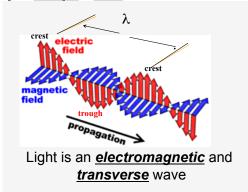
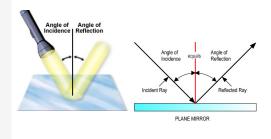
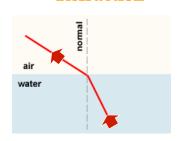
Light is an electromagnetic and transverse wave



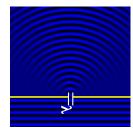


http://micro.magnet.fsu.edu/primer/lightandcolor/reflectionintro.htm

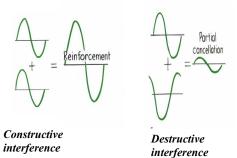
Refraction

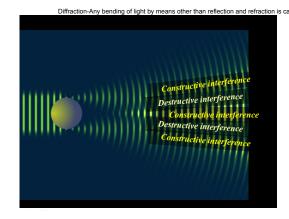


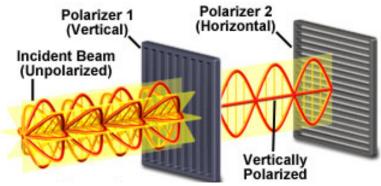


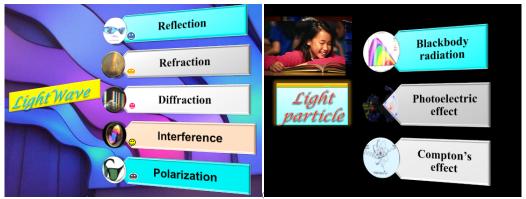


http://www.phy.ntnu.edu.tw/ntnujava/index.php? action=printpage:tonic=2314.0









Photoelectric effect is the result of the interaction between a single photon of light and a single electron. (Particle-Particle interaction)

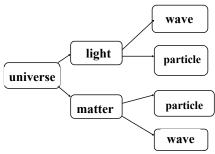
The photon is completely absorbed and its energy is transferred entirely to the electron.

Einstein's Explanation (Nobel Prize in 1921)

- 1. Light energy comes in DISCRETE packets (photons) and not continuously as waves.
- 2. Energy comes all at once in a bundle.

 $E = nhf \quad n = 1,2,3,...$

3. Energy is localized in a small volume of space.



Wave-particle duality

De Broglie Hypothesis

- A moving <u>particle</u> behaves in certain ways as though it has a <u>Wave</u> nature
- · Wavelength of a moving particle

$$\lambda = \frac{h}{mv}$$

 $h = Planck's constant = 6.63 \times 10^{-34} J-s$

Diffraction patterns





by X-RAYS

by ELECTRONS