Ptoblem 18-1.-1. $\frac{\alpha}{\alpha}$ $\frac{\alpha}{\alpha} = \frac{\alpha}{\alpha} - \frac{\alpha}{\alpha} - \frac{\alpha}{\alpha}$ Disopropyl ethor. 0 - 01/2-01/3. Cyclopenty - propy other. cyclopenty propy | p - Bromomethoxy benzone p-Bromo anisole. 1 - methoxy cyclo hexene.

Ptoblem 18-1-2 isobuty ethyl Ethy sobold ether. 1/2 C= cH-c/2-0-(cH=cH2) Ally Viny Ally I viny other.

Problem 18-3-1 as Methyl propy ether. methyl propyl. \rightarrow R-o and R-X 10 SN2 10 10 CM3 - Br and 0-CM= CM= CM3 Bromo methane propomolate Methanolate SHZ Bromopropone. Anisol (methy phony ethor) cbs phenyl methyl

SN2

Op o and olls-Br

Bromo methone

() Br and Ub-0 (X)

Problem 18-3-2

(6) Benzyl Isoptopyl ether. (0) c/2-0- c/1-d/3 Bon 2yl Isophopyl. Syz and 20-cH-cHs

Bromo metheyl benzene.

Syz and 20-cH-cHs

Ull3

Isoptopomolate

O-cH2-Q and 13+-cH-cH3 (X)

Phony methanolate

Bromo soptopane. (d) Ethyl 2.2-dhmethyl propyl ether. $CH_3 - CH_2 - O - CH_2 - C - CH_3$ ethyl 2.2 - dimethyl propyl.

SN2

CH3-CH2-B+ and $-\text{CH}_2$ $-\text{CH}_3$.

CH3-CH2-B+ and $-\text{CH}_2$ $-\text{CH}_3$.

CH3-CH2-B+ and $-\text{CH}_3$. Blomoethane 2,2-dimethy/proponolate.

CH3-CH2-Q and Br-CH2-c-CH3
Ethanolate

SH2 1-Bromo-2.2-dimethyl propane.

2- methoxy-2proparyl-bonzone or (2- methoxy propan -2-yl) bonzone

$$\rightarrow 0 + 4 - \ddot{0} - 4 + B + C$$

(ptop-1-en-2-yl)benzene.

$$\rightarrow CH_3 - CH_2 - CH_2 - CH_3 - CH_2 - CH_3 - CH_3$$

18-8.

2-methylpropene. Cyclohexanol.

problem 18-10

Problem 18-12.-1 > 2nd carbocation (a) HO. 3- golohexyl-2.2-dimethyloxirane SM 1. and Carbocation propan-2-01. major 04 01 2 -chloro-1-cyclo hexy! - 2-methy proponol.

Problem 18-12-2 1º carbocation. 3° carbocation (unstable) (1-chlorogodohexyl) methanol.

Problem 18-14-1 2-eth/-2-meth/oxtrane H2C -C - CH2 CH3 + H+ + (8H) under the basic condition. thing - opening Is SHZ txn. $H0^{18} - CH_2 - \frac{1}{5} - \frac{1}{5} - \frac{1}{5} = \frac{1}{5} - \frac{1}{5} - \frac{1}{5} - \frac{1}{5} - \frac{1}{5} = \frac{1}{5} - \frac{1}{5} - \frac{1}{5} = \frac{1}{5} - \frac{1}{5} = \frac{1}{5} - \frac{1}{5} = \frac{1}{5}$ 2-methyl butane-1,2-diol. H2 C- C- aladls

als

H3 0^{† 18}. (b) under the add and then. ting -opening is SN2 run or SNI run. Check the carbocation. 3º carbocation H2 C - c - cl/2 cl/3 + H+ + H2018 $+ \frac{64}{12} - \frac{18}{12} - \frac{18}{12} - \frac{1}{12} - \frac{1$

12 toblem 18-14-2.

$$||3| = \frac{6}{4} + \frac{6}{4}$$