

Primer on Semiconductors

Unit 1: Material Properties

Lecture 1.6: Doping

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Doping makes semiconductors useful

metal



gold (Au)

semiconductor



silicon (Si)

insulator



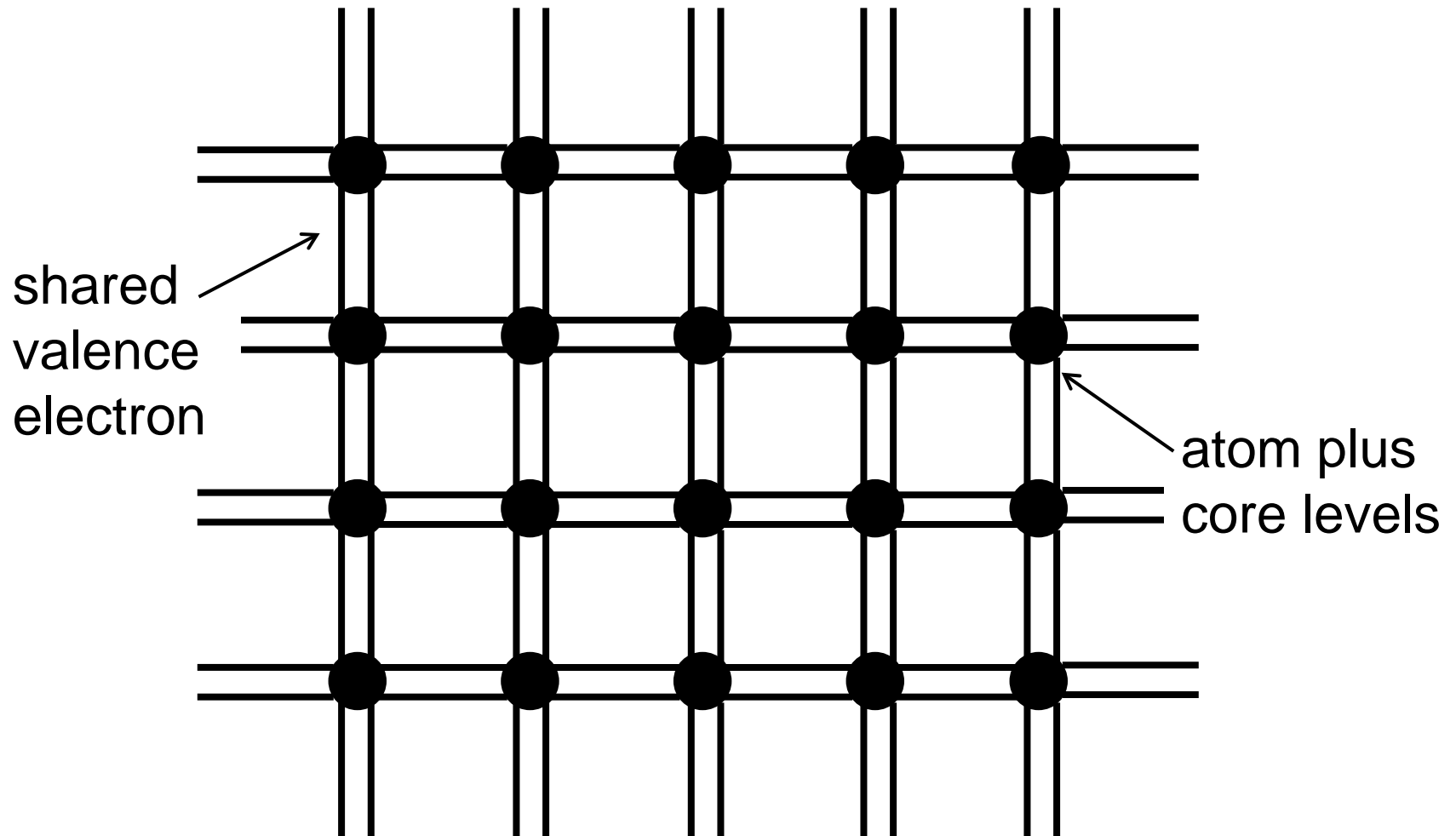
glass (SiO_2)

Things have changed

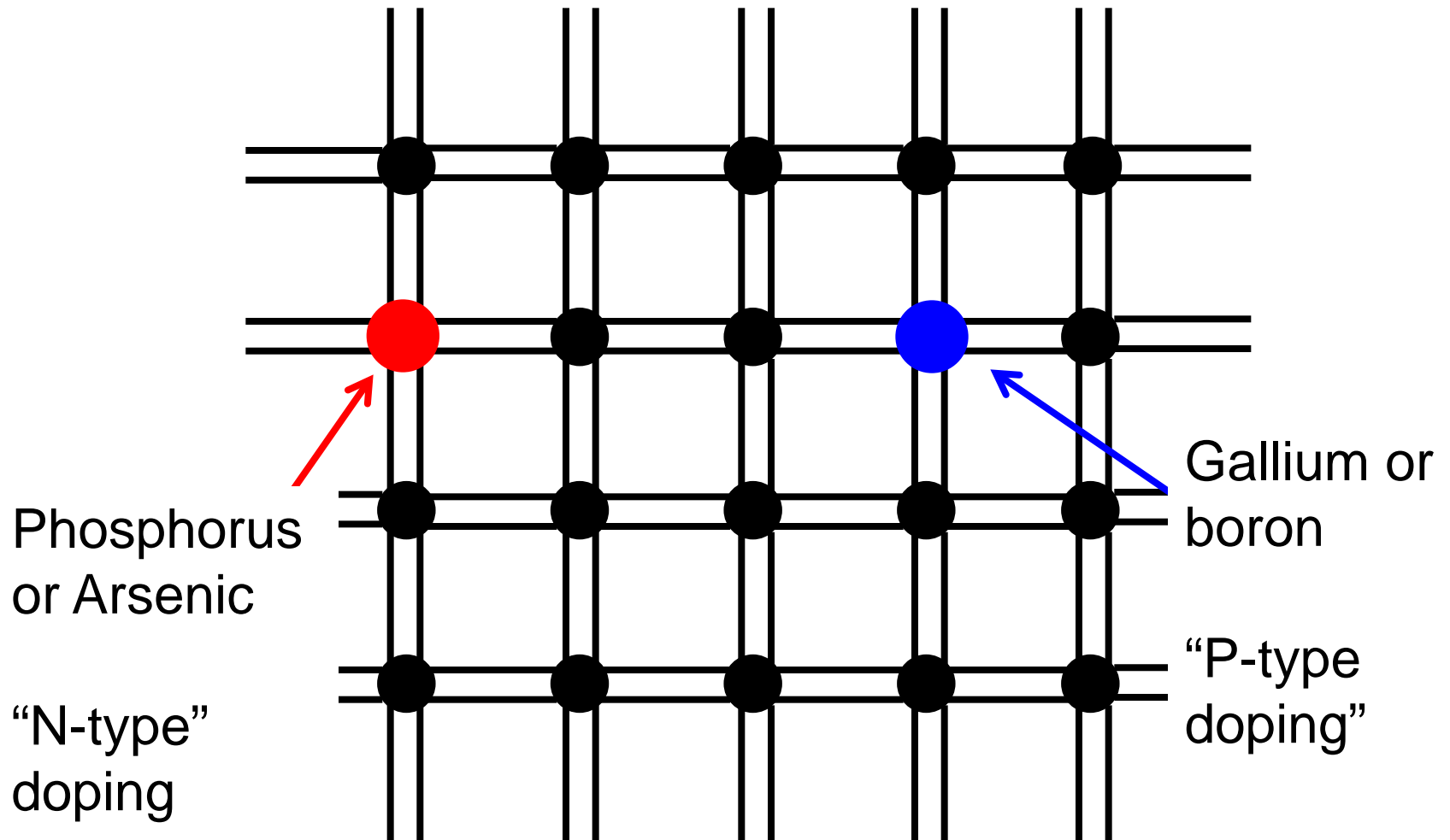
“One shouldn’t work on semiconductors, that is a filthy mess; who knows whether any semiconductors exist.”

Wolfgang Pauli, 1931

Recall: Bonding cartoon of Si lattice



Doping a semiconductor



Dopants in Si

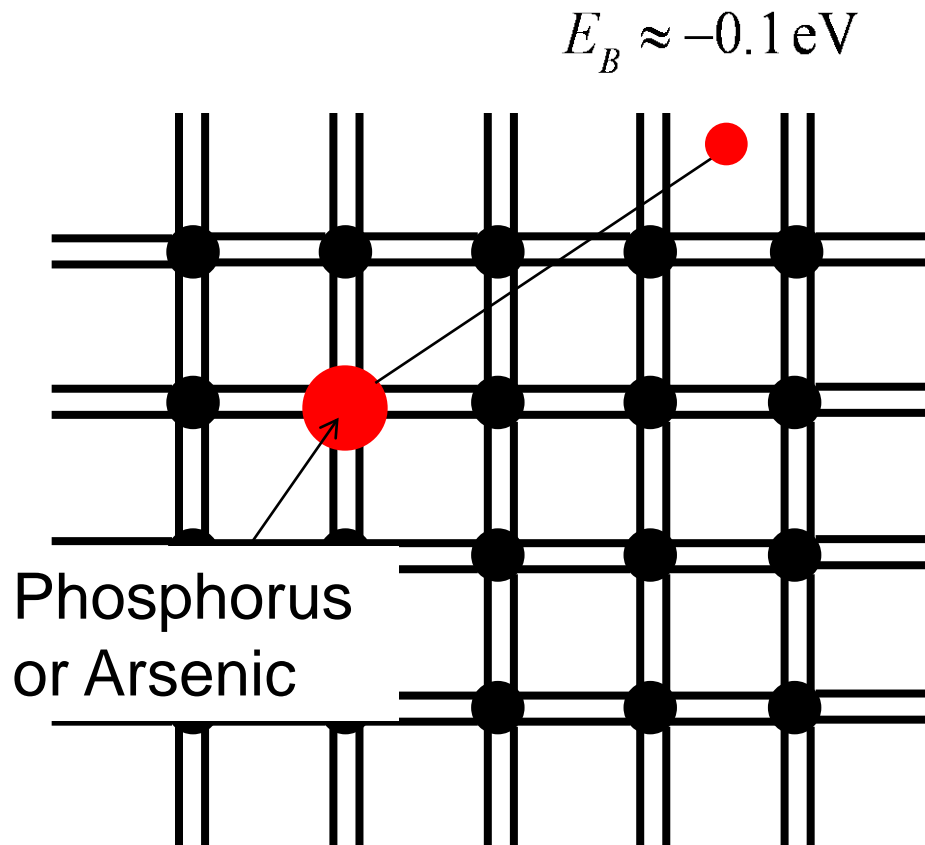
column IV

P-type dopants come from column III

N-type dopants come from column V

Group #	1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18
Period 1	1 H																	2 He
Period 2	3 Li	4 Be											5 B	6 C	7 N	8 O	9 F	10 Ne
Period 3	11 Na	12 Mg											13 Al	14 Si	15 P	16 S	17 Cl	18 Ar
Period 4	19 K									28 Ni	29 Cu	30 Zn	31 Ga	32 Ge	33 As	34 Se	35 Br	36 Kr
Period 5	37 Rb									46 Pd	47 Ag	48 Cd	49 In	50 Sn	51 Sb	52 Te	53 I	54 Xe
Period 6	55 Cs									78 Pt	79 Au	80 Hg	81 Tl	82 Pb	83 Bi	84 Po	85 At	86 Rn
Period 7	87 Fr	88 Ra	**	104 Rf	105 Db	106 Sg	107 Bh	108 Hs	109 Mt	110 Ds	111 Rg	112 Cn	113 Nh	114 Fl	115 Mc	116 Lv	117 Ts	118 Og
* Lanthanoids				57 La	58 Ce	59 Pr	60 Nd	61 Pm	62 Sm	63 Eu	64 Gd	65 Tb	66 Dy	67 Ho	68 Er	69 Tm	70 Yb	71 Lu
** Actinoids				89 Ac	90 Th	91 Pa	92 U	93 Np	94 Pu	95 Am	96 Cm	97 Bk	98 Cf	99 Es	100 Fm	101 Md	102 No	103 Lr

N-type doping

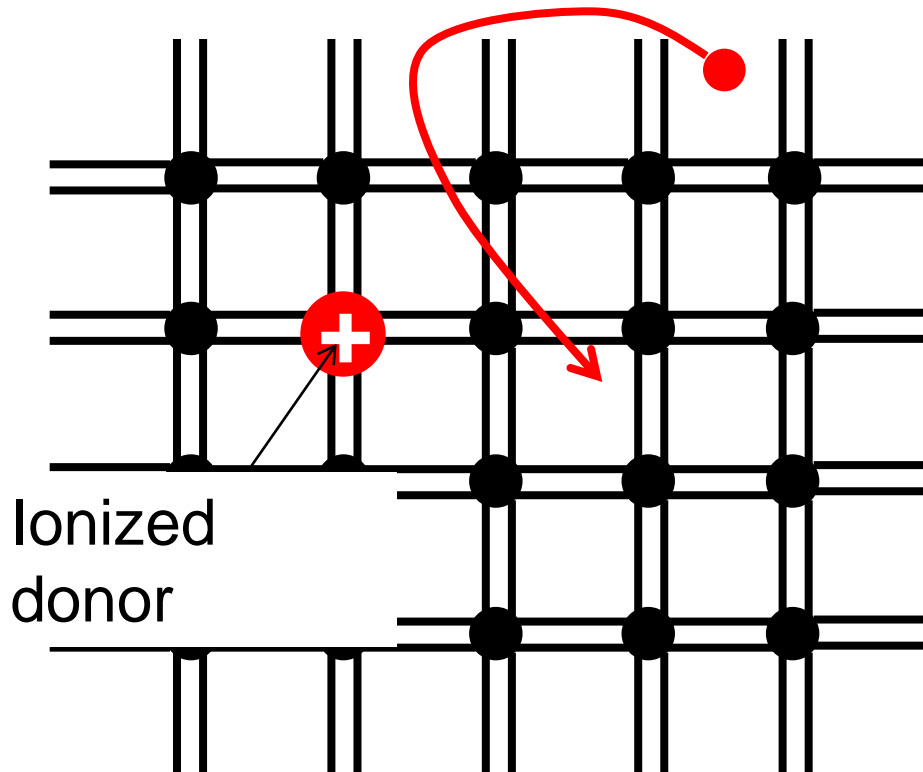


Weakly bound

Easily broken at room temperature

Produces an electron in the conduction band.

Ionized donor



Concentration of dopants:

$$N_D \text{ cm}^{-3}$$

Concentration of ionized donors:

$$N_D^+ \text{ cm}^{-3}$$

Concentration of electrons in the conduction band:

$$n \approx N_D^+ \text{ cm}^{-3}$$

Be careful about units!

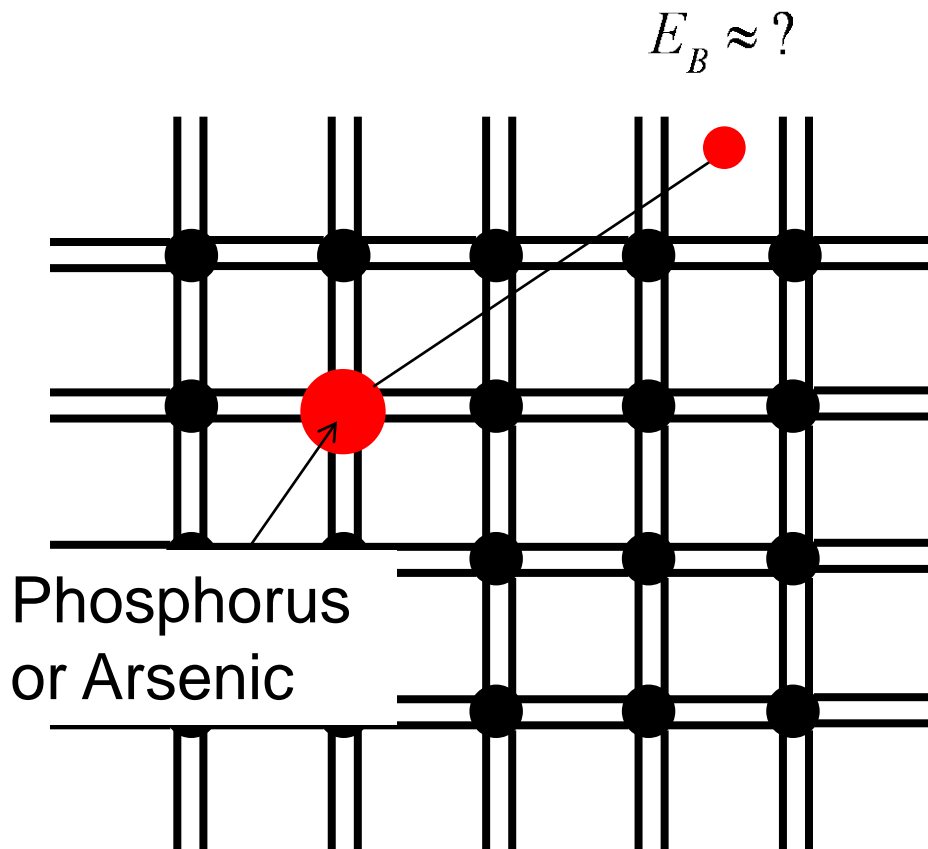
$$n \approx N_D^+ \text{ cm}^{-3}$$

We will be working in SI (MKS) units. The carrier concentration should be given per cubic **meter**, but semiconductor people like to mix their units.

It is safest to do the calculations in SI units, and then convert to cubic cm.

$$n = 10^{26} \text{ m}^{-3} \rightarrow n = 10^{26} \frac{1}{\text{m}^3} \times \left(\frac{10^{-2} \text{ m}}{\text{cm}} \right)^3 \rightarrow n = 10^{20} \text{ cm}^{-3}$$

Binding energy of the donor



Hydrogen atom:

$$E_B = -\frac{m_0 q^4}{2(4\pi\epsilon_0 \hbar)^2} \text{ eV}$$

$$E_B = -13.6 \text{ eV}$$

Donor in Si:

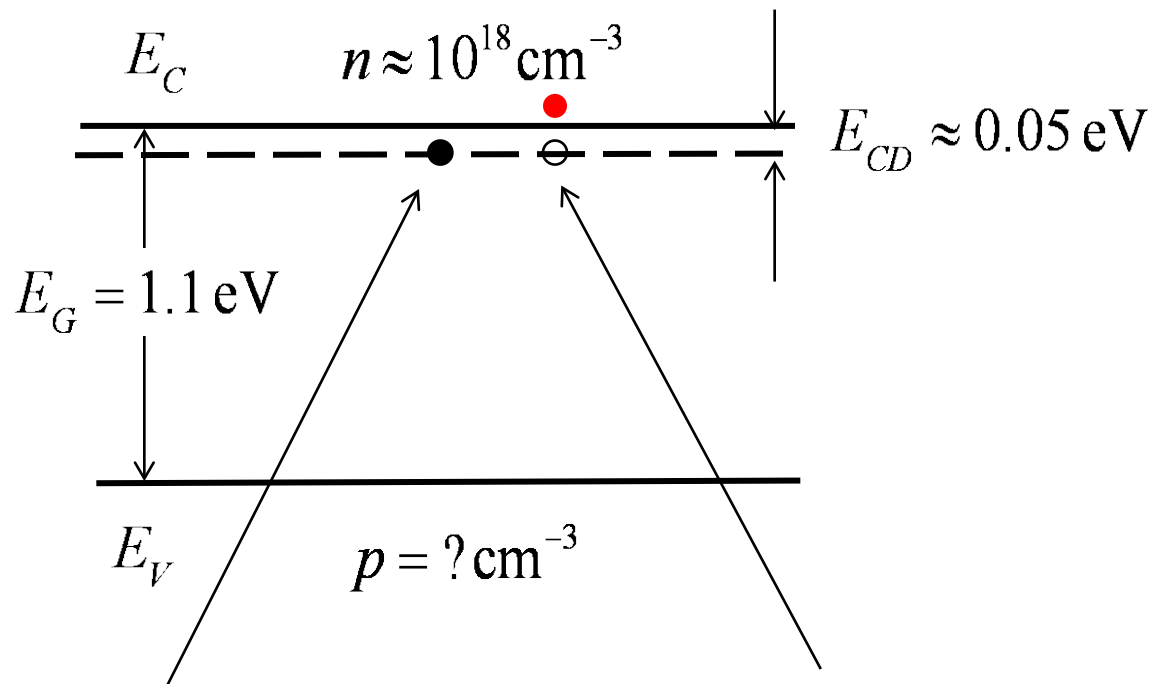
$$\begin{aligned} \epsilon &= K_S \epsilon_0 \\ &= 11.8 \times (8.854 \times 10^{-12} \text{ F/m}) \end{aligned}$$

$$m_n^* = 1.18 m_0$$

$$E_B \approx -0.1 \text{ eV}$$

Energy band view (n-type)

n-doped Si



neutral donor

ionized donor

Lundstrom: 2018

$$N_D = 10^{18} \text{ cm}^{-3}$$

$$N_D^+ = 10^{18} \text{ cm}^{-3}$$

$$n \approx N_D^+ = 10^{18} \text{ cm}^{-3}$$

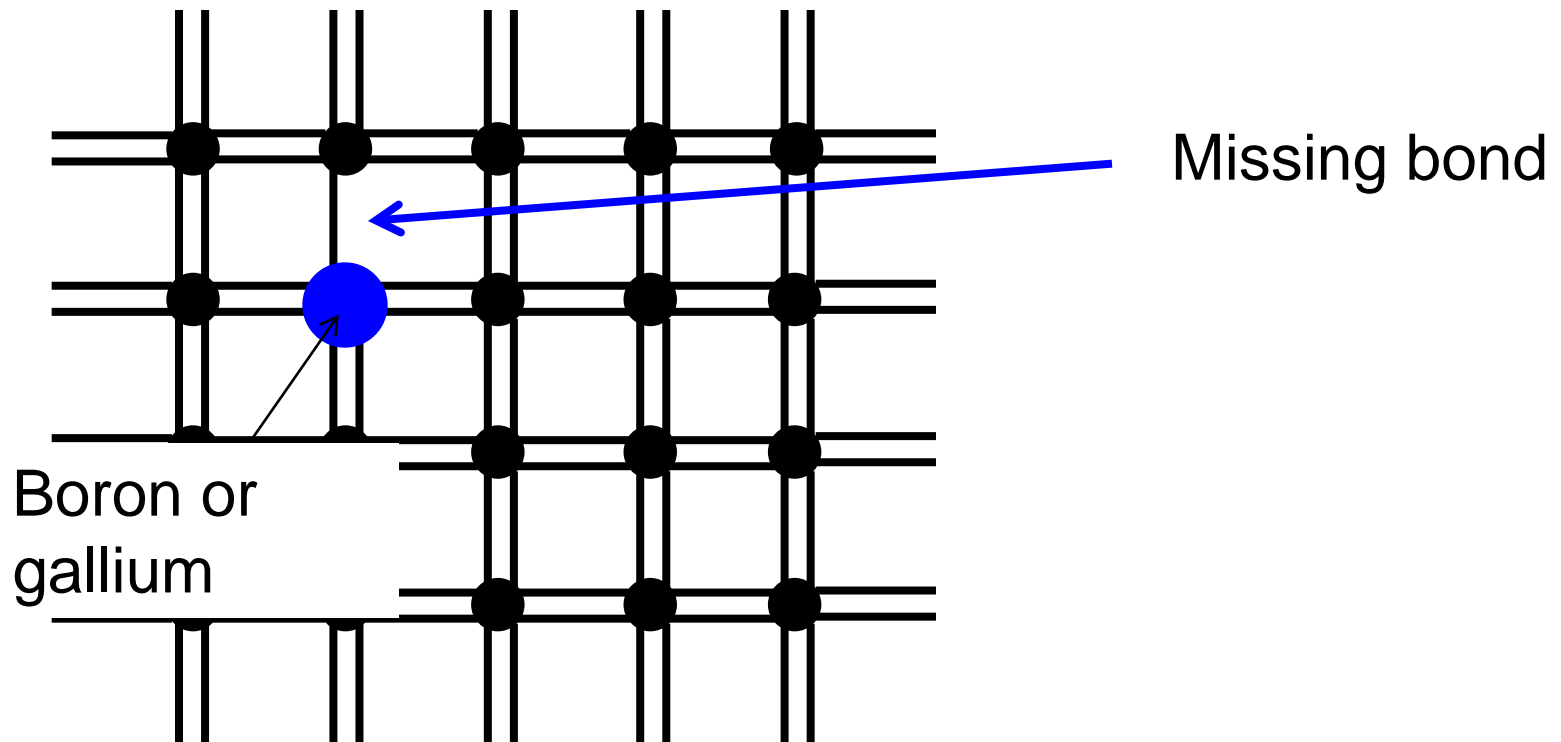
$$p \approx ?$$

$$(T = 300 \text{ K})$$

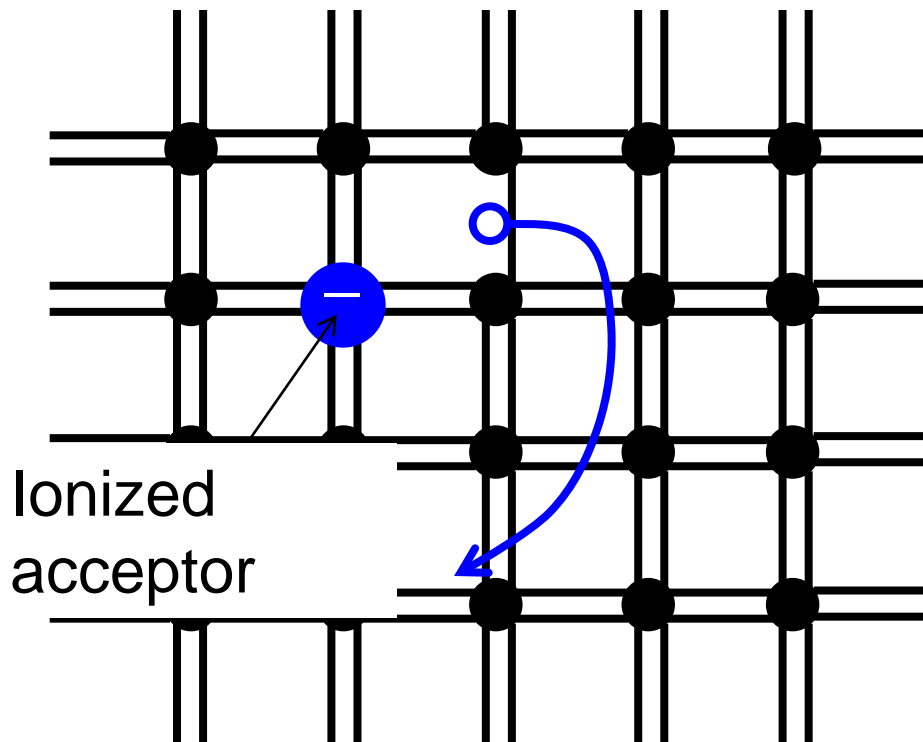
$$10^{14} \leq N_D \leq 10^{20} \text{ cm}^{-3}$$

$$N_{atoms} \approx 5 \times 10^{22} \text{ cm}^{-3}$$

P-type doping



Ionized acceptor



Concentration of dopants:

$$N_A \text{ cm}^{-3}$$

Concentration of ionized acceptors:

$$N_A^- \text{ cm}^{-3}$$

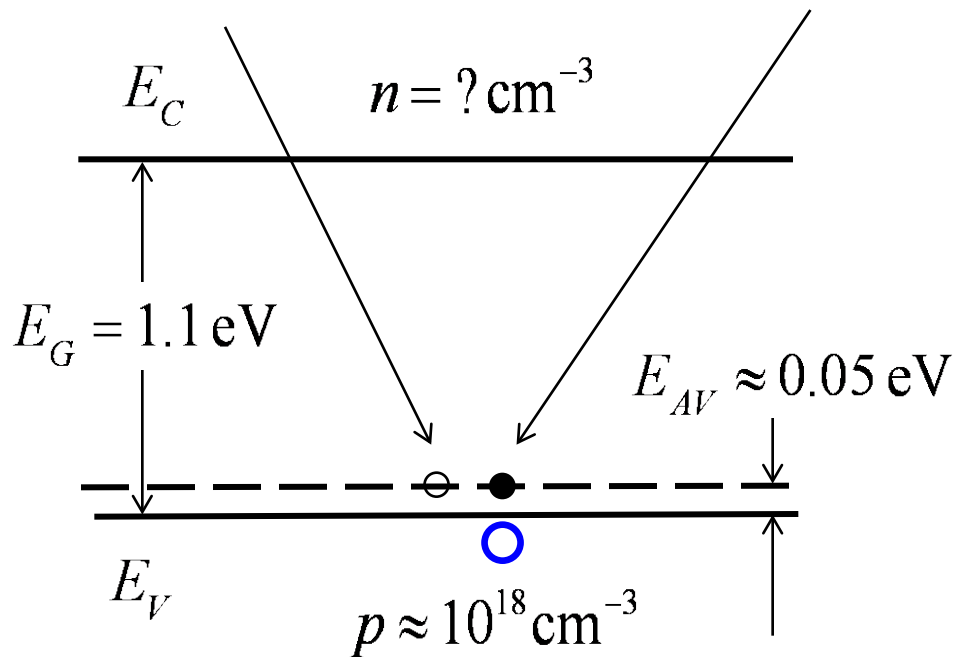
Concentration of “holes” in the valence band:

$$p \approx N_A^- \text{ cm}^{-3}$$

Energy band view (p-type)

p-doped Si

neutral acceptor ionized acceptor



$$N_A = 10^{18} \text{ cm}^{-3}$$

$$N_A^- = 10^{18} \text{ cm}^{-3}$$

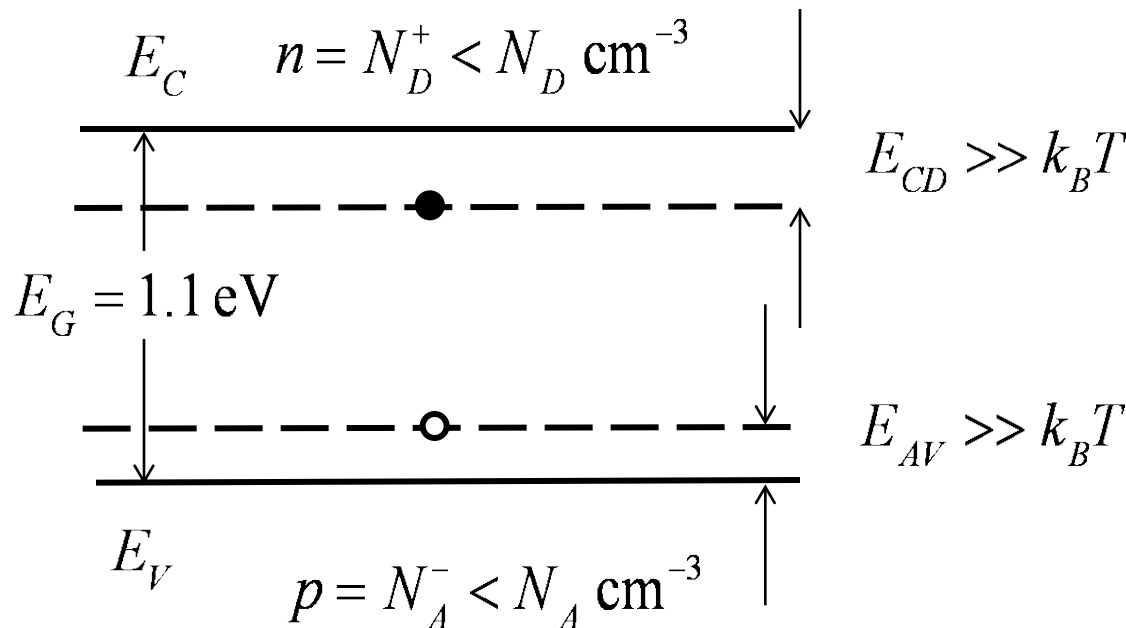
$$p \approx N_A^- = 10^{18} \text{ cm}^{-3}$$

$$n \approx ?$$

$$(T = 300 \text{ K})$$

$$10^{14} \leq N_A \leq 10^{20} \text{ cm}^{-3}$$

“Deep donors” and “deep acceptors”



If the donor or acceptor level is much larger than kT (at room temperature), we say that the level is “deep”).

Donors and acceptors will be partially ionized at room temperature.

“Shallow” donors and acceptors are fully ionized at room temperature.

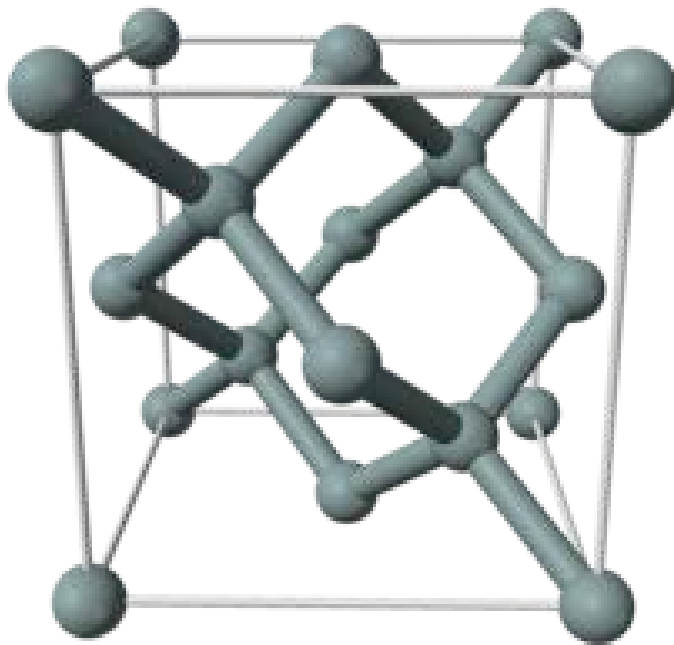
Question

1) What type of dopant is Si in GaAs?

- a) n-type
- b) p-type
- c) either n-type or p-type
- d) neither n-type nor p-type
- e) don't know

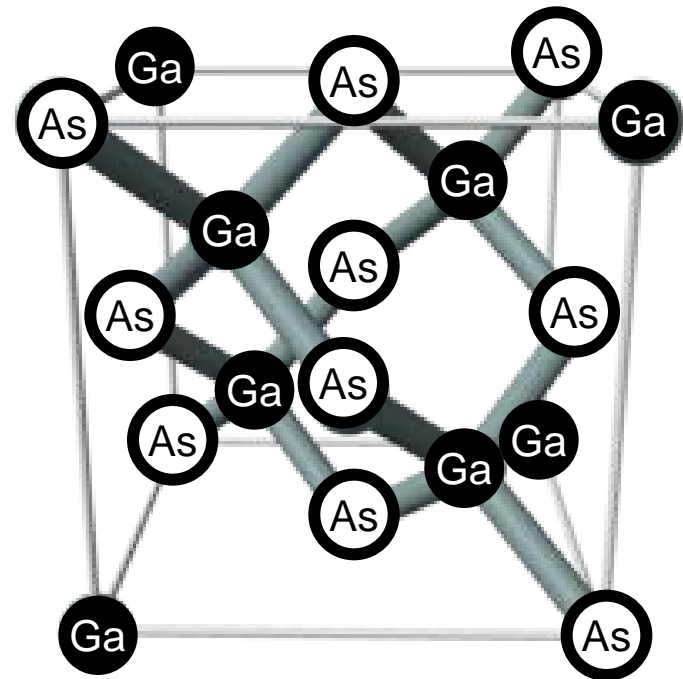
Si vs. GaAs crystals

Si



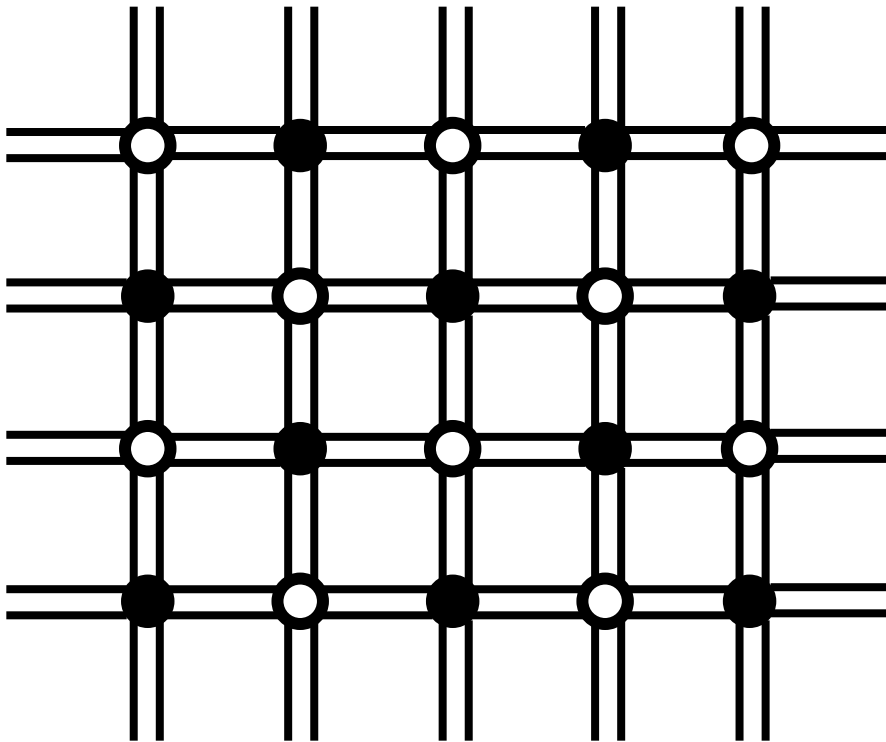
diamond lattice

GaAs



zinc blende lattice

GaAs bonding cartoon



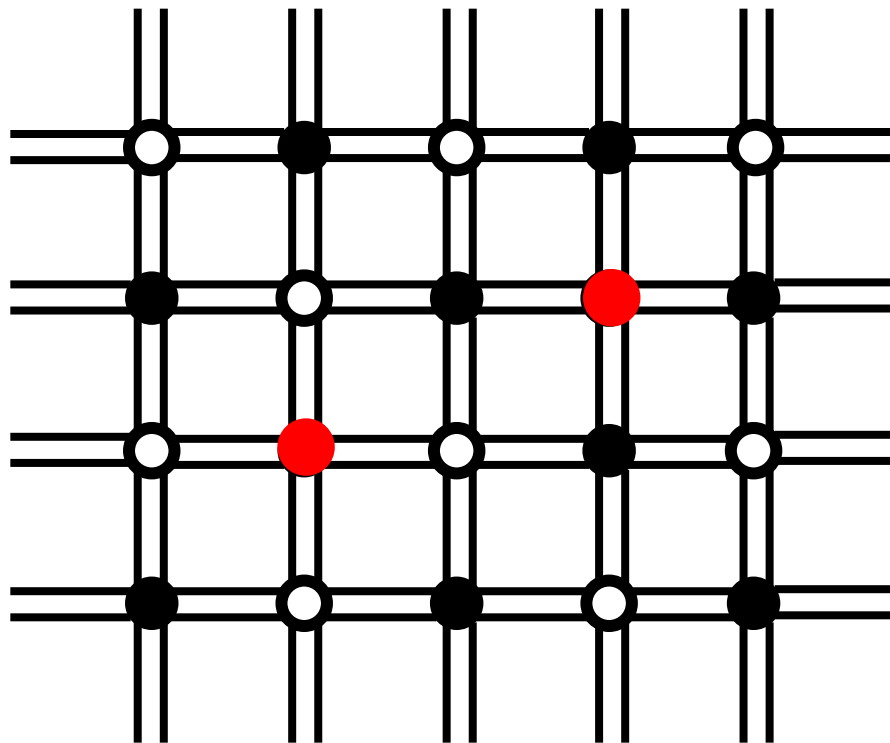
● Ga

○ As

Every Ga atom has
4 NN As atoms.

Every As atom has
4 NN Ga atoms.

Si in GaAs: Two possibilities



- Ga (III)
- As (V)
- Si (IV)

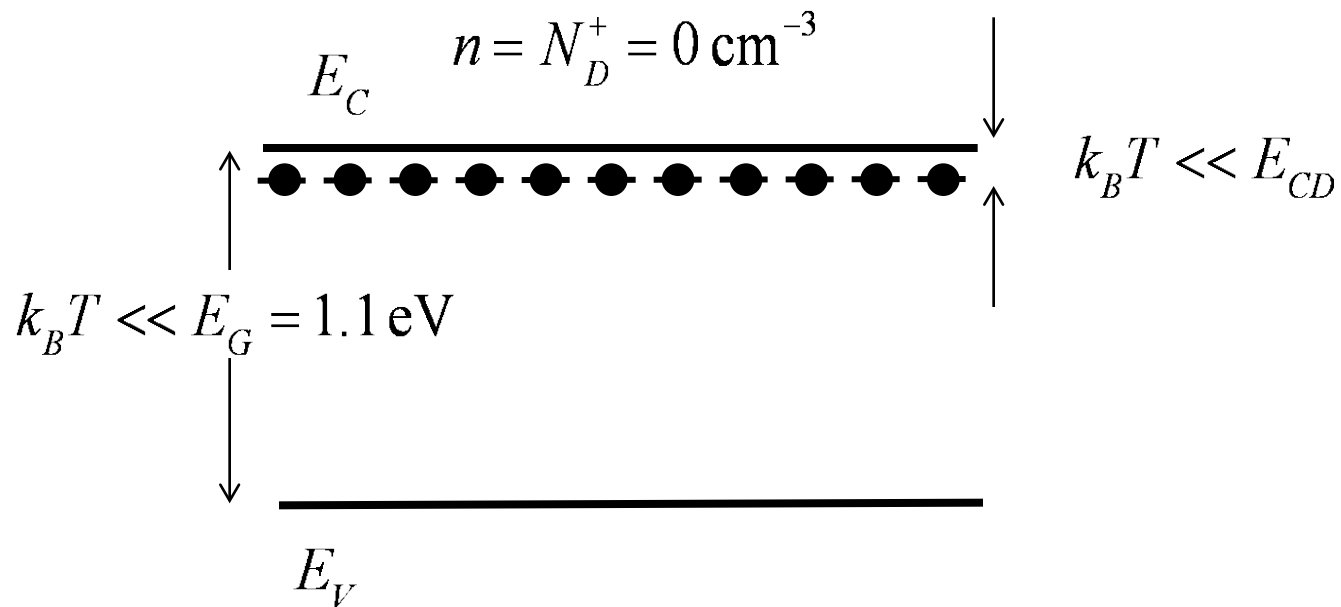
Si on a As site is an acceptor.

Si is an “amphoteric” dopant in GaAs.

Si on a Ga site is a donor.

How does the carrier concentration vary with T?

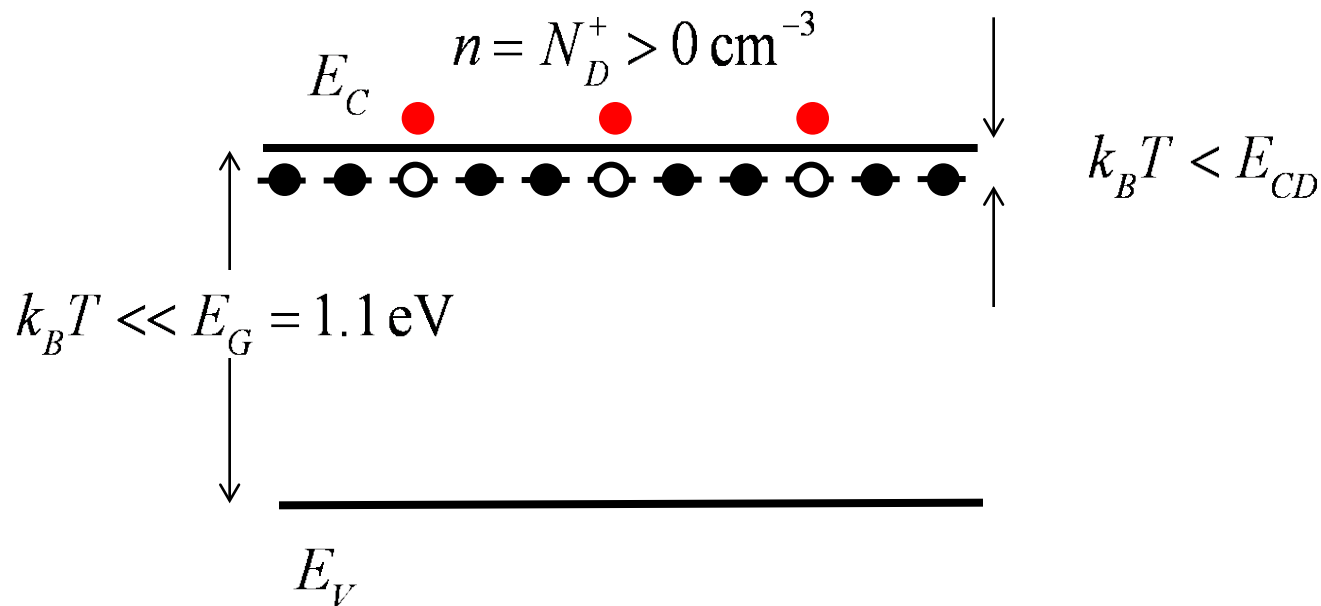
T = 0 K



No extrinsic carriers due to doping, and no intrinsic carriers due to thermal excitation across the band gap.

Low temperature

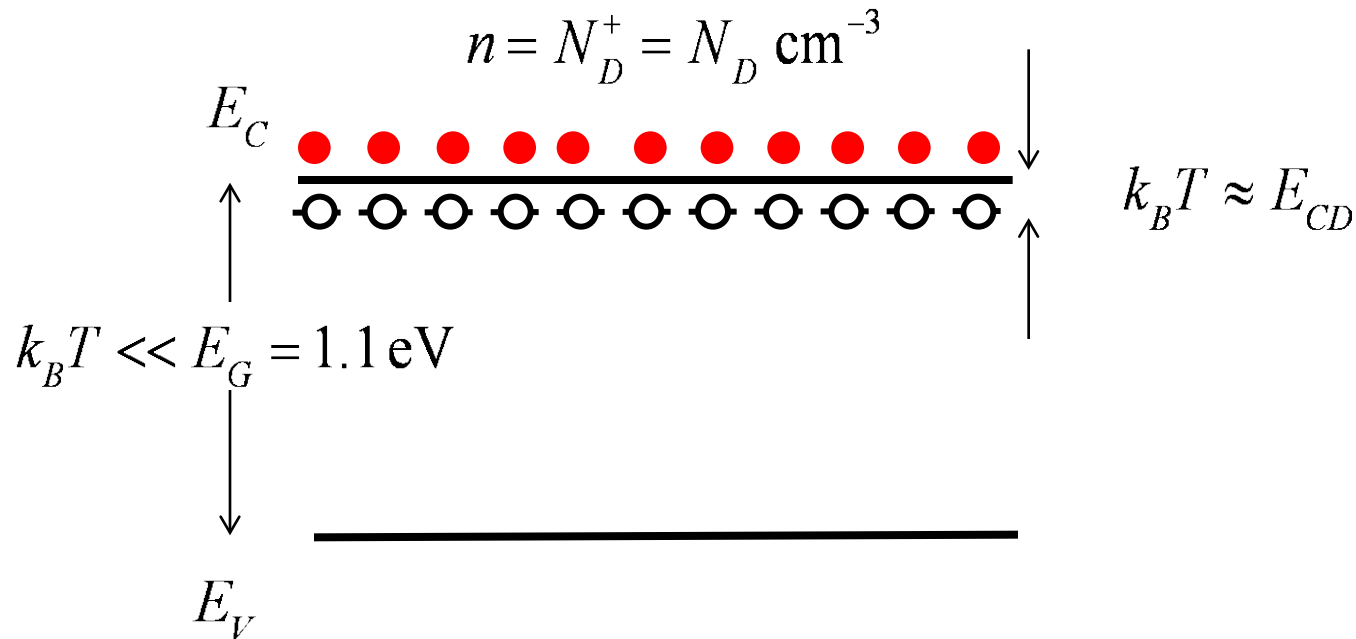
$T > 0 \text{ K}$



A few extrinsic carriers due to doping, but no intrinsic carriers due to thermal excitation across the band gap.

Room temperature (shallow dopants)

$T = 300 \text{ K}$

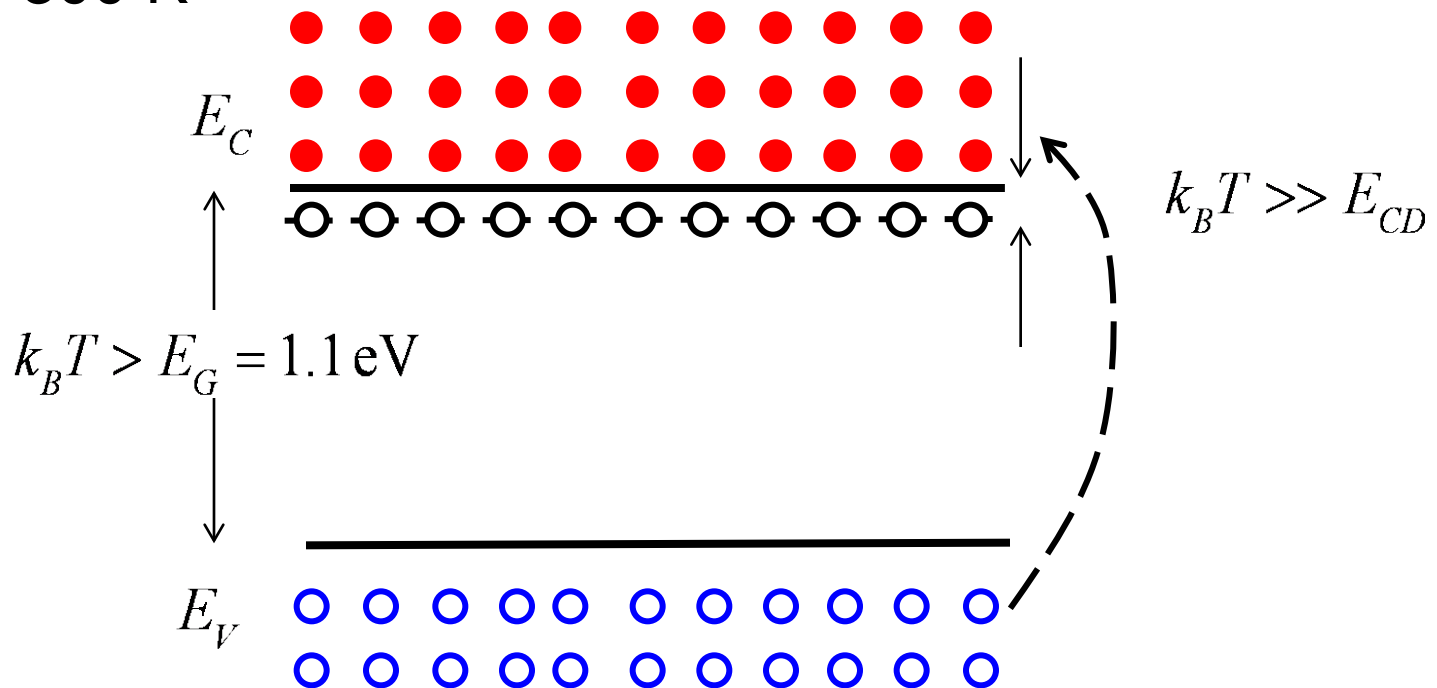


Many extrinsic carriers due to doping, but very few intrinsic carriers due to thermal excitation across the band gap.

High temperature

$T \gg 300 \text{ K}$

$$n = N_D^+ = N_D \text{ cm}^{-3}$$



Extrinsic carriers due to doping, but also many intrinsic carriers due to thermal excitation across the band gap. Also many holes in the valence band now.

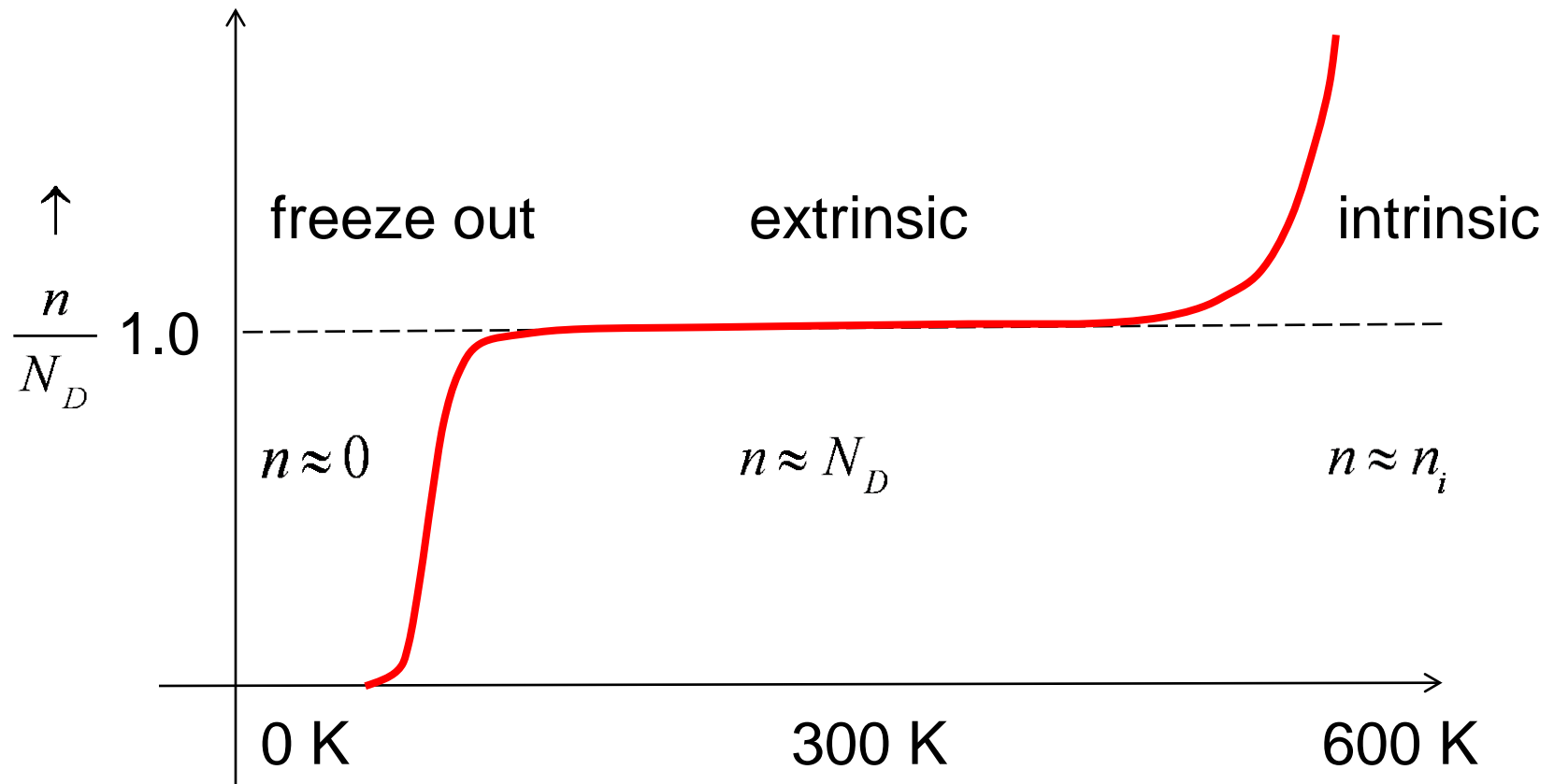
Comment

At high temperatures, the intrinsic carriers produced by thermal excitation across the bandgap overwhelm the extrinsic carriers produced by doping.

Because semiconductor devices are produced by selective doping, the device ceases to operate.

Semiconductor devices for operation at high temperature should be made with materials with large band gaps, such as SiC and GaN.

Carrier concentration vs. temperature



Summary

To dope a semiconductor, we replace a few atoms with atoms from a different column of the periodic table.

Ionized dopants produce electrons in the conduction band or holes in the valence band.

Dopants can be energetically shallow or deep.

The carrier concentration vs. temperature characteristic has freeze out, extrinsic, and intrinsic regions.

A low temperatures, semiconductors become insulators.