

## 18.303 Problem Set 3

Due Friday, 30 September 2016.

### Problem 1: Damping diffusion

When we put a strong acid, such as hydrochloric acid (HCl), into water, it does two things. First, it diffuses into the water, becoming more and more evenly distributed. Second, it *dissociates* by ionizing (releasing a proton, e.g. HCl becomes  $\text{Cl}^-$  plus a proton that attaches to nearby water molecules) and releasing heat. If the concentration of the acid is  $u(x, t)$ , the rate of dissociation is (to a first approximation, for low concentrations) proportional to  $u$ .

In this problem, you will be deriving the PDE describing this process by starting with a discrete system, applying the fact that a diffusion equation results from any microscopic picture in which the flow of solute is proportional to the gradient or difference in solute concentration (“Fick’s law”), and taking the continuum limit.

- (a) Consider the case of HCl diffusing in water within a narrow straw of length  $L$ , which we will approximate as a one-dimensional system. Imagine dividing the straw into  $M$  small pieces of length  $\Delta x$ , and call the mass of HCl in the  $m$ -th piece  $u_m \Delta x$  for  $m = 1, \dots, M$ , where  $u_m$  is a concentration (mass per  $\Delta x$ ). Suppose that the rate at which mass flows from one piece to the next is proportional to the difference in concentrations ( $u_m$ ), and is inversely proportional to the distance to the mass must travel ( $\Delta x$ ), so that the *mass per unit time* flowing from  $m + 1$  to  $m$  is given by:

$$D \frac{u_{m+1} - u_m}{\Delta x}$$

for some constant *diffusion coefficient*  $D$ . In addition to the HCl flowing out of each piece, the acid is also dissociating (the mass  $u_m \Delta x$  of acid is decreasing) at a rate  $R u_m \Delta x$  for some “rate” constant  $R > 0$ . Derive an equation for the net rate of change  $du_m/dt$  of the concentration in piece  $m$ . Take the limit  $\Delta x \rightarrow 0$  and find a PDE for  $u(x)$ , where  $u(m\Delta x) = u_m$ ; you should get a diffusion equation in the limit  $R \rightarrow 0$ . Don’t worry about the boundary conditions; just look at the interior  $1 < m < M$ .

- (b) Suppose that any HCl that reaches the ends of the straw is immediately removed (there is a little HCl-eating demon sitting at each end of the straw). What boundary condition on  $u_0$  and  $u_{M+1}$  does that imply in the discrete model? What boundary condition on  $u(x)$  in the continuum limit?
- (c) Suppose that we seal the ends of the straw, so that no HCl can enter or leave through the ends. Write down a boundary condition on  $u_0$  and  $u_{M+1}$  that reflects this situation in the discrete model, write the corresponding matrix  $A$  in the discrete system ( $d\mathbf{u}/dt = A\mathbf{u}$ ), and write the corresponding boundary condition on  $u(x)$  in the continuum limit.
- (d) For the two continuum-limit cases in your previous parts (b and c), show that the resulting PDE operator  $\hat{A}$  is Hermitian and negative-definite; what do you conclude about the solutions of  $\frac{\partial u}{\partial t} = \hat{A}u$  as  $t \rightarrow \infty$ ? What changes if you let  $R = 0$ ?
- (e) How does the PDE from part (a) change if  $D(x)$  and  $R(x)$  are (real, positive) functions of  $x$ ? In the discrete approximation, use  $D_{m+0.5} \frac{u_{m+1} - u_m}{\Delta x}$  and  $R_m u_m \Delta x$  terms, where  $D_{m+0.5} = D([m + 0.5]\Delta x)$  and  $R_m = R(m\Delta x)$ .