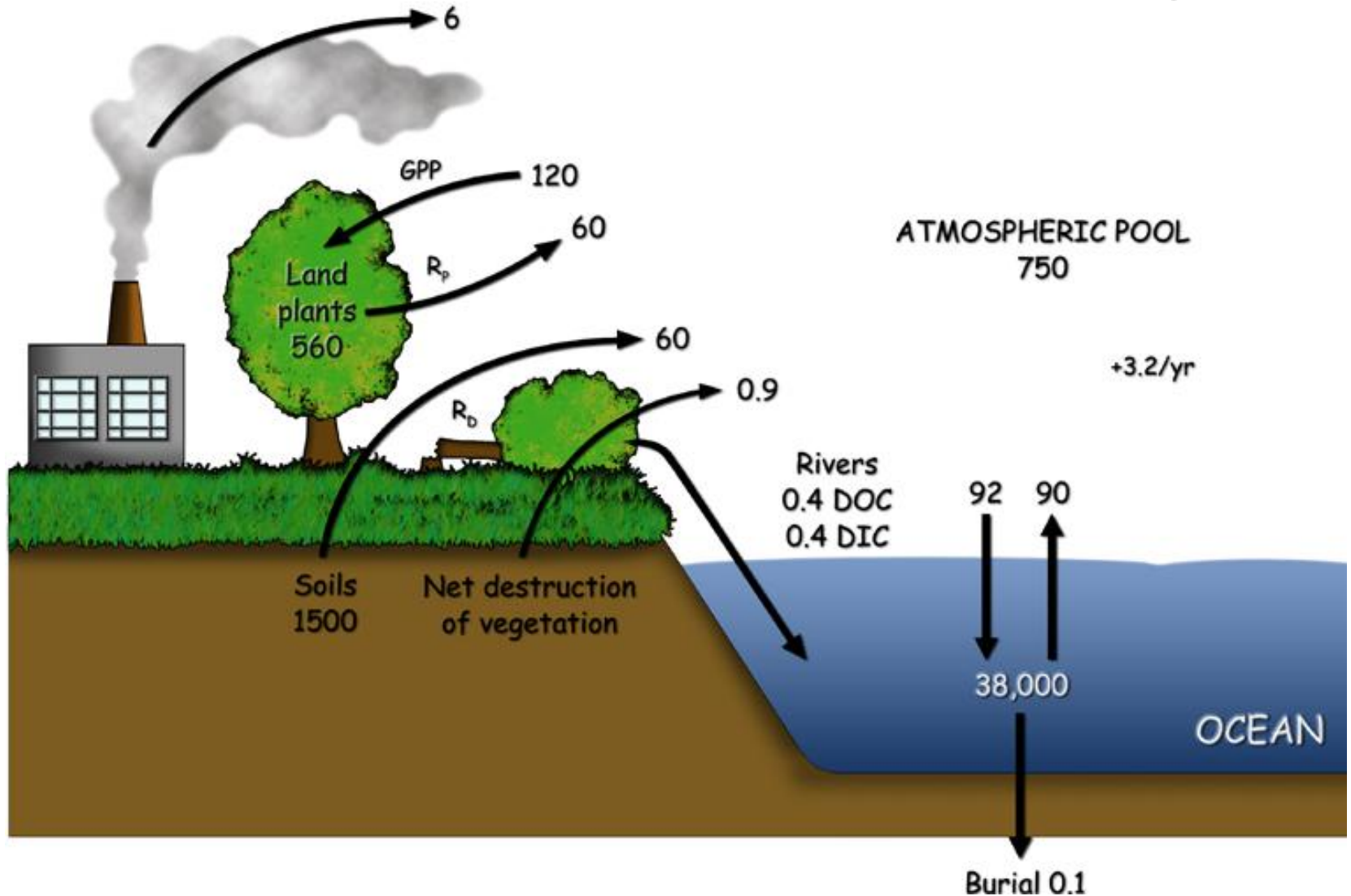




Theory of Leaf-Level Gas Exchange Measurements

Pat Morgan and Jason Hupp
Senior Application Scientists
LI-COR Biosciences
Lincoln, Nebraska, USA

Global Carbon cycle ($\times 10^{15}$ gC)

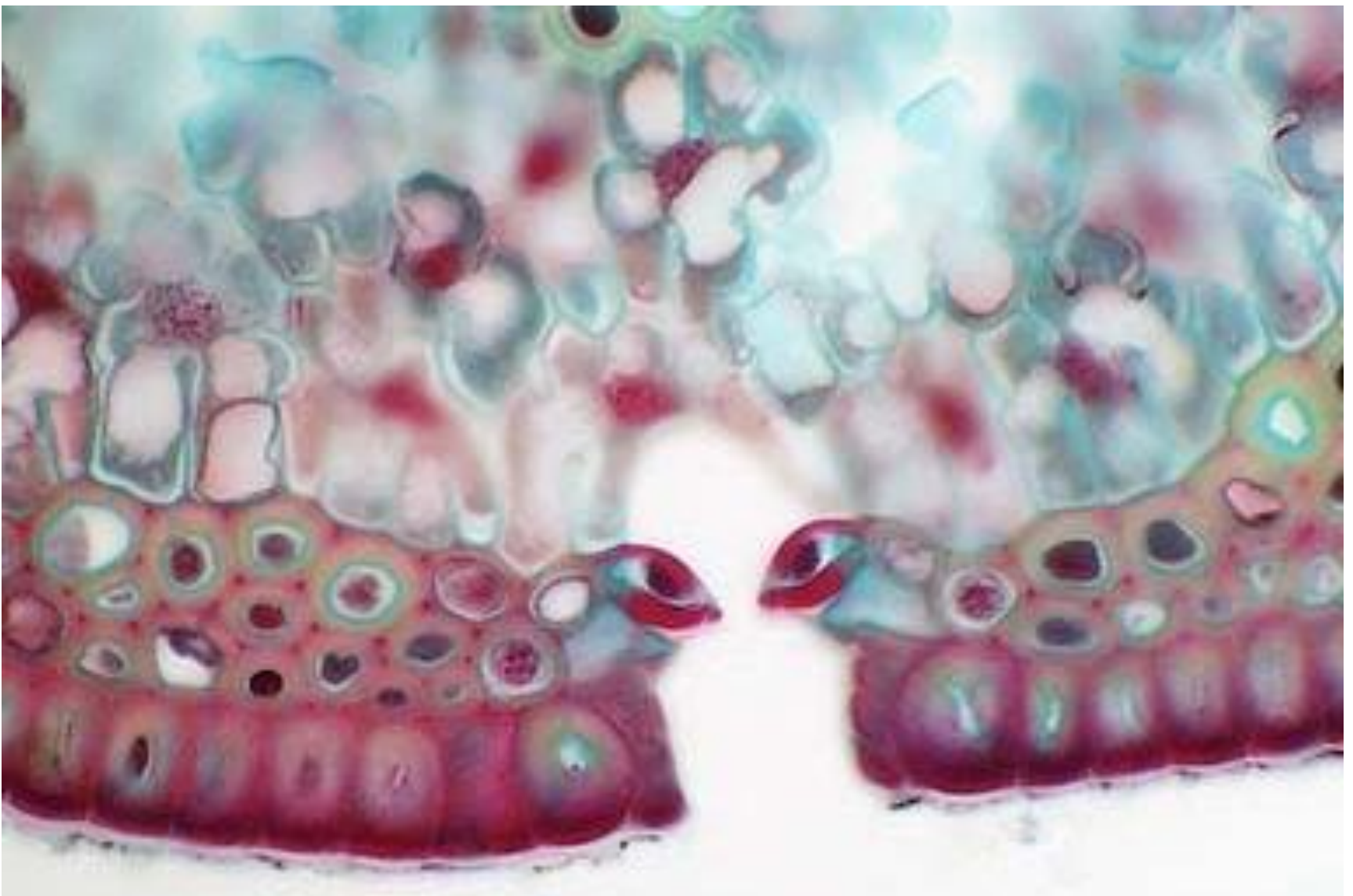


How is Vegetation Measured?



Plant and Atmosphere Interface





Stomata regulate H_2O loss from and CO_2 entry into leaf

Ohm's Law

$$\Delta E = IR$$

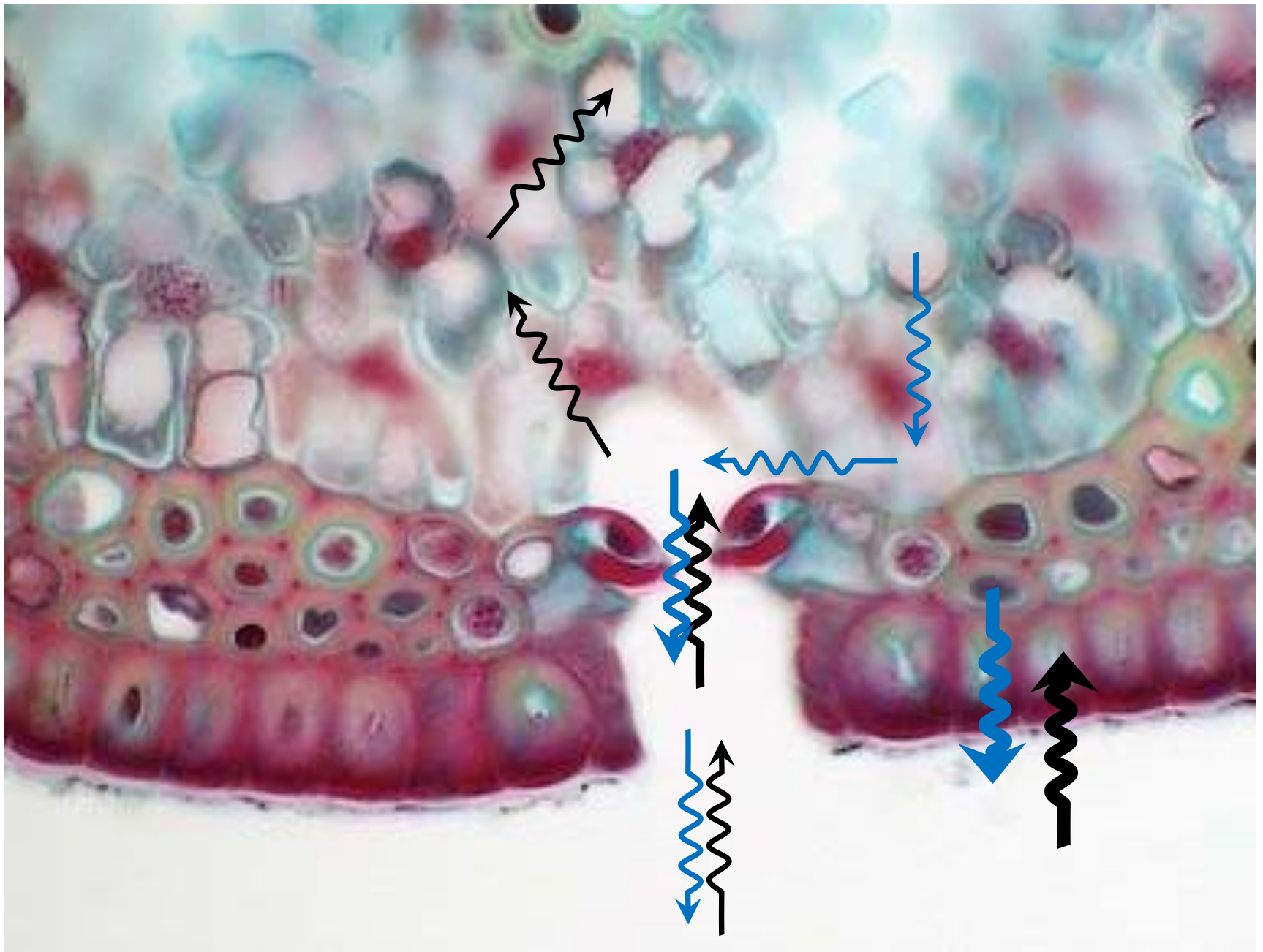
$$I = \Delta E / R$$

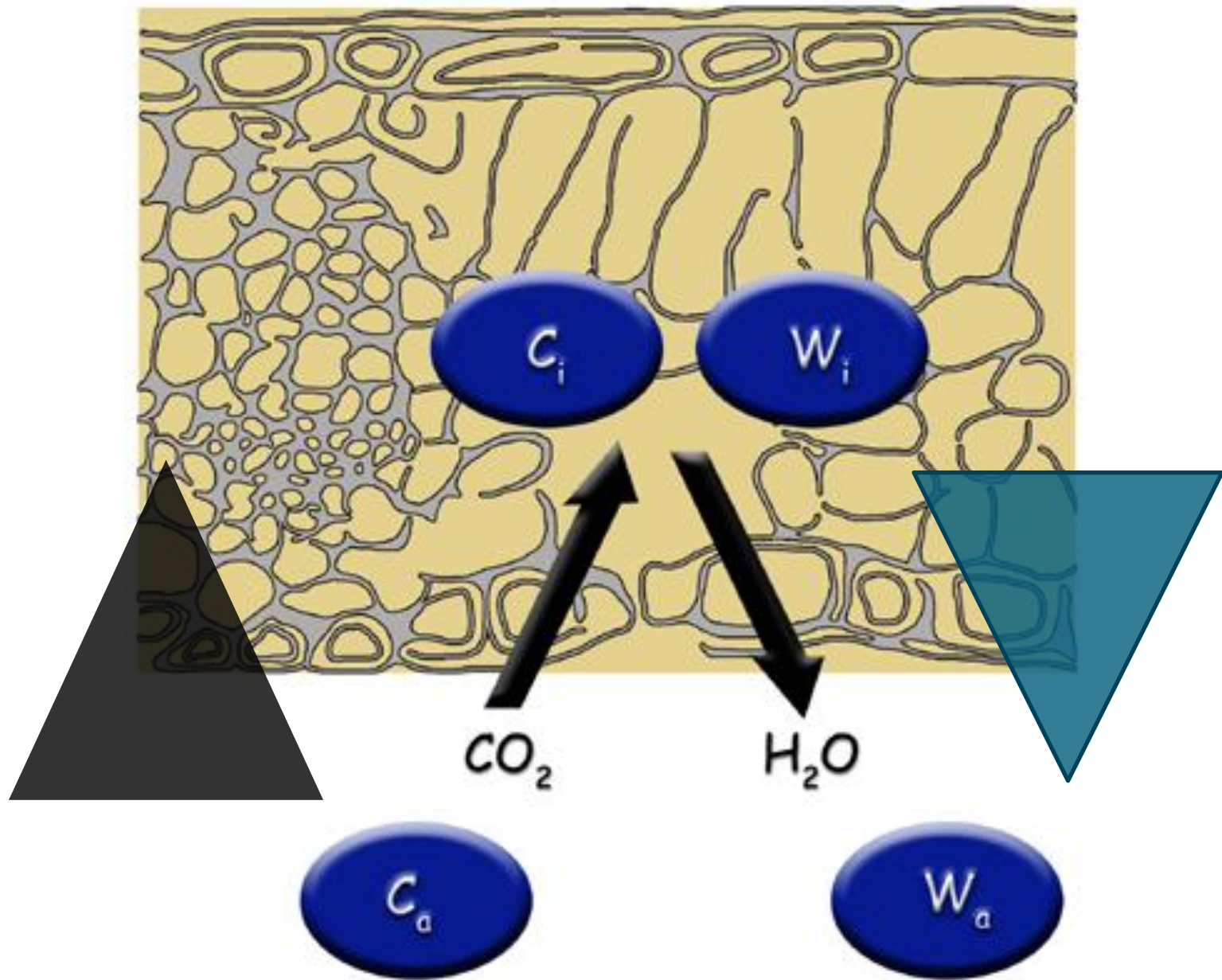
ΔE = driving force
(electric potential difference)

R = resistance

conductance = $1/R$

I = current





Fick's First Law

$$J_j = -D_j \frac{\partial c_j}{\partial x}$$
$$= g_j^{\text{bl}} \Delta c_j^{\text{bl}}$$

Δc_j = concentration gradient

g_j = conductance

J_j = flux

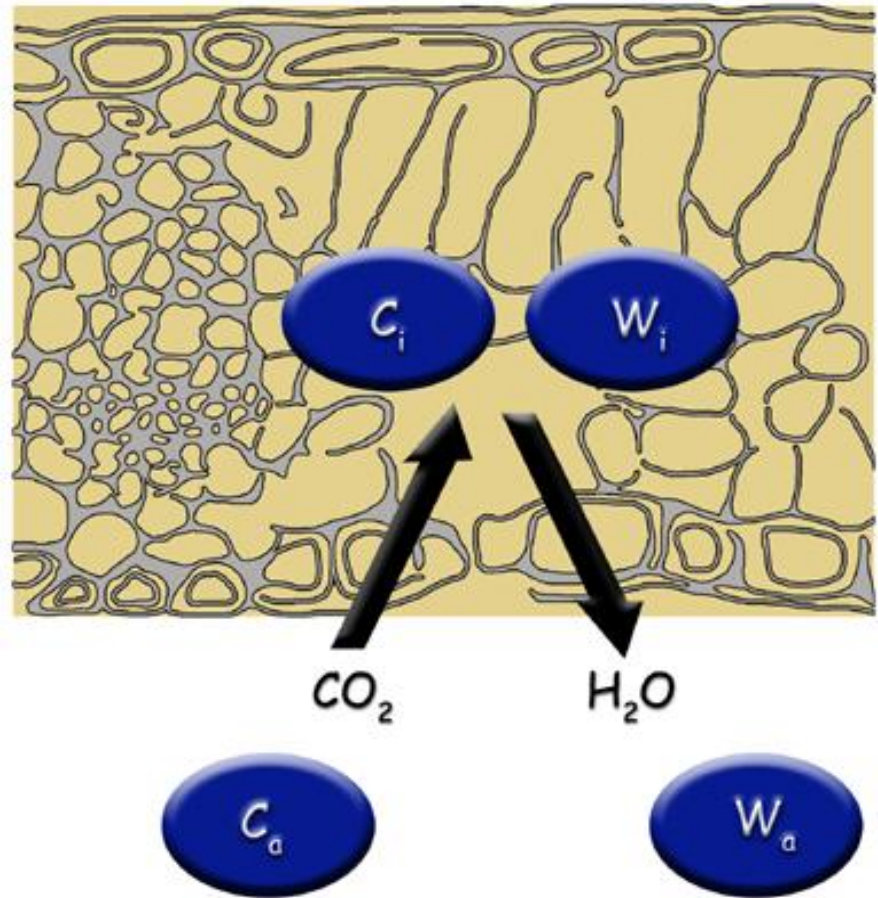
H₂O and CO₂ Flux

- Photosynthesis

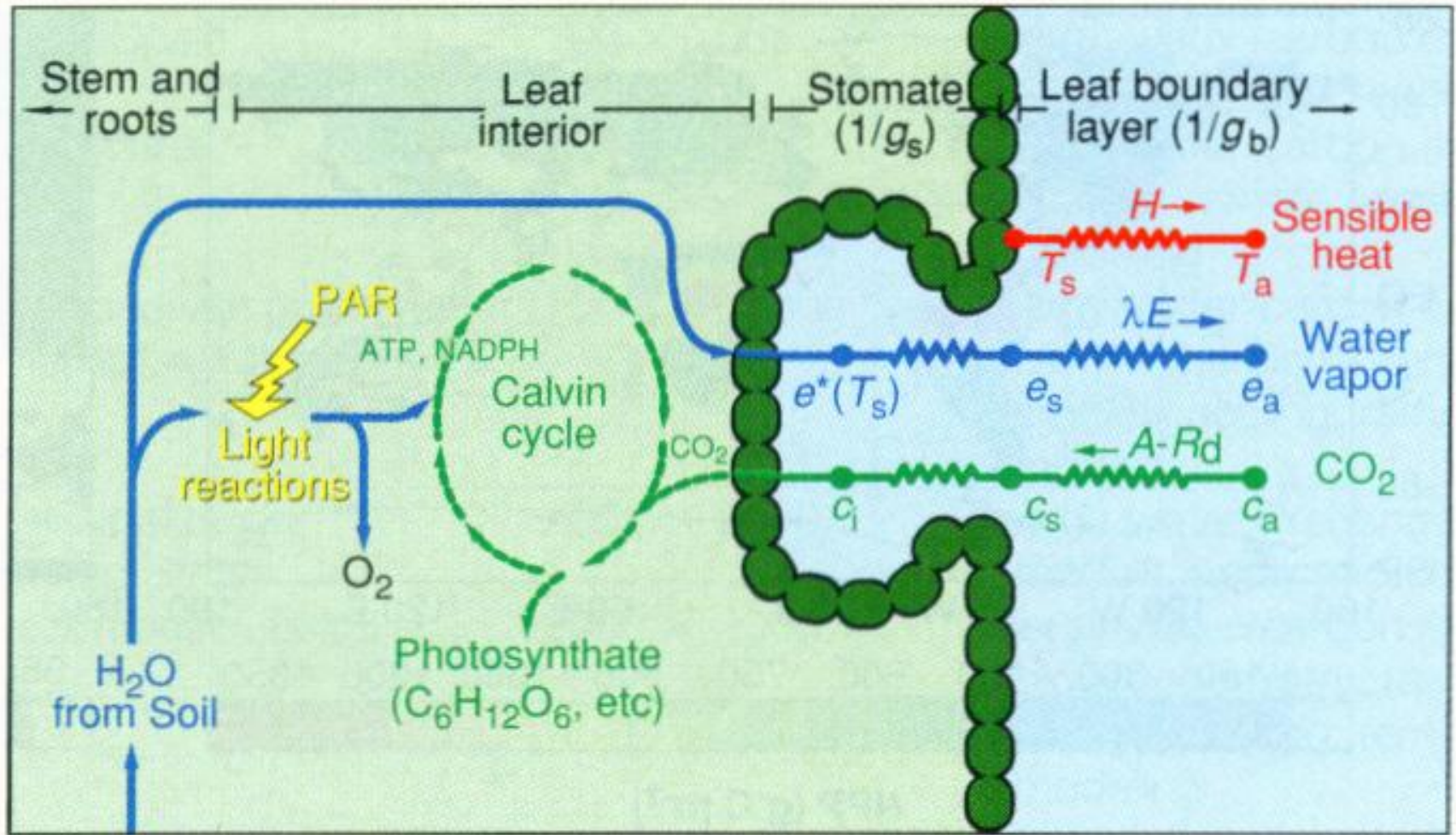
$$A = (c_a - c_i)g_{\text{CO}_2}$$

- Transpiration

$$E = (w_i - w_a)g_{\text{H}_2\text{O}}$$

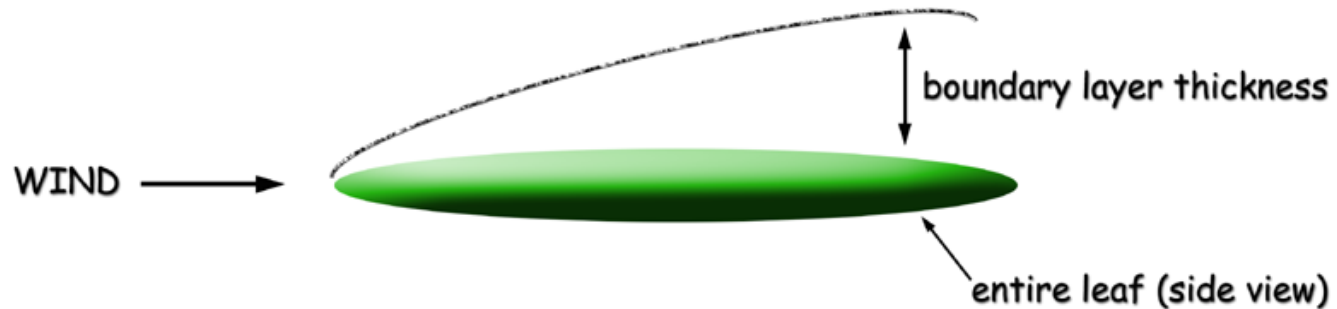


Leaf Conductance



$$r_t = r_{bl} + r_s + r_c + r_m$$

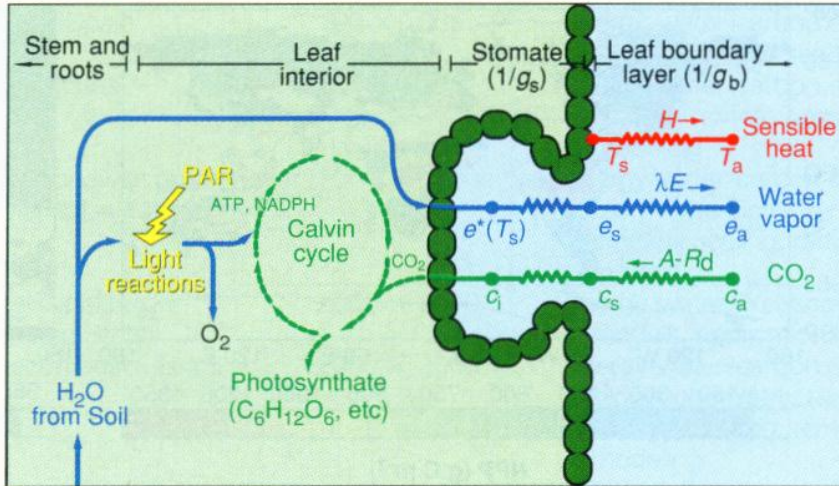
Boundary Layer



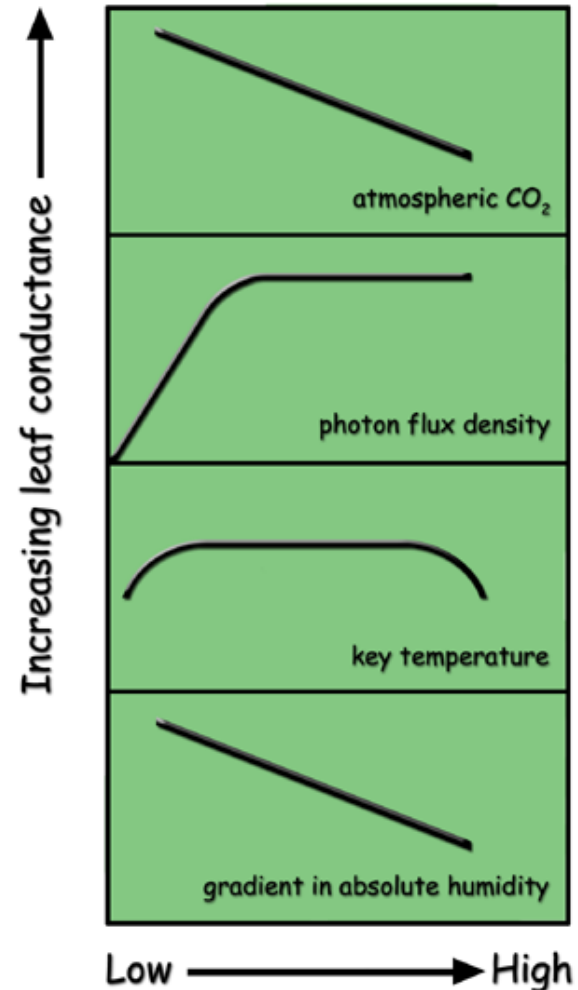
$$\delta_{(mm)}^{bl} = 4.0 \sqrt{\frac{l_{(m)}}{v_{(m\ s^{-1})}}}$$

The boundary layer retards the transfer of heat, CO₂, and H₂O from the leaf to the bulk air.

Stomatal Conductance (g_s)

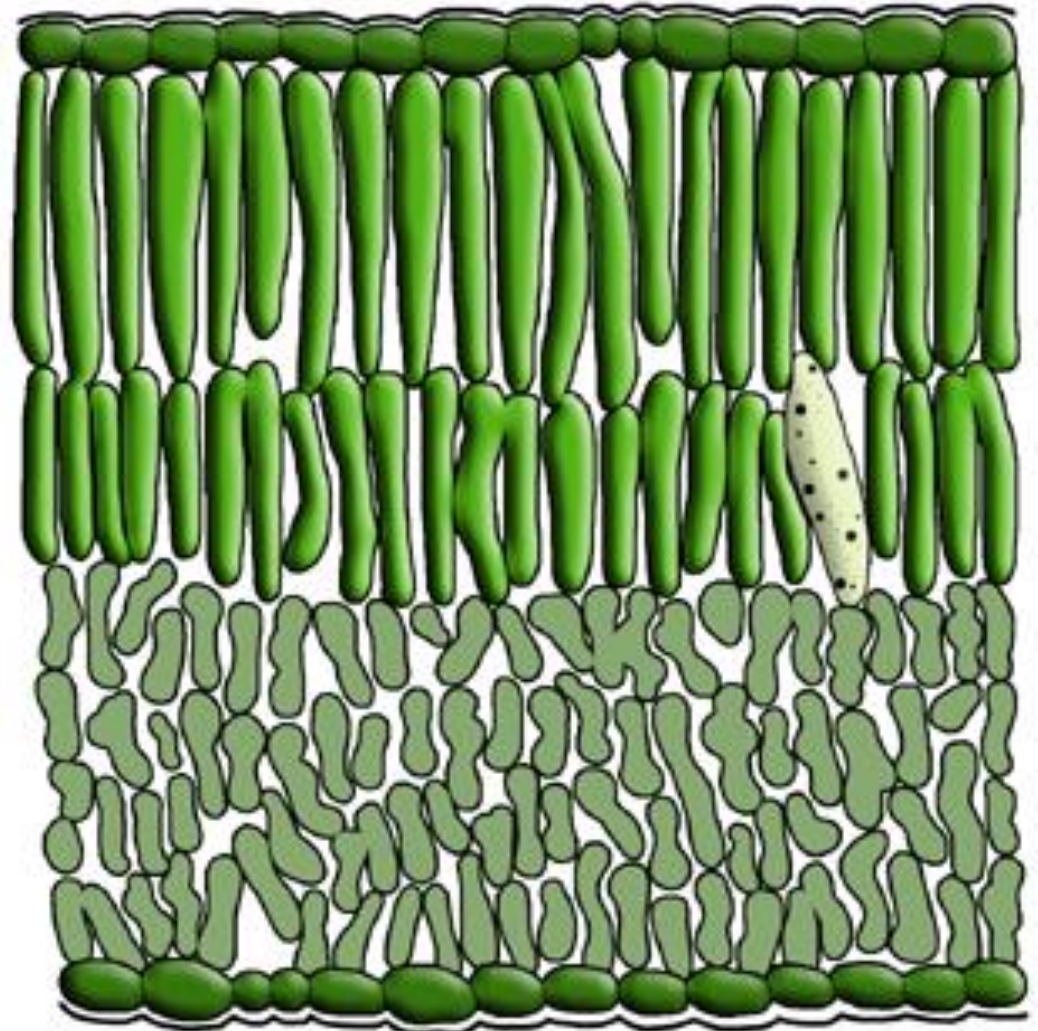


- Major resistance in leaf
 - Regulated to prevent H_2O loss while maximizing CO_2 influx



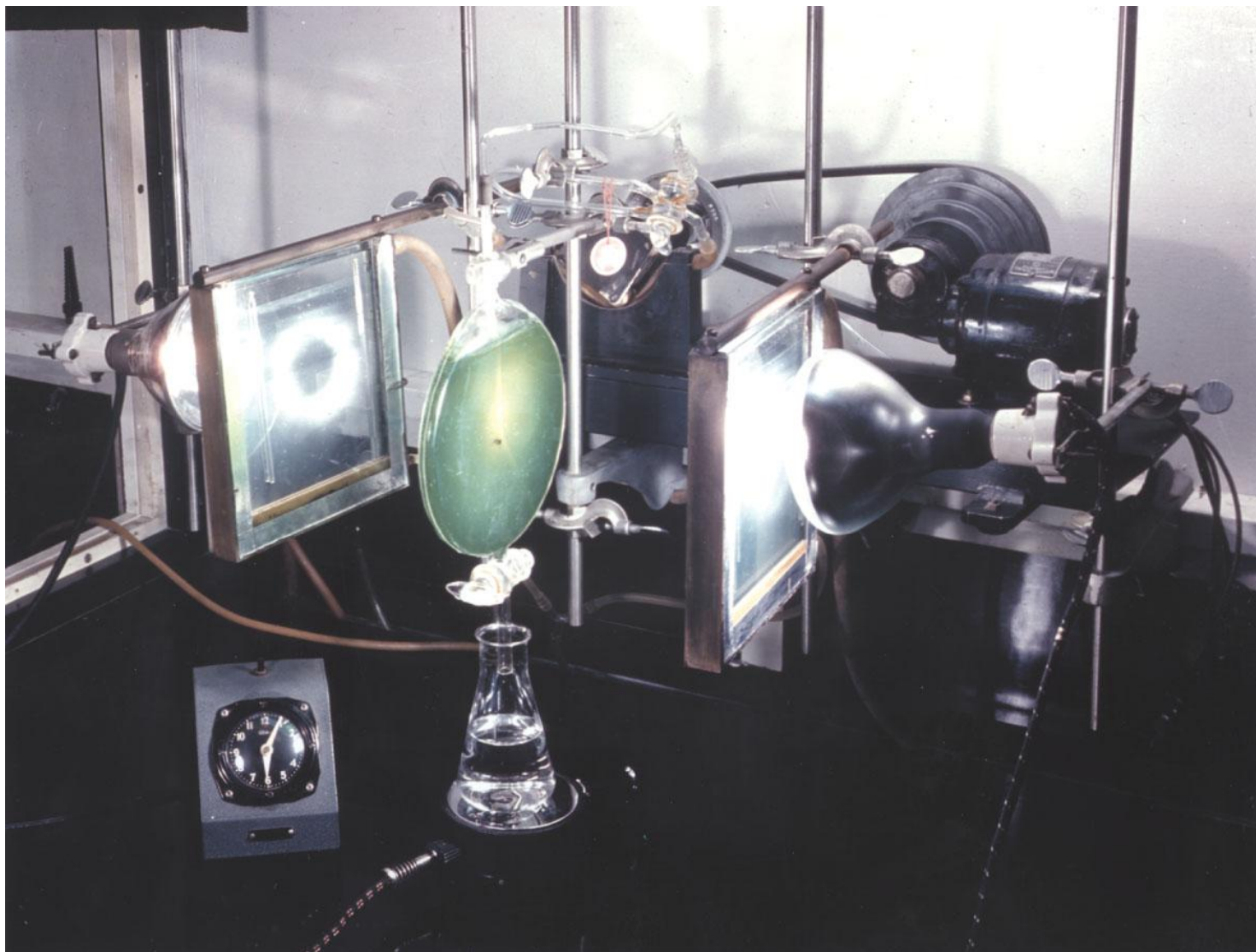
Resistance within Apoplast and Cell

- Apoplastic resistance due to diffusion

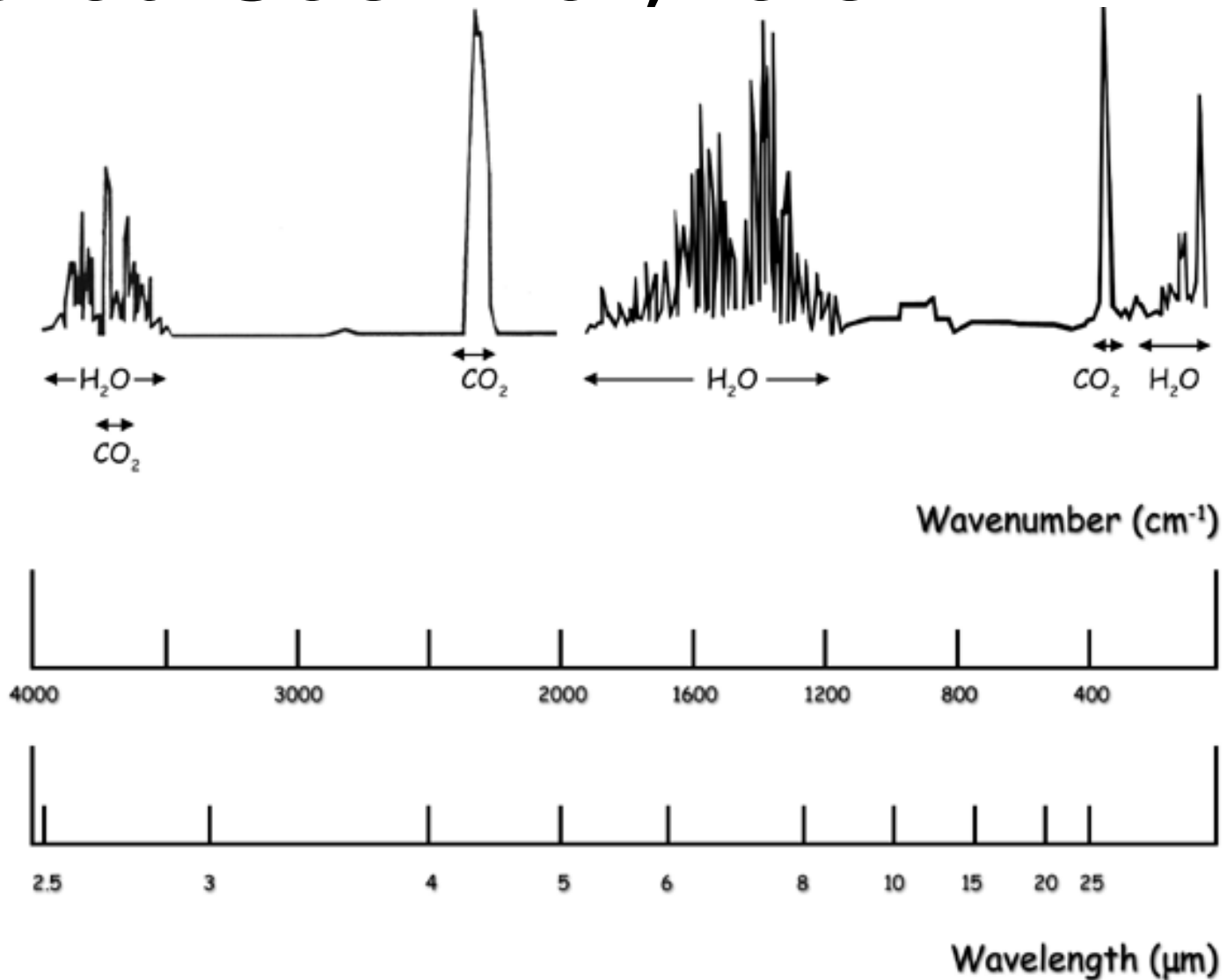


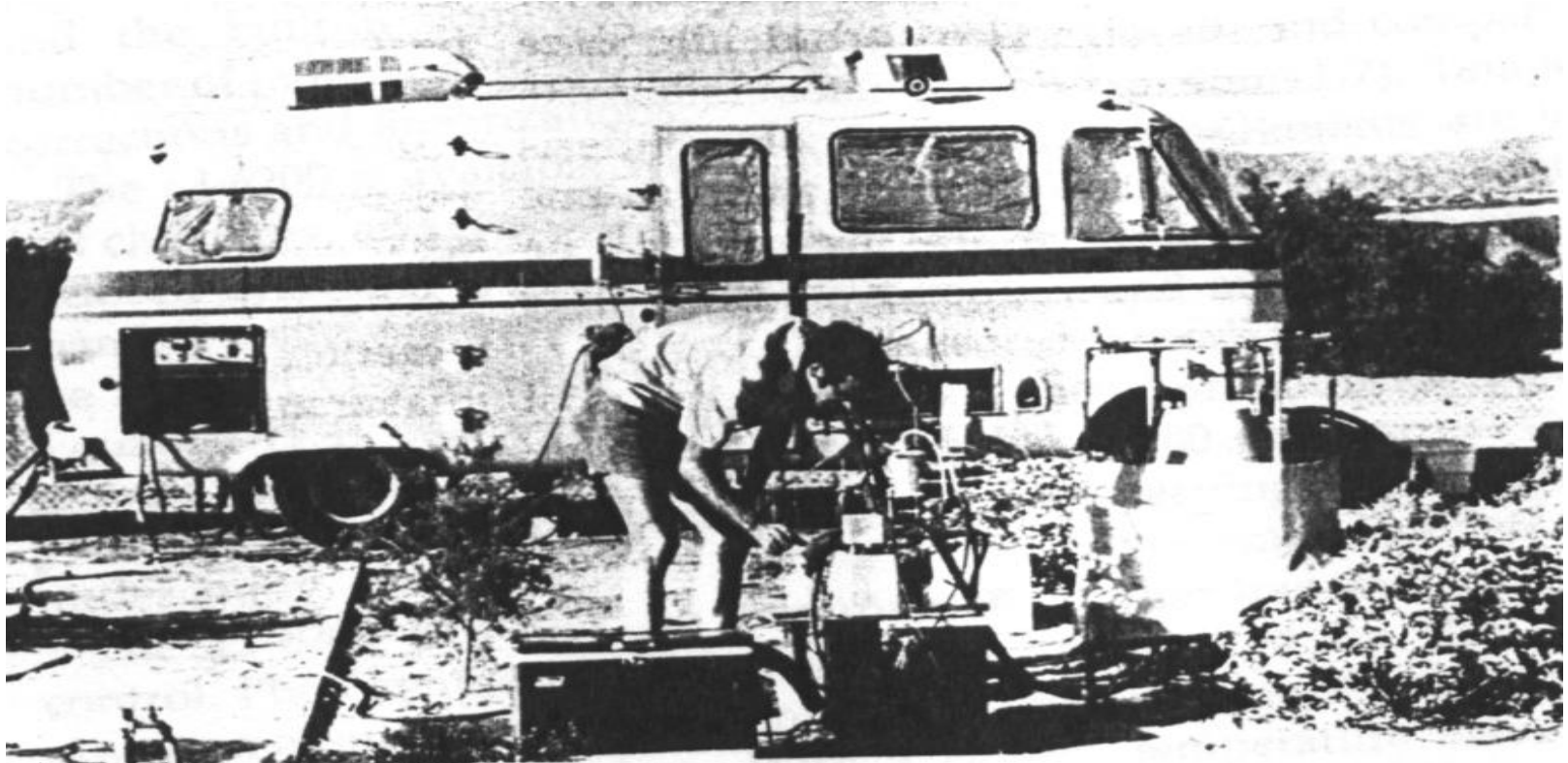
How Are Conductance and Concentrations Measured to Calculate Photosynthesis and Transpiration Fluxes?





Infrared Gas Analyzers



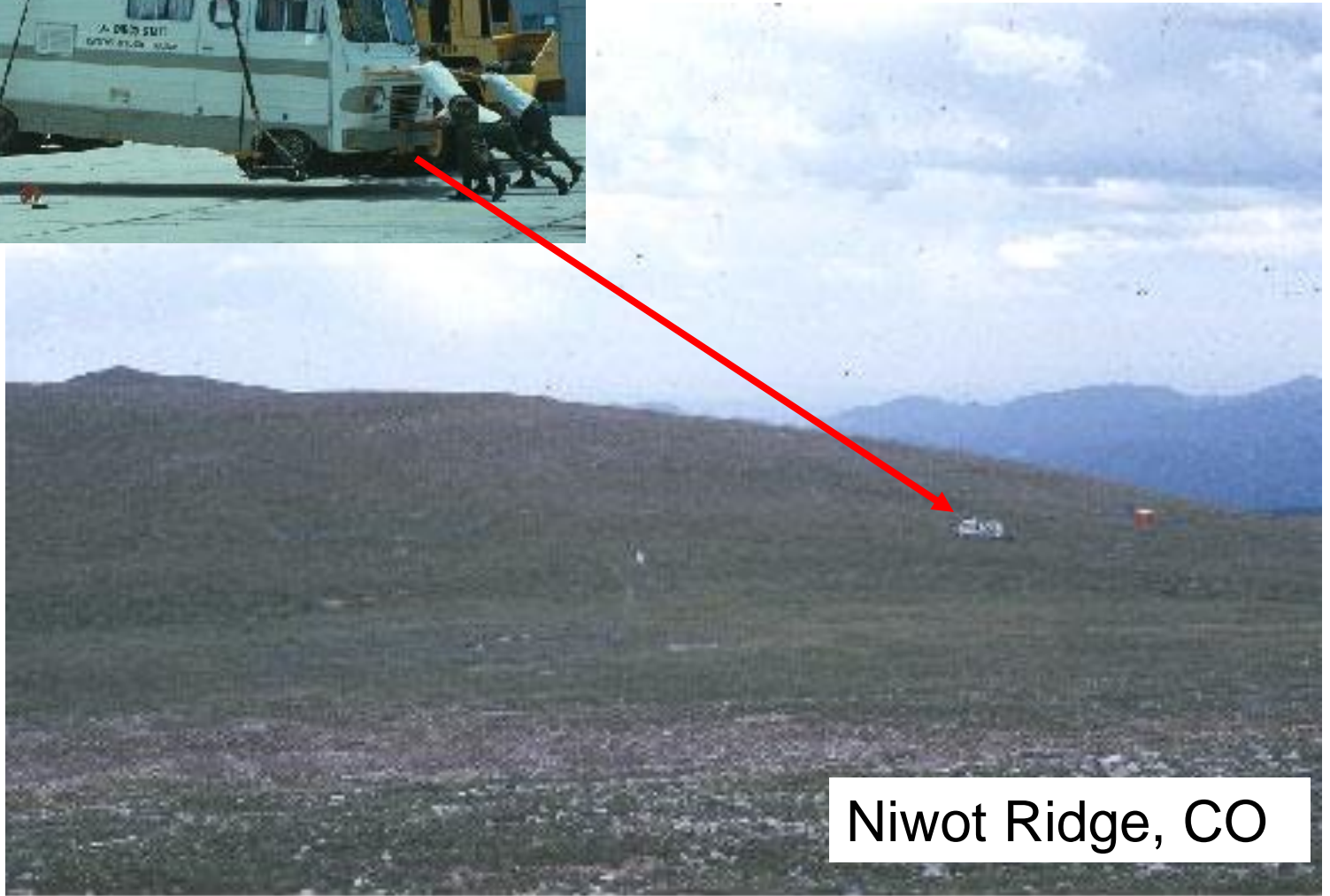


Mooney, H.A., et al. 1971. A mobile laboratory for gas exchange measurements. *Photosynthetica*, 5, 128-32.



LI-COR®

Field measurements,
then ...



Niwot Ridge, CO

Infrared Gas Analyzers



Mooney, H.A., et al. 1971. A mobile laboratory for gas exchange measurements. *Photosynthetica*, 5, 128-32.

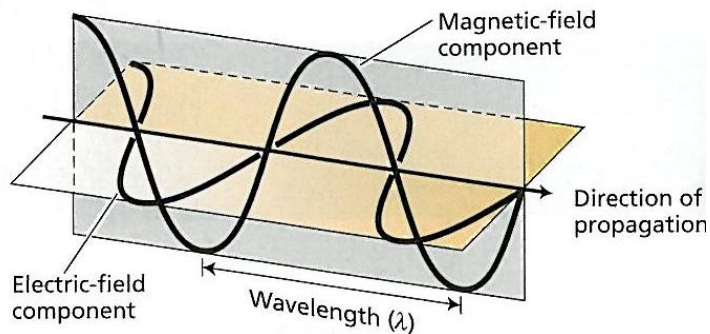
Then



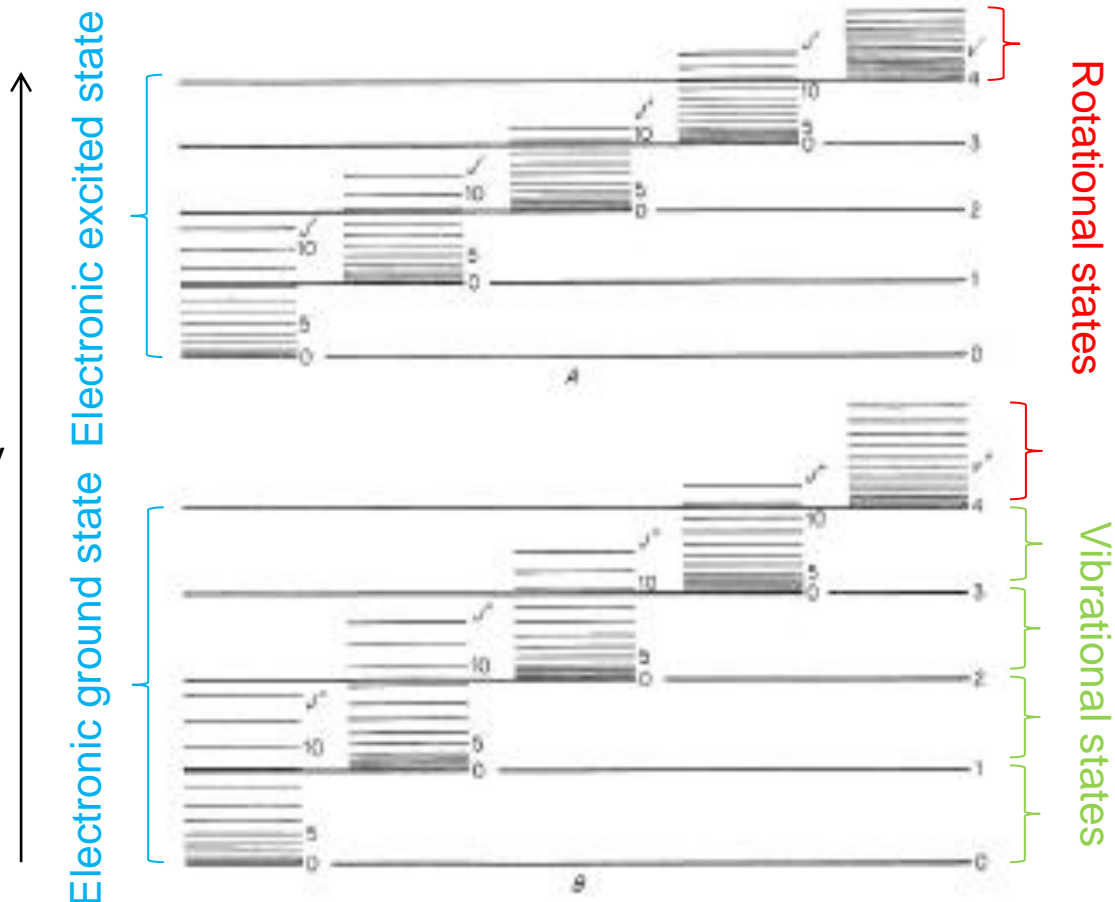
Now

Absorption of light by molecules

The energy of a molecule is the sum of electronic, vibrational, and rotational contributions.



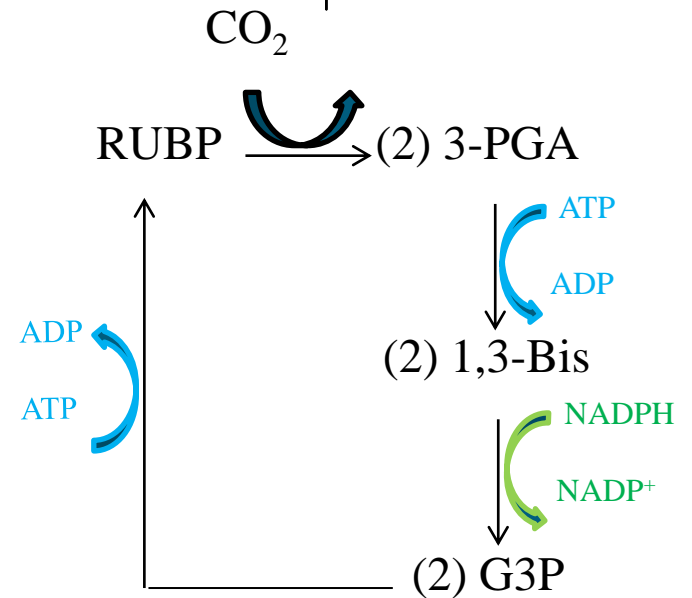
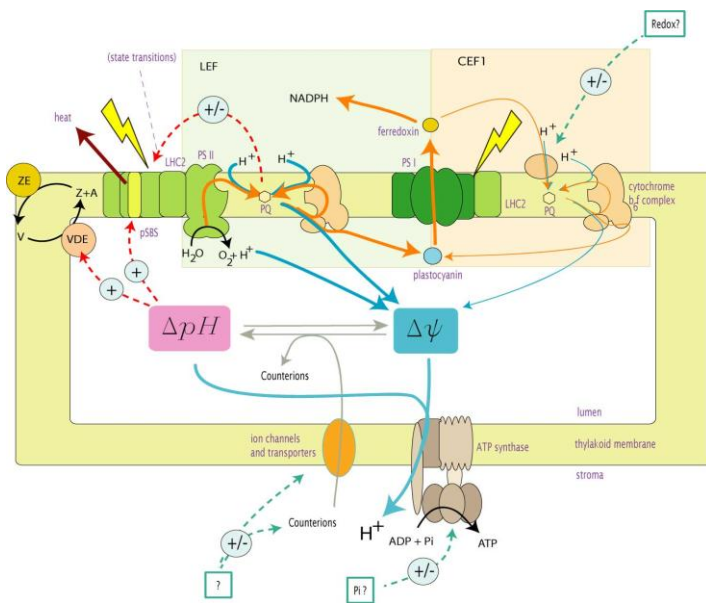
Energy



More, W.J. 1962

Using light to study photosynthesis

<http://cmb.physics.wisc.edu/tutorial/spectrum.htm>

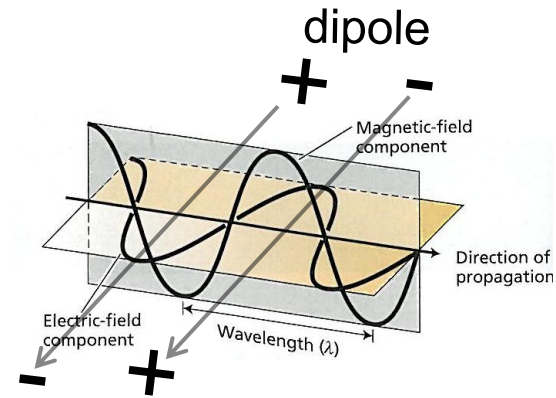
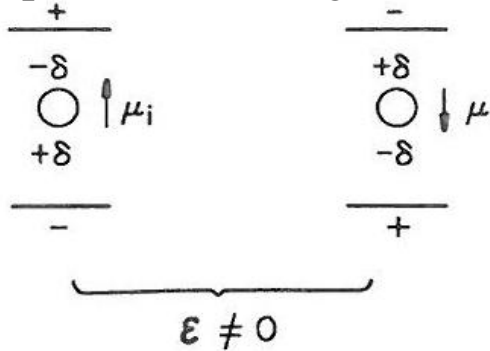


H₂O and CO₂ Absorption in the IR region involves rotations and vibrations

Induction of a dipole (separation of charge)

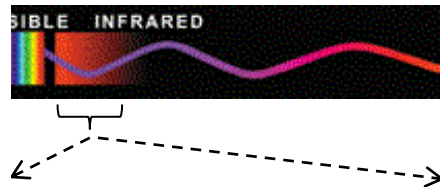
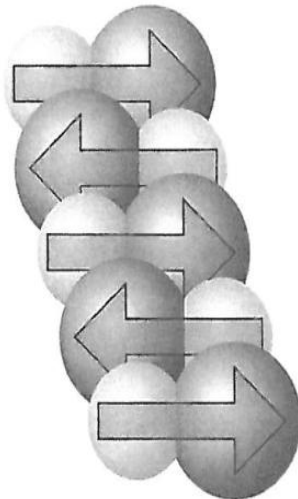
(μ_i = dipole) $\oplus \mu_i = 0$

(ϵ = electric field) $\epsilon = 0$

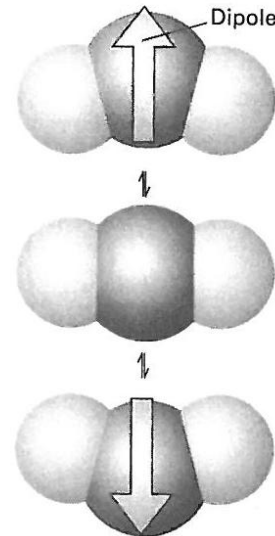


Oscillating dipole of a rotating polar molecule

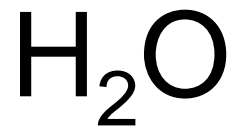
dipole
+ → -



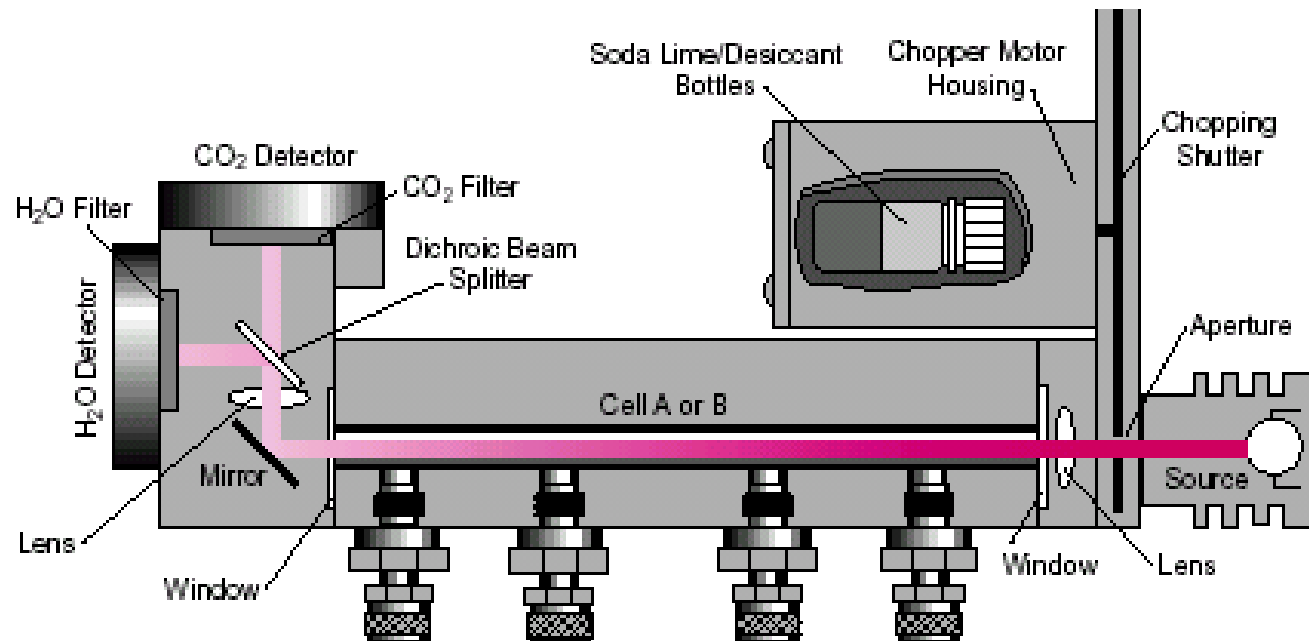
Oscillating dipole of a vibrating non-polar molecule



dipole
- ↑ +

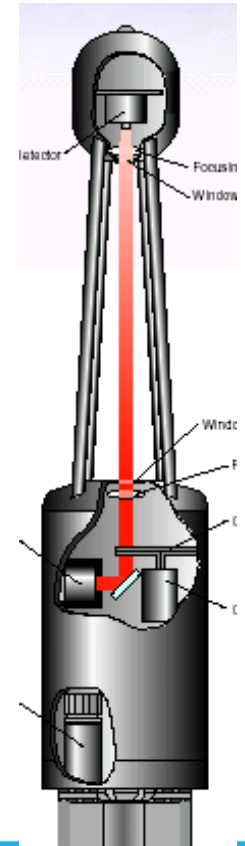
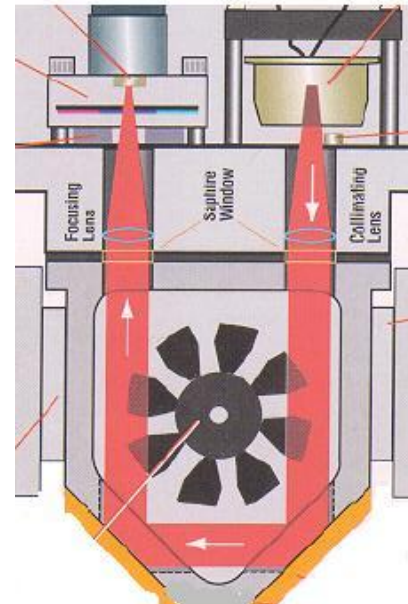
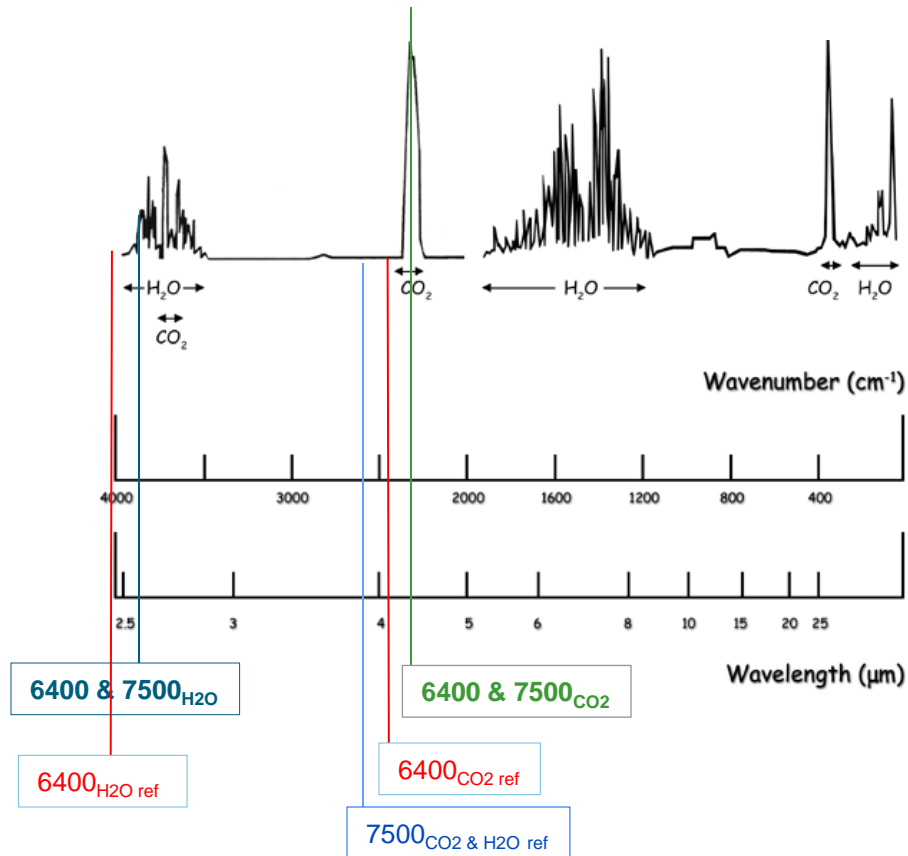


- H₂O absorbs infrared light about 2.4 μm

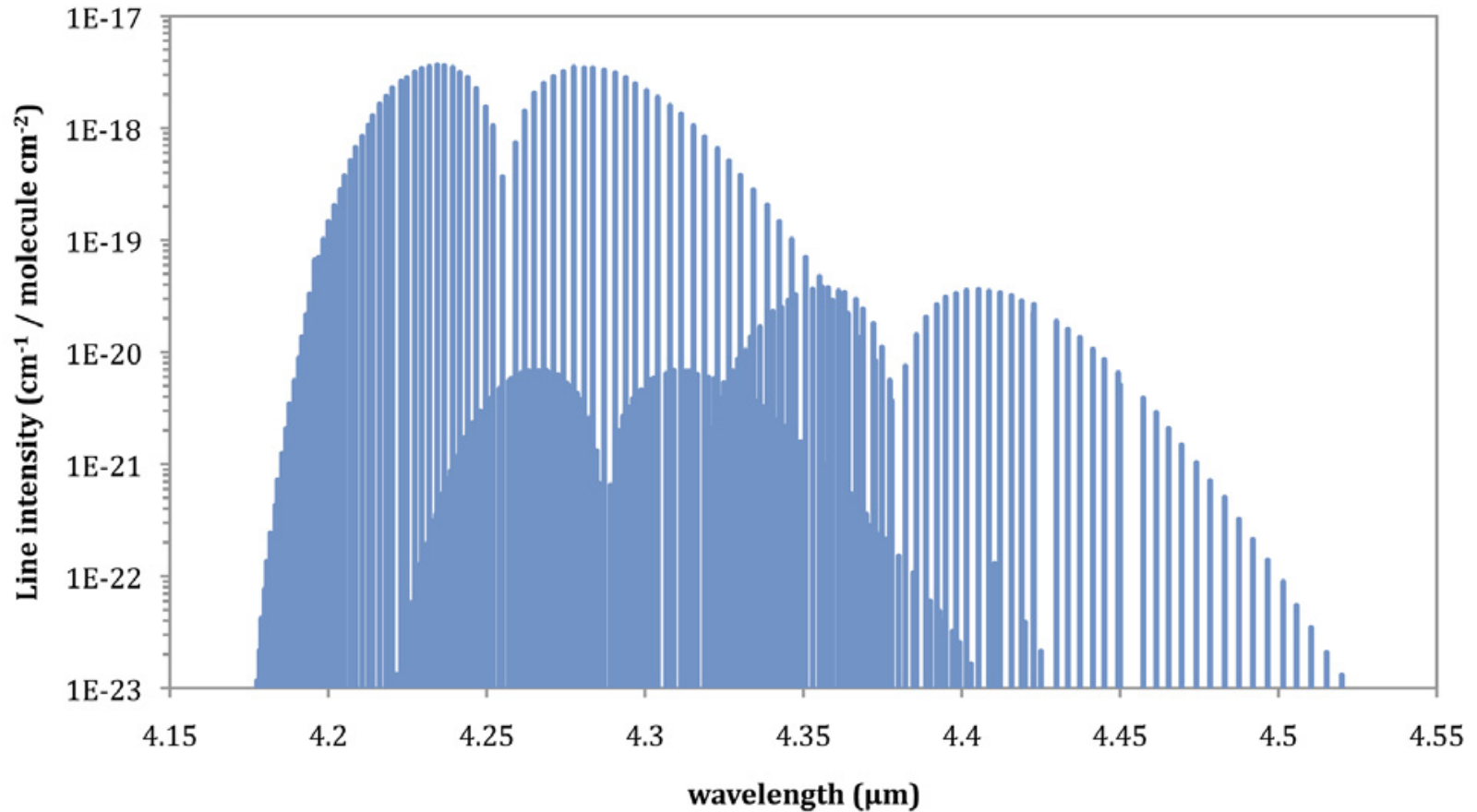




- CO₂ absorbs infrared light about 4.2 μm



CO₂ Absorption Spectrum



Band Broadening Corrections

- Intermolecular collisions with other gases in mixtures affect the IR absorption lines of CO₂.
 - Increasing collisions increases the total absorption without corresponding increase in concentration.

$$C = f\left(V \frac{P_o}{P}\right) \frac{T}{T_o}$$

- C is CO₂ in μmol mol⁻¹, f(x) is the fundamental LI-COR gas analyzer calibration function, P & T are the total pressure (kPa) and temperature (K), P_o = 101.3 kPa, T_o = 273K.

Effective Pressure Broadening

- Not all gases equally effective in causing pressure broadening of absorption lines.

$$P_e = P_{N_2} + \sum \alpha_i P_i + b_{CO_2}$$

- For example water and CO₂ in air would be:

$$P_e = P_{N_2} \alpha_{N_2} + P_{O_2} \alpha_{O_2} + P_w \alpha_w + b_{CO_2}$$

Pressure Broadening Cont.

- Substituting P_e for P in the CO_2 calculation gives the calibration equation that includes the pressure broadening effects of variable water vapor and oxygen:

$$C = c(w, x_o)F \left[\frac{VP_o}{P_c(w, x_o)} \right] \frac{T}{T_o}$$

Gas Exchange Techniques

■ Enclosure Methods

- Disturbs plants to some degree
- Possibility for controlled experiments
- Can measure individual components:
 - Different leaves
 - Roots
 - Soil

■ Micrometerological Methods

- Minimum disturbance
- Summary of large areas
- Difficult to do controlled experiments or isolate individual components



How Are Photosynthesis & Transpiration Measured in an Enclosure?

■ Closed System

$$A = \Delta \text{CO}_2 V (\Delta t S)^{-1}$$

■ No air enters or leaves the system

Leaks can cause large errors

■ Transient measurement

CO_2 , H_2O , T & P changes

■ Open System

$$A = (u_e c_e - u_o c_o) S^{-1}$$

■ Flow of air must be constant & known (accurate flow meter)

■ Steady-state measurement

controlling environmental variables

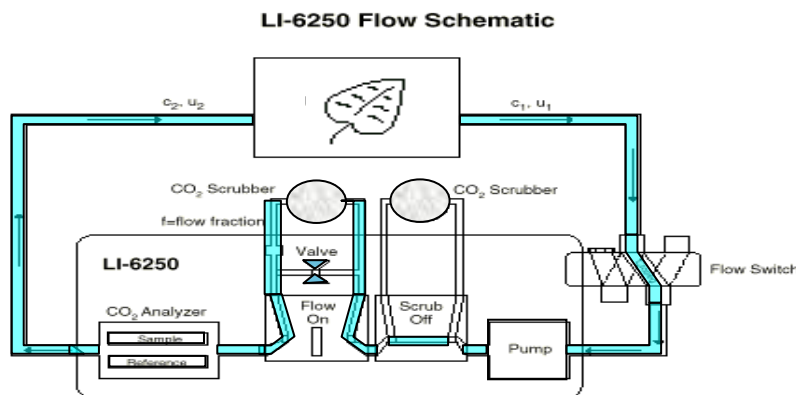
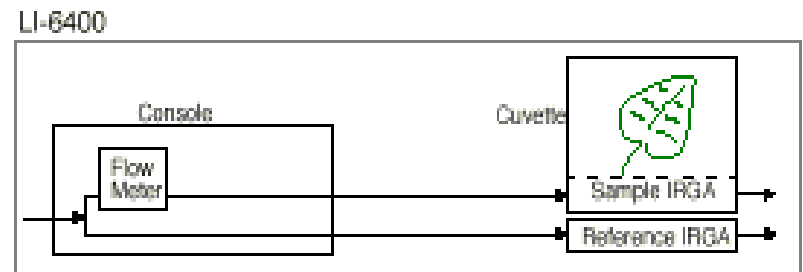
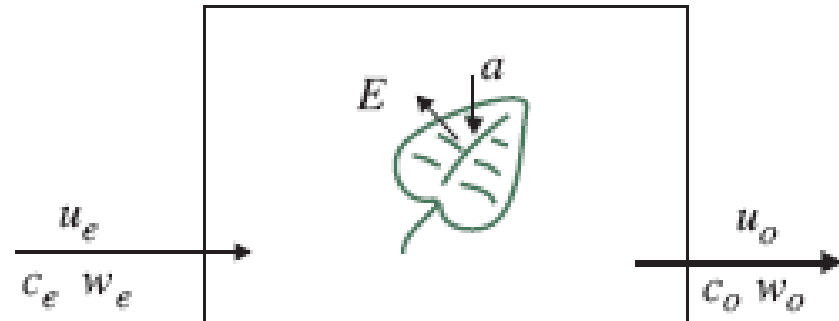


Figure 1. Schematic diagram of flow circuit.



Measured Variables

- CO_2
- H_2O
- Flow rate
- Light
- Temperature
 - Air and Leaf



Assumptions

- Flow rate into chamber equals flow rate out
- Leaf interior saturated $e(t_s)$
- g_c is negligible
- g_{bl} is known
- Diffusion difference between CO_2 and H_2O is 1.6

Leaf Transpiration

$$sE = (\mu_i + sE)w_o - \mu_e w_e$$

$$E = \frac{\mu_e(w_o - w_e)}{s(1 - w_o)}$$

- Transpiration (E) calculated from entering and exiting H₂O concentrations and flow rates.
- Normalized to leaf area (s).

Stomatal Conductance (g_{sw})

$$g_{sw} = \frac{1}{\frac{1}{g_{tw}} + \frac{k_f}{g_{bw}}}$$

$$g_{tw} = \frac{E \left(1 - \frac{w_l - w_e}{2} \right)}{w_l - w_e}$$

$$k_f = \frac{k^2 + 1}{(k + 1)^2}$$

- g_{tw} is calculated from the E , the measured H_2O in chamber (w_e) and assumed H_2O inside leaf (w_l)
- g_{sw} ($\text{mol } H_2O \text{ m}^{-2} \text{ s}^{-1}$) is calculated from total conductance
 - Boundary layer known
- k_f is a factor based on the estimate k
 - k = fraction of stomatal conductance of one side of the leaf to the other (*stomatal ratio*).

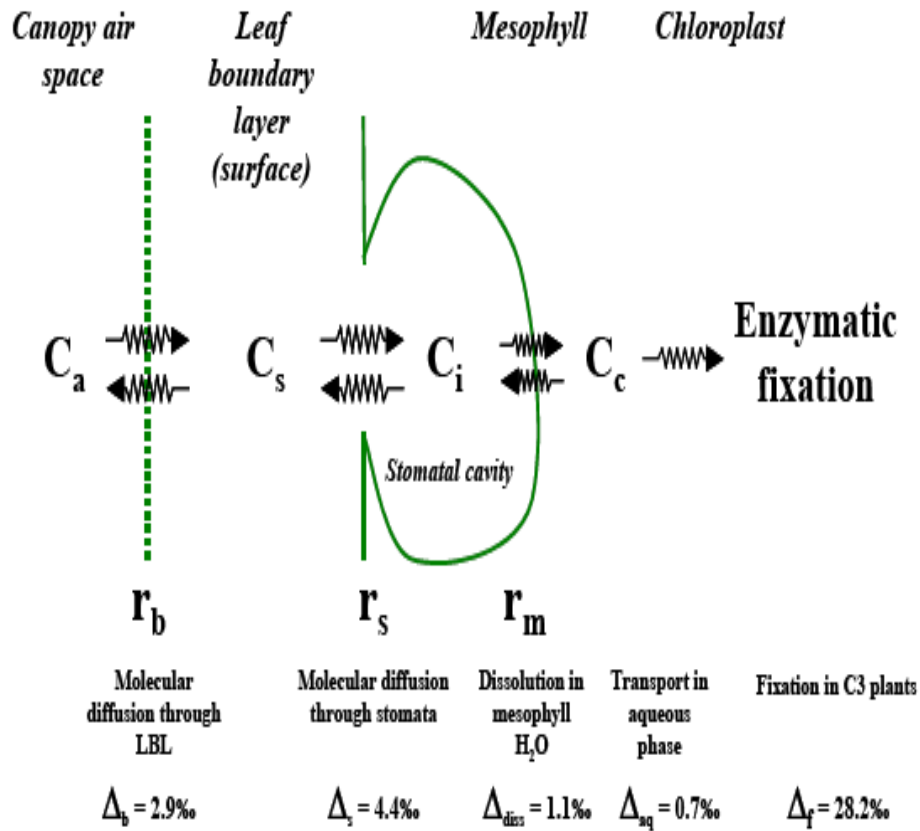
Photosynthesis

- $u_o = u_e + sE$
- $sA = u_e c_e - (u_e + sE) c_o$

$$A = \frac{u(C_r - C_s)}{100S} - C_s E$$

- μ is flow rate ($\mu\text{mol mol}^{-1}$), **C_r** and **C_s** are reference and sample CO_2 concentrations ($\mu\text{mol mol}^{-1}$), **S** is leaf area (m^2), and **E** is transpiration ($\text{mol m}^{-2} \text{s}^{-1}$)
- Dilution due to H_2O vapor

C_i

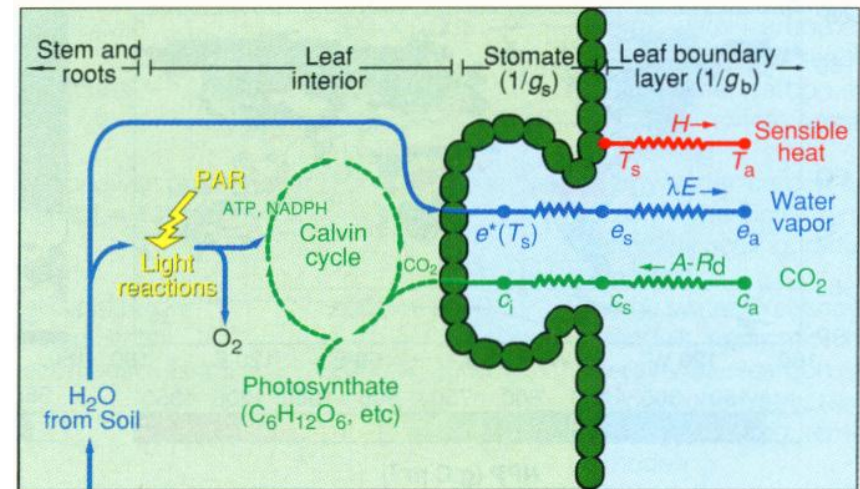


- $g_{sc} = 1.6 g_{sw}$

- $r_{sc} = 1 / g_{sc}$

- Fick Law

- $A = (c_a - c_i) / r_{sc}$



C_i cont'd.

$$g_{tc} = \frac{1}{\frac{1.6}{g_{sw}} + \frac{1.37k_f}{g_{bw}}}$$

$$C_i = \frac{(g_{tc} - \frac{E}{2})C_s - A}{g_{tc} + \frac{E}{2}}$$

- The ratio between the diffusivities of CO₂ and water in the boundary layer is 1.37, compared to 1.6 in air

Some Uses of Leaf-level Measures

- Plant growth models
- GCMs
 - Canopy A, E and g_s validation and modeling
- Eddy Covariance
 - Scale-up validation

What parameters do researchers want to measure?

- Net flux of CO_2 (photosynthesis or respiration)
- Transpiration (E) & Stomatal Conductance (g_s)
- Leaf internal CO_2 concentration (C_i) and CO_2 concentration inside the chloroplast (C_c)
- Electron Transport Rate (J)
- Velocity of Carboxylation (V_c)

Exploring limitations to photosynthesis

Journal of Experimental Botany, Vol. 54, No. 392, pp. 2393–2401, November 2003
DOI: 10.1093/jxb/erg262



REVIEW ARTICLE: FIELD TECHNIQUES

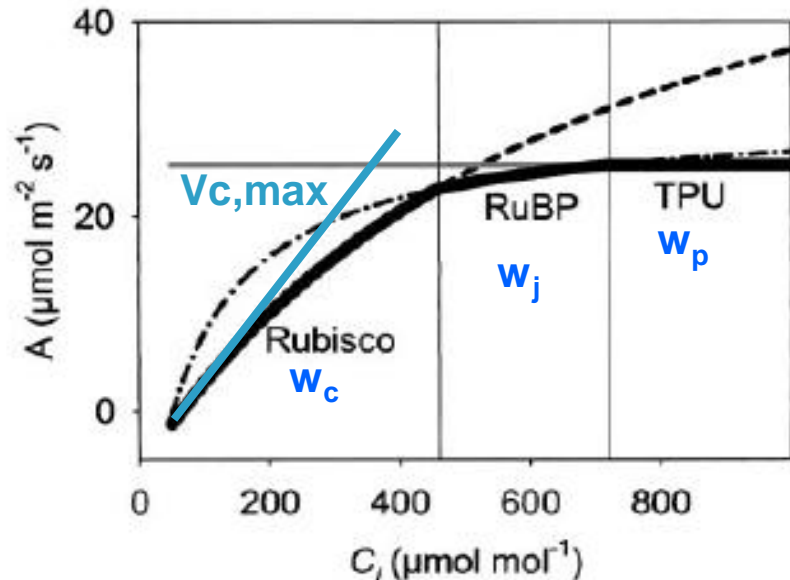
Gas exchange measurements, what can they tell us about the underlying limitations to photosynthesis? Procedures and sources of error

S. P. Long^{1,*} and C. J. Bernacchi²

¹ Departments of Crop Sciences and Plant Biology, University of Illinois at Urbana-Champaign, 379 Edward R. Madigan Laboratory, Urbana, IL 61801, USA

² USL

Receiv



A- C_i Curves

Farquhar, von Caemmerer & Berry analysis:

$$A = \min\{w_c, w_j, w_p\} \left(1 - \frac{\Gamma^*}{C_i}\right) - R_d$$

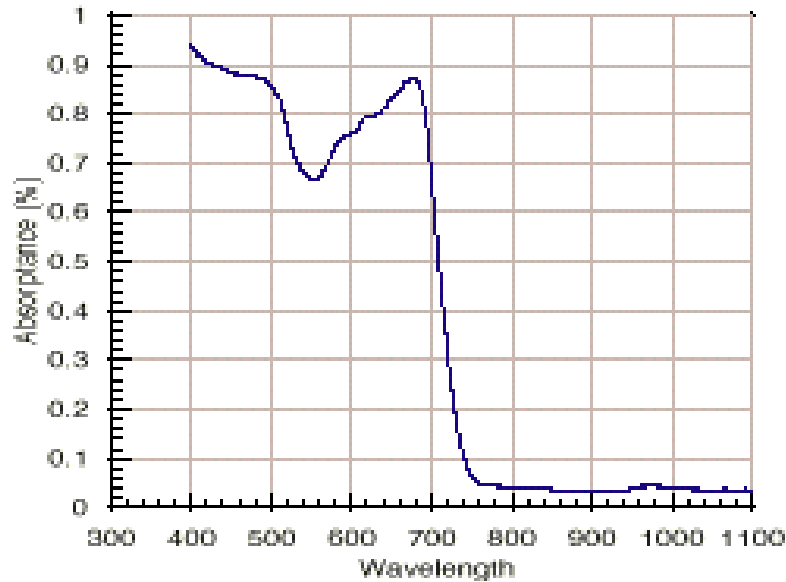
where:

$$w_c = \frac{V_{c,\max} C_i}{C_i + K_c(1 + O/K_o)}$$

$$w_j = \frac{J C_i}{4.5 C_i + 10.5 \Gamma^*}$$

$$w_p = \frac{3 V_{\text{tpu}}}{\left(1 - \frac{\Gamma^*}{C_i}\right)}$$

Exploring limitations to photosynthesis

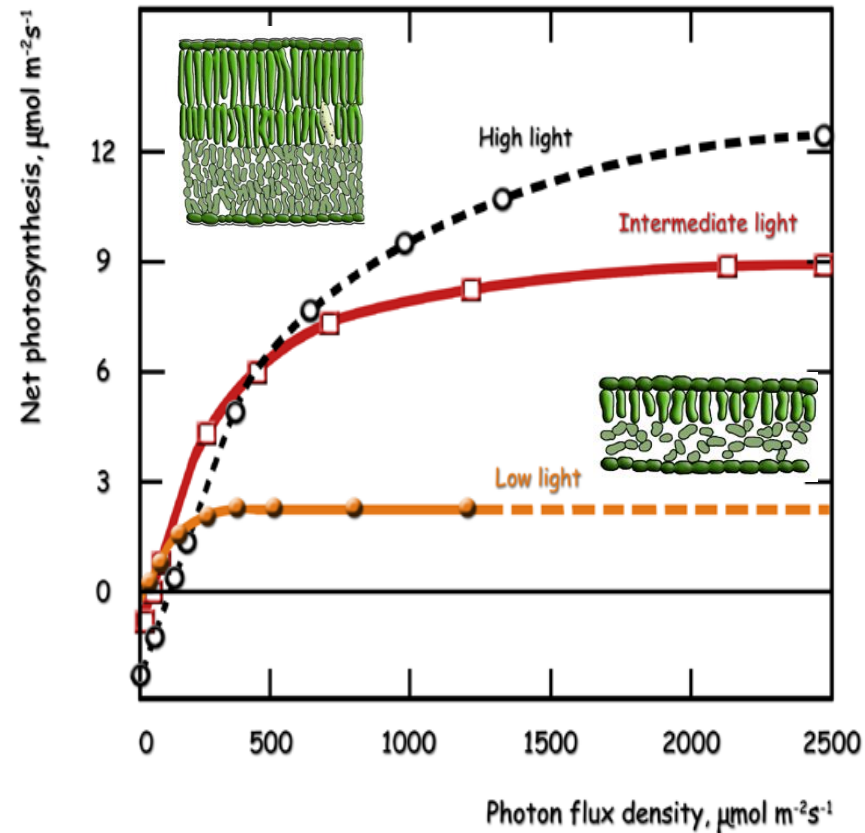
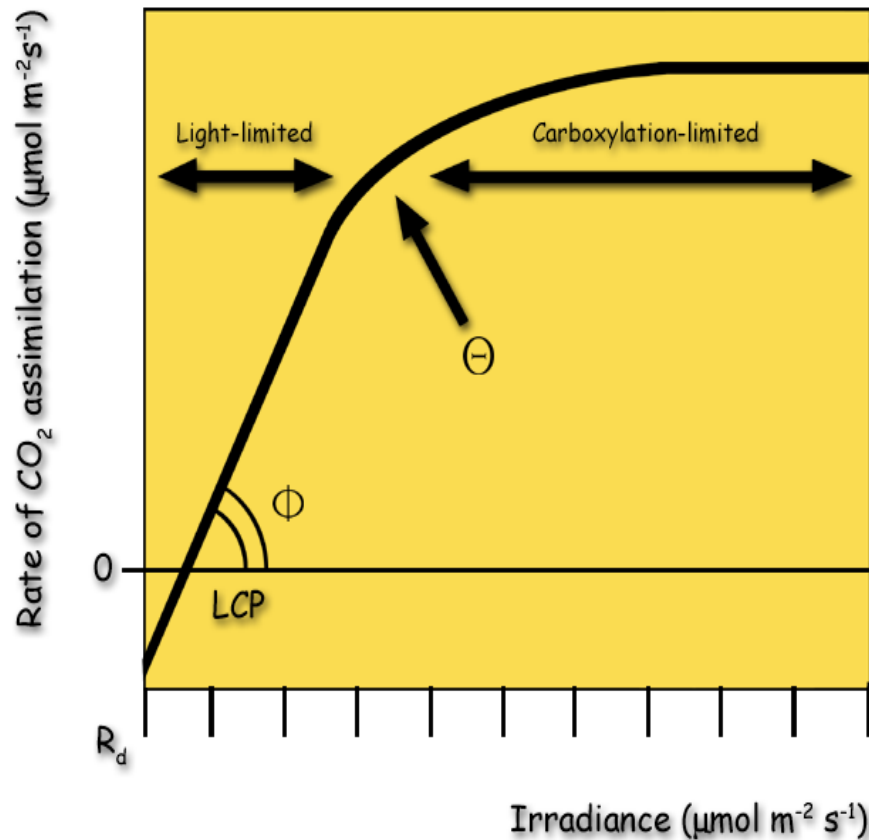


Photosynthesis depends more on the number of photons, rather than the amount of energy of the photons

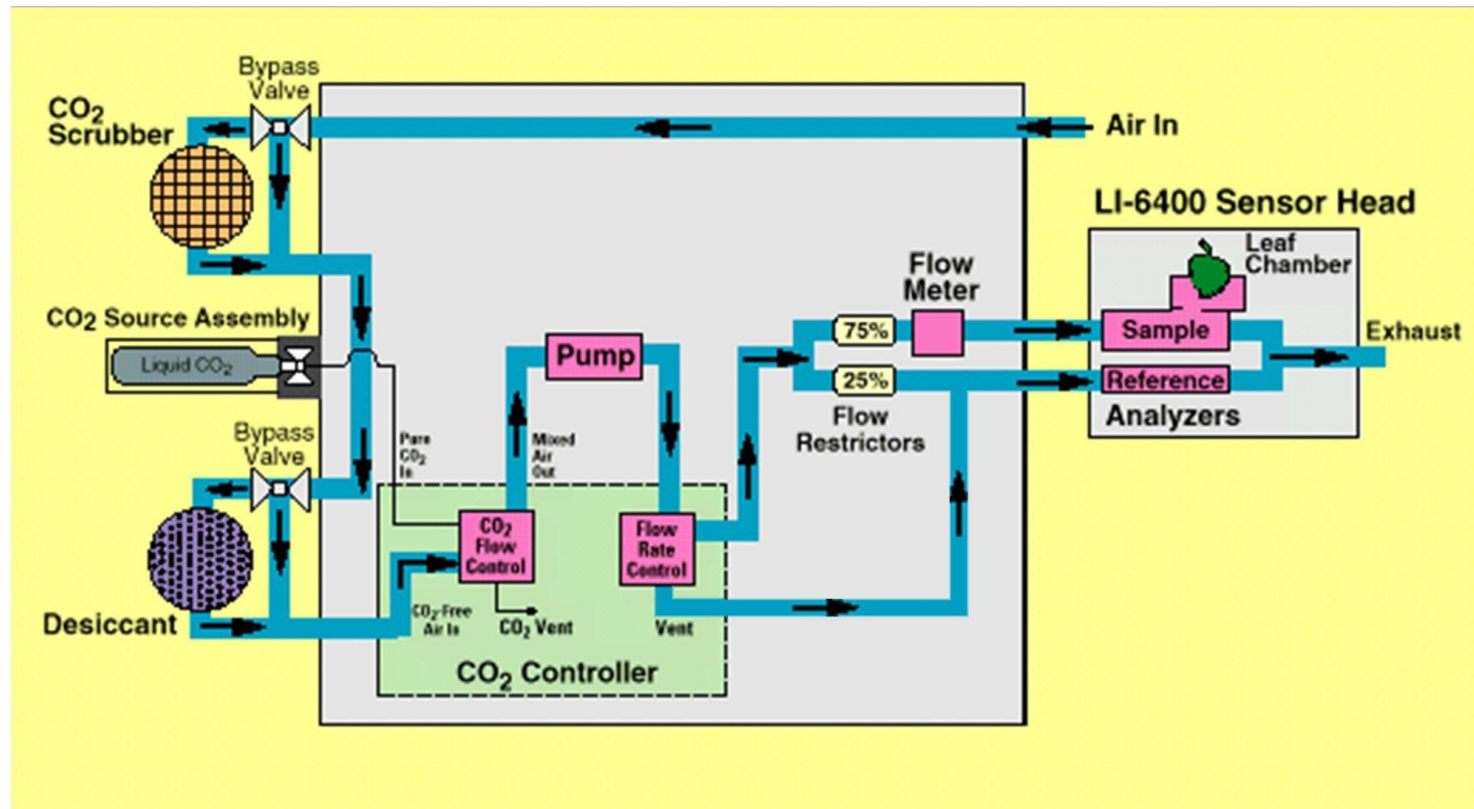
- Irradiance: Radiant flux incident on a surface (W m^{-2})
- Photosynthetically Active Radiation (PAR): Radiation 400 to 700 nm (W m^{-2}), about 50% of total radiation
- Photosynthetic Photon Flux Density (PPFD): Number of photons (400-700 nm) incident per unit time and area ($\mu\text{mol m}^{-2} \text{s}^{-1}$)

Exploring limitations to photosynthesis

Light Curves



Flow Path



Mixing and Analyzers

