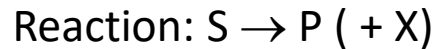


**Problem 4.1.**

Consider a cell culture with negligible growth ( $\mu \approx 0$ ) characterized by the following biological reaction:



Kinetic: 
$$v_s = \frac{v_{s,\max} S}{K_S + S}$$

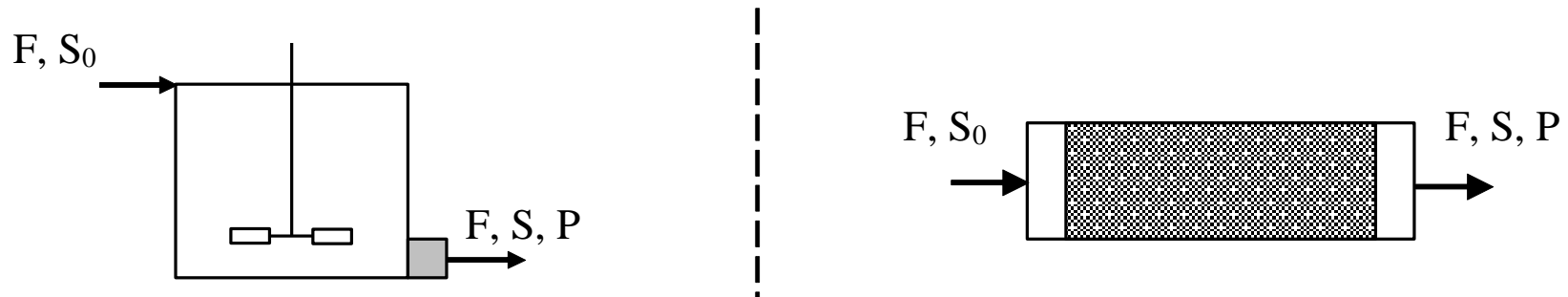
$$v_{s,\max} = 0.6 \text{ g-subst g-cel}^{-1} \text{ h}^{-1}$$

$$K_S = 0.01 \text{ g/L}$$

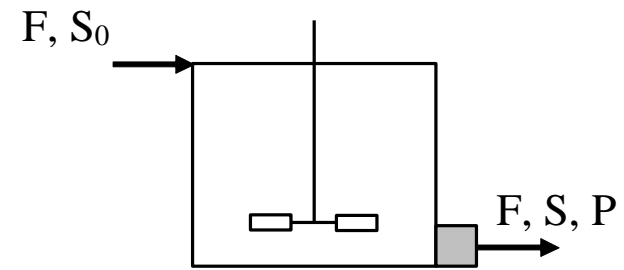
$$Y'_{p/s} = 0.2 \text{ g-prod/g-subst}$$

(negligible maintenance)

Consider the Bioreactors indicated in the figure. In the case of the CSTR, a filter in the output current prevents the Bioreactor from being washed. In the case of PFR, cells are immobilized on a solid support without diffusional limitations for the transport of 'S' and 'P'. In either case the cells are evenly distributed with concentration  $X=12.1 \text{ g/l}$ .



- a) Size the Bioreactors indicated in the figure for a productivity of 60 g product/h knowing that the input current has a substrate concentration  $S_0=13.2$  g/l and that its conversion is 95%. Consider that the substrate is in large excess in each of the reactors and that consequently the kinetics is of order 0.

**CSTR**

Balance to the substrate

$$\frac{F}{V} S_0 - \frac{F}{V} S - r_s = 0$$

$r_s$  = volumetric substrate consumption rate

$$F S_0 - F S - r_s V = 0$$

$$F (S_0 - S) - r_s V = 0$$

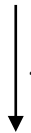
$$F (S_0 - S) - V_s X V = 0$$

$$r_s = V_s X$$

$V_s$  = specific substrate consumption rate

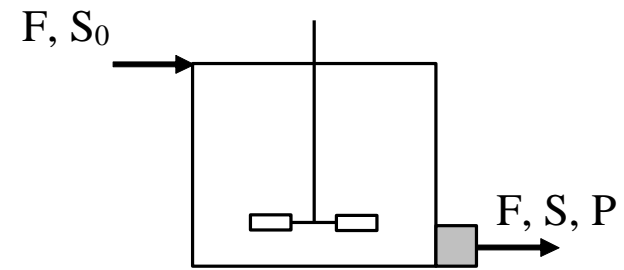
Calcular S

$$S_0 = 13,2 \text{ g/l}$$



*95% conversion*

$$S = 0,66 \text{ g/l}$$

**CSTR**

$$S \gg K_s$$

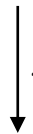
$$0.66 \text{ g/L} \gg 0.01 \text{ g/L}$$

$$V_s = V_{s,\max} \frac{S}{K_s + S}$$

$$V_s \approx V_{s,\max}$$

$$F(S_0 - S) - V_{s,\max} X V = 0$$

$$S_0 = 13,2 \text{ g/l}$$



*95% conversion*

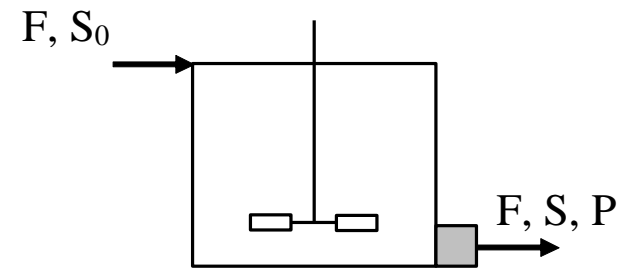
$$S = 0,66 \text{ g/l}$$

$$Y_{p/s} = \frac{\Delta P}{\Delta S} = \frac{P - P_0}{S_0 - S}$$

$$P = 2,51 \text{ g/l}$$

$$F P = 60 \text{ g/h}$$

$$F = 23,9 \text{ l/h}$$

**CSTR**

$$S \gg K_s \quad V_s = V_{s,\max} \frac{S}{K_s + S}$$

$$V_s \approx V_{s,\max}$$

$$F(S_0 - S) - V_{s,\max} X V = 0$$

$$V = 41,3 \text{ l}$$

**PFR**

Balance to the substrate

$$u \frac{dS}{dz} = -r_s$$

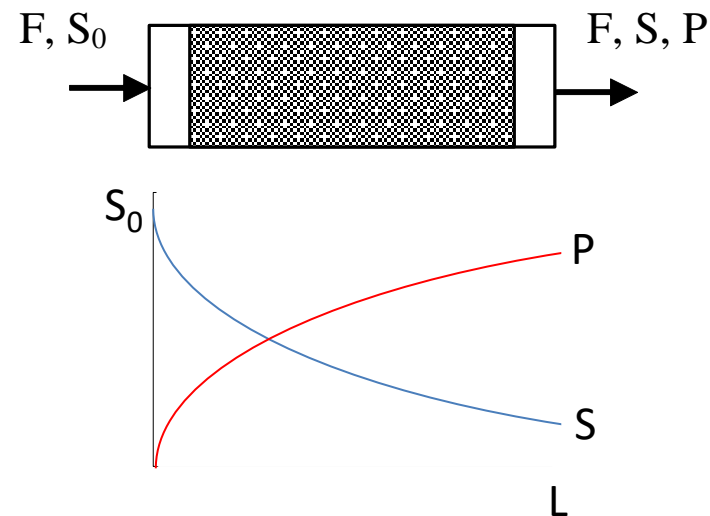
$$u \frac{dS}{dz} = -V_s X$$

$$u dS = -V_{s,max} X dz$$

$$u \int_{S_0}^S dS = -V_{s,max} X \int_0^L dz$$

$$u (S - S_0) = -V_{s,max} X (L - 0)$$

$$u (S - S_0) = -V_{s,max} X L$$



$$\frac{F}{A} (S - S_0) = -V_{s,max} X L$$

$$F (S - S_0) = -V_{s,max} X L A$$

$$F (S - S_0) = -V_{s,max} X V$$

$$V = 41,3 \text{ l}$$



- b) Size the reactors for the same conditions as in point a) considering that the saturation constant is  $K_s = 25 \text{ g/l}$ ,  $S_0 = 2.5 \text{ g/l}$  and  $X = 1.21 \text{ g/l}$ .

$$\begin{array}{c} S_0 = 2,5 \text{ g/l} \\ \downarrow \text{95\% conversion} \\ S = 0,125 \text{ g/l} \end{array}$$

$$Y_{p/s} = \frac{\Delta P}{\Delta S} = \frac{P - P_0}{S_0 - S}$$

$$P = 0,475 \text{ g/l}$$

$$60 \text{ g/h}$$

$$F P = 60 \text{ g/h}$$

$$F = 126,3 \text{ l/h}$$

CSTR

$$S \ll K_s$$

$$0.125 \text{ g/L} \ll 25 \text{ g/L}$$

$$V_s = V_{s,\max} \frac{S}{K_s + S}$$

$$V_s \approx \frac{V_{s,\max}}{K_s} S$$

$$V_s \approx 0,003 \text{ gS/gX.h}$$

$$F(S_0 - S) - V_s X V = 0$$

$$V = 83322 \text{ l} = 83,3 \text{ m}^3$$

**PFR**

Balance to the substrate

$$u \frac{dS}{dz} = -r_s$$

$$\frac{F}{A} \ln \frac{S}{S_0} = -\frac{V_{s,\max}}{K_s} \times L$$

$$u \frac{dS}{dz} = -V_s \times$$

$$F \ln \frac{S}{S_0} = -\frac{V_{s,\max}}{K_s} \times L \times A$$

$$u \frac{dS}{S} = -\frac{V_{s,\max}}{K_s} \times dz$$

$$F \ln \frac{S}{S_0} = -\frac{V_{s,\max}}{K_s} \times V$$

$$u \int_{S_0}^S \frac{dS}{S} = -\frac{V_{s,\max}}{K_s} \times \int_0^L dz$$

$$V = 13029 \text{ l} = 13,0 \text{ m}^3$$

$$u \ln \frac{S}{S_0} = -\frac{V_{s,\max}}{K_s} \times (L - 0)$$

c) Comment on the results obtained in a) and b)

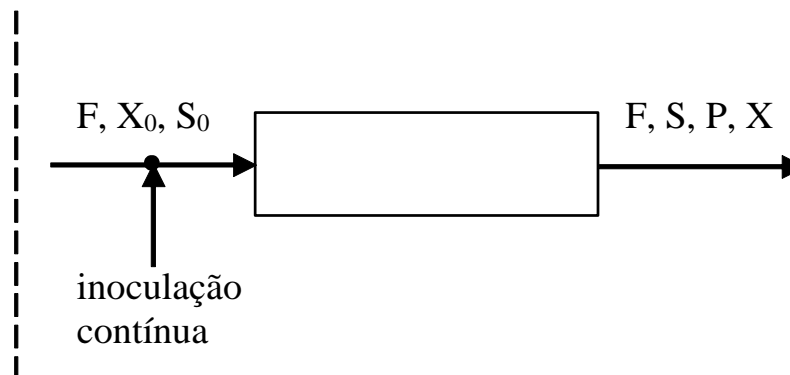
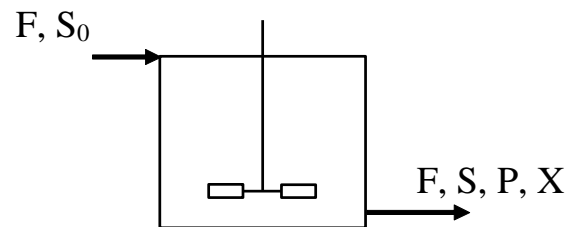
When cell growth is negligible

and  $S \gg K_s$   $V_{\text{CSTR}} \approx V_{\text{PFR}}$

and  $S \leq K_s$   $V_{\text{CSTR}} > V_{\text{PFR}}$

## Problema 4.2.

Consider the following Biorreactors



In which the biological reaction occurs

Reaction:  $S \rightarrow P + X$

kinetic: 
$$\mu = \frac{\mu_{\max} S}{K_S + S}$$

$$\mu_{\max} = 0.3 \text{ h}^{-1}$$

$$K_S = 0.01 \text{ g/L}$$

$$Y'_{xs} = 0.5 \text{ g cell/g subs}$$

$$Y'_{xp} = 4.3 \text{ g cell/g prod}$$

(negligible maintenance)

It is intended to dimension each one of the Bioreactors to treat a stream with  $S_0 = 150 \text{ g/l}$  and to convert 90% of substrate into product and cells. The size of the reactor must be such that the absolute productivity is 1 kg of product per hour.

- a) Size a CSTR for the specified conditions knowing that the reactor is initially inoculated with 1 g/l of cells after which the process converges to steady state.

steady-state

$$\mu = D = \frac{F}{V}$$

Calcular  $\mu$

$$\mu = \frac{\mu_{max} S}{K_s + S} = 0,2998 \text{ h}^{-1}$$

Calcular S

$$\begin{array}{c} S_0 = 150 \text{ g/l} \\ \downarrow 90\% \text{ conversion} \\ S = 15 \text{ g/l} \end{array}$$

Calcular F

$$1 \text{ kg/h}$$

$$F P = 1000 \text{ g/h}$$

$$F = 63,7 \text{ l/h}$$

Calcular P

$$Y_{x/p} = \frac{\Delta X}{\Delta P} = \frac{X - X_0}{P - P_0}$$

$$P = 15,7 \text{ g/l}$$

Calcular X

$$Y_{x/s} = \frac{\Delta X}{\Delta S} = \frac{X - X_0}{S_0 - S}$$

$$X = 67,5 \text{ g/l}$$

$$D = \frac{F}{V}$$

$$V = 213,8 \text{ l}$$



b) Size a PFR for the same conditions but with continuous inoculation with  $X_0 = 1$  g/l. Compare the result with the one obtained in point a) and comment.

Balance to the biomass

$$u \frac{dX}{dz} = r_x = \mu X$$

$$\mu = \frac{\mu_{\max} S}{K_s + S} = \frac{\mu_{\max} S}{K_s + S} \approx \mu_{\max}$$

$$u \frac{dX}{X} = \mu_{\max} dz$$

$$S \gg K_s$$

$$u \int_{X_0}^X \frac{dX}{X} = \mu_{\max} \int_0^L dz$$

$$F \ln \frac{X}{X_0} = \mu_{\max} L A$$

$$u \ln \frac{X}{X_0} = \mu_{\max} L$$

$$F \ln \frac{X}{X_0} = \mu_{\max} V$$

$$\frac{F}{A} \ln \frac{X}{X_0} = \mu_{\max} L$$

$$V = 897,6 \text{ l}$$

c) Determine the length and diameter of the PFR from point b). Consider that the physical properties of the culture medium are similar to those of water.

$$Re = \frac{\rho u d}{\mu}$$

Re = Reynolds number

Re < 2000 *plug flow*

$\rho$  = density = 1000 kg/m<sup>3</sup>

*water*

$\mu$  = viscosity = 10<sup>-3</sup> Pa.s

*water*

$$1 \text{ Pa.s} = 1 \text{ kg/m.s}$$

$$u = \frac{F}{A} = \frac{F}{\frac{\pi d^2}{4}} = \frac{4 F}{\pi d^2}$$

$$F = 63,7 \text{ l/h} = 1,769 \times 10^{-5} \text{ m}^3/\text{s}$$

$$Re = \frac{\rho u d}{\mu} = \frac{1000 \text{ kg/m}^3 \times \frac{4 F}{\pi d^2} \text{ m}^3/\text{s} \times d}{10^{-3} \text{ kg/m.s}} = \frac{0,071}{\pi \times 10^{-3} \times d}$$

$$Re < 2000 \quad \frac{0,071}{\pi \times 10^{-3} \times d} < 2000$$

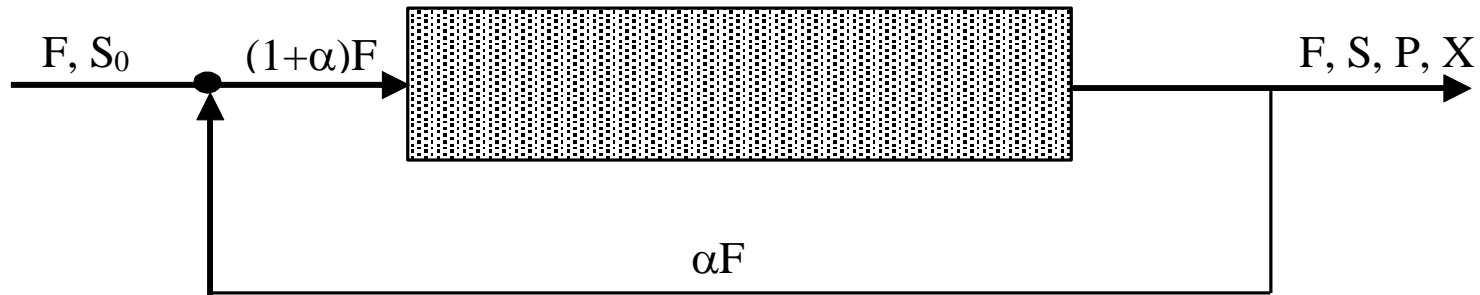
$$d > 1,1 \text{ cm}$$

$$V = A L$$

$$L = 9444 \text{ m}$$

**Problema 4.3.**

Consider a piston Bioreactor with backflow from the outlet to the inlet as indicated in the figure



with  $\alpha$  the ratio (reflow flow)/(inflow flow). The input current has  $S_0=100\text{g/l}$  and the flow through the system is  $F=10\text{ l/h}$ . The intended conversion is 90% of the input substrate. Consider that the Bioreactor is always operated with a large excess of substrate so  $\mu=\mu_{\max}=0.3\text{ h}^{-1}$ . The yields are  $Y'_{x/s}=0.5$  and  $Y'_{x/p}=7.1$  (negligible maintenance).

a) Calculate the concentrations of S, P and X at the exit of the system.

Calcular S

$$\begin{array}{c} S_0 = 100 \text{ g/l} \\ \downarrow 90\% \text{ conversion} \\ S = 10 \text{ g/l} \end{array}$$

Calcular X

$$Y_{x/s} = \frac{\Delta X}{\Delta S} = \frac{X - X_0}{S_0 - S}$$

$$X = 45 \text{ g/l}$$

Calcular P

$$Y_{x/p} = \frac{\Delta X}{\Delta P} = \frac{X - X_0}{P - P_0}$$

$$P = 6,34 \text{ g/l}$$

b) If you choose  $\alpha = 0.1$ , what volume is needed to achieve the desired conversion?

$$\alpha = 0,1 \qquad V = \frac{F \ln \frac{X}{X_0}}{\mu_{max}} \qquad F \rightarrow (1 + \alpha) F$$

Balance to the biomass

inlet  $\alpha X$

outlet  $X + \alpha X$

$$\frac{X_{out}}{X_{in}} = \frac{X + \alpha X}{\alpha X} = \frac{1 + \alpha}{\alpha}$$

$$V = \frac{(1 + \alpha) F \ln \frac{1 + \alpha}{\alpha}}{\mu_{max}} = 87,9 \text{ l}$$

c) If you choose  $\alpha = 2$ , what volume is needed to achieve the desired conversion?

$$V = \frac{(1 + \alpha) F \ln \frac{1 + \alpha}{\alpha}}{\mu_{\max}} = 40,5 \text{ l}$$

d) Compare and comment on the results of b) and c)

$$\alpha = 0,1 \quad V = 87,9 \text{ l}$$

$$\alpha = 2 \quad V = 40,5 \text{ l}$$

=> increasing  $\alpha \rightarrow$  reduction of the PFR volume