

# Homework 1

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## 1 Problem 4.2

Use separation of variable in *cartesian* coordinates to solve the infinite *cubical* well (or particle in a box):

$$V(x, y, z) = \begin{cases} 0, & \forall x, y, z \in [0, a] \\ \infty, & \forall x, y, z \notin [0, a] \end{cases}$$

1. Find the stationary states, and the corresponding energies.
2. Call the distinct energies  $E_1, E_2, \dots$  in order of increasing energy. Find  $E_1, E_2, E_3, E_4, E_5$  and  $E_6$ . Determine their degeneracies (that is, the number of different states that share the same energy).
3. What is the degeneracy of  $E_{14}$ , and why is this case interesting?

### Solution 1: Stationary states

To find the stationary states of the infinite cubical well, we are going to solve the time independent Schrödinger equation,

$$-\frac{\hbar^2}{2m} \nabla^2 \psi = E\psi, \quad \forall x, y, z \in [0, a],$$

with the following boundary conditions  $\psi(0, 0, 0) = \psi(a, a, a) = 0$ . To solve the equation we are going to use the method of separation of variables, that is, that we assume that the solution of the differential equation has the following form  $\psi(x, y, z) = X(x)Y(y)Z(z)$ . Substituting this solution to the differential equation, we can perform

some algebraic manipulation,

$$\begin{aligned}
-\frac{\hbar^2}{2m}\nabla^2\psi &= E\psi \\
\left(\frac{\partial^2}{\partial x^2} + \frac{\partial^2}{\partial y^2} + \frac{\partial^2}{\partial z^2}\right)X(x)Y(y)Z(z) &= -\frac{2m}{\hbar^2}EX(x)Y(y)Z(z) \\
Y(y)Z(z)\frac{\partial^2}{\partial x^2}X(x) + X(x)Z(z)\frac{\partial^2}{\partial y^2}Y(y) + X(x)Y(y)\frac{\partial^2}{\partial z^2}Z(z) &= -\frac{2m}{\hbar^2}EX(x)Y(y)Z(z) \\
\frac{1}{X(x)}\frac{\partial^2}{\partial x^2}X(x) + \frac{1}{Y(y)}\frac{\partial^2}{\partial y^2}Y(y) + \frac{1}{Z(z)}\frac{\partial^2}{\partial z^2}Z(z) &= -\frac{2m}{\hbar^2}E.
\end{aligned}$$

Now we can re-write this partial differential equation into three differential equations assuming that  $E = \frac{\hbar^2}{2m}(k_x^2 + k_y^2 + k_z^2)$ ,

$$\begin{aligned}
\frac{d^2X(x)}{dx^2} &= -k_x^2X(x) \rightarrow X(x) = A_x \sin[k_x x] + B_x \cos[k_x x], \\
\frac{d^2Y(y)}{dy^2} &= -k_y^2Y(y) \rightarrow Y(y) = A_y \sin[k_y y] + B_y \cos[k_y y], \\
\frac{d^2Z(z)}{dz^2} &= -k_z^2Z(z) \rightarrow Z(z) = A_z \sin[k_z z] + B_z \cos[k_z z].
\end{aligned}$$

In order to find the expression for the coefficients  $A_n$ ,  $B_n$  and  $k_n$ , we start by applying the boundary conditions. Since sin and cos are periodic functions, they satisfy  $f(0) = f(a)$ , however only the sin function satisfy the condition  $f(0) = f(a) = 0$ , hence, we set  $B_x = B_y = B_z = 0$  leading to,

$$X(x) = A_x \sin[k_x x], \quad Y(y) = A_y \sin[k_y y], \quad Z(z) = A_z \sin[k_z z].$$

Now we recall the fact that  $x, y$  and  $z$  have units of distance and that the argument of the sin function must be dimensionless, combining this restriction with the property of periodicity we can define the constants  $k_n$  as,  $k_x = n_x\pi/a$ ,  $k_y = n_y\pi/a$ ,  $k_z = n_z\pi/a$ , where  $(n_x, n_y, n_z) \in \mathbb{Z}^+$ . With this information we can re-write the solution as,

$$\psi(x, y, z) = A_x A_y A_z \sin\left[\frac{n_x\pi}{a}x\right] \sin\left[\frac{n_y\pi}{a}y\right] \sin\left[\frac{n_z\pi}{a}z\right],$$

with

$$E = \frac{\pi^2 \hbar^2}{2ma^2}(n_x^2 + n_y^2 + n_z^2), \quad (n_x, n_y, n_z) \in \mathbb{Z}^+.$$

Finally, in order to get the expression for  $A_x, A_y$  and  $A_z$  we apply the normalization restriction to each spatial dimension,

$$\int_0^a A_l^2 \sin^2\left[\frac{n_l\pi}{a}s\right] ds = A_l^2 \frac{a}{4} \left(2 - \frac{1}{\pi n} \sin[2\pi n]\right) = 1,$$

since  $n \in \mathbb{Z}^+$  we get that  $A_l = \sqrt{2/a}$ , therefore,

$$\psi(x, y, z) = \sqrt{\frac{8}{a^3}} \sin\left[\frac{n_x\pi}{a}x\right] \sin\left[\frac{n_y\pi}{a}y\right] \sin\left[\frac{n_z\pi}{a}z\right], \quad (n_x, n_y, n_z) \in \mathbb{Z}^+$$

**Solution 2: Energy analysis****Solution 3: Energy 14****2 Problem 4.3**

Use

$$P_l^m(x) \equiv (1-x^2)^{|m|/2} \left( \frac{d}{dx} \right)^{|m|} P_l(x)$$

$$P_l(x) \equiv \frac{1}{2^l l!} \left( \frac{d}{dx} \right)^l (x^2-1)^l$$

$$Y_l^m(\theta, \phi) = (-1)^m \sqrt{\frac{(2l+1)(l-|m|)!}{4\pi(l+|m|)!}} e^{im\phi} P_l^m(\cos[\theta])$$

to construct  $Y_0^0$  and  $Y_2^1$ . Check that they are normalized and orthogonal.

**Solution 4: Spherical harmonic**

We start with  $Y_0^0(\theta, \phi)$ ,  $m = l = 0$  substituting those values into the associate Legendre polynomials,

$$P_0(x) \equiv \frac{1}{2^0 0!} \left( \frac{d}{dx} \right)^0 (x^2-1)^0 = 1,$$

and

$$P_0^0(x) \equiv (1-x^2)^{|0|/2} \left( \frac{d}{dx} \right)^{|0|} P_0(x) = 1,$$

hence,

$$Y_0^0(\theta, \phi) = \frac{1}{\sqrt{4\pi}}.$$

Now, to check if it is normalize we integrate in spherical coordinates from  $\theta \in (0, \pi)$  and  $\phi \in (0, 2\pi)$ ,

$$\begin{aligned} \int_0^\pi \int_0^{2\pi} |Y_0^0(\theta, \phi)|^2 \sin \theta d\theta d\phi &= \int_0^\pi \int_0^{2\pi} \frac{1}{4\pi} \sin \theta d\theta d\phi \\ &= \frac{1}{4\pi} \left[ \int_0^\pi \sin \theta d\theta \right] \left[ \int_0^{2\pi} d\phi \right] \\ &= \frac{1}{4\pi} [2] [2\pi] \\ &= 1 \end{aligned}$$

$$\boxed{\int_0^\pi \int_0^{2\pi} |Y_0^0(\theta, \phi)|^2 \sin \theta d\theta d\phi = 1}$$

Now we do the same procedure with  $Y_2^1$ ,  $m = 1$  and  $l = 2$ , which gives that  $P_2^1(x) = \sqrt{1-x^2} \frac{d}{dx} P_2(x)$  and  $P_2(x) = 1/2(3x^2 - 1)$ , hence,

$$P_2(x) \equiv \frac{1}{2^2 2!} \left( \frac{d}{dx} \right)^2 (x^2 - 1)^2 = \frac{1}{2} (3x^2 - 1)$$

and

$$P_2^1(x) \equiv (1-x^2)^{|1|/2} \left( \frac{d}{dx} \right)^{|1|} P_2(x) = 3x\sqrt{1-x^2}$$

$$\begin{aligned} Y_2^1(\theta, \phi) &= -\sqrt{\frac{(2(2)+1)(2-|1|)!}{4\pi(2+|1|)!}} e^{im\phi} P_2^1(\cos[\theta]) \\ &= -\sqrt{\frac{5}{4\pi} \frac{1}{6}} e^{i\phi} 3 \cos[\theta] \sqrt{1-\cos^2[\theta]} = -\sqrt{\frac{5}{24\pi}} e^{i\phi} \sqrt{9} \cos[\theta] \sin[\theta] \\ &= -\sqrt{\frac{15}{8\pi}} e^{i\phi} \cos[\theta] \sin[\theta] \end{aligned}$$

Now we check if the function is normalize,

$$\begin{aligned} \int_0^\pi \int_0^{2\pi} |Y_2^1(\theta, \phi)|^2 \sin \theta d\theta d\phi &= \int_0^\pi \int_0^{2\pi} \frac{15}{8\pi} \cos^2[\theta] \sin^2[\theta] \sin \theta d\theta d\phi \\ &= \frac{15}{8\pi} \left[ \int_0^\pi \cos^2[\theta] \sin^2[\theta] \sin \theta d\theta \right] \left[ \int_0^{2\pi} d\phi \right] \\ &= \frac{15}{8\pi} \left[ \frac{4}{15} \right] [2\pi] \\ &= 1 \end{aligned}$$

$$\int_0^\pi \int_0^{2\pi} |Y_2^1(\theta, \phi)|^2 \sin \theta d\theta d\phi = 1$$

Finally, to check orthogonality we perform the following procedure,

$$\begin{aligned} \int_0^\pi \int_0^{2\pi} [Y_0^0(\theta, \phi)]^* Y_2^1(\theta, \phi) \sin \theta d\theta d\phi &= \int_0^\pi \int_0^{2\pi} \left( \frac{1}{\sqrt{4\pi}} \right)^* \left( -\sqrt{\frac{15}{8\pi}} e^{i\phi} \cos[\theta] \sin[\theta] \right) \sin \theta d\theta d\phi \\ &= -\frac{1}{\sqrt{4\pi}} \sqrt{\frac{15}{8\pi}} \left[ \int_0^\pi \cos[\theta] \sin[\theta] \sin \theta d\theta \right] \left[ \int_0^{2\pi} e^{i\phi} d\phi \right] \\ &= -\sqrt{\frac{15}{32\pi^2}} [0][0] \\ &= 0 \end{aligned}$$

$$\int_0^\pi \int_0^{2\pi} [Y_0^0(\theta, \phi)]^* Y_2^1(\theta, \phi) \sin \theta d\theta d\phi = 0$$

### 3 Problem 4.13

- Find  $\langle r \rangle$  and  $\langle r^2 \rangle$  for an electron in the ground state of hydrogen. Express your answers in terms of the Bohr radius ( $\rho$ ).
- Find  $\langle x \rangle$  and  $\langle x^2 \rangle$  for an electron in the ground state of hydrogen. *Hint:* this requires no new integration-note that  $r^2 = x^2 + y^2 + z^2$ , and exploit the symmetry of the ground state.
- Find  $\langle x^2 \rangle$  in the state  $n = 2, l = 1, m = 1$ . *Warning:* This state is not symmetrical in  $x, y, z$ . Use  $x = r \sin \theta \cos \phi$ .

#### Solution 5: Expected value of position.

By solving the Schrodinger equation in spherical coordinates with the Coulomb's law as the potential energy, the stationary states are in terms of the Bohr's radius,

$$\psi_{(n,m,l)}(r, \theta, \phi) = \frac{1}{r} \rho^{l+1} e^{-\rho} v(\rho) Y_l^m(\theta, \phi),$$

with  $v(\rho)$  being a polynomial of degree  $j_{\max} = n - l - 1$  with coefficients,

$$c_{j+1} = \frac{2(j+l+1-n)}{(j+1)(j+2l+2)} c_j.$$

For the ground state of the Hydrogen atom we set the parameters to

( $n = 1, l = 0, m = 0$ ), which gives,

$$\psi_{1,0,0}(r, \theta, \phi) = \frac{1}{\sqrt{\pi a^3}} e^{-r/a}.$$

To compute the expected value we perform the following operation,

$$\begin{aligned} \langle r \rangle &= \int_V r |\psi_{1,0,0}(r, \theta, \phi)|^2 dV \\ &= \int_0^\infty \int_0^\pi \int_0^{2\pi} r \frac{1}{\pi a^3} e^{-2r/a} r^2 \sin \theta d\theta d\phi dr \\ &= \frac{1}{\pi a^3} \int_0^\infty e^{-2r/a} r^3 dr \int_0^\pi \sin \theta d\theta \int_0^{2\pi} d\phi \\ &= \frac{1}{\pi a^3} \left( \frac{3}{8} a^4 \right) (2) (2\pi) \\ &= \frac{3}{2} a. \end{aligned}$$

Now, for  $\langle r^2 \rangle$ ,

$$\begin{aligned} \langle r^2 \rangle &= \int_V r^2 |\psi_{1,0,0}(r, \theta, \phi)|^2 dV \\ &= \int_0^\infty \int_0^\pi \int_0^{2\pi} r^2 \frac{1}{\pi a^3} e^{-2r/a} r^2 \sin \theta d\theta d\phi dr \\ &= \frac{1}{\pi a^3} \int_0^\infty e^{-2r/a} r^4 dr \int_0^\pi \sin \theta d\theta \int_0^{2\pi} d\phi \\ &= \frac{1}{\pi a^3} \left( \frac{3}{4} a^5 \right) (2) (2\pi) \\ &= 3a^2. \end{aligned}$$

Therefore,

$$\boxed{\langle r \rangle = \frac{3}{2} a, \quad \langle r^2 \rangle = 3a^2}$$

### **Solution 6: Expected values and standard deviation of the $x$ component.**

Recalling the hint,  $r^2 = x^2 + y^2 + z^2$  and using the symmetry of the ground state we can conclude that,

$$\begin{aligned} \langle x^2 \rangle &= \frac{1}{3} \langle r^2 \rangle \\ &= a^2. \end{aligned}$$

On the other hand, for  $\langle x \rangle$  we can write the integrals considering that  $x = r \sin \theta \cos \phi$ ,

$$\begin{aligned}\langle x \rangle &= \int_0^\infty \int_0^\pi \int_0^{2\pi} (r \sin \theta \cos \phi) \frac{1}{\pi a^3} e^{-2r/a} r^2 \sin \theta d\theta d\phi dr \\ &= \frac{1}{\pi a^3} \int_0^\infty e^{-2r/a} r^3 dr \int_0^\pi \sin^2 \theta d\theta \int_0^{2\pi} \cos \phi \\ &= \frac{1}{\pi a^3} \left( \frac{3}{8} a^4 \right) \left( \frac{\pi}{2} \right) (0) \\ &= 0\end{aligned}$$

Hence,

$$\boxed{\langle x \rangle = 0, \quad \langle x^2 \rangle = a^2}$$

### Solution 7: Stationary state ( $n = 2, l = 1, m = 1$ )

For this case we recall the stationary state of the hydrogen atom,

$$\psi_{(n,m,l)}(r, \theta, \phi) = \frac{1}{r} \rho^{l+1} e^{-\rho} v(\rho) Y_l^m(\theta, \phi),$$

with  $v(\rho)$  being a polynomial of degree  $j_{\max} = n - l - 1$  with coefficients,

$$c_{j+1} = \frac{2(j+l+1-n)}{(j+1)(j+2l+2)} c_j.$$

Applying the values of the parameters,

$$\psi_{(2,1,1)}(r, \theta, \phi) = -\frac{1}{\sqrt{\pi a}} \frac{1}{8a^2} r e^{-r/2a} e^{i\phi} \sin \theta.$$

Now we can perform the previous procedures to compute  $\langle x^2 \rangle$ ,

$$\begin{aligned}\langle x^2 \rangle &= \int_V r^2 |\psi_{1,0,0}(r, \theta, \phi)|^2 dV \\ &= \int_0^\infty \int_0^\pi \int_0^{2\pi} (r \sin \theta \cos \phi)^2 \frac{1}{64\pi a^5} r^2 e^{-r/a} \sin^2 \theta r^2 \sin \theta d\theta d\phi dr \\ &= \frac{1}{64\pi a^5} \int_0^\infty r^6 e^{-r/a} dr \int_0^\pi \sin^5 \theta d\theta \int_0^{2\pi} \cos^2 \phi d\phi \\ &= \frac{1}{64\pi a^5} (720 a^7) \left( \frac{16}{15} \right) (\pi) \\ &= 12a^2\end{aligned}$$

Therefore,

$$\boxed{\langle x^2 \rangle = 12a^2}$$

## 4 Problem 4.14

What is the *most probable* value of  $r$ , in the ground state of hydrogen? (The answer is not zero!) *Hint:* First you must figure out the probability that the electron would be found between  $r$  and  $r + dr$ .

### Solution 8: Most probable value of the position in the ground state.

To compute the most probable value of  $r$  we integrate thru space  $|\psi_{1,0,0}(r, \theta, \phi)|^2$  to get the probability density function.

$$\begin{aligned} p(r) &= \int_0^r \int_0^\pi \int_0^{2\pi} |\psi_{1,0,0}(r, \theta, \phi)|^2 r^2 \sin \theta d\theta d\phi dr \\ &= \frac{1}{\pi a^3} \int_0^\infty e^{-2r/a} r^2 dr \int_0^\pi \sin \theta d\theta \int_0^{2\pi} d\phi \\ &= \frac{1}{\pi a^3} () \end{aligned}$$

need to take into account that the probability density for the ground state is given by  $P = 4\pi$ , due to normalization restrictions. Now we can compute the mos probable value we perform the integral thru the radial component, because it has angular symmetry, Since it has angular symmetry we only integrate the radial component,

## 5 Problem 4.23

In problem 4.3 you showed that

$$Y_2^l(\theta, \phi) = -\sqrt{\frac{15}{8\pi}} \sin \theta \cos \theta e^{i\phi}.$$

Apply the raising operator to find  $Y_2^2(\theta, \phi)$ . Use equation  $A_l^m = \hbar \sqrt{l(l+1) - m(m \pm 1)} = \hbar \sqrt{(l \mp m)(l \pm m + 1)}$  to get the normalization.