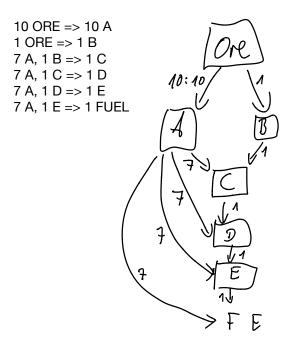
Each reaction gives specific quantities for its inputs and output; reactions cannot be partially run, so only whole integer multiples of these quantities can be used. (It's okay to have leftover chemicals when you're done, though.) For example, the reaction 1 A, 2 B, 3 C => 2 D means that exactly 2 units of chemical D can be produced by consuming exactly 1 A, 2 B and 3 C. You can run the full reaction as many times as necessary; for example, you could produce 10 D by consuming 5 A, 10 B, and 15 C.

Suppose your nanofactory produces the following list of reactions:

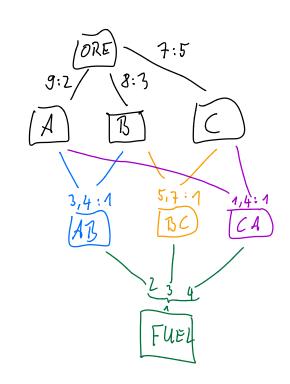


The first two reactions use only ORE as inputs; they indicate that you can produce as much of chemical A as you want (in increments of 10 units, each 10 costing 10 ORE) and as much of chemical B as you want (each costing 1 ORE). To produce 1 FUEL, a total of 31 ORE is required: 1 ORE to produce 1 B, then 30 more ORE to produce the 7 + 7 + 7 + 7 = 28 A (with 2 extra A wasted) required in the reactions to convert the B into C, C into D, D into E, and finally E into FUEL. (30 A is produced because its reaction requires that it is created in increments of 10.)

9 ORE => 2 A 8 ORE => 3 B 7 ORE => 5 C 3 A, 4 B => 1 AB 5 B, 7 C => 1 BC 4 C, 1 A => 1 CA 2 AB, 3 BC, 4 CA => 1 FUEL

The above list of reactions requires 165 ORE to produce 1 FUEL:

Consume 45 ORE to produce 10 A.
Consume 64 ORE to produce 24 B.
Consume 56 ORE to produce 40 C.
Consume 6 A, 8 B to produce 2 AB.
Consume 15 B, 21 C to produce 3 BC.
Consume 16 C, 4 A to produce 4 CA.
Consume 2 AB, 3 BC, 4 CA to produce 1 FUEL.



$$2AB = 2(3A + 4B) = 6A + BB$$

$$3BC = 3(5B + 7C) = 15B + 21C$$

$$4CA = 4(4C + 1A) = 16C + 4A$$

$$10A = 10(\frac{9 \text{ ORE}}{2A}) = 5.9 \text{ ORE} = 45 \text{ ORE}$$

$$23B = 23(\frac{8 \text{ ORE}}{3B}) = \frac{23}{3} = > \frac{24}{3}.8 \text{ ORE} = 64 \text{ ORE}$$

$$3FC = 3F(\frac{9 \text{ ORE}}{5C}) = \frac{37}{5} = > \frac{40}{5}.7 \text{ ORE} = 56 \text{ ORE}$$

$$45$$

$$64$$