Homework 1 Professor: Dr. Jaimes

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## 1 Problem 4.2

Use separation of variable in *cartesian* coordinates to solve the infinite *cubical* well (or particle in a box):

$$V(x,y,z) = \begin{cases} 0, & \forall x,y,z \in [0,a] \\ \infty, & \forall x,y,z \notin [0,a] \end{cases}$$

- 1. Find the stationary states, and the corresponding energies.
- 2. Call the distinct energies  $E_1, E_2, \ldots$  in order of increasing energy. Find  $E_1, E_2, E_3, E_4, E_5$  and  $E_6$ . Determine their degeneracies (that is, the number of different states that share the same energy).
- 3. What is the degeneracy of  $E_{14}$ , and why is this case interesting?

### **Solution 1: Stationary states**

To find the stationary states of the infinite cubical well, we are going to solve the time independent Schrödinger equation,

$$-\frac{\hbar^2}{2m}\nabla\psi=E\psi,\ \forall x,y,z\in[0,a],$$

with the following boundary conditions  $\psi(0,0,0) = \psi(a,a,a) = 0$ . To solve the equation we are going to use the method of separation of variables, that is, that we assume that the solution of the differential equation has the following form  $\psi(x,y,z) = X(x)Y(y)Z(z)$ . Substituting this solution to the differential equation, we can perform

some algebraic manipulation,

$$\begin{split} -\frac{\hbar^2}{2m}\nabla\psi &= E\psi\\ \left(\frac{\partial^2}{\partial x^2} + \frac{\partial^2}{\partial y^2} + \frac{\partial^2}{\partial z^2}\right) X(x)Y(y)Z(z) &= -\frac{2m}{\hbar^2} EX(x)Y(y)Z(z)\\ Y(y)Z(z)\frac{\partial^2}{\partial x^2} X(x) + X(x)Z(z)\frac{\partial^2}{\partial y^2} Y(y) + X(x)Y(y)\frac{\partial^2}{\partial z^2} Z(z) &= -\frac{2m}{\hbar^2} EX(x)Y(y)Z(z)\\ \frac{1}{X(x)}\frac{\partial^2}{\partial x^2} X(x) + \frac{1}{Y(y)}\frac{\partial^2}{\partial y^2} Y(y) + \frac{1}{Z(z)}\frac{\partial^2}{\partial z^2} Z(z) &= -\frac{2m}{\hbar^2} E. \end{split}$$

Now we can re-write this partial differential equation into three differential equations assumming that  $E = \frac{\hbar^2}{2m} \left( k_x^2 + k_y^2 + k_z^2 \right)$ ,

$$\frac{d^{2}X(x)}{dx^{2}} = -k_{x}^{2}X(x) \to X(x) = A_{x} \sin [k_{x}x] + B_{x} \cos [k_{x}x],$$

$$\frac{d^{2}Y(y)}{dy^{2}} = -k_{y}^{2}Y(y) \to Y(y) = A_{y} \sin [k_{y}y] + B_{y} \cos [k_{y}y],$$

$$\frac{d^{2}Z(z)}{dz^{2}} = -k_{z}^{2}Z(z) \to Z(z) = A_{z} \sin [k_{z}z] + B_{z} \cos [k_{z}z].$$

In order to find the expression for the coefficients  $A_n$ ,  $B_n$  and  $k_n$ , we start by applying the boundary conditions. Since sin and cos are periodic functions, they satisfy f(0) = f(a), however only the sin function satisfy the condition f(0) = f(a) = 0, hence, we set  $B_x = B_y = B_z = 0$  leading to,

$$X(x) = A_x \sin[k_x x], \quad Y(y) = A_y \sin[k_y y], \quad Z(z) = A_z \sin[k_z z].$$

Now we recall the fact that x, y and z have units of distance and that the argument of the sin function must be dimensionless, combining this restriction with the property of periodicity we can define the constants  $k_n$  as,  $k_x = n_x \pi/a$ ,  $k_y = n_y \pi/a$ ,  $k_z = n_z \pi/a$ , where  $(n_x, n_y, n_z) \in \mathbb{Z}^+$ . With this information we can re-write the solution as,

$$\psi(x, y, z) = A_x A_y A_z \sin\left[\frac{n_x \pi}{a} x\right] \sin\left[\frac{n_y \pi}{a} y\right] \sin\left[\frac{n_z \pi}{a} z\right],$$

with

$$E = \frac{\pi^2 \hbar^2}{2ma^2} \left( n_x^2 + n_y^2 + n_z^2 \right), \quad (n_x, n_y, n_z) \in \mathbb{Z}^+.$$

Finally, in order to get the expression for  $A_x, A_y$  and  $A_z$  we apply the normalization restiction to each spatial dimension,

$$\int_0^a A_l^2 \sin^2 \left[ \frac{n_l \pi}{a} s \right] ds = A_l^2 \frac{a}{4} \left( 2 - \frac{1}{\pi n} \sin \left[ 2\pi n \right] \right) = 1,$$

since  $n \in \mathbb{Z}^+$  we get that  $A_l = \sqrt{2/a}$ , therefore,

$$\psi(x,y,z) = \sqrt{\frac{8}{a^3}} \sin\left[\frac{n_x \pi}{a}x\right] \sin\left[\frac{n_y \pi}{a}y\right] \sin\left[\frac{n_z \pi}{a}z\right], \quad (n_x, n_y, n_z) \in \mathbb{Z}^+$$

### **Solution 2: Energy analysis**

### Solution 3: Energy 14

## 2 Problem 4.3

Use

$$P_l^m(x) \equiv \left(1 - x^2\right)^{|m|/2} \left(\frac{\mathrm{d}}{\mathrm{d}x}\right)^{|m|} P_l(x)$$

$$P_l(x) \equiv \frac{1}{2^l l!} \left(\frac{\mathrm{d}}{\mathrm{d}x}\right)^l \left(x^2 - 1\right)^l$$

$$Y_l^m(\theta, \phi) = \sqrt{\frac{(2l+1)}{4\pi} \frac{(l-|m|)!}{(l+|m|)!}} e^{im\phi} P_l^m(\cos[\theta])$$

to construct  $Y_0^0$  and  $Y_2^1$ . Check that they are normalized and orthogonal.

#### Solution 4: Spherical harmonic

We start with  $Y_0^0(\theta, \phi)$ , m = l = 0 which gives that,  $P_0^0(x) = P_0(x) = 1$ , hence,

$$Y_0^0(\theta,\phi) = \frac{1}{\sqrt{4\pi}}.$$

Now, to check if it is normalize we integrate in spherical coordinates from  $\theta \in (0, \pi)$  and  $\phi \in (0, 2\pi)$ ,

$$\int_0^{\pi} \int_0^{2\pi} \left| Y_0^0(\theta, \phi) \right|^2 \sin \theta d\theta d\phi = \int_0^{\pi} \int_0^{2\pi} \frac{1}{4\pi} \sin \theta d\theta d\phi$$
$$= \frac{1}{4\pi} \left[ \int_0^{\pi} \sin \theta d\theta \right] \left[ \int_0^{2\pi} d\phi \right]$$
$$= \frac{1}{4\pi} [2][2\pi]$$
$$= 1$$

$$\left| \int_0^{\pi} \int_0^{2\pi} \left| Y_0^0(\theta, \phi) \right|^2 \sin \theta d\theta d\phi = 1 \right|$$

Now we do the same procedure with  $Y_2^1$ , m=1 and l=2, which gives that  $P_2^1(x)=\sqrt{1-x^2}\frac{\mathrm{d}}{\mathrm{d}x}P_2(x)$  and  $P_2(x)=1/2(3x^2-1)$ , hence,

$$Y_2^1(\theta, \phi) = \sqrt{\frac{5}{4\pi}} \frac{1}{6} e^{i\phi} \sqrt{1 - x^2} \frac{d}{dx} \frac{1}{2} (3x^2 - 1) (\cos[\theta])$$
$$= \sqrt{\frac{5}{4\pi}} \frac{1}{6} e^{i\phi} \sqrt{1 - x^2} 3x$$

### 3 Problem 4.13

- Find  $\langle r \rangle$  and  $\langle r^2 \rangle$  for an electron in the ground state of hydrogen. Express your answers in terms of the Bohr radius.
- Find  $\langle x \rangle$  and  $\langle x^2 \rangle$  for an electron in the ground stat of hydrogen. *Hint*: this requires no noew integration-note that  $r^2 = x^2 + y^2 + z^2$ , and explot the symmetry of the ground state.
- Find  $\langle x^2 \rangle$  in the state n=2, l=1, m=1. Warning: This state is not symmetrical in x, y, z. Use  $x=r\sin\theta\cos\phi$ .

#### Solution 5

# 4 Problem 4.14

What is the *most probable* value of r, in the ground state of hydrogen? (The answer is not zero!) *Hint:* First ypu must figure out the probability that the electron would be found between r and r + dr.

# 5 Problem 4.23

In problem 4.3 you showed that

$$Y_2^l(\theta,\phi) = -\sqrt{\frac{15}{8\pi}}\sin\theta\cos\theta e^{i\phi}.$$

Apply the raising operator to find  $Y_2^2(\theta,\phi)$ . Use equation  $A_l^m = \hbar\sqrt{l(l+1) - m(m\pm 1)} = \hbar\sqrt{(l\mp m)(l\pm m+1)}$  to get the normalization.

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