



# Sri Chaitanya IIT Academy, India

A.P, TELANGANA, KARNATAKA, TAMILNADU, MAHARASHTRA, DELHI, RANCHI

A right Choice for the Real Aspirant  
ICON CENTRAL OFFICE, MADHAPUR-HYD

Sec: Sr.IPLCO

Time: 09:00 AM to 12:00 Noon

RPTA-10

Dt: 01-11-15

Max.Marks: 180

## PAPER-1

### KEY & SOLUTIONS

#### PHYSICS

1	ABD	2	BC	3	A	4	BC	5	D	6	ABC
7	AC	8	C	9	AD	10	AC	11	5	12	3
13	8	14	6	15	3	16	0	17	6	18	6
19	3	20	0								

#### CHEMISTRY

21	ABCD	22	CD	23	ABC	24	ABCD	25	ACD	26	ABD
27	ABD	28	AB	29	ABC	30	ABD	31	4	32	3
33	4	34	5	35	7	36	2	37	8	38	3
39	9	40	3								

#### MATHS

41	BCD	42	AC	43	ABC	44	ABD	45	ABCD	46	ABD
47	BCD	48	AB	49	AD	50	ABCD	51	3	52	5
53	2	54	7	55	1	56	6	57	1	58	9
59	2	60	3								

**CHEMISTRY**

21. All can oxidize  $I^-$  to  $I_2$
22.  $AgF$  and  $AgNO_3$  are soluble in water
23. dil. NaOH gives  $OF_2$ , conc. NaOH gives  $O_2$
24. All can give  $O_2$
25.  $XeO_3 + 2XeF_6 \rightarrow 3XeOF_4(p)$ ,  $XeO_3 + XeOF_4 \rightarrow 2XeO_2F_2$
26. No reaction with dry  $SiO_2$ .
27. (c) is  $Na_3(AlF_6)$
28. (C) and (D) have high BP.
29. Silica will provide acidic lining
30. Al is extracted by electro reduction.
31. Euchlorine :  $Cl_2 + ClO_2, O + 4 = 4$
32.  $3NaClO \rightarrow 2NaCl + NaClO_3$
33.  $AgF, AgCl, PbCl_2$  are white
34.  $CaF_2, Hg_2Cl_2, AgCl, TlCl, HgI_2 \rightarrow$  water insoluble.
35.  $XeF_6 + H_2O \rightarrow XeOF_4 + 2HF$
36.  $XeF_6 + SiO_2 \rightarrow XeOF_4(sp^3d^2) + SiF_4$   $6 - 4 = 2$
37.  $3Mn_3O_4 + 8Al \rightarrow 4Al_2O_3 + 9Mn$
38.  $Fe_2O_3 + 3CO \rightarrow 2Fe + 3CO_2$
39.  $4FeS_2 + 11O_2 \rightarrow 2Fe_2O_3 + 8SO_2$   
 $2 + 3 + 4 = 9$
40.  $[Cu_2S + FeS]$  Matte