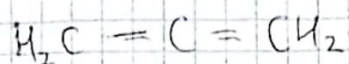
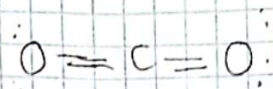


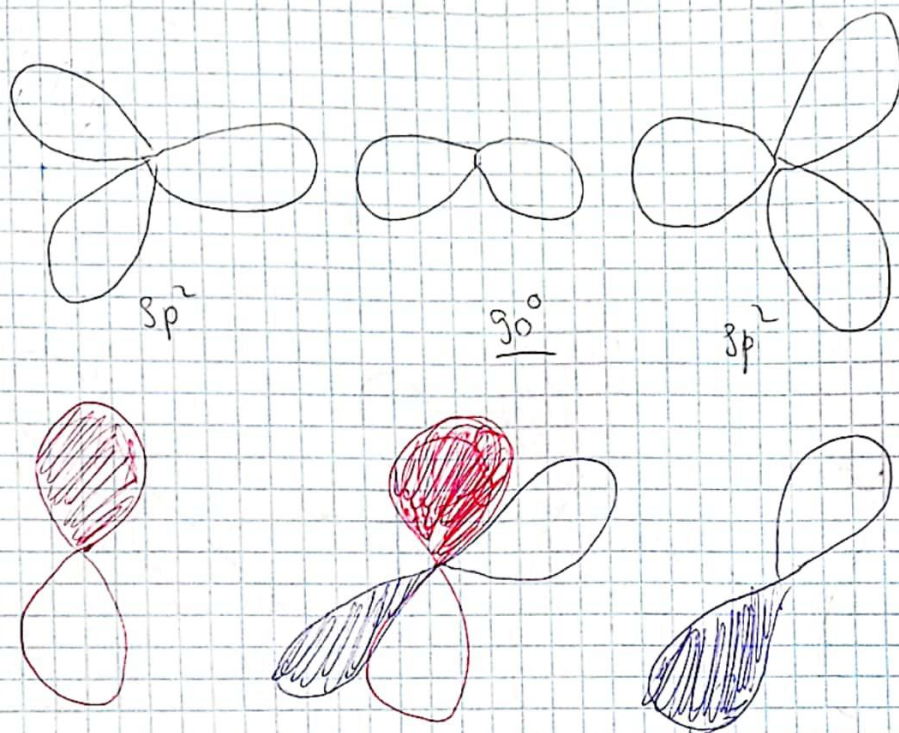
Problem 1-48



allene



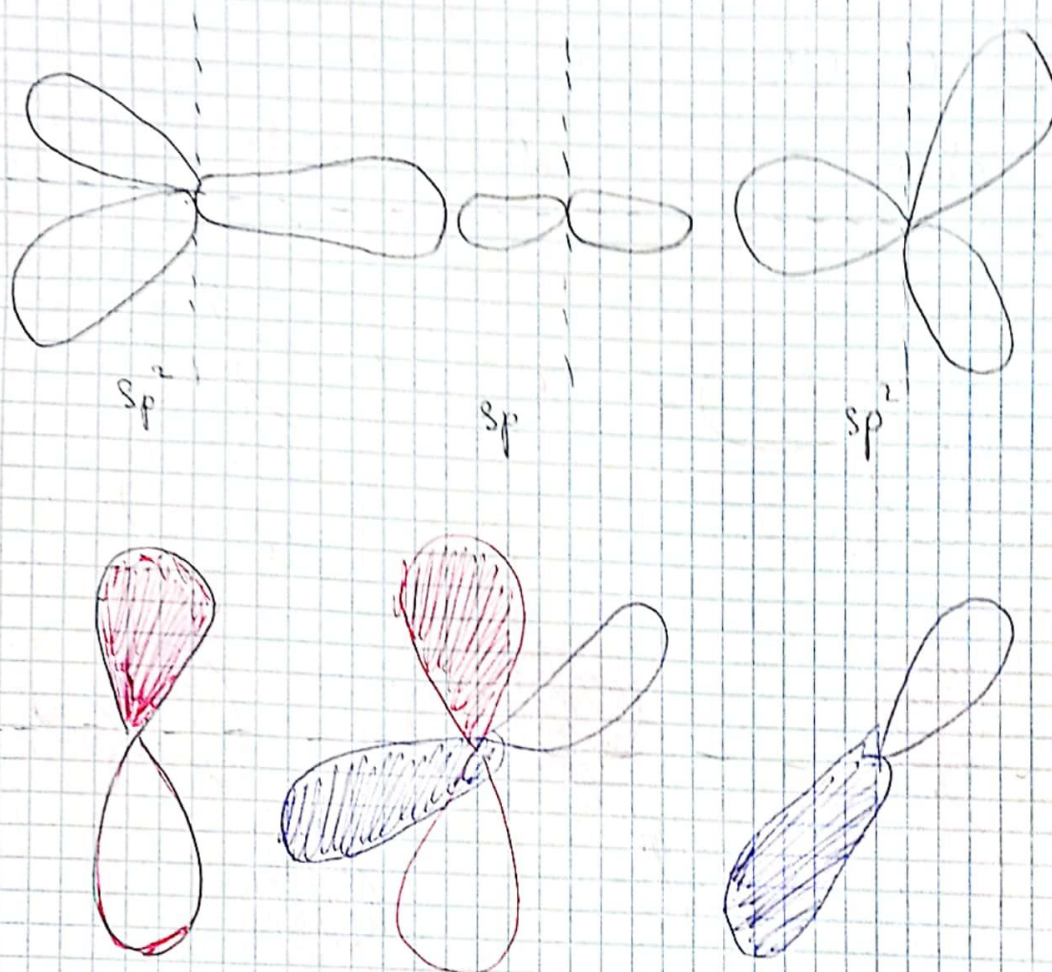
carbon dioxide



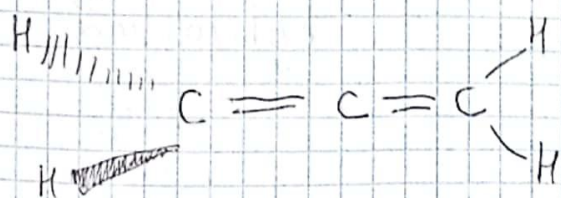
The central carbon atom makes a p bond to the oxygen atom on the left using the p orbitals that are oriented on the plane of the page.

The carbon atom of carbon dioxide is sp hybridized.

Problem 1-47



The hydrogen atoms bonded to the reaction on the left should be located on a plane perpendicular to the hydrogen atoms bonded to the carbon on the right.



Thank you

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