

# MULTI-DIMENSIONAL FEMTOSECOND-LASER INDUCED DYNAMICS OF CO ON METALS:

## ACCOUNTING FOR ELECTRONIC FRICTION AND SURFACE MOTION WITH COMBINED MODELS

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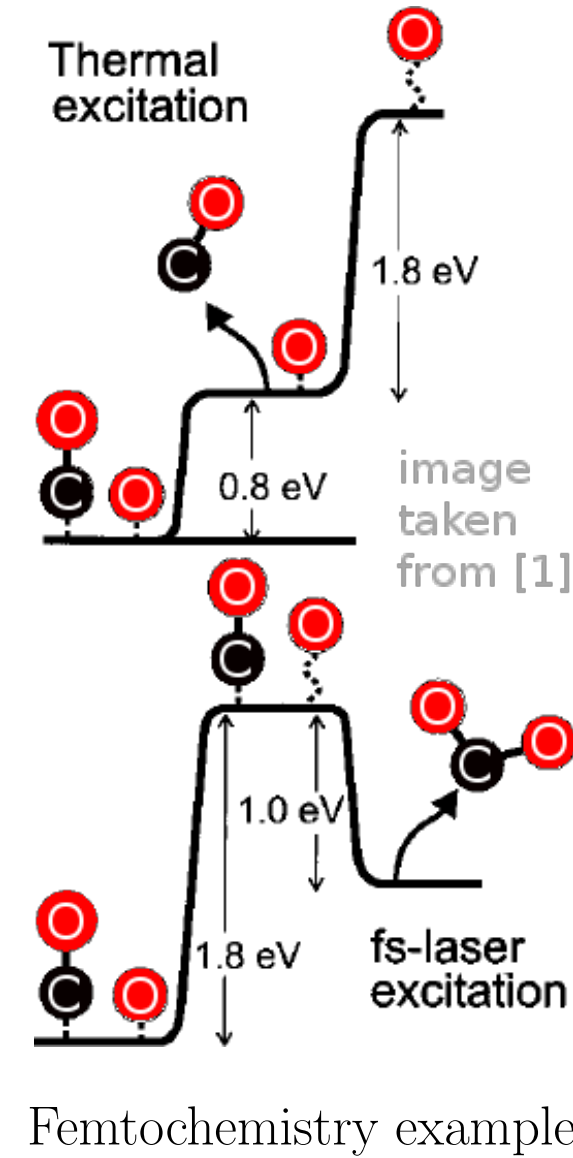
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## Introduction

### Motivation

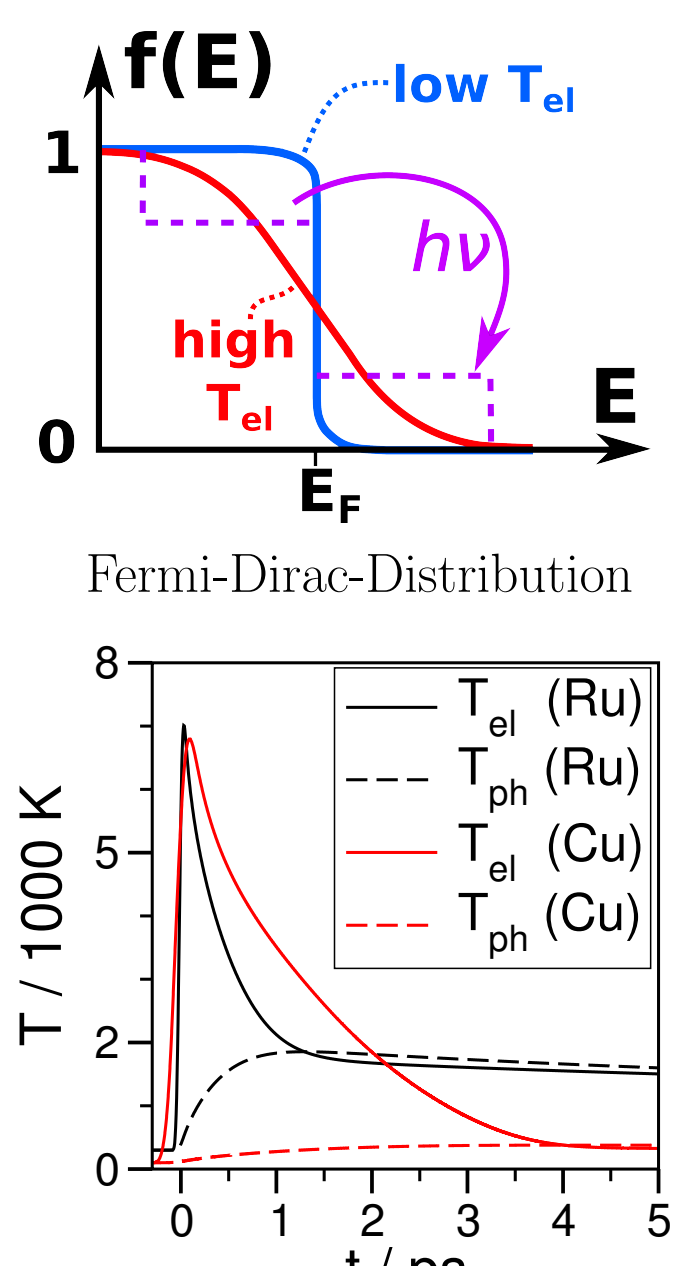
- **Aim:** gain **precise understanding of adsorbate bonding on metals**  
⇒ Important for **Catalysis**
- **Why femtosecond(fs)-lasers?**
  - **produce non-equilibrium 2-T-states**  
⇒ **different than normal heating**
  - further **tool** besides STM and scattering
  - **direct future applications** possible  
⇒ “**femtochemistry**”[1]
- **Why CO/Ru(001) and CO/Cu(100)?**
  - both are well studied **model systems**
  - recently, interesting **fs-laser experiments**[2][3]
  - also, **ab-initio based 6-dim. potentials** available [4][5]



Femtochemistry example

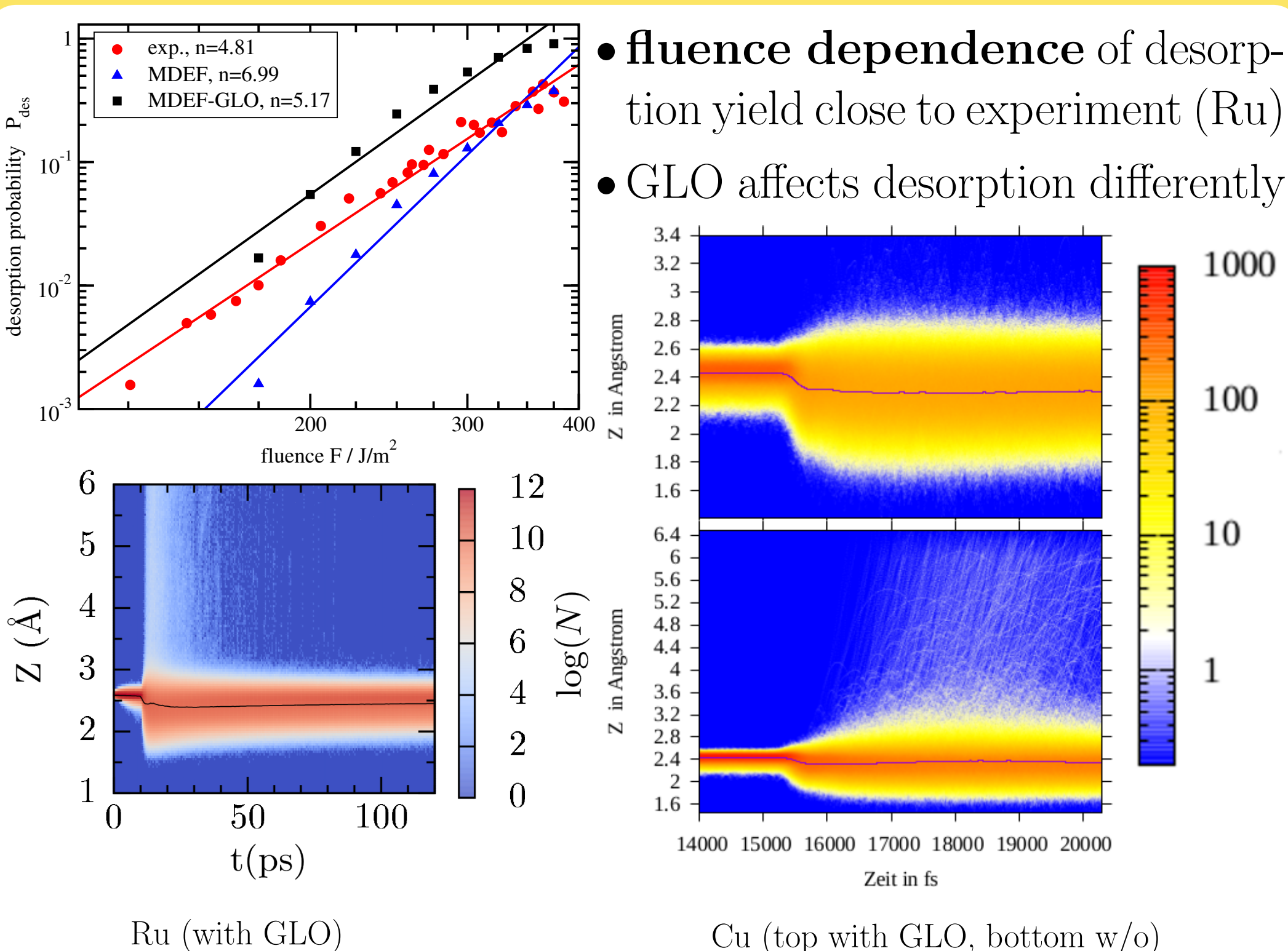
### How do fs-Lasers affect Adsorbate-Metal Systems?

- **fs-Laser** (purple wavy line) interacts with the **Ru (0001)** surface.
- **Desorption** (blue arrow) and **Diffusion** (yellow arrow) are possible outcomes.
- **Electron-phonon coupling** (1), **Electronic friction** (2), and **Phonon-adsorbate interaction** (3) are key processes.
- **only electrons of metal absorb laser**
- **electron-hole pairs** thermalize fast  
⇒ “**hot**” Fermi-Dirac-distribution
- electrons transfer energy to ion lattice, via **1) electron-phonon coupling**
- **equilibration within ps-timescale**  
⇒ Thus, for few ps **two temperatures**:
  - $T_{el}$  - electron temperature
  - $T_{ph}$  - phonon temperature
- both can **couple** to adsorbed **molecule**
- low electron heat **capacity** ⇒  $T_{el}$  higher



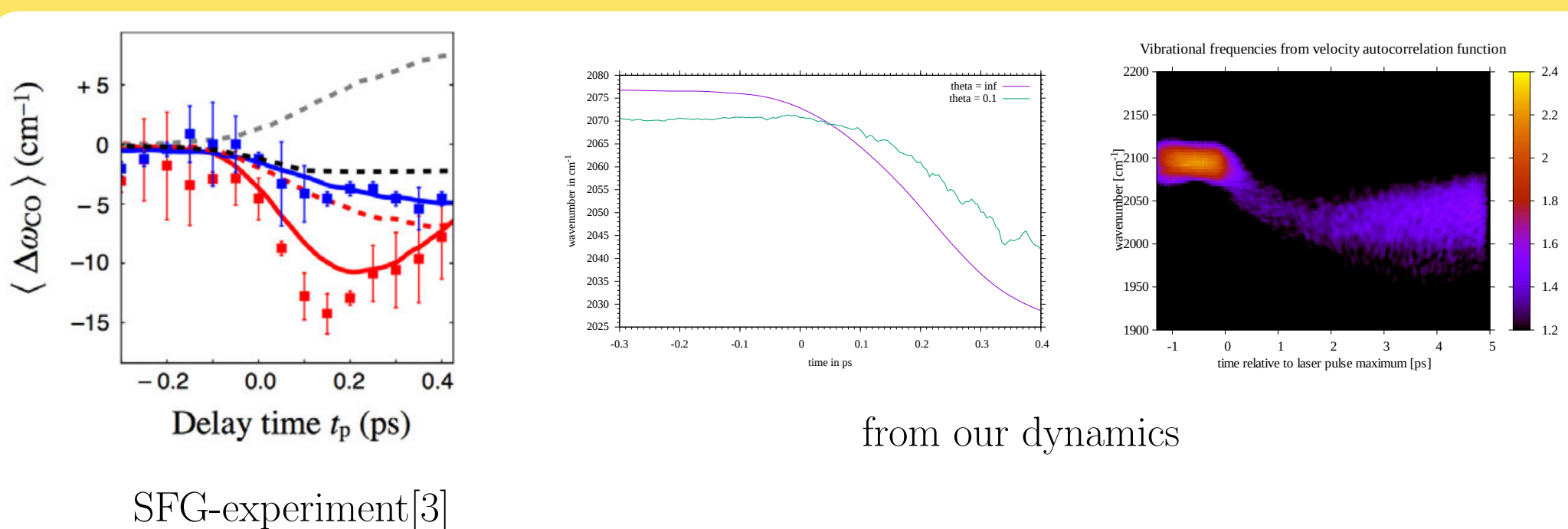
## Results

### Desorption

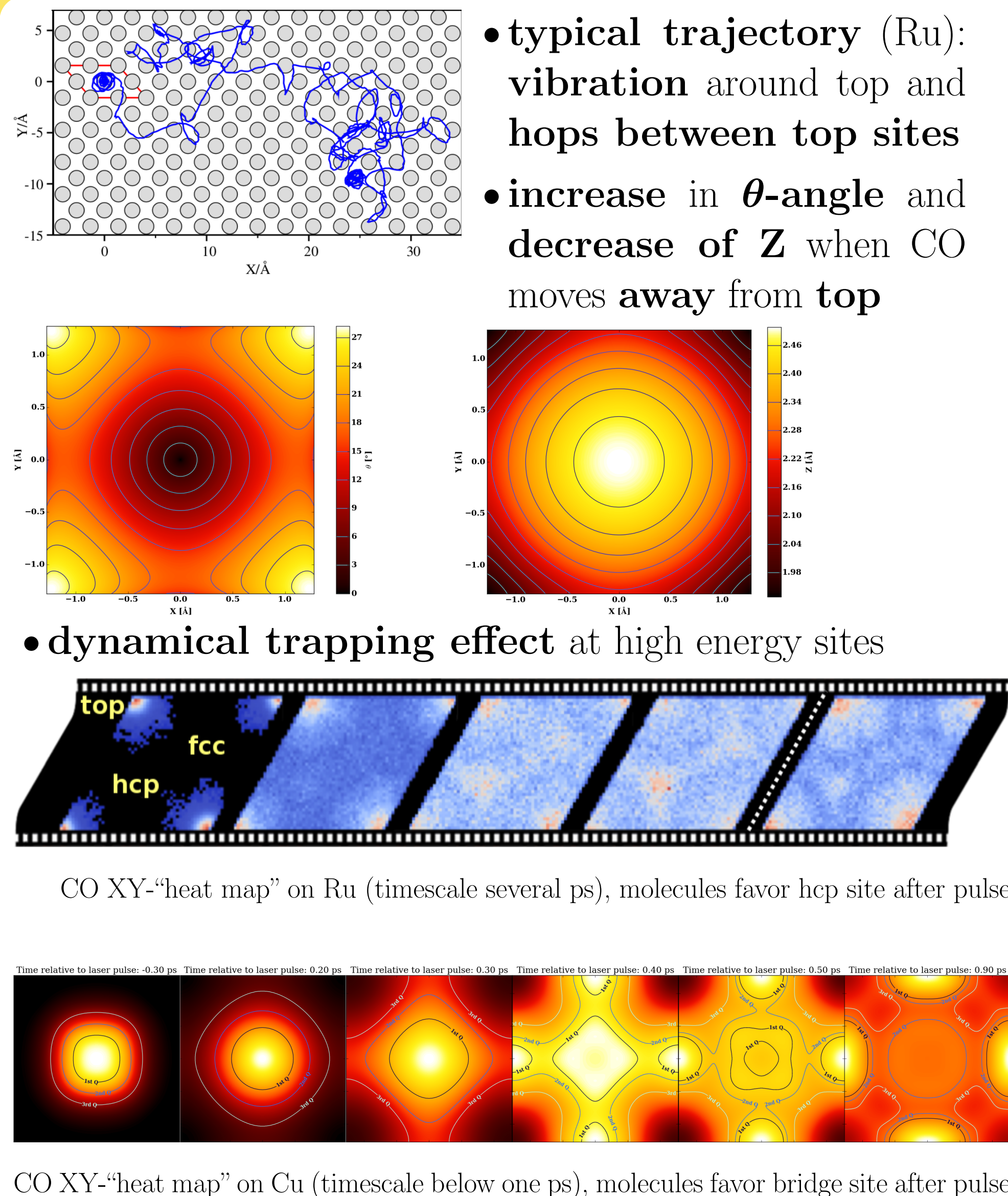


- **fluence dependence** of desorption yield close to experiment (Ru)
- GLO affects desorption differently

### CO-stretch Vibration (Data for Cu)



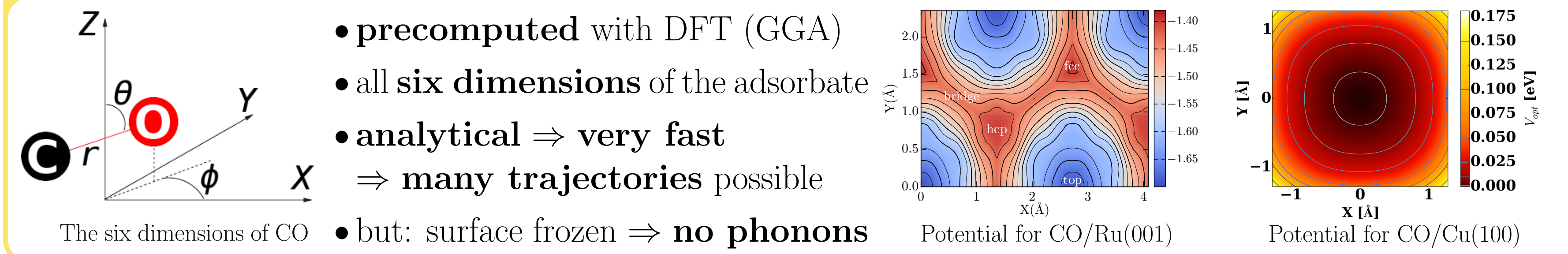
### Diffusion



- **typical trajectory (Ru):** vibration around top and hops between top sites
- increase in  $\theta$ -angle and decrease of  $Z$  when CO moves away from top
- **dynamical trapping effect** at high energy sites

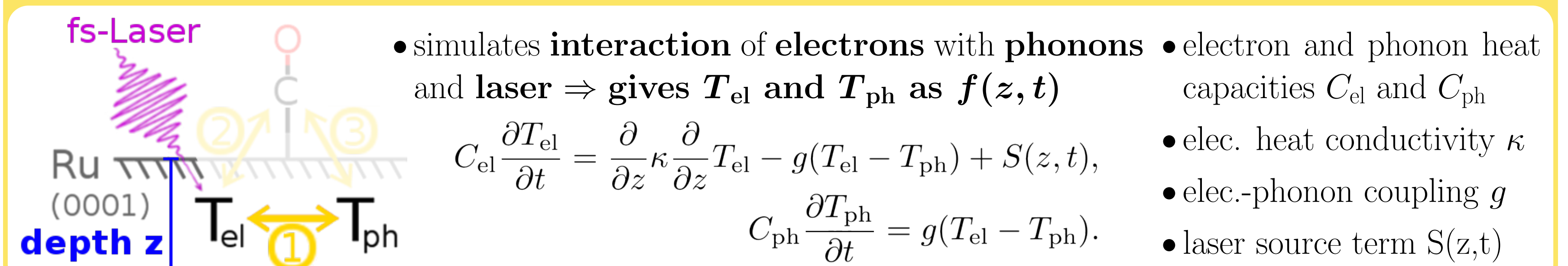
## Models and Methods

### Basis of the Dynamics: the Six-dimensional Potential Energy Surfaces (PES)[4][5]

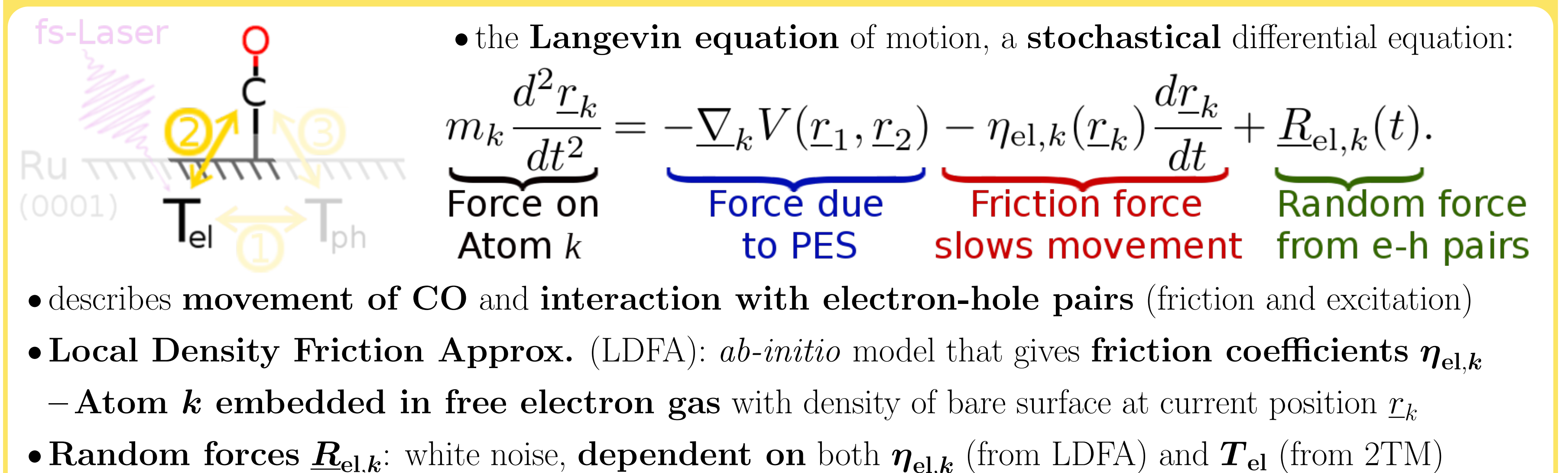


- **precomputed** with DFT (GGA)
- all **six dimensions** of the adsorbate
- **analytical** ⇒ **very fast**  
⇒ **many trajectories** possible
- but: surface frozen ⇒ **no phonons**

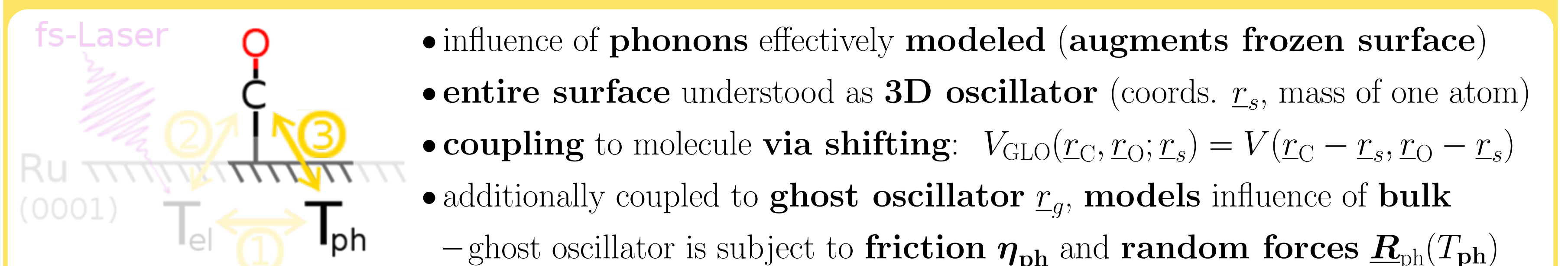
### Two-Temperature Model (2TM)[6]



### Electronic Friction: Langevin Dynamics[7] and Local Density Friction Approximation (LDFA)[8]



### Inclusion of Phonons: Generalized Langevin Oscillator(GLO)-model[9]



## Conclusions

- **system:** CO on Ru(001) or Cu(100)
- **full-dimensional (6D)** simulation of dynamics driven by **fs-laser irradiation**
- **first principles**, no “free” parameters
- accounting for **1) electronic friction** and excitation by hot electron-hole pairs (via LDFA and the Langevin approach) and **2) substrate motion** (via GLO)
- detailed time- and space-resolved insights

## Outlook

- better description of electronic friction:  $\eta_{el}$  as  $f(T_{el})$  and going **beyond LDFA** ⇒ Long term goal: **friction tensors**[10]
- further **2TM improvements**, consider **non-equilibrium lattice (NLM)**[11]
- directly model **SFG influence** by explicitly including **excitation from IR-pulse**
- **larger super cells** for other **coverages** and **intermolecular interaction**
- simulate **more complex systems**: hydrocarbons; coadsorbate CO + H or O

## References

- [1] M. Bonn, S. Funk, Ch. Hess, D.N. Denzler *et al.*, *Science* **285**, 1042 (1999).
- [2] M. Dell’Angela, T. Anniyev, M. Beye, R. Coffee *et al.*, *Science* **339**, 1302 (2013).
- [3] K. Inoue, K. Watanabe, T. Sugimoto *et al.*, *Phys.Rev.Lett.* **117**, 186101 (2016).
- [4] G. Füchsel, J.C. Tremblay, and P. Saalfrank, *J.Chem.Phys.* **141**, 094704 (2014).
- [5] R. Marquardt, F. Cuvelier, R.O. Olsen *et al.*, *J.Chem.Phys.* **132**, 074108 (2010).
- [6] S.I. Anisimov, B.L. Kapeliovich and T.L. Perelman, *Sov.Phys.-JETP* **39**, 375 (1974).
- [7] M. Head-Gordon and J.C. Tully, *J.Chem.Phys.* **103**, 10137 (1995).
- [8] J.I. Juaristi, M. Alducin, R. Díez Muiño *et al.*, *Phys.Rev.Lett.* **100**, 116102 (2008).
- [9] H.F. Busnengo, M.A. Di Césare, W. Dong *et al.*, *Phys.Rev.B* **72**, 125411 (2005).
- [10] R. J. Maurer, M. Askerka, V. S. Batista, J. C. Tully *Phys.Rev.B* **94**, 115432 (2016).
- [11] L. Waldecker, R. Bertoni, R. Ernstorfer, J. Vorberger *Phys.Rev.X* **6**, 021003 (2016).