Multi-dimensional Femtosecond-laser induced dynamics of CO on metals:

ACCOUNTING FOR ELECTRONIC FRICTION AND SURFACE MOTION WITH COMBINED MODELS

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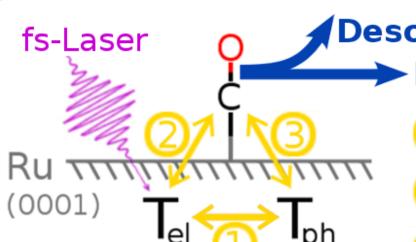
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Introduction

Motivation

- Aim: gain precise understanding of adsorbate bonding on metals ⇒ Important for Catalysis
- Why femtosecond(fs)-lasers?
- -produce non-equilibrium 2-T-states ⇒ different than normal heating
- -further **tool** besides STM and scattering -direct future applications possible \Rightarrow "femtochemistry"[1]
- Why CO/Ru(001) and CO/Cu(100)?
- -both are well studied **model systems**
- -recently, interesting **fs-laser experiments**[2][3]
- -also, **ab-initio** based **6-dim. potentials** available [4][5]

How do fs-lasers affect adsorbate-metal systems?



Desorption

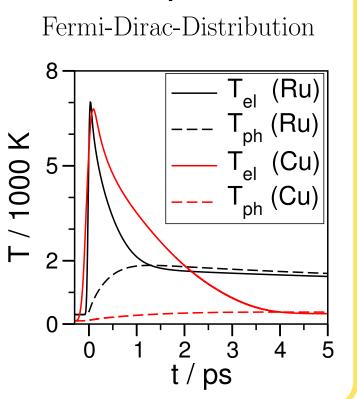
Diffusion (and possibly Reactions)

- Electron-phonon coupling
- **Electronic friction**
- Phonon-adsorbate interaction
- only electrons of metal absorb laser f(E) low T_{el}
- electron-hole pairs thermalize fast ⇒ "hot" Fermi-Dirac-distribution
- electrons transfer energy to ion lattice,
- via 1 electron-phonon coupling • equilibration within ps-timescale
- \Rightarrow Thus, for few ps **two temperatures**:
 - $-T_{\rm el}$ electron temperature
 - $-T_{\rm ph}$ phonon temperature
- both can **couple** to adsorbed **molecule**
- low electron heat **capacity** \Rightarrow $T_{\rm el}$ higher

Femtochemistry example

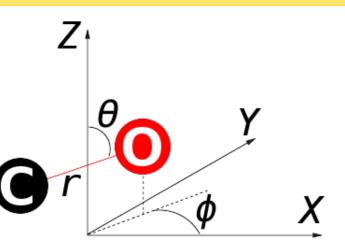


high



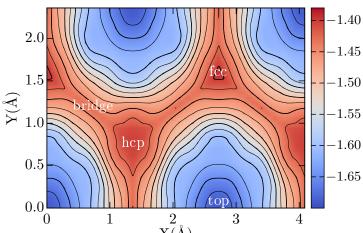
Models and Methods

Basis of the Dynamics: the Six-dimensional Potential Energy Surfaces (PES)[4][5]



The six dimensions of CO

- **precomputed** with DFT (GGA)
- all **six dimensions** of the adsorbate
- analytical \Rightarrow very fast ⇒ many trajectories possible
- but: surface frozen \Rightarrow **no phonons**



fs-Laser

Potential for CO/Ru(001)

Potential for CO/Ru(001)

Two-Temperature Model (2TM)[6]

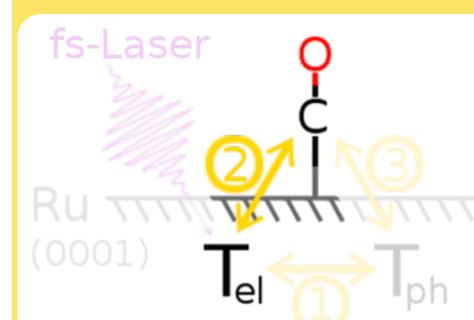
• describes interaction of electrons with phonons and laser

$$C_{\rm el} \frac{\partial T_{\rm el}}{\partial t} = \frac{\partial}{\partial z} \kappa \frac{\partial}{\partial z} T_{\rm el} - g(T_{\rm el} - T_{\rm ph}) + S(z, t),$$

$$C_{\rm ph} \frac{\partial T_{\rm ph}}{\partial t} = g(T_{\rm el} - T_{\rm ph}).$$
Ru Tel (0001)
$$C_{\rm ph} \frac{\partial T_{\rm ph}}{\partial t} = g(T_{\rm el} - T_{\rm ph}).$$
Ru Tel (10001)
$$C_{\rm ph} \frac{\partial T_{\rm ph}}{\partial t} = g(T_{\rm el} - T_{\rm ph}).$$

 \Rightarrow get $T_{\rm el}$ and $T_{\rm ph}$ as f(z,t) from laser parameters and material properties

Electronic Friction: Langevin Dynamics[7] and Local Density Friction Approximation (LDFA)[8]

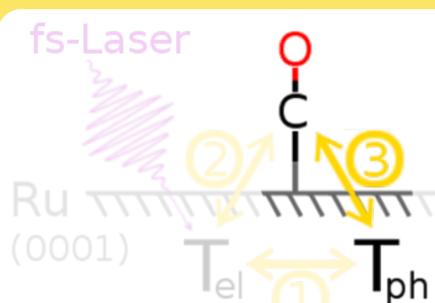


• Langevin equation of motion, a stochastical differential equation:

$$m_k \frac{d^2\underline{r}_k}{dt^2} = -\underline{\nabla}_k V(\underline{r}_1,\underline{r}_2) - \eta_{\mathrm{el},k}(\underline{r}_k) \frac{d\underline{r}_k}{dt} + \underline{R}_{\mathrm{el},k}(t).$$
 Force on Atom k Force due to PES slows movement from e-h pairs

- describes movement of CO and interaction with electron-hole pairs (friction and excitation)
- Local Density Friction Approx. (LDFA): simple model to get friction coefficients $\eta_{el,k}$
- -Atom k embedded in free electron gas with density of bare surface at current position \underline{r}_k
- Random forces $\underline{R}_{el,k}$: white noise, dependent on both $\eta_{el,k}$ (from LDFA) and T_{el} (from 2TM)

Inclusion of Phonons: Generalized Langevin Oscillator(GLO)-model[9]



- influence of phonons effectifely modeled (augments frozen surface)
- entire surface understood as 3D oscillator (coords. \underline{r}_s , mass 1 atom)
- coupling to molecule via shifting: $V_{\text{GLO}}(\underline{r}_{\text{C}},\underline{r}_{\text{O}};\underline{r}_{s}) = V(\underline{r}_{\text{C}} \underline{r}_{s},\underline{r}_{\text{O}} \underline{r}_{s})$
- additionally coupled to **ghost oscillator** \underline{r}_q , **models** influence of **bulk** -ghost oscillator is subject to friction $\eta_{\rm ph}$ and random forces $\underline{R}_{\rm ph}(T_{\rm ph})$

Results

• desorption mainly dur-5 ing first 50 ps Z (Å) • fluence dependence of 3 desorption yield close to experiment • no barrier in PMF t(ps)■ MDEF-GLO, n=5.17 T=1000 KT=4000 K

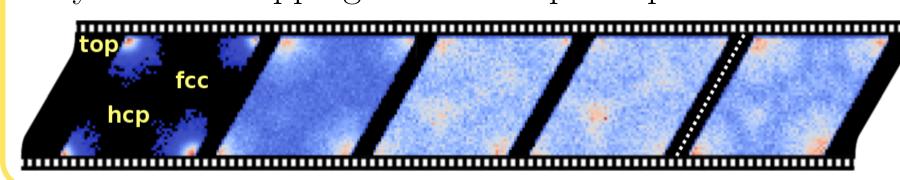
300

fluence F / J/m²

Desorption (Data for Ru)

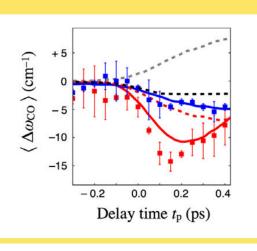
Diffusion (Data for Ru, but Cu similar)

- typical trajectory: hops between top sites and vibration
- increase in θ -angle when CO moves away from top
- overall, very large diffusion
- also, nonisotropic diffusion behaviour observed:
- dynamical trapping effect at hcp site predicted

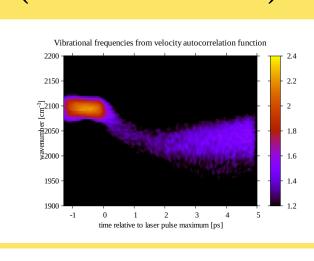


Vibrations (Data for Cu)

• Frequency-shift from time-resolved SFG



molecule-surface distance Z / Å



• Preliminary results from our dynamics

Conclusions

- 6D Langevin dynamics of CO on Ru and Cu
- based on first principles, no "free" parameters
- accounting for (via LDFA) electronic friction, hot electron excitation and (via GLO) substrate motion
- allows for detailed time- and space-resolved insights
- no physisorbed state, molecules desorp directly

Outlook

- better electronic friction ($\eta(T_{\rm el})$) and beyond LDFA) \Rightarrow Long term goal: use tensorial friction (exact)
- non-equilibrium lattice model (NLM) instead 2TM
- simulate other coveages (bigger super cells)
- simulate bigger systems (CO + H; hydrocarbons)
- include interaction between adsorbate molecules

References

- [1] M. Bonn, S. Funk, Ch. Hess, D.N. Denzler et al., Science 285, 1042 (1999).
- [2] M. Dell'Angela, T. Anniyev, M. Beye, R. Coffee et al., Science **339**, 1302 (2013).
- [3] K. Inoue, K. Watanabe, T. Sugimoto et al., Phys. Rev. Lett. 117, 186101 (2016).
- [4] G. Füchsel, J.C. Tremblay, and P. Saalfrank, *J. Chem. Phys.* **141**, 094704 (2014).

[5] R. Marquardt, F. Cuvelier, R.O. Olsen et al., J. Chem. Phys. 132, 074108 (2010).

- [7] M. Head-Gordon and J.C. Tully, *J. Chem. Phys.* **103**, 10137 (1995). [8] J.I. Juaristi, M. Alducin, R. Díez Muiño et al., Phys. Rev. Lett. 100, 116102 (2008).
- [6] S.I. Anisimov, B.L. Kapeliovich and T.L. Perelman, Sov. Phys. JETP 39, 375 (1974). [9] H.F. Busnengo, M.A. Di Césare, W. Dong et al., Phys. Rev. B 72, 125411 (2005).