

QOG – Anotações Slide 2

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Nomenclatura: Alcanos

1 Prefixo

Relacionado ao numero de carbonos

1. met	6. hex	11. undec	16. hexadec
2. et	7. hep	12. dodec	17. heptadec
3. prop	8. oct	13. tridec	18. octadec
4. but	9. non	14. tetradec	19. nonadec
5. pen	10. dec	15. pentadec	20. icos
30. tria			

2 Intermediários

referencia as ligações (suprime o “an”, ex: etano→etino)

an Somente ligações simples

in 1 ligação tripla

en 1 ligação dupla

diin 2 ligações triplas

dien 2 ligações duplas

enin 1 dupla e 1 tripla

3 Sufixos 1

o: Hidrocarboneto ---CH_2

ol: Álcool e enol ---OH

al: Aldeído $\text{---C}\begin{smallmatrix} \text{O} \\ \text{//} \\ \text{H} \end{smallmatrix}$

ona: Cetona =O

óico: Ácido Carboxilo $\text{---C}\begin{smallmatrix} \text{O} \\ \text{//} \\ \text{OH} \end{smallmatrix}$

amina: ---NH_2

amida: $\text{R}-\text{C}\begin{smallmatrix} \text{H} \\ \text{//} \\ \text{N} \end{smallmatrix}\begin{smallmatrix} \text{R} \\ \text{R}' \end{smallmatrix}$

sulfônico: Ácido Sulfônico $\text{R}-\text{S}\begin{smallmatrix} \text{O} \\ \text{||} \\ \text{OH} \\ \text{O} \end{smallmatrix}$

óxi: Éter $\text{R}_1-\text{O}-\text{R}_2$

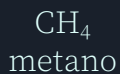
ato ... ila: Éster $\text{R}_1-\text{C}\begin{smallmatrix} \text{O} \\ \text{//} \\ \text{O} \end{smallmatrix}-\text{R}_2$

ato de ...: Sal Orgânico $\text{R}_1-\text{C}\begin{smallmatrix} \text{O} \\ \text{//} \\ \text{O} \end{smallmatrix}-\text{Metal}$

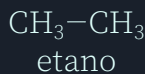
Exemplo 1

Hidrocarbonetos

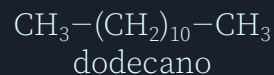
E1.1)



E1.2)

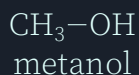


E1.3)

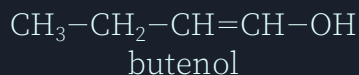


Álcools e Enols

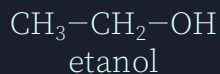
E1.4)



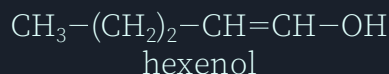
E1.6)

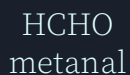
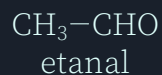
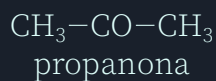
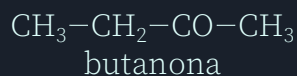
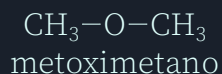
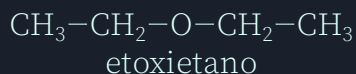
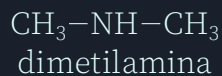
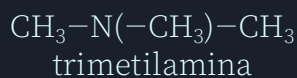
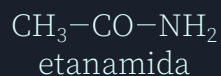
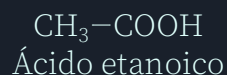


E1.5)



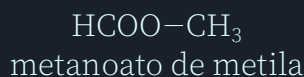
E1.7)



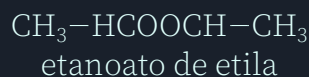
Aldeídos**E1.8)****E1.9)****Cetona****E1.10)****E1.11)****Eter****E1.12)****E1.13)****Amina****E1.14)****E1.15)****Amida****E1.16)****E1.17)****Ácido Carboxílico****E1.18)****E1.19)**

Éster

E1.20)

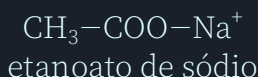


E1.21)

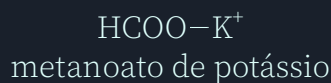


Sal Orgânico

E1.22)

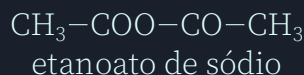


E1.23)

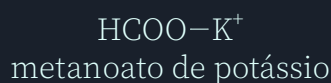


Anidrido de Ácido

E1.24)



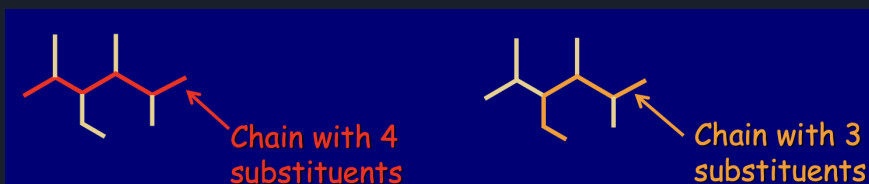
E1.25)



4 Radicais Substituintes

Quando há ramificações, para poder nomear precisa seguir o passo a passo

4.1 Identificar a cadeia mais longa



em empate escolhe a que tem mais substituintes

4.2 Nomear os substituintes

halogenio recebe o nome do halogenio Br,F,Cl,I

Cadeia org simples recebe prefixo “il”

Canonicos: Estruturas canonicas que devemos conhecer

Multiplos substituintes identicos se juntam recebendo um prefixo numérico

Exemplo



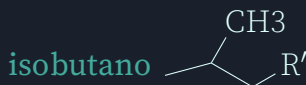
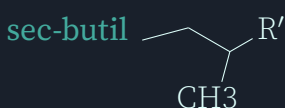
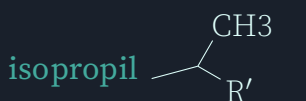
Complexidade da cadeia simples usa o prefixo numérico comum, complexa (multiplas bifurcações) usa os prefixos:

2. bis

3. tris

4. tetrakis

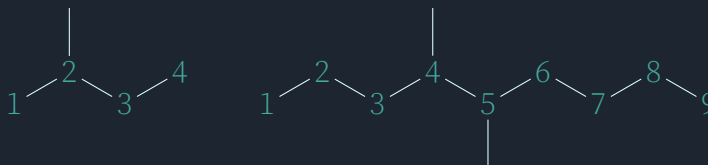
Estruturas Canonicas



4.3 Numerar a cadeia

Se nomeia da extremidade mais próxima aos substituintes

Exemplo

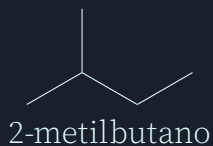


4.4 Nomear em ordem alfabética

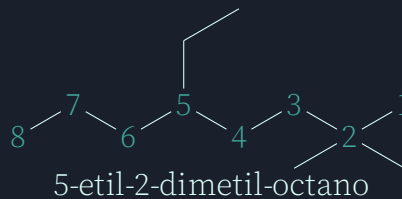
Não leva em conta o prefixo numérico de multiplas ramificações exceto quando as ramificações são complexas

Exemplo 2

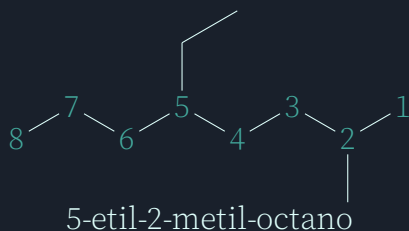
E2.1)



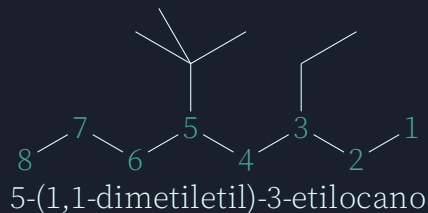
E2.3)



E2.2)



E2.4)



E2.5)

