properties

Scilab code Exa 21.1 Computation of the Absorption Coefficient for Glass

```
1 //Example 20.1 Calculation of absorption coefficient
3 clear all;
4 clc;
6 printf("\tExample 21.1\n");
8 // x is thickness of glass (mm)
9 x = 200;
10
11 //It is intensity of non-absorbed radiation
12 //Io is intensity of non-relected radiation
14 f=0.98; //f=It/Io
15
16 //b is absorption coefficient
17
18 b=-\log(f)/x;
19
20 printf("\nAbsorption coefficient is \%f mm^-1\n",b);
21
```

Scilab code Exa 21.2 Velocity of light in diamond

```
1 // Velocity of light in diamond
3 clear all;
4 clc;
5
6 printf("\tExample 21.2\n");
7
8 \text{ er=5.5};
                                  //Relative permitivity
                                  //Magnetic Suseptibility
9 \text{ xm} = -2.17D-5;
10
                                  //Permitivity in free
11 eo=8.85D-12;
      space
12 \text{ uo}=4*\%\text{pi}*10^-7;
                                  //Permeability
13
14 \text{ e=er*eo};
15 u=uo*(1+xm);
16
17 v=1/sqrt(u*e);
18
19 printf("\nVelocity in diamond is \%.2e m/s\n",v);
20
21 / End
```

Scilab code Exa 21.3 Suitable material required

```
1 // Suitable material required
2
3 clear all;
4 clc;
```

```
5
6 printf("\tExample 21.3\n");
8 R=0.05;
9 n=poly([0], 'x');
10
11 //R = (ns-1/ns+1)^2 leading to the equation
        0.95 \,\mathrm{ns} \, 2 - 2.1 \,\mathrm{ns} + 0.95 = 0
13 / HencE
14
15 ns=roots((0.95*n^2)-(2.1*n)+0.95);
16
17 disp(ns, 'The values of ns are: ");
18 printf("Thus, soda-lime glass, Pyrex glass, and
      polyprop would be suitable for this application
19
20 //End
```