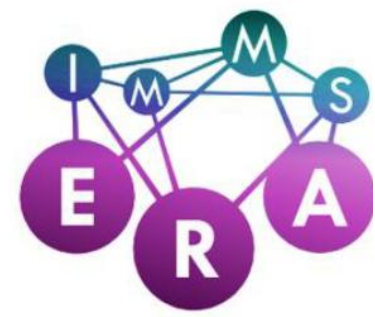


# 5<sup>th</sup> International Meeting on Materials Science for Energy Related Applications

September 25-26, 2025, Belgrade



## Automatizing data storage, analysis, and sharing for the electrochemical CO<sub>2</sub> reduction – the FAIR case of SuPERCO<sub>2</sub>

Dr. Federico Dattila

Department of Applied Science and Technology

Polytechnic of Turin



**Politecnico  
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SESSION 3 – (ELECTRO)CATALYSIS IN ENERGY  
CONVERSION & STORAGE

25/09/25, Belgrade, Serbia

# The Polytechnic of Turin

Located in Turin

Technical university since 1859

Both experimental and computational groups working on electrocatalysis

1.5h flight Milan-Belgrade, then 2h bus

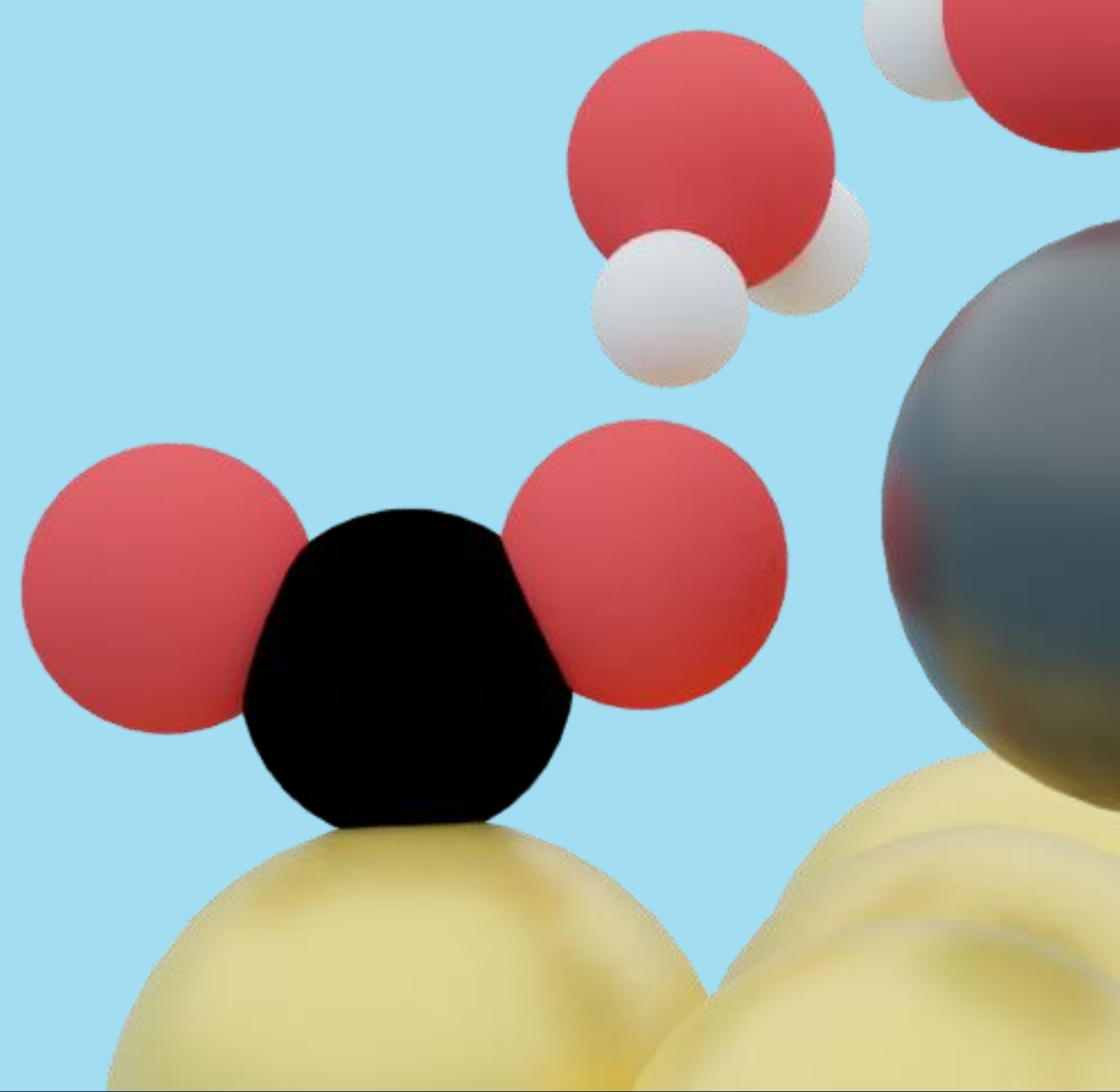


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di Scienza Applicata  
e Tecnologia

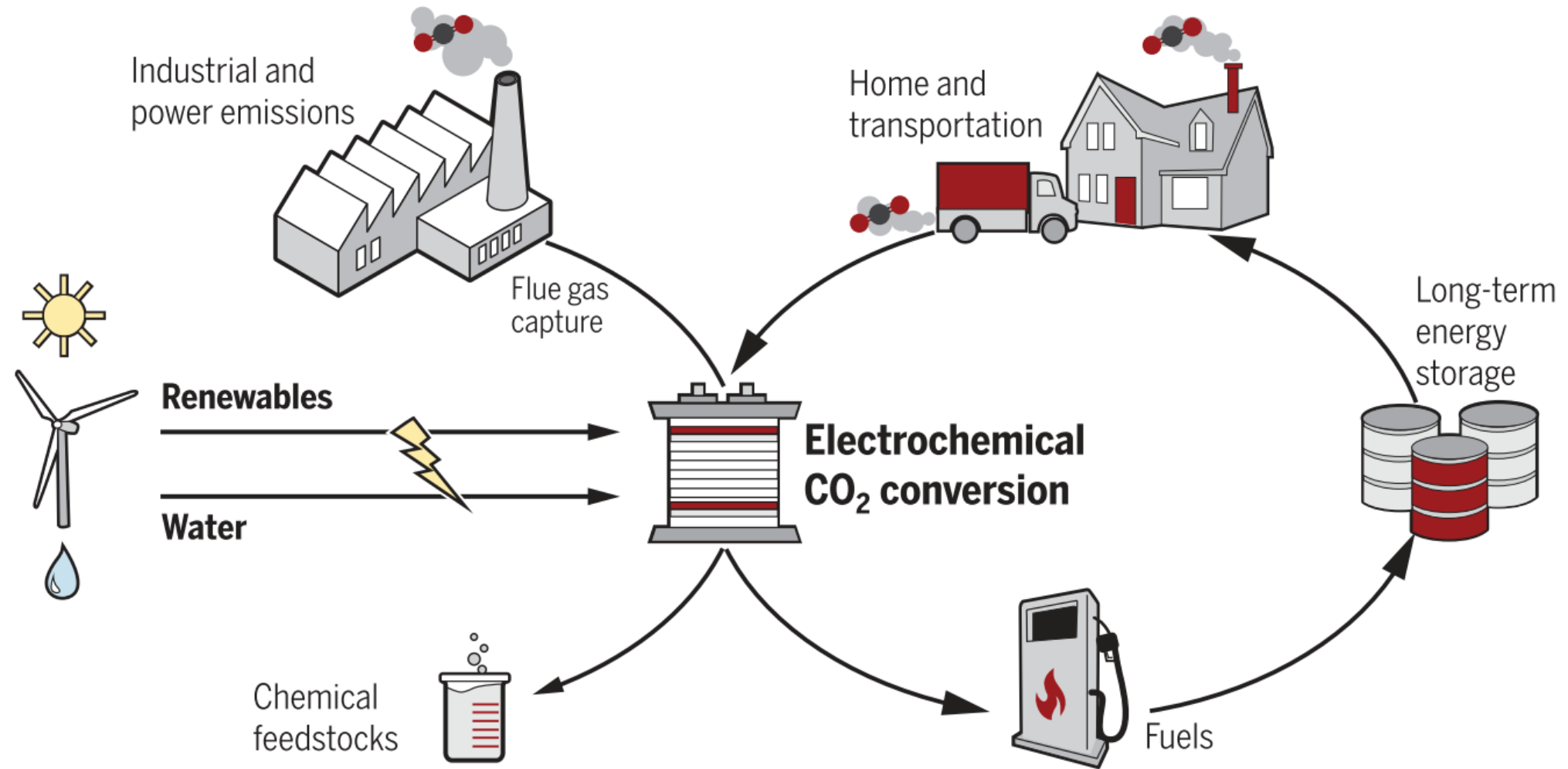


ECCELLENZA 2019 – 2027

An abstract graphic on the left side of the slide features several spheres of different colors (red, white, black, yellow, and grey) arranged in a way that suggests molecular structures or a network. The spheres are of varying sizes and are set against a light blue background. The overall composition is modern and scientific.

# The FAIR problem of CO<sub>2</sub>R

# Electrochemical CO<sub>2</sub> reduction



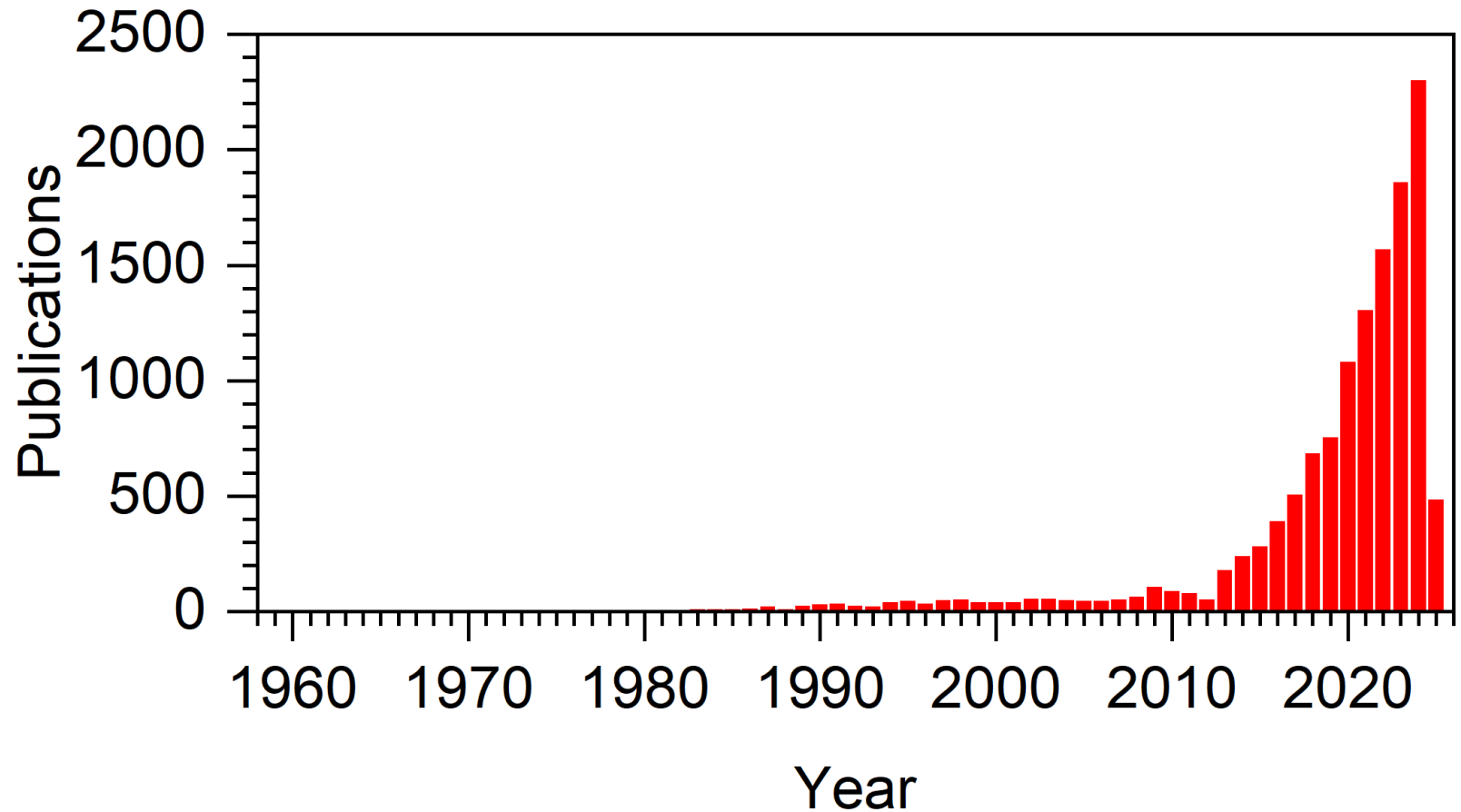
# Publications on CO<sub>2</sub>R

## POLAROGRAPHIC REDUCTION OF CARBON DIOXIDE

Sir:

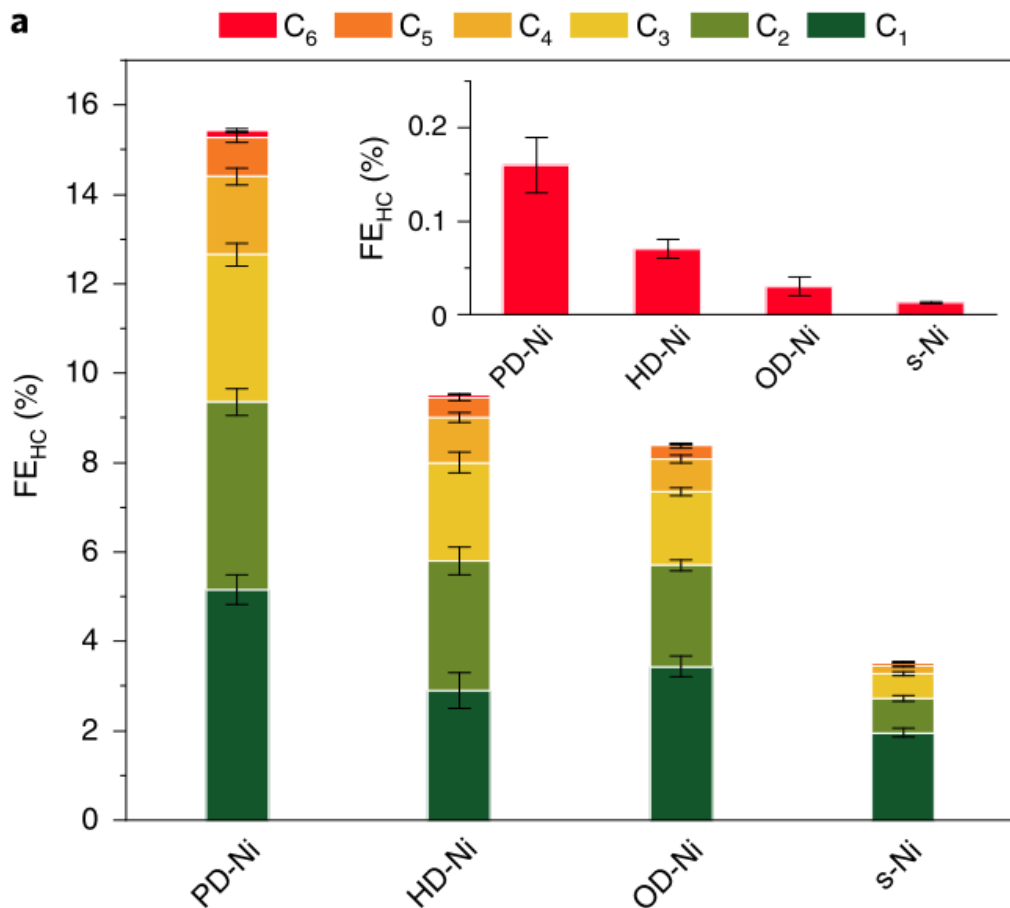
We have been engaged for some time in the study of the reduction of carbon dioxide at the dropping mercury cathode. On the basis of a large number of polarograms obtained by means of a Heyrovsky Polarograph of the Sargent Co. with solutions of carbon dioxide in 0.1 molar tetramethylammonium chloride, we have arrived at the following definite conclusions: carbon dioxide exhibits well-defined reduction waves with a half-wave potential of remarkable constancy:  $-2.24 \pm 0.01$  volt referred to the saturated calomel electrode. This is the average of ten independent determinations with varying amounts of carbon dioxide. When hydrogen is passed through the substituted ammonium salt solution before the dissolution of carbon dioxide, the waves are equally well defined, but the half-wave potential is somewhat less negative:  $-2.18 \pm 0.02$  volt referred to the saturated calomel electrode. This is the average of seven independent determinations, with varying amounts of hydrogen and carbon dioxide.

Van Rysselberghe, P.; Alkire, G. J.  
*J. Am. Chem. Soc.* **1944**, 66, 1801.



# The need for FAIR protocols

Findability / Accessibility / Interoperability / Reusability



Data to Zenodo

Figure 1 CO<sub>2</sub> electroreduction to hydrocarbons on Ni catalysts

Figure 1a FE of C1-C6 hydrocarbons formed on s-Ni, PD-Ni, HD-Ni and OD-Ni.xlsx

9.9 kB

Figure 1b Stability of the PD-Ni catalyst.xlsx

802.2 kB

Figure 1c Ratio of the FE of hydrocarbons formed on PD-Ni compared with that on s-Ni versus the carbon number.xlsx

8.6 kB

Figure 2 Hydrocarbon formation on Cu and PD-Ni

Figure 2 Hydrocarbon formation on Cu and PD-Ni.xlsx

8.8 kB

Figure 3 Redox response and operando X-ray absorption spectroscopy studies

Figure 3a Cyclic voltammograms of nickel phosphate under Ar and CO<sub>2</sub>.xlsx

142.5 kB

Files (1.2 MB)

Name

Size

Download all

Data to Zenodo.zip

md5:50fc947751f1d5ce4c5255f83b55745c

1.2 MB

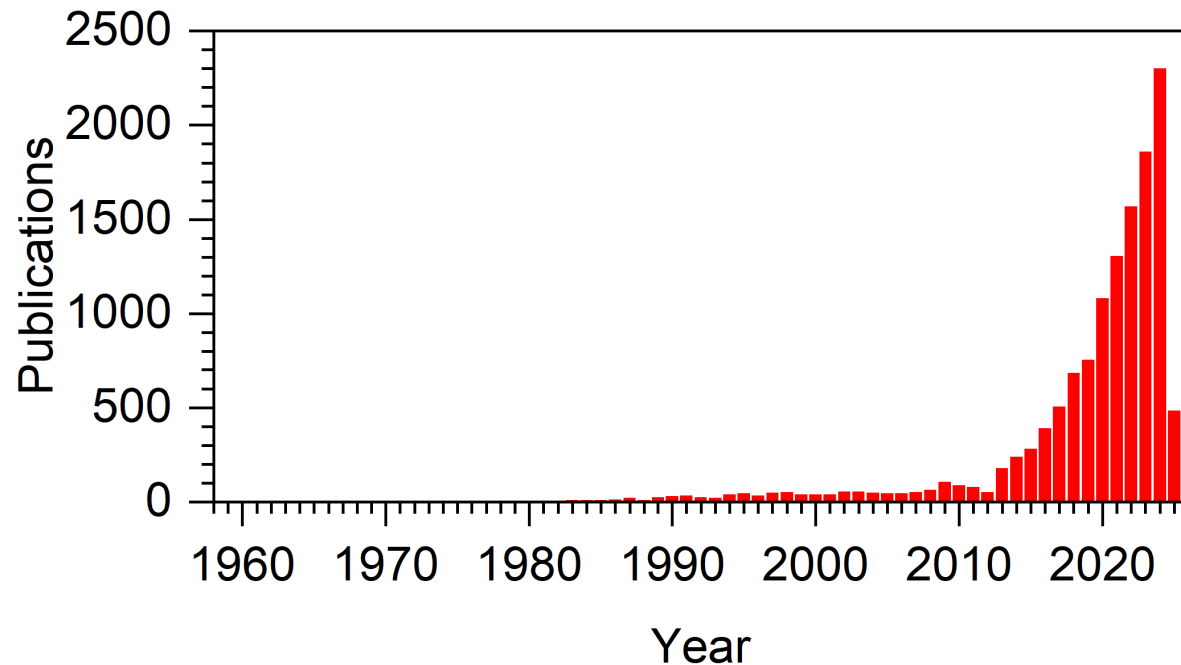
Preview

Download

# The limit of current repositories

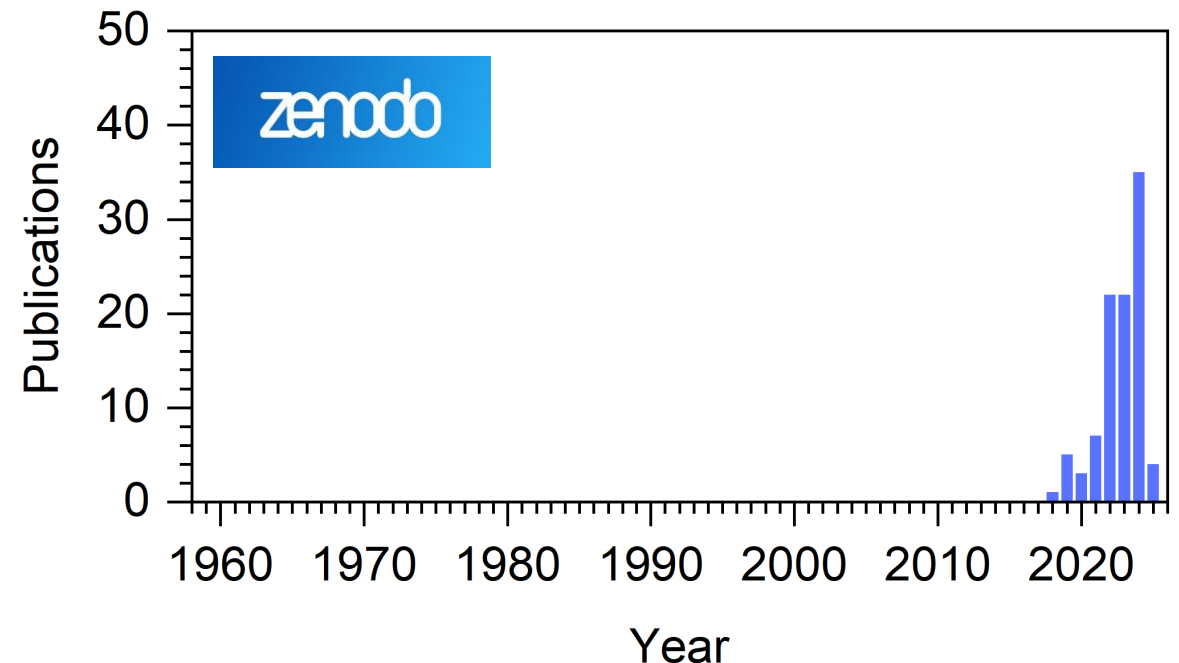
**Keywords:**

Electrochemical CO<sub>2</sub> reduction



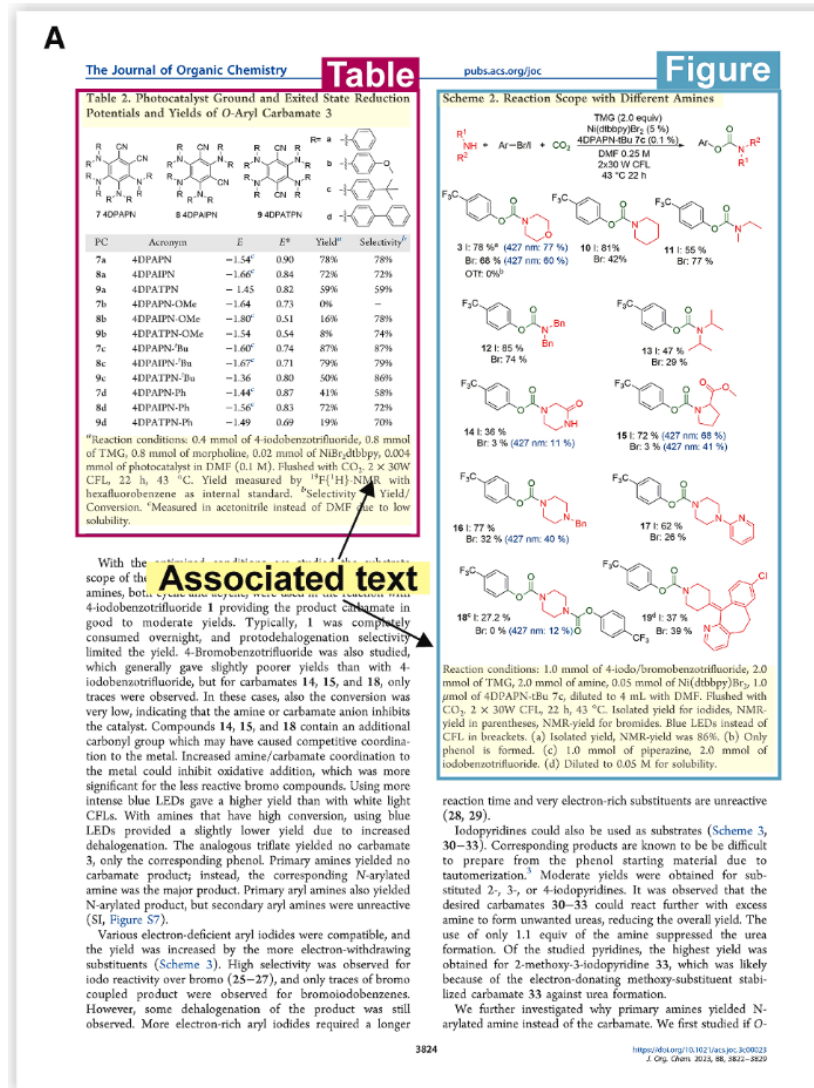
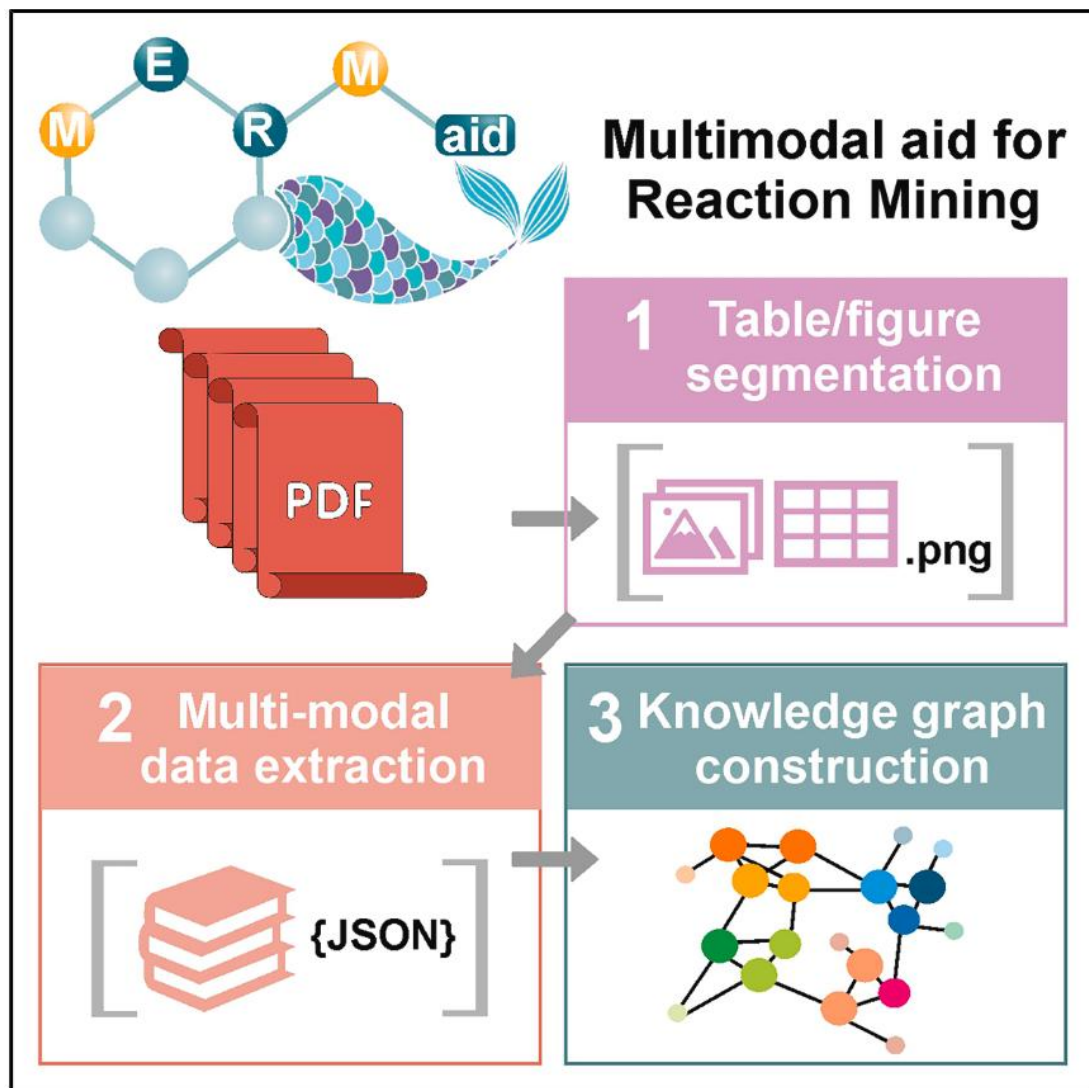
**Keywords:**

Electrochemical CO<sub>2</sub> reduction,  
Zenodo



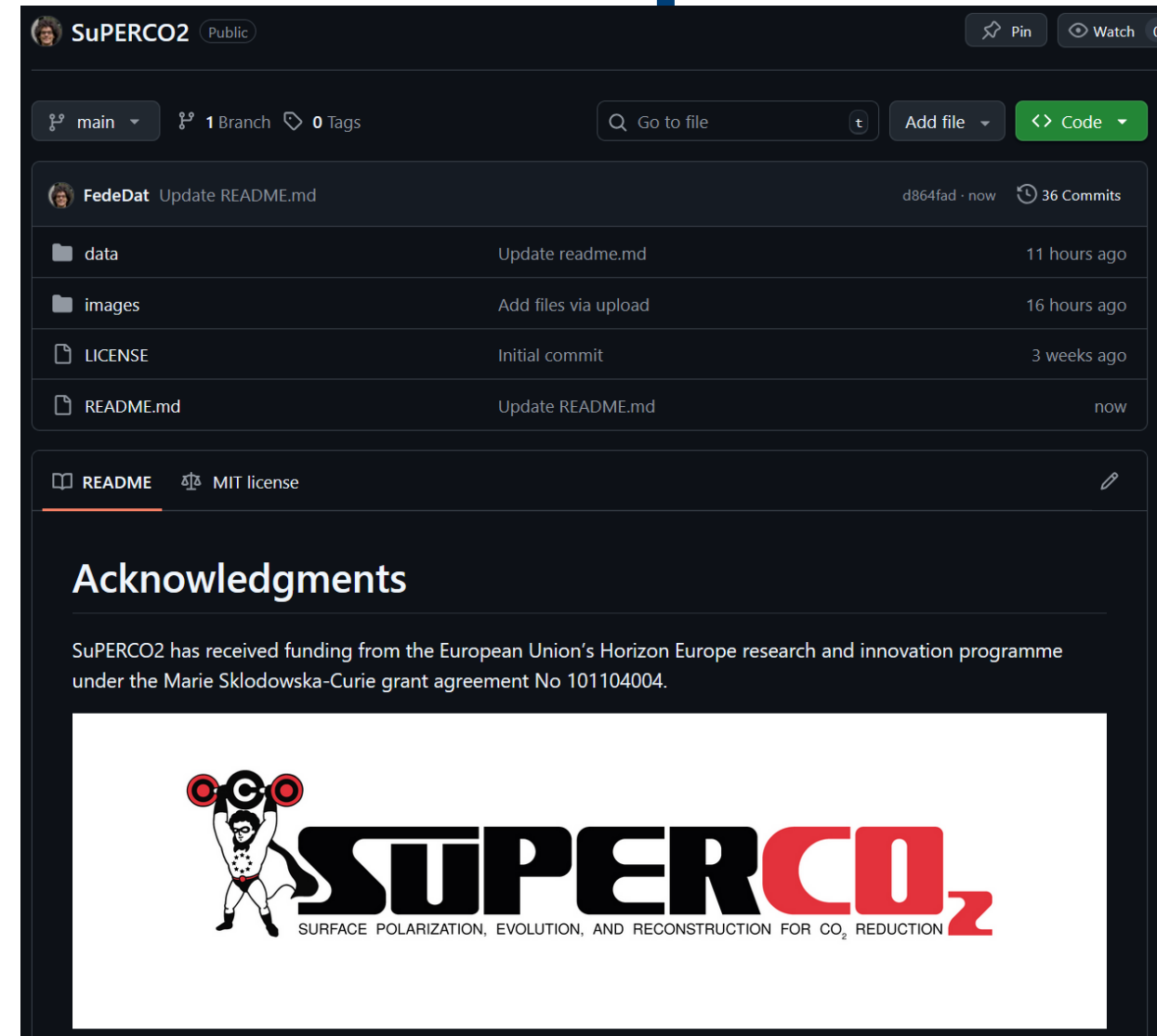
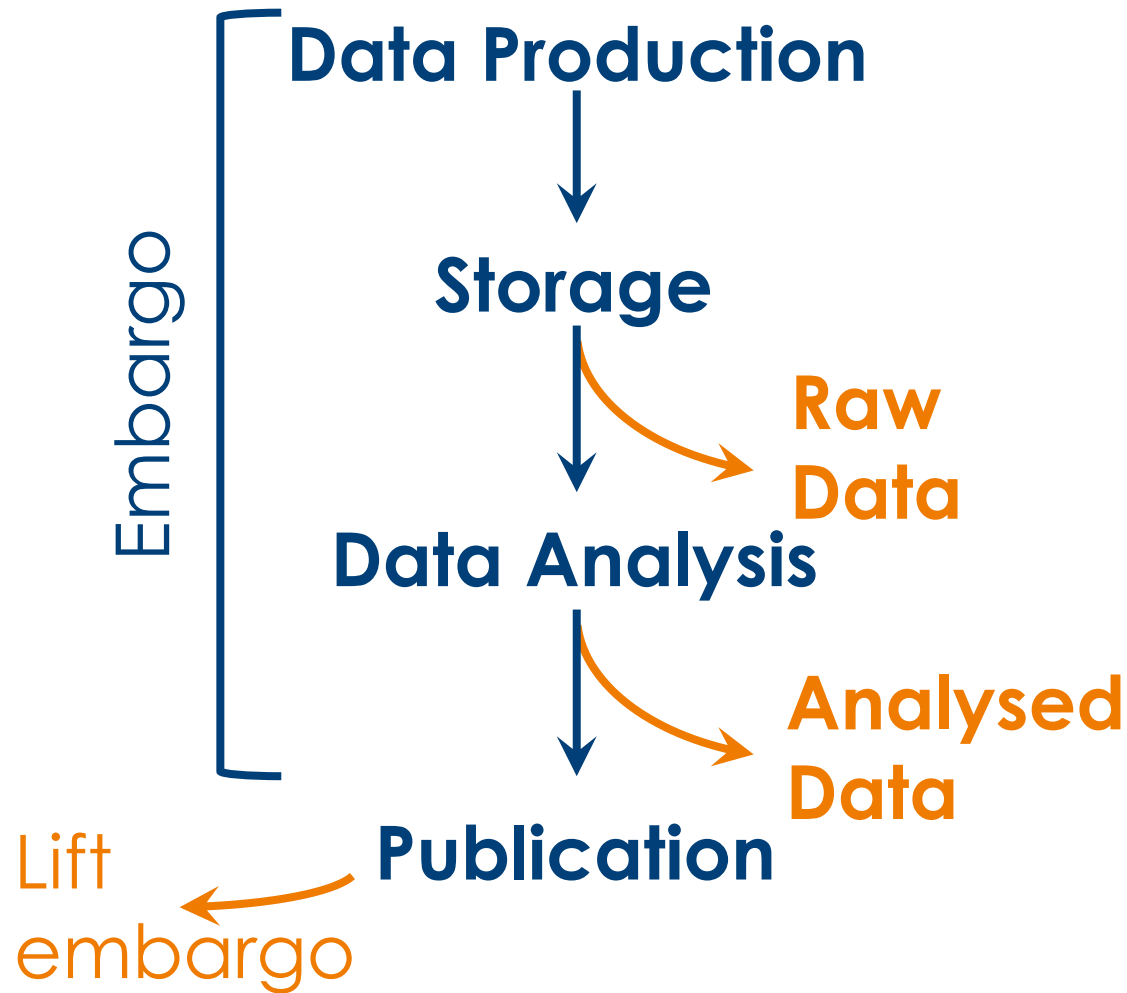


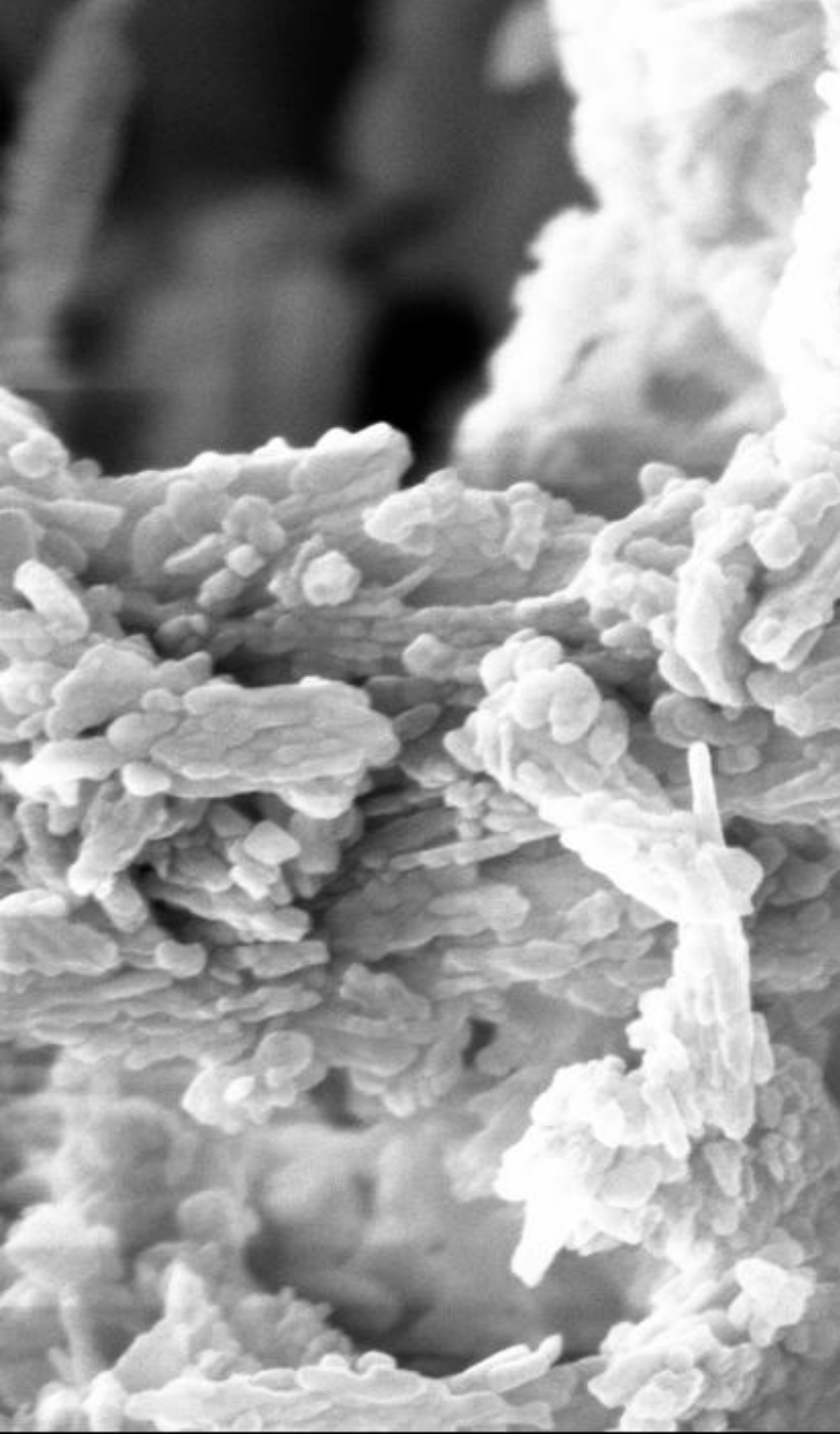
# AI-based solution: MERmaid





# The SuPERCO2 concept





# Characterization

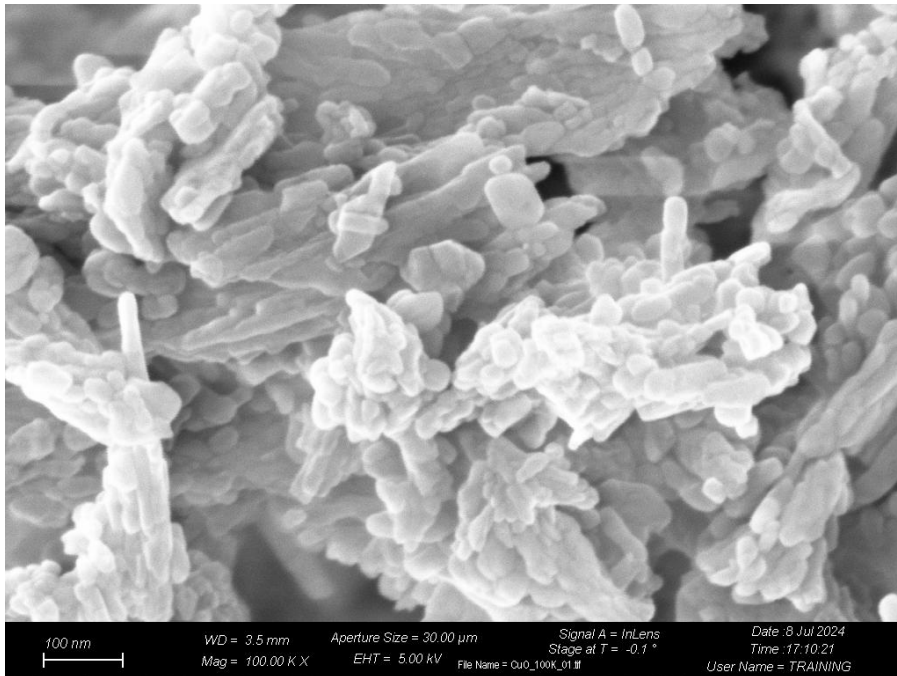
Repository: [github.com/FedeDat/SuPERCO2](https://github.com/FedeDat/SuPERCO2)

item: data/24-10-30\_CuO/characterization

# Scanning Electron Microscopy

## Post-processing

## Data analysis & Visualization



SuPERCO2 / data / 24-10-30\_CuO / characterization / fesem / [↑ Top](#)

|           |                      |                |
|-----------|----------------------|----------------|
| 100K.png  | Add files via upload | 11 hours ago   |
| 150K.png  | Add files via upload | 11 hours ago   |
| 25K.png   | Add files via upload | 11 hours ago   |
| 50K.png   | Add files via upload | 11 hours ago   |
| 5K.png    | Add files via upload | 11 hours ago   |
| readme.md | Update readme.md     | 11 minutes ago |

readme.md [✎](#) [☰](#)

### Methods

A ZEISS Supra 40 Field Emission Scanning Electron Microscopy (FESEM) was employed to characterize the morphology of the CuO powder electrode.


### 5K magnification

A high-magnification scanning electron microscopy (SEM) image showing the morphology of CuO powder. The image displays a dense collection of elongated, needle-like or rod-like structures.

# Energy Dispersive X-ray

## Post-processing

## Data analysis & Visualization

 **FedeDat** Add files via upload

**Preview** Code Blame 4 lines

Search this file

|   | Cu    | Fe | O     | C | Na | Cl |
|---|-------|----|-------|---|----|----|
| 1 |       |    |       |   |    |    |
| 2 | 42.59 |    | 57.41 |   |    |    |
| 3 | 42.77 |    | 57.23 |   |    |    |
| 4 | 41.37 |    | 58.63 |   |    |    |

```
def cm_to_inch(cm):
    inch=float(cm/2.54)
    return inch

def AgCl_to_RHE(V_AgCl,pH,ref):
    if ref == "Ag/AgCl":
        V_RHE=+0.197+V_AgCl+0.059*pH
    return V_RHE

# Read the content of the file
url_EDX = "https://raw.githubusercontent.com/FedeDat/SuPERCO2/main/data/24-10-30_CuO/characterization/EDX.csv"

data_EDX = pd.read_csv(url_EDX, usecols=lambda column: pd.notnull(column))
data_EDX.dropna(axis=1, how='all', inplace=True)

#Calculate ratio between metal and oxygen

data_EDX[''+str(data_EDX.columns[0])+'/'+str(data_EDX.columns[1])+'']=data_EDX[data_EDX.columns[0]]/data_EDX[data_EDX.columns[1]]

data_EDX.index=list(range(1, len(data_EDX)+1))

ax1 = data_EDX.iloc[:, :-1].plot(kind='bar', stacked=True, figsize=(cm_to_inch(10), cm_to_inch(8)))
plt.legend(loc='upper center')

ax2 = ax1.twinx() # Create a second y-axis
ax2.plot(data_EDX.index-1, data_EDX[data_EDX.columns[-1]], linestyle='None', marker='o', markersize=8, markerfacecolor='red',

# Adding titles and labels
ax1.set_xticklabels(list(range(1, len(data_EDX)+1)), rotation=0, ha='right')
```

# X-ray Diffraction

## Post-processing

## Data analysis & Visualization

SuPERCO2 / data / 24-10-30\_CuO / characterization / XRD.csv

|    | Preview     | Code      | Blame | 2694 lines (2694 loc) · 46 KB |
|----|-------------|-----------|-------|-------------------------------|
| 28 | Angle       | Intensity |       |                               |
| 29 | 20.01513028 | 862       |       |                               |
| 30 | 20.04139085 | 848       |       |                               |
| 31 | 20.06765141 | 881       |       |                               |
| 32 | 20.09391198 | 856       |       |                               |
| 33 | 20.12017254 | 853       |       |                               |
| 34 | 20.14643311 | 891       |       |                               |
| 35 | 20.17269367 | 876       |       |                               |

```
def AgCl_to_RHE(V_AgCl,pH,ref):  
    if ref == "Ag/AgCl":  
        V_RHE=+0.197+V_AgCl+0.059*pH  
        return V_RHE  
  
url_XRD = "https://raw.githubusercontent.com/FedeDat/SuPERCO2/main/data/24-10-30_CuO/characterization/XRD.csv"  
  
data_XRD = pd.read_csv(url_XRD, header=None, usecols=[0, 1], names=['Angle', 'Intensity'])  
  
k_XRD = float(data_XRD[data_XRD.apply(lambda row: row.astype(str).str.contains('K-Alpha1 wavelength')).any(axis=1)]['Intensity'])  
n_XRD = int(data_XRD[data_XRD.apply(lambda row: row.astype(str).str.contains('No. of points')).any(axis=1)]['Intensity'])  
  
data_XRD_s1 = data_XRD.iloc[(data_XRD[data_XRD.apply(lambda row: row.astype(str).str.contains('Angle')).any(axis=1)].index[0])]  
data_XRD_s1 = data_XRD_s1.astype(float)  
  
data_XRD_s1['Intensity'] = data_XRD_s1['Intensity'] - np.min(data_XRD_s1['Intensity'])  
data_XRD_s1['Intensity']=data_XRD_s1['Intensity']/np.max(data_XRD_s1['Intensity'])  
  
data_XRD_s1['Intensity_savgol'] = savgol_filter(data_XRD_s1['Intensity'], window_length=10, polyorder=1)  
  
k_XRD = float(data_XRD[data_XRD.apply(lambda row: row.astype(str).str.contains('K-Alpha1 wavelength')).any(axis=1)]['Intensity'])  
n_XRD = int(data_XRD[data_XRD.apply(lambda row: row.astype(str).str.contains('No. of points')).any(axis=1)]['Intensity'])
```





# Electrochemical testing

Repository: [github.com/FedeDat/SuPERCO2](https://github.com/FedeDat/SuPERCO2)

item: data/24-10-30\_CuO/EC



# Open Circuit Potential

## Post-processing

## Data analysis & Visualization

SuPERCO2 / data / 24-10-30\_CuO / EC / OCV.csv



FedeDat Add files via upload

Preview

Code

Blame

548 lines (548 loc)

Search this file

|   |          |           |
|---|----------|-----------|
| 1 | pH       | 6.8       |
| 2 | Ref      | Ag/AgCl   |
| 3 |          |           |
| 4 | t (min)  | Ewe (V)   |
| 5 | 0.00E+00 | -2.14E-01 |

```
# Read the content of the file

url_OCV = "https://raw.githubusercontent.com/FedeDat/SuPERCO2/main/data/24-10-30_CuO/EC/OCV.csv"

data_OCV = pd.read_csv(url_OCV, header=None, names=['time', 'V'])

pH_OCV=float(data_OCV['V'][0])
ref_OCV=str(data_OCV['V'][1])

data_OCV=data_OCV[4:]

data_OCV=data_OCV.astype(float)

data_OCV['V']=AgCl_to_RHE(data_OCV['V'],6.8,ref_OCV)

plt.figure(figsize=(cm_to_inch(12), cm_to_inch(8)))

plt.plot(data_OCV['time'],data_OCV['V'], )

plt.title('Open circuit potential')

plt.xlabel(r"$t$ (min)")
plt.ylabel(r"$V$ (vs RHE)")

plt.ylim(min(np.min(data_OCV['V']),0.5*np.mean(data_OCV['V'])),max(np.max(data_OCV['V']),1.5*np.mean(data_OCV['V'])))

plt.savefig('OCV.png', format='png', dpi=300, transparent=True, bbox_inches='tight')


plt.close()
```

# Cyclic voltammetry

## Post-processing

## Data analysis & Visualization

SuPERCO2 / data / 24-10-30\_CuO / EC / CV-CO2.csv

 **FedeDat** Add files via upload

**Preview** Code Blame 1810 lines (1810 loc)

Search this file

|   |           |            |
|---|-----------|------------|
| 1 | pH        | 6.8        |
| 2 | Ref       | Ag/AgCl    |
| 3 |           |            |
| 4 | Ewe (V)   | j (mA/cm2) |
| 5 | -1.45E+00 | -8.35E+00  |

```
data_CV_CO2['V']=AgCl_to_RHE(data_CV_CO2['V'],6.8,ref_CV_CO2)

url_CV_N2 = "https://raw.githubusercontent.com/FedeDat/SuPERCO2/main/data/24-10-30_CuO/EC/CV-N2.csv"
data_CV_N2 = pd.read_csv(url_CV_N2, header=None, names=['V', 'j'])

pH_CV_N2=float(data_CV_N2['j'][0])
ref_CV_N2=str(data_CV_N2['j'][1])

data_CV_N2=data_CV_N2[4:]

data_CV_N2=data_CV_N2.astype(float)

data_CV_N2['V']=AgCl_to_RHE(data_CV_N2['V'],6.8,ref_CV_N2)

plt.figure(figsize=(cm_to_inch(12), cm_to_inch(12)))

plt.plot(data_CV_CO2['V'],data_CV_CO2['j'], label=r"CO2 saturated",color='red')
plt.plot(data_CV_N2['V'],data_CV_N2['j'], label=r"N2 saturated",color='gray')

plt.title('Cyclic voltammetry')
plt.legend(loc='best')

plt.xlabel(r"$V$ (vs RHE)")
plt.ylabel(r"$j$ (mA/cm2)")

plt.savefig('CV.png', format='png', dpi=300, transparent=True, bbox_inches='tight')

plt.close()
```

### alculation of CO<sub>2</sub> reduction activity and selectivity

```
# Analyse Faradaic Efficiency
```

# Chronoamperometry

## Post-processing (CA)

SuPERCO2 / data / 24-10-30\_CuO / EC / CA1.csv

FedeDat Add files via upload

Preview Code Blame 21125 lines (21125 loc) · 392 Bytes

Search this file

|   |                     |            |
|---|---------------------|------------|
| 1 | pH                  | 6.8        |
| 2 | Ref                 | Ag/AgCl    |
| 3 | V                   | -1.199     |
| 4 | Flow Rate (NmL/min) | 20         |
| 5 | Type                | CA         |
| 6 |                     |            |
| 7 | t (s)               | j (mA/cm2) |
| 8 | 2.63E+00            | -2.43E-05  |

## Post-processing (GC)

SuPERCO2 / data / 24-10-30\_CuO / EC / GC1.csv

FedeDat Add files via upload

Preview Code Blame 11 lines (11 loc) · 392 Bytes

Search this file

|   |            |    |            |            |    |     |      |
|---|------------|----|------------|------------|----|-----|------|
| 1 | H2         | O2 | N2         | CH4        | CO | CO2 | C2H4 |
| 2 | 0          | 0  | 0.938709   | 0.00071317 | 0  | 0   | 0    |
| 3 | 0.08128675 | 0  | 0.94220475 | 0          | 0  | 0   | 0    |
| 4 | 0.15007389 | 0  | 0.93674202 | 0          | 0  | 0   | 0    |
| 5 | 0.14496378 | 0  | 0.93663998 | 0          | 0  | 0   | 0    |
| 6 | 0.14065497 | 0  | 0.92155004 | 0          | 0  | 0   | 0    |
| 7 | 0.14463469 | 0  | 0.91517259 | 0          | 0  | 0   | 0    |
| 8 | 0.14796505 | 0  | 0.91493298 | 0          | 0  | 0   | 0    |

# Activity & Selectivity

## Data analysis & Visualization

### Calculation of CO<sub>2</sub> reduction activity and selectivity

```
▶ # Analyse Faradaic Efficiency

# Github API URL for the folder
api_url = "https://api.github.com/repos/FedeDat/SuPERCO2/contents/data/24-10-30_CuO/EC"

# Get the file listing
response = requests.get(api_url)
files = response.json()

metadata_EC = pd.DataFrame(columns=['pH', 'ref', 'V'])
anl_EC = pd.DataFrame(columns=['V_RHE', 'Flow Rate', 'Mode', 'Average j', 'Q (mC)'])

data = pd.DataFrame()

n1=0

plt.figure(figsize=(cm_to_inch(18), cm_to_inch(12)))
```

## The FAIR CO<sub>2</sub>R problem

Open data is limited in CO<sub>2</sub>R

Zenodo limits interoperability

AI mining on published data

## The SuPERCO2 concept

Early data sharing

Characterization: EDX XRD

Electrochemistry: OCV CV CA

## Next steps

Polish & upload Python scripts

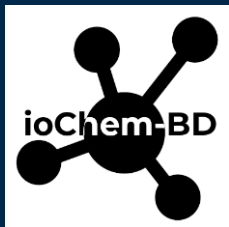
Provide a tutorial

Automatize upload & analysis

Create new modules: synthesis

# Acknowledgements

## Inspiration



M.Eng. Moisés Álvarez  
(ICIQ)

## Exp collaboration

M.Sc. Ali Zarei  
(catalyst synthesis, electrochemical testing)

Prof. Simelys Hernández  
(lab Access)



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## Financial support



SuPERCO2 has received funding from the European Union's Horizon Europe research and innovation programme under the Marie Skłodowska-Curie grant agreement No 101104004.



Thank you!  
Questions or feedbacks?

Here or via email  
([federico.dattila@polito.it](mailto:federico.dattila@polito.it))



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