Scilab Textbook Companion for Internal Combustion Engines by H. B. Keswani¹

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Book Description

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Scilab numbering policy used in this document and the relation to the above book.

Exa Example (Solved example)

Eqn Equation (Particular equation of the above book)

AP Appendix to Example(Scilab Code that is an Appednix to a particular Example of the above book)

For example, Exa 3.51 means solved example 3.51 of this book. Sec 2.3 means a scilab code whose theory is explained in Section 2.3 of the book.

Contents

Lis	st of Scilab Codes	4
1	INTRODUCTION	5
3	AIR STANDARD CYCLES	9
4	FUEL AIR CYCLES AND REAL CYCLES	35
5	VARIABLE SPECIFIC HEAT	37
6	COMBUSTION CHARTS	46
8	COMBUSTION	49
9	SPARK IGNITION ENGINES	72
11	FOUR STROKE SPARK IGNITION ENGINE	87
14	THE DIESEL ENGINE	94
15	FUEL INJECTION	96
16	COMBUSTION IN COMPRESSION IGNITION ENGINES	105
18	SUPERCHARGING	108
19	TWO STROKE ENGINES	113
23	TESTING OF ENGINES	115

25 GAS TURBINES	135
26 JET PROPULSION	159

List of Scilab Codes

Exa 1.1	MECHANICAL EFFICIENCY 5
Exa 1.2	MECHANICAL EFFICIENCY 5
Exa 1.3	BRAKE THERMAL EFFICIENCY 6
Exa 1.4	THERMAL EFFICIENCY
Exa 1.5	FUEL CONSUMPTION
Exa 3.1	AUR STANDARD EFFICIENCY 9
Exa 3.2	THERMAL EFFICIENCY
Exa 3.3	FUEL CONSUMPTION
Exa 3.4	RATIO
Exa 3.5	AIR STANDARD EFFICIENCY
Exa 3.6	HEAT AND WORK
Exa 3.7	AIR STANDARD EFFICIENCY
Exa 3.8	THERMAL EFFICIENCY
Exa 3.9	TEMPERATURE AND EFFICIENCY 15
Exa 3.10	PRESSURE AND HEAT 16
Exa 3.11	AIR STANDARD EFFICIENCY
Exa 3.12	AIR STANDARD EFFICIENCY
Exa 3.13	TEMPERATURE AND EFFICIENCY 18
Exa 3.14	AIR STANDARD EFFICIENCY 20
Exa 3.15	WORK DONE AND EFFICIENCY 21
Exa 3.16	TEMPERATURE AND EFFICIENCY 22
Exa 3.18	MEAN EFFECTIVE PRESSURE 24
Exa 3.19	CYLINDER SIZES
Exa 3.20	THERMAL EFFICIENCY
Exa 3.21	COMPRESSION RATIO
Exa 3.22	EFFICIENCY
Exa 3.26	PERCENTAGE CHANGE IN EFFICIENCY 28
Exa 3.30	MAXIMUM HORSE POWER 29

Exa 3.31	AIR FUEL RATIO	30
Exa 3.32	AIR GAS RATIO	31
Exa 3.33	FRICTION HORSE POWER	33
Exa 4.1	WORK DONE	35
Exa 5.2	CHANGE IN AIR STANDARD EFFICIENCY	37
Exa 5.3	WORK DONE	38
Exa 5.4	PERCENTAGE OF STROKE	38
Exa 5.5	PERCENTAGE OF STROKE	39
Exa 5.6	CHANGE IN ENTHALPY	40
Exa 5.7	TEMPERATURE AND PRESSURE	41
Exa 5.8	TEMPERATURE AND HEAT	42
Exa 5.10	PERCENTAGE CHANGE IN EFFICIENCY	44
Exa 6.1	EFFICIENCY	46
Exa 6.2	EFFICIENCY	47
Exa 6.3	EFFICIENCY	47
Exa 8.1	COMPOSITION	49
Exa 8.2	WEIGHT OF AIR	50
Exa 8.3	AIR AND VOLUME	51
Exa 8.4	WEIGHT OF AIR	52
Exa 8.5	MASS OF AIR	53
Exa 8.6	VOLUMETRIC ANALYSIS	54
Exa 8.7	PERCENTAGE COMBUSTION	55
Exa 8.8	PARTIAL PRESSURE	57
Exa 8.9	ANALYSIS BY WEIGHT	58
Exa 8.10	GRAVIMETRIC ANALYSIS	59
Exa 8.11	WEIGHT OF AIR	60
Exa 8.12	AIR SUPPLIED	61
Exa 8.13	PERCENTAGE COMPOSITION	62
Exa 8.14	EXCESS AIR	64
Exa 8.15		65
Exa 8.16	VOLUME OF AIR	66
Exa 8.17	VOLUMETRIC COMPOSITION	67
Exa 8.18	WEIGHT	68
Exa 8.19	WEIGHT OF AIR	70
Exa 9.1	WEIGHT	72
Exa 9.2	AIR FUEL RATIO	73
Exa 9.3	THEORETICAL RATIO OF DIAMETER	73
Exa 9.4	SECTION AT THE THROAT	74

Exa 9.5	SIZE OF THE VENTURI	75
Exa 9.6	FUEL CONSUMPTION	76
Exa 9.7	DIAMETER OF JET	77
Exa 9.8	AIR FUEL RATIO	78
Exa 9.9	VELOCITY	79
Exa 9.10	PETROL CONSUMPTION	79
Exa 9.11	AIR FUEL RATIO	80
Exa 9.12	THROAT PRESSURE	81
Exa 9.13	DROP IN PRESSURE	82
Exa 9.14	ORIFICE DIAMETER	83
Exa 9.15	ORIFICE DIAMETER	84
Exa 9.16	POWER DEVELOPED	85
Exa 11.1	WEIGHT	87
Exa 11.2	INDICATED THERMAL EFFICIENCY	88
Exa 11.3	PROPERTIES	89
Exa 11.4	WEIGHT OF AIR	90
Exa 11.5	CLEARANCE VOLUME	91
Exa 11.6	CYLINDER PROPERTIES	92
Exa 14.1	WEIGHT OF OIL	94
Exa 14.2	RATIO OF MASS	95
Exa 15.1	QUANTITY OF FUEL	96
Exa 15.2	QUANTITY OF FUEL	96
Exa 15.3	QUANTITY OF FUEL	97
Exa 15.4	SIZE OF PUMP	98
Exa 15.5	VELOCITY	99
Exa 15.6	PUMP DISPLACEMENT	99
Exa 15.7	PUMP DISPLACEMENT	100
Exa 15.8	PLUNGER STROKE	101
Exa 15.9	DIAMETER	102
Exa 15.10	AMOUNT OF FUEL	103
Exa 16.1	DELAY TIME	105
Exa 16.2	PRESSURE	106
Exa 16.3	BRAKE THERMAL EFFICIENCY	107
Exa 18.1	PRESSURE AND AIR CONSUMPTION	108
Exa 18.2	PERCENTAGE INCREASE IN IMEP	109
Exa 18.3		110
Exa 19.3		113
Exa 23.1		115

Exa 23.2	INDICATED HORSE POWER	116
Exa 23.3	CALORIFIC VALUE	116
Exa 23.4	BHP OF ENGINE	117
Exa 23.5	THERMAL EFFICIENCY	118
Exa 23.6	FUEL CONSUMPTION	119
Exa 23.7	PROPERTIES	120
Exa 23.8	HEAT BALANCE SHEET	121
Exa 23.9	HEAT BALANCE SHEET	122
Exa 23.10	VOLUMETRIC EFFICIENCY	123
Exa 23.11	HEAT BALANCE SHEET	124
Exa 23.12	HEAT BALANCE SHEET	126
Exa 23.13	BRAKE MEAN EFFECTIVE PRESSURE	127
Exa 23.14	HEAT BALANCE SHEET	128
Exa 23.15	HEAT BALANCE SHEET	130
Exa 23.16	HEAT BALANCE SHEET	132
Exa 23.17	HEAT LOST	133
Exa 25.2	OVERALL EFFICIENCY	135
Exa 25.3	EFFICIENCY	136
Exa 25.4	NET OUTPUT	137
Exa 25.5	TEMPERATURE	138
Exa 25.6	HP DEVELOPED	138
Exa 25.7	THERMAL EFFICIENCY	139
Exa 25.8	FLOW RATE	140
Exa 25.9	FLOW RATE	142
Exa 25.10	EFFICIENCY	143
Exa 25.11	EFFICIENCY	144
Exa 25.12	FLOW OF AIR	145
Exa 25.13	POWER OTPUT	147
Exa 25.14	EFFICIENCY	148
Exa 25.15	PRESSURE	149
Exa 25.16	EFFICIENCY	150
Exa 25.17	EFFICIENCY	151
Exa 25.18	EFFICIENCY	153
Exa 25.19	EFFICIENCY	155
Exa 25.20	HP	156
Exa 25.22	EFFICIENCY	157
Exa 26.1	PROPERTIES	159
Exa 26.2	AIR FUEL RATIO	159

Exa 26.3	PROPERTIES	160
Exa 26.4	AREA	161
Exa 26.5	POWER AND THRUST	162

Chapter 1

INTRODUCTION

Scilab code Exa 1.1 MECHANICAL EFFICIENCY

```
clc
clear
//Input data
HHP=51//Brake horse power in h.p
N=1000//Speed in r.p.m
FHP=17//Friction horse power in h.p

//Calculations
HHP=(BHP+FHP)//Indicated Horse power in h.p
mn=(BHP/IHP)*100//Mechanical efficiency in percent
//Output
printf('Mechanical efficiency of the engine is %i percent',mn)
```

Scilab code Exa 1.2 MECHANICAL EFFICIENCY

```
1 clc
```

```
2 clear
3 //Data taken from Ex.No.1
4 BHP=51//Brake horse power in h.p
5 N=1000/Speed in r.p.m
6 FHP=17//Friction horse power in h.p
7 //Input data
8 O1=BHP/2//Half of b.h.p output in h.p
9 02=10//Brake horse power in h.p.
10
11 // Calculations
12 // Case (i)
13 IHP1=(01+FHP)//Indicated Horse power in h.p.
14 mn1=(01/IHP1)*100//Mechanical efficiency in percent
15
16 // Case ( i i )
17 IHP2=(02+FHP)//Indicated Horse power in h.p.
18 mn2=(02/IHP2)*100//Mechanical efficiency in percent
19
20
21 // Output
22 printf ('Mechanical efficiency of the engine when it
      delivers \n (a) Half the b.h.p output is \%3.0 f
      percent \n (b) 10 b.h.p is \%3.0f percent', mn1, mn2
     )
```

Scilab code Exa 1.3 BRAKE THERMAL EFFICIENCY

```
1 clc
2 clear
3 //Input data
4 Fc=220//Fuel consumption in gm/(b.h.p*hr)
5 CV=10600//Calorific value in kcal/kg
6
7 //Calculations
8 hf=(Fc/1000)*CV//Heat supplied in kcal/hr
```

Scilab code Exa 1.4 THERMAL EFFICIENCY

```
1 clc
2 clear
3 //Input data
4 IHP=45//Indicated horse power in h.p
5 Fc=13//Fuel consumption in litres/hr
6 g=0.8//Specific gravity of oil
7 nm=80//Mechanical efficiency in percent
8 CV=10000//Calorific value of fuel in kcal/kg
9
10 // Calculations
11 BHP=(IHP*nm)/100//Brake horse power in h.p.
12 hi=(Fc*g*CV)//Heat supplied in kcal/hour
13 In=((IHP*4500*60)/(427*hi))*100//Indicated thermal
      efficiency in percent
14 Bn=(In*(nm/100))/Brake thermal efficiency in
     percent
15
16 //Output
17 printf ('Indicated thermal efficiency is \%3.2 f
      percent \n Brake thermal efficiency is \%3.2f
     percent', In, Bn)
```

Scilab code Exa 1.5 FUEL CONSUMPTION

```
1 clc
2 clear
3 //Input data
4 BHP=15//Brake horse power in h.p
5 In=28//Indicated thermal efficiency in percent
6 mn=75//Mechanical efficiency in percent
7 CV=10000//Calorific value of fuel in kcal/kg
9 // Calculations
10 Bn=((In/100)*(mn/100))*100//Brake thermal efficiency
      in percent
11 I = (BHP/(Bn/100))*((4500*60)/427)/Input in kcal/hr
12 Fc=(I/CV)//Fuel consumption in kg/hr
13
14 //Output
15 printf('Fuel consumption of the engine is %3.2 f kg/
     hr', Fc)
```

Chapter 3

AIR STANDARD CYCLES

Scilab code Exa 3.1 AUR STANDARD EFFICIENCY

```
clc
clear
//Input data
p=[1,8]//Pressure at the beginning and end of
    compression in kg/m^3
s=1.4//Ratio of specific heats

//Calculations
r=(p(2)/p(1))^(1/g)//Compression ratio
n=(1-(1/r)^(g-1))*100//Air standard efficiency in
    percent

//Output
printf('Air standard efficiency of an engine working
    on the Otto cycle is %3.1f percent',n)
```

Scilab code Exa 3.2 THERMAL EFFICIENCY

```
1 clc
2 clear
3 //Input data
4 D=0.25//Bore in m
5 L=0.45//Stroke in m
6 Cv=5//Clearance volume in litres
7 g=1.4//Ratio of specific heats
8 IHP=32//Indicated Horse power in h.p.
9 m=14//Gas consumption in m<sup>3</sup>/hr
10 CV=4000//Calorific value of gas in kcal/m<sup>3</sup>
11
12 // Calculations
13 Vs = (3.14/4) *D^2*L//Stroke volume in m^3
14 Vc=Cv/1000//Clearance volume in m<sup>3</sup>
15 r=(Vs+Vc)/Vc//Compression ratio
16 na=(1-(1/r)^{(g-1)})*100/Air standard efficiency in
      percent
17 q=(m*CV)/60//Heat supplied in kcal/min
18 aI = (IHP*4500)/427//Heat equivalent of I.H.P in kcal/
      min
  itn=(aI/q)*100//Indicated thermal efficiency in
19
      percent
20 rn=(itn/na)*100//Relative efficiency in percent
21
22 //Output
23 printf ('The air standard efficiency is \%3.1f percent
       \n Indicated thermal efficiency is \%3.1f percent
       \n Relative efficiency is \%3.1f percent', na, itn,
      rn)
```

Scilab code Exa 3.3 FUEL CONSUMPTION

```
1 clc
2 clear
3 //Input data
```

```
4 r=6//Compression ratio
5 It=0.6//Indicated thermal efficiency ratio
6 CV=10000//Calorific value in kcal/kg
7 g=1.4//Ratio of specific heats
8
9 //Calculations
10 an=(1-(1/r)^(g-1))*100//Air standard efficiency in percent
11 In=(It*(an/100))//Indicated thermal efficiency
12 SFC=((4500*60)/(427*CV*In))//Specific fuel consumption in kg/I.H.P.hr
13
14 //Output
15 printf('Specific fuel consumption is %3.3 f kg/I.H.P.hr',SFC)
```

Scilab code Exa 3.4 RATIO

```
clc
clear
//Input data
T=[100+273,473+273]//Temperatures at the beginning
    and at the end of adiabatic compression in K

g=1.4//Ratio of specific heats

//Calculations
an=(1-(T(1)/T(2)))*100//Air standard efficiency in
    percent
r=(T(2)/T(1))^(1/(g-1))//Compression ratio

//Output
printf('The compression ratio is %3.2 f \n Air
    standard efficiency is %i percent',r,an)
```

Scilab code Exa 3.5 AIR STANDARD EFFICIENCY

```
1 clc
2 clear
3 //Input data
4 T1=45+273//Temperature at the beginning of
      compression in K
  p1=1//Pressure at the beginning of compression in kg
     /\mathrm{cm}^2
  T2=325+273//Temperature at the end of compression in
  T3=1500+273 // Temperature at the end of constant
      volume heat addition in K
8 g=1.4//Ratio of specific heats
10 // Calculations
11 r=(T2/T1)^(1/(g-1))/Compression ratio
12 an=(1-(1/r)^{(g-1)})*100//Air standard efficiency in
      percent
13 p2=(p1*r^g)//Pressure at the end of compression in
     kg/cm^2
14 p3=(p2*(T3/T2))//Pressure at the end of constant
      volume heat addition in kg/cm<sup>2</sup>
15 p4=p3/p2//Pressure at the end of adiabatic expansion
       in kg/cm<sup>2</sup>
16 T4=T3/r^{(g-1)}/Temperature at the end of adiabatic
      expansion in K
17 t4=T4-273//Temperature at the end of adiabatic
      expansion in degree C
18
19 //Output
20 printf ('The air standard efficiency is %3.1f percent
       \n Temperature at the end of adiabatic expansion
       is %i degree C \n Pressure at the end of
```

Scilab code Exa 3.6 HEAT AND WORK

```
1 clc
2 clear
3 //Input data
4 T1=40+273//Temperature at the beginning of
      compression in K
5 p1=1//Pressure at the beginning of compression in kg
     /\mathrm{cm}^2
6 p2=15//Pressure at the end of adabatic compression
     in kg/cm<sup>2</sup>
  T3=2000+273//Maximum temperature during the cycle in
8 Cv=0.17//Specific heat at constant volume in kJ/kg.K
9 g=1.4//Ratio of specific heats
10
11 // Calculations
12 T2=T1*(p2/p1)^((g-1)/g)/Temperature at the end of
      adabatic compression in K
13 na=(1-(T1/T2))*100//Air standard efficiency in
      percent
14 q=(Cv*(T3-T2))//Heat added in kcal/kg of air
15 W=((na/100)*q)//Workdone per kg of air in kcal
16 W1 = (4.28*W) / Workdone per kg of air in kg.m
17 p3=(p2*(T3/T2))//Pressure at the end of constant
      volume heat addition in kg/cm<sup>2</sup>
18 p4=(p3*p1)/p2//Pressure at the end of adiabatic
      expansion in kg/cm<sup>2</sup>
19
20 //Output
21 printf('(a) The heat supplied is \%3.0 f kcal/kg of
      air \n (b) The workdone is %i kcal/kg of air \n (
     c) The pressure at the end of adiabatic expansion
```

Scilab code Exa 3.7 AIR STANDARD EFFICIENCY

Scilab code Exa 3.8 THERMAL EFFICIENCY

```
10 g=1.4//Ratio of specific heats
11
12 // Calculations
13 r1=(p2/p1)^(1/g)/Compression ratio
14 k=r1/r//Cutoff ratio
15 T2=(p2/p1)^{(g-1)/g}*T1//Temperature at the end of
      adiabatic compression in K
16 T3=T2*k//Temperature at the end of constant pressure
      heat addition in K
  T4=T3*(1/r)^{(g-1)}/Temperature at the end of
      adiabatic expansion in K
18 qa=(Cp*(T3-T2))//Heat added in kcal/kg of air
19 qre=(Cv*(T4-T1))//Heat rejected in kcal/kg of air
20 nt=((qa-qre)/qa)*100//Ideal thermal efficiency in
     percent
21
22 // Output
23 printf ('The ideal thermal efficiency is %3.1 f
     percent', nt)
```

Scilab code Exa 3.9 TEMPERATURE AND EFFICIENCY

```
10 g=1.41//Ratio of specific heats
11
12 // Calculations
13 T2=T1*r^{(g-1)}/Temperature at the end of adiabatic
      compression in K
14 s = (((T3/T2)-1)/(r-1))*100//Percentage of the stroke
      at which cut off occurs
15 r1=(r/(T3/T2))/Expansion ratio
16 T4=T3/(r1)^{(g-1)}/Temperature at the end of
      adiabatic expansion in K
17 qa=(Cp*(T3-T2))//Heat added in kcal/kg of air
18 qre=(Cv*(T4-T1))//Heat rejected in kcal/kg of air
19 nt=((qa-qre)/qa)*100//Air standard efficiency in
      percent
20
21 // Output
22 printf('(a) The percentage of stroke at which cut
      off takes place is \%3.2 \,\mathrm{f} percent \n (b) The
      temperature at the end of expansion stroke is \%3
      .0 f K \setminus n (c) The theoretical efficiency is \%3.0 f
      percent', s, T4, nt)
```

Scilab code Exa 3.10 PRESSURE AND HEAT

```
9 g=1.4//Ratio of specific heats
10
11 // Calculations
12 p2=p1*r^g//Pressure at the end of compression in kg/
     cm^2
13 s = (((r*(p4/p2)^{(1/g)})^{-1})/(r-1))*100//Percentage of
     stroke when the fuel is cut off in percent
14 T2=(T1*(p2/p1))/r//Temperature at the end of
      compression in K
  T3=(T2*r*(p4/p2)^(1/g))/Temperature at the end of
15
      adiabatic expansion in K
16 q = (Cp*(T3-T2)) // Heat supplied in kcal/kg
17
18 //Output
19 printf('(a) The maximum pressure attained during the
       cycle is \%3.1 f kg/cm^2 \n (b) The percentage of
     working stroke at which the heat supply to the
     working fluid ceases is \%3.2f percent \n (c) The
     heat received per kg of woring substance during
     the cycle is %3.0 f kcal/kg',p2,s,q)
```

Scilab code Exa 3.11 AIR STANDARD EFFICIENCY

```
1 clc
2 clear
3 d=0.25//Diameter of the cylinder in m
4 L=0.35//Stroke in m
5 Cv=1500//Clearance volume in c.c
6 s=5//cut off ratio takes place at 5 percent of stroke
7 a=1.4//Explosion ratio
8 g=1.4//Ratio of specific heats for air
9
10 //Calculations
11 Vs=(3.14/4)*d^2*L//Stroke volume in m^3
```

```
12  r=(Vs*10^6+Cv)/Cv//Compression ratio
13  k=(Cv+((s/100)*Vs*10^6))/Cv//Cut off ratio
14  na=(1-((1/(r^(g-1)))*((a*k^g-1)/((a-1)+a*g*(k-1)))))
      *100//Air standard efficiency in percent
15
16  //Output
17  printf('The air standard efficiency of the engine is %3.1 f percent',na)
```

Scilab code Exa 3.12 AIR STANDARD EFFICIENCY

```
1 clc
2 clear
3 //Input data
4 d=0.2//Diameter of the cylinder in m
5 L=0.4//Stroke in m
6 r=13.5//Compression ratio
7 a=1.42//Explosion ratio
8 s=5.1//Cut off occurs at 5.1 percent of the stroke
9 g=1.4//Ratio of specific heats for air
10
11 // Calculations
12 Vs = (3.14/4)*d^2*L*10^-6//Stroke volume in c.c
13 Vc=Vs/r//Clearance volume in c.c
14 k = (((s/100)*Vs)+Vc)/Vc//Cut off ratio
15 ASE=(1-((1/(r^(g-1)))*((a*k^g-1)/((a-1)+a*g*(k-1))))
     )*100//Air standard efficiency in percent
16
17 //Output
18 printf ('The air standard efficiency of the engine is
      %3.1f percent', ASE)
```

Scilab code Exa 3.13 TEMPERATURE AND EFFICIENCY

```
1 clc
2 clear
3 //Input data
4 = \frac{2}{3}, \frac{1}{3} /The dual cycle atkes two-thirds of its
      total heat supply at constant volume and one-
      third at constant pressure
5 r=13//Compression ratio
6 p3=43//Maximum pressure of the cycle in kg/cm<sup>2</sup>
7 p1=1//Pressure at intake in kg/cm<sup>2</sup>
8 T1=15+273//Intake temperature in K
9 Cp=0.24//Specific heat at constant pressure in kJ/kg
10 Cv=0.17//Specific heat at constant volume in kJ/kg.K
11 g=1.41//Ratio of specific heats
12
13 // Calculations
14 T2=T1*r^{(g-1)}/Temperature at the end of compression
15 p2=(p1*r^g)//Pressure at the end of compression in
     kg/cm^2
16 T3=T2*p3/p2//Temperature at the end of constant
      volume heat addition in K
17 q23=Cv*(T3-T2)//Heat added at constant volume in
      kcal/kg
18 q34=(1/2)*q23//Heat added at constant pressure in
     kcal/kg
19 T4=(q34/Cp)+T3//Temperature at the end of constant
      pressure heat supply in K
20 T5 = (T4*((p1*(T4/T3))/r)^(g-1))/Temperature at the
     end of expansion in K
21 \quad na = (1 - ((Cv*(T5-T1))/((Cv*(T3-T2)) + (Cp*(T4-T3)))))
      *100//Efficiency in percent
T = [T1-273, T2-273, T3-273, T4-273, T5-273] / Temperature
      at the five cardinal points in degree C
23
24 //Output
25 printf('(a) The temperature at the five cardinal
      points of the cycle are : \n point 1 is \%3.0f
```

degree C \n point 2 is %3.0f degree C \n point 3
is %3.0f degree C \n point 4 is %3.1f degree C \n
point 5 is %3.0f degree C \n\n (b) The ideal
thermal efficiency of the cycle is %3.1f percent'
,T(1),T(2),T(3),T(4),T(5),na)

Scilab code Exa 3.14 AIR STANDARD EFFICIENCY

```
1 clc
2 clear
3 //Input data
4 p1=1//Pressure at intake in kg/cm<sup>2</sup>
5 T1=100+273//Intake temperature in K
6 r=10//Compression ratio
7 p3=70//Maximum pressure of the cycle in kg/cm<sup>2</sup>
8 q=400//Amount of heat added in kcal/kg of air
9 Cp=0.24//Specific heat at constant pressure in kJ/kg
10 Cv=0.17//Specific heat at constant volume in kJ/kg.K
11 g=1.41//Ratio of specific heats
12
13 // Calculations
14 T2=(T1*r^{(g-1)})/Temperature at the end of
      compression in K
15 p2=(p1*r^g)//Pressure at the end of compression in
     kg/cm^2
16 T3=T2*(p3/p2)//Temperature at the end of constant
     volume heat addition in K
17 qv = (Cv * (T3 - T2)) / Heat added at constant volume in
     kcal/kg
18 qp=(q-qv)//Heat added at constant pressure in kcal/
19 T4=(qp/Cp)+T3//Temperature at the end of constant
      pressure heat supply in K
20 k=(T4/T3)//Cut off ratio
```

```
T5=T4/(r/k)^(g-1)//Temperature at the end of
        expansion in K

22 qv2=Cv*(T5-T1)//Heat added at constant volume in
        kcal/kg

23 W=q-qv2//Workdone in kcal/kg of air
24 na=(W/q)*100//Air standard efficiency in percent

25 //Output

27 printf('The temperature at the five cardinal points
        of the cycle are : \n point 1 is %3.0 f K \n point
        2 is %3.0 f K \n point 3 is %3.0 f K \n point 4 is
        %3.0 f K \n point 5 is %3.0 f K \n\n The air
        standard efficiency of the engine is %3.1 f
        percent', T1, T2, T3, T4, T5, na)
```

Scilab code Exa 3.15 WORK DONE AND EFFICIENCY

```
1 clc
2 clear
3 //Input data
4 r=12//Compression ratio
5 p1=0.955//Pressure at the start of compression in kg
      /\mathrm{cm}^2
6 T1=85+273//Temperature at the start of compression
7 p3=55//Maximum pressure of the cycle in kg/cm<sup>2</sup>
8 x=(1/30) // Constant pressure heat reception continues
      for 1/30 of the stroke
9 Cp=0.238//Specific heat at constant pressure in kJ/
     kg.K
10 Cv=0.17//Specific heat at constant volume in kJ/kg.K
11 g=1.4//Ratio of specific heats
12
13 // Calculations
14 T2=T1*r^{(g-1)}/Temperature at the end of compression
```

```
in K
15 p2=(p1*r^g)//Pressure at the end of compression in
     kg/cm^2
  T3=T2*(p3/p2)//Temperature at the end of constant
16
     volume heat addition in K
  T4=(T3*((p1+x*(r-1))/p1))//Temperature at the end of
17
      constant pressure heat supply in K
  T5=T4*((p1+x*(r-1))/r)^(g-1)/Temperature at the end
      of expansion in K
  qs = (Cv*(T3-T2)) + (Cp*(T4-T3)) / Heat supplied in kcal /
     kg of air
20 qre=(Cv*(T5-T1))//Heat rejected in kcal/kg of air
21 W=(qs-qre)//Workdone in kcal/kg of air
22 an=((qs-qre)/qs)*100//Air standard efficiency in
     percent
23
24 //Ouptut
25 printf ('The wordone per kg of air is \%3.2 f kcal \n
     The ideal thermal efficiency is %3.1f percent', W,
     an)
```

Scilab code Exa 3.16 TEMPERATURE AND EFFICIENCY

```
1 clc
2 clear
3 //Input data
4 p1=1//Pressure of air intake in kg/cm^2
5 T1=50+273//Temperature of air intake in K
6 v=(1/14)//Volume compresses by it adiabatically of its original volume
7 x=2//During the first stage, the pressure increases at constant volume to twice the pressure of the adiabate compression, and during the second stage the volume is increased twice the clearance volume at constant pressure
```

```
8 r = (1/v) // Compression ratio
9 Cp=0.237//Specific heat at constant pressure in kJ/
10 Cv=0.169//Specific heat at constant volume in kJ/kg.
11 g=1.4//Ratio of specific heats for air
12
13 // Calculations
14 T2=T1*r^{(g-1)}/Temperature at the end of compression
15 p2=(p1*r^g)//Pressure at the end of compression in
      kg/cm^2
  p3=x*p2//Pressure at the end of the heat addition at
       constant volume in kg/cm<sup>2</sup>
  T3=T2*(p3/p2)//Temperature at the end of constant
      volume heat addition in K
18 T4=(T3*x)//Temperature at the end of constant
      pressure heat supply in K
  T5=T4/(r/x)^{(g-1)}/Temperature at the end of
      expansion in K
20 qs=(Cv*(T3-T2))+(Cp*(T4-T3))/Heat supplied in kcal/
      kg of air
21 gre=(Cv*(T5-T1))//Heat rejected in kcal/kg of air
22 na=((qs-qre)/qs)*100//Air standard efficiency in
      percent
23
  T = [T1-273, T2-273, T3-273, T4-273, T5-273] / Temperature
      at the five key points in degree C
24
25 //Output
26 printf('(a) The temperature at the five key points
      of the cycle are : \n point 1 is \%3.0 \, f \, K = \%3.0 \, f
      degree C \n point 2 is \%3.0 \, \text{f K} = \%3.0 \, \text{f degree C} \setminus
      n point 3 is \%3.0 \, \text{f K} = \%3.0 \, \text{f degree C} \setminus \text{n point 4}
      is \%3.0 \, \text{f K} = \%3.0 \, \text{f degree C} \setminus \text{n point 5 is } \%3.0 \, \text{f K}
       = \%3.0 \, \text{f} degree C \n\n (b) The ideal thermal
      efficiency of the cycle is \%3.2f percent', T1, T(1)
      ,T2,T(2),T3,T(3),T4,T(4),T5,T(5),na)
```

Scilab code Exa 3.18 MEAN EFFECTIVE PRESSURE

```
1 clc
2 clear
3 //Input data
4 n=6//Six cylinder engine
5 r=5//Compression ratio
6 Vc=110//Clearance volume in c.c
7 a=0.66//Efficiency ratio referred to the air
      standard cycle
8 N=2400/Speed in r.p.m
9 m=9.9//Mass of petrol in kg
10 CV=10600//Calorific value of fuel in kcal/kg
11 g=1.4//Ratio of specific heats
12
13 // Calculations
14 Vs=(r*Vc-Vc)//Swept Volume in c.c
15 na=(1-(1/r)^{(g-1)})*100/Air standard efficiency in
      percent
16 nt = (na/100)*a//Thermal efficiency
17 IHP=(nt*CV*m*427)/(4500*60)//Indicated Horse Power
     in h.p
18 pm = (((IHP/n)*4500*100*2)/(Vs*N))/Average indicated
     mean effective pressure in kg/cm<sup>2</sup>
19
20 //Output
21 printf ('The average indicated mean effective
      pressure in each cylinder is \%3.3 f kg/cm^2',pm)
```

Scilab code Exa 3.19 CYLINDER SIZES

1 clc

```
2 clear
3 //Input data
4 n=4//Four cylinder engine
5 BHP=40//Brake horse power in h.p
6 N=3000/Speed in r.p.m
7 nm=70//Mechanical efficiency in percent
8 pm=13.5//Indicated mean effective pressure in kg/cm
9 //Bore is equal to stroke
10
11 // Calculations
12 // case(i)
13 d1=((BHP*100*4500*n*2)/(n*(nm/100)*pm*N*3.14))^(1/3)
     //Cylinder bore or stroke length in cm
14
15 // Case ( i i )
16 d2=((BHP*100*4500*n)/(n*(nm/100)*pm*N*3.14))^(1/3)
     *10//Cylinder bore or stroke length in cm
17
18 //Output
19 printf ('The cylinder sizes for a bore equal to
     stroke of a four cylinder in case of \n (i) Four
     stroke engine is %3.1f cm \n (ii)Two stroke
     engine is %3.0 f mm', d1, d2)
```

Scilab code Exa 3.20 THERMAL EFFICIENCY

```
1 clc
2 clear
3 //Input data
4 T=[50+273,345+273]//Temperatures at the beginning and end of compression in K
5 g=1.4//ratio of specific heats
6 IHP=25//Indicated horse power in h.p
7 m=5.44//Mass of fuel consumed per hour in kg
```

```
8 CV=10300//Calorific value in kcal/kg
10 // Calculations
11 na=(1-(T(1)/T(2)))*100//Air standard efficiency in
      percent
12 r = (T(2)/T(1))^{(1/(g-1))}/Compression ratio
13 qIHP = (IHP * 4500) / 427 / / Heat equivalent of I.H.P in
      kcal/min
14 q=(m*CV)/60//Heat supplied per minute in kcal/min
15 Ith=(qIHP/q)*100//Indicated thermal efficiency in
      percent
16 nr=(Ith/na)*100//Efficiency ratio
17
18 //Output
19 printf ('The air standard efficiency is %3.1f percent
       \n The compression ratio is \%3.2 f \n Indicated
      thermal efficiency is %3.1f percent \n Efficiency
       ratio is %3.1f percent', na, r, Ith, nr)
```

Scilab code Exa 3.21 COMPRESSION RATIO

```
12 v=(p(2)/p(1))^(1/n)//Ratio of specific volumes
13 r=((pe(2)/100)*v-(pe(1)/100))/((pe(2)/100)-((pe(1)/100)*v))//Compression ratio
14 na=(1-(1/r)^(g-1))*100//Air standard efficiency in percent
15 ith=(rn*na)/100//Indicated thermal efficiency in percent
16 q=(4500*60)/(427*(ith/100))//Heat supplied in kcal/i.h.p.hr
17 Sc=(q/CV)//Specific consumption in kg/i.h.p.hr
18 //Output
19 printf('Compression ratio is %3.2 f \n Specific consumption is %3.3 f kg/i.h.p.hr',r,Sc)
```

Scilab code Exa 3.22 EFFICIENCY

```
1 clc
2 clear
3 //Input data
4 n=4//Four cylinder four stroke
5 d=7.5//Bore in cm
6 L=8.75//Stroke in cm
7 r=6//Compression ratio
8 n1=55//Efficiency in percent
9 g=1.4//ratio of specific heats
10 N=2400/Speed in r.p.m
11 pm=7//Brake mean effective pressure in kg/cm<sup>2</sup>
12 m=9//Mass of fuel per hour in kg
13 CV=10500//Calorific Value in kcal/kg
14
15 // Calculations
16 an=(1-(1/r)^{(g-1)})*100//Air standard efficiency in
      percent
17 In=(an*n1)/100//Indicated thermal efficiency in
```

Scilab code Exa 3.26 PERCENTAGE CHANGE IN EFFICIENCY

```
1 clc
2 clear
3 //Input data
4 r=7//Compression ratio
5 v=1//Specific heat at constant volume increases by 1
      percent
6 g=1.4//Ratio of specific heats
8 // Calculations
9 e=(1-(1/r^{(g-1))})/Air standard efficiency
10 dee = -(((1-e)*(g-1)*log(r)*(v/100))/e)*100//Change in
      efficiency to the original efficiency
11 x=-(dee)//For Output purpose
12
13 //Output
14 printf ('Percentage change is efficiency is \%3.2 f
     percent i.e., a decrease of \%3.2f percent', dee,x)
```

Scilab code Exa 3.30 MAXIMUM HORSE POWER

```
1 clc
2 clear
3 //Input data
4 N=210//Speed in r.p.m
5 d=0.3//Diameter of the piston in m
6 L=0.4//Stroke in m
7 v=2.5 // Clearance volume is 2.5\% of the swept volume.
      But in textbook it is given wrong as 25\%
8 CO=19.7//Percentage of CO gas
9 H2=28.8//Percentage of H2 gas
10 CO2=14.4//Percentage of CO2 gas
11 N2=37.1//Percentage of N2 gas
12 x=0.875//Total mixture at N.T.P admitted per suction
      stroke is 0.875 of the total volume behind the
     piston at the end of the stroke
13 tn=35//Thermal efficiency in percent
14 CVH2=13200//Calorific value of H2 per kg in kcal
15 CVC=2540 // Calorific value of carbon burning from CO
     to CO2 in kcal/kg
16 de=1.293//Density of air in kg/m^3
17 mC=12//Molecular weight of carbon
18 mO2=32//Molecular weight of O2
19 mH2=2//Molecular weight of H2
20 mCO=28//Molecular weight of CO
21
22 // Calculations
23 a = ((100/21)*((CO2/100)+((CO/2)/100)))/Air per cu.m
     of gas in cu.m
24 Vm=(a+1)//Volume of mixture per cu.m of gas in cu.m
25 Vs = ((3.14/4)*d^2*L) / Swept volume in cu.m
26 Vc=(Vs*v)/100//Clearance volume in cu.m
27 V=Vc+Vs//Total volume in cu.m
```

```
28 VC=V*x//Volume of charge admitted per stroke in cu.m
29 VM=VC*(N/2)//Charge volume per minute in cu.m
30 VG=(VM/Vm)//cu.m of gas per minute
31 vH2=(VG*(H2/100))/Volume of H2 per minute in cu.m
32 vCO = (VG * (CO/100)) / Volume of CO per minute in cu.m
33 CVH2cum = (mH2*CVH2)/(vH2*1000)//Calorific value of H2
       per cu.m in kcal
34 CVCO = (CVC * (2 * mC) / (2 * mCO)) / Calorific value of CO per
      kg in kcal
  CVCOcum=(mCO*CVCO)/(vH2*1000)//Calorific value of CO
       per cu.m in kcal
36 qH2=(16.09*CVH2cum)//Heat in charge due to H2 in
      kcal
37 qCO=(11*CVCOcum)//Heat in charge due to CO in kcal
38 qt=(qH2+qC0)//Heat supplied per minute in kcal
39 qu=(qt*(tn/100))/Heat utilised in kcal
40 hp=(qu*427)/4500//H.P. developed
41
42 // Output
43 printf ('Maximum horse power that can be developed is
      \%3.1 f H.P', hp)
```

Scilab code Exa 3.31 AIR FUEL RATIO

```
1 clc
2 clear
3 //Input data
4 vCH4=65//Composition by volume of CH4
5 vH2=2//Composition by volume of H2
6 vN2=2//Composition by volume of N2
7 vCO2=31//Composition by volume of CO2
8 O2=5.3//Composition of O2 in dry exhaust gases when analysed in orsat apparatus
9 N2=83//Composition of N2 in dry exhaust gases when analysed in orsat apparatus
```

```
10 CO=0.3//Composition of CO in dry exhaust gases when
      analysed in orsat apparatus
11 CO2=11.4//Composition of CO2 in dry exhaust gases
     when analysed in orsat apparatus
12 an=79//Air contains 79% by volume of nitrogen
13
14 // Calculations
15 a=(100/(100-an))*(((vCH4/100)*2)+((vN2/100)*(1/2)))
     //Total air required for complete combustion of 1
      cu.m of gas in cu.m
16 xCO=(CO/2)/O2 required to burn the CO in cu.m.
17 \text{ xCO2=CO}/\text{CO2} formed in cu.m
18 t02=02-xC0//Total O2 in cu.m
19 tN2=N2//Total N2 in cu.m
20 tCO2 = CO2 + xCO2 / Total CO2 in cu.m
21 T=t02+tN2+tC02//Total mixture in cu.m
22 pCO2=(tCO2*100)/T//Percentage of CO2 in percent
23 mm=(a*100)//Minimum air supply required for complete
      combustion of 100 cu.m of the gas in cu.m
24 an2=(an/100)*mm//N2 for this air in cu.m
25 tn2=(an2+vN2)//Total N2 in cu.m
26 v=(((vCH4+vCO2)*100)/pCO2)-(vCH4+vCO2+tn2)//Increase
      in air supply for reduction in percentage of CO2
      in cu.m
27 pea=(v*100)/mm//Percentage of excess air. In
     textbook it is given wrong as 26.7 percent
28
29 //Output
30 printf('(a) the air-fuel ratio by volume to give
     complete combustion is %3.3 f \n (b) the
     percentage of excess air actually used in the
      test is %3.1f percent',a,pea)
```

Scilab code Exa 3.32 AIR GAS RATIO

```
1 clc
2 clear
3 //Input data
4 Vs=9.45//Swept volume in litres
5 Vc=2.32//Clearance volume in litres
6 m=4.25//Consumption of gas per hour in cu.m
7 N=165/Speed in r.p.m
8 bhp=5.62//Brake horse power in h.p.
9 nm=73.4//Mechanical efficiency in percent
10 CV=3500//Calorific value in kcal per cubic meter
11 vn=0.87//Volumetric efficiency
12 g=1.4//Ratio of specific heats
13
14 // Calculations
15 tV = (Vs + Vc) *1000 / Total volume in c.c
16 rc=(tV/Vc)//Compression ratio
17 na=(1-(1/rc^{(g-1)}))*100//Air at and ard efficiency in
      percent
18 W=(bhp*4500)/427//Workdone per minute in kcal
19 Iw=(W/(nm/100))//Indicated work in kcal/min
20 q=(m/60)*CV//Heat supplied in kcal/min
21 ith=(Iw/q)*100//Indicated thermal efficiency in
      percent
22 rn=(ith/na)*100//Relative efficiency in percent
23 Vm=(Vs*1000)*vn//Volume of mixture taken in per
      stroke in c.c
24 \text{ Vg}=(m*2*10^6)/(60*N)//\text{Volume of gas taken in per}
      stroke in c.c
25 Va=(Vm-Vg)//Volume of air taken in per stroke in c.c
26 agr = (Va/Vg) // Air gas ratio
27 CVc=(CV/(agr+1))//Calorific value of charge in kcal
28
29 // Output
30 printf('Ratio of air to gas used is \%3.2 f \n
      Calorific value of 1 cu.m of the mixture in the
      cylinder is %3.1f kcal', agr, CVc)
```

Scilab code Exa 3.33 FRICTION HORSE POWER

```
1 clc
2 clear
3 //Input data
4 d=18//Bore in cm
5 1=37.5//Stroke in cm
6 \text{ N=} 220 // \text{Speed in r.p.m}
7 //Mean effective pressure in kg/cm<sup>2</sup>
8 // Firing
9 pp=5.9//Positive loop
10 pn=0.248//Negative loop
11 // Missing
12 \text{ nn=0.432}//\text{Negative loop}
13 bhp=8.62//Brake horse power in h.p
14 ex=100//Explosions per minute
15 vg=0.101//Gas used in cu.m per minute
16
17 // Calculations
18 tc=(N/2) //The number of cycles
19 nw=ex//Number of working cycles
20 nm=(tc-nw)//Number of missing cycles
21 ihp=((1/100)*(3.14/4)*(d^2/4500))*((pp-pn)*(100-nn))
      //Net I.H.P in h.p
22 fhp=(ihp-bhp)//Friction horse power in h.p.
23 W=((pp-pn)*(3.14/4)*(d^2*(1/100)))/Workdone per
      firing done in kg.m
24 Wp = (nn*(3.14/4)*d^2*(1/100)) / Workdone per pumping
      stroke in kg.m
25 n=((fhp*4500)+(Wp*tc))/(W+Wp)/Number of strokes)
26 gf=(vg/nw)//Gas per firing stroke in cu.m
27 gl=(n*gf)//Gas per minute at no load in cu.m
28
29 //Output
```

30 printf('Friction horse power of the engine is $\%3.2\,\mathrm{f}$ \n Gas consumption at no load is $\%3.3\,\mathrm{f}$ cu.m/min', fhp,gl)

Chapter 4

FUEL AIR CYCLES AND REAL CYCLES

Scilab code Exa 4.1 WORK DONE

```
1 clc
2 clear
3 //Input data
4 d=20/Bore in cm
5 1=38//Stroke in cm
6 Vc=900//Clearance volume in c.c
7 p1=1//Pressure at the start of compression stroke in
      kg/cm^2
8 T1=90+273//Temperature at the start of compression
     stroke in K
9 x=0.75//Piston travelled 0.75 of the compression
     stroke
10 n=1.32//Compression curve index
11 wa=0.0125//Weight of air in kg
12
13 // Calculations
14 Vs = (3.14/4)*d^2*1//Swept volume in c.c
15 V1 = (Vs + Vc) / Volume in c.c
16 V2=(1-x)*Vs+Vc//Volume in c.c
```

```
17 p2=p1*(V1/V2)^n//Pressure in kg/cm^2
18 T2=(T1*(p2/p1)*(V2/V1))//Temperature in K
19 W=((p1*V1-p2*V2)/(n-1))*10^-2//Workdone in kg.m
20 dI=wa*0.17*(T2-T1)//Change in internal energy in kcal
21 q=(dI+(W/427))//Heat in kcal
22
23 //Output
24 printf('When the cylinder has travelled %3.2 f of the compression stroke, \n The volume is %3.0 f c.c \n The pressure is %3.2 f kg/cm^2 \n Temperature is %3.0 f K \n\n The workdone on the gas is %3.2 f kg .m \n\n Change in internal energy between the two points is %3.3 f kcal',x,V2,p2,T2,W,dI)
```

Chapter 5

VARIABLE SPECIFIC HEAT

Scilab code Exa 5.2 CHANGE IN AIR STANDARD EFFICIENCY

```
1 clc
2 clear
3 //Input data
4 r=8//Compression ratio
5 n=1.41//Adiabatic index of the medium
6 cv=0.17//Mean Specific heat at constant volume in
     kcal/kg/degree C
7 x=2//Percentage with which spcific heat at constant
     volume increases
8 R=29.3//Characteristic gas constant in mkg/kg/degree
9 J=427//Mechanical equivalent of heat in kg.m/kcal
10
11 // Calculations
12 e=(1-(1/r^{(n-1)}))/Air standard efficiency
     neglecting the variation in specific heat
13 debye=((x/100)*((1-e)/e)*(R/(J*cv))*log(r))*100//
     Ratio of de and e in percent
14
15 // Output
16 printf ('The change in air standard efficiency of the
```

Scilab code Exa 5.3 WORK DONE

```
clc
clear
//Input data
//Cv=0.125+0.000005T where Cv is Specific heat at
    constant volume and T is the temperature in K
R=28.9//Characteristic gas constant in mkg/kg/degree
    C
T=[100+273,50+273]//Temperature in K
J=427//Mechanical equivalent of heat in kg.m/kcal
//Calculations
function y=f(x),y=0.125+(0.00005*x),endfunction
I=J*intg(303,373,f)
//Output
printf('The work done is %i m.kg/kg of gas',I)
```

Scilab code Exa 5.4 PERCENTAGE OF STROKE

```
7 p1=1//Pressure at the beginning of compression in kg
     /\mathrm{cm}^2
8 T1=153+273//Temperature at the beginning of
     compression in K
9 CV=10000//Heating value of fuel in kcal/kg
10 n=1.35//Adiabatic constant
11 R=29//Characteristic gas constant in mkg/kg.K
12 J=427 // Mechanical equivalent of heat in kg.m/kcal
13
14 // Calculations
15 T2=(T1*r^(n-1))/Temperature at the end of
     compression in K
16 a = (cv(2)/2) / For solving T3
17 b=cv(1)+(R/J)//For solving T3
18 c=(-T2*cv(1))-((cv(2)/2)*T2^2)-((R/J)*T2)-(CV/(af+1)
     )//Foe solving T3
19 T3=(-b+sqrt(b^2-(4*a*c)))/(2*a)//Soving for T3 in K
20 pc = (((T3/T2)-1)/(r-1))*100//Percentage cut off
21
22 //Output
23 printf ('The percentage of stroke at which the
      constant pressure combustion stops is %i percent'
      ,pc)
```

Scilab code Exa 5.5 PERCENTAGE OF STROKE

```
8 T2=800+273//Temperature at the end of compression in
9 R=29//Characteristic gas constant in mkg/kg/degree C
10 J=427 // Mechanical equivalent of heat in kg.m/kcal
11
12 // Calculations
13 CVm=(CV/(af+1))//Calorific value of mixture in kcal/
      kg
14 cpv=(R/J)//Difference in mean specific heats in kcal
     /kg mol.K
15 a = (cv(2)/2) // For solving T3
16 b = cpv + cv(1) // For solving T3
17 c=(-T2*(cpv+cv(1)))-((cv(2)/2)*T2^2)-CVm//Foe
      solving T3
18 T3=(-b+sqrt(b^2-(4*a*c)))/(2*a)//Soving for T3 in K
19 s=((T3/T2)/(r-1))*100//Percentage of the stroke
20
21 // Output
22 printf ('The percentage of the stroke at which the
      combustion will be complete is \%3.2 f percent', s)
```

Scilab code Exa 5.6 CHANGE IN ENTHALPY

```
1 clc
2 clear
3 //Input data
4 T=[500,2000]//Change in temperature in K
5 x=[11.515,-172,1530]//Cp=11.515-172/sqrt(T)+1530/T
    in kcal/kg mole.K
6 m02=32//Molecular weight of oxygen
7
8 //Calculations
9 function y=f(T),y=(x(1)+(x(2)/sqrt(T))+(x(3)/T)),
    endfunction
10 I=-intg(T(2),T(1),f)//Integration
```

```
11 dh=(I/m02)//Change in enthalpy in kcal/kg
12
13 //Output
14 printf('The change in enthalpy is %3.1 f kcal/kg',dh)
```

Scilab code Exa 5.7 TEMPERATURE AND PRESSURE

```
1 clc
2 clear
3 //Input data
4 r=14//Compression ratio
5 \text{ s=}5//\text{Fuel injection stops} at 5\% \text{ stroke} after inner
      head centre
6 pm=50//Maximum pressure in kg/cm<sup>2</sup>
  p4=1//Pressure at the end of suction stroke in kg/cm
  T4=90+273//Temperature at the end of suction stroke
  R=29.3//Characteristic gas constant in mkg/kg/degree
      \mathbf{C}
10 cv = [0.171, 0.00003] / Cv = 0.171 + 0.00003T where Cv is
      Specific heat at constant volume and T is the
      temperature in K
11 J=427 // Mechanical equivalent of heat in kg.m/kcal
12
13 // Calculations
14 a=(R/J)+cv(1)/a value in kcal/kg.mole.K
15 g=(a+cv(2)*T4)/(cv(1)+cv(2)*T4)/Adiabatic index of
      compression
16 z=1.3//Rounding off 'z' value to one decimal.
17 T5=(T4*r^(z-1))/Temperature in K
18 p5=(p4*r^g)/Pressure in kg/cm^2
19 T1=T5*(pm/p5)//Tmperature in K
20 T2=(T1*(1+(s/100)*(r-1)))//Temperature in K
21 T3=(T2*((1+(s/100)*(r-1))/r)^(g-1))//Temperature in
```

```
K
22 p3=(p4*(T3/T4))/Pressure in kg/cm^2
23 function y=f1(T), y=cv(1)+(cv(2)*T), endfunction
24 I1=intg(T5,T1,f1)
25 function y=f2(T), y=(a+(cv(2)*T)), endfunction
26 I2=intg(T1,T2,f2)//I2 answer is given wrong in the
      textbook
27 qs=(I1+I2)//Heat supplied per kg of air in kcal/kg
28 function y=f3(T), y=a+(cv(2)*T), endfunction
29 qre=intg(T4,T3,f3)//Heat required per kg of air in
      kcal/kg
30 nth=((qs-qre)/qs)*100//Thermal efficiency in percent
31
32 // Output
33 printf('The tempertautes and pressures at salient
       points of the cycle are : \n T1 = \%3.0 \, f \, K \, p1 =
      \%3.1 \text{ f kg/cm}^2 \text{ n T2} = \%3.0 \text{ f K p2} = \%3.1 \text{ f kg/cm}^2
       n T3 = \%3.0 f K p3 = \%3.1 f kg/cm^2 n T4 = \%3.0
      f K p4 = \%3.1 \, \text{f} \, \text{kg/cm}^2 \, \text{n} \, \text{T5} = \%3.0 \, \text{f} \, \text{K} \, \text{p5} = \%3
      .1f kg/cm<sup>2</sup> \n\n Heat supplied per kg of air is
      \%3.1 \, \text{f kcal/kg } \ \text{n} \ \text{The thermal efficiency of the}
      cycle is %3.1 f percent', T1, pm, T2, pm, T3, p3, T4, p4,
      T5, p5, qs, nth)
34 //Textbook answers are given wrong
```

Scilab code Exa 5.8 TEMPERATURE AND HEAT

```
compression in K
 8 p3=50//Maximum pressure in kg/cm<sup>2</sup>
 9 R=29.3//Characteristic gas constant in mkg/kg/degree
10 cv = [0.171, 0.00003] / Cv = 0.171 + 0.00003T where Cv is
                Specific heat at constant volume and T is the
               temperature in K
11 g1=1.4//Ratio of specific heats
12 J=427 // Mechanical equivalent of heat in kg.m/kcal
13
14 // Calculations
15 T2x=(T1*r^(g1-1))//Temperature in K
16 function y=f1(T), y=cv(1)+(cv(2)*T), endfunction
17 I1=intg(T1, T2x, f1)
18 Cv = (1/(T2x-T1))*I1//Mean value of Cv in kJ/kg.K
19 Cp = (Cv + (R/J)) / Mean value of Cp in kJ/kg.K
20 g=1.35//(Cp/Cv) value and rounded off to 2 decimal
               places for calculation purpose. Ratio of specific
                  heats
21 T2=(T1*r^(g-1))/Temperature in K
22 I2=intg(T1,T2,f1)
23 CV = (1/(T2-T1))*I2//Maen value of Cv in kJ/kg.K
24 CP = (Cv + (R/J)) / Mean value of Cp in kJ/kg.K
25 \text{ g2=1.36//(Cp/Cv)} value and rounded off to 2 decimal
                places for calculation purpose. Ratio of specific
               heats
26 T2a=(T1*r^(g2-1))//Temperature in K
27 p2=(p1*r*(T2a/T1))/Pressure in kg/cm^2
28 T3=(T2a*(p3/p2))//Temperature in K
29 T4 = (((r-1)*(c/100))+1)*T3//Temperature in K
30 g3=1.3//Assuming gamma as 1.3 for process 4-5
31 T5=(T4/(r/(((r-1)*(c/100))+1))^(g3-1))/Temperature
               in K
32 \text{ cV} = \text{cv}(1) + (\text{cv}(2)/2) * (\text{T5} + \text{T4}) / \text{Mean value of Cv in kJ} / 
33 cP=cV+(R/J)/Mean value of Cp in kJ/kg.K
34 g4=(cP/cV)//Ratio of specific heats
35 T5a=(T4/(r/(((r-1)*(c/100))+1))^(g4-1))
```

```
36    I3=intg(T2a,T3,f1)
37    function y=f2(T),y=cv(1)+(R/J)+(cv(2)*T),endfunction
38    I4=intg(T3,T4,f2)//Textbook answer is wrong
39    q=I3+I4//Heat supplied per kg of working substance
        in kcal/kg
40
41    //Output
42    printf('(a) Temperatures at all the points of the
        cycle are: \n T1 = %i K \n T2 = %3.0 f K \n T3 =
        %3.0 f K \n T4 = %3.0 f K \n T5 = %i K \n\n (b)
        heat supplied per kg of the working substance is
        %3.1 f kcal/kg',T1,T2a,T3,T4,T5a,q)
43    //Textbook answer is wrong
```

Scilab code Exa 5.10 PERCENTAGE CHANGE IN EFFICIENCY

```
1 clc
2 clear
3 //Input data
4 r=20//Compression ratio
5 c=5/Cut off at 5\%
6 dc=1//Specific heat at constant volume increases by
     1%
7 Cv=0.171//pecific heat at constant volume in kJ/kg.K
8 R=29.3//Characteristic gas constant in mkg/kg/degree
      \mathbf{C}
9 k=1.95/k can be obtained from relation de/e=-dcv/cv
     *(1-e/e)*(g-1)*((1/g)+ln(r)-(k^g*lnk)/(k^g-1))
10 J=427 // Mechanical equivalent of heat in kg.m/kcal
11
12 // Calculations
13 g=(R/(J*Cv))+1//Ratio of specific heats
14 e=(1-((1/g)*(1/r^(g-1))*((k^g-1)/(k-1))))/Air
     standard efficiency of the cycle
15 dee = ((-(dc/100)*((1-e)/e)*(g-1)*((1/g)+log(r)-((k^g*
```

```
log(k))/(k^g-1))))*100)//Change in efficiency due
to 1% change in cv

16
17 //Output
18 printf('Percentage change in air standard efficiency
is %3.3 f percent \n This indicates that there is
a decrease in efficiency', dee)
```

Chapter 6

COMBUSTION CHARTS

Scilab code Exa 6.1 EFFICIENCY

```
1 clc
2 clear
3 //Input data
4 r=7.5//Compression ratio
5 //Data from combustion chart
6 p=[1,15.1,26.95,1.95] // Pressure of air fuel mixture
      in kg/cm<sup>2</sup>
7 T=[60,460,1150,435]//Temperature of air fuel mixture
       in K
8 V = [16.98, 2.264, 2.264, 16.98] / Volume in m<sup>3</sup>/kg
9 U=[17,78.8,212,80] //Internal energy in kcal/kg
10 S=[0.07,0.07,0.22,0.22]//Entropy in kcal/kg.degree C
11 g=1.4//Ratio of specific heats
12
13 // Calculations
14 \quad n = (((U(3) - U(4)) - (U(2) - U(1))) / (U(3) - U(2))) *100 / /
      Thermal efficiency in percent
15 na=(1-(1/r)^{(g-1)})*100//Air standard efficiency in
      percent
16
17 // Output
```

18 printf('Thermal efficiency is %3.1 f percent \n Air standard efficiency is %3.1 f percent',n,na)

Scilab code Exa 6.2 EFFICIENCY

```
1 clc
2 clear
3 //Input data
4 //Data from combustion chart
5 p=[1,33,33,1]//Pressure of air fuel mixture in kg/cm
6 T=[65,600,1450,725] // Temperature of air fuel mixture
       in K
7 V = [16, 1.23, 3.45, 16] / Volume in m^3/kg
8 U=[11.8,110,295,140]//Internal energy in kcal/kg
9 H=[22.7,150,395,225]//Enthalpy in kcal/kg
10 S = [0.068, 0.068, 0.264, 0.264] / Entropy in kcal/kg.
     degree C
11
12 // Calculations
13 r=(V(1)/V(2))/Compression ratio
14 q=(H(3)-H(2))/Heat supplied in kcal/kg
15 qre=(U(4)-U(1))//Heat rejected in kcal/kg
16 nt=((q-qre)/q)*100//Thermal efficiency in percent
17
18 //Output
19 printf('(a) Compression ratio is %3.0 f \n (b) Heat
     supplied to the cycle is \%3.0 f kcal/kg \n (c)
     Heat rejected by the cycle is \%3.2f kcal/kg \n (d)
       Thermal efficiency is %3.2f percent',r,q,qre,nt
     )
```

Scilab code Exa 6.3 EFFICIENCY

```
1 clc
2 clear
3 //Input data
4 //Data from combustion chart
5 p=[1,51.5,77.25,77.25,3.75]//Pressure of air fuel
      mixture in kg/cm<sup>2</sup>
6 T=[16,1,1,1.5,16] // Temperature of air fuel mixture
      in K
7 V = [65,745,1400,2200,1030] / Volume in <math>m^3/kg
8 U=[14.7,135,275,475,197]//Internal energy in kcal/kg
9 H=[21.9,85,372,625,280]/Enthalpy in kcal/kg
10 S = [0.068, 0.068, 0.19, 0.32, 0.32] // Entropy in kcal/kg.
      degree C
11
12 // Calculations
13 nth = (((U(3)-U(2))+(H(4)-H(3))-(U(5)-U(1)))/((U(3)-U(3)))
      (2) + (H(4) - H(3))) *100 / Thermal efficiency in
      percent
14
15 //Output
16 printf('Thermal efficiency is %3.2f percent',nth)
```

Chapter 8

COMBUSTION

Scilab code Exa 8.1 COMPOSITION

```
1 clc
2 clear
3 //Input data
4 C=88.6//Composition of C in percent
5 H2=11.4//Composition of H2 in percent
6
7 // Calculations
8 w1=(C/100)//Weight per kg of fuel of C in kg
9 w2=(H2/100)//Weight per kg of fuel of H2 in kg
10 O1=(8/3)//Oxygen required per kg of constituent for
     C in kg
11 O2=8//Oxygen required per kg of constituent for H2
12 O11=(w1*O1)//Oxygen required per kg of fuel for C in
13 O22=(w2*O2)//Oxygen required per kg of fuel for H2
14 T=(011+022)//Total Oxygen required per kg of fuel in
15 P1=(w1+011)//Composition of CO2 in kg
16 P2=(w2+022)//Composition of H2O in kg
```

```
17 w=(T*(100/23))//Weight of air required in kg per kg
      of fuel
18
19 //Output
20 printf('The weight of air required for complete
      combustion of liquid fuel is %3.2 f kg per kg of
      fuel \n \n The composition of CO2 is %3.3 f kg \n
      The composition of H2O is %3.3 f kg',w,P1,P2)
```

Scilab code Exa 8.2 WEIGHT OF AIR

```
1 clc
2 clear
3 //Input data
4 C=12//Molecular weight of carbon
5 H2=2//Molecular weight of H2
6 02=32//Molecular weight of O2
8 // Calculations
9 C7H16=(7*C+8*H2)//Molecular weight of C7H16
10 02x=(11*02) // Molecular weight of 2202
11 wt = (02x/C7H16)*(100/23.2)//Weight of air in kg per
     kg of fuel
12 //Now in actual experiment, we have
13 //1 [C7H16] +x [O2] +... [N2] = a [CO2] +8[H2O] +a [O2]
      + \dots [N2]
14 //This is the new equation written in volumes. The
     volumes of CO2 and O2 being equal, with no CO
     present, and the usual assumption that all the
     hydrogen is burnt to H2O
15 //Now, if all the carbon is burnt, we must have 7
     mols of CO2
16 x=(7+4+7)/Total number of mols from CO2, H2O and O2
     terms respectively
17 W=((x*02)/100)*(100/23.2)/Weight of air in kg per
```

```
kg of fuel

18
19 //Output
20 printf('The weight of air is %3.1f kg per kg of fuel which would just suffice for theoretically complete combustion \n The ratio of air to fuel by weight as actually supplied is %3.1f kg of air per kg of fuel', wt, W)
```

Scilab code Exa 8.3 AIR AND VOLUME

```
1 clc
2 clear
3 //Input data
4 H=15//Percentage of Hydrogen by volume
5 CO=25//Percentage of carbon monoxide by volume
6 CH4=4//Percentage of methane by volume
7 CO2=4//Percentage of carbon dioxide by volume
8 O2=2//Percentage of oxygen by volume. In textbook it
       is given wrong as 25
9 N2=50//Percentage of nitrogen by volume
10
11 // Calculations
12 021=(H/100)*(1/2)//Volume of oxygen required in m<sup>3</sup>
      by 15 percent of H2
13 022=(C0/100)*(1/2)/Volume of oxygen required in m<sup>3</sup>
       by 25 percent of CO
14 CO21 = (CO/100) *1//Volume of CO2 produced in m<sup>3</sup> by 25
       percent of CO
15 O23=(CH4/100)*2//Volume of oxygen required in m<sup>3</sup> by
       4 percent of CH4
16 CO22 = (CH4/100)*1/Volume of CO2 produced in m^3 by 4
       percent of CH4
17 H201=(CO/100)*2//Volume of H2O produced in m<sup>3</sup> by 4
      percent of CH4
```

```
18 T02 = (021 + 022 + 023 - (02/100)) / Total vol. of oxygen in
      m^3
19 wa=(T02*(100/21))//Theoretical volume of air
      required in m<sup>3</sup>
20 vN2=(wa*(79/100))/Volume of N2 present in air in m
21 TvN2=(vN2+(N2/100))//Total volume of N2 after
      combustion of 1 m<sup>3</sup> of fuel in m<sup>3</sup>
22 \times CO2 = (CO21 + CO22) / / CO2  produced due to combustion of
      fuel in m<sup>3</sup>
23 TCO2 = (xCO2 + (CO2/100)) / Total volume of CO2 in the
      flue gas in m<sup>3</sup>
24
25 //Output
26 printf ('The air required for complete combustion of
      one m<sup>3</sup> of the fuel is %3.3f cu.m \n The dry flue
       gas contains %3.3 f cu.m volume of N2 and %3.2 f
      cu.m volume of CO2', wa, TvN2, TCO2)
```

Scilab code Exa 8.4 WEIGHT OF AIR

```
1 clc
2 clear
3 //Input data
4 C=88.1//Composition of C in percent
5 H2=10.7//Composition of H2 in percent
6 O2=1.2//Composition of O2 in percent
7
8 //Calculations
9 w1=(C/100)//Weight per kg of fuel of C in kg
10 w2=(H2/100)//Weight per kg of fuel of H2 in kg
11 w3=(O2/100)//Weight per kg of fuel of O2 in kg
12 O1=(8/3)//Oxygen required per kg of constituent for C in kg
13 O2=8//Oxygen required per kg of constituent for H2
```

```
in kg
14 O11=(w1*O1)//Oxygen required per kg of fuel for C in
15 O22=(w2*O2)//Oxygen required per kg of fuel for H2
     in kg
16 T=(011+022-w3)//Total Oxygen required per kg of fuel
      in kg
17 P1=(w1+011)//Composition of CO2 in kg
18 P2=(w2+022)/Composition of H2O in kg
19 w=(T*(100/23))/Weight of air required in kg per kg
      of fuel
20 wN2 = (w*(77/100)) / Weight of N2 in 'w' kg of fuel in
21 T1=(P1+P2+wN2)//Total weight of all products of
     combustion in kg
22 pCO2 = (P1/T1)*100//Percentage composition of CO2 by
      weight
23 pH2O=(P2/T1)*100//Percentage composition of H2O by
      weight
24 pN2=(wN2/T1)*100//Percentage composition of N2 by
     weight
25
26 // Output
27 printf ('The weight of air required to burn one kg of
      the fuel is \%3.1 f kg \n The composition of
     products of combustion by weight is %3.2f percent
      of CO2, %3.2f percent of H2O and %3.2f percent
     of N2', w, pCO2, pH2O, pN2)
```

Scilab code Exa 8.5 MASS OF AIR

```
1 clc
2 clear
3 //Input data
4 C=85//Composition of C in percent
```

```
5 H2=15//Composition of H2 in percent
  6 CV=10600//Calorific value in kcal/kg
  7 e02=60//Percentage of air in excess
  8 bhp=240//Brake horse power in h.p
  9 nth=30//Thermal efficiency in percent
10 02=23 // Percentage of oxygen contained in air by
                     weight
11 wC=12//Molecular weight of carbon
12 wH2=2//Molecular weght of H2
13 w02=32//Molecular weight of O2
14
15 // Calculations
16 mma = (100/23) * (((C/100) * (w02/wC)) + ((H2/100) * (w02/(wH2)) + ((H2/100) * (w02/(wH2)) + ((H2/100) * (w02/(wH2)) + ((H2/100) * (w02/wC)) + ((
                     *2))))//Minimum air in kg per kg oil
17 aa=((100+e02)/100)*mma//Actual air supplied in kg
                     per kg oil
18 q = ((bhp*(4500/427))/(nth/100))//Heat supplied in
                     kcal/min
19 mf = (q/CV) // Mass of fuel supplied in kg/min
20 ma=(aa*mf)//Mass of air supplied in kg/min
21
22 //Output
23 printf('The weight of air is %3.2 f kg/min', ma)
```

Scilab code Exa 8.6 VOLUMETRIC ANALYSIS

```
1 clc
2 clear
3 //Input data
4 //C + O2 -> CO2
5 // 12 + 32 -> 44
6 C=12//Molecular weight of carbon
7 O2=32//Molecular weight of O2
8 CO2=44//Molecular weight of CO2
9 N2=28//Molecular weight of N2
```

```
10
11 // Calculations
12 wair=(02/C)*(100/23) // Air required per kg of C in kg
13 wN2=(02/C)*(77/23) // N2 associated with the air in kg
14 pC02=(C02/C)/C02// Parts by volume/k for CO2
15 pN2=(wN2/N2) // Parts by volume/k for N2
16 Tv=(pC02+pN2) // Total parts by volume
17 ppC02=(pC02/Tv)*100// Percentage volume of CO2
18 ppN2=(pN2/Tv)*100// Percentage volume of N2
19
20 // Output
21 printf('The volumetric analysis of the flue gas when pure carbon is burnt with a minimum quantity of air is given by \n CO2 -> %3.1 f percent \n N2 -> %3.1 f percent', ppC02, ppN2)
```

Scilab code Exa 8.7 PERCENTAGE COMBUSTION

```
1 clc
2 clear
3 //Input data
4 C=90//Percentage composition of C
5 H2=3.3//Percentage composition of H2
6 02=3//Percentage composition of O2
7 N2=0.8//Percentage composition of N2
8 S=0.9//Percentage composition of S
9 Ash=2//Percentage composition of Ash
10 e02=50//Percentage of excess air
11 mC=12//Molecular weight of carbon
12 mS=32//Molecular weight of sulphur
13 mCO2=44//Molecular weight of CO2
14 mO2=32//Molecular weight of O2
15 mSO2=64//Molecular weight of SO2
16 mN2=28//Molecular weight of N2
17
```

```
18 // Calculations
19 w1=(C/100)//Weight per kg of fuel of C in kg
20 w2=(H2/100)/Weight per kg of fuel of H2 in kg
21 w3=(S/100)//Weight per kg of fuel of S in kg
22 O1=(8/3)//Oxygen required per kg of constituent for
     C in kg
23 O2=8//Oxygen required per kg of constituent for H2
     in kg
24 O3=1//Oxygen required per kg of constituent for S in
  011=(w1*01)//Oxygen required per kg of fuel for C in
26
  022=(w2*02)//Oxygen required per kg of fuel for H2
      in kg
  033=(w3*03)//Oxygen required per kg of fuel for S in
27
28 T = (011 + 022 + 033 - (02/100)) / Total Oxygen required per
     kg of fuel in kg
29 ma=(T*(100/23))/Minimum air required in kg
30 aN2 = (ma*((100+e02)/100)*(77/100))/N2 in actual air
      supply in kg
31 TN2=(aN2+(N2/100))//Total N2 in kg
32 \text{ wt=(ma*(e02/100)*(23/100))//Weight of air due to}
      excess O2 in kg
33 TSO2=(w3*(mSO2/mS))//Total SO2 in kg
34 TCO2 = (w1*(mCO2/mC))/Total CO2 in kg
35 pCO2=(TCO2/mCO2)//Parts by volume of CO2
36 \text{ pSO2}=(\text{TSO2/mSO2})//\text{Parts} by volume of SO2
37 pO2=(wt/mO2)//Parts by volume of O2
38 pN2=(TN2/mN2)//Parts by volume of N2
39 Tv=(pCO2+pSO2+pN2+pO2)//Total parts by volume
40 ppCO2=(pCO2/Tv)*100//Percentage volume of CO2
41 ppSO2=(pSO2/Tv)*100//Percenatge volume of SO2
42 pp02=(p02/Tv)*100//Percentage volume of O_2
43 ppN2=(pN2/Tv)*100//Percentage volume of N2
44
45 // Output
46 printf ('Percentage combustion of the dry flue gases
```

by volume is $\n CO2 \%3.2 f$ percent $\n SO2 \%3.2 f$ percent $\n O2 \%3.1 f$ percent $\n N2 \%3.2 f$ percent', ppCO2,ppSO2,ppN2)

Scilab code Exa 8.8 PARTIAL PRESSURE

```
1 clc
2 clear
3 //Input data
4 C=85//Composition of C in percent
5 H2=12.3//Composition of H2 in percent
6 i=2.7//Incombustible residue composition in percent
7 ma=25//Mass of air supplied in kg of air per kg of
      fuel
8 p02=23//Percentage of oxygen in gemetric analysis of
9 pN2=77//Percentage of nitrogen in gemetric analysis
      of air
10 p=1.03//Total pressure of the exhaust gases in kg/cm
11 mC=12//Molecular weight of carbon
12 mO2=32//Molecular weight of O2
13 mCO2=44//Molecular weight of CO2
14 mH2=2//Molecular weight of H2
15 mH2O=18//Molecular weight of H2O
16 mN2=28//Molecular weight of N2
17
18 // Calculations
19 xCO2 = ((C/100) * (mCO2/mC)) / per kg of fuel, the
      products formed in kg
20 \text{ xH2O} = ((\text{H2}/100)*((2*\text{mH2O})/(2*\text{mH2})))/\text{per kg of fuel},
      the products formed in kg
21 \times 02 = (((C/100) * (m02/mC)) + ((H2/100) * (m02/(2*mH2)))) / 
      Oxygen used in kg
22 xN2=(pN2/pO2)*xO2//Associated nitrogen in kg
```

```
23 mma=(x02+xN2)//Minimum air required in kg
24 ea=(ma-mma)//Excess air supplied in kg
25 XO2=((pO2/100)*ea)//Mass of O2 in excess air in kg
26 XN2=((pN2/100)*ea)//Mass of N2 in excess air in kg
27 wCO2=xCO2/mCO2//Parts by volume for CO2
28 \text{ wO2=XO2/mO2//Parts} by volume for O2
29 wN2=((XN2+xN2)/mN2)//Parts by volume for N2
30 wH20=(xH20/mH20)//Parts by volume for H2)
31 Tv = (wCO2 + wO2 + wN2 + wH2O) / Total parts by volume
32 ppCO2=(wCO2/Tv)*100//Percentage volume of CO2
33 pp02=(w02/Tv)*100//Percentage volume of O2
34 ppN2=(wN2/Tv)*100//Percentage volume of N2
35 ppH20=(wH20/Tv)*100//Percenatage volume of H2O
36 Tv1=(wCO2+wO2+wN2)//Total parts by volume for dry
     products
37 pp1CO2=(wCO2/Tv1)*100//Percentage volume of CO2 for
     dry analysis
 pp102=(w02/Tv1)*100//Percentage volume of O2 for dry
      analysis
  pp1N2=(wN2/Tv1)*100//Percentage volume of N2 for dry
      analysis
40 papH20=(ppH20/100)*p//Partial pressure of H2O in kg/
41
42 //Output
43 printf ('The volumetric analysis for wet products
     volumetric analysis for dry products gives (in
     \%3.1 f \ \  The partial pressure of the vapour is
     \%3.4\,\mathrm{f} kg/cm^2, ppCO2, ppO2, ppN2, ppH2O, pp1CO2, pp1O2
     ,pp1N2,papH20)
```

Scilab code Exa 8.9 ANALYSIS BY WEIGHT

```
1 clc
2 clear
3 //Input data
4 CO2=15//Volumetric analysis composition in percent
5 CO=2.2//Volumetric analysis composition in percent
6 02=1.6//Volumetric analysis composition in percent
7 N2=81.2//Volumetric analysis composition in percent
8 m02=32//Molecular weight of O2
9 mCO2=44//Molecular weight of CO2
10 mCO=28//Molecular weight of CO2
11 mN2=28//Molecular weight of N2
12
13 // Calculations
14 pCO2=(CO2/100)*mCO2//Proportional weight for CO2
15 pCO=(CO/100)*mCO//Proportional weight for CO
16 p02=(02/100)*m02//Proportional weight for O2
17 pN2=(N2/100)*mN2//Proportional weight for N2
18 T=(pCO2+pCO+pO2+pN2)//Total proportional weight
19 ppCO2=(pCO2/T)*100//Weight per kg of exhaust gas for
      CO2
20 ppCO=(pCO/T)*100//Weight per kg of exhaust gas for
21 ppO2=(pO2/T)*100//Weight per kg of exhaust gas for
22 ppN2=(pN2/T)*100//Weight per kg of exhaust gas for
23
24 printf ('The analysis by weight is given by (in
     \%3.1 \, \mathrm{f} \, \ln \, \mathrm{N2} \rightarrow \%3.1 \, \mathrm{f}, ppCO2, ppCO, ppO2, ppN2)
```

Scilab code Exa 8.10 GRAVIMETRIC ANALYSIS

```
1 clc
2 clear
```

```
3 CO2=10.9//Volumetric analysis composition in percent
4 CO=1//Volumetric analysis composition in percent
5 02=7.1//Volumetric analysis composition in percent
6 N2=81//Volumetric analysis composition in percent
7 m02=32//Molecular weight of O2
8 mCO2=44//Molecular weight of CO2
9 mCO=28//Molecular weight of CO
10 mN2=28//Molecular weight of N2
11
12 // Calculations
13 pCO2=(CO2/100)*mCO2//Proportional weight for CO2
14 pCO=(CO/100)*mCO//Proportional weight for CO
15 p02=(02/100)*m02//Proportional weight for O2
16 pN2=(N2/100)*mN2//Proportional weight for N2
17 T=(pCO2+pCO+pO2+pN2)//Total proportional weight
18 ppCO2=(pCO2/T)*100//Weight per kg of exhaust gas for
       CO2
19 ppCO=(pCO/T)*100//Weight per kg of exhaust gas for
20 ppO2=(pO2/T)*100//Weight per kg of exhaust gas for
21 ppN2=(pN2/T)*100//Weight per kg of exhaust gas for
22
23 printf ('The gravimetric analysis is given by (in
      \%3.2 \, \mathrm{f} \, \backslash \mathrm{n} \, \mathrm{N2} \, -> \, \%3.2 \, \mathrm{f} \, , \mathrm{ppCO2}, \mathrm{ppCO}, \mathrm{ppO2}, \mathrm{ppN2})
```

Scilab code Exa 8.11 WEIGHT OF AIR

```
1 clc
2 clear
3 //Input data
4 CO2=10//Volumetric analysis composition in percent
5 N2=80//Volumetric analysis composition in percent
```

```
6 C=80//Carbon content of the fuel in percent
7 m02=32//Molecular weight of O2
8 mCO2=44//Molecular weight of CO2
9 mN2=28//Molecular weight of N2
10 mC=12//Molecular weight of carbon
11
12 // Calculations
13 O2=100-(N2+CO2)//Volumetric analysis composition in
14 pCO2=(CO2/100)*mCO2//Proportional weight for CO2
15 p02=(02/100)*m02//Proportional weight for O2
16 pN2=(N2/100)*mN2//Proportional weight for N2
17 T=(pCO2+pO2+pN2)//Total proportional weight
18 ppCO2=(pCO2/T)//Weight per kg of exhaust gas for CO2
19 ppO2=(pO2/T)//Weight per kg of exhaust gas for O2
20 ppN2=(pN2/T)//Weight per kg of exhaust gas for N2
21 wC=(ppCO2*(mC/mCO2))//Weight of carbon per kg of
     exhaust gases in kg
22 WC=((C/100)/wC)//Weight of exhaust gases per kg of
     fuel burned in kg
va=(WC-(ppCO2+ppO2+ppN2))/Weight of air supplied
     per kg fuel in kg
24
25 // Output
26 printf ('Weight of air supplied per kg of fuel is %i
     kg',wa)
```

Scilab code Exa 8.12 AIR SUPPLIED

```
1 clc
2 clear
3 //Input data
4 CO2=12//Volumetric analysis composition in percent
5 CO=4//Volumetric analysis composition in percent
6 N2=84//Volumetric analysis composition in percent
```

```
7 mO2=32//Molecular weight of O2
8 mCO2=44//Molecular weight of CO2
9 mCO=28//Molecular weight of CO
10 mN2=28//Molecular weight of N2
11 mC=12//Molecular weight of carbon
12 mH2=2//Molecular weight of H2
13
14 // Calculations
15 pCO2=(CO2/100)*mCO2//Proportional weight for CO2
16 pCO=(CO/100)*mCO//Proportional weight for CO
17 pN2=(N2/100)*mN2//Proportional weight for N2
18 T=(pCO2+pCO+pN2)//Total proportional weight
19 ppCO2=(pCO2/T)//Weight per kg of exhaust gas for CO2
20 ppCO=(pCO/T)//Weight per kg of exhaust gas for CO
21 ppN2=(pN2/T)//Weight per kg of exhaust gas for N2
22 wC=((ppCO2*(mC/mCO2))+(ppCO*(mC/mCO)))//Weight of
     carbon per kg of flue gases
23 pC=((6*mC)/(6*mC+7*mH2))//Percentage by weight of
     carbon in C6H14
24 we=(pC/wC)//Weight of exhaust gases per kg of fuel
     in kg
25 wa=(we-(ppCO2+ppCO+ppN2))//Weight of air supplied
     per kg of fuel in kg
26 \text{ tw} = ((100/23)*(((m02/mC)*pC)+((m02/(2*mH2))*0.163)))
     //Theoretical amount of air required for complete
      combustion of C6H14 in kg
27 exc=(wa-tw)//Excess air supplied per kg of fuel in
     kg
28
29 //Output
30 printf ('Excess air supplied per kg of fuel is %3.1 f
     kg(deficient)', exc)
```

Scilab code Exa 8.13 PERCENTAGE COMPOSITION

```
1 clc
2 clear
3 //Input data
4 C=84//Gravimetric analysis composition in percent
5 H2=12//Gravimetric analysis composition in percent
6 S=1.5//Gravimetric analysis composition in percent
7 02=1.5//Gravimetric analysis composition in percent
8 ma=20//Mass of air in kg
9 pC=4//Percent of carbon in the fuel which is burnt
      to form CO
10 mO2=32//Molecular weight of O2
11 mCO2=44//Molecular weight of CO2
12 mCO=28//Molecular weight of CO
13 mN2=28//Molecular weight of N2
14 mC=12//Molecular weight of carbon
15 mH2=2//Molecular weight of H2
16 mS=32//Molecular weight of S
17 mSO2=64//Molecular weight of SO2
18
19 // Calculations
20 \quad mm = ((100/23)*((C/100)*(mO2/mC)+(H2/100)*(mO2/(2*mH2))
     +(S/100)*(m02/mS)-(02/100))/Minimum air in kg/
     kg of fuel
21 //When 20 kg of air is supplied
22 \text{ xCO2} = ((\text{C}/100) * (\text{mCO2/mC})) / \text{Mass of CO2 in kg}
23 xSO2 = ((S/100) * (mSO2/mS)) / Mass of SO2 in kg
24 x02 = ((23/100) * (ma-mm)) / Mass of O2 in kg
25 xN2 = ((77/100) * ma) / Mass of N2 in kg
26 nCO2 = (xCO2/mCO2) // Parts by volume of CO2
27 nSO2 = (xSO2/mSO2) // Parts by volume of SO2
28 nO2=(xO2/mO2)/Parts by volume of O2
29 nN2 = (xN2/mN2) / Parts by volume of N2
30 T=(nCO2+nSO2+nO2+nN2)//Total parts by volume
31 pCO2=(nCO2/T)*100//Percentage volume of CO2
32 pSO2=(nSO2/T)*100//Percentage volume of SO2
33 pO2=(nO2/T)*100//Percentage volume of O2
34 pN2=(nN2/T)*100//Percentage volume of N2
35 //4% of available carbon is burnt to CO then per kg
```

```
of fuel
36 \text{ yCO2} = ((\text{C}/100)/(1+(\text{pC}/100)))*(\text{mCO2/mC})/\text{Mass of CO2}
37 \text{ yCO} = (((C/100) - ((C/100) / (1+(pC/100))))*(mCO/mC)) / (1+(pC/100)))
      Mass of CO in kg
38 y02 = ((C/100) * (m02/mC)) / Mass of O2 in kg
39 e02 = (y02 - (((C/100)/(1+(pC/100)))*(m02/mC)+(((C/100)))
       -((C/100)/(1+(pC/100)))*(mO2/(2*mC))))
40 nnCO2=(yCO2/mCO2)//Parts by volume of CO2
41 nnCO=(yCO/mCO)//Parts by volume of CO
42 nnSO2 = (xSO2/mSO2) / Parts by volume of SO2
43 nnO2=((xO2+eO2)/mO2)/Parts by volume of O2
44 nnN2=(xN2/mN2)//Parts by volume of N2
45 TT = (nnCO2 + nnCO + nnSO2 + nnO2 + nnN2) / Total parts by
       volume
46 ppCO2=(nnCO2/TT)*100//Percentage volume of CO2
47
48 //Output
49 printf ('Minimum weight of air required for complete
       combustion of 1 kg of the fuel is \%3.1f kg/kg of
       fuel \n\n Percentage composition by volume when
       % kg of air is supplied (in percent) n CO2 \rightarrow
      \%3.1 \text{ f} \setminus \text{n SO2} \longrightarrow \%3.1 \text{ f} \setminus \text{n O2} \longrightarrow \%3.1 \text{ f} \setminus \text{n N2} \longrightarrow \%3
       .1f \n\n The percentage volume of CO2 when %i
       percent of the carbon in the fuel is burnt to
      form CO is %3.1 f percent', mm, ma, pCO2, pSO2, pO2, pN2
       ,pC,ppCO2)
```

Scilab code Exa 8.14 EXCESS AIR

```
1 clc
2 clear
3 //Input data
4 C=85//Composition by weight in percent
5 H2=14//Composition by weight in percent
```

```
6 x=50//Percentage of excess air
7 Ta=[70+273,500+273] //Temperature of air entering and
      leaving in K
8 Cp=0.24//Mean specific heat of air in kJ/kg.K
9 qC=8080//Heat liberated in kcal/kg
10 qH2=34250//Heat liberated in kcal/kg
11 a=23//Air contains 23% by weight of O2
12 mO2=32//Molecular weight of O2
13 mCO2=44//Molecular weight of CO2
14 mCO=28//Molecular weight of CO
15 mN2=28//Molecular weight of N2
16 mC=12//Molecular weight of carbon
17 mH2=2//Molecular weight of H2
18
19 // Calculations
20 mm = ((100/a)*((C/100)*(mO2/mC)+(H2/100)*(mO2/(2*mH2))
     ))//Minimum air required in kg/kg of fuel
21 Q1=((C/100)*qC+(H2/100)*qH2)/Heat in kcal/kg fuel
22 ea=((x/100)*mm)//Excess air supplied in kg/kg fuel
23 Q2=((mm/2)*Cp*(Ta(2)-Ta(1)))/Heat in kcal/kg fuel
24
25 // Output
26 printf('(a) Minimum quantity of air necessary for
     the complete combustion of 1 kg of fuel is \%3.2 f
     kg/kg of fuel \n (b) Heat released per kg of fuel
      when the carbon is burnt to CO2 and hydrogen is
     burnt to H2O is %3.0f kcal/kg fuel \n (c) Heat
     carried away by the excess air is \%3.0 f kcal/kg
     fuel', mm, Q1, Q2)
```

Scilab code Exa 8.15 VOLUME OF GAS

```
1 clc
2 clear
3 //Input data
```

```
4 CO=17//Percentage composition by volume
5 H2=53.4//Percentage composition by volume
6 CH2=28.8//Percentage composition by volume
7 02=0.8//Percentage composition by volume
8 ea=30//Percentage of excess air
9 \text{ v=1//Volume in m}^3
10
11 // Calculations
12 ta=((100/21)*((CO/100)/2+(H2/100)/2+(CH2/100)*2-(O2)
      /100)))//Theoretical air in m<sup>3</sup>/m<sup>3</sup> of gas
13 aa=((1+(ea/100))*ta)//Actual air in m^3/m^3 of gas
14 Vg=(v+aa)//Volume of gas air mixture in m<sup>3</sup>/m<sup>3</sup> of
      gas
15
16 //Output
17 printf('Total quantity of air required is %3.2f m^3/
      m<sup>3</sup> of gas \n The volume of gas air mixture is \%3
      .2 \, f \, m^3/m^3 \, of \, gas', ta, Vg)
```

Scilab code Exa 8.16 VOLUME OF AIR

```
14 mm02=((2*CH4)+(3*C2H4)+(H2/2)+(CO/2))-O2//Minimum
      moles of O2
15 mCO2 = (CH4 + (2*C2H4) + CO + CO2) / Moles of CO2
16 \text{ mH2O} = ((2*CH4) + (2*C2H4) + H2) / Moles of H2O
17 mN2 = (N2 + (79/21) * mmO2) / Moles of N2
18 ma=((100/21)*(mm02/100))/Minimum air in m^3/m^3 of
      gas
19 ea=(v-ma)//Excess air in m^3/m^3 of gas
20 tm=(mCO2+mN2+ea)*2//Total moles of dry products per
      100 moles of gas
21 pCO2=(mCO2/tm)*100//Percentage of CO2 by volume in
      dry flue gases
22
23 // Output
24 printf ('Minimum volume of air necessary for the
      complete combustion of 1 m<sup>3</sup> of coal gas is %3.2 f
       m<sup>3</sup>/m<sup>3</sup> of gas \n Percentage volume of CO2 in
      dry flue gases is %3.2f percent', ma, pCO2)
```

Scilab code Exa 8.17 VOLUMETRIC COMPOSITION

```
clc
clear
//Input data
C=86//Percentage of carbon
H2=14//Percentage of Hydrogen
ea=20//Percentage of excess air
O2=23//Weight of oxygen in air in percent
m02=32//Molecular weight of O2
mC02=44//Molecular weight of CO2
mC0=28//Molecular weight of CO
mN2=28//Molecular weight of N2
mC=12//Molecular weight of N2
mC=12//Molecular weight of H2
MH20=18//Molecular weight of H2
```

```
15
16 // Calculations
17 ma = ((100/02)*((C/100)*(m02/mC)+(H2/100)*(m02/(2*mH2))
       )))/Minimum weight of air required in kg/kg
       petrol
18 //Products of combustion by weight per kg-petrol
19 XCO2 = (C/100) * (mCO2/mC) / (CO2) in kg
20 \text{ XH2O} = (\text{H2}/100) * (\text{mH2O}/\text{mH2}) / /\text{H2O} \text{ in kg}
21 XO2 = (XCO2 + XH2O - 1) * (ea/100) / O2 in kg
22 XN2 = (ma*(1+(ea/100))*((100-02)/100))/N2 in kg
23 XT = (XCO2 + XH2O + XO2 + XN2) / Total weight in kg
24 // Percentage analysis by weight
25 \text{ xCO2} = (\text{XCO2}/\text{XT}) * 100 //\text{CO2}
26 \text{ xH20} = (\text{XH20}/\text{XT}) * 100 / /\text{H2O}
27 \times 02 = (X02/XT) \times 100 / / O2
28 \times N2 = (XN2/XT) * 100 / / N2
29 // Percentage by weight to molecular weight
30 \quad xxCO2 = (xCO2/mCO2) //CO2
31 \text{ xxH20} = (\text{xH20/mH20}) / /\text{H2O}
32 \times x \times 02 = (x \cdot 02 / m \cdot 02) / / 02
33 xxN2 = (xN2/mN2) / N2
34 xxt=(xxCO2+xxH2O+xxO2+xxN2)//Total percentage by
       weight to molecular weight
35 // Percentage by volume
36 \text{ pCO2} = (\text{xxCO2}/\text{xxt}) * 100 / /\text{CO2}
37 pH20=(xxH20/xxt)*100//H2O
38 p02=(xx02/xxt)*100//O2
39 pN2=(xxN2/xxt)*100//N2
40
41 //Output
42 printf ('Volumetric composition of the products of
       \%3.1 \text{ f} \setminus \text{n} \cup \text{O2} \longrightarrow \%3.2 \text{ f} \setminus \text{n} \cup \text{N2} \longrightarrow \%3.2 \text{ f}', pCO2, pH2O
       ,p02,pN2)
```

Scilab code Exa 8.18 WEIGHT

```
1 clc
2 clear
3 //Input data
4 bhp=20//Brake horse in h.p
5 N=320/Speed in r.p.m
6 C=84//Percentage of carbon
7 H2=16 // Percentage of hydrogen
8 CV=10800//Calorific value in kcal/kg
9 bth=30//Brake thermal efficiency in percent
10 mO2=32//Molecular weight of O2
11 mCO2=44//Molecular weight of CO2
12 mCO=28//Molecular weight of CO
13 mN2=28//Molecular weight of N2
14 mC=12//Molecular weight of carbon
15 mH2=2//Molecular weight of H2
16 mH2O=18//Molecular weight of H2O
17
18 // Calculations
19 W=(bhp*4500)/427//Work done in kcal
20 Wc = (W*2)/N//Work done per cycle in kcal
21 qs=(Wc/(bth/100))//Heat supplied per cycle in kcal
22 wf = (qs/CV) // Weight of fuel used per cycle in kg
23 t02 = ((C/100) * (m02/mC) + (H2/100) * (m02/(2*mH2))) / Total
      O2/kg fuel in kg
24 mw = (t02/(23/100))/Minimum weight of air required in
      kg/kg fuel
  aw=(mw*2)//Actual weight of air supplied in kg/kg
25
      fuel
  wac=(aw*wf)//Wt. of air supplied/ cycle in kg. In
     textbook, it is given wrong as 0.1245 kg
27
28 //Output
29 printf('(a) the weight of fuel used per cycle is %3
      .6 f kg \n (b) the actual weight of air taken in
     per cycle is %3.4 f kg', wf, wac)
```

Scilab code Exa 8.19 WEIGHT OF AIR

```
1 clc
2 clear
3 //Input data
4 CO2=8.85//Percentage composition by volume
5 CO=1.2//Percentage composition by volume
6 02=6.8//Percentage composition by volume
7 N2=83.15//Percentage composition by volume
8 C=84//Percentage composition by weight
9 H2=14//Percentage composition by weight
10 a02=2//Percentage composition by weight
11 mO2=32//Molecular weight of O2
12 mCO2=44//Molecular weight of CO2
13 mCO=28//Molecular weight of CO
14 mN2=28//Molecular weight of N2
15 mC=12//Molecular weight of carbon
16 mH2=2//Molecular weight of H2
17 mH2O=18//Molecular weight of H2O
18
19 // Calculations
20 //O2 required per kg of fuel
21 \text{ xC} = ((\text{C}/100) * (\text{mO2/mC})) / \text{C}
22 xH2 = ((H2/100) * (mO2/(2*mH2))) //H2
23 \times 02 = -(a02/100)//02
24 tt02=(xC+xH2-x02)//Theoretical total oxygen required
       in kg/kg fuel
25 twa=(tt02/(23/100))//Theoretical weight of air in kg
      /kg fuel
  //Conversion of volumetric analysis of the flue gas
26
      into a weight analysis
27 // Percenatge by volume * mol. wt
28 \text{ xxCO2} = (\text{CO2} * \text{mCO2}) / / \text{CO2}
29 xxCO = (CO*mCO) / /CO
```

```
30 \text{ } \text{xxO2} = (02*\text{mO2}) //\text{O2}
31 \times xN2 = (N2*mN2) / N2
32 xxt = (xxCO2 + xxCO + xxO2 + xxN2) / Total
33 //Percentage by weight
34 yCO2=(xxCO2/xxt)*100//CO2
35 yCO = (xxCO/xxt)*100//CO
36 \text{ yO2} = (xxO2/xxt)*100//O2
37 yN2 = (xxN2/xxt)*100//N2
38 wcd=((yCO2/100)*(mC/mCO2))+((yCO/100)*(mC/mCO))//
      Weight of carbon/ kg of dry flue gas in kg
39 wdf = ((C/100)/wcd)/Wt. of dry flue gas/kg fuel in kg
40 wxf = (wdf * (y02/100)) / Weight of excess O2/kg fuel in
      kg
41 we02 = (wxf/(23/100))/Weight of excess air in kg/kg
42 was=(twa+we02)//Weight of air supplied/kg fuel in kg
43
44 //Output
45 printf ('Weight of air supplied per kg fuel burnt is
      \%3.2 f \text{ kg}', \text{was}
```

Chapter 9

SPARK IGNITION ENGINES

Scilab code Exa 9.1 WEIGHT

```
1 clc
2 clear
3 //Input data
4 d=0.001//Diameter of the jet in m
5 vd=104//Venturi depression in cm of water. In
     textbook, it is given as 10 cm
6 Cd=0.65//Coefficient of discharge
7 g=0.76//Specific gravity of petrol
8 w=1000//Weight of water per one cu.m in kg
9
10 // Calculations
11 pa=(vd/100)*w//Venturi depression in kg/m^2
12 dp=(g*w)//Density of petrol in kg/m^3
13 wf = (((3.14*d^2)/4)*Cd*sqrt(2*9.81*dp*pa))/10^-3//
      Petrol discharge in gm/sec neglecting nozzle lip
14
15 //Output
16 printf ('The weight of petrol discharged is %3.2 f gm/
     sec', wf)
```

Scilab code Exa 9.2 AIR FUEL RATIO

```
1 clc
2 clear
3 //Input data
4 d1=0.075//Throat diameter in m
5 Ca=0.93//Coefficient of air flow
6 d2=0.005//Orifice diameter in m
7 Cf=0.68//Coefficient of fuel discharge
8 ap=1//Approach factor
9 dp=0.15//Pressure drop in kg/cm<sup>2</sup>
10 da=1.29 // Density of air in kg/m^3
11 df = 720 // Density of fuel in kg/m^3
12
13 // Calcultions
df)//The air-fuel ratio neglecting the nozzle lip
15
16 // Output
17 printf('The air-fuel ratio neglecting the nozzle lip
      is %3.1 f', w)
```

Scilab code Exa 9.3 THEORETICAL RATIO OF DIAMETER

```
1 clc
2 clear
3 //Input data
4 af=15//Air fuel ratio
5 dp=753//Density of petrol in kg/m^3
6 da=1.28//Density of air in kg/m^3
7 C=[0.84,0.7]//Coefficient of discharge for air and fuel respectively
```

```
8
9 //Calculations
10 A=1/(af*(C(2)/C(1))*sqrt(dp/da))//Ratio of areas
11 d=sqrt(A)//Ratio of diamter of jet to diameter of
    venturi
12 x=(1/d)//Reverse of ratio
13
14 //Output
15 printf('The ratio of diameter of jet to diameter of
    venturi is 1 : %3.1f',x)
```

Scilab code Exa 9.4 SECTION AT THE THROAT

```
1 clc
2 clear
3 //Input data
4 D=[10,12]//Dimensions of four cylinder in 10 cm * 12
      cm
5 n=4//Four cylinder
6 N=2000/Speed in r.p.m
7 d=0.03//Venturi throat in m
8 nv=70//Volumetric efficiency of the engine in
      percent
9 Ca=0.8//Coefficient of air flow
10 da=1.29 // Density of air in kg/m^3
11
12 // Calculations
13 Vs = (3.14/4) * (D(1)/100)^2 * (D(2)/100) / Stroke volume
      of one cylinder in m<sup>3</sup>
14 Va=(Vs*n*(nv/100)*(N/2))/Volume of air drawn per
      minute in m<sup>3</sup>
15 w=(Va*da)/60//Weight of air drawn per sec
16 dp = ((w/((3.14/4)*d^2*Ca))^2/(2*9.81*da))/Venturi
      depression in kg/cm<sup>2</sup>
17
```

```
18 //Output
19 printf('The venturi depression is %3.1 f kg/m^2',dp)
```

Scilab code Exa 9.5 SIZE OF THE VENTURI

```
1 clc
2 clear
3 //Input data
4 D=[8.25,11.5]/Dimensions of four cylinder in 8.25
     cm* 11.5 cm
5 n=4//Four cylinder
6 N=3000/Speed in r.p.m
7 v=150//Venturi depression in cm of water
8 nv=80//Volumetric efficiency of the engine in
      percent
9 af=14//Air fuel ratio
10 Ca=0.84//Coefficient of air flow
11 Cf=0.7//Coefficient of fuel orifice
12 da=1.29//Density of air in kg/m^3
13 df = 700 // Density of fuel in kg/m^3
14 dw=1000//Density of water in kg/m<sup>3</sup>
15
16 // Calculations
17 Va=((3.14/4)*(D(1)/100)^2*(D(2)/100)*n*(nv/100)*(N)
     /(2*60)))//Maximum amount of air passing through
      the venturi in m<sup>3</sup>
18 vd=(v/100)*dw//Venturi depression in kg/m^2
19 va=(Ca*sqrt((2*9.81*vd)/da))/Velocity of air in m/s
20 d=sqrt((Va/va)*(4/3.14))//Throat diameter of venturi
       in m
21 Af=1/(af*(va/Va)*sqrt(df/da)*(Cf/Ca))//Area of
      orifice in m<sup>2</sup>
22 df = sqrt((Af*4)/3.14)*1000//Diameter of orifice in mm
23
24 // Output
```

25 printf('The size of venturi is %i kg/m^2 \n The diameter of fuel orifice is %3.2 f mm', vd, df)

Scilab code Exa 9.6 FUEL CONSUMPTION

```
1 clc
2 clear
3 D=[7.5,10]/Dimensions of four cylinder in 7.5 cm
     diameter and 10 cm stroke
4 n=6//Six cylinder
5 pC=84//Percentage of carbon in volatile fuel
6 pH2=16//Percentage of hydrogen in volatile fuel
7 dc=(38.5/1000)//Diameter of the throat of the choke
     tube in m
8 N=3000/Speed in r.p.m
9 nv=0.8//Volumetric efficiency in ratio
10 p=0.914//Pressure at the throat of the choke tube in
      kg/cm^2
11 T=15.5+273//Temperature at the throat of the choke
     tube in K
12 Ts=273//Temperature of 0 degree C in K
13 ps=1.027//Atmospheric pressure in kg/cm<sup>2</sup>
14 Ra=29.27 // Universal gas constant for air in kg.m/kg.
     K
15 Rf=9.9//Gas constant for fuel in kg.m/kg.K
16 p02=23//Composition by weight of oxygen in air in
      percent
17 pN2=77//Composition by weight of nitogen in air in
      percent
18 mO2=32//Molecular weight of O2
19 mH2=2//Molecular weight oh H2
20 mC=12//Molecular weight of carbon
21
22 // Calculations
23 Vm = ((3.14/4)*(D(1)/100)^2*(D(2)/100)*n*(N/2)*nv)//
```

```
Volume of mixture supplied per sec in m<sup>3</sup>
24 qa = ((100/p02)*(((pC/100)*(m02/mC))+((pH2/100)*(m02/mC))
      /(2*mH2)))))//Quantity of air required for
      complete combustion of fuel in kg
25 vf=(Rf*Ts)/(ps*10^4)//Specific volume of volatile
      fuel in m<sup>3</sup>/kg
26 \text{ va}=(\text{Ra}*\text{Ts})/(\text{ps}*10^4)//\text{Specific volume of air in m}^3/
      kg
27 wf=(Vm/(ga*va+vf))//Flow rate of fuel in kg/min
28 Fc=(wf*60)//Fuel consumption in kg/hour
29 da=(p*10^4)/(Ra*T)//Density of air at the throat of
      the choke in kg/m<sup>3</sup>
30 Va=((qa*wf)/((3.14/4)*dc^2*da*60))/Speed of air at
      throat in m/s
31
32 //Output
33 printf('(a) The fuel consumption is \%3.1 \,\mathrm{f}\,\mathrm{kg/hr}\,\mathrm{n} (
      b) The speed of the air through the choke is \%3.1
      f m/s', Fc, Va)
```

Scilab code Exa 9.7 DIAMETER OF JET

```
clear
clear
//Input data
mf=7.5//Consumption of petrol per hour
gf=0.75//Specific gravity of fuel
Tf=25+273//Temperature of fuel in K
af=15//Air fuel ratio
dc=22//diameter of choke tube in mm
l=4//Top of the jet is 4 mm above the petrol level
    in the float chamber
Ca=0.82//Coefficient of discharge for air
Cf=0.7//Coefficient of discharge for fuel
R=29.27//Characteristic gas constant for air in kg.m
```

Scilab code Exa 9.8 AIR FUEL RATIO

```
1 clc
2 clear
3 //Input data
4 td=7.5//Throat diameter in cm
5 Ca=0.85//Coefficient of air flow
6 fd=0.5//Diameter of fuel orifice in cm
7 Cd=0.7//Coefficient of discharge
8 1=5//Nozzle lip in mm
9 x=1//Approach factor
10 dpa=0.15//Pressure drop in kg/cm<sup>2</sup>
11 da=1.29 // Density of air in kg/m^3
12 dp=720//Density of fuel in kg/m^3
13
14 // Calculations
15 afr1=(((3.14*td^2)/(3.14*fd^2))*(Ca/Cd)*sqrt(da/dp))
     //Air fuel ratio
16 afr2 = ((3.14*td^2)/(3.14*fd^2))*(Ca/Cd)*sqrt((da*dpa)
```

```
/(dp*(dpa-((1/100)*(dp/10^6)))))//Air fuel ratio
17
18 //Output
19 printf('The air fuel ratio \n (a) neglecting nozzle
    lip is %3.2 f \n (b) nozzle lip is taken into
    account is %3.2 f', afr1, afr2)
```

Scilab code Exa 9.9 VELOCITY

```
1 clc
2 clear
3 //Input data
4 td=7.5//Throat diameter in cm
5 Ca=0.85//Coefficient of air flow
6 fd=0.5//Diameter of fuel orifice in cm
7 Cd=0.7//Coefficient of discharge
8 1=5//Nozzle lip in mm
9 x=1//Approach factor
10 dpa=0.15//Pressure drop in kg/cm<sup>2</sup>
11 da=1.29 // Density of air in kg/m^3
12 dp=720//Density of fuel in kg/m^3
13
14 // Calculations
15 v = sqrt(2*9.81*(1/1000)*(dp/da)) / Velocity of air in
     m/s
16
17 // Output
18 printf('Velocity of air flow is %3.1 f m/s',v)
```

Scilab code Exa 9.10 PETROL CONSUMPTION

```
1 clc
2 clear
```

```
3 //Input data
4 x=2.8//Height above the nozzle in mm
5 va=58//Velocity of air in m/s
6 da=1.28//Density of air in kg/m<sup>3</sup>
7 dp=750//Density of petrol in kg/m<sup>3</sup>
8 An=1.8//Area of cross section of nozzle in mm<sup>2</sup>
9 Cd=0.6//Coefficient of discharge of nozzle
10 Ca=0.84//Coefficient of discharge of air
11
12 // Calculations
13 dpa=((va/Ca)^2*(da/(2*9.81)))/Change in pressure in
       kg/m^2
14 wf = ((An*10^-6)*Cd*sqrt(2*9.81*dp*(dpa-((x/1000)*dp)))
      ))//Petrol consumption in kg/sec
15
16 // Output
17 printf ('Petrol consumption is \%3.4 f kg/sec', wf)
```

Scilab code Exa 9.11 AIR FUEL RATIO

```
clc
clear
//Input data
d=0.155//Diameter of orifice in mm
Cd=0.94//Coefficient of discharge
td=3.18//Throat diameter in cm
Ca=0.84//Coefficient of discharge
x=29//Venturi depression
dw=0.89//Minimum depression of water in cm
sa=1.1//Specific weight of air in kg/m^3
sg=0.72//Specific gravity of petrol
cyd=[7.75,10.75]//Cylinder dimensions in cm
fc=10.9//Fuel consumption in kg/hr
N=3200//Speed in r.p.m
n=4//Number of cylinders
```

Scilab code Exa 9.12 THROAT PRESSURE

```
1 clc
2 clear
3 //Input data
4 af=0.066//Air fuel ratio
5 p=0.83//Pressure at the venturi throat in kg/cm<sup>2</sup>
6 pd=0.04//Pressure drop in kg/cm<sup>2</sup>
7 va=245//Air flow at sea level in kg per hour
8
9 // Calculations
10 dpa=1.03-p//Pressure at air cleaner in kg/cm<sup>2</sup>
11 d=(1.03-pd-dpa)//Throat pressure when the air
      cleaner is fitted in kg/cm<sup>2</sup>
12 naf = (af * sqrt ((1.03-d)/dpa)) / New air fuel ratio
13
14 // Output
15 printf('(a) Throat pressure when the air cleaner is
      fitted is %3.2 f kg/cm<sup>2</sup> \n (b) New air fuel ratio
       is \%3.4 \, \mathrm{f}, d, naf)
```

Scilab code Exa 9.13 DROP IN PRESSURE

```
1 clc
2 clear
3 //Input data
4 x=3//Petrol stands 3 mm below
5 Ta=15.5+273//Temperature of air in K
6 pa=1.027//Pressure of air in kg/cm<sup>2</sup>
7 R=29.27 // Characteristic gas constant in kg.m/kg.K
8 sg=0.76//Specific gravity of fuel
9 fc=6.4//Consumption of fuel in kg/hour
10 jd=1.27//Jet diameter in mm
11 Cd=0.6//Nozzle discharge coefficienct
12 Ca=0.8//Discharge coefficient of air
13 af=0.066//Air fuel ratio
14
15 // Calculations
16 df = (sg*1000) // Density of fuel in kg/m^3
17 da=(pa*10^4)/(R*Ta)//Density of air in kg/m^3
18 va=Ca*sqrt((2*9.81*x*df)/(da*1000))//Critical
      velocity of air in m/s
19 dpa=(((fc/(60*60))/((3.14/4)*(jd/1000)^2*Cd))
      ^2/(2*9.81*df))+((x/1000)*df)/Drop in pressure
     in kg/m^3
20 dpaw=(dpa/1000)*100//Drop in pressure in cm of water
21 dj = sqrt(((fc/(3600*af))/(Ca*sqrt(2*9.81*da*dpa)))
     /(3.14/4))*1000//Effective throat diameter in mm
22
23 //Output
24 printf ('Critical air velocity is \%3.2 f m/sec \n
      Effective throat diameter of the venturi is \%3.1f
      mm \n The drop in pressure in the venturi is \%3
      .2 f cm of water', va, dj, dpaw)
```

Scilab code Exa 9.14 ORIFICE DIAMETER

```
1 clc
2 clear
3 //Input data
4 ma=6.11//Flow rate of air in kg/min
5 mf=0.408//Flow arte of petrol in kg/min
6 dp=768//Density of petrol in kg/m<sup>3</sup>
7 Ta=15.5+273//Temperature of air in K
8 pa=1.027//Pressure of air in kg/cm<sup>2</sup>
9 R=29.27 // Characteristic gas constant in kg.m/kg.K
10 va=97.5//Speed of air in m/sec
11 Cv=0.84//Velocity coefficient
12 g=1.4//Ratio of specific heats
13 x=0.8//pressure at the venturi is 0.8 of the
      pressure drop at the choke
14 Cd=0.66 // Coefficient of discharge
15
16 // Calculations
17 rp=(1-((va/Cv)^2/(((2*9.81*g)/(g-1))*R*Ta)))^(g/(g
     -1))//Pressure ratio
18 p2=(pa*rp)//Pressure in kg/cm^2
19 T2=(Ta*rp^{(g-1)/g}))/Temperature in K
20 da=(p2/(R*T2))*10^4/Density in kg/m^3
21 daa=sqrt((ma/(60*va*da))/(3.14/4))*1000//Throat
      diameter in mm
22 df = sqrt((mf/(60*Cd*sqrt(2*9.81*dp*x*(pa-p2)*10^4)))
      /(3.14/4))*1000//Orifice diameter in mm
23
24 //Output
25 printf ('Throat diameter of the choke is %i mm \n The
       orifice diameter is %3.2 f mm', daa, df)
```

Scilab code Exa 9.15 ORIFICE DIAMETER

```
1 clc
2 clear
3 //Input data
4 ma=6.8//Mass flow rate of air in kg/min
5 mf=0.45//Mass flow rate of petrol in kg/min
6 pa=1.033//Pressure of air in kg/cm<sup>2</sup>
7 Ta=20+273//Temperature of air in K
8 va=97.5//Velocity of air in m/s
9 Cv=0.8//Velocity coefficient
10 g=1.4//Ratio of specific heats
11 R=29.27 // Characteristic gas constant in kg.m/kg.K
12 x=0.75//pressure at the venturi is 0.8 of the
      pressure drop at the choke
13 Cd=0.65 // Coefficient of discharge
14 pw=800//Weight of petrol in kg per cu.m
15
16 // Calculations
17 rp=(1-((va/Cv)^2/(((2*9.81*g)/(g-1))*R*Ta)))^(g/(g
     -1))//Pressure ratio
18 p2=(pa*rp)//Pressure in kg/cm^2
19 T2=(Ta*rp^{(g-1)/g})/Temperature in K
20 da=(p2/(R*T2))*10^4/Density in kg/m^3
21 daa=sqrt((ma/(60*va*da))/(3.14/4))*100//Throat
      diameter in mm
22 df = sqrt((mf/(60*Cd*sqrt(2*9.81*pw*x*(pa-p2)*10^4)))
     /(3.14/4))//Orifice diameter in mm
23
24 //Output
25 printf ('Throat diameter of the choke is %3.2 f cm \n
     The orifice diameter is %3.5 f m', daa, df)
```

Scilab code Exa 9.16 POWER DEVELOPED

```
1 clc
2 clear
3 //Input data
4 n=6//Number of cylinders
5 d=100//Diameter in mm
6 L=100//Stroke in mm
7 N=1500/Speed in r.p.m
8 ap=13.5//Air fuel ratio
9 Ta=80+273//Temperature of air in K
10 x = (7/8) / Ratio of volume drawn
11 nth=22//Thermal efficiency in percent
12 p=76 // Pressure in cm of mercury
13 CV=9000//Calorific value of petrol in kcal/kg
14 1=1524// Altitude in m
15 dp=2.54//Drop in pressure in cm of barometric
      reading
16 lx=274//Altitude rise in m
17
18 // Calculations
19 Vs = (3.14/4)*(d/10)^2*(L/10) / Swept volume in c.c
20 Va=(x*Vs)//Volume of air drawn in per cylinder per
     stroke in c.c
21 wa=(Va*10^-6*(N/2)*n)//Weight of air supplied to the
      engine per minute in kg
22 wf=(wa/ap)//Weight of fuel consumed per minute in kg
23 q=(wf*CV)//Heat supplied to the engine per minute in
      kcal
P = (q*(nth/100)*427)/4500//Power developed at ground
      level in H.P
25 db=(1/lx)*dp//Drop in barometric reading at an
      altitude of 1524 m in cm
26 Pd=(P/p)*(p-db)//Power developed at 1524 m altitude
```

```
in H.P
27
28 //Output
29 printf('Power developed at the ground level is %i H.
    P \n Power developed at an altitude of %i m is %i
    H.P',P,1,Pd)
```

Chapter 11

FOUR STROKE SPARK IGNITION ENGINE

Scilab code Exa 11.1 WEIGHT

```
1 clc
2 clear
3 //Input data
4 d=0.0625//Diameter in m
5 L=0.09//Stroke in m
6 nv=0.75//Volumetric efficiency
7 p=1.03//Pressure at N.T.P in kg/cm<sup>2</sup>
8 T=273//Temperature at N.T.P in K
9 R=29.27 // Characteristic gas constant in kg.m/kg.
      degree C
10
11 // Calculations
12 Vs = ((3.14/4)*d^2*L) / Swept volume in cu.m
13 V=(nv*Vs)//Volume of charge at N.T.P in cu.m
14 w=(p*10^4*V)/(R*T)/Weight of the charge in kg/cycle
15
16 //Output
17 printf('The weight of the charge is %3.6 f kg/cycle',
```

Scilab code Exa 11.2 INDICATED THERMAL EFFICIENCY

```
1 clc
2 clear
3 //Input data
4 n=9//Number of cylinder
5 d=0.145 // Bore in m
6 l=0.19//Stroke in m
7 r=5.9//Compression ratio
8 bhp=460//Brake horse power in B.H.P
9 N=2000/Speed in r.p.m
10 x=20//Percentage rich in mixture
11 CV=11200//Calorific value in kcal/kg
12 pC=85.3//Percentage of carbon
13 pH2=14.7//Percentage of Hydrogen
14 nv=70//Volumetric efficiency in percent
15 T=15+273//Temperature in K
16 nm=90//Mechanical efficiency in percent
17 w02=23.3 // Percentage of oxygen by weight in air
18 da=1.29//Density of air in kg/m^3
19 mC=12//Molecular weight of carbon
20 mO2=32//Molecular weight of O2
21 mH2=2//Molecular weight of H2
22
23 // Calculations
24 hihp=((bhp/(nm/100))*(4500/427))/Heat equivalent in
25 Vs = ((3.14/4)*d^2*1*(N/2)*n) / Swept volume in c.m per
       min
26 cw=(Vs/da)//Charge weight of air per minute in kg
27 \text{ ma} = (100/\text{w}02)*((\text{pC}/100)*(\text{m}02/\text{mC})+(\text{pH}2/100)*(\text{m}02/(2*))
      mH2)))//Wt. of air required per kg of fuel in kg
28 mf = (cw/ma) //Minimum fuel inkg
29 ith=(hihp/(mf*(100+x)/100*CV))*100//Indicated
```

```
thermal efficiency in percent'

30

31 //Output

32 printf('Indicated thermal efficiency of the engine is %3.1f percent',ith)
```

Scilab code Exa 11.3 PROPERTIES

```
1 clc
2 clear
3 //Input data
4 n=8//Number of cylinders
5 d=8.57//Bore in cm
6 1=8.25//Stroke in cm
7 r=7//Compression ratio
8 N=4000/Speed in r.p.m
9 la=53.35//Length of the arm in cm
10 t=10//Test duration in min
11 br=40.8//Beam reading in kg
12 gas=0.455//gasoline in kg. In textbook, it is given
     wrong as 4.55
13 CV=11400//Calorific value in kcal/kg
14 Ta=21+273//Temperature of air in K
15 pa=1.027//Pressure of air in kg/cm<sup>2</sup>
16 wa=5.44//Rate of air in kg/min
17 J=427 // Mechanical equivalent of heat in kg.m/kcal
18 R=29.27 // Characteristic gas constant in kg.m/kg.K
19
20 // Calculations
21 bhp = (2*3.14*N*br*la)/(4500*100)/Brake horse power
     in B.H.P
22 pb=(bhp*4500)/((n/2)*(1/100)*(3.14/4)*d^2*N)/Brake
     mean effective pressure in kg/cm<sup>2</sup>
23 bsfc=(gas*60)/bhp//Brake specific fuel consumption
     in kg/b.h.p.hr
```

```
24 bsac=((wa*60)/bhp)//Brake specific fuel consumption
      in kg/b.h.p.hr
25 nb = ((bhp*4500)/(J*gas*CV))*100//Brake thermal
      efficiency in percent
26 Vd = ((3.14/4)*d^2*1) // Piston displacement in c.c/
  Pd=(Vd/10^6)*(N/2)*n//Piston displacement in m^3/min
27
28 Va=((wa*R*Ta)/(pa*10^4))/Volume of air used in m^3/
29 nv=(Va/Pd)*100//Volumetric efficiency in percent
30 af = (wa/gas) // Air fel ratio
31
32 //Output
33 printf('(a) the B.H.P delivered s \%3.0 \,\mathrm{f} h.p \n (b)
      the b.m.e.p is \%3.1 \, \text{f kg/cm}^2 \, \text{n} (c) the b.s.f.c
      is %3.3 f kg/b.h.p.hr \n (d) the brake specific
      air consumption is \%3.3 \, f \, kg/b.h.p.hr \setminus n (e) the
      brake thermal efficiency is \%3.1f percent \n (f)
      the volumetric efficiency is \%3.0f percent \n (g)
       the air fuel ratio is \%3.2\,\mathrm{f}, bhp, pb, bsfc, bsac, nb
      ,nv,af)
```

Scilab code Exa 11.4 WEIGHT OF AIR

```
1 clc
2 clear
3 //Input data
4 n=4//Number of cylinders
5 N=2000//Speed in r.p.m
6 m=13.15//Mass of fuel in kg/hour
7 Vd=655.5//Displacement volume in c.c
8 da=1.2//Density of air in kg/m^3
9 d=12.7//Manometer depression in cm
10 //Qa=0.231*sqrt(ha); Qa is the flow rate in cu.m/min and ha is the pressure difference in metres
```

Scilab code Exa 11.5 CLEARANCE VOLUME

```
1 clc
2 clear
3 //Input data
4 d=10//Diameter in cm
5 l=15//Stroke in cm
6 r=6//Compression ratio
7 ihp=20//Indicated horse power in h.p
8 N=1000/Speed in r.p.m
9 n=4//Number of cylinders
10 nt=30//Thermal efficiency in percent
11 CV=10000//Calorific value in kca/kg
12 g=1.4//Ratio of specific heats
13
14 //Output
15 Vs = ((3.14/4)*d^2*1) / Swept volume in c.c
16 Vc = (Vs/(r-1)) / Clearance volume in c.c
17 na=(1-(1/r)^{(g-1)})*100//Air standard efficiency in
```

```
percent
18 pm=((ihp*4500)/((1/100)*(3.14/4)*(d/100)^2*(N/2)*n))
    //Pressure in kg/cm^2
19 pc=(ihp*4500*60)/(427*(nt/100)*CV)//Petrol
    consumption in kg/hr
20
21 //Output
22 printf('Clearance volume is %3.1 f c.c \n The air
    standard efficiency is %3.1 f percent \n Petrol
    consumption is %3.2 f kg/hr', Vc, na, pc)
```

Scilab code Exa 11.6 CYLINDER PROPERTIES

```
1 clc
2 clear
3 //Input data
4 n=6//Number of cylinders
5 \text{ P=}62//\text{Power in HP}
6 N=3000/Speed in r.p.m
7 nv=85//Volumetric efficiency in percent
8 nt=25//Thermal efficiency in percent
9 CV=10500//Calorific value in kcal/kg
10 af=15//Air fuel ratio
11 T=273//Standard atmosphere temperature in K
12 p=1.03//Standard atmosphere pressure in kg/cm<sup>2</sup>
13 R=29.27 // Characteristic gas constant in kg.m/kg.K
14 J=427 // Mechanical equivalent of heat in kg.m/kcal
15
16 // Calculations
17 q=(P*4500)/(J*(nt/100))//Heat supplied in kcal/min
18 F=(q/CV)//Fuel supplied per minute in kg
19 Fc=(F/N)*(2/n)//Fuel supplied per cycle per cylinder
       in kg
20 wt=(af*Fc)//Weight of air supplied per cycle in kg
21 d = ((((wt)*R*T)/(p*10^4*(3.14/4)*(nv/100)))^(1/3))
```

```
*100//Diameter in cm

22
23 //Output
24 printf('Cylinder bore = stroke = %3.2 f cm',d)
```

Chapter 14

THE DIESEL ENGINE

Scilab code Exa 14.1 WEIGHT OF OIL

```
1 clc
2 clear
3 //Input data
4 r=14.3//Compression ratio
5 c=5//Fuel cutoff in percent of stroke
6 w=0.006//Weight of charge in kg
7 T4=912//Final temperature in degree C abs
8 q=8300//Heat in kcal
9 x=[0.258,0.000048]//Temperature expression is 0.258T
     +0.000048T^2, where T is in degree C abs
10
11 // Calculations
12 v4=1//Assuming clearance volume as unity
13 v1=1.665//v1 from fig. 14.2 on page no. 352
14 T1=(T4*v1)/v4//Temperature in degree C abs
15 qp1=(x(1)*T1+x(2)*T1^2)/Constant pressure heat of
     mixture at temperature T1 in kcal/kg
16 qp4=(x(1)*T4+x(2)*T4^2)/Constant pressure heat of
     mixture at temperature T4 in kcal/kg
17 qre=(qp1-qp4)//Heat required by the mixture in kcal/
     kg
```

Scilab code Exa 14.2 RATIO OF MASS

```
1 clc
2 clear
3 //Input data
4 r=14//Compression ratio
5 p=1.2//Induction pipe pressure in kg/cm<sup>2</sup>
6 bp=0.65//Exhaust back pressure in kg/cm<sup>2</sup>
7 Tc=87+273//Charge temperature in K
8 Te=850+273//Exhaust temperature in K
9 T1=111+273//Temperature at the beginning of
      compression in K
10 g=1.2//Ratio of specific heats
11
12 // Calculations
13 Cw1 = ((bp*10^4)/Te)//specific heat in kJ/kg.K
14 Cw2=((p*10^4*(r-1))/Tc)//specific heat in kJ/kg.K
15 T3=((g*Te*Cw1+Cw2*Tc)/(Cw1*g+Cw2))//Temperature in K
16 t3=T3-273//Temperature in degree C
17 rw=(Cw1/Cw2)//Ratio of specific heats
18
19 //Output
20 printf ('The ratio of the mass residuals to fresh
      charge is %3.4 f', rw)
```

Chapter 15

FUEL INJECTION

Scilab code Exa 15.1 QUANTITY OF FUEL

Scilab code Exa 15.2 QUANTITY OF FUEL

```
1 clc
2 clear
3 //Input data
4 n=4//Number of cylinders
5 fc=0.215//Brake specific fuel consumption in kg/B.H.
     P hour
6 BHP=400//Brake horse power in B.H.P
7 \text{ N=}250/\text{Speed in r.p.m}
8 sg=0.9//Specific gravity
10 // Calculations
11 Fc=(fc*BHP)//Fuel consumption per hour in kg/hr
12 Fcy=(Fc/n)//Fuel consumption per cylinder in kg/hr
13 Fcyc=((Fcy/(60*(N/2)))/(sg*1000))*10^6//Fuel
      consumption per cycle in kg. In textbook it is
      given wrong as 0.0287 instead of 3.185
14
15 //Output
16 printf ('The quantity of fuel to be injected per
      cycle per cylinder is %3.3 f c.c', Fcyc)
```

Scilab code Exa 15.3 QUANTITY OF FUEL

```
1 clc
2 clear
3 //Input data
4 n=4//Number of cylinders
5 p=450//Brake Horse power in B.H.P
6 N=200//Speed in r.p.m
7 f=0.2//Fuel rate in kg per horse power hour
8 g=0.9//Specific gravity of fuel
9
10 //Output
11 Fc=(p*f)//Fuel consumption per hour in kg/hr
12 Fcy=(Fc/n)//Fuel consumption per cylinder in kg/hr
```

Scilab code Exa 15.4 SIZE OF PUMP

```
1 clc
2 clear
3 //Input data
4 //Data from problem 1
5 n=6//Number of cylinders
6 p=720//Horse power in h.p
7 N=180/Speed in r.p.m
8 f=250//Fuel rate in gm per horse power hour
9
10 Vo=20//Volume of oil in the suction chamber in c.c
11 dp=80//Discharge pressure in kg/cm<sup>2</sup>
12 voi=6//Volume of oil in the injector in c.c
13 g=0.9//Specific gravity of oil
14 b=78.8*10^-6//Coefficient of compressibility in cm
      ^2/kg when pressure is taken as atmospheric
15
16 // Calculations
17 w = (((f/1000)*p)/((N/2)*60*n))*1000/Weight of fuel
     per cycle in gm/cycle
18 Va=(w/g)//Volume of air per cycle in c.c
19 V1=(Vo+Va)//Initial volume in c.c
20 dV12=(b*V1*dp)//Change in volume in c.c
21 //Assuming in accordance with average practice that
     s=2d, nv=0.94 and full load in this pump type x
```

```
=0.5

22 d=((voi+dV12)/((3.14/4)*2*0.94*0.5))^(1/3)//Diameter

in cm

23 l=(2*d)//Stroke in cm

24

25 //Output

26 printf('The diameter of the pump is %3.2 f cm \n The

total stroke is %3.2 f cm',d,1)
```

Scilab code Exa 15.5 VELOCITY

Scilab code Exa 15.6 PUMP DISPLACEMENT

```
1 clc
2 clear
3 //Input data
4 Vf=6.2//Volume of fuel in c.c
5 l=65//Length of fuel line in cm
```

```
6 di=2.5//Inner diameter in mm
7 V=2.75//Volume of fuel in the injector valve in c.c
8 Vd=0.15//Volume of fuel to be delivered in c.c. In
      textbook it is given wrong as 0.047
9 p=140 // Pressure in kg/cm<sup>2</sup>
10 pp=1//Pump pressure in kg/cm<sup>2</sup>
11 patm=1.03//Atmospheric pressure in kg/cm<sup>2</sup>
12 b=78.8*10^-6//Coefficient of compressibility in cm
      ^2/kg when pressure is taken as atmospheric
13
14 // Calculations
15 V1 = (Vf + (3.14/4) * (di/10)^2 * 1 + V) / Initial volume in c.
16 dV = ((b*V1*(p-pp)/patm)) / Change in volume in c.c
17 d=(dV+Vd)//Total displacement of the plunger in c.c
18
19 //Output
20 printf ('The total displacement of the plunger is \%3
      .3 f c.c',d)
```

Scilab code Exa 15.7 PUMP DISPLACEMENT

```
clear
//Input data
Vf=6.75//Volume of fuel in c.c

1=65//Length of fuel line in cm
di=2.5//Inner diameter in mm
V=2.45//Volume of fuel in the injector valve in c.c
Vd=0.15//Volume of fuel to be delivered in c.c.
p=150//Pressure in kg/cm^2
pp=1//Pump pressure in kg/cm^2
patm=1.03//Atmospheric pressure in kg/cm^2
b=78.8*10^-6//Coefficient of compressibility in cm^2/kg when pressure is taken as atmospheric
```

Scilab code Exa 15.8 PLUNGER STROKE

```
1 clc
2 clear
3 //Input data
4 Vf=6.75//Volume of fuel in c.c
5 1=65//Length of fuel line in cm
6 di=2.5//Inner diameter in mm
7 V=2.45//Volume of fuel in the injector valve in c.c
8 Vd=0.15//Volume of fuel to be delivered in c.c.
9 p=150//Pressure in kg/cm^2
10 pp=1//Pump pressure in kg/cm<sup>2</sup>
11 patm=1.03//Atmospheric pressure in kg/cm<sup>2</sup>
12 b=78.8*10^-6//Coefficient of compressibility in cm
      ^2/kg when pressure is taken as atmospheric
13 dp=0.75//Diameter of the plunger in cm
14
15 // Calculations
16 V1 = (Vf + (3.14/4) * (di/10)^2 * 1 + V) / Initial volume in c.
17 dV = ((b*V1*(p-pp)/patm)) / Change in volume in c.c.
18 d=(dV+Vd)//Total displacement of the plunger in c.c
19 s=((4/3.14)*(d/dp^2))*10//Stroke in mm
20
```

```
21 //Output
22 printf('The effective plunger stroke is %3.1 f mm',s)
```

Scilab code Exa 15.9 DIAMETER

```
1 clc
2 clear
3 //Input data
4 n=6//Number of cylinders
5 p=300//Horse power in H.P
6 \text{ N=} 1200 // \text{Speed in r.p.m}
7 f=0.2//Fuel rate in kg per B.H.P hour
8 ip=200//Injection pressure in kg/cm<sup>2</sup>
9 cp=40//Pressure in the combustion chamber in kg/cm<sup>2</sup>
10 pic=33//Period of injection of the crank angle in
      degrees
11 g=0.83//Specific gravity of fuel. In textbook, it is
       given wrong as 0.89
12 Cd=0.9//Coefficient of discharge
13
14 //Output
15 Fc=(p*f)//Fuel consumption per hour in kg/hr
16 Fcy=(Fc/n)//Fuel consumption per cylinder in kg/hr
17 Fcyc=(Fcy/(60*(N/2)))/Fuel consumption per cycle in
18 q=(Fcyc/(g*1000))*10^6/Quantity of fuel injected
      per cylinder per cycle in c.c
19 I=((pic/360)*(1/N)*60)//Injection period in sec
20 df = (g/1000) / Density of fuel in kg/m<sup>3</sup>
v=sqrt(2*981*((ip-cp)/df))/Velocity of fuel through
       orifice in m/s
22 A=(q/(Cd*v*I))/Area of orifice in cm<sup>2</sup>
23 d = sqrt(A/(3.14/4))*10//Diameter in mm
24
25 //Output
```

26 printf('The diameter of the single orifice injector is %3.2 f mm',d)

Scilab code Exa 15.10 AMOUNT OF FUEL

```
1 clc
2 clear
3 //Input data
4 n=6//Number of cylinders
5 d=11.5//Bore in cm
6 l=14//Stroke in cm
7 af=16//Air fuel ratio
8 pa=1.03//Pressure of air intake in kg/cm<sup>2</sup>
9 Ta=24+273//Temperature of air intake in K
10 nv=76.5//Volumetric efficiency in percent
11 R=29.27//Characteristic gas constant in kg.m/kg.K
12 N=1500/Speed in r.p.m
13 ip=125//Injection pressure in kg/cm<sup>2</sup>
14 cp=40//Compression pressure in kg/cm<sup>2</sup>
15 q=18.5//Fuel injection occupies 18.5 degrees of
      crenk travel
16 fsw=760//Fuel specific weight in kg/m<sup>2</sup>
17 dc=0.94//Orifice discharge coefficient
18
19 // Calculations
20 Vs = ((3.14/4)*d^2*1) / Stroke volume in c.c
21 Va=(Vs*(nv/100))/Volume of air supplied in c.c.
22 wa=((pa*10^4*Va*10^-6)/(R*Ta))/Weight of air
      supplied per cylinder per cycle in kg
23 wf=(wa/af)//Weight of fuel injected per cylinder per
       cycle in kg
24 I=((60*q)/(N*360))/Injection time per cycle in sec
25 F=(wf/I)//Fuel injected per cylinder per sec in kg/
26 Af=(F/(dc*sqrt(2*9.81*fsw*(ip-cp)*10^4)))/Area of
```

```
orifice in sq.m
27 df=sqrt(Af/(3.14/4))*1000//Diameter of orifice in mm
28
29 //Output
30 printf('Maximum amount of fuel injected per cylinder per sec is %3.2 f kg/sec \n Diameter of orifice is %3.3 f mm', F, df)
```

Chapter 16

COMBUSTION IN COMPRESSION IGNITION ENGINES

Scilab code Exa 16.1 DELAY TIME

```
1 clc
2 clear
3 //Input data
4 s=0.005//Delay in sec
5 d=30//Bore in cm
6 \text{ N=}600/\text{Speed in r.p.m}
7 dx=[10,15,20]//Bore diameters in cm
9 // Calculations
10 t=(s/d)*dx//Time of delay in sec. In textbook, t(2)
      is given wrong as 0.00025 sec instead of 0.0025
      sec
11
12 //Output
13 printf ('The delay time for %i cm diameter bore is %3
      .5f sec \n The delay time for %i cm diameter bore
       is %3.5f sec \n The delay time for %i cm
```

```
diameter bore is \%3.5 \, \text{f} \, \sec^{\prime}, dx(1), t(1), dx(2), t(2), dx(3), t(3))
```

Scilab code Exa 16.2 PRESSURE

```
1 clc
2 clear
3 //Input data
4 d=[15,60]/Bore in cm
5 N = [1600, 400] / Speed in r.p.m respectively
6 q=30//Injection of oil occupies 30 degrees of crank
      travel in each case
7 pc=30//Compression pressure in kg/cm<sup>2</sup>
8 d=0.001//Delay time in sec
9 rp=5//Rapid combustion period is 5 degree of crank
      travel
10 pe=60//Compression pressure at the end of rapid
      compression in kg/cm<sup>2</sup>
11
12 // Calculations
13 //For small engine
14 It1=(60/N(1))*(q/360)//Injection time in sec
15 pf1=((d/It1)+(rp/pc))*100//Percent fuel
16 //For large engine
17 It2=(60/N(2))*(q/360)//Injection time in sec
18 pf2=((d/It2)+(rp/pc))*100//Percent fuel
19 pr=(pc*(pf2/pf1))/Pressure rise in kg/cm^2
20 mp=(pc+pr)//Maximum pressure in kg/cm<sup>2</sup>
21
22 //Output
23 printf ('Pressure in the large engine is \%3.1 f kg/cm
      ^2, mp)
```

Scilab code Exa 16.3 BRAKE THERMAL EFFICIENCY

```
1 clc
2 clear
3 //Input data
4 n=4//Number of cylinders
5 d=105//Bore in mm
6 l=127//Stroke in mm
7 BHP=63//Brake horse power in h.p
8 N=1800/Speed in r.p.m
9 t=15//Test time in min
10 mf = 2.75 //Mass of fuel in kg
11 CV=11000//Calorific value in kcal/kg
12 \text{ af} = 14.8 // \text{Air fuel ratio}
13 v=0.805//Specific volume in m^3/kg
14 nv=80//Volumetric efficiency in percent
15 J=427 // Mechanical equivalent of heat in kg.m/kcal
16
17 // Calculations
18 bth=((BHP*4500)/(J*(mf/t)*CV))*100//Brake thermal
      efficiency in percent
19 Vs = ((3.14/4)*(d/10)^2*(1/10)) / Stroke volume in c.c
20 Vsw = (Vs*n*(N/2)*t) / Swept volume in c.c.
21 Va=(Vsw*10^-6*(nv/100))/Volume of air sucked in m^3
22 wa=(Va/v)//Weight of air sucked in kg
23 wr=(af*mf)//Weight of air reqired in kg
24 pei=(wr/wa)*100//Percentage of air available for
      combustion
25
26 //Output
27 printf ('Brake thermal efficiency is %3.1f percent \n
      The percentage of air used for combustion is %i
      percent', bth, pei)
```

Chapter 18

SUPERCHARGING

Scilab code Exa 18.1 PRESSURE AND AIR CONSUMPTION

```
1 clc
2 clear
3 //Input data
4 n=6//Number of cylinders
5 d=9//Bore in cm
6 l=10//Stroke in cm
7 N=2500/Speed in r.p.m
8 Ta=25+273//Temperature of air entering the
     compressor in K
9 q=16800//Heat rate in kcal/hour
10 T=60+273 // Temperature of air leaving the cooler in K
11 p=1.6//Pressure of air leaving the cooler in kg/cm<sup>2</sup>
12 t=14.5//Engine torque in kg.m
13 nv=75//Volumetric efficiency in percent
14 nm=74//Mechanical efficiency in percent
15 R=29.27 // Characteristic gas constant in kg.m/kg.K
16 Cp=0.24//Specific heat at constant pressure n kcal/
     kg.K
17
18 // Calculations
19 BHP=(2*3.14*N*t)/4500//Brake horse power in B.H.P
```

```
20 IHP=(BHP/(nm/100))//Indicated horse power in I.H.P
21 pm=((IHP*4500)/((1/100)*(3.14/4)*d^2*(N/2)*n))/Mean
       effective pressure in kg/cm<sup>2</sup>
22 Vs = (n*(3.14/4)*(d/100)^2*(1/100)*(N/2))/Swept
      volume in m<sup>3</sup>/min
23 Va=(Vs*(nv/100))//Aspirated Volume of air into
      engine in m<sup>3</sup>/min
24 ma=(p*10^4*Va)/(R*T)/Aspirated mass flow into the
      engine in kg/min
  mcdT=((BHP*4500/427)/Cp)//Product of mass flow rate
      and change in temperature
26 msdT = ((q/60)/Cp)//Product of mass flow rate and
      change in temperature
27 \text{ x=(mcdT/msdT)//Ratio}
28 T2=((Ta-(x*T)))/(1-x)//Temperature in K
29 mc = (msdT/(T2-T))//Air flow in kg/min
30
31 //Output
32 printf('(a) the mean effective pressure is \%3.2 f kg/
      cm<sup>2</sup> \n (b) the air consumption is \%3.3 \,\mathrm{f} \,\mathrm{kg/min} \
      n (c) the air flow into the compressor is \%3.2 f
      kg/min',pm,ma,mc)
```

Scilab code Exa 18.2 PERCENTAGE INCREASE IN IMEP

```
compression in kg/cm<sup>2</sup>
8 T1=100+273//Charge temperature at the beginning of
      compression in K
9 pm=0.91//Mean pressure during the conduction stroke
     in kg/cm<sup>2</sup>
10 bn=70//Blower adiabatic efficiency in percent
11 T2=50//Temperature of the charge after delivery by
      the blower in degree C
12 dp=0.07 // Pressure drop in kg/cm<sup>2</sup>
13 pi=1.47//Charge pressure in the cylinder during the
      induction stroke in kg/cm<sup>2</sup>
14 Ta=15+273 // Atomspheric temperature in K
15 pa=1.03//Atmospheric pressure in kg/cm<sup>2</sup>
16 g=1.4//Ratio of specific heats
17
18 // Calculations
19 T2x = ((((pi/pa)^((g-1)/g)-1)/(bn/100))+1)*Ta+T2//
      Temperature in K
20 rIMEP=((pi/pa)*(T1/T2x))/Ratio of I.M.E.P
21 gIMEP=(rIMEP*IMEP)//Gross I.M.E.P in kg/cm^2
22 nsIMEP=(gIMEP+(pi-pa))//Net I.M.E.P supercharged in
     kg/cm^2
23 nuIMEP=(IMEP-pIMEP)//Net I.M.E.P unsupercharged in
     kg/cm^2
24 iIMEP=(nsIMEP-nuIMEP)//Increase in I.M.E.P in kg/cm
25 pei=(iIMEP*100)/nuIMEP//Percentage increase
26
27 //Output
28 printf ('Percentage increase in the net I.M.E.P due
      to supercharging is %3.1f percent', pei)
```

Scilab code Exa 18.3 PERCENTAGE INCREASE IN BHP

1 clc

```
2 clear
3 //Input data
4 1=4.5//Capacity in litres
5 P=20//Power in H.P per m<sup>3</sup> of free air induced per
      minute
6 N=1700/Speed in r.p.m
7 nv=75//Volumetric efficiency in percent
8 Ta=27+273//Atomspheric temperature in K
9 pa=1.03//Atmospheric pressure in kg/cm<sup>2</sup>
10 pr=1.75//Pressure ratio
11 ie=70//Isentropic efficiency in percent
12 nm=75//Mechanical efficiency in percent
13 g=1.4//Ratio of specific heats
14 nb=80//Efficiency of blower in percent
15 R=29.27 // Characteristic gas constant in kg.m/kg.K
16 Cp=0.24//Specific heat at constant pressure in kJ/kg
      .K
  J=427 // Mechanical equivalent of heat in kg.m/kcal
17
18
19 // Calculations
20 Vs=(1/1000*(N/2))/Swept volume in m<sup>3</sup>/min
21 uVs=((nm/100)*Vs)//Unsupercharged swept volume in m
      ^3/\min
22 dp=(pr*pa)//Blower delivery pressure in kg/cm<sup>2</sup>
23 Tc=(Ta*pr^((g-1)/g))//Temperature after isentropic
      compression in K
24 dT=(Ta+(Tc-Ta)/(ie/100))//Blow delivery temperature
      in K
  Va=(Vs*(dp*Ta)/(pa*dT))//Equivalent volume at free
      air condition in m<sup>3</sup>/min
  iiv=(Va-uVs)//Increase in the induced volume in m<sup>3</sup>/
26
      min
27 iIHP=(P*iiv)//ncrease in I.H.P
28 iBHP=(iIHP*(nm/100))//Increase in B.H.P
29 ma=(dp*10^4*Vs)/(R*dT)/Mass of air delivered by
      blower in kg/min
30 HP=(ma*Cp*(dT-Ta)*J)/(4500*(80/100))/H.P. required
      for blower
```

Chapter 19

TWO STROKE ENGINES

Scilab code Exa 19.3 PROPERTIES

```
1 clc
2 clear
3 //Input data
4 d=11.25//Bore in cm
5 l=15//Stroke in cm
6 r=7//Compression ratio
7 N=1800/Speed in r.p.m
8 a=4.5//Air supply in kg/min
9 Ta=72+273//Temperature of air in K
10 af=14.3//Air fuel ratio
11 ep=1//Exhaust pressure in kg/cm<sup>2</sup>
12 R=29.27 // Characteristic gas constant in kg.m/kg.
      degree C
13
14 // Calculations
15 Vc = ((r/(r-1))*(3.14/4)*(d/100)^2*(1/100))/Swept
      volume in m<sup>3</sup>
16 Wa=(Vc*N*ep*10^4)/(R*Ta)//Ideal air capacity in kg/
      min
17 sr=(a/Wa)//Scavenging ratio
18 \operatorname{sn}=(1-\exp(-\operatorname{sr}))//\operatorname{Scavenging} efficiency
```

```
19 nt=(sn/sr)//Trapping efficiency
20
21 //Output
22 printf('(a) Ideal air capacity is %3.2 f kg/min \n (b
      ) Scavenging ratio is %3.2 f \n (c) Scavenging
      efficiency is %3.3 f \n (d) Trapping efficiency is
      %3.2 f', Wa, sr, sn, nt)
```

Chapter 23

TESTING OF ENGINES

Scilab code Exa 23.1 THERMAL EFFICIENCY

```
1 clc
2 clear
3 //Input data
4 n=4//Four cylinder engine
5 N=1200/Speed in r.p.m
6 BHP1=26.3//Brake horse power in B.H.P
7 T=11.3//Average torque in kg
8 CV=10000//Calorific value of the fuel in kcal/kg
9 m=270//Flow rate in gm of petrol per B.H.P hour
10
11 // Calculations
12 BHP2=(T*2*3.14*N)/4500//Average B.H.P on 3 cylinders
13 IHP=BHP1-BHP2//Average I.H.P of one cylinder
14 TIHP = (n*IHP) // Total I.H.P
15 p=((m/1000)*BHP1)/TIHP//Petrol used in kg/I.H.P hr
16 nth=((4500*60)/(427*p*CV))*100//Indicated Thermal
      efficiency in percent
17
18 //Output
19 printf('Thermal efficiency is %3.1f percent',nth)
```

Scilab code Exa 23.2 INDICATED HORSE POWER

```
1 clc
2 clear
3 n=4//Four cylinder engine
4 d=0.1//Diameter of piston in m
5 l=0.15//Stroke in m
6 RPM=1600/Speed in r.p.m
7 ap=(5.76*10^-4)//Area of positive loop of indicator
     diagram in sq.m
  an=(0.26*10^-4)//Area of negative loop of indicator
     diagram in sq.m
9 L=0.055//Length of the indicator diagram in m
10 k=(3.5/10^-6)/Spring constant in kg/m^2 per m
11
12 // Calculations
13 NA=(ap-an)//Net area of the indicator diagram in sq.
14 h=(NA/L)//Average height of diagram in m
15 Pm=(h*k)//Mean effective pressure in kg/m<sup>2</sup>
16 IHP=(Pm*1*(3.14/4)*d^2*RPM*n)/4500//Indicated Horse
     Power
17
18 //Output
19 printf ('Indicated horse power of a four cylinder two
      stroke petrol engine is %3.1f', IHP)
```

Scilab code Exa 23.3 CALORIFIC VALUE

```
1 clc
2 clear
3 //Input data
```

```
4 n=6//Number of cylinders
5 d=0.089 //Bore in m
6 l=0.1016//Stroke in m
7 vc=3.183//Compression ratio
8 rn=55//Relative efficiency in percent
9 m=0.218//Petrol consumption in kg/hp.hr
10 Pm=(8.4/10^-4)//Indicated mean effective pressure in
      kg/m^2
11 N=2500/Speed in r.p.m
12
13 // Calculations
14 an=(1-(1/(vc-1)))*100//Air standard efficiency in
     percent
15 nth=(rn*an)/100//Thermal efficiency in percent
16 CV = ((4500*60) / (m*(nth/100)*427)) / Calorific value in
17 IHP=((Pm*(3.14/4)*d^2*l*N*n)/(4500*2))/Indicated
     horse power
18 p=(m*IHP)//Petrol consumption in kg/hour
19
20 //Output
21 printf('(1) The calorific value of petrol is %i kcal
     /kg \n (2) Corresponding petrol consumption is \%3
     .1 f kg/hour', CV,p)
```

Scilab code Exa 23.4 BHP OF ENGINE

```
1 clc
2 clear
3 //Input data
4 n=4//Number of cylinders
5 d=0.2//Bore in m
6 l=0.3//Stroke in m
7 N=300//Speed in r.p.m
8 af=5//Air to fuel ratio by volume. In textbook it is
```

```
given as 4 which is wrong
9 nv=78//Volumetric efficiency in percent
10 CV=2200//Calorific value in kcal/cu.m at N.T.P
11 bth=23//Brake thermal efficiency in percent
12
13 // Calculations
14 Vs = ((3.14/4)*d^2*1) / Swept volume in cu.m
15 c=((nv/100)*Vs)//Total charge per stroke in cu.m
16 Vg=((c/af)*N)//Volume of gas used per min in cu.m at
      N.T.P
17 q=(CV*Vg)//Heat supplied in kcal/min
18 BHP=((bth/100)*q)/(4500/427)//Brake horse power
19
20 //Output
21 printf ('The volume of gas used per min is \%3.3 f cu.m.
       at N.T.P \n B.H.P of engine is \%3.1 f', Vg, BHP)
```

Scilab code Exa 23.5 THERMAL EFFICIENCY

```
1 clc
2 clear
3 //Input data
4 d=20/Bore in cm
5 1=37.5//Stroke in cm
6 r=6//Compression ratio
7 IPm=5//Indicated Mean effective pressure in kg/cm<sup>2</sup>
8 ag=6//Air to gas ratio
9 CV=3070//Calorific value of gas in kcal/cu.m
10 T=75+273//Temperature in K
11 p=0.975//Pressure in kg/cm^2
12 RPM=240//Speed in r.p.m
13 g=1.4//Ratio of specific heats
14
15 // Calculations
16 Vs = (3.14/4)*d^2*1//Stroke Volume in cu.m
```

```
17 Vg = (1/(r+1))*Vs//Volume of gas in cylinder in cu.m
      per cycle
18 x = (Vg*(p/1.03)*(273/T)) / Volume at 'Vg' cu.m at 'p'
     kg/cm<sup>2</sup> and 'T' K are equivalent in cu.m
19 q=(CV*x)/10^6//Heat added in kcal per cycle
20 IHP=(IPm*(Vs/100)*(RPM/2))/4500//Indicated horse
     power
21 nth = ((IHP*4500)/(427*q*(RPM/2)))*100//Thermal
      efficiency in percent
22 na=(1-(1/r^{(g-1))})*100/Air standard efficiency in
      percent
23 rn=(nth/na)*100//Relative effeciency in percent
24
25 // Output
26 printf ('The thermal efficiency is \%3.1f percent \n
     The relative efficiency is %3.1f percent \n
      Indicated horese power is %3.1f H.P', nth, rn, IHP)
```

Scilab code Exa 23.6 FUEL CONSUMPTION

```
1 clc
2 clear
3 //Input data
4 n=4//Number of cylinders
5 d=6.25//Diametre in cm
6 l=9.5//Stroke in cm
7 t=678//Torque in kg.m
8 N=3000//Speed in r.p.m
9 Vc=60//Clearance volume in c.c
10 be=0.5//Brake efficiency ratio based on the air standard cycle
11 CV=10000//Calorific value in kcal/kg
12 g=1.4//Ratio of specific heats
13
14 //Calculations
```

```
15 Vs=(3.14/4)*d^2*1//Stroke volume in c.c per cylinder
16 r=((Vs+Vc)/Vc)//Compression ratio
17 na=(1-(1/r^(g-1)))//Air standard efficiency
18 bth=(be*na)*100//Brake thermal efficiency in percent
19 bhp=((t/100)*2*3.14*N)/4500//B.H.P in H.P
20 q=(bhp*(4500/427))/(bth/100)//Heat supplied in kcal/min
21 F=(q*60)/CV//Fuel consumption in kg/hour
22 P=(bhp*4500*2*100)/(n*Vs*N)//pressure in kg/cm^2
23
24 //Output
25 printf('The fuel consumption is %3.2 f kg/hour \n The brake mean effective pressure is %3.2 f kg/cm^2', F,P)
```

Scilab code Exa 23.7 PROPERTIES

```
1 clc
2 clear
3 //Input data
4 n=1//Number of cylinders
5 t=30//Trail time in min
6 m=5.6//Oil consumption in 1
7 CV=9980 // Calorific value of oil in kcal/kg
8 g=0.8//Specific gravity of oil
9 a=8.35//Average area of indicator diagram in sq.cm
10 1=8.4//Length of the indicator diagram in cm
11 is=5.5//Indicator spring scale
12 L=147.5//Brake load in kg
13 sp=20//Spring balance reading in kg
14 d=1.5//Effective brake wheel diameter in m
15 N=200/Speed in r.p.m
16 cyd=30 // Cylinder diameter in cm
17 \ 11=45//Stroke in cm
18 mw=11//Jacket cooling water in kg/min
```

```
19 Tc=35+273//Temperature rise of cooling water in K
20
21 // Calculations
22 mp=(a/1)*is//Mean effective pressure
23 ihp=((mp*(11/100)*(3.14/4)*cyd^2*(N/2))/4500)//
      Indicated horse power in h.p.
24 bhp=(L*3.14*d*N)/4500//Brake horse power in h.p
25 nm=(bhp/ihp)*100//Mechanical efficiency in percent
26 F=(m*(60/t)*g)//Fuel consumption in kg/hour
27 Fc=(F/bhp)//Specific fuel consumption in kg/B.H.P/
      hour
  ith = ((ihp*(4500/427))/((F/60)*CV))*100//Indicated
      thermal efficiency in percent
29
30 //Output
31 printf('(a) I.H.P is \%3.1 f \setminus n (b) B.H.P is \%3.1 f \setminus n
      (c) Mechanical efficiency is \%3.1f percent \n (d)
       Specific fuel consumption is \%3.2 f kg/B.H.P/hour
       \n (e) Indicated thermal efficiency is \%3.1f
      percent', ihp, bhp, nm, Fc, ith)
```

Scilab code Exa 23.8 HEAT BALANCE SHEET

```
12 db=1.78//Diameter of the brake wheel drum in m
13 dr=4//Diameter of rope in cm
14 cw=545//Cooling water circulated in litres
15 Tc=45 // Cooling water temperature rise in degree C
16 g=0.8//Specific gravity of oil
17
18 // Calculations
19 ihp=((imep-pimep)*(1/100)*3.14*d^2*n)/(4500*60)//I.H
     .P in h.p
20 q=(f*g*CV)/60//Heat supplied in kcal/min
21 bhp=(L*3.14*(db+(dr/100))*n)/(4500*60)/B.H.P in h.p
22 qbhp=(bhp*4500)/427//Heat equivalent of B.H.P in
     kcal/min
23 qw=(cw*Tc)/60//Heat lost to jacket cooling water in
     kcal/min
24 dq=(q-(qbhp+qw))//Heat unaccounted in kcal/min
25
26 //Output
27 printf ('Heat supplied is %3.0f kcal/min \n Heat
      equivalent of B.H.P is \%3.0f kcal/min \n Heat
     lost to jacket cooling water is \%3.0f kcal/min \n
      Heat unaccounted is \%3.0 f kcal/min',q,qbhp,qw,dq
     )
```

Scilab code Exa 23.9 HEAT BALANCE SHEET

```
1 clc
2 clear
3 //Input data
4 d=27//Diameter in cm
5 l=45//Stroke in cm
6 db=1.62//Effective diameter of the brake in m
7 t=(38*60+30)//Test duration in sec
8 CV=4650//Calorific value in kcal/m^3 at N.T.P
9 n=8080//Total no. of revolutions
```

```
10 en=3230 // Total number of explosions
11 p=5.75//Mean effective pressure in kg/cm<sup>2</sup>
12 V=7.7/Gas used in m^3
13 T=15+273 // Atmospheric temperature in K
14 pg=135//pressure of gas in mm of water above
      atmospheric pressure
15 hb=750//Height of barometer in mm of Hg
16 L=92//Net load on brake in kg
17 w=183//Weigh of jacket cooling water in kg
18 Tc=47 // Cooling water temperature rise in degree C
19
20 // Calculations
21 ihp=(p*(1/100)*(3.14/4)*d^2*en)/(4500*(t/60))/I.H.P
       in h.p
22 bhp=(L*3.14*db*n)/(4500*(t/60))/B.H.P in h.p
23 pa=(hb+(pg/13))//Pressure of gas supplied in mm of
     Hg
V_g = (V * (273/T) * (pa/760)) / V_{olume} \text{ of gas used at N.T.P}
       in m^3
25 q=(Vg*CV)/(t/60)//Heat supplied per minute in kcal
26 qbhp=(bhp*4500)/427//Heat equivalent of B.H.P in
      kcal/min
27 qc = (w/(t/60))*Tc//Heat lost to jacket cooling water
     in kcal/min
28 qra=(q-(qbhp+qc))//Heat lost to exhaust, etc in kcal
     /min
29
30 //Output
31 printf('Heat supplied is %3.1f kcal/min \n Heat
      equivalent of B.H.P is %3.0f kcal/min \n Heat
      lost to jacket cooling water is \%3.1f kcal/min \n
      Heat lost to exhaust radiation etc. is %3.1f
      kcal/min',q,qbhp,qc,qra)
```

Scilab code Exa 23.10 VOLUMETRIC EFFICIENCY

```
1 clc
2 clear
3 //Input data
4 d=25//Bore in cm
5 1=50//Stroke in cm
6 \text{ N=}240 // \text{Speed in r.p.m}
7 n=100//Number of times fires per minute
8 qc=0.3//Quantity of coal gas used in cu.m per minute
9 h=100//Head in mm of water
10 bp=1.03//Barometric pressure in kg/cm<sup>2</sup>
11 T=15+273//Temperature in K
12 ma=2.82//Mass of air used in kg per minute
13 R=29.45 // Characteristic gas constant in kg.m/kg.K
14
15 // Calculations
16 gp=(bp+(100/13.6)*(bp/76))/Gas pressure in kg/cm<sup>2</sup>
17 Vc = (qc*(gp/bp)*(273/T))/Volume of coal gas at N.T.P
       in cu.m per minute
18 Vce=(Vc/n)//Volume of coal gas per explosion in cu.m
       at N.T.P
19 va=(ma*R*273)/(bp*10^4)/Volume of air taken in at N
      .T.P in cu.m per min
20 V = ((va - (((N/2) - n) * Vce)) / (N/2)) / Volume in cu.m
21 tV=(V+Vce)//Total volume of charge in cu.m at N.T.P
22 Vs = ((3.14/4)*(d^2*1)*10^-6) / Swept volume in cu.m
23 nv=(tV/Vs)*100//Volumetric efficiency in percent
24
25 //Output
26 printf('(a) the charge of air per working cycle as
      measured at N.T.P is %3.5f cu.m \n (b) the
      volumetric efficiency is %3.1f percent',tV,nv)
```

Scilab code Exa 23.11 HEAT BALANCE SHEET

1 clc

```
2 clear
3 //Input data
4 d=18//Diameter in cm
5 1=24//Stroke in cm
6 t=30//Duration of trail in min
7 r=9000//Total number of revolutins
8 e=4445//Total number of explosions
9 mep=5.85//Mean effective pressure in kg/cm<sup>2</sup>
10 N1=40//Net load on brake wheel in kg
11 ed=1//Effective diameter of brake wheel in meter
12 tg=2.3//Total gas used at N.T.P in m<sup>3</sup>
13 CV=4600//Calorific value of gas in kcal/m<sup>3</sup> at N.T.P
14 ta=36//Total air used in m<sup>3</sup>
15 pa=720 // Pressure of air in mm of Hg
16 Ta=18+273 // Temperature of air in K
17 da=1.293//Density of air at N.T.P in kg/m<sup>3</sup>
18 Te=350+273//Temperature of exhaust gases in K
19 Tr=18+273//Room temperature in K
20 Cp=0.24//Specific heat of exhaust gases in kJ/kg.K
21 twc=81.5//Total weight of cylinder jacket cooling
     water in kg
22 dT=33//Rise in temperature of jacket cooling water
     in degree C
23 R=29.45//Characteristic gas constant in kg.m/kg.
     degree C
24
25
  // Calculations
26 ihp=(mep*(1/100)*(3.14/4)*d^2*(e/t))/4500/Indicated
      horse power in h.p
  bhp=(N1*3.14*r*ed)/(4500*t)/Brake horse power in h.
27
28
  qs=(tg/t)*CV//Heat supplied at N.T.P in kcal
29 qbhp=(bhp*4500)/427//Heat equivalent of B.H.P in
     kcal/min
30 ql=(twc/t)*dT//Heat lost to cylinder jacket cooling
     water in kcal/min
31 VA=(ta*(273/Ta)*(pa/760))/Volume of air used at N.T
      .P in m^3
```

```
32 WA=(VA*da)/t//Weight of air used per min in kg
33 WG = (1.03 * tg * 10^4) / (R * 273) / Weight of gas in kg
34 Wg=(WG/t)//Weight of gas per minute in kg
35 We=(WA+Wg)//Total weight of exhaust gases in kg
36 qle=(We*(Te-Tr)*Cp)//Heat lost of exhaust gases in
      kcal/min
37 qra=(qs-(qbhp+ql+qle))//Heat lost by radiation in
      kcal/min
38 nm=(bhp/ihp)*100//Mechanical efficiency in percent
39 ith=((ihp*4500)/(427*qs))*100//Indicated thermal
      efficiency in percent
40
41 // Output
42 printf('
                       HEAT BALANCE SHEET\n
                          — \n\n Heat supplied per minute
      is \%3.1 f kcal/min \n Heat expenditure
                       (kcal per minute) \n 1. Heat
      equivalent of B.H.P is
                                            \%3.1 f \setminus n 2. Heat
      lost to jacket cooling water is %3.1 f \n 3. Heat
                                            \%3.1 \, \text{f} \setminus \text{n} \quad 4. \, \text{Heat}
      lost in exhaust gases is
      lost by radiation, etc., is
                                            \%3.1 \,\mathrm{f} \,\, \backslash \mathrm{n}
                                                   \%3.1 \text{ f} \n
      ,qbhp,ql,qle,qra,qs)
```

Scilab code Exa 23.12 HEAT BALANCE SHEET

```
1 clc
2 clear
3 //Input data
4 gu=0.135//Gas used in m^3/min at N.T.P
5 CV=3990//Calorific value of gas in kcal/m^3 at N.T.P
6 dg=0.64//Density of gas in kg/m^3 at N.T.P
7 au=1.52//Air used in kg/min
```

```
8 C=0.24//Specific heat of exhaust gases in kJ/kg.K
9 Te=397+273//Temperature of exhaust gases in K
10 Tr=17+273//Room temperature in K
11 cw=6//Cooling water per minute in kg
12 rT=27.5//Rise in temperature in degree C
13 ihp=12.3//Indicated horse power in h.p
14 bhp=10.2//Brake horse power in h.p
15
16 // Calculations
17 qs=(gu*CV*60)//Heat supplied in kcal/hour
18 qbhp = ((bhp*4500*60)/427)/Heat equivalent of B.H.P
     in kcal/hr
19 ql=(cw*60*rT)//Heat lost in jacket cooling water in
     kcal/hr
20 mg=(gu*dg)//Mass of gas used per minute in kg
21 me=(mg+au)//Mass of exhaust gases per minute in kg
22 qe = (me * C * (Te - Tr) * 60) / Heat carried away by exhaust
     gases in kcal/hour
23 qun=(qs-(qbhp+ql+qe))//Heat unaccounted in kcal/hour
24
25 // Output
26 printf ('Heat supplied is %3.0 f kcal/hour \n Heat
      equivalent of B.H.P is %3.0f kcal/hr \n Heat lost
      in jacket cooling water is \%3.0 f kcal/hr \n Heat
       carried away by exhaust gases is \%3.0 f kcal/hour
       \n Heat unaccounted is \%3.0f kcal/hour', qs, qbhp,
     ql,qe,qun)
```

Scilab code Exa 23.13 BRAKE MEAN EFFECTIVE PRESSURE

```
1 clc
2 clear
3 //Input data
4 n=4//Number of cylinders
5 r=1//Radius in metre
```

```
6 \text{ N=1400//Speed in r.p.m}
7 bl=14.5//Net brake load in kg
8 P=[9.8,10.1,10.3,10]/Loads on the brake in kg
9 d=9/Bore in cm
10 l=12//Stroke in cm
11
12 // Calculations
13 bhp=(bl*2*3.14*r*N)/4500//Brake horse power in h.p.
14 bhp1=(P(1)*2*3.14*r*N)/4500//Brake horse power in h.
15 bhp2=(P(2)*2*3.14*r*N)/4500//Brake horse power in h.
16 bhp3=(P(3)*2*3.14*r*N)/4500//Brake horse power in h.
17 bhp4=(P(4)*2*3.14*r*N)/4500//Brake horse power in h.
18 ihp1=bhp-bhp1//Indicated horse power in h.p.
19 ihp2=bhp-bhp2//Indicated horse power in h.p.
20 ihp3=bhp-bhp3//Indicated horse power in h.p.
21 ihp4=bhp-bhp4//Indicated horse power in h.p.
22 ihp=(ihp1+ihp2+ihp3+ihp4)//Indicated horse power in
23 nm=(bhp/ihp)*100//Mechanical efficiency in percent
pm = ((4500*bhp)/((1/100)*(3.14/4)*d^2*(N/2)))/Brake
     mean effective pressure in kg/cm<sup>2</sup>
25
26 //Output
27 printf('I.H.P is %3.1f h.p \n Mechanical efficiency
      is %3.1f percent \n Brake mean effective pressure
       is \%3.0 \, \text{f} \, \text{kg/cm}^2, ihp,nm,pm)
```

Scilab code Exa 23.14 HEAT BALANCE SHEET

```
1 clc
2 clear
```

```
3 //Input data
4 N=350/Speed in r.p.m
5 L=60//Net brake load in kg
6 mep=2.75//Mean effective pressure in kg/cm<sup>2</sup>
7 oc=4.25//Oil consumption in kg/hour
8 jcw=490//Jacket cooling water in kg/hour
9 Tw=[20+273,45+273] //Temperature of jacket water at
     inlet and outlet in K
10 au=31.5//Air used per kg of oil in kg
11 Ta=20+273//Temperature of air in the test room in K
12 Te=390+273 // Temperature of exhaust gases in K
13 d=22//Cylinder diameter in cm
14 l=28//Stroke in cm
15 bd=1//Effective brake diameter in m
16 CV=10500//Calorific value of oil in kcal/kg
17 pH2=15//Proportion of hydrogen in fuel oil in
     percent
18 C=0.24//Mean specific heat of dry exhaust gases
19 Cs=9.5//Specific heat of steam in kJ/kg.K
20
21 // Calculations
22 ibp=(mep*(1/100)*(3.14/4)*d^2*N)/4500//Indicated
     brake power in h.p
23 bhp=(L*3.14*N*bd)/4500/Brake horse power in h.p.
24 qs=(oc*CV)/60//Heat supplied per minute in kcal
25 qbhp=(bhp*4500)/427//Heat equivalent of B.H.P in
     kcal/min
26 pqbhp=(qbhp/qs)*100//Percenatge of heat
27 ql = (jcw/60)*(Tw(2)-Tw(1))/Heat lost to cooling
     water in kcal/min
28 pql=(ql/qs)*100//Percenatge of heat
29 wH20=(9*(pH2/100)*(oc/60))/Weight of H2O produced
     per kg of fuel burnt in kg/min
30 twe=(oc*(au+1))/60//Total weight of wet exhaust
     gases per minute in kg
31 twd=(twe-wH20)//Weight of dry exhaust gases per
     minute in kg
32 qle=(twd*C*(Te-Ta))/Heat lost to dry exhaust gases/
```

```
min in kcal
33 pqle=(qle/qs)*100//Percenatge of heat
34 \text{ qx} = (100+538.9+0.5*(Te-373))/\text{Heat in kcal/kg}
35 qst=(wH2O*qx)//Heat to steam in kcal/min
36 pqst=(qst/qs)*100//Percenatge of heat
37 qra=(qs-(qbhp+ql+qle+qst))//Heat lost by radiation
       in kcal/min
38 pqra=(qra/qs)*100//Percenatge of heat
39
40 //Output
41 printf('
                          HEAT BALANCE SHEET\n
                            --- \n\n Heat supplied per minute
                    \%3.0 \, \text{f} \, \text{kcal/min} \, 100 \, \text{percent} \, \ln n \, \text{Heat}
                                        (kcal per minute)
       expenditure
       percent) \n 1. Heat equivalent of B.H.P is
                                      \%3.1 \, f \setminus n \, 2. \, \text{Heat lost to}
                        \%3.1 \text{ f}
       cooling water is
                                            \%3.0 \text{ f}
                                                              \%3.1 \,\mathrm{f} \, \setminus \mathrm{n}
        3. Heat lost to dry exhaust gases is
                                                                 %3.1 f
               %3.1 f\n 4. Heat lost of steam in exhaust
       gases is %3.0 f
                                       \%3.1 \text{ f} \text{ n} 5. Heat lost by
       radiation, etc., is
                                            \%3.0 \text{ f}
                                                              \%3.1 \,\mathrm{f} \,\, \backslash \mathrm{n}
                                            Total
                                               \%3.0 \text{ f}
                                                                %3.0 f
       \backslash n
                         -----',qs,qbhp,pqbhp,ql,pql,qle,pqle
       ,qst,pqst,qra,pqra,qs,(pqbhp+pql+pqle+pqst+pqra))
```

Scilab code Exa 23.15 HEAT BALANCE SHEET

```
1 clc
2 clear
3 //Input data
4 d=20//Diameter in cm
5 l=40//Stroke in cm
```

```
6 mep=5.95//Mean effective pressure in kg/cm<sup>2</sup>
7 bt=41.5//Brake torque in kg.m
8 N=250/Speed in r.p.m
9 oc=4.2//Oil consumption in kg per hour
10 CV=11300//Calorific value of fuel in kcal/kg
11 jcw=4.5//Jacket cooling water in kg/min
12 rT=45 // Rise in temperature in degree C
13 \text{ au}=31//\text{Air used in kg}
14 Te=420//Temperature of exhaust gases in degree C
15 Tr=20//Room temperature in degree C
16 Cm=0.24//Mean specific heat of exhaust gases in kJ/
     kg.K
17
18 // Calculations
19 ihp=(mep*(1/100)*(3.14/4)*d^2*(N/2))/4500//Indicated
      horse power in h.p
20 bhp=(bt*2*3.14*N)/4500//Brake horse power in h.p.
21 q=(oc*CV)//Heat supplied in kcal/hour
22 qbhp=(bhp*4500*60)/427//Heat equivalent of B.H.P in
      kcal/hour
23 qfhp=((ihp-bhp)*4500*60)/427//Heat equivalent F.H.P
      in kcal/hour
  qc=(jcw*rT*60)//Heat lost in cooling water in kcal/
25 qe=(oc*32*Cm*(Te-Tr))//Heat lost in exhaust gases in
      kcal/hour
26 hu=(q-(qbhp+qfhp+qc+qe))//Heat unaccounted in kcal/
     hour
27
28 //Output
29 printf ('Indicated horse power is %3.1f h.p \n Brake
      horse power is %3.2 f h.p \n Heat supplied is %3.0
      f kcal/hour \n Heat equivalent of B.H.P is %3.0 f
      kcal/hour \n Heat equivalent of F.H.P is \%3.0 f
      kcal/hour \n Heat lost in cooling water is %3.0 f
      kcal/hour \n Heat lost in exhaust gases is \%3.0 f
      kcal/hour \n Heat unaccounted is \%3.0 f kcal/hour'
      , ihp, bhp,q,qbhp,qfhp,qc,qe,hu)
```

Scilab code Exa 23.16 HEAT BALANCE SHEET

```
1 clc
2 clear
3 //Input data
4 ihp=45//Indicated horse power in h.p.
5 bhp=37//Brake horse power in h.p
6 fu=8.4//Fuel used in kg/hour
7 CV=10000//Calorific value in kcal/kg
8 Tc=[15,70]//Inlet and outlet temperatures of
      cylinders in degree C
9 cj=7//Rate of flow of cylinder jacket in kg/min
10 Tw=[15,55] //Inlet and outlet temperatures of water
     in degree C
11 rw=12.5//Rate of water flow in kg per minute
12 Te=82//Final temperature of exhaust gases in degree
13 Tr=17//Room temperature in degree C
14 af=20//Air fuel ratio
15 Cm=0.24//Mean specific heat of exhaust gases in kJ/
     kg.K
16
17 // Calculations
18 q=(fu/60)*CV//Heat supplied in kcal/min
19 qbhp = (bhp*4500)/427//Heat equivalent of B.H.P in
     kcal/min
20 ql=(cj*(Tc(2)-Tc(1)))/Heat lost to cylinder jacket
     cooling water in kcal/min
21 qe=(rw*(Tw(2)-Tw(1)))/Heat lost by exhaust gases in
      kcal/min
22 qee=(Te-Tr)*Cm*(af+1)*fu/60//Heat of exhaust gas in
     kcal/min
23 te=(qe+qee)//Total heat lost to exhaust gases in
     kcal/min
```

```
24 hra=(q-(qbhp+ql+te))//Heat lost to radiation in kcal
     /min
25 ith=((ihp*4500)/(427*q))*100//Indicated thermal
      efficiency in percent
26 bth=((bhp*4500)/(427*q))*100//Brake thermal
      efficiency in percent
  nm=(bhp/ihp)*100//Mechanical efficiency in percent
27
28
29 // Output
30 printf ('Heat supplied is %3.0 f kcal/min \n Heat
      equivalent of B.H.P is %3.0f kcal/min \n Heat
     lost to cylinder jacket cooling water is \%3.0f
     kcal/min \n Total heat lost to exhaust gases is
     \%3.1f kcal/min \n Heat lost to radiation is \%3.1f
      kcal/min \n Indicated thermal efficiency is \%3.1
     f percent \n Brake thermal efficiency is \%3.1f
      percent \n Mechanical efficiency is \%3.1f percent
      ',q,qbhp,ql,te,hra,ith,bth,nm)
```

Scilab code Exa 23.17 HEAT LOST

```
1 clc
2 clear
3 //Input data
4 Vs=0.0015//Stroke volume in cu.m
5 rc=5.5//Volume compression ratio
6 p2=8//Pressure at the end of compression stroke in kg/cm^2
7 T2=350+273//Temperature at the end of compression stroke in K
8 p3=25//Pressure in kg/cm^2
9 x=(1/30)//Fraction of distance travelled by piston
10 pa=1/16//Petrol air mixture ratio
11 R=29.45//Characteristic gas constant in kg.m/kg degree C
```

```
12 CV=10000//Calorific value of fuel in kcal per kg
13 Cv=0.23//Specific heat in kJ/kg.K
14
15 // Calculations
16 V2 = (Vs * 10^6) / (rc - 1) / Volume in c.c
17 V3 = (Vs * 10^6) * x + V2 / Volume in c.c
18 T3=(T2*p3*V3)/(p2*V2)//Temperature in K
19 W = ((p3+p2)/2)*(V3-V2)/Workdone in kg.cm
20 mM = ((p2*V2)/(T2*R*100))/Mass of mixture present in
      kg
21 dE=(mM*Cv*(T3-T2))/Change in energy in kcal
22 q = (dE + (W/(427*100))) / Heat in kcal
23 qc = (1/(1+(1/pa)))*mM*CV //Heat in kcal
24 ql=(qc-q)/mM//Heat lost in kcal per kg of charge
25
26 // Output
27 printf('Heat lost per kg of charge during explosion
      is %3.0 f kcal',q1)
```

Chapter 25

GAS TURBINES

Scilab code Exa 25.2 OVERALL EFFICIENCY

```
1 clc
2 clear
3 //Input data
4 p=4//Pressure ratio
5 T3=1000//Turbine inlet temperature in K
6 T1=15+273//Inlet temperature in K
7 p1=1//Inlet pressure in kg/cm<sup>2</sup>
8 m=11//Mass flow rate of air in kg/s
9 Cp=0.24//Specific heat at constant pressure in kJ/kg
     .K
10 R=29.27//haracteristic gas constant in kg.m/kg.K
11 g=1.4//Ratio of specific heats
12
13 // Calculations
14 Pc = (m*R*T1*(p^((g-1)/g)-1))/75//Power consumed by
     the compressor in H.P.
15 Pt = (m*R*T3*(1-(1/p)^((g-1)/g)))/75/Power consumed
     by the turbine in H.P
16 N=(Pt-Pc)//Net output of the plant in H.P. In
      textbook answer is given wrong
17 T2=(T1*(p)^((g-1)/g))/Temperature at the end of
```

Scilab code Exa 25.3 EFFICIENCY

```
1 clc
2 clear
3 //Input data
4 T1=15+273//Temperature of air entering the
     compressor in K
5 rp=5//Pressure ratio
6\ T3=700+273//Temperature of air after heating in K
7 g=1.4//Ratio of specific heats
8 Cp=0.24//Specific heat at constant pressure in kJ/kg
     .K
9
10 // Calculations
11 T2=(T1*rp^{(g-1)/g}))/Temperature of air after
     compression in K
12 T4=(T3/rp^{(g-1)/g})/Temperature of air after
     expansion in K
13 Wc=(Cp*(T2-T1))//Workdone in the compressor in kcal/
     kg of air
14 Wt=(Cp*(T3-T4))//Workdone in the turbine in kcal/kg
15 N=(Wt-Wc)//Net workdone in kcal/kg of air
16 SHP=(N*427)/75//Shaft horse power in H.P per kg of
     air/sec
```

Scilab code Exa 25.4 NET OUTPUT

```
1 clc
2 clear
3 //Input data
4 g=1.4//Ratio of specific heats
5 Cp=0.24//Specific heat at constant pressure in kJ/kg
     .K
6 m=20.5//Air flow rate in kg/sec
7 p=[5.85,1.03,1.03,5.85]//Inlet and outlet pressure
     of turbine and compressor respectively in kg/cm<sup>2</sup>
8 T = [20+273, 250+273, 600+273, 360+273] / Inlet and outlet
      temperatures of turbine and compressor
      respectively in degree C. In textbook instead of
     360 degree C, 375 degree C is given
10 // Calculations
11 T2=(T(1)*(p(4)/p(3))^((g-1)/g))/Temperature at the
      outlet of compressor in ideal cycle in K
12 T4=(T(3)/(p(1)/p(2))^{(g-1)/g)}/Temperature at the
      outlet of turbine in ideal cycle in K
13 ic=((T2-T(1))/(T(2)-T(1)))*100//Isentropic
      efficiency of compressor in percent
14 it=((T(3)-T(4))/(T(3)-T4))*100//Isentropic
      efficiency of turbine in percent
15 Wc = (Cp*(T(2)-T(1))) / Workdone in compressor in kcal/
     kg of air
```

Scilab code Exa 25.5 TEMPERATURE

```
1 clc
2 clear
3 //Input data
4 rp=5//Pressure ratio
5 T1=15+273//Inlet temperature in K
6 nc=80//Adiabatic efficiency of the compressor in
      percent
7 n=1.4//Adiabatic index
9 // Calculations
10 T2=(T1*rp^{(n-1)/n})/Temperature at the outlet of
      compressor in ideal cycle in K. The textbook
      answer is wrong. Instead of 456 K, it is given as
      452K
11 T2i = (((nc/100)*T1)+T2-T1)/(nc/100)//Temperature at
      the outlet of the compressor in the actual cycle
     in K
12
13 //Output
14 printf ('The temperature at the end of compression is
      \%3.0 \text{ f K}', T2i)
```

Scilab code Exa 25.6 HP DEVELOPED

```
1 clc
2 clear
3 //Input data
4 p1=5.62//Pressure of gas entering the turbine in kg/
5 T1=1000+273//Temperature of gas entering the turbine
       in K
6 p2=1.124//Pressure of gas leaving the turbine in kg/
      cm<sup>2</sup>. In textbook it is given as 1.24 instead of
      1.124
7 n1=0.8//Isotropic efficiency of the turbine in ratio
8 n=1.36//Adiabatic index
9 Cp=0.25//Specific heat at constant pressure in kJ/kg
      .Κ
10
11 // Calculations
12 T2=(T1/(p1/p2)^{((n-1)/n)})/Temperature at the end of
       adiabatic expansion in K
13 dt=(T1-T2)//Isentropic temperature drop in K
14 adt=(n1*dt)//Actual temperature drop in K
15 T2i=(T1-adt)//Temperature at the end of actual
      expansion in K
16 W=(Cp*(T1-T2i))//Workdone per kg of gas in kcal
17 q = (W*427)/4500//H.P developed per kg of gas per
18 t2i=(T2i-273)//Exhaust gas temperature in degree C
19
20 //Output
21 printf('(1) H.P developed per kg of gas per min is
      \%3.2 \,\mathrm{f} \, \backslash \mathrm{n} \, (2) Exhaust gas temperature is \%3.1 \,\mathrm{f}
      degree C',q,t2i)
```

Scilab code Exa 25.7 THERMAL EFFICIENCY

1 clc

```
2 clear
3 //Input data
4 pt1=[1,15+273] // Pressure and temperature at the
      inlet of compressor in kg/cm<sup>2</sup> and K respectively
  pt3=[4,650+273] // Pressure and temperature at the
      inlet of turbine in kg/cm<sup>2</sup> and K respectively
6 n=[85,80]//Isentropic efficiencies of turbine and
      compressor respectively in percent
7 g=1.4//Ratio of specific heats
8 Cp=0.24//Specific heat at constant pressure in kJ/kg
      .K
10 // Calculations
11 T2=(pt1(2)*(pt3(1)/pt1(1))^((g-1)/g))//Temperature
      at the end of adiabatic compression in K
  T2i = (pt1(2) + ((T2-pt1(2))/(n(2)/100)))/Temperature
      at the end of actual compression in K
13 T4=(pt3(2)/(pt3(1)/pt1(1))^((g-1)/g))/Temperature
      at the end of adiabatic expansion in K
14 T4i = (pt3(2) - ((pt3(2) - T4) * (n(1)/100))) / Temperature
      at the end of actual expansion in K
  Wt = (Cp*(pt3(2)-T4i))/Workdone in turbine in kcal/kg
       of air
16 Wc=(Cp*(T2i-pt1(2)))//Workdone in compressor in kcal
     /kg of air
17 N=(Wt-Wc)//Net workdone in kcal/kg of air
18 q=(Cp*(pt3(2)-T2i))//Heat supplied in kcal/kg of air
19 nt = (N/q) *100 / Thermal efficiency in percent
20
21 // Output
22 printf ('Thermal efficiency of the cycle is \%3.2 f
      percent', nt)
```

Scilab code Exa 25.8 FLOW RATE

```
1 clc
2 clear
3 //Input data
4 p1=1.03//Inlet air pressure in kg/cm<sup>2</sup>
5 T1=15.5+273//Inlet temperature of air in K
6 rp=5//Compression ratio
7 nc=85//Isentropic efficiency of the compressor in
      percent
  T3=540+273 // Temperature of the gas entering the
      turbine in K
9 p3=1.03//Pressure of gas entering the turbine in kg/
10 nt=80//Isentropic efficiency of the turbine in
      percent
11 0=2500/Net output in H.P
12 fp=0.07//Fall of pressure through the combustion
     chamber in kg/cm<sup>2</sup>
13 g=1.4//Ratio of specific heats for both air and gas
14 Cp=0.24//Specific heat at constant pressure in kJ/kg
      .K for both air and gas
15
16 // Calculations
17 T2=(T1*rp^{(g-1)/g}))/Temperature of air at the end
      of adiabatic compression in K
18 T2i = (T1 + ((T2 - T1) / (nc/100))) / Temperature of air at
      the end of actual compression in K
19 T4=(T3/((rp*p3-fp)/p3)^((g-1)/g))//Temperature of
      air at the end of adiabatic compression in K
  T4i = (T3 - ((T3 - T4) * (nt/100))) / Temperature of air at
      the end of actual compression in K
21 Wt=(Cp*(T3-T4i))//Workdone in turbine in kcal/kg of
22 Wc = (Cp*(T2i-T1))/Workdone in compressor in kcal/kg
      of air
23 N=(Wt-Wc)//Net workdone in kcal/kg of air
24 F1=(0*4500)/(427*N*60)//Flow rate for 2500 H.P in kg
      /sec
25
```

```
26 //Output
27 printf('Flow rate of air is %3.1f kg/sec for a net
      output of %i H.P',Fl,0)
```

Scilab code Exa 25.9 FLOW RATE

```
1 clc
2 clear
3 //Input data
4 p1=1//Inlet air pressure in kg/cm<sup>2</sup>
5 T1=16+273//Inlet temperature of air in K
6 rp=3.5//Pressure ratio
7 nc=85//Isentropic efficiency of the compressor in
      percent
  T3=500+273//Temperature of the gas entering the
      turbine in K
9 nt=80//Isentropic efficiency of the turbine in
      percent
10 mc=4//Mass of air entering the compressor in tonnes/
     hour
11 g=1.4//Ratio of specific heats
12 Cp=0.24//Specific heat at constant pressure in kJ/kg
      . K
13
14 // Calculations
15 T2=(T1*rp^{(g-1)/g}))/Temperature of air at the end
      of adiabatic compression in K
16 dt=(T2-T1)//Isentropic temperature rise in K
17 adt=(dt/(nc/100))//Actual temperature rise in K
18 T2i = (T1 + ((T2 - T1)/(nc/100))) / Temperature of air at
      the end of actual compression in K
19 T4=(T3/rp^{(g-1)/g})/Temperature of air at the end
      of adiabatic compression in K
20 T4i = (T3 - ((T3 - T4) * (nt/100))) / Temperature of air at
      the end of actual compression in K
```

```
21 Wt=(Cp*(T3-T4i))//Workdone in turbine in kcal/kg of air
22 Wc=(Cp*(T2i-T1))//Workdone in compressor in kcal/kg of air
23 N=(Wt-Wc)//Net workdone in kcal/kg of air
24 q=(Cp*(T3-T2i))//Heat supplied in kcal/kg of air
25 NHP=(N*427*mc*1000)/(60*4500)//Net Horse Power available in H.P
26 nt=(N/q)*100//Thermal efficiency in percent
27
28 //Output
29 printf('(i) The net Horse power available from this unit is %3.1 f H.P \n (ii) The thermal efficiency of the plant is %3.2 f percent', NHP,nt)
```

Scilab code Exa 25.10 EFFICIENCY

```
1 clc
2 clear
3 //Input data
4 p1=1.02//Inlet air pressure in kg/cm<sup>2</sup>
5 T1=27+273//Inlet temperature of air in K
6 p2=4.08//Pressure after compression in kg/cm<sup>2</sup>
7 nc=80//Isentropic efficiency of compressor in
      percent
8 mf=1//Mass of fuel in kg
9 ma=80//Mass of air in kg
10 nt=85//Isentropic efficiency of the turbine in
      percent
11 CV=10000//Calorific value of fuel n kcal per kg
12 g=1.4//Ratio of specific heats
13 Cp=0.24//Specific heat at constant pressure in kJ/kg
      .Κ
14
15 // Calculations
```

```
16 rp=(p2/p1)//Pressure ratio
17 T2=(T1*rp^{(g-1)/g}))/Temperature of air at the end
      of adiabatic compression in K
18 dt=(T2-T1)//Isentropic temperature rise in K
19 adt=(dt/(nc/100))//Actual temperature rise in K
20 T2i = (T1 + ((T2 - T1)/(nc/100))) / Temperature of air at
     the end of actual compression in K
21 q=(mf/ma)*CV//Heat supplied per kg of air in kcal
22 T3=(q/Cp)+T2i//Temperature of gas at the inlet of
     the turbine in K
23 T4=(T3/rp^{((g-1)/g)})/Temperature of air at the end
      of adiabatic expansion in K
24
  T4i = (T3 - ((T3 - T4) * (nt/100))) / Temperature of air at
      the end of actual expansion in K
  Wt = (Cp*(T3-T4i)*((ma+mf)/ma))/Workdone in turbine
     in kcal/kg of air
26 Wc=(Cp*(T2i-T1))//Workdone in compressor in kcal/kg
     of air
27 N=(Wt-Wc)//Net workdone in kcal/kg of air
28 nt=(N/q)*100//Thermal efficiency in percent
29
30 //Output
31 printf('(a) The net work output of installation is
     %3.2f kcal/kg of air \n (b) Overall efficiency of
       the plant is %3.1f percent', N, nt)
```

Scilab code Exa 25.11 EFFICIENCY

```
1 clc
2 clear
3 //Input data
4 rp=5//Pressure ratio
5 T3=580+273//Temperature of gas at the inlet of the turbine in K
6 p1=1.03//Inlet air pressure in kg/cm<sup>2</sup>
```

```
7 T1=15+273//Inlet temperature of air in K
8 nc=80//Isentropic efficiency of compressor in
      percent
9 no=18//Overall efficiency of the plant in percent
10 Cpa=0.239//Specific heat of air at constant pressure
      in kJ/kg.K
11 Cpg=0.261//Specific heat of gas at constant pressure
      in kJ/kg.K
12 R=29.27 // haracteristic gas constant in kg.m/kg.K
13 g=1.4//Ratio of specific heats for air
14 g1=1.355//Ratio of specific heats for gas
15
16 // Calculations
17 T2=(T1*rp^{(g-1)/g}))/Temperature of air at the end
     of adiabatic compression in K
  T2i = (T1 + ((T2 - T1)/(nc/100)))//Temperature of air at
      the end of actual compression in K
19 q=(Cpg*(T3-T2i))//Heat supplied in kcal/kg of air
20 Wc=(Cpa*(T2i-T1))//Workdone in compressor in kcal/kg
      of air
  Wt=((no/100)*q)+Wc//Turbine work output in kcal/kg
      of air
  T4i=(T3-(Wt/Cpg))//Temperature of air at the end of
22
      actual expansion in K
23 T4=(T3/rp^{(g1-1)/g1)}/Temperature of air at the
     end of adiabatic expansion in K
24 nt = ((T3-T4i)/(T3-T4))*100//Isentropic efficiency of
      turbine in percent
25
26 //Output
27 printf ('Isentropic efficiency of turbine is %3.1 f
     percent', nt)
```

Scilab code Exa 25.12 FLOW OF AIR

```
1 clc
2 clear
3 //Input data
4 p1=1.03//Inlet air pressure in kg/cm<sup>2</sup>
5 T1=15+273//Inlet temperature of air in K
6 rp=5//Pressure ratio
7 nc=85//Isentropic efficiency of the compressor in
      percent
  T3=540+273 // Temperature of the gas entering the
      turbine in K
9 nt=80//Isentropic efficiency of the turbine in
      percent
10 NHP=2000//Net horse power in H.P.
11 fp=0.1//Fall of pressure through the combustion
      system in kg/cm<sup>2</sup>
12 g=1.4//Ratio of specific heats for both air and gas
13 Cp=0.25//Specific heat at constant pressure in kJ/kg
      .K for both air and gas
14
15 // Calculations
16 T2i = (T1*rp^{(g-1)/g}) / Temperature of air at the end
       of adiabatic compression in K
17 dt=(T2i-T1)//Isentropic temperature rise in K
18 adt=(dt/(nc/100))/Actual temperature rise in K
19 Wc=(Cp*adt)//Workdone in compressor in kcal/kg of
      air
20 e=((rp*p1-fp)/p1)/(Expansion ratio)
21 T4i = (T3/e^{(g-1)/g})/Temperature of air at the end
      of adiabatic expansion in K
22 dt1=(T3-T4i)//Isentropic temperature rise in K
23 adt1=(dt1/(nt/100))//Actual temperature rise in K
24 Wt=(Cp*adt1)//Workdone in turbine in kcal/kg of air
25 N=(Wt-Wc)//Net workdone in kcal/kg of air
26 \text{ w} = (\text{NHP} * 75) / (427 * 9.8) / / \text{Flow rate in kg of air per sec}
27
28 //Output
29 printf('Flow rate is %3.2 f kg of air per sec',w)
```

Scilab code Exa 25.13 POWER OTPUT

```
1 clc
2 clear
3 //Input data
4 nc=75//Isentropic efficiency of the compressor in
      percent
  nt=85//Isentropic efficiency of the turbine in
      percent
6 nm=98//Mechanical efficiency in percent
7 rp=6//Pressure ratio
8 T3=727+273 // Temperature of the gas entering the
      turbine in K
9 p1=1//Inlet air pressure in kg/cm<sup>2</sup>
10 T1=15.5+273//Inlet temperature of air in K
11 m=2.2//Mass flow rate in kg/sec
12 Cpa=0.24//Specific heat of air at constant pressure
      in kJ/kg.K
13 Cpg=0.276//Specific heat of gas at constant pressure
      in kJ/kg.K
14 g=1.4//Ratio of specific heats for air
15 g1=1.33//Ratio of specific heats for gas
16
17 // Calculations
18 T2=(T1*rp^{(g-1)/g}))/Temperature of air at the end
      of adiabatic compression in K
19 T2i = (T1 + ((T2 - T1))/(nc/100)))/Temperature of air at
      the end of actual compression in K
20 T4=(T3/rp^{(g1-1)/g1)}/Temperature of air at the
      end of adiabatic compression in K
21 T4i = (T3 - ((T3 - T4) * (nt/100))) / Temperature of air at
      the end of actual compression in K
22 Wt=(Cpg*(T3-T4i))//Workdone in turbine in kcal/kg of
       air
```

Scilab code Exa 25.14 EFFICIENCY

```
1 clc
2 clear
3 //Input data
4 T1=15+273//Inlet temperature of air in K
5 rp=4//Pressure ratio
6 T4=560+273//Maximum temperature of the cycle in K
7 nc=83//Isentropic efficiency of the compressor in
     percent
8 nt=86//Isentropic efficiency of the turbine in
     percent
9 x=75//Heat exchanger making use of heat available in
      percent
10 g=1.4//Ratio of specific heats
11
12 // Calculations
13 T5i = (T4*(1/rp)^{((g-1)/g)}) / Temperature in K
14 dt=(T4-T5i)//Isometric temperature drop through
      turbine in degree C
15 ta=((nt/100)*dt)//Actual temperature drop in degree
16 T5=(T4-ta)//Temperature in K
17 T2i = (T1*rp^{(g-1)/g})/Temperature in K
18 tc=(T2i-T1)//Temperature change in degree C
19 T2=(tc/(nc/100))+T1//Temperature in K
```

```
20 q=(T5-T2)//Available heat in exchanger in kcal per
     kg *Cp
21 T3 = ((q*(x/100)) + T2) / Temperature in K
22 //Without heat exchanger
23 qw=(T4-T2)//Heat supplied *Cp in kcal/kg
24 tw=(T4-T5)//Turbine work *Cp in kcal/kg
25 cw=(T2-T1)//Compressor work *Cp in kcal/kg
26 nw=(tw-cw)//Net workdone *Cp in kcal/kg
27 no=(nw/qw)*100//Overall efficiency in percent
28 //With heat exchanger
29 qs=(T4-T3)//Heat supplied *Cp in kcal/kg
30 no1=(nw/qs)*100//Overall efficiency in percent
31
32 //Output
33 printf('The overall efficiency \n (a) without heat
     exchanger is \%3.1f percent \n (b) with heat
     exchanger making use of %i percent of heat
      available is %3.1f percent', no, x, no1)
```

Scilab code Exa 25.15 PRESSURE

```
1 clc
2 clear
3 //Input data
4 p1=1//Initial pressure in kg/cm^2
5 T1=15+273//Initial temperature in K
6 p2=5.5//Pressure after compression in kg/cm^2
7 T3=750+273//Temperature at the entrance of turbine in K
8 v=225//Speed in m/s
9 x=70//Percentage
10 in=75//Isentropic efficiency of compressor in percent
11 Cp=0.24//Specific heat at constant pressure in kJ/kg.K
```

```
12 g=1.4//Ratio of specific heats
13
14 // Calculations
15 T2=(T1*(p2/p1)^((g-1)/g))/Temperature in K
16 at=(T2-T1)/(in/100)//Actual temperature rise in the
      compressor in K
17 T2i=(T1+at)//Temperature in K
18 T4=(T3/(p2/p1)^{((g-1)/g)})/Temperature in K
19 to=(Cp*(T3-T4))//Theoritical turbine output in kcal/
     kg of air
20 ci=(Cp*(T2i-T1))//Actual compressor input in kcal/kg
       of air
21 ke = (v^2/(2*9.81*427)) / Kinetic energy in gas leaving
       the exhaust annulus in kcal/kg
  dT34 = (ci+ke)/((x/100)*Cp)//Change in temperature in
23 r=1/(1-(dT34/T3))^{(g/(g-1))}/Ratio of pressures
24 p4=(r/p2)//Pressure in kg/cm^2
25
26 //Output
27 printf ('The pressure of the gases in the turbine
      exhaust annulus is %3.1 f kg/cm<sup>2</sup>',p4)
```

Scilab code Exa 25.16 EFFICIENCY

```
1 clc
2 clear
3 //Input data
4 p=[1,5]//Pressures in atm
5 T1=288//Temperature in K
6 T3=650+273//Temperature in K
7 er=0.85//Efficiency ratio
8 x=0.72//Effectiveness of heat exchanger
9 Cp=0.24//Specific heat at constant pressure in kJ/kg
.K
.K
```

```
10 g=1.4//Ratio of specific heats
11
12 //Calculations
13 T2=(T1*(p(2)/p(1))^((g-1)/g))//Temperature in K
14 T2i=(T1+((T2-T1)/er))//Temperature in K
15 T4=(T3/(p(2)/p(1))^((g-1)/g))//Temperature in K
16 T4i=(T3-(er*(T3-T4)))//Temperature in K
17 Tc=((x*(T4i-T2i))+T2i)//Temperature in K
18 W=((Cp*((T3-T4i)-(T2i-T1))))//Workdone in kcal/kg
19 q=(Cp*(T3-Tc))//Heat supplied in kcal/kg
20 n=(W/q)*100//Efficiency in percent
21
22 //Output
23 printf('The heat efficiency of the plant is %3.1f percent',n)
```

Scilab code Exa 25.17 EFFICIENCY

```
1 clc
2 clear
3 //Input data
4 T1=15+273//Inlet temperature of air in K
5 p1=1.03//Inlet pressure of air in kg/cm<sup>2</sup>
6 rp=5//Pressure ratio
7 T3=815+273//Temperature of air entering the turbine
     in K
8 nc=0.83//Adiabatic efficiency of the compressor
9 nt=0.92//Internal engine efficiency of the turbine
10 nr=0.65//Effectiveness of regenerator
11 p2=2.45//Pressure in kg/cm^2
12 T6=T1//Temperature in K
13 T9=T3//Temperature in K
14 Cp=0.24//Specific heat at constant pressure in kJ/kg
15 g=1.4//Ratio of specific heat
```

```
16
17 // Calculations
18 T2=(T1*rp^((g-1)/g))/Temperature in K
19 T4=(T3/rp^{(g-1)/g})/Temperature in K
20 Wt=(Cp*(T3-T4))//Isentropic work done in the turbine
       in kcal/kg of air
21 Wc = (Cp*(T2-T1)) / Isentropic work done in the
      compressor in kcal/kg of air
22 Wr = (Wt/Wc) //Work ratio
23 qa=(Cp*(T3-T2))/Heat added in kcal/kg of air
24 nth=((Wt-Wc)/qa)*100//Thermal efficiency in percent
25 T2i=(T1+((T2-T1)/nc))//Temperature in K
26 T4i=(T3-(nt*(T3-T4)))//Temperature in K
27 Wti=(Cp*(T3-T4i))//work done in the turbine in kcal/
     kg of air
  Wci=(Cp*(T2i-T1))//work done in the compressor in
28
      kcal/kg of air
29 Wri=(Wti/Wci)//Work ratio
30 qai=(Cp*(T3-T2i))//Heat added in kcal/kg of air
31 nthi=((Wti-Wci)/qai)*100//Thermal efficiency in
      percent
32 T2ii=(T2i+((T4i-T2i)*nr))//Temperature in K
33 qaii=(Cp*(T3-T2ii))/Heat added in kcal/kg of air
34 nthii=((Wti-Wci)/qaii)*100//Thermal efficiency in
      percent
35 T5=(T1*(p2/p1)^((g-1)/g))/Temperature in K
36 T5i = (T1 + ((T5 - T1)/nc)) / Temperature in K
37 T7 = (T1*((rp*p1)/p2)^((g-1)/g))/Temperature in K
38 T7i = (T6 + ((T7 - T6)/nc))//Temperature in K
39 T7ii = (T7i + ((T4i - T7i) * nr)) / Temperature in K
40 Wcomp = (Cp*((T5i-T1)+(T7i-T6)))/(Compressor work in
      kcal/kg of air
41 Wratio=(Wti/Wcomp)//Work ratio
42 qaa=(Cp*(T3-T7ii))/Heat added in kcal/kg of air
43 nthe = ((Wti-Wcomp)/qaa)*100//Thermal efficiency in
      percent
44 T8=(T3*(p2/(rp*p1))^((g-1)/g))//Temperature in K
45 T8i = (T3 - ((T3 - T8) * nt)) / / Temperature in K
```

```
46 T10=(T9/(p2/p1)^((g-1)/g))//Temperature in K
47 T10i = (T9 - ((T9 - T10) * nt)) / Temperature in K
48 T2iii=(T2i+((T10i-T2i)*nr))//Temperature in K
49 Wturb=(Cp*((T3-T8i)+(T3-T10i)))//Compressor work in
       kcal/kg of air
50 \text{ Wratioi} = (\text{Wturb/Wci}) / / \text{Work ratio}
51 qaai = (Cp*((T3-T2iii)+(T9-T8i)))/Heat added in kcal/
       kg of air
52 nthei=((Wturb-Wci)/qaai)*100//Thermal efficiency in
       percent
53 T7iii = (T7i + ((T10i - T7i) * nr)) / Temperature in K
54 Wratioii=(Wturb/Wcomp)//Work ratio
55 qaaii=(Cp*((T3-T7iii)+(T9-T8i)))/Heat added in kcal
       /kg of air
56 ntheii=((Wturb-Wcomp)/qaaii)*100//Thermal efficiency
        in percent
57
58 //Output
59 printf ('Condition
                               Work ratio
                                                  Thermal
       efficiency (in percent)\n
                                         (a)
                                                          \%3.3 \text{ f}
                         \%3.1 \text{ f} \n
                                      (b)
                                                        \%3.2 \text{ f}
                           \%3.1 \text{ f} \n
                                                         \%3.2 \text{ f}
                                       (c)
                           \%3.1 \text{ f} \n
                                       (d)
                                                         \%3.2 \text{ f}
                           \%3.1 \text{ f} \n
                                                         \%3.3 \text{ f}
                                        (e)
                         \%3.1 \text{ f} \n
                                      (f)
                         \%3.1 \, \mathrm{f}', Wr, nth, Wri, nthi, Wri, nthii,
       Wratio, nthe, Wratioi, nthei, Wratioii, ntheii)
```

Scilab code Exa 25.18 EFFICIENCY

```
1 clc
2 clear
3 //Input data
4 p=[1,9]//Pressures in ata
5 T=[25+273,1250+273]//Minimum and maximum
```

```
temperatures in K
6 n=0.83//Compressor and turbine efficiencies
7 Cp=0.24//Specific heat at constant pressure in kJ/kg
     .K
8 g=1.4//Ratio of specific heats
9 x=0.65//Cycle with 65% regeneration
10
11 // Calculations
12 //(a) Without regeneration
13 ip=sqrt(p(1)*p(2))/Intermediate pressure in ata
14 T2=(T(1)*(ip/p(1))^((g-1)/g))/Temperature in K
15 T3=(T(1)+((T2-T(1))/n))/Temperature in K
16 T4=T(1)//Temperature in K
17 T5=T2//Temperature in K
18 T6=T3//Temperature in K
19 T7=T(2)//Temperature in K
20 T8=T7/(ip/p(1))((g-1)/g)/Temperature in K
21 T9=(T7-((T7-T8)*n))/Temperature in K
22 T10=T7 // Temperature in K
23 T11=T8//Temperature in K
24 T12=T9//Temperature in K
25 Wc=(2*Cp*(T3-T(1)))/Work of compression in kcal/kg
     of air
26 We=(2*Cp*(T7-T8))/Work of expansion in kcal/kg of
27 NW=(We-Wc)//Net output in kcal/kg of air
28 qi = (Cp*((T7-T6)+(T10-T9))) // Heat input in kcal/kg of
29 nth=(NW/qi)*100//Thermal efficiency in percent
30
31 //(b) Cycle efficiency with 65\% regeneration
32 Tg = (T6 + (x*(T12-T6))) / Temperature in K
33 q=(Cp*((T7-Tg)+(T10-T9)))//Heat input in kcal/kg of
      air
34 nthi=(NW/q)*100//Thermal efficiency in percent
35
36 //(c) Cycle efficiency with ideal regeneration
37 Eg=T12//Temperature in K
```

Scilab code Exa 25.19 EFFICIENCY

```
1 clc
2 clear
3 //Input data
4 p1=1//Inlet pressure of compressor in atm
5 T1=27+273//Inle temperature of compressor in K
6 ic=0.8//Isentropic efficiency of compressor
7 ma=20.5//Mass flow rate of air in kg/sec
8 T3=650+273//Inlet temperatures of both turbines in K
9 p2=5//Inlet pressure of turbine in atm
10 it=0.92//Internal engine efficiency for both the
     turbines
11 CV=10000//Calorific value in kcal/kg
12 Cpa=0.24//Specific heat at constant pressure of air
     in kJ/kg.K
13 ga=1.4//Ratio of specific heats for air
14 Cpg=0.276//Specific heat at constant pressure of gas
      in kJ/kg.K
15 gs=1.33//Ratio of specific heats for gas
16
17 // Calculations
18 T2=(T1*(p2/p1)^((ga-1)/ga))/Temperature in K
19 T2i=(T1+((T2-T1)/ic))//Temperature in K
20 T4=(T3/(p2/p1)^((gs-1)/gs))/Temperature in K
```

```
21 T4i=(T3-((T3-T4)*it))//Temperature in K
22 Wc=(Cpa*(T2i-T1))//Work of compression in kcal/kg of air
23 We=(Cpg*(T3-T4i))//Work of expansion in kcal/kg of air
24 mx=(Wc/We)//Gas required per kg of air compressed in kg
25 F=((Cpa*T2i)-(Cpg*T3))/(Cpg*T3-CV)//Amount of fuel supplied per kg of air in kg
26 Wg=1+F//Weight of gases per kg of air in kg
27 Wt=(Wg-mx)//Gases supplied to turbine in kg
28 Ot=((Wt*ma*427*We)/75)//Output of turbine in H.P
29 nth=((Wt*We)/(CV*F))*100//Thermal efficiency in percent
30
31 //Output
32 printf('Output is %3.0 f H.P \n Thermal efficiency is %3.2 f percent',Ot,nth)
```

Scilab code Exa 25.20 HP

Scilab code Exa 25.22 EFFICIENCY

```
1 clc
2 clear
3 //Input data
4 p=[5,20]//Pressure limits in atm
5 T3=650+273//Temperature in K
6 T1=60+273//Temperature in K
7 T2=T1//Temperature in K
8 Cp=0.24//Specific heat at constant pressure in kJ/kg
     . K
9 g=1.4//Ratio of specific heats
10 R=29.27 // Characteristic gas constant in kg.m/kg.K
11 J=427 // Mechanical equivalent of heat in kg.m/kcal
12
13 // Calculations
14 T4=T3/(p(2)/p(1))^{((g-1)/g)}/Temperature in K
15 Wc = ((R*T1)/J)*log(p(2)/p(1))/Compression work in
     kcal/kg
16 qs=(Cp*(T3-T2))//Heat supplied at constant pressure
```

```
in kcal/kg
17 qre=(Cp*(T4-T1))//Heat ejected during process 4-1 in kcal/kg
18 nth=((qs-Wc-qre)/(qs-qre))*100//Thermal efficiency in percent
19 nc=((T3-T1)/T3)*100//Carnot efficiency in percent
20 r=(nth/nc)*100//Ratio of air standard efficiency to carnot efficiency in percent
21
22 //Output
23 printf('(a) air standard efficiency of the cycle is %3.1 f percent \n (b) carnot efficiency is %3.0 f percent \n (c) Ratio of air standard efficiency to carnot efficiency is %3.1 f percent',nth,nc,r)
```

Chapter 26

JET PROPULSION

Scilab code Exa 26.1 PROPERTIES

```
1 clc
2 clear
3 //Input data
4 Ve=2700//Jet exit velocity in m/s
5 Vf=1350//Forward flight velocity in m/s
6 m=78.6//Propellant consumption in kg/s
8 // Calculations
9 T = ((m/9.81) * (Ve - Vf)) / Thrust in kg
10 TH=((T*Vf)/75)/10^5//Thrust horse power in HP*<math>10^5
11 pn=(2/(1+(Ve/Vf)))*100//Propulsive efficiency in
      percent
12
13 //Output
14 printf('(i) Thrust is %3.0 f kg \n (ii) Thrust horse
      power is \%3.3 \, f*10^5 \, H.P \setminus n (iii) Propulsive
      efficiency is %3.1f percent', T, TH, pn)
```

Scilab code Exa 26.2 AIR FUEL RATIO

```
1 clc
2 clear
3 //Input data
4 CV=10000//Calorific value in kcal/kg
5 F=1.4//Fuel consumption in kg per hour per kg of
      thrust
6 T=900//Thrust in kg
7 Va=425//Aircraft velocity in m/s
8 w=19.5//Weight of air in kg/sec
10 // Calculations
11 af = (w/((F*T)/3600))//Air fuel ratio
12 nv = ((T*Va*3600)/(427*F*T*CV))*100//Overall
      efficiency in percent
13
14 // Output
15 printf('Air fuel ratio is %3.1f \n Overall
      efficiency is %3.1f percent', af, nv)
```

Scilab code Exa 26.3 PROPERTIES

```
1 clc
2 clear
3 //Input data
4 a=11500//Altitude in m
5 n=123//Number of passengers
6 c=3//Cargo in tonnes
7 Va=650//Velocity of air craft in km/hour
8 d=640//Drag in kg
9 pe=50//Propulsion efficiency in percent
10 oe=18//Overall efficiency in percent
11 CV=10000//Calorific value in kcal/kg
12 da=0.0172//Density of air at 11500 m in kg/cm^2
13
14 //Calculations
```

```
15 Vp = ((Va*1000)/3600) / Velocity of aeroplane in m/s
16 Vr = ((2/(pe/100)) - 1) * Vp / Velocity of working medium
      in m/s
17 nhp = ((d*Vp)/(75*(pe/100)))/Net horse power in H.P
18 wf = ((nhp*75*3600)/((oe/100)*427*CV))/Mass flow rate
       in kg/hr
19 thp=((Va*Vp)/75)//Thrust horse power in H.P
20 F=(wf/thp)//Fuel consumption per thrust H.P hour in
21 W=((Va*9.81)/Vr)//Air flow in kg/sec
22 va=(W/da)//Volume of air in cu.m/sec
23 aa=(va/(3*Vr))/Area of jet in m^2
24 d=sqrt((4*aa)/3.14)*100//Diameter of jet in cm
25 af = ((W*3600)/wf)//Air fuel ratio
26
27 // Output
28 printf('(a) Absolute velocity of the jet is \%3.1f m/
      sec \n (b) Net horse power of the gas plant is \%3
      .0 f H.P \n (c) Fuel consumption per thrust H.P
     hour is %3.3 f kg \n (d) Diameter of the jet is %3
      .1 f cm \n (e) Air-fuel ratio of the engine is \%3
      .1 f', Vr, nhp, F, d, af)
```

Scilab code Exa 26.4 AREA

```
1 clc
2 clear
3 //Input data
4 p1=7//Pressure of gas before expansion in kg/cm^2
5 p2=5//Pressure of gas after expansion in kg/cm^2
6 T1=250+273//Temperature of gas before expansion in K
7 Cp=0.24//Specific heat at constant pressure in kJ/kg .K
8 Cv=0.17//Specific heat at constant volume in kJ/kg.K
9 nv=0.8//Nozzle efficiency
```

```
10
11 // Calculations
12 R=427*(Cp-Cv)//Characteristic gas constant in kg.m/
      kg.K
13 g=(Cp/Cv)//Ratio of specific heats
14 V1 = (R*T1)/(p1*10^4)/Volume in cu.m per kg
15 V2 = (V1 * (p1/p2)^(1/g)) / Volume in cu.m per kg
16 Wd = (g/(g-1))*(p1*V1-p2*V2)*10^4/Work done in m.kg
      per kg
17 KE=(nv*Wd)//Kinetic energy at exit in m.kg per kg
18 v3=sqrt(2*9.81*nv*Wd)//Velocity in m/s
19 T2=(T1*(p2/p1)*(V2/V1))/Temperature in K
20 T3=(((1-nv)*Wd)/(427*Cp))+T2//Temperature in K
21 V3=(V2*(T3/T2))/Volume in cu.m per kg
22 qa=(V3/v3)*10^4//Discharge area unit rate of mass
      flow in cm<sup>2</sup>
23
24 //Output
25 printf ('Area of discharge per unit rate of mass flow
       is \%3.2 \, \text{f} \, \text{sq.cm}', \text{qa})
```

Scilab code Exa 26.5 POWER AND THRUST

```
clc
clear
//Input data
p=3.5//Pressure at the delivery is 3.5 times that at
    entrance

T=1.15//Temperature rise during compression is 1.15
    times that for frictionless adiabatic compression
    . In textbook it is given wrong as 1.5
T3=500+273//Temperature of products of combustion in
    K
pa=1//Atmospheric pressure in kg/cm<sup>2</sup>
Ta=15+273//Atmospheric temperature in K
```

```
9 Cp=0.24//Specific heat at constant pressure in kJ/kg
      . K
10 g=1.4//Ratio of specific heats
11 J=427 // Mechanical equivalent of heat in kg.m/kcal
12
13 // Calculations
14 p2=p*pa//Pressure in kg/cm<sup>2</sup>
15 T2a=(Ta*(p2/pa)^((g-1)/g))/Temperature in K
16 T2=(T2a-Ta)*T+Ta//Temperature in K
17 wcomp=(Cp*(T2-Ta))//Work done by compressor in kcal/
     kg
18 T5=T3/(p2/pa)^{((g-1)/g)}/Temperature in K
19 dh35=(Cp*(T3-T5))/(Change in enthalpy in kcal/kg)
20 dhnozzle=(dh35-wcomp)//Change in enthalpy of nozzle
     in kcal/kg
21 v5=sqrt(2*9.81*J*dhnozzle)//Velocity at the nozzle
      exit in m/sec
22 Th=(v5/9.81)//Thrust in kg per kg of air/sec
23
24 // Output
25 printf('(a) the power required to drive the
      compressor per kg of air per second is %3.1f kcal
     /kg \n (b) Static thrust developed per kg of air
     per second is %3.1 f kg', wcomp, Th)
```