

EB – Exercises

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Questão 1

A 150 m^3 reactor is operated at 35°C to produce biomass from glucose. The O_2 consumption rate is $1.5 \text{ kg m}^{-3} \text{ h}^{-1}$. The stirrer dissipates heat at the speed of 1 kW m^{-3} . The cooling water flows at a temperature of 10°C and at a flow rate of $60 \text{ m}^3/\text{h}$, it passes inside a coil placed inside the reactor. If the system operates in steady state, determine the temperature of the cooling water leaving the reactor.

Data:

$$\bullet Q = 460 \text{ kJ mol (O}_2\text{)}^{-1}$$

$$\bullet C_{p\text{H}_2\text{O}} = 75.4 \text{ J mol}^{-1} \text{ }^\circ\text{C}^{-1}$$

Resposta

$$T_1 = T_0 + \Delta T;$$

$$\Delta H = \Delta H_{rxn} + W_s;$$

$$\Delta H_{rxn} = Q * C_{\text{O}_2} * V = \begin{pmatrix} 460 \text{ kJ mol}^{-1} & * \\ 1.5 \text{ kg m}^{-3} \text{ h}^{-1} & * \\ 3600^{-1} \text{ h s}^{-1} & * \\ 32^{-1} \text{ mol g}^{-1} & * \\ 150 \text{ m}^3 & \end{pmatrix} \cong 898.438 \text{ kJ s}^{-1};$$

$$W_s = 1 \text{ kW m}^{-3} 150 \text{ m}^3 = 150 \text{ kW};$$

$$\Delta H \cong 898.438 \text{ kJ s}^{-1} + 150 \text{ kJ s}^{-1} \cong 1048.438 \text{ kJ s}^{-1} =$$

$$= M C_p \Delta T =$$

$$= (v * \rho_{\text{H}_2\text{O}}) \left(75.4 \text{ J mol}^{-1} \text{ }^\circ\text{C}^{-1} \frac{M_{w\text{H}_2\text{O}} \text{ g}}{\text{mol}} \right) \Delta T =$$

$$= (60 * 1000 \text{ kg/h}) \left(\frac{75.4}{18} \text{ kJ kg}^{-1} \text{ }^\circ\text{C}^{-1} \right) \Delta T \Rightarrow$$

$$\Rightarrow T_1 / ^\circ\text{C} = T_0 + \Delta T \cong 10 + \left(\frac{1048.438 \text{ kJ s}^{-1}}{\frac{75.4 * 60000}{18} \text{ kJ }^\circ\text{C}^{-1} \text{ h}^{-1} \frac{\text{h}}{3600 \text{ s}}} \right) \cong 25.017$$

Questão 2

A fermenter used to produce an antibiotic should have a temperature of 27°C . After considering the oxygen requirements of the microorganisms and the heat dissipated by the stirrer, the maximum amount of heat to be transferred was estimated at 550 kW . The cooling water enters at a temperature of 10°C and leaves at 25°C . The heat transfer coefficient in the fermentation fluid was estimated at $2150\text{ W m}^{-2}\text{ }^{\circ}\text{C}^{-1}$ and the heat transfer coefficient of the cooling water has the value of $14000\text{ W m}^{-2}\text{ }^{\circ}\text{C}^{-1}$. The steel cooling coil has an outer diameter of 8 cm and a thickness of 5 mm , the thermal conductivity of steel is $60\text{ W m}^{-1}\text{ }^{\circ}\text{C}^{-1}$. Calculate the length of coil needed under these conditions.

Resposta

$$\Delta T = \frac{2T_{\text{fluido}} - (T_1 + T_2)}{2} = \frac{2 * 28 - (T_1 + T_2)}{2} = 9.5^{\circ}\text{C};$$

$$Q = h A \Delta T;$$

$$h^{-1} = h_i^{-1} + h_e^{-1} + h_w^{-1} B = 2150^{-1} + 14000^{-1} + 60^{-1} * 5 \text{ E}^{-3} \cong 1614 \text{ m}^{-2}$$

$$A = \frac{Q}{h \Delta T} \cong \frac{550}{1614 * 9.5} \cong 35.870 \implies$$

$$\implies A = 2 \pi r l \implies l = \frac{A}{2 \pi r} \cong \frac{35.870}{2 \pi (8/2)} \cong 142.8 \text{ m}$$