



Sri Chaitanya IIT Academy, India

A.P, TELANGANA, KARNATAKA, TAMILNADU, MAHARASHTRA, DELHI, RANCHI

A right Choice for the Real Aspirant

ICON CENTRAL OFFICE, MADHAPUR-HYD

Sec: Sr.IPLCO
Time: 3 Hours

JEE-ADVANCE
2012-P2-Model

Date: 16-08-15
Max Marks: 198

KEY & SOLUTIONS

PHYSICS

1	A	2	D	3	D	4	C	5	A	6	D
7	D	8	C	9	A	10	C	11	A	12	A
13	B	14	C	15	AC	16	ACD	17	BC	18	AB
19	CD	20	CD								

CHEMISTRY

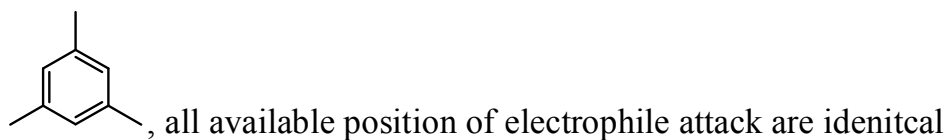
21	B	22	A	23	C	24	A	25	A	26	B
27	D	28	B	29	B	30	C	31	A	32	C
33	B	34	C	35	CD	36	ACD	37	ABCD	38	ABC
39	BC	40	AC								

MATHS

41	D	42	D	43	C	44	B	45	A	46	A
47	C	48	D	49	D	50	C	51	A	52	D
53	A	54	C	55	BCD	56	AC	57	ABC	58	BD
59	BC	60	ABD								

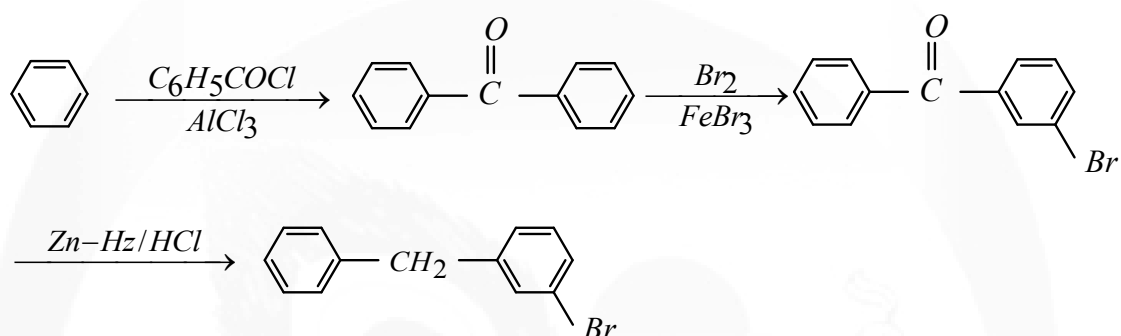
CHEMISTRY

21.



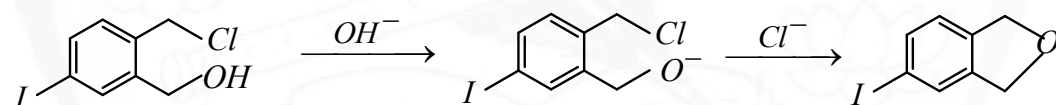
22. Electron withdrawing groups increases reactivity towards Nucleophile

23.

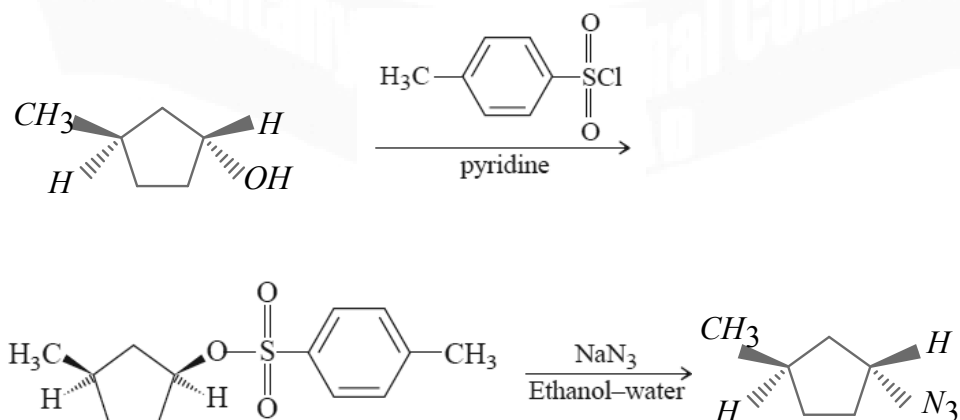


24. Intermolecular nucleophilic substitution

25.

26. S_N2 : Inverted product27. S_N2 : reactivity is more with good leaving group

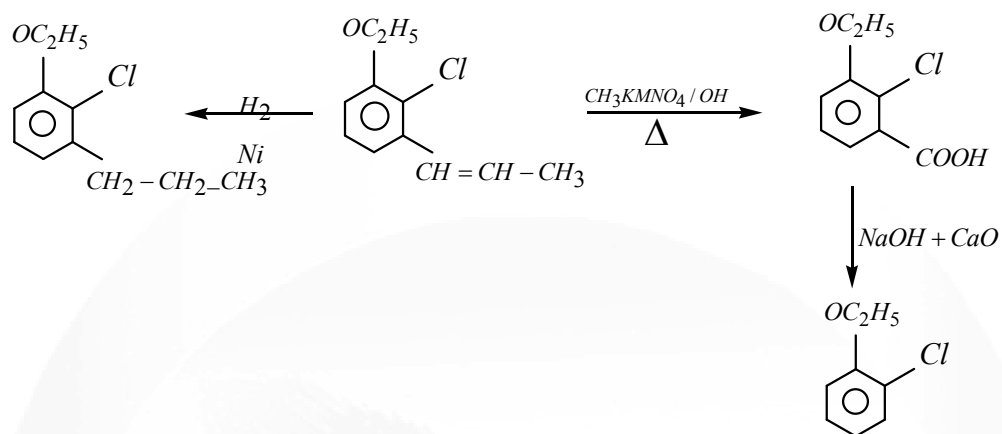
28.



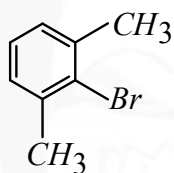
29, 30

Orientation effects

31, 32.

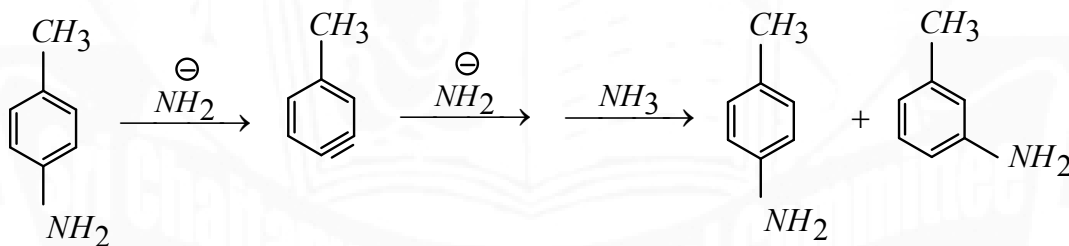


33.

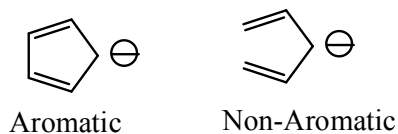


No hydrogens on adjacent carbons to one which is bonded to halogen. So elimination-addition mechanism is not possible.

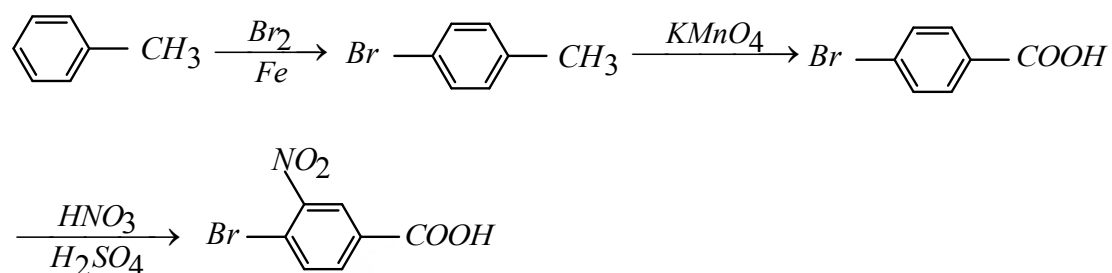
34.

35. $-N=O \rightarrow$ Deactivating, O/P-directing.

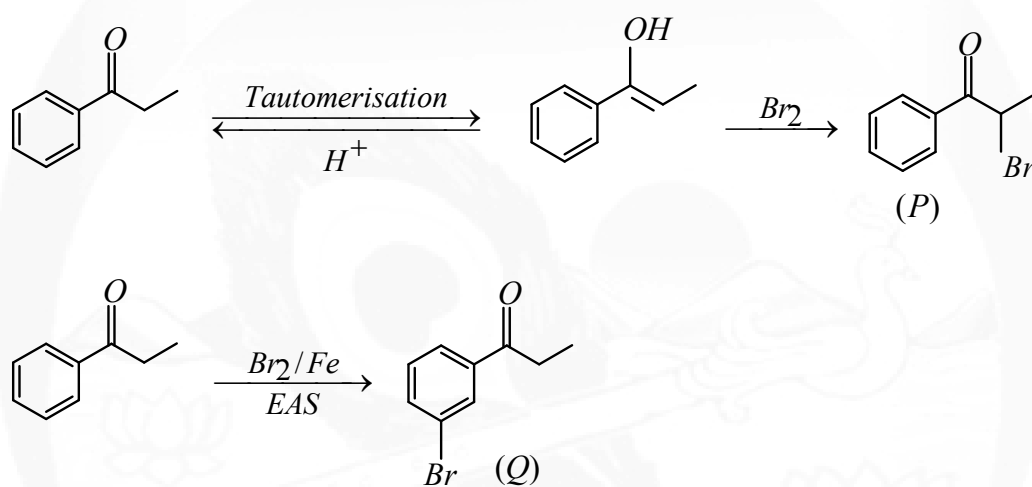
36.



37.

38. $-\text{OH}$ is converted into good leaving group by converting into phosphate, esters.

39.



40.

