P510/1 Physics Paper 1 July/August 2019 2½ hours

BUGANDA EXAMINATIONS COUNCIL MOCKS

Uganda Advanced Certificate of Education

PHYSICS

PAPER 1

2HOURS 30 MINUTES

INSTRUCTIONS TO CANDIDATES:

- o Answer **five** questions, including at least **one**, but not more than **two** from each of the sections **A**, **B** and **C**.
- Any additional question(s) answered will not be marked.
- o Non-programmable scientific calculators may be used.
- o Assume where necessary;

-	Acceleration due to gravity	=	9.81ms ⁻²
-	Electron charge, e	=	$1.6 \times 10^{-19} \text{C}$
-	Electron mass	=	$9.11 \times 10^{-31} \text{kg}$
-	Mass of the earth	=	$5.97 \times 10^{24} \text{kg}$
-	Plank's constant, h	=	$6.6 \times 10^{-34} \text{Js}$
-	Stefan's-Boltzmann's constant, δ	=	$5.67 \times 10^{-8} \text{Wm}^{-2} \text{K}^{-4}$
-	Radius of the earth	=	$6.4 \times 10^6 \text{m}$
-	Radius of the sun	=	$7.0 \times 10^8 \text{m}$
-	Radius of earth's orbit about the sun	=	$1.5 \times 10^{11} \text{m}$
-	Speed of light in a Vacuum, C	=	$3.0 \times 10^8 \text{ms}^{-1}$

- Thermal conductivity of copper = 390Wm⁻¹K⁻¹ - Thermal conductivity of aluminium = 210Wm⁻¹K⁻¹

- Universal gravitational constant, G = $6.67 \times 10^{-11} \text{Nm}^{-2} \text{kg}^{-2}$

- Avogadro's number, N_A = $6.02 \times 10^{23} \text{mol}^{-1}$

- Density of water = 1000kgm^{-3}

- Gas constant, \mathbf{R} = 8.31 $\mathrm{Jmol}^{-1}\mathrm{k}^{-1}$

- Charge to mass ratio, e/m = 1.8 x 10¹¹Ckg⁻¹

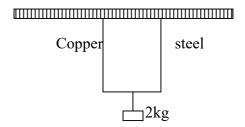
- The constant $\frac{1}{4\pi\varepsilon_0}$ = 9.0 x 10⁹F⁻¹m

- Faraday's constant, \mathbf{F} = $9.65 \times 10^4 \text{Cmol}^{-1}$

1(a) Define the elastic and inelastic collisions.

(02marks)

- (b) In an experiment to measures its velocity, a bullet of mass 10g is fired at short range from a gun of mass 0.8kg into a block of wood of mass 390g suspended from a vertical string. On striking the block, the bullet is embedded and the block rises to a vertical height of 6.0cm above its rest position. Calculate the;
 - (i) speed of the bullet (05marks)
 - (ii) the recoil velocity of the gun (02marks)
- (c)(i) Distinguish between conservative and non-conservative forces and state an example of each. (03marks)
- (ii) Show that in a system where the only forces acting are conservative forces, mechanical energy is conserved. (04marks)
- (d) A car of mass 1.0 x 10³ moves with uniform velocity of 36kmh⁻¹ up a straight track inclined at an angle of 20° to the horizontal. The total frictional resistance to the motion of the car is 248N. Calculate the power developed in the engine. (04marks)
- 2(a) Define the following terms
 - (i) Elastic limit (01mark)
 - (ii) Young's modulus (01mark)
 - (iii) Tensile stress (01mark)
- (b) Derive the expression for the energy stored on a unit volume of an elastic material in terms of its Young's modulus E and its strain. (04marks)
- (c) Two wires of copper and steel of the same length and of diameter 1.0mm are suspended side by side as shown in the figure below



A mass of 2kg is hung at the lower ends of the wires. If Young's Moduli for copper and steel are 1.0×10^{11} Pa and 2.0×10^{11} pa respectively find;

(i) the tension in each wire

(4marks)

(ii) the strain in the copper wire

(3marks)

- (d) A uniform metal bar of length 1.0m and diameter 2.0cm is fixed between two rigid supports at 25°C. If the temperature of the bar is raised to 75°C, find:-
 - (i) the force exerted on the supports.

(3marks)

(ii) the energy stored in the bar at 75°C.

(3marks)

(Young's modulus for the metal = 2.0×10^{11} pa, coefficient of linear expansion = 1.0×10^{-5} k⁻¹)

3a(i) State Bernoulli's principle.

(1mark)

- (ii) Explain why it is dangerous to stand close to a railway line to which a fast moving train is passing. (3marks)
- (b)(i) Explain the temperature dependence of viscosity of a liquid.

(3marks)

- (ii) Water of negligible viscosity flows steadily through a horizontal pipe of varying cross-sectional area. At a point A, of cross-sectional area 10cm^2 , the velocity is 0.2ms^{-1} . What is the pressure difference between A and B if the cross section area of point B is 2.5cm^2 ? (Given that the density of water = 10^3kgm^{-3}) (4marks)
- (c)(i) Explain the origin of surface tension

(4marks)

- (ii) Describe an experiment to measure the surface tension of a liquid by the capillary tube method. (5marks)
- 4a(i) State Keplar's laws of planetary motion.

(3marks)

(ii) Define the following; angular momentum and parking orbit.

(2marks)

(iii) What is meant by uniform circular motion?

(1mark)

- (b) A motor cyclist rides round a circular track of radius r inclined at angle Θ to the vertical.
- (i) Draw a sketch diagram of the motorist and show the forces acting on the motorist.

(2marks)

(ii) Explain why the motor cyclist leans towards the centre of the circular path. (2marks)

- (iii) If the coefficient of friction between the tyres and the ground is μ , derive the expression for which the motorcyclist will ride safely round the track. (3marks)
- (c) A small mass of 20.0g is attached to one end of an inextensible string of length 0.8m. The mass is whirled through a vertical circle of radius 0.8m. If the maximum tension in the string is 0.2N, find:-
 - (i) the period of motion of the mass. (3marks)
 - (ii) the tensions in the string when it makes an angle of 30° with the vertical.(4marks)

SECTION B

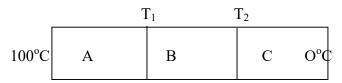
5a(i) Define critical temperature of a gas. (1mark)

(ii) Distinguish between a vapour and a gas. (2marks)

- (b)(i) Draw a sketch graph of pressure against temperature for a saturated vapour in a sealed container whose temperature is gradually raised until all the liquid in the container is evaporated. (2marks)
- (ii) A horizontal tube of uniform bore, closed at one end has some air trapped by small quantity of water. The length of the enclosed air column is 20.0cm at 12°C and 23.04cm at 38°C. Given that the saturation vapour pressure of water at 12°C is 10.5mm of mercury, find the saturation vapour pressure of water at 38°C.

 (Atmospheric pressure = 75.0cmHg). (4marks)
- (c)(i) With use of a labeled diagram, explain how a constant volume gas thermometer is used to measure the Kelvin temperature of a liquid. (5marks)
- (ii) Mention any two sources of error in the constant-volume gas thermometer and state how each one can be minimized. (2marks)
- (d) 15129 joules of heat is needed to raise the temperature of 288g of a gas by 41°C at constant volume. If the molar mass is 16g, find the molar heat capacity of this gas at constant pressure. (4marks)
- 6(a)(i) State Wein's displacement law. (1mark)
- (ii) Explain the mechanism of heat transfer by convection. (3marks)
- (b) A metal sphere of density 8.93 x 10³kgm⁻³ and specific heat capacity 370Jkg⁻¹k⁻¹ is placed on an enclosure maintained at a temperature of 300K. When the temperature of the sphere is 150K, it changes at a rate of 0.065Ks⁻¹.
- (i) Calculate the diameter of the sphere. (4marks)
- (ii) State any assumption made. (1mark)

- (c) With the use of a diagram, explain how a thermopile is used to detect radiation. (4marks)
- (d) A car engine has four cylinders. At a certain speed, a piston in a cylinder executes twenty-four power strokes per minute. To keep the temperature of the engine constant, water circulates around the engine at a rate of 200g per second and its temperature rises by 10.0K. Calculate;
- (i) the heat generated by each power stroke. (3marks)
- (ii) the power input of the car engine if its efficiency is 0.75. (3marks)
- 7a(i) Define cooling correction. (1mark)
- (ii) State Newton's law of cooling. (1mark)
- (b)(i) Using a well labeled diagram, describe an experiment to determine the specific latent heat of vapourisation of water by the method of mixtures. (6marks)
- (ii) State two advantages of the electrical method over the method of mixtures in determining the specific latent heat of vapourisation of a liquid. (2marks)
- (c) Three metallic conductors A, B and C of equal lengths and cross section area are joined to form an insulated composite rod as shown in the figure below.



The exposed end of A and C are maintained at 100° C and $^{\circ}$ C respectively. If the ratio of the thermal conductivities A, B and C is 1.5: 2: 2.5 respectively, find the steady temperature of the interfaces T_1 and T_2 . (4marks)

- (ii) Draw a sketch graph of temperature against length of the composite rod. (2marks)
- (d) Explain why;
- (i) one feels cool after sweating. (2marks)
- (ii) greenhouse effect causes global warming. (2marks)

SECTION C

- 8(a) Define binding energy of a nuclide. (1mark)
- (b)(i) Sketch a graph showing how binding energy per nucleon varies with mass number.

(1mark)

(ii) describe the main features of the graph in b(i) above. (3marks)

- (c) Distinguish between nuclear fission and nuclear fusion and account for energy released. (3marks)
- (d) With the aid of a labeled diagram, describe the working of an ionization chamber. (6marks)
- (e)(i) What is meant by half-life and decay constant as applied to radioactivity. (2marks)
- (ii) A Gerger Muller (GM) tube placed 20cm from a 2.0g of Randon ²²²₈₆Rn gives a count rate of 85 counts per second. If the entrance window of the GM tube has an area of 10cm², calculate the half-life of radon. (4marks)
- 9(a) Describe an experiment which justifies the existence of a small nucleus at the centre of an atom. (6marks)
- (b)(i) State Bohr's postulates of a hydrogen atom. (2marks)
 - (ii) State the limitations of Bohr's model of the atom. (2marks)
- Use Bohr's model to show that the total, E, of an electron in an atom is given by $E = \frac{me^4}{8\varepsilon n^2 n^2 h^2}, \text{ where m is the mass of the electron, } \varepsilon_o \text{ is permittivity of free space, n is principal quantum number and h is Planck's constant.}$ (5marks)
- (d) The ionization energy of helium is 24.6eV. An electron from an excitation level of helium of energy -21.4eV falls to the ground state.
- (i) What is ionization energy? (1mark)
- (ii) Calculate the wave length of the radiation emitted in the above transition to the ground state. (3marks)
- (iii) In what region of the spectrum does the radiation emitted lie? (1mark)
- 10a(i) Define stopping potential and threshold wavelength. (2marks)
- (ii) Explain why the wave theory of light fails to account for the photo electric emission of light. (4marks)
- (iii) Describe an experiment to determine Plank's constant. (5marks)
- (b) In X-ray production explain;
- (i) the difference between occurrence of line and continuous spectra. (3marks)
- (ii) the difference between hard and soft x-rays. (2marks)
- (c) An x-ray tube is operated at 20KV with an electron current of 16mA in the tube Estimate the:
- (i) number of electrons hitting the target per second. (2marks)
- rate of flow of cooling water needed to maintain a temperature difference of 12K between the inflow and outflow assuming 99.5% of the kinetic energy of electron is converted to heat. (2marks)

END