

# Selective photoinactivation of protein function through environment-sensitive switching of singlet oxygen generation by photosensitizer

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Chromophore-assisted light inactivation is a promising technique to inactivate selected proteins with high spatial and temporal resolution in living cells, but its use has been limited because of the lack of a methodology to prevent nonspecific photodamage in the cell owing to reactive oxygen species generated by the photosensitizer. Here we present a design strategy for photosensitizers with an environment-sensitive off/on switch for singlet oxygen ( $^1\text{O}_2$ ) generation, which is switched on by binding to the target, to improve the specificity of protein photoinactivation.  $^1\text{O}_2$  generation in the unbound state is quenched by photoinduced electron transfer, whereas  $^1\text{O}_2$  generation can occur in the hydrophobic environment provided by the target protein, after specific binding. Inositol 1,4,5-trisphosphate receptor, which has been suggested to have a hydrophobic pocket around the ligand binding site, was specifically inactivated by an environment-sensitive photosensitizer-conjugated inositol 1,4,5-trisphosphate receptor ligand without  $^1\text{O}_2$  generation in the cytosol of the target cells, despite light illumination, demonstrating the potential of environment-sensitive photosensitizers to allow high-resolution control of generation of reactive oxygen species in the cell.

activatable photosensitizer | boron dipyrromethene derivative |  
electron transfer | inositol 1,4,5-trisphosphate receptor

Chromophore-assisted light inactivation (CALI)(1) is a technique with great potential to inactivate proteins with high spatial and temporal resolution by using an antibody to direct a suitable fluorophore specifically to the protein of interest. Illumination induces local generation of reactive oxygen species (ROS), which react chemically with the adjacent antigen and inactivate it. Although CALI is a powerful technique, its use has been limited by the complexity of the procedures (i.e., the need to deliver a labeled antibody into cells or to use a laser as the light source). Several groups have reported alternative approaches. Genetically targeted CALI is one such method, in which the target protein is tagged with a tetracysteine tag that is recognized by a membrane-permeant biarsenical chromophore (FIAsH) (2, 3), or tagged with GFP (4–6). However, these methods also cause nonspecific damage, owing to the nonspecific binding of the biarsenical chromophore to cysteine-rich proteins (3, 7) in FIAsH-mediated photoinactivation, or to the use of a relatively high-power laser in EGFP-mediated CALI (4, 5). Current implementations of the CALI technique leave much to be desired, and highly specific inactivation of a protein of interest would require a methodology to control ROS generation by the photosensitizer in the cells with high spatial resolution.

We present here an approach for designing photosensitizers with an environment-controlled off/on switch for singlet oxygen ( $^1\text{O}_2$ ) generation to improve the specificity of CALI. We have developed environment-sensitive photosensitizers (ESPers), which are acti-

vated by recognition of the hydrophobic (low-polarity) environment of the target protein, i.e., recognition of the appropriate environment switches on local generation of  $^1\text{O}_2$ , whereas  $^1\text{O}_2$  is not generated in the polar cytosolic environment (Fig. 1*a*). The value of ESPers to control tightly the specificity of protein photoinactivation in the CALI technique was demonstrated by applying one of our ESPers for highly specific inactivation of inositol 1,4,5-trisphosphate receptor (IP<sub>3</sub>R).

## Results

**Photoinduced Electron Transfer as a Mechanism to Control Photosensitization.** We previously developed small molecule-based CALI for IP<sub>3</sub>R by using a conventional photosensitizer (malachite green), and the physiological function of IP<sub>3</sub>R was analyzed (8, 9). However, it would be advantageous for further biological applications if an activatable photosensitizer, which generates ROS only when it binds to IP<sub>3</sub>R, could be developed. With this in mind, we focused on hydrophobic environment as a putative on-switch for an activatable photosensitizer. Intracellular proteins (e.g., the receptor of interest) usually consist of both hydrophobic and hydrophilic domains, and we considered that a hydrophobic photosensitizer moiety, if it is conjugated to a specific ligand of the target protein by a suitable linker, could be delivered to a hydrophobic domain near the ligand binding site inside the target protein by hydrophobic interaction. In the case of IP<sub>3</sub>R, it has been suggested that IP<sub>3</sub>R has a hydrophobic pocket around the binding site, based on the finding that IP<sub>3</sub> derivatives bearing a hydrophobic moiety have high binding affinity (10, 11). Thus, photosensitizers that could generate ROS only when they are activated by recognition of the hydrophobic environment around IP<sub>3</sub>R should cause little or no nonspecific damage in the cytosol, where the environment is polar (Fig. 1*a*). To realize this concept, we selected photoinduced electron transfer (PeT) as a switch mechanism to control the  $^1\text{O}_2$  generation of photosensitizers. PeT is a well known mechanism through which the fluorescence of a fluorophore is quenched by electron transfer from the PeT donor to the lowest singlet-excited fluorophore (12, 13). Photosensitization is well known to

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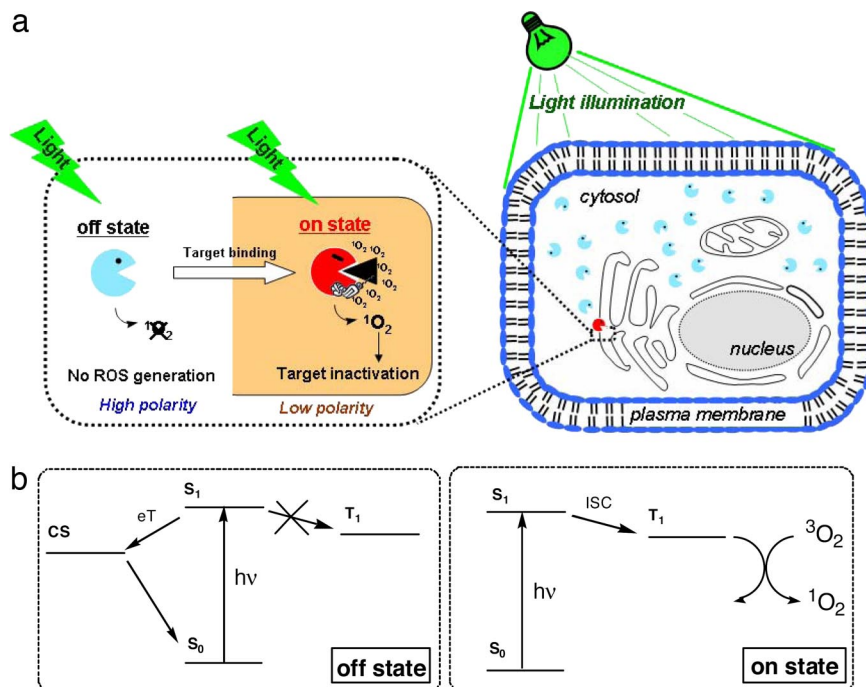
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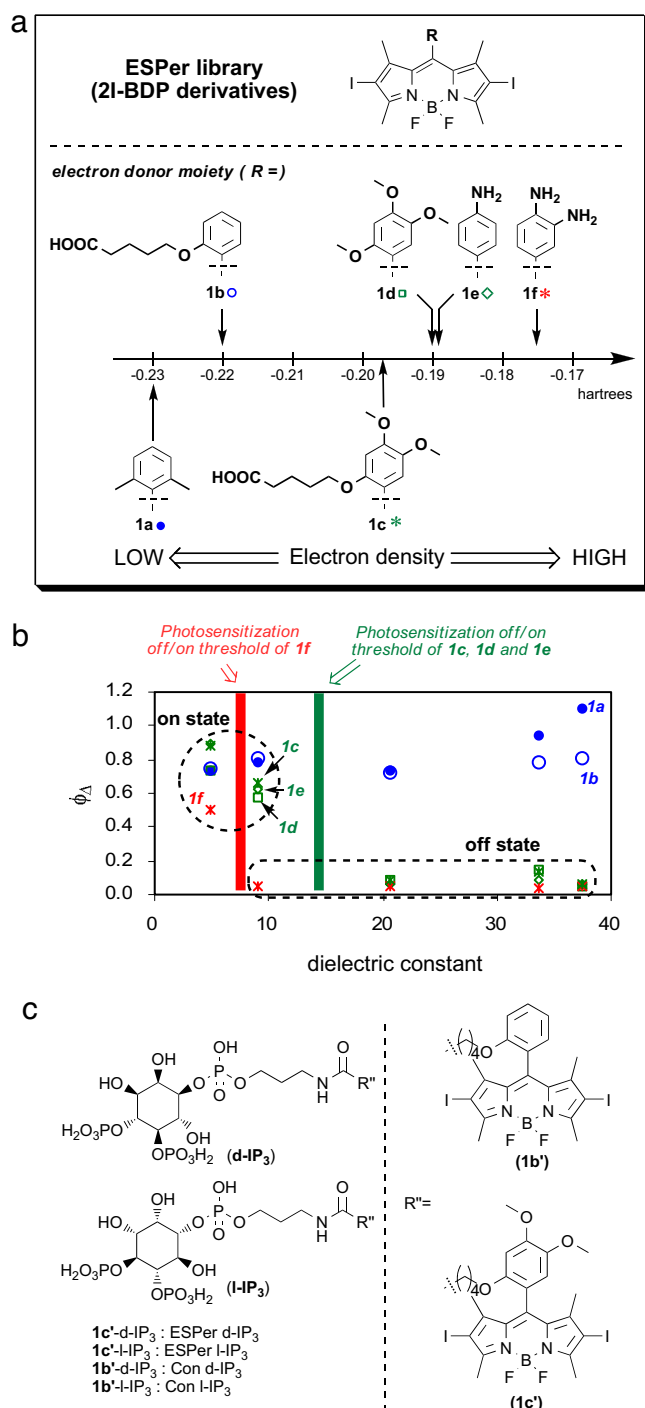
**Fig. 1.** Photoinduced electron transfer as a mechanism for controlling photosensitization in biological systems. (a) Schematic representation of protein photoinactivation by using ESPers. In the cytosolic polar environment, ESPers are in the off state of photosensitization (indicated by light blue) and do not generate  $^1O_2$ . In the hydrophobic environment of the target, ESPers are in the on state of photosensitization (indicated by red) and generate  $^1O_2$ , inducing inactivation of the nearby protein of interest. (b) Energy diagram of the ESPers. ISC, intersystem crossing; CS, charge separation; eT, electron transfer;  $S_0$ , singlet ground state;  $S_1$ , lowest singlet excited state;  $T_1$ , lowest triplet excited state.

proceed also via the lowest singlet excited state ( $S_1$ ), so it seemed reasonable that the photosensitization process could be controlled by PeT (Fig. 1b). Indeed, our preliminary results showed that  $^1O_2$  generation of erythrosin derivatives upon light illumination was efficiently quenched by introducing electron-donating substituents into the benzene moiety (S. Kamakura, Y.U., and T.N., unpublished results). The PeT process is known to depend on the highest occupied molecular orbital (HOMO) energy level of the electron donor and the solvent polarity, so we designed and synthesized a series of photosensitizer derivatives by attaching an electron donor moiety to a photosensitizing chromophore for the specific inactivation of IP<sub>3</sub>R in a hydrophobic environment.

**Design and Synthesis of a Library of Candidate ESPers.** We recently reported development of 4,4-difluoro-2,6-diiodo-1,3,5,7-tetramethyl-4-bora-3a,4a-diaza-s-indacene (2I-BDP) as a photosensitizer that generates  $^1O_2$  uniformly in various solvents ranging from H<sub>2</sub>O to *n*-hexane (14). We hypothesized that introduction of a suitable electron-donor moiety into 2I-BDP would allow the photosensitizing ability to be controlled precisely by means of PeT. To test this principle, we synthesized various 2I-BDP derivatives bearing a substituted benzene moiety (1a–1f), in which the use of various substituents (methyl, methoxy, or amino) as electron donors was expected to allow the HOMO energy level to be finely tuned (Fig. 2a). These 2I-BDP derivatives have high values of extinction coefficient ( $\epsilon \approx 1 \times 10^5$  M<sup>-1</sup>·cm<sup>-1</sup>) at  $\approx 530$  nm [supporting information (SI) Fig. 5]. We then examined the ability of these 2I-BDP derivatives to generate  $^1O_2$  in various solvents, from polar to nonpolar, by observing the disappearance of 1,3-diphenylisobenzofuran (DPBF), which is known to react with  $^1O_2$  under light illumination (15) (for detailed experimental procedures, see SI Materials and Methods). The relative efficiencies of  $^1O_2$  generation ( $\phi_\Delta$ ) of 1a–1f in each solvent are summarized in Fig. 2b. These 2I-BDP deriva-

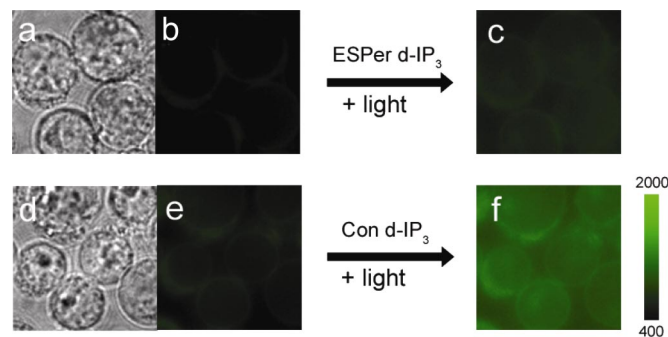
tives had almost identical absorbance maximum wavelengths (SI Fig. 5), but the  $\phi_\Delta$  of certain 2I-BDP derivatives, i.e., those having an electron donor moiety with high HOMO energy (1c–1f), was significantly quenched in polar solvents such as CH<sub>3</sub>CN and MeOH, compared with that of 1a and 1b, whose benzene moiety has a relatively low HOMO energy. These results suggested that  $\phi_\Delta$  of these 2I-BDP derivatives could indeed be controlled by PeT. We then examined the effect of solvent polarity. We have recently developed environment-sensitive fluorescence probes based on the boron dipyrromethene (BODIPY) fluorophore, whose fluorescence properties were shown to be controlled by PeT and by solvent effects on the PeT (see also SI Fig. 6). Because 2I-BDP derivatives are iodinated derivatives of BODIPY, they may behave similarly. Fig. 2b shows that  $\phi_\Delta$  of 1c–1f was dependent on the solvent polarity (i.e., dielectric constant, DC). Compounds 1c–1e were unable to generate  $^1O_2$  (i.e., they were in the off state of photosensitization) in solvents more polar than acetone (DC  $\approx 20.7$ ), whereas the  $^1O_2$ -generating ability was restored (i.e., the compounds were in the on state of photosensitization) in solvents less polar than CH<sub>2</sub>Cl<sub>2</sub> (DC  $\approx 9.14$ ). On the other hand, 1f was still in the off state in CH<sub>2</sub>Cl<sub>2</sub>, and was switched on only in solvents less polar than CHCl<sub>3</sub> (DC  $\approx 4.81$ ), indicating that the threshold for off–on switching of  $^1O_2$  generation is dependent on the HOMO energy level of the benzene moiety. These results indicated that appropriately designed 2I-BDP derivatives, having a HOMO energy around  $-0.17$  to  $-0.19$  hartree, could be used as environment-sensitive photosensitizers (ESPers), which would be activated by recognizing a hydrophobic environment, and would generate  $^1O_2$  in such an environment, but not in a polar environment. We then examined whether ESPers could be activated by recognizing a hydrophobic environment at the cellular level.

**Design of ESPer-Conjugated IP<sub>3</sub> Ligand for Specific Photoinactivation of IP<sub>3</sub>R.** We next designed and synthesized an ESPer-conjugated IP<sub>3</sub>R ligand for photoinactivation of IP<sub>3</sub>R and examined whether



**Fig. 2.** Development of a library of ESPers making use of the solvent effect on PeT. (a) Structures of 2I-BDP derivatives with various benzene moieties. (b) Relationship between the relative efficiency of <sup>1</sup>O<sub>2</sub> generation ( $\phi_A$ ) of 2I-BDP derivatives (1a–1f) and the dielectric constant of the solvent. Solvents used in this study were CH<sub>3</sub>CN, MeOH, acetone, CH<sub>2</sub>Cl<sub>2</sub>, and CHCl<sub>3</sub>, whose dielectric constants are 37.5, 33.6, 20.7, 9.14, and 4.81, respectively. Filled blue circle, 1a; open blue circle, 1b; green star, 1c; open green square, 1d; open green diamond, 1e; red star, 1f. (c) Structures of ESPer 1c-conjugated IP<sub>3</sub> derivatives and control (Con) photosensitizer 1b-conjugated IP<sub>3</sub> derivatives; d- and l- refer to the D and L absolute configurations of inositol 1,4,5-trisphosphate.

it could generate <sup>1</sup>O<sub>2</sub> specifically in the vicinity of IP<sub>3</sub>R. IP<sub>3</sub>R has been suggested to have a hydrophobic pocket around the binding site, which was estimated to be similar in polarity to CH<sub>2</sub>Cl<sub>2</sub>



**Fig. 3.** Imaging of photosensitizer-induced nonspecific cytosolic <sup>1</sup>O<sub>2</sub> generation in the presence of ESPer d-IP<sub>3</sub>, compared with that in the presence of Con d-IP<sub>3</sub>, using a fluorescence probe for <sup>1</sup>O<sub>2</sub> (DMAX-2). Permeabilized DT40 cells (wild type) were loaded with 3  $\mu$ M ESPer d-IP<sub>3</sub> or Con d-IP<sub>3</sub> and 10  $\mu$ M DMAX-2, followed by illumination with green light (BP530–550 nm, 1.5 mW/cm<sup>2</sup>, 20 sec). Fluorescence images of DMAX-2 were acquired with a fluorescence microscope by excitation with blue light (BA470–490) before and after green light illumination. (a–c) Single DT40 cells under transmitted light (a) and fluorescence image excited with blue light before (b) and after (c) excitation of ESPer d-IP<sub>3</sub> with green light. (d–f) Single DT40 cells under transmitted light (d) and fluorescence image excited with blue light before (e) and after (f) excitation of Con d-IP<sub>3</sub> with green light. Color scale on the right is the relative fluorescence intensity. (Scale bar, 10  $\mu$ m.)

(DC  $\approx$  9.14) by the use of environment-sensitive fluorescence probes (SI Materials and Methods and SI Fig. 6). Thus, we considered that 1c and 1d (Fig. 2a), which have a trimethoxybenzene moiety, might be suitable ESPers to recognize the environment around IP<sub>3</sub>R. We synthesized an IP<sub>3</sub> derivative bearing 1c (Fig. 2c). As a control (Con) compound, we also synthesized an IP<sub>3</sub> derivative bearing 1b, whose off/on switch for <sup>1</sup>O<sub>2</sub> generation is constitutively on (Fig. 2a and b), regardless of the environment. d-IP<sub>3</sub> derivatives were designed as IP<sub>3</sub> ligands that bind to IP<sub>3</sub>R, and l-IP<sub>3</sub> derivatives (optical isomers of d-IP<sub>3</sub>), which are expected to have the same photochemical properties as those of d-IP<sub>3</sub> derivatives, although with much weaker agonistic effects on IP<sub>3</sub>R (8), were also synthesized (Fig. 2c). These photosensitizer-conjugated d-IP<sub>3</sub> derivatives induced Ca<sup>2+</sup> release via IP<sub>3</sub>R in a dose–response manner with an EC<sub>50</sub> of 3  $\mu$ M for both ESPer d-IP<sub>3</sub> and Con d-IP<sub>3</sub>, whereas the photosensitizer-conjugated l-IP<sub>3</sub> derivatives had almost no Ca<sup>2+</sup> release activity (SI Fig. 7).

**Imaging of Photosensitizer-Induced Nonspecific Damage Caused by ESPer d-IP<sub>3</sub> and Con d-IP<sub>3</sub> in the Cytosolic Polar Environment.** We examined whether ESPer d-IP<sub>3</sub> outside the hydrophobic IP<sub>3</sub>R environment (e.g., in the cytosolic polar environment) indeed lacked <sup>1</sup>O<sub>2</sub>-generating ability (off state of photosensitization) despite light illumination, whether <sup>1</sup>O<sub>2</sub>-generating ability was recovered in the vicinity of IP<sub>3</sub>R (on state of photosensitization), and whether IP<sub>3</sub>R could be inactivated by the <sup>1</sup>O<sub>2</sub> generated (as illustrated in Fig. 1a). We first studied whether ESPer d-IP<sub>3</sub> generated <sup>1</sup>O<sub>2</sub> in the cytosol in the cells by fluorescence imaging of <sup>1</sup>O<sub>2</sub> generation with a fluorescent probe for <sup>1</sup>O<sub>2</sub> (DMAX-2) (16) in permeabilized DT40 cells (wild type). Permeabilized DT40 cells were loaded with 3  $\mu$ M ESPer d-IP<sub>3</sub> (or Con d-IP<sub>3</sub>) and 10  $\mu$ M DMAX-2, followed by illumination with green light (BP530–550 nm, 1.5 mW/cm<sup>2</sup>, 20 sec) to excite the photosensitizer. ESPer d-IP<sub>3</sub>-loaded cells showed little, if any, fluorescence increase of DMAX-2 upon light illumination (Fig. 3a–c). On the other hand, Con d-IP<sub>3</sub>-loaded cells showed a marked fluorescence increase of DMAX-2 in cytosol (Fig. 3d–e), owing to nonspecific <sup>1</sup>O<sub>2</sub> generation in the polar cytosolic environment, as shown in Fig. 2b. The fluorescence increase induced by DMAX-2 in Con d-IP<sub>3</sub>-loaded cells was inhibited by adding a <sup>1</sup>O<sub>2</sub> quencher





**Fluorescence Imaging.** Fluorescence images were acquired with an inverted microscope (IX71; Olympus), equipped with a cooled CCD camera (Cool Snap HQ; Roper Scientific) and a xenon lamp (AH2-RX; Olympus). The whole system was controlled with MetaFluor 6.1 software (Universal Imaging). For imaging  $^1\text{O}_2$  generation by DMAX-2 (for details of the characteristics of DMAX-2, see also SI Fig. 10), DT40 cells (wild type) attached to the coverslips were permeabilized with 60  $\mu\text{M}$   $\beta$ -escin for 1–2 min. Cells were loaded with 3  $\mu\text{M}$  Con d-IP<sub>3</sub> or ESPer d-IP<sub>3</sub> and 10  $\mu\text{M}$  DMAX-2, and fluorescence images (excitation filter, BA470–490; dichroic mirror, DM505; emission filter, BA510–550; Olympus) were acquired immediately after illumination with green light (BP530–550 nm, 1.5 mW/cm<sup>2</sup>, 20 sec).

**Luminal  $\text{Ca}^{2+}$  Imaging of DT40 Cells.**  $\text{Ca}^{2+}$  imaging of DT40 cells was performed as described previously (17) (see also SI Materials and Methods). Briefly, DT40 cells (wild type) were loaded with Fura2/AM, a membrane-permeant, low-affinity  $\text{Ca}^{2+}$  indicator, which enters both the cytosol and organelles. Fura2-loaded cells were then permeabilized with  $\beta$ -escin so that Fura2 was retained only in the ER, enabling us to continuously monitor luminal  $\text{Ca}^{2+}$  concentration ( $[\text{Ca}^{2+}]_l$ ) within the store. An increase in  $[\text{Ca}^{2+}]_l$  was observed

upon activation of sarco/endoplasmic reticulum  $\text{Ca}^{2+}$ -ATPase (SERCA) with application of both  $\text{Ca}^{2+}$  and MgATP, followed by a decrease upon addition of IP<sub>3</sub>. This  $\text{Ca}^{2+}$  loading and release procedure can be repeated reproducibly in the same cells, and the rate constant of IP<sub>3</sub>-induced  $\text{Ca}^{2+}$  release (IICR) was used as an index of IP<sub>3</sub>R activity.

**Light Illumination for Photoinactivation of IP<sub>3</sub>R.** First, by monitoring  $[\text{Ca}^{2+}]_l$  within permeabilized DT40 cells, we measured the IICR rate at 10  $\mu\text{M}$  IP<sub>3</sub>. The cells were then illuminated for 5 sec in the presence of 2  $\mu\text{M}$  test photosensitizer with a xenon lamp (AH2-RX), which was filtered to  $\approx 535 \pm 25$  nm by an excitation filter (HQ535/50; Chroma Technology) through the objective lens under the fluorescence microscope. The photosensitizer was washed out, and the IICR rate was measured again and compared with the pretreatment value.

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