

# VERIFICATION OF LAWS OF PHOTOELECTRIC EFFECT USING V-LAB SIMULATIONS

*Submitted in partial fulfillment of the requirement for the degree of*

**Bachelor of Science**

**in**

**PHYSICS**

*by*

**RITHIK VINAYARAJ**

PR18CPHR17

Under the guidance of

**Dr.Deepa K**

HOD

PAZHASSI RAJA NSS COLLEGE, MATTANUR

KANNUR UNIVERSITY



## **DECLARATION**

I hereby declare that the project report entitled “ **VERIFICATION OF PHOTOELECTRIC EFFECT USING V-LAB SIMULATIONS**” submitted for the award of the degree of *Bachelor of Science in Physics* to Kannur University is a record of original and bonafide work done by our project group carried out by the members under the guidance and supervision of **Dr.Deepa K**

I further declare that the Information and Data given in this project is authentic and true to the best of my knowledge. The work report in this project has not been submitted and will not be submitted, either in part or in full, for the award of any other degree or diploma or any other fellowship in this institute or any other institute or university.

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**DEPARTMENT OF PHYSICS**  
**PAZHASSI RAJA N.S.S COLLEGE, MATTANUR**



(AFFILIATED TO KANNUR UNIVERSITY)

**CERTIFICATE**

This is to certify that the Project report entitled “ **VERIFICATION OF PHOTOELECTRIC EFFECT USING V-LAB SIMULATOR**” is a bonafide work submitted by **ASWATHI.P, RITHIK VINAYARAJ** and **AMALENDU RAGHUNATH** of VI<sup>th</sup> semester in partial fulfillment of the requirement for the awarding of the degree of Bachelor of Science in Physics, Kannur University.

The project (Core XVIII- 6B18PHY) is a record of original work done by them under my supervision, as per the K.U code of academic & research ethics.

The project is not submitted to any other institute for the award of any other degree in any university.

The project fulfills the requirement and regulation of the university and in my opinion meets the necessary standard for submission.

Place:

Date :

Signature of Guide

*Dr. Deepa K*

HOD-PHYSICS

Internal Examiner

External Examiner

## **ACKNOWLEDGEMENT**

I'm very glad to express my deep sense of gratitude to all the staff of Dept.Of.Physics - PRNSS College, Mattanur for their encouragement and support in presenting the project. I'm highly indebted to Dr.Deepa.K – Assistant Professor of Physics (my guide) for her guidance and constant supervision as well as for providing necessary information regarding the project and also for her support in completing the project.

I can't forget to offer my sincere thanks to my astoundingly supportive fellow colleague for their kind co-operation & encouragement which helped me in completion of this project. My thanks & appreciation also for my colleague in developing the project and people who have willingly helped me out with their abilities.

# **CONTENT**

➤ Abstract

➤ Introduction

- Objective
- Motivation
- Background

➤ Theory

➤ Project Descriptions & Goals (Methodology)

➤ Results and Discussion

➤ Conclusions

➤ Reference

➤ Appendix1: Github repository

## **ABSTRACT**

This project report is on the experimental verifications of **Laws of Photoelectric Effect** using Amritha Virtual lab simulations. The laws of Photoelectric Effect have been verified for various metals like Copper, Zinc and Platinum. The experimental curves could be reproduced for these metals for various intensities and frequencies of the incident radiation.

The curves are plotted using Visualization with Python module namely **Matplotlib** in **Jupyter notebook** – open source Computational notebook ,using **Anaconda Navigator-3 Distribution Package** of the python for scientific computing as well as plotting manually in a graph.

**The Work Function ( $\Phi$ )** of the metals has also been calculated.

# **INTRODUCTION**

## **OBJECTIVE :**

1. To understand the phenomenon- Photoelectric Effect as a whole.
2. To draw the Kinetic Energy ( $KE_{\max}$ ) of photoelectrons as a functions of frequency of incident radiation.
3. To plot a graph connecting photocurrent and applied potential
4. To determine the stopping potential from the photocurrent Vs applied potential graph.

## **MOTIVATIONS :**

Physicists at the end of the nineteenth century believed that most of the fundamental physical law had been worked out. They expected only minor refinement to get “an extra decimal place” of accuracy. As it turns out , the field of physics was transformed profoundly in the early twentieth century by Einstein’s discovery of relativity and by the development of Quantum Mechanics. While relativity has had daily little impact on chemistry, all of theoretical chemistry is founded upon Quantum Mechanics.

The Development of Quantum Mechanics was initially motivated by two observations which demonstrated the inadequacy of classical physics. These are the “Ultraviolet Catastrophe” and “**The Photoelectric Effect**”.

## **BACKGROUND :**

The photoelectric effect was discovered in 1887 by the German physicist Heinrich Rudolf Hertz. In connection with work on radio waves, Hertz observed that, when ultraviolet light shines on two metal electrodes with a voltage applied across them, the light changes the voltage at which sparking takes place. This relation between light and electricity (hence photoelectric) was clarified in 1902 by another German physicist, Philipp Lenard. He demonstrated that electrically charged particles are liberated from a metal surface when it is illuminated and that these particles are identical to electrons, which had been discovered by the British physicist Joseph John Thomson in 1897.

The experimental observations with respect to variation in intensity and frequency of incident radiation were compiled by Lenard and are often referred to as '*Law of Photoelectric Effect*'. As the first experiment demonstrated the Quantum Theory of Energy levels, photoelectric experiment is of great historical importance.

In this work, attempts have been made to verify the laws of photoelectric effect using Amritha V-lab simulations. Virtual lab provides remote-access to labs in various disciplines of Science and Engineering, and helps in learning basic and advanced concepts through remote experimentation. This also helps to share costly equipments and resources, which are otherwise available to limited number of users due to constraints on time and geographical distance.



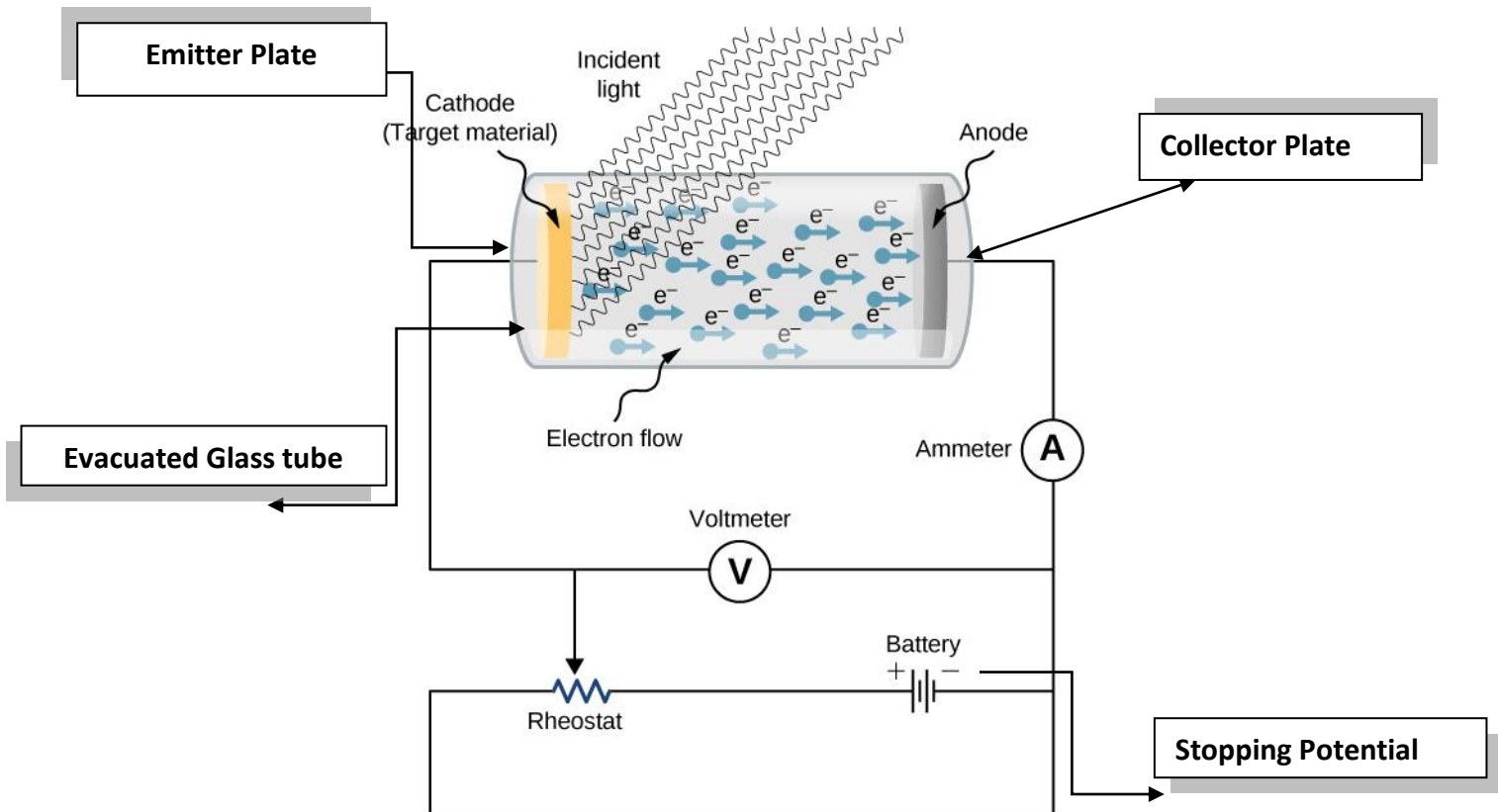
# THEORY

When Electromagnetic radiation ( $EM_{\text{wave}}$ ) of high frequency get incident on clean metal surface, electron is emitted from the surface of the metal. This phenomenon is called **Photoelectric Effect** and the emitted electrons are called **Photoelectrons**. The UV-light, X-ray,  $\gamma$ -ray, Visible-light and InfraRed produces this effect.

The phenomenon is studied in condensed matter physics, and solid state and quantum chemistry to draw inferences about the properties of atoms, molecules and solids. The effect has found use in electronic devices specialized for light detection and precisely timed electron emission.

## Experimental arrangement of Photoelectric Effect

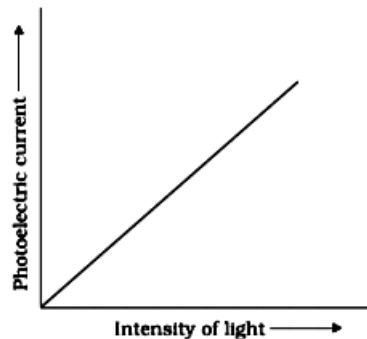
fig1.



## Laws of Photoelectric Effect.

The experimental observations were compiled by Philip Lenard and are known as “*Laws of Photoelectric Effect*” for which he won Nobel Prize in Physics-1905.

- 1) For given sensitive surface there is a minimum frequency of the incident radiation below which there is no emission of photoelectrons (photons). This is known as **Threshold frequency** ( $\nu_0$ ).
- 2) For a given frequency  $\nu$ , the photoelectric current under a constant accelerating potential is directly proportional to the intensity of the incident radiation. The variation is shown in the **fig2**.



Current-Intensity Graph @ constant potential

Fig 2. Variation of Photocurrent with intensity of radiation

- 3) The stopping potential  $V_0$  and hence the maximum kinetic energy  $KE_{\max}$  of the emitted electrons is independent of the intensity of the incident light. **Fig3**. Shows the plot for various intensities of incident radiation.

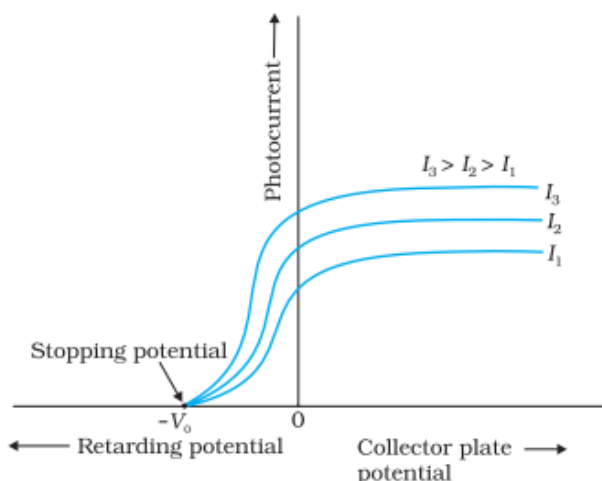
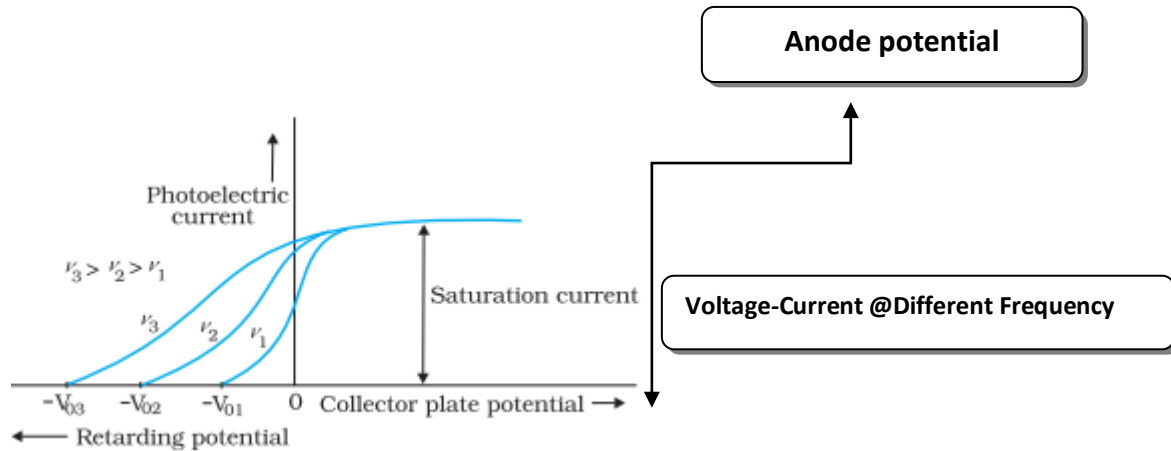


Fig3. Variation of photocurrent with retarding potential for various intensity

Voltage-Current @Different Intensity

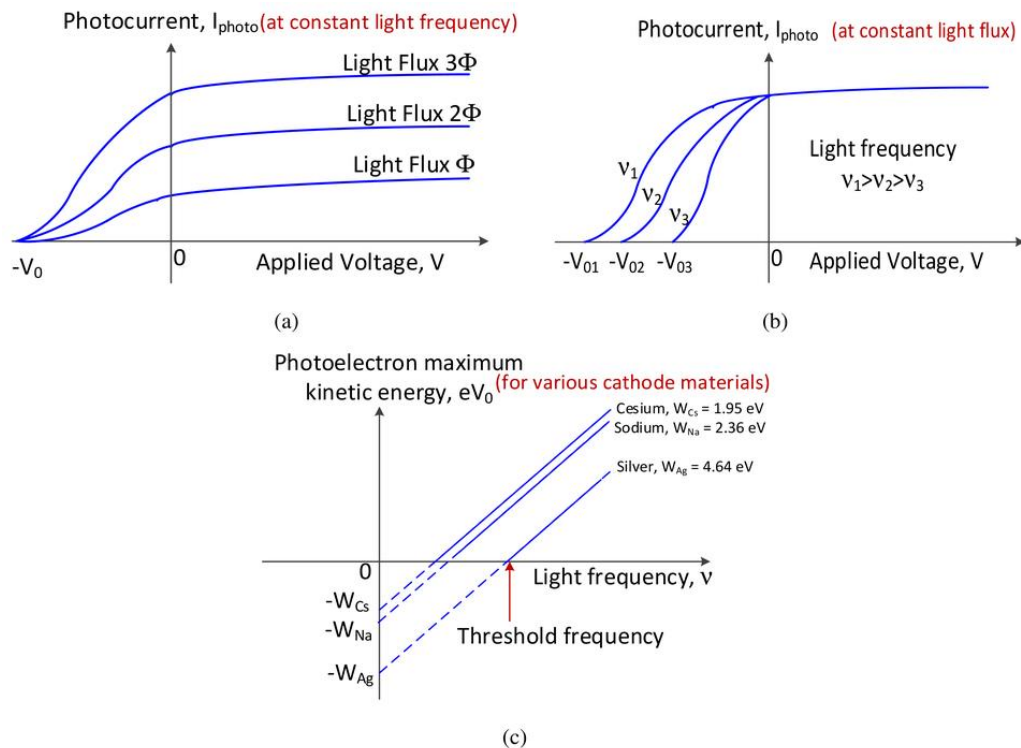
- 4) The maximum kinetic energy ( $KE_{\max}$ ) of the electron emitted from a given surface increases linearly with the frequency radiation as shown in **fig4**.



**Variation of Photoelectric Current with potential for various frequency of incident radiation**

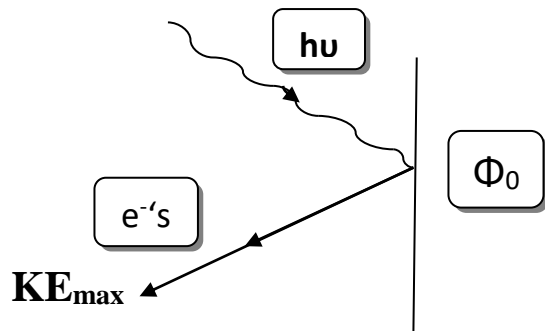
5. The emission of photo electrons from the photo sensitive surface is an instantaneous process.

### Overview of Graph of Photoelectric Effect -various condition.



## Einstein's Theory of Photoelectric Effect

In 1905, Einstein proposed a theory of the photoelectric effect using a concept first put forward by Max Planck that light consists of tiny packets of energy known as **photons** or light quanta. Each packet carries energy  $h\nu$  that is proportional to the frequency  $\nu$  of the corresponding electromagnetic waves. The proportionality constant  $h$  has become known as the *Planck's constant*.



When an electromagnetic radiation of energy ' $h\nu$ ' is incident on a metal surface. It transfers its entire energy to the metal in a single step. The metal consumes this energy to increase the **Work Function  $\Phi_0$**  and the remaining energy remains with  $e^-$  as  $KE_{\max}$

By the Laws of Conservation of Energy:-

$$*h\nu = \phi_0 + KE_{\max} \text{ -----(1)} \quad ; \phi_0 - \text{Work Function}$$

to get the value of  $\phi_0$  we assume incident frequency equal to threshold frequency  
when the incident frequency  $\nu$  decreases, then  $KE_{\max}$  also decreases.

$$* \nu = \nu_0 ; KE_{\max} = 0$$

$$\therefore h\nu_0 = \phi_0 \text{ -----(2)}$$

substituting in eq(1) =>

$$h\nu = h\nu_0 + KE_{\max}$$

$$KE_{\max} = h\nu - h\nu_0$$

Formula for maximum Kinetic energy of the ejected electron;

$$KE_{\max} = h(\nu - \nu_0)$$

;  $\nu \gg \nu_0$  to emit electrons  $\nu$  must be always greater than

$\nu_0$ . If not  $\nu - \nu_0$  becomes negative & KE becomes -ve.

Energy cannot be -ve.

**KE** is +ve ,when  $\nu \gg \nu_0$  and is required for the photoelectric effect to occur. The frequency  $\nu_0$  is the threshold frequency for the given material, Above that frequency the maximum KE of the photoelectrons as well as the stopping

Voltage in the experiment  $V_0 = \left(\frac{h}{e}\right) (\nu - \nu_0)$  rise linearly with the frequency and have no dependence on the number of photons and the intensity of the impinging monochromatic light. Einstein's formula, however explain the phenomenon of the photoelectric effect , and had far reaching consequences in the development of **quantum mechanics**.

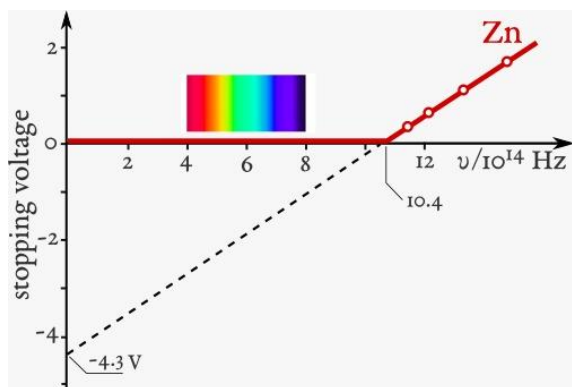


Diagram of the maximum kinetic energy as a function of the frequency of light on zinc.

## Work Function

It is the minimum amount of energy required by an electron to liberate from a metal surface

$$\phi_0 = h\nu_0$$

but,  $\nu = \frac{c}{\lambda}$

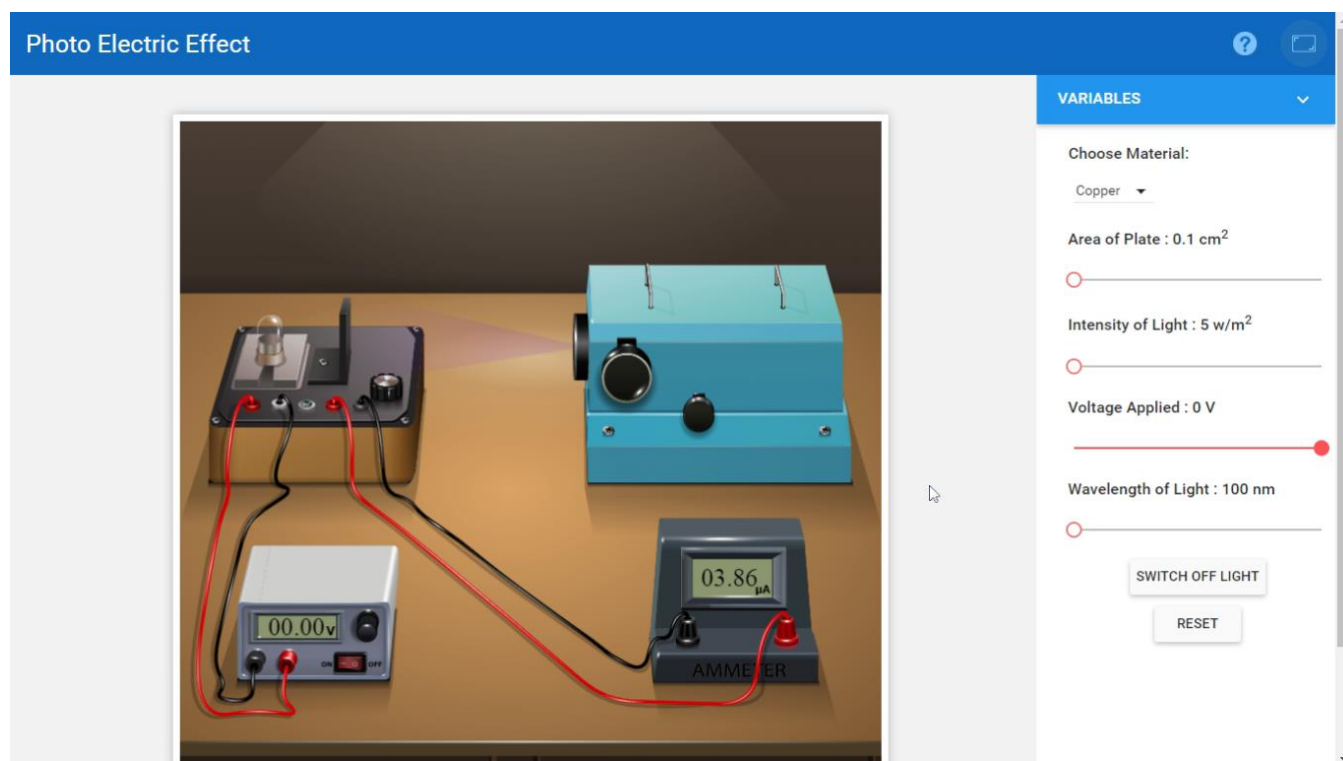
$$\therefore \phi_0 = h \frac{c}{\lambda_0}$$

❖ If work function is more ,then the material is said to be photosensitive. Of all alkali metal poses high photosensitivity.

# METHODOLOGY

## Project Description & Goal .

The measurements of photoelectric current and accelerating potential was done using Amritha V-lab simulations . The experimental set up is used is shown below



V-lab experimental arrangement for photoelectric effect

- **To study the Effect of Intensity of the incident light on the current:**  
By keeping the potential difference between the plates and the frequency of the incident light constant, the photo electric current for various intensities of the incident light.
- **To study the Effect of Potential difference on current:**  
Keeping the frequency and intensity of the incident light constant the potential difference between the electrodes is varied.
- **To study the Effect of frequency on the stopping potential :**  
Adjusting the positive potential (accelerating potential) of the electrode for the saturation current for the given intensity and frequency of the incident radiation. Repeating the experiment for different frequencies for

the same saturation current and in all the cases stopping potentials are determined.

- **To determine the Work Function of given metals experimentally.**  
The work function  $\phi_0$  of the Copper, Zinc and Platinum is determined experimentally – from the plotted graph, calculated using python program (user input) as well by determining the  $\phi_0$  via direct mathematical calculation.

# RESULTS AND DISCUSSION

The observed values of photoelectric potential and accelerating potential for different metals are tabulated below.

## **Metal : COPPER**

### **1). Effect of Frequency.**

**\*Intensity:- 5w/m<sup>2</sup> ; \*Area of plate:- 0.1cm<sup>2</sup>**

**Table.1 Values of photoelectric current and potential for *Different Frequencies*.**

Freq $\nu = 3 \times 10^6$ Hz		Freq $\nu = 2.5 \times 10^6$ Hz		Freq $\nu = 2.14 \times 10^6$ Hz	
Wav $\lambda = 100$ nm		Wav $\lambda = 120$ nm		Wav $\lambda = 140$ nm	
Potential (V)	Current(A)	Potential (V)	Current(A)	Potential (V)	Current(A)
3.8	$1.96 \times 10^{-6}$	1.7	$1.97 \times 10^{-6}$	0.2	$1.98 \times 10^{-6}$
4.3	$1.71 \times 10^{-6}$	2.3	$1.67 \times 10^{-6}$	0.5	$1.83 \times 10^{-6}$
4.8	$1.46 \times 10^{-6}$	2.7	$1.47 \times 10^{-6}$	1.2	$1.48 \times 10^{-6}$
5.3	$1.21 \times 10^{-6}$	3.3	$1.17 \times 10^{-6}$	1.7	$1.23 \times 10^{-6}$
5.8	$0.96 \times 10^{-6}$	3.7	$0.97 \times 10^{-6}$	2.2	$0.98 \times 10^{-6}$
6.3	$0.71 \times 10^{-6}$	4.3	$0.67 \times 10^{-6}$	2.7	$0.73 \times 10^{-6}$
6.8	$0.46 \times 10^{-6}$	4.7	$0.47 \times 10^{-6}$	3.2	$0.48 \times 10^{-6}$
7.3	$0.21 \times 10^{-6}$	5.3	$0.17 \times 10^{-6}$	3.7	$0.23 \times 10^{-6}$
7.8	0	5.7	0	4.2	0

### **2).Effect of Intensity**

**\*Wavelength  $\lambda = 100$ nm ; \*Area of plate :- 0.1cm<sup>2</sup>**

**Table.2 Values of photoelectric current and potential for *Different Intensities***

Intensity : 10w/m <sup>2</sup>		Intensity : 15w/m <sup>2</sup>		Intensity : 20w/m <sup>2</sup>	
Potential (V)	Current(A)	Potential (V)	Current(A)	Potential (V)	Current(A)
0	$7.72 \times 10^{-6}$	0	$11.57 \times 10^{-6}$	0	$15.43 \times 10^{-6}$
1	$6.72 \times 10^{-6}$	1	$10.07 \times 10^{-6}$	1	$13.43 \times 10^{-6}$
2	$5.72 \times 10^{-6}$	2	$8.57 \times 10^{-6}$	2	$11.43 \times 10^{-6}$
3	$4.72 \times 10^{-6}$	3	$7.07 \times 10^{-6}$	3	$9.43 \times 10^{-6}$
4	$3.72 \times 10^{-6}$	4	$5.57 \times 10^{-6}$	4	$7.43 \times 10^{-6}$
5	$2.72 \times 10^{-6}$	5	$4.07 \times 10^{-6}$	5	$5.43 \times 10^{-6}$
6	$1.72 \times 10^{-6}$	6	$2.57 \times 10^{-6}$	6	$3.43 \times 10^{-6}$
7	$0.72 \times 10^{-6}$	7	$1.07 \times 10^{-6}$	7	$1.43 \times 10^{-6}$
8	0	8	0	8	0



### 3).Effect of intensity on current

\*Potential V:- 1V ; \*Frequency  $\nu = 3 \times 10^6$  ; \*Area:-  $0.1 \text{ cm}^2$  ; \*Wavelength  $\lambda = 100 \text{ nm}$

**Table.3 Values of intensity of incident radiation and photocurrent**

Intensity $\text{W/m}^2$	Current ( $\mu\text{A}$ )
5	3.36
10	6.72
15	10.07
20	13.43
25	16.79
30	20.15

### **Plotting the Graph – Computational representation using Python**

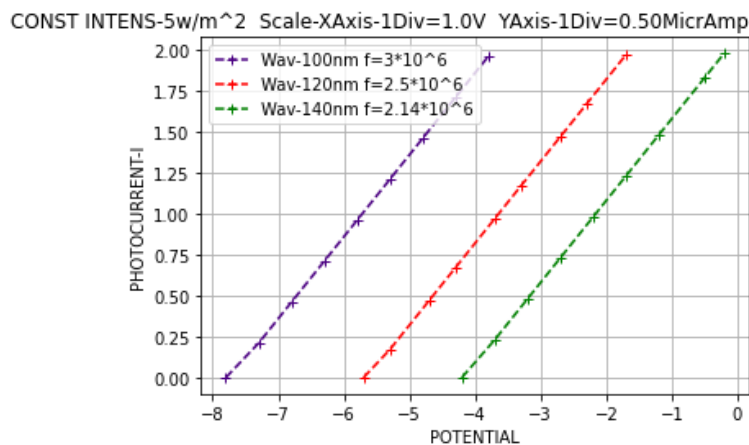
The graphs are plotted with above data computationally by using Python. Using the **matplotlib** in python the visualization of above data is achieved. For that, **Anaconda3** package which consist of comprehensive library for creating static visualization in python-coded in **Jupyter notebook** used for data analysis.

## Metal : COPPER

### 1). Effect of Frequency.

#### Type-1(without extending towards the +ve abscissa)

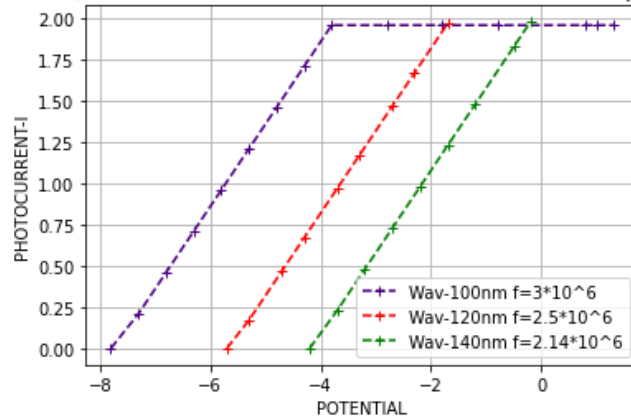
```
1 #GRAPH - 1 COPPER
2 import matplotlib.pyplot as plt
3 # values of x and y axis
4 x1=[-7.8,-7.3,-6.8,-6.3,-5.8,-5.3,-4.8,-4.3,-3.8] #abscissa
5 y1=[0.00,0.21,0.46,0.71,0.96,1.21,1.46,1.71,1.96] #ordinate
6 x2=[-5.7,-5.3,-4.7,-4.3,-3.7,-3.3,-2.7,-2.3,-1.7]
7 y2=[0.00,0.17,0.47,0.67,0.97,1.17,1.47,1.67,1.97]
8 x3=[-4.2,-3.7,-3.2,-2.7,-2.2,-1.7,-1.2,-0.5,-0.2]
9 y3=[0.00,0.23,0.48,0.73,0.98,1.23,1.48,1.83,1.98]
10 plt.plot(x1,y1, linestyle='--', marker='+', color='indigo',label="Wav-100nm f=3*10^6") #+-plus
11 plt.plot(x2,y2, linestyle='--', marker='+', color='red',label="Wav-120nm f=2.5*10^6")
12 plt.plot(x3,y3, linestyle='--', marker='+',color='green',label="Wav-140nm f=2.14*10^6")
13 plt.grid(True) #grid
14 plt.xlabel('POTENTIAL') #abscissa_name
15 plt.ylabel('PHOTOCURRENT-I') #ordinate_name
16 plt.title("CONST INTENS-5w/m^2" " " "Scale-XAxis-1Div=1.0V" " ""YAxis-1Div=0.50MicrAmp",loc='right',rotation=0)
17 plt.legend(loc = "best") #Legend
18 plt.show()
```



#### Type-2(extending towards +ve abscissa-replicating the last value)

```
1 #GRAPH - 1 COPPER
2 import matplotlib.pyplot as plt
3 # values of x and y axis
4 x1=[-7.8,-7.3,-6.8,-6.3,-5.8,-5.3,-4.8,-4.3,-3.8,-2.8,-1.8,-0.8,0.8,1.0,1.3] #abscissa
5 y1=[0.00,0.21,0.46,0.71,0.96,1.21,1.46,1.71,1.96,1.96,1.96,1.96,1.96,1.96,1.96] #ordinate
6 x2=[-5.7,-5.3,-4.7,-4.3,-3.7,-3.3,-2.7,-2.3,-1.7]
7 y2=[0.00,0.17,0.47,0.67,0.97,1.17,1.47,1.67,1.97]
8 x3=[-4.2,-3.7,-3.2,-2.7,-2.2,-1.7,-1.2,-0.5,-0.2]
9 y3=[0.00,0.23,0.48,0.73,0.98,1.23,1.48,1.83,1.98]
10 plt.plot(x1,y1, linestyle='--', marker='+', color='indigo',label="Wav-100nm f=3*10^6") #+-plus
11 plt.plot(x2,y2, linestyle='--', marker='+', color='red',label="Wav-120nm f=2.5*10^6")
12 plt.plot(x3,y3, linestyle='--', marker='+',color='green',label="Wav-140nm f=2.14*10^6")
13 plt.grid(True) #grid
14 plt.xlabel('POTENTIAL') #abscissa_name
15 plt.ylabel('PHOTOCURRENT-I') #ordinate_name
16 plt.title("CONST INTENS-5w/m^2" " " "Scale-XAxis-1Div=1.0V" " ""YAxis-1Div=0.50MicrAmp",loc='right',rotation=0)
17 plt.legend(loc = "best") #Legend
18 plt.show()
```

CONST INTENS-5w/m<sup>2</sup> Scale-XAxis-1Div=1.0V YAxis-1Div=0.50MicrAmp



## 2).Effect of Intensity

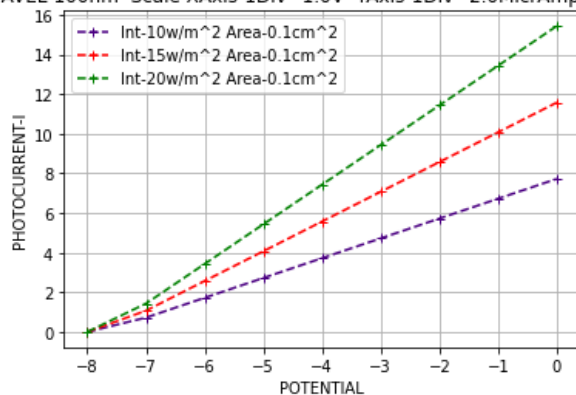
Type-1(without extending towards the +ve abscissa)

```

1 #GRAPH - 2 COPPER
2 import matplotlib.pyplot as plt
3 # values of x and y axis
4 x1=[-8,-7,-6,-5,-4,-3,-2,-1,0]
5 y1=[0.00,0.72,1.72,2.72,3.72,4.72,5.72,6.72,7.72]
6 x2=[-8,-7,-6,-5,-4,-3,-2,-1,0]
7 y2=[0.00,1.07,2.57,4.07,5.57,7.07,8.57,10.07,11.57]
8 x3=[-8,-7,-6,-5,-4,-3,-2,-1,0]
9 y3=[0.00,1.43,3.43,5.43,7.43,9.43,11.43,13.43,15.43]
10 plt.plot(x1,y1, linestyle='--', marker='+', color='indigo',label="Int-10w/m^2 Area-0.1cm^2") #+-plus
11 plt.plot(x2,y2, linestyle='--', marker='+', color='red',label="Int-15w/m^2 Area-0.1cm^2")
12 plt.plot(x3,y3, linestyle='--', marker='+',color='green',label="Int-20w/m^2 Area-0.1cm^2")
13 plt.grid(True)
14 plt.xlabel('POTENTIAL')
15 plt.ylabel('PHOTOCURRENT-I')
16 plt.title("CONST WAVEL-100nm" " " "Scale-XAxis-1Div=1.0V" " " "YAxis-1Div=2.0MicrAmp",loc='right',rotation=0)
17 plt.legend(loc = "best")
18 plt.show()

```

CONST WAVEL-100nm Scale-XAxis-1Div=1.0V YAxis-1Div=2.0MicrAmp

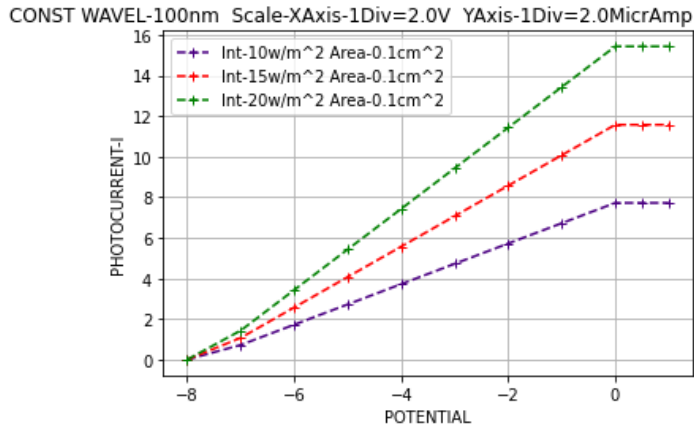


## Type-2(extending towards +ve abscissa-replicating the last value)

```

1 #GRAPH - 2 COPPER
2 import matplotlib.pyplot as plt
3 # values of x and y axis
4 x1=[-8,-7,-6,-5,-4,-3,-2,-1,0,0.5,1]
5 y1=[0.00,0.72,1.72,2.72,3.72,4.72,5.72,6.72,7.72,7.72,7.72]
6 x2=[-8,-7,-6,-5,-4,-3,-2,-1,0,0.5,1]
7 y2=[0.00,1.07,2.57,4.07,5.57,7.07,8.57,10.07,11.57,11.57,11.57]
8 x3=[-8,-7,-6,-5,-4,-3,-2,-1,0,0.5,1]
9 y3=[0.00,1.43,3.43,5.43,7.43,9.43,11.43,13.43,15.43,15.43,15.43]
10 plt.plot(x1,y1, linestyle='--', marker='+', color='indigo',label="Int-10w/m^2 Area-0.1cm^2")
11 plt.plot(x2,y2, linestyle='--', marker='+', color='red',label="Int-15w/m^2 Area-0.1cm^2")
12 plt.plot(x3,y3, linestyle='--', marker='+',color='green',label="Int-20w/m^2 Area-0.1cm^2")
13 plt.grid(True)
14 plt.xlabel('POTENTIAL')
15 plt.ylabel('PHOTOCURRENT-I')
16 plt.title("CONST WAVEL=100nm" " " "Scale-XAxis-1Div=2.0V" " " "YAxis-1Div=2.0MicrAmp",loc='right',rotation=0)
17 plt.legend(loc = "best")
18 plt.show()

```



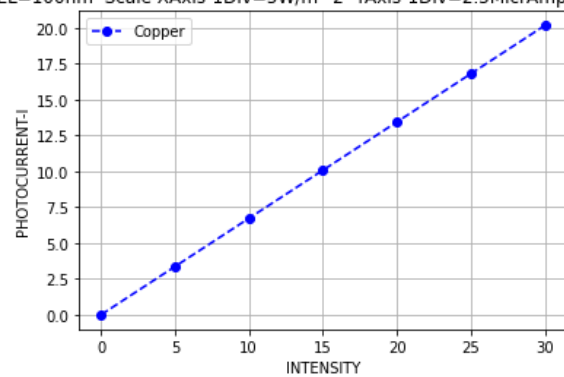
## 3).Effect of intensity on current

```

1 #GRAPH - 3 COPPER
2 import matplotlib.pyplot as plt
3 # values of x and y axis
4 x=[0,5,10,15,20,25,30]
5 y=[0,3.36,6.72,10.07,13.43,16.79,20.15]
6 plt.plot(x,y, linestyle='--', marker='o', color='b',label="Copper")
7 plt.grid(True)
8 plt.xlabel('INTENSITY')
9 plt.ylabel('PHOTOCURRENT-I')
10 plt.title("CONST POTEN=1V & WAVEL=100nm" " " "Scale-XAxis-1Div=5W/m^2" " " "YAxis-1Div=2.5MicrAmp",loc='right',rotation=0)
11 plt.legend(loc = "best")
12 plt.show()

```

CONST POTEN=1V & WAVEL=100nm Scale-XAxis-1Div=5W/m<sup>2</sup> YAxis-1Div=2.5MicrAmp



## Metal : PLATINUM

### 1). Effect of Frequency.

\*Intensity:-  $5\text{w/m}^2$  ; \*Area of plate:-  $0.1\text{cm}^2$

**Table.1** Values of photoelectric current and potential for *Different Frequencies*.

Freq $\nu = 3 \times 10^6 \text{ Hz}$		Freq $\nu = 2.5 \times 10^6 \text{ Hz}$		Freq $\nu = 2.14 \times 10^6 \text{ Hz}$	
Wav $\lambda = 100\text{nm}$		Wav $\lambda = 120\text{nm}$		Wav $\lambda = 140\text{nm}$	
Potential (V)	Current(A)	Potential (V)	Current(A)	Potential (V)	Current(A)
3.6	$1.23 \times 10^{-6}$	1.5	$1.25 \times 10^{-6}$	0	$1.26 \times 10^{-6}$
4.1	$0.98 \times 10^{-6}$	2	$1 \times 10^{-6}$	0.6	$0.96 \times 10^{-6}$
4.6	$0.73 \times 10^{-6}$	2.5	$0.75 \times 10^{-6}$	1.3	$0.61 \times 10^{-6}$
5.1	$0.48 \times 10^{-6}$	3	$0.5 \times 10^{-6}$	1.6	$0.46 \times 10^{-6}$
5.6	$0.23 \times 10^{-6}$	3.5	$0.25 \times 10^{-6}$	1.9	$0.31 \times 10^{-6}$
6.1	0	4	0	2.6	0

### 2).Effect of Intensity

\*Wavelength  $\lambda = 100\text{nm}$  ; \*Area of plate :-  $0.1\text{cm}^2$

**Table.2** Values of photoelectric current and potential for *Different Intensities*

Intensity : $10\text{w/m}^2$		Intensity : $15\text{w/m}^2$		Intensity : $20\text{w/m}^2$	
Potential (V)	Current(A)	Potential (V)	Current(A)	Potential (V)	Current(A)
0	$9.10 \times 10^{-6}$	0	$12.13 \times 10^{-6}$	0	$6.07 \times 10^{-6}$
0.1	$8.95 \times 10^{-6}$	0.1	$11.93 \times 10^{-6}$	0.1	$5.97 \times 10^{-6}$
1.1	$7.45 \times 10^{-6}$	1.1	$9.93 \times 10^{-6}$	1.1	$4.97 \times 10^{-6}$
2.1	$5.95 \times 10^{-6}$	2.1	$7.93 \times 10^{-6}$	2.1	$3.97 \times 10^{-6}$
3.1	$4.45 \times 10^{-6}$	3.1	$5.93 \times 10^{-6}$	3.1	$2.97 \times 10^{-6}$
4.1	$2.95 \times 10^{-6}$	4.1	$3.93 \times 10^{-6}$	4.1	$1.97 \times 10^{-6}$
5.1	$1.45 \times 10^{-6}$	5.1	$1.93 \times 10^{-6}$	5.1	$0.97 \times 10^{-6}$
6.1	0	6.1	0	6.1	0

### 3).Effect of intensity on current

\*Potential V:- 1V ; \*Frequency  $\nu = 3 \times 10^6$  ; \*Area:-  $0.1 \text{ cm}^2$  ; \*Wavelength  $\lambda = 100 \text{ nm}$

Table.3 Values of intensity of incident radiation and photocurrent

Intensity $\text{w/m}^2$	Current ( $\mu\text{A}$ )
5	2.53
10	5.07
15	7.60
20	10.13
25	12.66
30	15.20

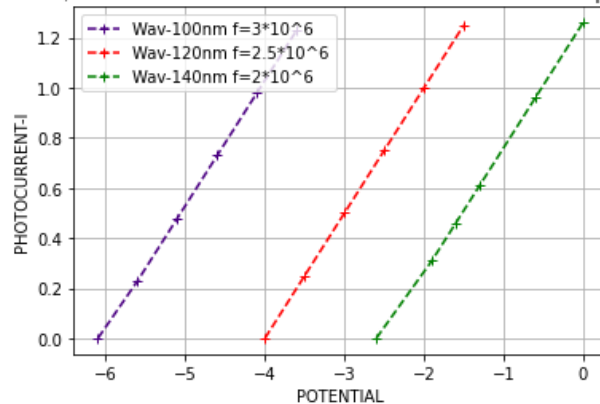
Metal : PLATINUM

#### 1). Effect of Frequency.

Type-1(without extending towards the +ve abscissa)

```
1 #GRAPH - 1 PLATINUM
2 import matplotlib.pyplot as plt
3 # values of x and y axis
4 x1=[-6.1,-5.6,-5.1,-4.6,-4.1,-3.6]
5 y1=[0.00,0.23,0.48,0.73,0.98,1.23]
6 x2=[-4.0,-3.5,-3.0,-2.5,-2.0,-1.5]
7 y2=[0.00,0.25,0.50,0.75,1.00,1.25]
8 x3=[-2.6,-1.9,-1.6,-1.3,-0.6,0.0]
9 y3=[0.00,0.31,0.46,0.61,0.96,1.26]
10 plt.plot(x1,y1, linestyle='--', marker='+', color='indigo',label="Wav-100nm f=3*10^6") #+-plus
11 plt.plot(x2,y2, linestyle='--', marker='+', color='red',label="Wav-120nm f=2.5*10^6")
12 plt.plot(x3,y3, linestyle='--', marker='+',color='green',label="Wav-140nm f=2.14*10^6")
13 plt.grid(True)
14 plt.xlabel('POTENTIAL')
15 plt.ylabel('PHOTOCURRENT-I')
16 plt.title("CONST INTENS-5w/m^2 " " " "Scale-XAxis-1Div=1.0V" " ""YAxis-1Div=0.50MicrAmp",loc='right',rotation=0)
17 plt.legend(loc = "best")
18 plt.show()
```

CONST INTENS-5w/m<sup>2</sup> Scale-XAxis-1Div=1.0V YAxis-1Div=0.50MicrAmp

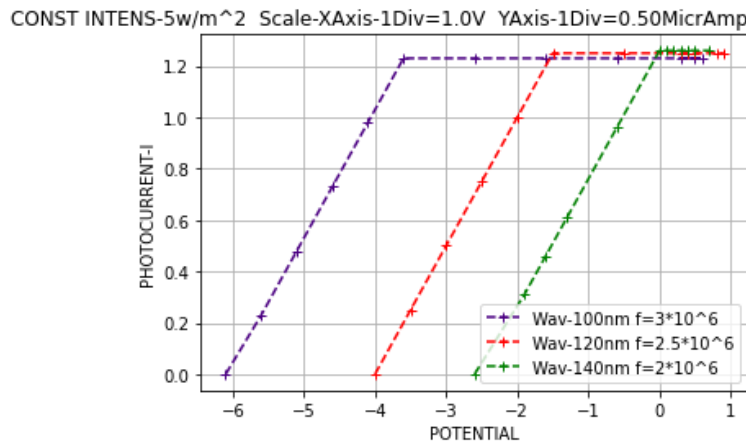


## Type-2(extending towards +ve abscissa-replicating the last value).

```

1 #GRAPH - 1 PLATINUM
2 import matplotlib.pyplot as plt
3 # values of x and y axis
4 x1=[-6.1,-5.6,-5.1,-4.6,-4.1,-3.6,-2.6,-1.6,-0.6,0.6,0.3,0.5]
5 y1=[0.00,0.23,0.48,0.73,0.98,1.23,1.23,1.23,1.23,1.23,1.23,1.23]
6 x2=[-4.0,-3.5,-3.0,-2.5,-2.0,-1.5,-0.5,0.3,0.4,0.5,0.8,0.9]
7 y2=[0.00,0.25,0.50,0.75,1.00,1.25,1.25,1.25,1.25,1.25,1.25,1.25]
8 x3=[-2.6,-1.9,-1.6,-1.3,-0.6,0.0,0.1,0.2,0.3,0.4,0.5,0.7]
9 y3=[0.00,0.31,0.46,0.61,0.96,1.26,1.26,1.26,1.26,1.26,1.26,1.26]
10 plt.plot(x1,y1, linestyle='--', marker='+', color='indigo',label="Wav-100nm f=3*10^6")
11 plt.plot(x2,y2, linestyle='--', marker='+', color='red',label="Wav-120nm f=2.5*10^6")
12 plt.plot(x3,y3, linestyle='--', marker='+',color='green',label="Wav-140nm f=2.14*10^6")
13 plt.grid(True)
14 plt.xlabel('POTENTIAL')
15 plt.ylabel('PHOTOCURRENT-I')
16 plt.title("CONST INTENS-5w/m^2 " " "Scale-XAxis-1Div=1.0V" " "YAxis-1Div=0.50MicrAmp",loc='right',rotation=0)
17 plt.legend(loc = "best")
18 plt.show()

```



## 2).Effect of Intensity

### Type-1(without extending towards the +ve abscissa)

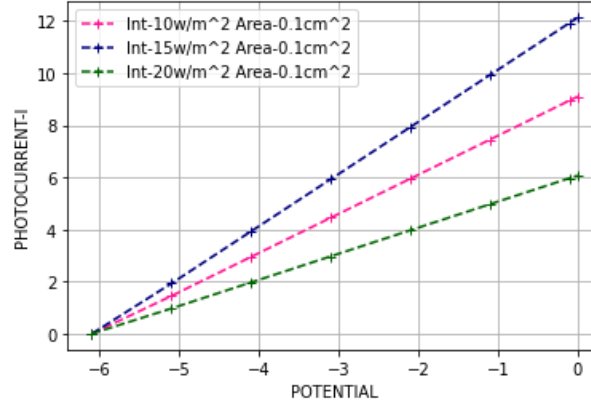
```

1 #GRAPH - 2 PLATINUM
2 import matplotlib.pyplot as plt
3 # values of x and y axis
4 x1=[-6.1,-5.1,-4.1,-3.1,-2.1,-1.1,-0.1,0.0]
5 y1=[0.00,1.45,2.95,4.45,5.95,7.45,8.95,9.10]
6 x2=[-6.1,-5.1,-4.1,-3.1,-2.1,-1.1,-0.1,0.0]
7 y2=[0.00,1.93,3.93,5.93,7.93,9.93,11.93,12.13]
8 x3=[-6.1,-5.1,-4.1,-3.1,-2.1,-1.1,-0.1,0.0]
9 y3=[0.00,0.97,1.97,2.97,3.97,4.97,5.97,6.07]
10 plt.plot(x1,y1, linestyle='--', marker='+', color='deeppink',label="Int-10w/m^2 Area-0.1cm^2")
11 plt.plot(x2,y2, linestyle='--', marker='+', color='navy',label="Int-15w/m^2 Area-0.1cm^2")
12 plt.plot(x3,y3, linestyle='--', marker='+',color='darkgreen',label="Int-20w/m^2 Area-0.1cm^2")
13 plt.grid(True)
14 plt.xlabel('POTENTIAL')
15 plt.ylabel('PHOTOCURRENT-I')
16 plt.title("CONST WAVELENGTH-100nm " " "Scale-XAxis-1Div=1.0V" " "YAxis-1Div=2.0MicrAmp",loc='right',rotation=0)
17 plt.legend(loc = "best")
18 plt.show()

```



CONST WAVEL-100nm Scale-XAxis-1Div=1.0V YAxis-1Div=2.0MicrAmp



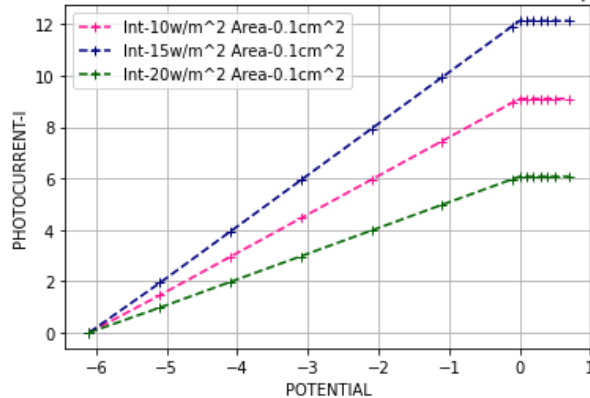
Type-2(extending towards +ve abscissa-replicating the last value)

```

1 #GRAPH - 2 PLATINUM
2 import matplotlib.pyplot as plt
3 # values of x and y axis
4 x1=[-6.1,-5.1,-4.1,-3.1,-2.1,-1.1,-0.1,0.0,0.1,0.2,0.3,0.4,0.5,0.7] #abscissa
5 y1=[0.00,1.45,2.95,4.45,5.95,7.45,8.95,9.10,9.10,9.10,9.10,9.10,9.10,9.10] #ordinate
6 x2=[-6.1,-5.1,-4.1,-3.1,-2.1,-1.1,-0.1,0.0,0.1,0.2,0.3,0.4,0.5,0.7]
7 y2=[0.00,1.93,3.93,5.93,7.93,9.93,11.93,12.13,12.13,12.13,12.13,12.13,12.13,12.13]
8 x3=[-6.1,-5.1,-4.1,-3.1,-2.1,-1.1,-0.1,0.0,0.1,0.2,0.3,0.4,0.5,0.7]
9 y3=[0.00,0.97,1.97,2.97,3.97,4.97,5.97,6.07,6.07,6.07,6.07,6.07,6.07,6.07]
10 plt.plot(x1,y1, linestyle='--', marker='+', color='deeppink',label="Int-10w/m^2 Area-0.1cm^2") #+-plus
11 plt.plot(x2,y2, linestyle='--', marker='+', color='navy',label="Int-15w/m^2 Area-0.1cm^2")
12 plt.plot(x3,y3, linestyle='--', marker='+',color='darkgreen',label="Int-20w/m^2 Area-0.1cm^2")
13 plt.grid(True) #grid
14 plt.xlabel('POTENTIAL') #abscissa_name
15 plt.ylabel('PHOTOCURRENT-I') #ordinate_name
16 plt.title("CONST WAVEL-100nm " " " "Scale-XAxis-1Div=1.0V" " " "YAxis-1Div=2.0MicrAmp",loc='right',rotation=0)
17 plt.legend(loc = "best") #Legend
18 plt.show()

```

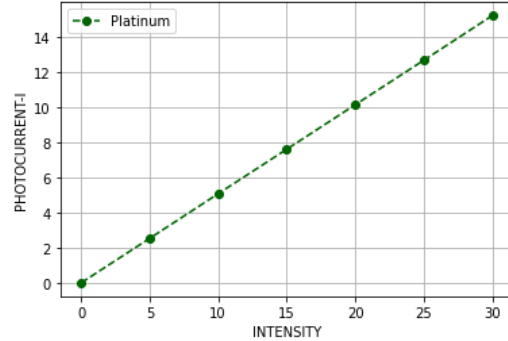
CONST WAVEL-100nm Scale-XAxis-1Div=1.0V YAxis-1Div=2.0MicrAmp



### 3).Effect of intensity on current

```
1 #GRAPH - 3 PLATINUM
2 import matplotlib.pyplot as plt
3 # values of x and y axis
4 x=[0,5,10,15,20,25,30]                                #abscissa
5 y=[0,2.53,5.07,7.60,10.13,12.66,15.20]                #ordinate
6 plt.plot(x,y, linestyle='--', marker='o', color='darkgreen',label="Platinum") #o-round
7 plt.grid(True)                                         #grid
8 plt.xlabel('INTENSITY')                                #abscissa_name
9 plt.ylabel('PHOTOCURRENT-I')                           #ordinate_name
10 plt.title("CONST POTEN=1V & WAVEL=100nm" " " "Scale-XAxis-1Div=5W/m^2" " ""YAxis-1Div=2.5MicrAmp",loc='right',rotation=0)
11 plt.legend(loc = "best")                               #Legend
12 plt.show()
```

CONST POTEN=1V & WAVEL=100nm Scale-XAxis-1Div=5W/m<sup>2</sup> YAxis-1Div=2.5MicrAmp



## Metal : ZINC

### 1). Effect of Frequency.

\*Intensity:-  $5\text{w/m}^2$  ; \*Area of plate:-  $0.1\text{cm}^2$

**Table.1** Values of photoelectric current and potential for *Different Frequencies*

Freq $\nu = 3 \times 10^6 \text{ Hz}$		Freq $\nu = 2.5 \times 10^6 \text{ Hz}$		Freq $\nu = 2.14 \times 10^6 \text{ Hz}$	
Wav $\lambda = 100\text{nm}$		Wav $\lambda = 120\text{nm}$		Wav $\lambda = 140\text{nm}$	
Potential (V)	Current(A)	Potential (V)	Current(A)	Potential (V)	Current(A)
3.7	$2.21 \times 10^{-6}$	1.6	$2.22 \times 10^{-6}$	0	$2.26 \times 10^{-6}$
4.2	$1.96 \times 10^{-6}$	2.1	$1.97 \times 10^{-6}$	0.6	$1.98 \times 10^{-6}$
4.7	$1.71 \times 10^{-6}$	2.6	$1.72 \times 10^{-6}$	1	$1.78 \times 10^{-6}$
5.2	$1.46 \times 10^{-6}$	3.1	$1.47 \times 10^{-6}$	1.6	$1.48 \times 10^{-6}$
5.7	$1.21 \times 10^{-6}$	3.6	$1.22 \times 10^{-6}$	2	$1.28 \times 10^{-6}$
6.2	$0.96 \times 10^{-6}$	4.1	$0.97 \times 10^{-6}$	2.6	$0.98 \times 10^{-6}$
6.7	$0.71 \times 10^{-6}$	4.6	$0.72 \times 10^{-6}$	3	$0.78 \times 10^{-6}$
7.2	$0.46 \times 10^{-6}$	5.1	$0.47 \times 10^{-6}$	3.6	$0.48 \times 10^{-6}$
7.7	$0.21 \times 10^{-6}$	5.6	$0.22 \times 10^{-6}$	4	$0.28 \times 10^{-6}$
8.2	0	6.1	0	4.6	0

### 2).Effect of Intensity

\*Wavelength  $\lambda = 100\text{nm}$  ; \*Area of plate :-  $0.1\text{cm}^2$

**Table.2** Values of photoelectric current and potential for *Different Intensities*

Intensity : $10\text{w/m}^2$		Intensity : $15\text{w/m}^2$		Intensity : $20\text{w/m}^2$	
Potential (V)	Current(A)	Potential (V)	Current(A)	Potential (V)	Current(A)
1.2	$6.92 \times 10^{-6}$	1.2	$10.37 \times 10^{-6}$	1.2	$13.83 \times 10^{-6}$
2.2	$5.92 \times 10^{-6}$	2.2	$8.87 \times 10^{-6}$	2.2	$11.83 \times 10^{-6}$
3.2	$4.92 \times 10^{-6}$	3.2	$7.37 \times 10^{-6}$	3.2	$9.83 \times 10^{-6}$
4.2	$3.92 \times 10^{-6}$	4.2	$5.87 \times 10^{-6}$	4.2	$7.83 \times 10^{-6}$
5.2	$2.92 \times 10^{-6}$	5.2	$4.37 \times 10^{-6}$	5.2	$5.83 \times 10^{-6}$
6.2	$1.92 \times 10^{-6}$	6.2	$2.87 \times 10^{-6}$	6.2	$3.83 \times 10^{-6}$
7.2	$0.92 \times 10^{-6}$	7.2	$1.37 \times 10^{-6}$	7.2	$1.83 \times 10^{-6}$
8.2	0	8.2	0	8.2	0

### 3).Effect of intensity on current

\*Potential V:- 1V ; \*Frequency  $\nu = 3 \times 10^6$  ; \*Area:-  $0.1 \text{ cm}^2$  ; \*Wavelength  $\lambda = 100 \text{ nm}$

Table.3 Values of intensity of incident radiation and photocurrent

Intensity $\text{w/m}^2$	Current ( $\mu\text{A}$ )
5	3.56
10	7.12
15	10.67
20	14.23
25	17.79
30	21.35

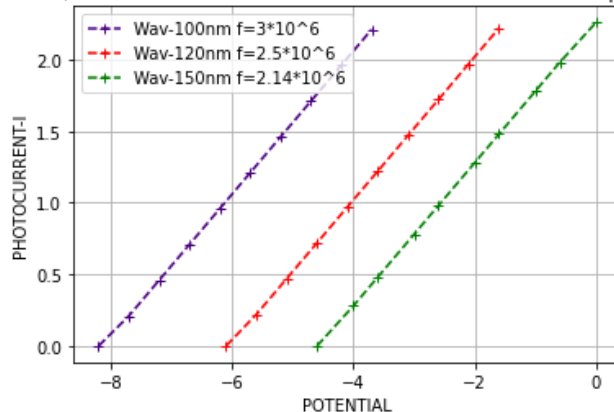
Metal : ZINC

#### 1). Effect of Frequency.

Type-1(without extending towards the +ve abscissa)

```
1 #GRAPH - 1 ZINC
2 import matplotlib.pyplot as plt
3 # values of x and y axis
4 x1=[-8.2,-7.7,-7.2,-6.7,-6.2,-5.7,-5.2,-4.7,-4.2,-3.7]
5 y1=[0.00,0.21,0.46,0.71,0.96,1.21,1.46,1.71,1.96,2.21]
6 x2=[-6.1,-5.6,-5.1,-4.6,-4.1,-3.6,-3.1,-2.6,-2.1,-1.6]
7 y2=[0.00,0.22,0.47,0.72,0.97,1.22,1.47,1.72,1.97,2.22]
8 x3=[-4.6,-4,-3.6,-3,-2.6,-2,-1.6,-1,-0.6,0]
9 y3=[0.00,0.28,0.48,0.78,0.98,1.28,1.48,1.78,1.98,2.26]
10 plt.plot(x1,y1, linestyle='--', marker='+', color='indigo',label="Wav-100nm f=3*10^6") #+-plus
11 plt.plot(x2,y2, linestyle='--', marker='+', color='red',label="Wav-120nm f=2.5*10^6")
12 plt.plot(x3,y3, linestyle='--', marker='+',color='green',label="Wav-150nm f=2.14*10^6")
13 plt.grid(True)
14 plt.xlabel('POTENTIAL')
15 plt.ylabel('PHOTOCURRENT-I')
16 plt.title("CONST INTENS-5w/m^2 " " "Scale-XAxis-1Div=1.0V" " ""YAxis-1Div=0.50MicrAmp",loc='right',rotation=0)
17 plt.legend(loc = "best")
18 plt.show()
```

CONST INTENS-5w/m<sup>2</sup> Scale-XAxis-1Div=1.0V YAxis-1Div=0.50MicrAmp

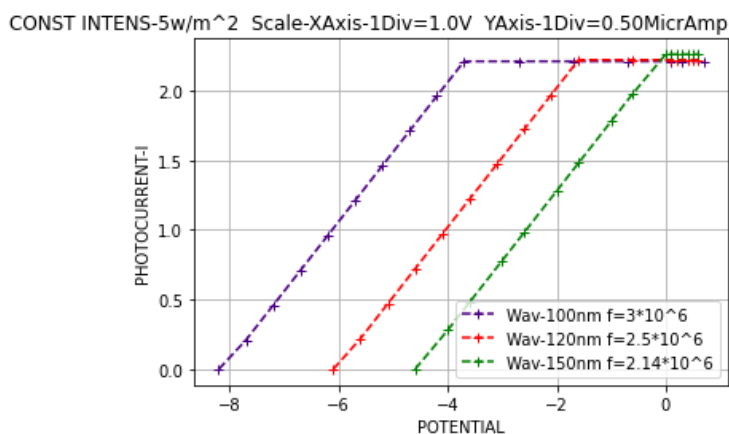


## Type-2(extending towards +ve abscissa-replicating the last value).

```

1 #GRAPH - 1 ZINC
2 import matplotlib.pyplot as plt
3 # values of x and y axis
4 x1=[-8.2,-7.7,-7.2,-6.7,-6.2,-5.7,-5.2,-4.7,-4.2,-3.7,-2.7,-1.7,-0.7,0.1,0.3,0.7] #abscissa
5 y1=[0.00,0.21,0.46,0.71,0.96,1.21,1.46,1.71,1.96,2.21,2.21,2.21,2.21,2.21,2.21,2.21] #ordinate
6 x2=[-6.1,-5.6,-5.1,-4.6,-4.1,-3.6,-3.1,-2.6,-2.1,-1.6,-0.6,0.1,0.2,0.4,0.5,0.6]
7 y2=[0.00,0.22,0.47,0.72,0.97,1.22,1.47,1.72,1.97,2.22,2.22,2.22,2.22,2.22,2.22,2.22]
8 x3=[-4.6,-4,-3.6,-3,-2.6,-2,-1.6,-1,-0.6,0,0.1,0.2,0.3,0.4,0.5,0.6]
9 y3=[0.00,0.28,0.48,0.78,0.98,1.28,1.48,1.78,1.98,2.26,2.26,2.26,2.26,2.26,2.26,2.26]
10 plt.plot(x1,y1, linestyle='--', marker='+', color='indigo',label="Wav-100nm f=3*10^6") #+-plus
11 plt.plot(x2,y2, linestyle='--', marker='+', color='red',label="Wav-120nm f=2.5*10^6")
12 plt.plot(x3,y3, linestyle='--', marker='+',color='green',label="Wav-150nm f=2.14*10^6")
13 plt.grid(True) #grid
14 plt.xlabel('POTENTIAL') #abscissa_name
15 plt.ylabel('PHOTOCURRENT-I') #ordinate_name
16 plt.title("CONST INTENS-5w/m^2 " " " "Scale-XAxis-1Div=1.0V" " " "YAxis-1Div=0.50MicrAmp",loc='right',rotation=0)
17 plt.legend(loc = "best") #Legend
18 plt.show()

```



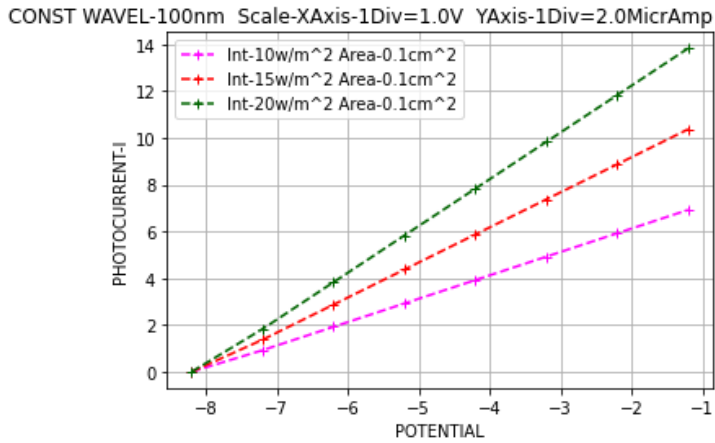
## 2).Effect of Intensity

### Type-1(without extending towards the +ve abscissa)

```

1 #GRAPH - 2 ZINC
2 import matplotlib.pyplot as plt
3 # values of x and y axis
4 x1=[-8.2,-7.2,-6.2,-5.2,-4.2,-3.2,-2.2,-1.2] #abscissa
5 y1=[0.00,0.92,1.92,2.92,3.92,4.92,5.92,6.92] #ordinate
6 x2=[-8.2,-7.2,-6.2,-5.2,-4.2,-3.2,-2.2,-1.2]
7 y2=[0.00,1.37,2.87,4.37,5.87,7.37,8.87,10.37]
8 x3=[-8.2,-7.2,-6.2,-5.2,-4.2,-3.2,-2.2,-1.2]
9 y3=[0.00,1.83,3.83,5.83,7.83,9.83,11.83,13.83]
10 plt.plot(x1,y1, linestyle='--', marker='+', color='magenta',label="Int-10w/m^2 Area-0.1cm^2") #+-plus
11 plt.plot(x2,y2, linestyle='--', marker='+', color='red',label="Int-15w/m^2 Area-0.1cm^2")
12 plt.plot(x3,y3, linestyle='--', marker='+',color='darkgreen',label="Int-20w/m^2 Area-0.1cm^2")
13 plt.grid(True) #grid
14 plt.xlabel('POTENTIAL') #abscissa_name
15 plt.ylabel('PHOTOCURRENT-I') #ordinate_name
16 plt.title("CONST WAVEL-100nm" " " "Scale-XAxis-1Div=1.0V" " " "YAxis-1Div=2.0MicrAmp",loc='right',rotation=0)
17 plt.legend(loc = "best") #Legend
18 plt.show()

```

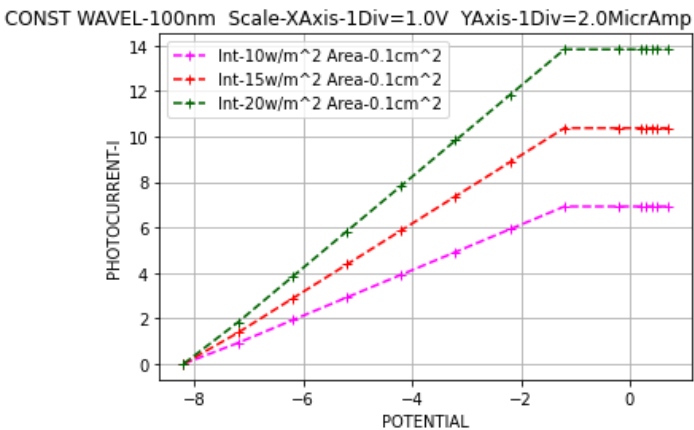


**Type-2(extending towards +ve abscissa-replicating the last value)**

```

1  #GRAPH - 2 ZINC
2  import matplotlib.pyplot as plt
3  # values of x and y axis
4  x1=[-8.2,-7.2,-6.2,-5.2,-4.2,-3.2,-2.2,-1.2,-0.2,0.2,0.3,0.4,0.5,0.7]
5  y1=[0.00,0.92,1.92,2.92,3.92,4.92,5.92,6.92,6.92,6.92,6.92,6.92,6.92,6.92]
6  x2=[-8.2,-7.2,-6.2,-5.2,-4.2,-3.2,-2.2,-1.2,-0.2,0.2,0.3,0.4,0.5,0.7]
7  y2=[0.00,1.37,2.87,4.37,5.87,7.37,8.87,10.37,10.37,10.37,10.37,10.37,10.37,10.37]
8  x3=[-8.2,-7.2,-6.2,-5.2,-4.2,-3.2,-2.2,-1.2,-0.2,0.2,0.3,0.4,0.5,0.7]
9  y3=[0.00,1.83,3.83,5.83,7.83,9.83,11.83,13.83,13.83,13.83,13.83,13.83,13.83,13.83]
10 plt.plot(x1,y1, linestyle='--', marker='+', color='magenta',label="Int-10w/m^2 Area-0.1cm^2")
11 plt.plot(x2,y2, linestyle='--', marker='+', color='red',label="Int-15w/m^2 Area-0.1cm^2")
12 plt.plot(x3,y3, linestyle='--', marker='+',color='darkgreen',label="Int-20w/m^2 Area-0.1cm^2")
13 plt.grid(True)
14 plt.xlabel('POTENTIAL')
15 plt.ylabel('PHOTOCURRENT-I')
16 plt.title("CONST WAVEL-100nm" " " "Scale-XAxis-1Div=1.0V" " ""YAxis-1Div=2.0MicrAmp",loc='right',rotation=0)
17 plt.legend(loc = "best")
18 plt.show()

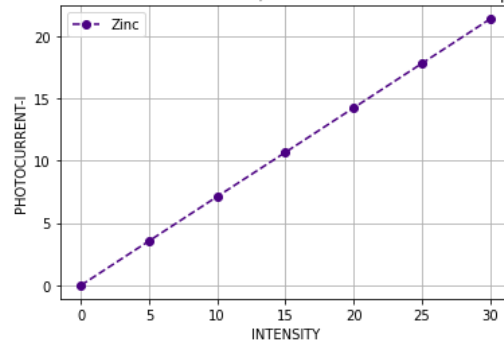
```



### 3).Effect of intensity on current

```
1 #GRAPH - 3 ZINC
2 import matplotlib.pyplot as plt
3 # values of x and y axis
4 x=[0,5,10,15,20,25,30]
5 y=[0,3.56,7.12,10.67,14.23,17.79,21.35]
6 plt.plot(x,y, linestyle='--', marker='o', color='indigo',label="Zinc")
7 plt.grid(True)
8 plt.xlabel('INTENSITY')
9 plt.ylabel('PHOTOCURRENT-I')
10 plt.title("CONST POTEN=1V & WAVEL=100nm" " " "Scale-XAxis-1Div=5W/m^2" " " "YAxis-1Div=2.5MicrAmp",loc='right',rotation=0)
11 plt.legend(loc = "best")
12 plt.show()
```

CONST POTEN=1V & WAVEL=100nm Scale-XAxis-1Div=5W/m<sup>2</sup> YAxis-1Div=2.5MicrAmp



# COPPER

## WORK FUNCTION- $\phi_0$ - Using PYTHON

```
1 # Python
2 #Photoelectric effect- Calculation of WorkFunction
3 #Error Calculation from Theoretical and Calculated Value
4 # Amrutha V-Lab Simulation Result
5
6 #COPPER #COPPER #COPPER #COPPER #COPPER #COPPER
7
8 #Variable declaration
9
10
11 e = 1.6e-19; # Charge on an electron, C
12 h = 6.626e-34; # Planck's constant, Js
13 c = 3.0e+08; # Speed of light in vacuum, m/s
14 W_Theo = 4.70; # Theoretical Value of Work Function of Cu in eV
15 lamb = float(input("Enter the Wavelength (in meter) :")); # Wavelength of incident light (meter)
16 V_0 = float(input("Enter the Stopping potential (-ve) :")); # Stopping potential for emitted electrons, V
17
18
19 #Calculation
20
21 f = c/lamb; # Frequency of incident radiation , Hz
22 E = (h*f); # Energy carried by one photon from Planck's Law, J
23 K_max = (e*V_0); # Maximum kinetic energy of electrons, J
24 # We have, WorkFunction W = E-K_max
25 W_in_joule = ((h*f)-(e*V_0));
26 #Converting to eV, Dividing by e=1.6e-19 to get WorkFunction
27 W_in_eV = (W_in_joule/e)
28
29
30 #Result
31
32
33 print("The work function of Cu metal (in joule) = ")
34 print(W_in_joule , "joule")
35 print("The work function of Cu metal (in eV) = ")
36 print(W_in_eV,"eV")
37
38 #Error Calculation
39 print("Theoretical Value of WorkFunction of Cu:-", W_Theo ,"eV" )
40 print("Calculated Value of WorkFunction of Cu:-",W_in_eV,"eV")
41
42 #Error=(|theoreticalValue-CalculatedValue|)/theoreticalValue
43 Error=(W_Theo-W_in_eV)/(W_Theo)
44 print("ErrorCalculated=",Error)
45
46 #Error%=Error*100
47 Error_Percent = Error*100
48 print("ErrorPercentage=",Error_Percent,"%")
```

Enter the Wavelength (in meter) :100e-09

Enter the Stopping potential (-ve) : 8

## Output

```
Enter the Wavelength (in meter) :100e-09
Enter the Stopping potential (-ve) :8
The work function of Cu metal (in joule) =
7.078e-19 joule
The work function of Cu metal (in eV) =
4.42375 eV
Theoretical Value of WorkFunction of Cu:- 4.7 eV
Calculated Value of WorkFunction of Cu:- 4.42375 eV
ErrorCalculated= 0.05877659574468087
ErrorPercentage= 5.877659574468088 %
```



## WORK FUNCTION - $\phi_0$ - Using Einstein's photoelectric eq<sup>n</sup>.

G.E :-

$$E = K_{\max} + \phi_0$$

Diagram showing the equation  $E = K_{\max} + \phi_0$  with arrows pointing from  $E$  to  $h\nu$  and from  $\phi_0$  to  $eV_0$ .

\* $V_0$ (from graph2)= 8eV

\* $C = 3 \times 10^8 \text{ m/s}$

\* $\lambda = 100 \text{ nm} = 100 \times 10^{-9} \text{ m}$

\* $h = 6.626 \times 10^{-34} \text{ Joule-Sec}$

\* $e = 1.6 \times 10^{-19} \text{ coulomb}$

$$\phi_0 = h\nu - eV_0$$

but  $\nu = c/\lambda$

$$= h \frac{c}{\lambda} - eV_0$$

$$= \frac{(6.626 \times 10^{-34}) \times (3 \times 10^8)}{(100 \times 10^{-9})} - 1.6 \times 10^{-19} \times 8.0$$

$$= 7.078 \times 10^{-19} \text{ Joule}$$

$$= \frac{7.078 \times 10^{-19}}{1.6 \times 10^{-19}} \text{ eV}$$

Converting to eV ,dividing by e

$$= \underline{\underline{4.42375 \text{ eV}}} \quad \text{Calculated Value of Copper -Work Function.}$$

- Theoretical Work Function of Cu = **4.7eV**
- Calculated Work Function of Cu = **4.42375 eV**

• Error =  $\frac{|\text{accepted value} - \text{experiment value}|}{\text{accepted value}}$

$$= \frac{|\text{Theoretical value} - \text{Calculated value}|}{\text{Theoretical value}}$$

$$= \frac{4.70 \text{ eV} - 4.42375 \text{ eV}}{4.70 \text{ eV}} = 0.05877$$

• Error% = Error  $\times 100$  = **5.87%**

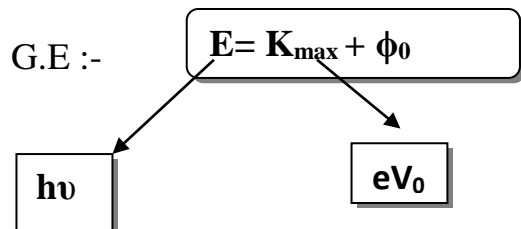
# PLATINUM

## WORK FUNCTION- $\phi_0$ - Using PYTHON

```
1 # Python
2 #Photoelectric effect- Calculation of WorkFunction
3 #Error Calculation from Theoretical and Calculated Value
4 # Amrutha V-Lab Simulation Result
5
6 #PLATINUM #PLATINUM #PLATINUM #PLATINUM #PLATINUM #PLATINUM
7
8 #Variable declaration
9
10
11 e = 1.6e-19; # Charge on an electron, C
12 h = 6.626e-34; # Planck's constant, Js
13 c = 3.0e+08; # Speed of light in vacuum, m/s
14 W_Theo = 6.35; # Theoretical Value of Work Function of Pt in eV
15 lamb = float(input("Enter the Wavelength (in meter) :")); # Wavelength of incident light (meter)
16 V_0 = float(input("Enter the Stopping potential (-ve) :")); # Stopping potential for emitted electrons, V
17
18
19 #Calculation
20
21 f = c/lamb; # Frequency of incident radiation , Hz
22 E = (h*f); # Energy carried by one photon from Planck's Law, J
23 K_max = (e*V_0); # Maximum kinetic energy of electrons, J
24 # We have, WorkFunction W = E-K_max
25 W_in_joule = ((h*f)-(e*V_0));
26 #Converting to eV, Dividing by e=1.6e-19 to get WorkFunction
27 W_in_eV = (W_in_joule/e)
28
29
30 #Result
31
32
33 print("The work function of Pt metal (in joule) = ")
34 print(W_in_joule , "joule")
35 print("The work function of Pt metal (in eV) = ")
36 print(W_in_eV,"eV")
37
38 #Error Calculation
39 print("Theoretical Value of WorkFunction of Pt:-", W_Theo , "eV" )
40 print("Calculated Value of WorkFunction of Pt:-",W_in_eV,"eV")
41
42 #Error=(|theoreticalValue-CalculatedValue|)/theoreticalValue
43 Error=(W_Theo-W_in_eV)/(W_Theo)
44 print("ErrorCalculated=",Error)
45
46 #Error%=Error*100
47 Error_Percent = Error*100
48 print("ErrorPercentage=",Error_Percent,"%")
```

```
Enter the Wavelength (in meter) :100e-09
Enter the Stopping potential (-ve) :6.1
The work function of Pt metal (in joule) =
1.0118e-18 joule
The work function of Pt metal (in eV) =
6.32375 eV
Theoretical Value of WorkFunction of Pt:- 6.35 eV
Calculated Value of WorkFunction of Pt:- 6.32375 eV
ErrorCalculated= 0.004133858267716412
ErrorPercentage= 0.4133858267716412 %
```

## WORK FUNCTION - $\phi_0$ - Using Einstein's photoelectric eq<sup>n</sup>.



\* $V_0$ (from graph2)= **6.1eV**

\* $C = 3 \times 10^8 \text{ m/s}$

\* $\lambda = 100 \text{ nm} = 100 \times 10^{-9} \text{ m}$

\* $h = 6.626 \times 10^{-34} \text{ Joule-Sec}$

\* $e = 1.6 \times 10^{-19} \text{ coulomb}$

$$\phi_0 = h\nu - eV_0$$

but  $\nu = c/\lambda$

$$= h \frac{c}{\lambda} - eV_0$$

$$= \frac{(6.626 \times 10^{-34}) \times (3 \times 10^8)}{(100 \times 10^{-9})} - 1.6 \times 10^{-19} \times 6.1$$

$$= 1.0118 \times 10^{-18} \text{ Joule}$$

$$= \frac{1.0118 \times 10^{-18}}{1.6 \times 10^{-19}} \text{ eV}$$

$$= \underline{\underline{6.32375 \text{ eV}}} \quad \text{Calculated Value of Platinum -Work Function.}$$

Converting to eV ,dividing by e

- Theoretical Work Function of Pt= **6.35eV**
- Calculated Work Function of Pt = **6.32375 eV**

• Error =  $\frac{|\text{accepted value} - \text{experiment value}|}{\text{accepted value}}$

$$= \frac{|\text{Theoretical value} - \text{Calculated value}|}{\text{Theoretical value}}$$

$$= \frac{6.35 \text{ eV} - 6.32375 \text{ eV}}{6.35 \text{ eV}} = 0.004133858268$$

• Error% = Error  $\times 100$  = **0.41%**

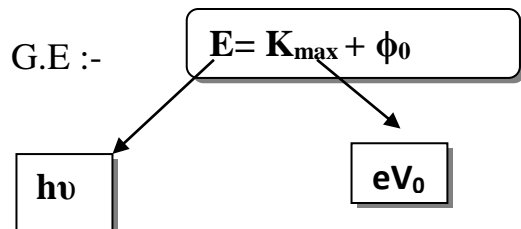
# ZINC

## WORK FUNCTION- $\phi_0$ - Using PYTHON

```
1 # Python
2 #Photoelectric effect- Calculation of WorkFunction
3 #Error Calculation from Theoretical and Calculated Value
4 # Amrutha V-lab Simulation Result
5
6 #ZINC #ZINC #ZINC #ZINC #ZINC #ZINC
7
8 #Variable declaration
9
10
11 e = 1.6e-19; # Charge on an electron, C
12 h = 6.626e-34; # Planck's constant, Js
13 c = 3.0e+08; # Speed of light in vacuum, m/s
14 W_Theo = 4.31; # Theoretical Value of Work Function of Zn in eV
15 lamb = float(input("Enter the Wavelength (in meter) :")); # Wavelength of incident light (meter)
16 V_0 = float(input("Enter the Stopping potential (-ve) :")); # Stopping potential for emitted electrons, V
17
18
19 #Calculation
20
21 f = c/lamb; # Frequency of incident radiation , Hz
22 E = (h*f); # Energy carried by one photon from Planck's Law, J
23 K_max = (e*V_0); # Maximum kinetic energy of electrons, J
24 # We have, WorkFunction W = E-K_max
25 W_in_joule = ((h*f)-(e*V_0));
26 #Converting to eV, Dividing by e=1.6e-19 to get WorkFunction
27 W_in_eV = (W_in_joule/e)
28
29
30 #Result
31
32
33 print("The work function of Zn metal (in joule) = ")
34 print(W_in_joule , "joule")
35 print("The work function of Zn metal (in eV) = ")
36 print(W_in_eV,"eV")
37
38 #Error Calculation
39 print("Theoretical Value of WorkFunction of Zn:-", W_Theo ,"eV" )
40 print("Calculated Value of WorkFunction of Zn:-",W_in_eV,"eV")
41
42 #Error=(|theoreticalValue-CalculatedValue|)/theoreticalValue
43 Error=(W_Theo-W_in_eV)/(W_Theo)
44 print("ErrorCalculated=",Error)
45
46 #Error%=Error*100
47 Error_Percent = Error*100
48 print("ErrorPercentage=",Error_Percent,"%")
```

```
Enter the Wavelength (in meter) :100e-09
Enter the Stopping potential (-ve) :8.2
The work function of Zn metal (in joule) =
6.758000000000001e-19 joule
The work function of Zn metal (in eV) =
4.223750000000001 eV
Theoretical Value of WorkFunction of Zn:- 4.31 eV
Calculated Value of WorkFunction of Zn:- 4.223750000000001 eV
ErrorCalculated= 0.020011600928073976
ErrorPercentage= 2.001160092807398 %
```

## WORK FUNCTION - $\phi_0$ - Using Einstein's photoelectric eq<sup>n</sup>.



\* $V_0$ (from graph2)= **8.2eV**

\* $C = 3 \times 10^8 \text{ m/s}$

\* $\lambda = 100 \text{ nm} = 100 \times 10^{-9} \text{ m}$

\* $h = 6.626 \times 10^{-34} \text{ Joule-Sec}$

\* $e = 1.6 \times 10^{-19} \text{ coulomb}$

$\phi_0 = h\nu - eV_0$

but  $\nu = c/\lambda$

$= h \frac{c}{\lambda} - eV_0$

$= \frac{(6.626 \times 10^{-34}) \times (3 \times 10^8)}{(100 \times 10^{-9})} - 1.6 \times 10^{-19} \times 8.2$

$= 6.758 \times 10^{-19} \text{ Joule}$

$= \frac{6.758 \times 10^{-19}}{1.6 \times 10^{-19}} \text{ eV}$

$= 4.22375 \text{ eV}$     Calculated Value of Zinc -Work Function.

Converting to eV ,dividing by e

- Theoretical Work Function of Zn= **4.31eV**
- Calculated Work Function of Zn = **4.22375 eV**

• Error =  $\frac{|\text{accepted value} - \text{experiment value}|}{\text{accepted value}}$

$= \frac{|\text{Theoretical value} - \text{Calculated value}|}{\text{Theoretical value}}$

$= \frac{4.31 \text{ eV} - 4.22375 \text{ eV}}{4.31 \text{ eV}} = 0.02001160093$

• Error% = Error  $\times 100$  = **2.0011%**

## **Overview of Work Function**

The Work Function of the 3 metals viz, **Cu , Pt ,Zn** are calculated by knowing the  $V_0$  stopping potential.

The results are determined by both scientific computations – using **Python** program as well as by mathematical formulation- using **Einstein's Photoelectric Equation**.

### **1. COPPER**

- ◆ Theoretical Work Function = **4.7eV**
- ◆ Calculated Work Function = **4.42375eV**
- ◆ Error% = **5.87%**

### **2. PLATINUM**

- Theoretical Work Function = **6.35eV**
- Calculated Work Function = **6.32375eV**
- Error% = **0.41%**

### **3. ZINC**

- Theoretical Work Function = **4.31eV**
- Calculated Work Function = **4.22375eV**
- Error% = **2.0011%**

## **CONCLUSION**

The values of Work Function were calculated using Einstein's Photoelectric Equation. The values obtained using data from Virtual- lab simulation is in slight disagreement with the typical values in literature. Its accuracy is calculated by means of Error percentage.

\*\*\*\*\*

# **REFERENCES**

## **Bibliography:**

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2. [en.wikipedia.org/wiki/Photoelectric\\_effect](http://en.wikipedia.org/wiki/Photoelectric_effect)
3. <http://www.youtube.com/watch?v=UWT-bgYRh7g>
4. <http://www.youtube.com/watch?v=ITxs1usXSmk&feature=related>
5. [https://phys.libretexts.org/Bookshelves/University\\_Physics/Book%3A\\_University\\_Physics\\_\(OpenStax\)/Book%3A\\_University\\_Physics\\_III\\_-\\_Optics\\_and\\_Modern\\_Physics\\_\(OpenStax\)/06%3A\\_Photons\\_and\\_Matter\\_Waves/6.03%3A\\_Photoelectric\\_Effect](https://phys.libretexts.org/Bookshelves/University_Physics/Book%3A_University_Physics_(OpenStax)/Book%3A_University_Physics_III_-_Optics_and_Modern_Physics_(OpenStax)/06%3A_Photons_and_Matter_Waves/6.03%3A_Photoelectric_Effect)



## **APPENDIX-1**

### **Github- repository**

The python code included in the report, for all the metal's different graph and its Work Functions are hosted in **Github** – coding platform.

Pull request is allowed to discuss changes in repositories with the owner for collaborator.

- ❖ Repository               :-Photoelectric-Effect--Graph-and-Work-Function--PYTHON-CODE
  - ❖ Web URL(HTTPS):- <https://github.com/FoxTrot-dev/Photoelectric-Effect--Graph-and-Work-Function--PYTHON-CODE.git>
  - ❖ Github CLI               :- gh repo clone FoxTrot-dev/Photoelectric-Effect--Graph-and-Work-Function--PYTHON-CODE
  - ❖ Language                :- Python
  - ❖ Owner                    :- Rithik Vinayaraj
- Id: - FoxTrot-dev©

Nb: - The new code owner and collaborator-individuals are responsible for the section Of new code, if you make a pull request on that code.  
The Owner of the Github retain all the Copyright © of the code