## Femtosecond-laser induced dynamics of CO on Ru(0001):

### NEW INSIGHTS FROM A HOT-ELECTRON, ELECTRONIC FRICTION MODEL INCLUDING SURFACE MOTION

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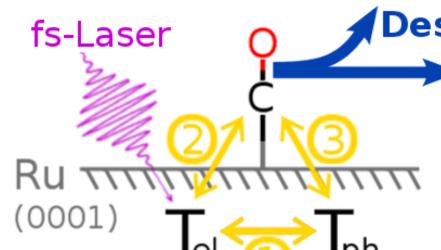
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#### Introduction

#### Motivation

- research on small molecules adsorbed to metals is important for:
- -catalytic applications
- -fundamental understanding of bonding
- femtosecond(fs)-lasers  $\Rightarrow$  very valuable tool -allow for investigations on small timescales - open up new processes compared to heating or "normal" lasers  $\Rightarrow$  femtochemistry, e.g. [1]:
- may enable specific control over catalytic reactions  $\Rightarrow$  photocatalysis
- specific motivation for system CO/Ru(0001) -experimentally well studied regarding fs-laser irradiation, e.g. [2, 3] -fulldimensional ab-initio potential recently developed in our group [4] -details of this indicate interpretation of experiment [3] may be wrong

#### How does fs-laser-irradiation affect metal surfaces?



Desorption **Diffusion** (and possibly Reactions)

- (1) Electron-phonon coupling (2) Electronic friction
- 3 Phonon-adsorbate interaction

Thermal

excitation

0.8 eV

1.8 eV

t/ps

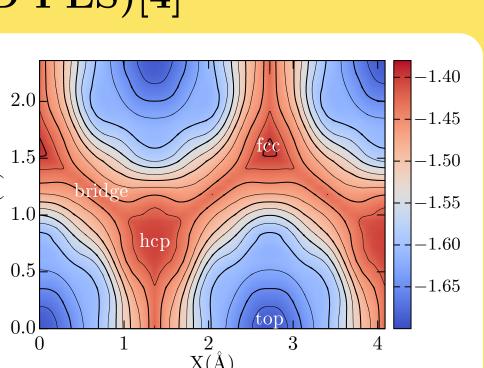
from [1]

- metals: ion lattice plus quasi-free electron gas
- visible light is absorbed only by the electrons
- produced electron-hole pairs thermalize quickly  $\Rightarrow$  "hot" Fermi-Dirac-distribution (after  $\sim 10 \text{ fs}$ )
- electrons transfer part of energy to ion lattice, via (1) electron-phonon coupling (phonons = lattice vibrations; quasi-particles) -electrons couple to phonons as their fast movement causes "shockwaves" in ion lattice -equilibration process completes after  $\sim 1 \text{ ps}$
- $\Rightarrow$  Thus, with fs-lasers, two different temperatures:  $-T_{\rm el}$  - electron temperature
  - $-T_{\rm ph}$  phonon temperature
- can be simulated using a Two-Temperature Model (2TM)[5] (see right)

#### Models and Methods

#### Six-dimensional Potential Energy Surface (6D PES)[4]

- Basis for dynamics: precomputed PES from DFT (rPBE + D2)
  - all 6 dimensions of the adsorbate
  - $\bullet$  analytical PES and gradients  $\Rightarrow$  very fast
  - ⇒ number and length of trajectories can be large
  - downsides: surface atoms frozen  $\Rightarrow$  no phonons
  - had to be constructed first



#### Two-Temperature Model (2TM)[5]

• describes interaction of metal with laser, using two differential equations:

$$C_{\rm el} \frac{\partial T_{\rm el}}{\partial t} = \frac{\partial}{\partial z} \kappa \frac{\partial}{\partial z} T_{\rm el} - g(T_{\rm el} - T_{\rm ph}) + S(z, t),$$
$$C_{\rm ph} \frac{\partial T_{\rm ph}}{\partial t} = g(T_{\rm el} - T_{\rm ph}).$$

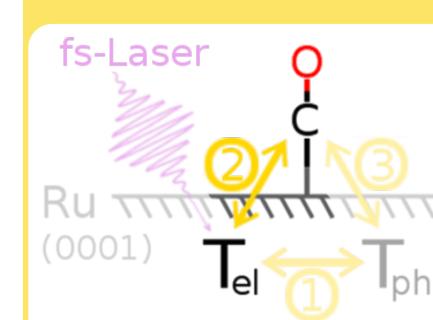
depth z  $\Rightarrow$  get  $T_{\rm el}$  and  $T_{\rm ph}$  as f(z,t) from laser parameters and material properties: -electron and phonon heat capacities  $C_{\rm el}$  and  $C_{\rm ph}$ 

- electron heat conductivity  $\kappa$ 

fs-Laser

- -laser wavelength  $\lambda$  (affects penetretion depth into material) - (effective) absorbed fluence F (energy/area)
- -pulse duration  $\tau$  (all three appear in the "source term" S(z,t)) - electron-phonon coupling constant g

#### Electronic Friction: Langevin Dynamics[6] and Local Density Friction Approximation (LDFA)[7]

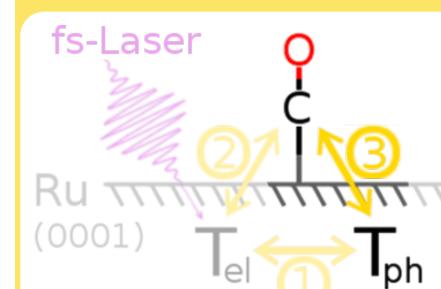


• Langevin equation of motion, a stochastical differential equation:

$$m_k \frac{d^2\underline{r}_k}{dt^2} = -\underline{\nabla}_k V(\underline{r}_1,\underline{r}_2) - \eta_{\mathrm{el},k}(\underline{r}_k) \frac{d\underline{r}_k}{dt} + \underline{R}_{\mathrm{el},k}(t).$$
 Force on Atom  $k$  Force due to PES slows movement from e-h pairs

- describes movement of CO on the PES and interaction with electron-hole pairs (friction and excitation)
- Local Density Friction Approx. (LDFA): most simple model to calculate friction coefficients  $\eta_{el,k}$ -Atom k embedded in free electron gas with density of bare surface at current position  $\underline{r}_k$
- Random forces  $\underline{R}_{el,k}$ : gaussian white noise, dependent on both  $\eta_{el,k}$  (from LDFA) and  $T_{el}$  (from 2TM) -justified by the 2. fluctuation dissipation theorem [11], which relates friction and thermal movement

#### Inclusion of Phonons: Generalized Langevin Oscillator(GLO)-model[8, 9, 10]



- influence of phonons modeled in an effective way (augments frozen surface)
- entire surface understood as 3D oscillator (coordinates  $\underline{r}_s$ , mass  $m_s = m_{\rm Ru}$ )
- coupling to molecule via shifting:  $V_{\text{GLO}}(\underline{r}_{\text{C}},\underline{r}_{\text{O}};\underline{r}_{s}) = V(\underline{r}_{\text{C}} \underline{r}_{s},\underline{r}_{\text{O}} \underline{r}_{s})$
- additionally coupled to ghost oscillator  $\underline{r}_q$  to model influence of the bulk
- -ghost oscillator is subject to friction  $\eta_{\rm ph}$  and random forces  $\underline{R}_{\rm ph}(T_{\rm ph})$

#### Results

#### Desorption • desorption mainly 10 during first 50 ps • fluence dependence (Å)of desorption yield Nclose to experiment • no "precursor state" as suggested by [3] t(ps)• no barrier in PMF exp., n=4.81▲ MDEF, n=6.99 ■ MDEF-GLO, n=5.17 — T=0 K ─ T=300 K - T=500 K — T=1000 K T=1500 KT=2000 K— T=4000 K 200 300 fluence F / J/m<sup>2</sup> molecule-surface distance Z / Å

# Diffusion

- typical trajectory: hops between top sites and vibration • increase in  $\theta$ -angle when CO moves away from top -• overall, very large diffusion
- also, nonisotropic diffusion behaviour observed:
- dynamical trapping effect at hcp site predicted
- $\Rightarrow$  possible alternative explanation to "precursor state" [3]

#### Conclusions

- 6D Langevin dynamics of CO on Ru(0001)
- based on first principles, no "free" parameters
- accounting for (via LDFA) electronic friction, hot electron excitation and (via GLO) substrate motion
- allows for detailed time- and space-resolved insights
- no physisorbed state, molecules desorp directly

#### Outlook

- calculation of RIXS spectra currently performed
- employ better electonic friction (beyond LDFA)
- enhance 2TM with electron-electron-scattering
- simulate other coverages (here only 0.25ML)
- include interaction between adsorbate molecules

#### References

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