ERQ I – Teste 2023.2 Resolução

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Conteúdo

Questão 1

- A *⇒* B
 - fase gásosa
- - Reator tubular adiabático
- - · A (10%)
 - $T_0 = 200 \, ^{\circ}\text{C} = 473.15 \, \text{K}$
- - $v_0 = 10 \,\mathrm{dm}^3/\mathrm{s} = 600 \,\mathrm{L/min}$
 - $R = 1.987 \, \text{cal mol}^{-1} \, \text{K}^{-1}$ • gráfico $X \times T$
 - 0.9 0.8 0.7 0.6 **x** 0.5 0.4

• $C_{pA} = C_{pB} = 5 \text{ cal/mol K}$

• $C_{pI} = 12 \, \text{cal/mol K}$

• $k_{d(473 \text{ K})} = 1.17 \, \text{min}^{-1}$

• $Ea = 20 \, \text{kcal/mol}$

• $K_{e(473 \text{ K})} = 40$

0.3 0.2 0.1 470 520 570 620 670 720 770 820 870 920 970 1020 T(K)

Exotérmica pois $X \propto T^{-1}$ ou seja diminui conforme a temperatura almenta.

Q1 b.

Q1 a.

Resposta

Calor de reação

 $\Rightarrow \Delta H_R = \frac{-R}{(T^{-1} - T_P^{-1})} \ln \frac{K_{e(T)}}{K_{e(T_P)}};$

Endotermica/exotermica

Resposta $\Delta H_R : K_{e(T)} = K_{e(T_R)} \exp\left(-\frac{\Delta H_R}{R}(T^{-1} - T_R^{-1})\right) \implies$

$K_{e\,(T)}:K_{e\,(T)}=\frac{p_{p\,B}}{p_{p\,A}}=\frac{C_{p\,B}\,R\,T}{C_{p\,A}\,R\,T}=\frac{C_{p\,A_0}\,X}{C_{p\,A_0}\,(1-X)}=\frac{1}{1/X-1}$

$$\implies X = \frac{1}{1 + 1/K_e} \implies X = 0.5 \begin{cases} K_e = 1 \\ T \cong 619 \text{ K} \end{cases}$$

Resposta

 $\cong -14.701 \, \text{kcal/mol}$

Q1 c.
$$X_{eq} \wedge T_{eq}$$

 $X_{eq}\cong 0.79 \quad \wedge \quad T_{eq}\cong 546\, \mathrm{K}$

 $\therefore \Delta H_R = \frac{-R}{(T^{-1} - T_R^{-1})} \ln \frac{K_{e(T)}}{K_{e(T_R)}} \cong \frac{-1.987}{(619^{-1} - 473^{-1})} \ln \frac{1}{40} \cong$

$X_{(520 \, \mathrm{K})} = \frac{(C_{p\,A} + \theta_I \, C_{p\,I})(T - T_0)}{-\Delta H_R} = \frac{(C_{p\,A} + \frac{Y_{I\,0}}{Y_{A\,0}} \, C_{p\,I})(T - T_0)}{-\Delta H_P} \cong$ $\cong \frac{\left(5 + \frac{0.9}{0.1} \cdot 12\right) \left(520 - 473\right)}{14.701 \cdot E^3} \cong 0.360$ $\begin{cases} X_0 = 0; & T_0 = 473 \\ X_1 \cong 0.360; & T_1 = 520 \end{cases}$

Q1 d.

 $V = \int_{0}^{X} \frac{F_{A0} dX}{-r_{A}} = \int_{0}^{X} \frac{C_{A0} v_{0} dX}{-r_{A}};$

 $-r_A = k(C_A - C_B/K_e) =$

Volume do reator para $X_1 = 95\% X_e$

$= k \left(\left(\frac{C_{A0}(1-X)}{1+\varepsilon X} \frac{T_0}{T} \right) - \left(\frac{C_{A0}X}{1+\varepsilon X} \frac{T_0}{T} \right) / K_e \right) =$ $=k\left(\frac{C_{A0}(1-X(1-1/K_e))}{1+\varepsilon X}\frac{T_0}{T}\right)=$

$$= k \left(\frac{C_{A0}(1 - X(1 - 1/K_e))}{1 + y_{A0} \delta X} \frac{T_0}{T} \right) =$$

$$= k \left(\frac{C_{A0}(1 - X(1 - 1/K_e))}{1 + y_{A0} (1 - 1) X} \frac{T_0}{T} \right) =$$

$$= k C_{A0} (1 - X(1 - K_e)) \frac{T_0}{T} \implies$$

$$\implies V = \int_0^X \frac{C_{A0} v_0 \, dX}{k C_{A0} (1 - X(1 - K_e)) \frac{T_0}{T}} =$$

$$= \int_0^X \frac{v_0 \, dX}{k (1 - X(1 - K_e)) \frac{T_0}{T}} =$$

$$= \int_0^{.95*.79} \frac{600 \, dX}{k (1 - X(1 - K_e)) \frac{473}{T}} =$$

$$\cong \int_0^{0.7505} \frac{1.268 \, dX}{k (1 - X(1 - K_e))/T}$$

$\implies T = T_0 - \frac{X \Delta H_R}{C_{nA} + \theta_I C_{nI}} \cong 473 - \frac{-X 14.701 E^3}{5 + \frac{.9}{7} 12} \cong$ $\cong 473 + X 130.094;$

Simpson:

 $f(X) = \frac{1.268 \text{ d}X}{k(1 - X(1 - K_e))/T};$

 $T: X = \frac{(C_{pA} + \theta_I C_{pI})(T - T_0)}{-\Lambda H_D} \Longrightarrow$

$$k_{(T)} = k_{(T_R)} \exp\left(-\frac{Ea}{R}(T^{-1} - T_R^{-1})\right) =$$

$$\approx 1.17 \exp\left(-\frac{20 E^3}{1.987}(T^{-1} - 473^{-1})\right) \approx$$

$$\approx 1.17 \exp\left(-10.064 E^3(T^{-1} - 473^{-1})\right);$$

$$\cong 40 \exp \left(7.398 \,\mathrm{E}^3 \left(T^{-1} - 473^{-1}\right)\right)$$

$$h = \frac{0.7505}{2} = 0.37525 \begin{cases} X_0 = 0 \\ X_1 = 0.37525 \\ X_2 = 0.7505 \end{cases}$$

 $K_{e(T)} = K_{e(T_R)} \exp\left(-\frac{\Delta H}{R} (T^{-1} - T_R^{-1})\right) \cong$

 $\cong 40 \exp \left(\frac{14.701 \,\mathrm{E}^3}{1.987} \,(T^{-1} - 473^{-1})\right) \cong$

$$\therefore V = \frac{h}{3} \left(f_{(X_0)} + 4 \, f_{(X_1)} + f_{(X_2)} \right)$$
 Encontramos com os 3 pontos de X os valores para T, $k_{(T)}$,

mente são usados para encontrar V

Q1 e. Y_{A0} para $X_{eq} = 90\%$

 $K_{e(T)}$ que podem ser usados para encontrar f(X) que final-

$Y_{A0}: -X_{(T)} \Delta H_R = (C_{pA} + \theta_I C_{pI}) (T - T_0) =$

Resposta

$$= \left(C_{pA} + \frac{Y_{I0}}{Y_{A0}}C_{pI}\right)(T - T_0) =$$

$$= \left(C_{pA} + \frac{1 - Y_{A0}}{Y_{A0}} C_{pI}\right) (T - T_0);$$

$$= \left(C_{pA} + \frac{1}{Y_{A0}}C_{pI}\right)(I - I_{0});$$

$$T_{eq} \cong 473 + 0.9 * 130.094 \cong 590.085 \implies$$

$$\implies Y_{A0} = \left(1 + \frac{\frac{-X_{(T)}\Delta H_{R}}{T - T_{0}} - C_{pA}}{C_{pI}}\right)^{-1} =$$

$$= \left(1 + \frac{\frac{0.9 * 14.701 E^{3}}{590.085 - 473} - 8}{12}\right)^{-1} \cong 0.103$$

Questão 2

• A → B

CSTR não adiabático

• $V = 6 \, \text{m}^3$

· Em estado estacionário

• $v_0 = 6 \, \text{L/min} = 360 \, \text{L/h}$

• $Y_{A0} = 0.1$

• $\Delta H_R = -100.7 \,\mathrm{kJ/mol}$

• $C_{pA} = C_{pB} = 34.6 \, \text{J/mol K}$

• $C_{pI} = 75.4 \,\mathrm{J/mol}\,\mathrm{K}$

• $R = 8.314 \,\mathrm{I} \,\mathrm{mol}^{-1} \,\mathrm{K}^{-1}$

• $C_{A0} = 1 \, \text{M}$

• $T_0 = 298 \, \text{K}$

• $k_{(298 \text{ K})} = 0.0227 \,\text{h}^{-1}$

• com parede envolvida até 85% com agua a $100\,^{\circ}\text{C} = 373\,\text{K}$

Q2 a.

Equações das curvas

Resposta

$$G_{(T)} = -\Delta H_R X$$

$$X: r_{(T)} = G_{(T)} = \frac{Ua}{F_{A0}} (T - T_0) + \sum \theta_i Cp i(T - T_0) =$$

$$= \frac{\left(\frac{Q}{T_0 - T}\right)}{F_{A0}} (T - T_0) + \frac{1 - Y_{A0}}{Y_{A0}} C_{pB} (T - T_0) =$$

$$= \frac{-Q}{C_{A0} v_0} + (1/Y_{A0} - 1) C_{pA} X(T - T_0) \implies$$

$$\implies X = \frac{G_{(T)} + \frac{Q}{C_{A0} v_0}}{(1/Y_{A0} - 1) C_{pA} (T - T_0)} \implies$$

$$\Rightarrow G_{(T)} =$$

$$= -\Delta H_R \left(\frac{Q^{\cdot}}{C_{A0} v_0}\right) \left((1/Y_{A0} - 1) C_{pA} (T - T_0) + \Delta H_R \right)^{-1} =$$

$$= -100.7 \,\mathrm{E}^3 \left(\frac{Q^{\cdot}}{1 * 360}\right) \left((1/0.1 - 1) 1 (T - 298) - 100.7 \,\mathrm{E}^3 \right)^{-1} \cong$$

$$\cong \frac{-31.080 \, Q^{\cdot}}{T - 11.487 \,\mathrm{E}^3}$$

Por não conseguir calcular Q^\cdot deixei como variável, se mostra necessário uma vez que a equação final deve ser uma reta, $Q^\cdot \propto T^2$ o que nao verificou

Q2 d.

Ea usando gráfico