IEQB - Resolução de Questões

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3.10)

•
$$1 = 1i$$
 • $3 = 1o2$ • $4 = 2o1$ • $x_B = B$ • $x_X = X$

•
$$2 = 101$$
 • $3 = 2i$ • $5 = 202$ • $x_T = T$

•
$$m_{1i} = 1000 \, mol/h$$
 • $X_{1o1} = 0$ • $B_{2o1} = 0.08$

•
$$B_{1i} = 0.2$$
 • $B_{1o2} = 0.025$ • $T_{2o1} = 0.72$

•
$$T_{1i} = 0.3$$
 • $T_{1o2} = 0.35$ • $X_{2o1} = 0.2$

•
$$X_{1i} = 0.5$$
 • $X_{1o2} = 0.625$ • $B_{2o2} = 0$

$$m_{1o2} X_{1o2} = m_{1i} X_{1i} \implies m_{1o2} = \frac{1000 * 0.5}{0.625} \cong 800 \, mol/h$$

$$m_{1o1} = m_{1i} - m_{1o2} = 200 \, mol/h$$

$$m_{2o1} B_{2o1} = m_{2i} B_{2i} = m_{1o2} B_{2o1} \implies m_{2o1} = \frac{800 * 0.025}{0.08} \cong 250 \, mol/h$$

. . .

3.11)

•
$$1 = 1i1$$
 • $4 = 1o2$ • $6 = 2o2$ • $x_C = Glic$

•
$$2 = 101$$
 • $4 = 2i$ • $x_A = Agua$

•
$$3 = 1i2$$
 • $5 = 2o1$ • $x_B = \text{NaCl} \bullet x_D = \text{Alc}$

	1i1	1i2	101	1o2 / 2i	201	202
$^{\mathrm{m}}$	1000 kg/h	$1000 \mathrm{\ Kg/h}$				
Agua	$0.20 \; p/p$	$0.87 \; p/p$			$0.75 \; { m p/p}$	$\mid 0.05 \; \mathrm{p/p} \mid$
Alc	$0.98 \; {\rm p/p}$	-	$0.01 \; {\rm p/p}$		-	$0.95 \mathrm{\ p/p}$
NaCl	-	$0.03 \; {\rm p/p}$			-	-
Glic	-	$0.10 \; p/p$	$0.01~\mathrm{p/p}$		$0.25 \; p/p$	-

Produção horária de glicerina = $Glic_{2o1} m_{2o1}$; $m_{2o1} + m_{2o2} = m_{2i} = m_{1o2}$; $m_{1o2} + m_{1o1} = m_{1i1} + m_{1i2}$; $m_{2o2} Alc_{2o2} + m_{1o1} Alc_{1o1} = m_{1i1} Alc_{1i1} + m_{1i2} Alc_{1i2}$; $m_{1o1} = ?$

 m_{2i}