# CHEMISTRY: Max. Marks: 60

# SECTION – I (MULTIPLE CORRECT CHOICE TYPE)

This section contains **8 multiple choice questions.** Each question has 4 choices (A), (B), (C) and (D) for its answer, out of which **ONE OR MORE** is/ are correct

- 21. Aldehydes and ketones give crystalline product with NaHSO<sub>3</sub>. The correct statements about it are
  - A) all carbohydrates also give the addition product with NaHSO<sub>3</sub>
  - B) Carbonyl compounds can be regenerated from the addition product by hydrolysis
  - C) Oxygen in SO<sub>3</sub> connects to carbonyl carbon
  - D) Sulphur in the -SO<sub>3</sub>H connects to the carbonyl compound
- 22. The product of hydrolysis of A and B can be distinguished with

(A) 
$$CH_2 = C O - COCH_3$$
 $CH_3$ 

CH<sub>3</sub>
|
(B) 
$$C = C - O - COCH_3$$
|
H H

- A) I<sub>2</sub>/NaOH
- C) NaHSO<sub>3</sub>

- B) Fehlings solution
- D) 2,4-DNPH

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- 23. Which of the following react with ethanolic KCN?
  - A) ethyl chloride

B) chloro benzene

C) benzaldehyde

- D) 2,4-dinitro chloro benzene
- 24. Among the following which will react with acetone and form a product with

- A)  $C_6H_5NH_2$
- B)  $(CH_3)_2 N$
- C)  $NH_2 NH_2$
- $\begin{array}{c|c}
  CH_3 N CH_3 \\
  \downarrow \\
  D) & H
  \end{array}$
- 25. The chemicals used in preparing Fehlings solution are.
  - A) CuSO<sub>4</sub>

B) sodium potassium tartrate

C)NaOH

D) sodium citrate

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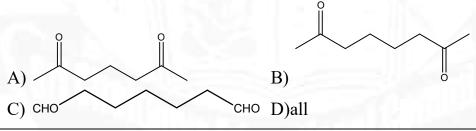
The reagents which bring about the following reaction are 26.

- A) CrO3, HCl, pyridine
- B)  $CrO_3$ ,  $H_2SO_4$ , acetone

- D) 1% KMnO<sub>4</sub>
- In which of the following a pair of diastereomers is formed? 27.

$$+NH_2OH \xrightarrow{H^+}_{\Delta}$$

- A)  $CH_3CHO + NH_2OH \xrightarrow{H^+}_{\Delta}$
- C)  $CH_3COCH_2CH_3 \xrightarrow{NH_2OH} \rightarrow$
- D)  $HCHO + NH_2OH \xrightarrow{H^+}$
- Which of the following when treated with aq.NaOH give a product with six 28. membered ring?



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# SECTION - II (COMPREHENSION TYPE)

This section contains **4 groups of questions**. Each group has 2 multiple choice questions based on a paragraph. Each question has 4 choices A), B), C) and D) for its answer, out of which **ONLY ONE** is **correct**.

## Paragraph for Questions 29 and 30

When an aldehyde with no  $\alpha$ -hydrogen is treated with conc alkali it undergoes disproportionation. The products are sodium salt of an acid and alcohol. It can be intramolecular also.

29. The product formed in the following reaction in

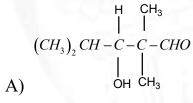
 $C_6H_5COCHO \xrightarrow{OH^-}$ 

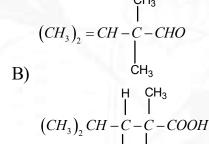
A)  $C_6H_5COCOO^-$ 

B)  $C_6H_5CHOHCHO$ 

$$C_6H_5CH-COO^-$$

- C)  $C_6H_5COOH + C_6H_5CH_2OH$
- D) OH
- 30.  $(CH_3)_2 CHCHO \xrightarrow{NaOH}$  the products are





OH CH<sub>3</sub>

C) 
$$(CH_3)_2 CHCH_2OH + (CH_3)_2 CHCOO^-Na^+ D$$

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31.

## Paragraph for Questions 31 and 32

Aldol condensation takes place in aldehydes or ketones having at least one  $\alpha$  H-atom when treated with NaOH. The product is  $\beta$  hydroxy aldehyde or ketone respectively. On heating, dehydration takes place and  $\alpha, \beta$ -unsaturated aldehydes or ketones are formed.

32. Which of the following pair gives a single product in crossed aldol condensation

A) 
$$C_6H_5 - CH_2CHO, +C_6H_5CHO$$

B) 
$$C_6H_5CHO + HCHO$$

$$C_6H_5CHO + (CH_3)_2 - CCHO$$
  
C)

D) 
$$CH_3CHO + (CH_3), CH.CHO$$

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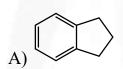
# Paragraph for Questions 33 and 34

In the following reactions sequences, B is an intermediate which releases CO<sub>2</sub> with NaHCO<sub>3</sub> and decolorises Baeyer's reagent

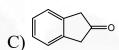
$$A \xrightarrow{(CH_3,CO)_2O} B \xrightarrow{1.H_2,Pd-C} CH_3COONa \rightarrow B \xrightarrow{2.SOCl_2} 3.AlCl_3$$

- 33. A is
  - A)  $C_6H_5CHO$
- B)  $C_6H_5CH_2OH$
- C)  $C_6H_5COCH_3$  D)  $C_6H_5CH = CH_3$

34. C is



B)



# Paragraph for Questions 35 and 36

5 isomeric p-disubstituted aromatic compounds have the formula  $C_8H_8O_2$ 

One isomer gives positive iodoform test it is 35.

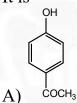


COCH<sub>3</sub> B)

OCH = CH<sub>2</sub> C)

D)

One compound reduces ammonical AgNO<sub>3</sub> and gives Color with neutral FeCl<sub>3</sub>. 36. It is



B)

OCH = CH<sub>2</sub>

C)

CH<sub>2</sub>CHO

CH2CH2OH D)

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#### **SECTION - III**

#### (MATRIX MATCH TYPE)

This section contains 4 multiple choice questions. Each question has matching lists. The codes for the lists have choices (A), (B), (C), and (D) out of which ONLY ONE is correct.

# 37. Match the following

## Column-I

#### Column-II

(Reaction)

(Name of the reaction in which it is involved)

A) 
$$RCOOOH$$
 $CH_2OH$ 
 $Quinone$ 
 $Al \ tert \ but oxide$ 
 $CHO$ 

P) pinacol-pinacolone rearrangement

RCOOR 
$$\xrightarrow{CH_2\text{-COOEt}}$$
 OH  $\xrightarrow{Br}$  R - C - CH<sub>2</sub>COOEt  $\xrightarrow{Zn, \text{ ether}}$  R

Q)Reformatsky reaction

C) D) R) Oppenauer oxidation

- A)  $A \rightarrow S$ ;  $B \rightarrow R$ ;  $C \rightarrow Q$ ;  $D \rightarrow P$
- C)  $A \rightarrow Q$ ;  $B \rightarrow S$ ;  $C \rightarrow Q$ ;  $D \rightarrow P$
- S) Baeyer-villiger oxidation
- B)  $A \rightarrow S$ ;  $B \rightarrow Q$ ;  $C \rightarrow P$ ;  $D \rightarrow S$
- D)  $A \rightarrow R$ ;  $B \rightarrow PR$ ;  $C \rightarrow Q$ ;  $D \rightarrow S$

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## 38. Column-I

Column-II

(Reaction)

(Product)

$$A) \stackrel{\mathsf{OH}}{\longrightarrow} \xrightarrow{H_2CrO_4}$$

P)  $C_6H_5COCH_3$ 

$$B) \stackrel{\mathsf{CH}_3}{\longleftrightarrow} \xrightarrow{CrO_2Cl_2}$$

Q)  $C_6H_5CH_2CHO$ 

C) 
$$C_6H_5 - C \equiv CH \xrightarrow{H_2SO_4} HgSO_4 \rightarrow$$

R)  $C_6H_5CHO$ 

D) 
$$C_6H_5CH_2Cl \xrightarrow{1.KCN} {2.SnCl_2,HCl \atop 3.H_2O} \rightarrow$$

S)

A) A 
$$\rightarrow$$
P;B  $\rightarrow$ R;C  $\rightarrow$ Q;D  $\rightarrow$ P

B)A  $\rightarrow$ P;B  $\rightarrow$ Q;C  $\rightarrow$ Q;D  $\rightarrow$ S

C) A 
$$\rightarrow$$
 S;B  $\rightarrow$  P;C  $\rightarrow$  Q;D  $\rightarrow$  Q

 $D)A \rightarrow S;B \rightarrow R;C \rightarrow P;D \rightarrow Q$ 

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39.	Column-I	Column-II
	(Compound) (N	Name of the reaction in which it is involved)
	A) $C_6H_5CHO$	P) Aldol condensation
	В) НСНО	Q) Cannizzaro reaction
	C) $CH_3COCH_3$	R) Benzoin condensation
	D) CH <sub>3</sub> CHO	S) Iodoform
	A) A $\rightarrow$ QR;B $\rightarrow$ Q;C $\rightarrow$ PS;D $\rightarrow$ PS B) A $\rightarrow$ PS;B $\rightarrow$ Q;C $\rightarrow$ PS;D $\rightarrow$ PQ	
	C) A $\rightarrow$ PR;B $\rightarrow$ Q;C $\rightarrow$ S;D $\rightarrow$ P	D) A $\rightarrow$ PS;B $\rightarrow$ P;C $\rightarrow$ S;D $\rightarrow$ S
40.	Column-I	Column-II
	(Compound)	(Reaction)
	A) $C_6H_5CHO$	P) positive iodoform test
	B) CH <sub>3</sub> CHO	Q) Reduces Fehlings solution
	C) HCHO	R) Reduces Tollens reagent
	D) CH <sub>3</sub> COCH <sub>3</sub>	S) 2,4-DNPH
	A) A $\rightarrow$ RS;B $\rightarrow$ PQRS;C $\rightarrow$ QRS;D $\rightarrow$ S	
	B) A $\rightarrow$ S;B $\rightarrow$ P;C $\rightarrow$ QRS;D $\rightarrow$ PQ	
	C) A $\rightarrow$ R;B $\rightarrow$ P;C $\rightarrow$ RS;D $\rightarrow$ P	
	D) A $\rightarrow$ PR;B $\rightarrow$ PRS;C $\rightarrow$ QS;D $\rightarrow$ S	
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