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Enhancement and modulation of valley polarization in Janus CrSSe with internal and external electric fields†

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The valley polarization, induced by the magnetic proximity effect, in monolayer transition metal dichalcogenides (TMDCs), has attracted significant attention due to the intriguing fundamental physics. However, the enhancement and modulation of valley polarization for real device applications is still a challenge. Here, using first-principles calculations we investigate the valley polarization properties of monolayer TMDCs CrS₂ and CrSe₂ and how to enhance the valley polarization by constructing Janus CrSSe (with an internal electric field) and modulate the polarization in CrSSe by applying external electric fields. Janus CrSSe exhibits inversion symmetry breaking, internal electric field, spin-orbit coupling, and compelling spin-valley coupling. A magnetic substrate of the MnO₂ monolayer can induce a modest magnetic moment in CrSe₂, CrSe₂, and CrSSe. Notably, the Janus structure with an internal electric field has a much larger valley *p* compared with its non-Janus counterparts. Moreover, the strength of valley polarization can be further modulated by applying external electric fields. These findings suggest that Janus materials hold promise for designing and developing advanced valleytronic devices.

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1 Introduction

Valley, a new degree of freedom, has emerged as a promising avenue with significant potential for information storage, processing, and transmission, offering faster processing speeds and lower energy consumption than conventional electronics. The inherent breaking of spatial inversion symmetry and the strong spin–orbit coupling effect (SOC) in valley materials lead to the disruption of spin degeneracy in the non-equivalent K and K' valleys, 2,3 establishing a unique coupling between spin and valley. Furthermore, breaking the time inversion symmetry allows for the perturbation of carrier transport equilibrium within the non-equivalent K valley, resulting in valley polarization. Given this unique physical property, numerous novel phenomena have been observed experimentally, including the valley Hall effect and valley-dependent optical selection rules, 6,7

leading to the exploration of various device applications, such

polarization. 9,10 However, its semi-metallic nature hinders its integration into valleytronics devices. Unlike graphene, some new valley materials with band gaps have been proposed, such as MoS₂, 11-13 SnS, 9,14 MoSi₂N₄, and TcIrGe₂S₆. 15 The transition metal dichalcogenides (TMDCs) with appropriate band gaps and Dirac-type valleys are suitable for valleytronics devices, which attracted widespread attention. 12,16 Recently, a new member of 2D TMDCs, Janus MoSSe, has been successfully synthesized and has attracted great attention. 17 Janus MoSSe is prepared from MoS₂, on which an S-layer is replaced by Se atoms, breaking the out-ofplane structural symmetry. Such an asymmetric configuration provides additional freedom that plays a crucial role in introducing novel physical properties. 17-23 For instance, ZrSSe exhibits excellent optical properties, 24 and MoSSe displays a strong Rashba effect. 25,26 Additionally, the Janus structures show a strong valley polarization effect and a large valley polarization compared to the common TMDCs.27,28

The essence of utilizing the valley as an information carrier is to eliminate valley degeneracy, thereby inducing valley polarization. In recent years, several methods for generating valley polarization have been proposed, including optical pumping, ^{29–31} breaking the crystal symmetry, ³² applied magnetic fields, ^{33,34} magnetic atom doping, ^{35,36} and the magnetic proximity effect. ^{10,11,13,37–43} Among them, the magnetic proximity effect is a particularly effective and

as valley spin valves and valley filters.^{3,8}
Graphene is the first promising 2D material for valley

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accessible method for introducing large valley polarization in valley-polarized materials. For instance, the magnetic proximity effect of ferromagnetic substrates, such as Fe₃GeTe₂, ¹¹ MnPS₃, ¹⁰ yttrium iron garnet, ⁴³ ScI₂, ⁴² CoO, and MnO, ^{44,45} induces large valley polarization.

In the present work, we employ first-principles calculations to explore the electronic structure and valley polarization properties of transition metal dichalcogenides CrX₂ (X = S, Se) and the Janus structure CrSSe. The results demonstrate that the monolayer Janus structure CrSSe is an exceptional candidate with inherent spin-valley coupling. Moreover, a more significant valley polarization of Janus CrSSe is obtained with an MnO₂ magnetic substrate than that of CrS₂ and CrSe₂ under identical conditions. The monolayer Janus structure CrSSe undergoes evident polarization upon incorporating a magnetic substrate, thus paving the way for a new platform in the advancement of valleytronics.

2 Computational methods

We performed first-principles calculations using the software package Vienna Ab initio Simulation Package (VASP)46 based on density functional theory. The Perdew-Burke-Ernzerhof (PBE) function was selected as the exchange-correlation function. 47,48 To avoid artificial periodic interactions, a vacuum space larger than 25 Å was adopted. The cutoff energy of the plane wave was set to 500 eV. In structural optimization and electronic structure calculations, 15 \times 15 \times 1 and 18 \times 18 \times 1 K-grid densities were used in the first Brillouin zone, respectively. The convergence thresholds for energy and force were set at less than 10⁻⁵ eV and 0.01 eV Å^{-1} , respectively. The DFT-D3 van der Waals correction and Coulomb interaction 49 of $U_{\text{eff}} = 3.9$ eV for Mn 3d orbitals were applied in the CrX₂/MnO₂ heterostructure calculations.⁵⁰ The Berry curvature of monolayer CrSSe was performed by the maximally localized Wannier function method, implemented in the WANNIER90 package.51 The phonon dispersion spectrum was calculated using PHONOPY software based on density functional perturbation theory with a 3 \times 3 \times 1 supercell.⁵²

3 Results and discussion

The optimized atomic structure of Janus CrSSe is shown in Fig. 1(a). Through the calculation, it can be found that CrSSe is a non-magnetic semiconductor like CrS2 and CrSe2.53 The Janus structure is derived from the monolayer CrS₂ by replacing the bottom S atoms with Se. The bond length of Cr-Se (2.421 Å) is slightly longer than that of Cr-S (2.300 Å) in Janus CrSSe. The point group of hexagonal CrS_2 , denoted as D_{3h} transforms the triangular C_{3v} symmetry for Janus CrSSe due to the breaking of mirror symmetry upon Se substitution. The relaxed lattice constants of CrS₂, CrSe₂, and CrSSe are 3.04 Å, 3.21 Å, and 3.12 Å, respectively. As can be seen, the lattice constant of Janus CrSSe is in between CrS₂ and CrSe₂.

Fig. 1b shows the phonon band structure, which has no apparent imaginary frequency in the entire Brillouin zone (BZ),

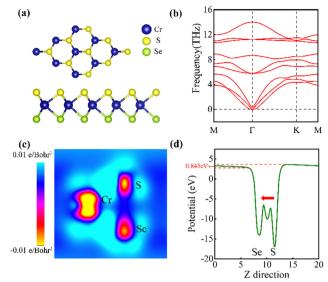


Fig. 1 (a) Schematic crystal structure of the optimized CrSSe, with yellow, green, and blue spheres representing S, Se, and Cr atoms, respectively. (b) Phonon spectrum of monolayer CrSSe. (c) Charge difference density. (d) Electrostatic potential distribution along the Z-direction.

verifying the dynamic stability of Janus CrSSe. Additionally, we calculate the formation energy to evaluate its energetic stability, employing the following formula:54

$$E_{\rm f} = E_{\rm total} - E_{\rm Cr} - E_{\rm S} - E_{\rm Se} \tag{1}$$

where E_{total} is the energy of monolayer CrSSe, and E_{Cr} , E_{Se} , and $E_{\rm S}$ are the energy of single atom Cr, Se, and S, respectively. The calculated formation energies of CrS₂, CrSe₂, and CrSSe are -7.66, -8.05 and -7.43 eV per unit cell, respectively. These results indicate that Janus CrSSe has excellent energy stability as its parents.

Fig. 1c shows the charge density difference of CrSSe on the [110] crystal face. The charge density distributions are noticeably different between the Se and S atomic layers. Specifically, the charge is more concentrated around the S atom than around the Se atom. Furthermore, a Bader charge analysis is performed to provide further insights into the charge transfer process of the Janus CrSSe system. The S atom obtains 0.53 e from the Cr atom, and the Se atom gains 0.39 e from the Cr atom. The trend aligns with the electronegativity sequence of S > Se > Cr. This indicates the occurrence of charge rearrangement in the CrSSe system.

Fig. 1d presents the average electrostatic potential along the Z direction in Janus CrSSe. The red arrow's direction indicates the local electrostatic potential between Cr and Se (S) atoms. The vacuum energy level distributions between the upper and lower surfaces are different, which signifies the work function distribution is different. The lower potential distribution and higher work function on the S atom side reflect the greater electronegativity compared with Se. These results indicate that a built-in electric field from Se to S is established in Janus CrSSe.

Fig. 2 shows the band structures of CrS₂, CrSe₂, and CrSSe with spin-orbit coupling (SOC). CrS2, CrSe2, and CrSSe are non-magnetic **PCCP** Paper

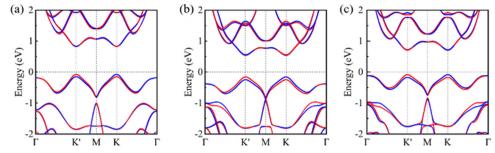


Fig. 2 (a) Energy band diagrams of CrS₂, (b) CrSe₂, and (c) CrSSe considering spin-orbit coupling (SOC), the red and blue lines correspond to spin-up and spin-down

semiconductors that are consistent with previous research results.⁵³ The valence band maximum (VBM) and conduction band minimum (CBM) are located at the K/K' valley, highlighting their direct bandgap semiconducting nature. The band gaps of CrS₂, CrSe₂, and CrSSe are 0.91 eV, 0.70 eV, and 0.79 eV, respectively. Due to the strong spin-orbit coupling, notable spin splitting (69 meV, 90 meV, and 81 meV) appears at the K and K' valleys near the VBM for CrS_2 , $CrSe_2$, and the Janus CrSSe monolayer, respectively. The upper (lower) band is dominated by the spin-down (spin-up) component at the K point and the spin-up (spin-down) component at the K' point, indicating spin-valley coupling. However, the presence of time-reversal symmetry ensures that the energy of the K and K' valleys remains energetically degenerate.

To induce the valley polarization, a magnetic substrate is introduced to provide a magnetic proximity effect. Here, we choose semiconducting MnO₂ as the magnetic substrate, which has been reported as a two-dimensional magnetic semiconductor with intrinsic ferromagnetism.⁵⁵ We constructed a CrSSe/ MnO₂ heterojunction, which breaks the time-reversal symmetry

and allows the manipulation of valley polarization in the CrSSe monolayer. The introduced magnetic substrate produces a distinct energy difference between the K and K' valleys, leading to valley polarization at K and K'. The magnitude of the valley polarization is defined as the energy difference between the extreme of the valence band between K and K' valleys, denoted as $\Delta_{KK'} = E_K - E_{K'}$. Fig. 3 shows the six heterojunction models with different stacking orders when Se atoms are located at the contact interface. Diverse stacking arrangements produced a wide range of valley polarization, ranging from 2 to 80 meV, as listed in the ESI,† Table S1. It is indicated that the stacking configuration directly affects the coupling between magnetic atoms and transition metal ions, significantly influencing valley polarization. When S atoms are located at the contact interface as shown in the ESI,† Fig. S1, the valley polarization ranges from 8 to 23 meV (Fig. S2, ESI†), which is smaller than that for Se at the interface. Hence, we chose the II stacking configuration when Se is close to the interface with the most stable structure. Fig. 4 illustrates the spin projection bands with SOC for the CrS₂/MnO₂, CrSe₂/MnO₂, and CrSSe/MnO₂

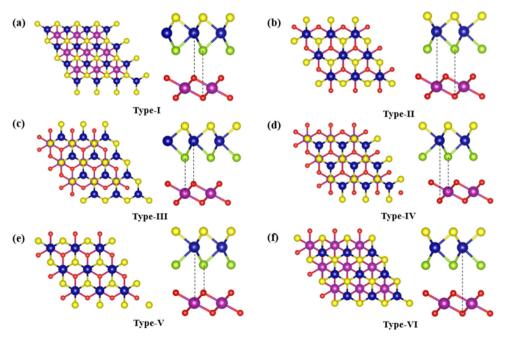
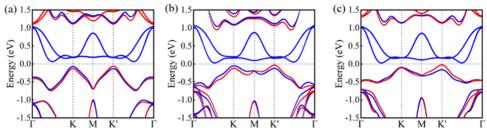


Fig. 3 CrSSe/MnO₂ heterojunction with different stacking orders.





The spin projected band structure of (a) CrS₂/MnO₂, (b) CrSe₂/MnO₂, (c) CrSSe/MnO₂.

heterojunctions. The valley polarization of the initial valence band is 8, 20, and 71 meV, respectively. Hence, valley polarization is significantly amplified in the Janus monolayer.

To unravel the intrinsic mechanism of the enhanced valley polarization in the Janus system, we calculate the charge density difference of CrS₂/MnO₂, CrSe₂/MnO₂, and CrSSe/ MnO₂ heterojunctions to analyse the coupling effect between chromium dichalcogenides and the magnetic substrate. 56,57 As shown in Fig. 5a-c, one can notice that the charges are accumulated on the MnO2 and the TMDC sides, suggesting substantial charge transfer from the TMDCs to the MnO2 substrate. The planar average charge density difference $\rho(z)$ in the vertical direction reveals a significant disparity in charge distribution, as shown in Fig. 5d-f. Notably, MnO2 exhibits a positive charge density, while the TMDCs display a negative charge density, confirming the occurrence of charge transfer between the two layers. Bader charges analysis evidence that charge transfers of 0.044e, 0.086e, and 0.078e occurred from the TMDs to the substrate in the CrS₂/MnO₂, CrSe₂/MnO₂, and CrSSe/MnO₂ systems, respectively. The charge transfer between the Se atom and the S atom in CrSSe/MnO2 is larger than that in CrS₂/MnO₂ and CrSe₂/MnO₂, illustrating that the coupling effect between the Janus CrSSe monolayer and the magnetic substrate MnO₂ is stronger than that in CrS₂/MnO₂ and CrSe₂/ MnO₂. The most robust coupling provides the most substantial magnetic proximity effect, resulting in a distinct improvement in valley polarization.

In fact, the presence of a magnetic substrate not only disrupts the system's inversion symmetry but also triggers weak magnetism within the transition metal atoms.^{39,58} In the CrS₂/MnO₂, CrSe₂/MnO₂, and CrSSe/MnO₂ heterojunctions, the magnetic moments of the Cr ions are $-0.017\mu_{\rm B}$, $-0.03\mu_{\rm B}$ and $0.037\mu_{\rm B}$, respectively. The magnetization of Cr atoms induced by a magnetic substrate can lead to the occurrence of spin splitting in band structures without SOC, as depicted in Fig. 6a-c. In CrSSe/MnO₂, the magnetic substrate triggers a more considerable magnetism in CrSSe, so the spin splitting near VBM is the most obvious in CrSSe/MnO₂ heterojunctions. When taking simultaneously SOC and magnetic proximity effects into consideration, as shown in Fig. 6d, valley polarization will appear in these heterojunctions. In CrSSe/MnO₂ heterojunctions, the most significant spin splitting near VBM induced by the magnetism from the magnetic substrate leads to enhanced valley polarization.

To gain comprehensive insights into the spin-valley locked physical properties at the K and K' valleys for the CrSSe monolayer and the CrSSe/MnO2 heterojunction, we employ the Kubo formula to calculate the Berry curvature, which involves summation from all occupied band contributions:

$$\Omega(k) = -\sum_{n} \sum_{n \neq n'} f(n) \frac{2Im\phi_{nk} |v_x| \phi_{n'k} \phi_{n'k} |v_y| \phi_{nk}}{(E_n - E_{n'})^2}$$
(2)

where n and n' are band structure indices, f(n), $\nu_{x/y}$, φ_{nk} , and $\varphi_{n'k}$ are the Fermi-Dirac distribution functions of the *n*th band

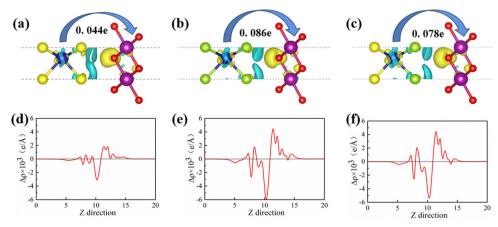


Fig. 5 (a) and (d) CrS₂/MnO₂, (b) and (e) CrSe₂/MnO₂, and (c) and (f) CrSSe/MnO₂ differential charge and planar average charge density difference. The yellow and blue regions represent the areas of charge accumulation and disappearance, respectively

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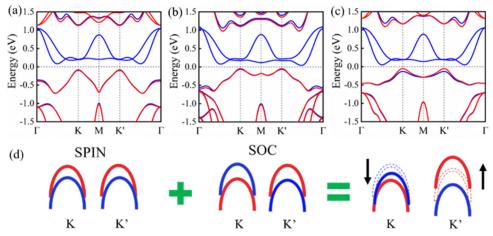


Fig. 6 The spin polarization band of (a) CrS₂/MnO₂, (b) CrSe₂/MnO₂, and (c) CrSSe/MnO₂ without SOC. (d) A schematic diagram describing the contributions to the valley polarization of the CrX₂/MnO₂ system under SOC and magnetic coexistence.

at point k, the Dirac electrons along the x/y direction, and the periodic part of the Bloch wave function with eigenvalues E_n and $E_{n'}$, respectively.

The Berry curvatures of CrSSe and CrSSe/MnO2 along high symmetry points are plotted in Fig. 7a. One can see that the Berry curvature of the CrSSe monolayer is ± 60.19 bohr² at K/K', which is comparable to the Berry curvature of monolayer MoS₂^{14,15} as listed in Table S2 (ESI†). Hence, the absolute values of the K and K' valley are identical. In this case, the time-reversal symmetry requires $\Omega_z(-k) = -\Omega_z(k)$; hence, the Berry curvature $\Omega_z(k)$ is an odd function of the momentum space k. The peaks of Berry curvatures are located at K and K'valleys with equal values but opposite in sign. The Berry curvature of CrSSe/MnO₂ is -32.94/30.83 bohr² at K/K' as shown in Fig. 7a and b. The absolute values of the K and K'valleys are not identical due to the breaking of the time-reversal symmetry along with valley polarization induced by the MnO₂ magnetic substrate. The identical absolute values for Berry curvatures will cause unequal carriers in the K and K' valleys. When doping with a small p-type, holes appear at VBM, which will cause an imbalance of carriers in the K and K' valleys. Then, a valley-polarized current will be generated driven by the bias voltage. Moreover, it is clear that the value of Berry curvature has decreased significantly from CrSSe to CrSSe/MnO2, which is similar to the decrease in MoS₂ and MoS₂/CoO.⁵⁹ The decrease is mainly due to the increase in the bandgap from 0.79 eV for CrSSe to 1.16 eV for CrSSe/MnO₂ as shown in the layer-resolved band structure in Fig. S3 (ESI†).

Finally, previous studies have shown that electric fields can effectively regulate the coupling between heterojunctions.³⁷ It is expected to tune the magnetic proximity effects and valley polarization of the CrSSe/MnO₂ heterojunction through the electric field. Thus, we applied electric fields on the CrSSe/ MnO₂ heterojunction to further investigate the electric field effect on valley polarization of the CrSSe/MnO₂ heterojunction. The electric field is added from -0.6 eV Å^{-1} to 0.6 eV Å^{-1} as shown in Fig. 8. Positive values represent the electric field along the +z direction from the MnO₂ layer to the CrSSe layer, and negative values represent the electric field along the -z direction. When a positive electric field was imposed, the valley polarization diminished as the electric field intensity increased, as depicted in Fig. 8. At a voltage of 0.6 V \mathring{A}^{-1} , the valley polarization splitting decreased to 34 meV. In contrast, upon reversing the electric field direction, the valley polarization surged to 81 meV with an electric field intensity of -0.4 V Å^{-1} . Hence, the electric fields are a practical scheme to tune the valley polarization of heterojunctions. In order to analyse the regulation mechanism of the electric field, we calculated the charge difference and Bader charge analysis. One can notice that the charge transfer changes with the change of electric

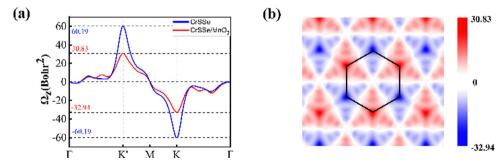


Fig. 7 (a) Berry curvature of Janus CrSSe and CrSSe/MnO₂ along high symmetry points. (b) Berry curvature of Janus CrSSe/MnO₂ in the 2D Brillouin.

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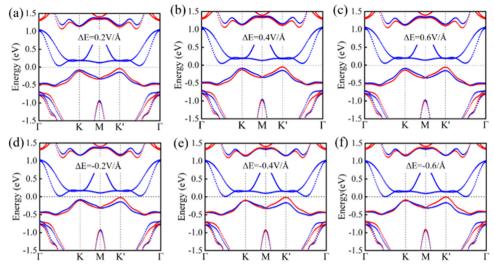


Fig. 8 Spin projection energy bands of CrSSe/MnO₂ heterojunction under different electric field strengths (a) 0.2 V Å^{-1} , (b) 0.4 V Å^{-1} , (c) 0.6 V Å^{-1} , (d) -0.2 V Å^{-1} , (e) -0.2 V Å^{-1} , (f) -0.2 V Å^{-1} .

field as shown in the ESI,† Fig. S4. Hence, the evolution of valley polarization with the external electric field is mainly due to the effective regulation of the interlayer coupling by the electric field, thereby changing the magnetic proximity effect and valley polarization.

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4 Conclusions

In summary, we have investigated the electronic structure and valley-contrasting physics of CrS2, CrSe2, and CrSSe monolayers by first-principles calculations. Our results indicate that CrS₂, CrSe₂, and CrSSe monolayers have intrinsic direct bandgaps of 0.91 eV, 0.70 eV, and 0.79 eV, respectively. The spin-valley coupling in Janus CrSSe is due to the breaking of inversion symmetry and SOC. Furthermore, introducing a magnetic substrate will induce valley polarization because of the magnetic proximity effect. The valley polarization in the CrSSe/MnO₂ heterojunction is more significant than that in CrS₂/MnO₂ and CrS₂/MnO₂ heterojunctions because of the stronger coupling with the magnetic substrate of Janus CrSSe by its high internal electronic field. Furthermore, it was found that the external electric field can further effectively tune the valley polarization splitting. Our results establish a novel foundation for the prospective utilization of valley polarization.

Conflicts of interest

There are no conflicts to declare.

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References

- 1 R. Peng, Y. Ma, X. Xu, Z. He, B. Huang and Y. Dai, Phys. Rev. B, 2020, 102, 035412.
- 2 A. R. Akhmerov and C. W. Beenakker, Phys. Rev. Lett., 2007, 98, 157003.
- 3 A. Rycerz, J. Tworzydło and C. W. J. Beenakker, Nat. Phys., 2007, 3, 172-175.
- 4 L. Ju, L. Wang, X. Li, S. Moon, M. Ozerov, Z. Lu, T. Taniguchi, K. Watanabe, E. Mueller, F. Zhang, D. Smirnov, F. Rana and P. L. McEuen, Nat. Commun., 2020, 11, 2941.
- 5 D. Xiao, G. B. Liu, W. X. Feng, X. D. Xu and W. Yao, Phys. Rev. Lett., 2012, 108, 5.
- 6 H. L. Zeng, J. F. Dai, W. Yao, D. Xiao and X. D. Cui, Nat. Nanotechnol., 2012, 7, 490-493.
- 7 T. Cao, G. Wang, W. P. Han, H. Q. Ye, C. R. Zhu, J. R. Shi, Q. Niu, P. H. Tan, E. Wang, B. L. Liu and J. Feng, Nat. Commun., 2012, 3, 5.
- 8 L. L. Tao and E. Y. Tsymbal, Phys. Rev. B, 2019, 100, 161110.
- 9 M. K. Mohanta, F. Is and A. De Sarkar, Phys. Rev. B, 2023, 107, 035429.
- 10 X.-J. Dong, K. Jia, W.-X. Ji, S.-S. Li and C.-W. Zhang, ACS Appl. Electron. Mater., 2023, 5, 2046-2054.
- 11 J. X. Li, W. Q. Li, S. H. Hung, P. L. Chen, Y. C. Yang, T. Y. Chang, P. W. Chiu, H. T. Jeng and C. H. Liu, Nat. Nanotechnol., 2022, 17, 721-728.
- 12 Z. Lin, Y. Liu, Z. Wang, S. Xu, S. Chen, W. Duan and B. Monserrat, Phys. Rev. Lett., 2022, 129, 027401.

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13 E. C. Castro, D. S. Brandão, H. Bragança, A. S. Martins,

- F. Riche, A. C. Dias, J. H. Zhao, A. L. A. Fonseca and F. Qu, *Phys. Rev. B*, 2023, **107**, 035439.
- 14 F. Is, R. Ahammed, P. Nandi, A. Rawat and A. De Sarkar, *Appl. Surf. Sci.*, 2023, **611**, 155675.
- 15 J. Zhao, T. Zhang, R. Peng, Y. Dai, B. Huang and Y. Ma, J. Phys. Chem. Lett., 2022, 13, 8749–8754.
- 16 Z. Huang, J. Wu, C. Wang, S. Yang and F. Ma, *Phys. Rev. B*, 2021, **104**, L041103.
- 17 J. Zhang, S. Jia, I. Kholmanov, L. Dong, D. Er, W. Chen, H. Guo, Z. Jin, V. B. Shenoy, L. Shi and J. Lou, *ACS Nano*, 2017, 11, 8192–8198.
- 18 J. Yuan, Y. Yang, Y. Cai, Y. Wu, Y. Chen, X. Yan and L. Shen, *Phys. Rev. B*, 2020, **101**, 094420.
- 19 C. Zhang, Y. Nie, S. Sanvito and A. Du, *Nano Lett.*, 2019, 19, 1366–1370.
- 20 C. Luo, X. Peng, J. Qu and J. Zhong, *Phys. Rev. B*, 2020, 101, 245416.
- 21 Y. C. Cheng, Z. Y. Zhu, M. Tahir and U. Schwingenschlögl, *Europhys. Lett.*, 2013, **102**, 5.
- 22 Q.-F. Yao, J. Cai, W.-Y. Tong, S.-J. Gong, J.-Q. Wang, X. Wan, C.-G. Duan and J. H. Chu, *Phys. Rev. B*, 2017, **95**, 165401.
- 23 A. Kandemir and H. Sahin, *Phys. Chem. Chem. Phys.*, 2018, 20, 17380–17386.
- 24 T. V. Vu, H. D. Tong, D. P. Tran, N. T. T. Binh, C. V. Nguyen, H. V. Phuc, H. M. Do and N. N. Hieu, RSC Adv., 2019, 9, 41058–41065.
- 25 H. U. Din, M. Idrees, A. Albar, M. Shafiq, I. Ahmad, C. V. Nguyen and B. Amin, *Phys. Rev. B*, 2019, **100**, 165425.
- 26 T. Hu, F. Jia, G. Zhao, J. Wu, A. Stroppa and W. Ren, *Phys. Rev. B*, 2018, **97**, 125401.
- 27 W. Y. Tong, S. J. Gong, X. Wan and C. G. Duan, *Nat. Commun.*, 2016, 7, 13612.
- 28 X.-W. Shen, W.-Y. Tong, S.-J. Gong and C.-G. Duan, *2D Mater.*, 2017, 5, 011001.
- 29 Y. T. Wang, C. W. Luo, A. Yabushita, K. H. Wu, T. Kobayashi, C. H. Chen and L. J. Li, *Sci. Rep.*, 2015, 5, 8289.
- 30 K. F. Mak, K. He, J. Shan and T. F. Heinz, *Nat. Nanotechnol.*, 2012, 7, 494–498.
- 31 H. Zeng, J. Dai, W. Yao, D. Xiao and X. Cui, *Nat. Nanotechnol.*, 2012, 7, 490–493.
- 32 Y. Wang, W. Wei, F. Li, X. Lv, B. Huang and Y. Dai, *Mater. Horiz.*, 2021, **8**, 244–249.
- 33 G. Aivazian, Z. Gong, A. M. Jones, R.-L. Chu, J. Yan, D. G. Mandrus, C. Zhang, D. Cobden, W. Yao and X. Xu, *Nat. Phys.*, 2015, 11, 148–152.
- 34 D. MacNeill, C. Heikes, K. F. Mak, Z. Anderson, A. Kormanyos, V. Zolyomi, J. Park and D. C. Ralph, Phys. Rev. Lett., 2015, 114, 037401.
- 35 R. Peng, Y. Ma, S. Zhang, B. Huang and Y. Dai, *J. Phys. Chem. Lett.*, 2018, **9**, 3612–3617.

- 36 Y. C. Cheng, Q. Y. Zhang and U. Schwingenschlögl, *Phys. Rev. B: Condens. Matter Mater. Phys.*, 2014, **89**, 155429.
- 37 I. Khan, B. Marfoua and J. Hong, *npj 2D Mater. Appl.*, 2021, 5, 10.
- 38 T. Norden, C. Zhao, P. Zhang, R. Sabirianov, A. Petrou and H. Zeng, *Nat. Commun.*, 2019, **10**, 4163.
- 39 Z. Zhang, X. Ni, H. Huang, L. Hu and F. Liu, *Phys. Rev. B*, 2019, **99**, 115441.
- 40 J. Qi, X. Li, Q. Niu and J. Feng, *Phys. Rev. B: Condens. Matter Mater. Phys.*, 2015, **92**, 121403.
- 41 Q. Zhang, S. A. Yang, W. Mi, Y. Cheng and U. Schwingenschlogl, *Adv. Mater.*, 2016, 28, 959–966.
- 42 X. Wang and J.-Y. You, *Mater. Today Electron.*, 2023, 5, 100051.
- 43 H. Zheng, B. Wu, C.-T. Wang, S. Li, J. He, Z. Liu, J.-T. Wang, G. Yu, J.-A. Duan and Y. Liu, *Nano Res.*, 2023, 16, 10580–10586.
- 44 L. Xu, M. Yang, L. Shen, J. Zhou, T. Zhu and Y. P. Feng, *Phys. Rev. B*, 2018, **97**, 041405.
- 45 G. Yang, J. Li, H. Ma, Y. Yang, C. Li, X. Mao and F. Yin, *Phys. Rev. B*, 2018, **98**, 235419.
- 46 G. Kresse and J. Furthmuller, *Phys. Rev. B: Condens. Matter Mater. Phys.*, 1996, 54, 11169–11186.
- 47 P. E. Blochl, *Phys. Rev. B: Condens. Matter Mater. Phys.*, 1994, **50**, 17953–17979.
- 48 J. P. Perdew, K. Burke and M. Ernzerhof, *Phys. Rev. Lett.*, 1996, 77, 3865–3868.
- 49 V. I. Anisimov, A. I. Poteryaev, M. A. Korotin, A. O. Anokhin and G. Kotliar, *J. Phys.: Condens. Matter*, 1997, **9**, 7359–7367.
- 50 Q. Wei, D. Chen, Y. Cai, L. Shen, J. Xu, J. Yuan, Y. Chen and X. Yan, *J. Supercond. Novel Magn.*, 2022, 35, 787–794.
- 51 A. A. Mostofi, J. R. Yates, Y.-S. Lee, I. Souza, D. Vanderbilt and N. Marzari, *Comput. Phys. Commun.*, 2008, **178**, 685–699.
- 52 A. Togo and I. Tanaka, Scr. Mater., 2015, 108, 1-5.
- 53 K. Chen, J. Deng, Y. Yan, Q. Shi, T. Chang, X. Ding, J. Sun, S. Yang and J. Z. Liu, *npj Comput. Mater.*, 2021, 7, 79.
- 54 H. Zhang, L.-M. Liu and W.-M. Lau, J. Mater. Chem. A, 2013,1, 10821–10828.
- 55 D. Wines, K. Saritas and C. Ataca, *J. Phys. Chem. C*, 2022, **126**, 5813–5821.
- 56 Q. Li, K.-Q. Chen and L.-M. Tang, *Phys. Rev. Appl.*, 2020, 13, 014064.
- B. Zhou, Z. Li, J. Wang, X. Niu and C. Luan, *Nanoscale*, 2019,
 11, 13567–13575.
- 58 T. Hu, G. Zhao, H. Gao, Y. Wu, J. Hong, A. Stroppa and W. Ren, *Phys. Rev. B*, 2020, **101**, 125401.
- 59 G. Yang, J. Li, Z. Liu, C. Li and X. Mao, *Phys. Chem. Chem. Phys.*, 2019, **21**, 15151–15156.