

EWALDI COMMUNITY SCHOOL

P. O. BOX 33197, KAMPALA-UGANDA TEL: 0787900299/0200907504 PRIMARY FIVE SCIENCE LESSON NOTES FOR TERM TWO 2023

CONTENT BREAKDOW FOR P.5 SCIENCE TERM TWO 2023

THEME 1: THE ENVIRONMENT

13. SOIL FERTILITY
THEME 2: MATTER AND ENERGY

TOPIC: COMPONENTS OF ENVIRONMENT (SOIL)

- 1. TERMS USED IN SOIL
- 2. TYPES OF SOIL
- 3. FORMATION OF SOIL
- 4. COMPONENTS OF SOIL
- 5. IMPORTANCE OF SOIL
- 6. SOIL EROSION
- 7. TYPES OF SOIL EROSION
- 8. CAUSES OF SOIL EROSION
- 9. AGENTS OF SOIL EROSION
- 10. PREVENTION OF SOIL EROSION
- 11. EFFECTS OF HARMFUL MATERIALS ON THE SOIL
- 12. SOIL CONSERVATION

THEME 3: SCIENCE IN HUMAN ACTIVITIES AND OCCUPATION

TOPIC: OCCUPATION IN OUR COMMUNITY (CROP GROWING)

- 1. ROOT TUBERS
- 2. STEM TUBERS
- 3. CARE FOR TUBER CROPS
- 4. COMMON PESTS AND DISEASES OF TUBER CROPS
- 5. CHARACTERISTICS OF COMMON TUBER CROP PEST
- 6. EFFECTS OF PESTS AND DISEASES ON TUBER CROPS
- 7. HARVESTING AND STORAGE
- 8. SCIENCE BASED CLUBS

THEME 4: THE WORLD OF LIVING THINGS

TOPIC: BACTERIA AND FUNGI

- 1. TYPES OF BACTERIA
- 2. NATURE OF BACTERIA

1. MATTER

TOPIC: HEAT ENERGY

- 2. STATES OF MATTER
- 3. MIXTURES
- 4. ENERGY
- 5. OTHER FORMS OF ENERGY
- 6. HEAT ENERGY
- 7. EFFECTS OF HEAT GAIN AND LOSS
- 8. USES OF DIFFERENT METHODS OF HEAT TRANSFER
- 9. HEAT AND TEMPERATURE
- 10. CONVERSIONS FROM CELSIUS TO FAHRENHEIT AND VICE VERSA
- 3. PREVENTION AND TREATMENT OF BACTERIAL DISEASES
- 4. FUNGI KINGDOM
- 5. DEFINITIONS OF KEY TERMS
- 6. BREEDING, NATURE AND GROWTH OF FUNGI
- 7. EXAMPLES OF FUNGI
- 8. USEFUL OR HARMLESS FUNGI
- 9. USELESS OR HARMLESS FUNGI
- 10. DANGERS OF FUNGI
- 11. WAYS OF CONTROLLING BACTERIAL AND FUNGI DISEASES
- 12. FACTS ABOUT BACTERIA AND FUNGI.

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THEME: COMPONENTS OF THE ENVIRONMENT

<u>WEEK 1</u> LESSON 1

			No. of pupils		ils		Teacher's name
Date	class	subject				Time	
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Vocabularies/key words

- 1. Soil erosion
- 2. Weathering
- 3. Decomposition
- 4. Organic matter
- 5. Components

- 6. Leaching
- 7. Soil texture
- 8. Soil profile
- 9. Soil capillarity
- 10.Soil structure

SOIL

Soil is the top layer of the earth's surface.

HOW SOIL IS FORMED

- > By weathering
- > By decomposition of organic matter.

Weathering is the physical and chemical breakdown of rocks into small particles to form soil.

Decomposition is the rotting of dead organic matter.

TYPES OF SOIL

There are three types of soil

- i. Clay soil.
- ii. Loam soil.
- iii. Sand soil.

a) CLAY SOIL

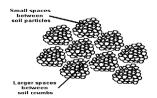
Characteristics of clay soil

- It has fine particles.
- It has closely packed particles.
- It does not allow water to pass through it very fast.
- It has a high water retention capacity so it easily becomes water logged.
- Clay soil has the highest rate of capillarity

Importance of clay soil

- Clay soil is good for pottery work (making pots, ceramics and modeling).
- Clay soil is good for making bricks for building.
- Clay soil is good for making tiles for roofing.

Illustration of arrangement of particles in clay soil.



Activity

- 1. What is soil?
- 2. Identify any three components of soil.
- 3. How is soil formed?
- 4. Mention the three types of soil.
- 5. How is clay useful to man?

GENERAL EVALUATION

SELF EVALUATION	STRENGTH	WEAKNESS	WAY FORWARD		

LESSON 2

Date	class	subject	No. of pupils			No. of pupils Time		Time	Teacher's name
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b) **LOAM SOIL**

Loam soil is a mixture of clay soil and organic matter Organic matter (humus) consists of decayed plants and animal matter Loam soil usually has adequate water, air and humus to sustain plant growth.

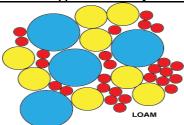
Characteristics of loam soil.

- It's particles are fairly arranged
- It contains both clay and sand particles.
- It has a lot of humus for plant growth.
- Has fairly larger air spaces as compared to clay soil

Importance of loam soil

- It is good for crop growing.

Illustration of arrangement of particles in loam soil.

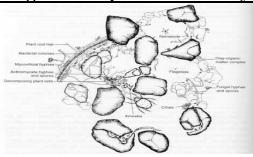


1. Why is loam soil the best for plant growth?

- ✓ It contains a lot of humus.
- ✓ It contains balanced particles of sand and clay.

c) SAND SOIL

Illustration of arrangement of particles in sandy soil.



- Has large particles that make it to be well aerated
- Water passes through it easily
- Has poor water retention capacity
- Has high water drainage
- It is easy to dig
- Has a poor rate of capillarity

Capillarity is the uptake of water through the soil particles. /

Capillarity is the tendency of water to rise through small narrow spaces.

Drainage is the capacity of the soil to allow water to pass through it.

Importance of sandy soil.

- Used for building.
- Used for making glass and sand papers.

Nb. It is not good for crop growing because it has a low water holding capacity.

Activity

- 1. why is loam soil the best for crop growing?
- 2. mention one characteristic of loam soil
- 3. identify the type of soil which used for making glass.
- 4. what is capillarity?
- 5. which type of soil is porous?

GENERAL EVALUATION

SELF EVALUATION	STRENGTH	WEAKNESS	WAY FORWARD

LESSON 3

			No. of pupils		ils		Teacher's name
Date	class	subject				Time	
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COMPONENTS OF THE SOIL

These are things which make up soil, they include:

- Air.
- water
- humus
- rock particles
- living organisms eg bacteria, insects, earthworms etc.

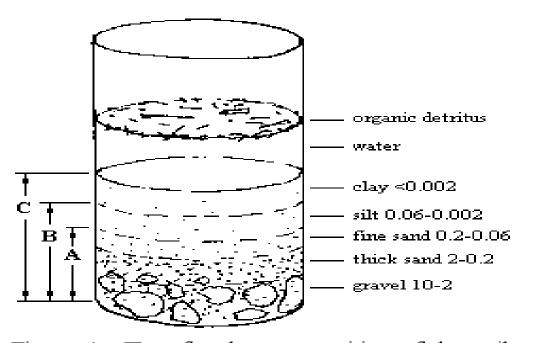


Figure 1 - Test for the composition of the soil.

(The sizes are in mm)

Air is used by animals in the soil to respirati on.Air is used during germina

tion.

Import ance of compo nents

of soil

a) Air

b) Water

- Water is used by plants for germination
- Making starch (it is a raw material for photosynthesis)

- Promoting decay of matter
- c) Rock particles (inorganic materials like: sand, gravels, clay formed by weathering)
- Provide space for air to occupy

d) Humus – dead decayed plants and animal matter

- Provide plant nutrients.
- Improve soil fertility
- Makes the soil appear dark in colour
- e) Living organisms

Examples of animals that live in the soil.

- ✓ Bacteria
- ✓ Moles
- ✓ Porcupines
- ✓ earth worms
- ✓ ants
- Bacteria like nitrogen fixing bacteria fix nitrogen in the soil and hence improving on soil fertility.

Earthworms

- Aerate the soil.
- Softens the soil /plough the soil
- Add soil fertility by breaking down dead plants and animal remains.

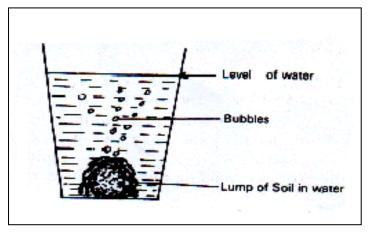
NB: Why do you think earthworms come out of the soil after raining?

- To breathe /take in oxygen.

PROPERTIES OF SOIL.

- It has air.
- It has water.
- It contains mineral salts.

EXPERIMENT ABOUT SOIL CONTAINS AIR.



SOIL PROFILE

Soil profile is the vertical arrangement of soil layers.

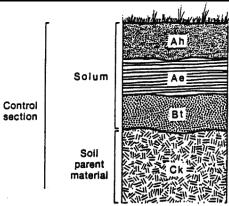
Or

Soil profile is the arrangement of soil layers from top to bottom.

Areas where one can clearly see soil profile.

- ✓ Pit latrines.
- ✓ In trenches.

ILLUSTRATION OF THE LAYERS OF THE SOIL.



Importance of the top most layer.

- It contains most of the nutrients for plant growth.

Briefly explain the following terms.

- **Soil texture.** These are the different sizes of soil particles in an area.
- **Soil structure.** Is the way soil particles are grouped/ arranged.

Activity

- 1. What is soil profile?
- 2. Name the soil layer which good for plant growth.
- 3. Suggest any place where one can clearly soil profile.
- 4. What is soil texture?
- 5. Name any two organisms found in the soil.

GENERAL EVALUATION

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SELF EVALUATION	STRENGTH	WEAKNESS	WAY FORWARD						

LESSON 4 &5

Date	class	subject	No. of pupils			Time	Teacher's name
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SOIL EROSION

Soil erosion is the removal of top soil by its agents.

AGENTS OF SOIL EROSION

These are forces that carry away top soil from one place to another. These include:

- Fast flowing water
- Strong wind
- Moving animals

CAUSES OF SOIL EROSION

These are main activities that enable the agents to take away top soil.

How?

- They expose the soil to agents of soil erosion.

These include

- Deforestation
- Overgrazing
- bush burning
 - Monoculture (mono-cropping)
 - Ploughing down slopes.
 - Over cultivation.

TYPES OF SOIL EROSION

- a) Sheet erosion: Top soil is washed away uniformly by fast flowing water.
- b) Gulley erosion (deep channels)
- c) Rill erosion (shallow channels)
- d) Splash erosion /raindrop erosion
- e) Stream /river bank erosion
- f) Wind erosion

Effects of soil erosion.

- Leads to soil exhaustion.
- It affects soil texture.

PREVENTION AND CONTROL OF SOIL EROSION.

- 1. **Terracing:** reduces the speed of running water
- 2. **Strip cropping**: reduces the speed of running water
- 3. **Contour ploughing**: is the ploughing across a slope.

It helps to reduces the speed of running water

- 4. **Afforestation**: is planting of trees where they have ever existed. This also keeps the soil covered from direct rain drops.
- 5. **Re-afforestation**: is the planting of trees where they have been ever existed. This also keeps the soil covered from direct rain drops.
- 6. Cover cropping: planting cover crops between plants that take long to mature.

Cover crops

These are crops that are planted between plants that take long to mature

Qn: How does cover cropping prevent soil erosion?

Cover crops protect soil from direct rain drops.

Qn: How does inter-cropping reduce soil erosion?

Reduces the speed of fast flowing water

- 7. **Bush fallowing:** resting period of land to regain its fertility **Importance:** enables the land to regain its fertility.
- 8. **Mulching:** is the covering of top soil in the garden with dry plant materials.

Materials used as mulch.

- ✓ Coffee husks
- ✓ Saw dust
- ✓ Dry banana leaves
- ✓ Crushed maize cobs
- ✓ Wood shavings
- ✓ Dry grass
- ✓ Straw

ADVANTAGES OF MULCHING.

- Controls soil erosion. How? By reducing the speed of fast flowing water.
- Maintains soil fertility. How? by reducing soil erosion and mulches rot to from humus.
- Keeps water in the soil. How? by condensing evaporating air from the soil
- Increases on the crop yields.
- Reduces the rapid growth of the weeds.

Disadvantages of mulching

- Mulches keep pests.
- Dry mulches can be fire hazards.
- Some mulches can grow into weeds.

How soil loses its fertility

Through:

- Leaching. It is the sinking of plant nutrients deeper into the soil where plant roots can't reach.
- Soil erosion
- Monoculture (mono-culture)
- Bush burning

How can we improve soil fertility?

- Mulching
- Crop rotation
- Bush fallowing
- Addition of fertilizers
- By terracing
- A forestation

Activity

1. What is soil erosion?

- 1. Mention the three agents of soil erosion.
- 2. give any three causes of soil erosion
- 3. Name two types of soil erosion.
- 4. Identify any one effect of soil erosion.

GENERAL EVALUATION

SELF EVALUATION	STRENGTH	WEAKNESS	WAY FORWARD		

WEEK TWO LESSON 1&2

			No. of pupils		ils		Teacher's name
Date	class	subject				Time	
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FERTILISERS.

These are substances put in the soil to increase its fertility.

Types fertilizer

- Artificial fertilizers.
- Natural fertilizers.

ARTIFICIAL FERTILIZERS

Are fertilizers got from inorganic matter artificially.

TYPES OF ARTIFICIAL FERTILIZERS

- Straight fertilizers.
- Compound fertilizers.

Straight fertilizers.

These are fertilizers that supply one nutrient to the soil.

Examples:

- ✓ SSP (single supper phosphate)
- ✓ Nitrogen
- ✓ Phosphorus
- ✓ Potassium.

Compound fertilizers.

These are fertilizers that supply more than one nutrient to the soil.

Examples:

✓ Ammonium phosphate

Advantages of using artificial fertilizers

- They have a high nutrient content needed by the plants.
- They are easy to handle, use and store.
- They help to make plants resistant to diseases as they grow.
- They provide the needed minerals with out fail.
- They supply a large quantity of mineral salts.
- They provide freedom of choice to farmers as they buy the type that they need.
- They improve crop yield.
- They supply the correct a mount of mineral salts to soil to meet the requirement of crops.

Disadvantages of using artificial fertilizers

- They are expensive to buy.
- -They require skilled labour to apply it.
- They stay in the soil for a shorter period.
- They are poisonous to animals, people and birds.
- They pollute water sources when washed there.
- They destroy soil texture and structure.
- They make the soil acidic.
- They kill organisms in the soil.

Natural fertilizers (manure)

These are fertilizers made from decayed plant and animal materials [humus)

TYPES OF NATURAL FERTILIZERS (MANURE)

There are four types of manure

- 1. Compost manure
- 2. Green manure
- 3. Farm yard manure
- 4. Organic mulch.

ACTIVITY

- 1. What are fertilizers?
- 2. Name the two types of fertilizers.
- 3. Write in full SSP.
- 4. A part from straight fertilizer, mention another type of artificial fertilizer.

GENERAL EVALUATION

SELF	STRENGTH	WEAKNESS	WAY
EVALUATION			FORWARD

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LESSON 3 &4

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Date	class	subject				Time	
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Compost manure

Compost manure is got from both plant and animal organic waste matter left to decay.

Things used to make compost manure.

- banana peeling.
- dry grass
- maize stalks
- leaves.
- animal wastes[dung and urine]

COMPOST HEAP.



Importance of compost manure

- Adds humus to the soil (improves soil fertility)
- Controls leaching
- Improves soil structure

Advantages of compost manure

- It is cheap.
- Its nutrients last for a longer time
- It is not poisonous to people, animals and birds
- It does not require skilled labour to apply it.
- Provides many nutrients to the soil at the same time.

NB: When making compost manure, water should be added to the compost heap to make the rotting faster.

Disadvantages of compost manure

- Produces bad smell

- It takes a lot of space
- Soil nutrients take long to be released into the soil
- It is not easy to tell which nutrients is present in the compost manure

Green manure

It is made from green crops especially legumes that are ploughed back into the soil at the flowering stage.

Why legumes? They can rot in a shorter time.

Advantages of green manure

- It lasts for several seasons in the soil.
- It rots and mixes easily into the soil.
- Its nutrients last for a longer time
- It is not poisonous to people, animals and birds
- It does not require skilled labour to apply it.

Disadvantages of green manure

- It is time wasting.
- It requires a lot of labour to plough it back into the soil.
- It is not easy to tell which nutrient is present in the green manure.

Farm yard manure

Is the manure got from animal wastes like dung and urine mixed with their bedding materials?

Advantages of farm yard manure

- It is cheap
- Its nutrients last for a longer time
- It is not poisonous to people, animals and birds
- It does not require skilled labour to apply it.
- Provides many nutrients to the soil at the same time

Disadvantages of farm yard manure

- It contains little amount of mineral salts.
- Produces bad smell
- It takes a lot of space
- Soil nutrients take long to be released into the soil
- It is not easy to tell which nutrient is present in the compost manure.

Effects of harmful materials on the soil

- Non degradable materials do not rot or decay
- They prevent or block air and water from entering the soil
- They lead to soil exhaustion.

These materials are called pollutants

Examples of soil pollutants.

- Polythene papers (buveera)
- Plastic materials

- Rubber materials (old shoes)
- Glass and broken bottles
- Metallic materials like nails, tins, etc
- Concrete from broken buildings.

Ways of properly handling wastes.

- Reusing plastic materials like jerry cans for other purposes like fetching water.
- Making new things from some wastes (recycle)
- Returning some wastes like bottles to the factories that make sodas.
- Rejecting/ refusing the use of non biodegradable materials like polythene bags.
- Reducing on the production of non biodegradable materials.

ACTIVITY

- 1. What is compost manure?
- 2. State two advantages of compost manure.
- 3. Suggest any two disadvantages of green manure.
- 4. What are soil pollutants?
- 5. Give two examples of soil pollutants.

GENERAL EVALUATION

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SELF EVALUATION	STRENGTH	WEAKNESS	WAY FORWARD						

LESSON 5

Date	class	subject	No. of pupils			Time	Teacher's name
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SOIL BARRIERS

These are things (materials) that prevent soil from being carried away by the agents of soil erosion.

Examples of soil barriers

- Grass (cover crops)
- Trees /roots
- Rocks
- Concrete walls
- Contours
- Gabions, Etc

SOIL CONSERVATION

Soil conservation is the maintaining soil fertility.

Methods of soil conservation

- Afforestation /re-afforestation
- Bush fallowing
- Terracing hilly areas
- Rotational grazing
- Contour ploughing
- Strip cropping
- Crop rotation
- Mulching
- Cover cropping
- Mixed farming
- Manuring /application of the fertilizers
- Agro-forestry.

Importance of soil conservation

- Retains soil fertility
- Retains soil moisture
- Prevents spread of diseases and pests.

ACTIVITY

- 1. What are soil barriers?
- 2. give any four examples of soil barriers
- 3. What is mulching?
- 4. State any two advantages and disadvantages of mulching.

GENERAL EVALUATION

SELF EVALUATION	STRENGTH	WEAKNESS	WAY FORWARD

WEEK 3 LESSON 1 & 2

Date	class	subject	No. of pupils			No. of pupils		oils	Time	Teacher's name
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MATTER

Matter is anything that has mass and volume.

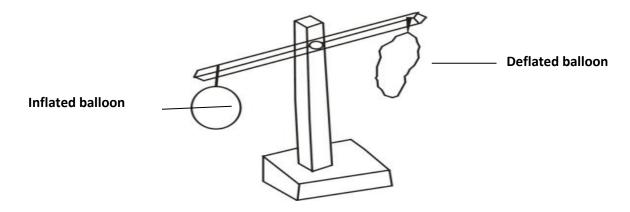
OR

Matter is anything that has weight and occupies space.

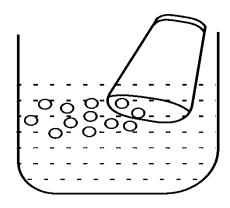
Properties of matter

- **❖** Matter occupies space.
- ❖ Matter has weight.
- **❖** Matter exerts pressure.

Experiment to show that air has weight.



Experiment to show that air occupies space.



The meaning of each of the following

- a. Mass: is the amount of matter in an object.
- b. **Volume:** Is the space occupied by an object.
- c. **Molecules:** molecules are the smallest particles of matter.
- **d. Weight:** is the gravitational force exerted on an object by the earth.

STATES OF MATTER

There are three different states of matter

Solids/solid state

Liquids/liquid state

Gases/gaseous state

SOLIDS

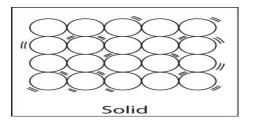
Examples of solids.

- Wood
- Rubber
- Glass
- Plastic

Characteristics of solids.

- Molecules of solids are closely packed.
- Particles of solids are held together very tightly.
- Molecules of solids do not move from position but vibrate.
- Solids have shape, size and volume apart from irregular objects.

Diagram to show the arrangement of molecules.



LIQUIDS

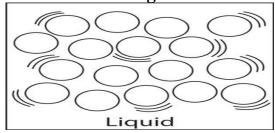
Examples of liquids.

- Water
- Soda
- Oils
- Juice

Characteristics of liquids.

- Molecules of liquids are spaced.
- Molecules of liquds are loosely held together.
- Liquids have a proper volume (capacity)
- Liquids have no definite shape (take up the shape of the container in which they are poured)

Diagram to show the arrangement of molecules.



GASES

Examples of gases.

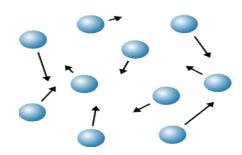
- Nitrogen

- Carbon dioxide
- Oxygen
- Rare gases

Characteristics of gases.

- Molecules of gases are far apart
- Gases have no definite shape
- Gases have a particular volume.
- Molecules move freely.

Diagram to show the arrangement of molecules.



ACTIVITY

- 1. what is energy?
- 2. what is matter?
- 3. name the three states of matter.
- 4. in which state of matter are the following
 - a) water
 - b) metal
 - c) oxygen

GENERAL EVALUATION

SELF EVALUATION	STRENGTH	WEAKNESS	WAY FORWARD

LESSON3&4

			No. of pupils		ils		Teacher's name
Date	class	subject				Time	
			В	G	T		

FORMS OF ENERGY

Energy is the ability of the body to do work.

Forms of energy.

- Heat energy
- Sound energy
- Light energy
- Sound energy
- Electric energy
- Magnetic energy
- Chemical energy
- Solar energy.

Types of energy

- Kinetic energy
- Potential energy

Potential energy

Potential energy is the energy that is stored by an object at rest.

Examples of potential energy

- A baby being asleep in a cot
- A car standing still at traffic lights
- A pupil sitting and listening to the teacher
- A stone / book ruler resting on a table /ground /cupboard etc

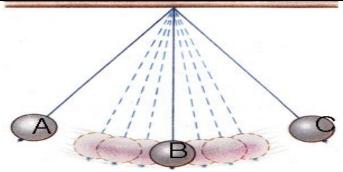
Kinetic energy

This is the energy possessed by a moving object or an object in motion. It is also referred to as the *energy in motion*

Examples of kinetic energy

- A girl running along the road
- An arrow flying through the air
- A stone thrown up in air
- A brick dropping from a wall
- A leaf falling to the ground from a tree

Think of a stone or a pendulum swinging in air



At A, the stone possess potential energy At B, the stone possess kinetic energy

Activity

- 1. Mention the two types of energy.
- 2. Define the following types of energy
 - **a.** potential energy.
 - **b.** kinetic energy.
- 3. List down any three forms of energy.

GENERAL EVALUATION

SELF EVALUATION	STRENGTH	WEAKNESS	WAY FORWARD

LESSON 5&6

Date	class	subject	No. of pupils		ils	Time	Teacher's name
			В	G	T		

HEAT ENERGY

Heat energy is a form of energy that increases temperature of an object.

Standard units for measuring heat.

Heat Calories

Instruments used to measure heat.

Heat Calorimeter

sources of heat

These are objects that produce heat.

Types of sources of heat.

- i. Natural sources of heat
- ii. Artificial sources of heat

Natural sources

Natural sources are sources of heat provided by nature.

Examples.

- ❖ Sun (main natural source of heat)
- **❖** Food
- Stars

Erupting volcanoes

Artificial sources.

These are sources of heat made by people.

Examples of artificial sources of heat.

- Lamps
- Bulbs
- Candles

Uses of heat to man

- used for ironing clothes
- Heat is used to run some machines e.g. steam engines. Diesel engines, rockets etc.
- used to dry harvested crops before storage.
- Heat evaporates the water in water bodies and plants during rain formation
- Heat enables us to cook our food
- Heat can act as a disinfectant.

Effects of heat on matter

Heat causes the following effects to matter.

- Melting.
- Expansion.
- Evaporation.
- Sublimation.

Diagram to show physical changes of state of matter.

- **A** Melting physical change from solid to liquid
- **B** Evaporation physical change from liquid to gas.
- **C** Freezing physical change from liquid to solid.
- **D** Condensation Physical change from gas to liquid
- E Sublimation physical change from solid to gas.

F - Sublimation physical change from gas to solid.

ACTIVITY

- 1. What is heat energy?
- 2. Name two the types of sources of heat in the environment.
- 3. Give two examples of each of the following sources of heat
 - a. Natural source of heat.
 - b. Artificial source of heat.
- 4. Identify any two effects of heat on a state of matter.

GENERAL EVALUATION

SELF EVALUATION	STRENGTH	WEAKNESS	WAY FORWARD

WEEK4

LESSON 1&2

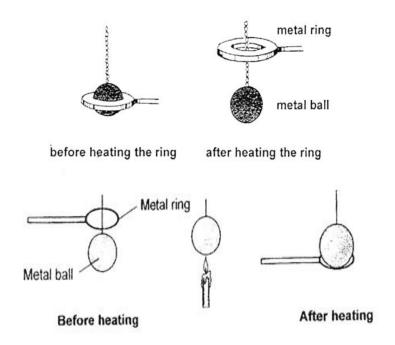
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EFFECTS OF HEAT ON EACH STATE OF MATTER. SOLIDS.

- Metals expand.

Metallic ball ring experiment

a). Before heating the metallic ball goes through the ring.



b). after heating the metallic ball does not pass through the ring

Reasons why metallic ball did not go through the ring after heating

The metallic ball had expanded.

Conclusion

Metals expand when heated.

GENERAL EVALUATION

SELF EVALUATION	STRENGTH	WEAKNESS	WAY FORWARD

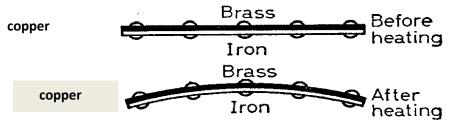
LESSON 3&4

			No. of pupils				Teacher's name
Date	class	subject				Time	
			В	G	Т		

What do you think can happen to the metallic ball if dipped into cold water? The hot metallic ball will contract and pass through the ring again.

Bimetallic strip.

Before heating and after heating



Bimetallic strip

Observation

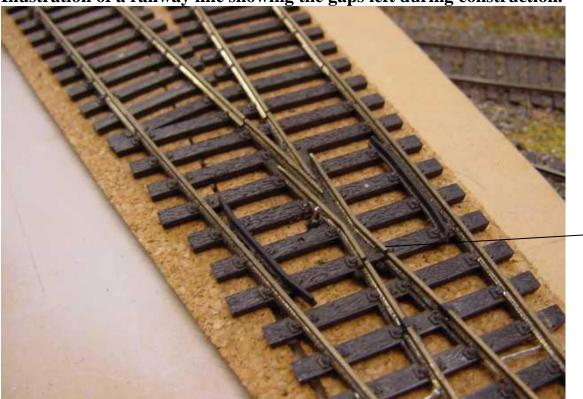
The copper strip heats up and expands faster than the iron strip hence bending to wards the copper.

Qn: 1. Which of the two metals is the best conductor of heat? Copper

IMPORTANCE OF A BIMETALLIC.

- Bimetallic strips are used in automatic switches of electric kettles, flat irons, fridges, freezers etc.

Illustration of a railway line showing the gaps left during construction.



Qn: What happens if gaps were not left between rails during construction? *The rails would expand on hot days bend and cause railway accidents.*

2. Why are gaps left between railways during construction? To leave room for expansion of metals on a hot day.

Diagrams to show the effect of heat on electric / telephone wires On a hot day.



Electric / telephone wires expand become loose and starts sagging / slacking. **On a cold day** wire contract and become shorter appearing relatively tight.



Qn; 1. Why are electric / telephone wires left sagging/slacking during construction?

To allow room for contraction of wires on a cold day.

2. What would happen to the wires when tied tightly fixed on the poles? *The wires would break due to contraction on cold days.*

Activity

- 1. Why gaps left between the railway lines during construction?
- 2. Why telephone wires are loosely fixed on the poles?
- 3. what happens to electric wires on the following
 - a. Cold days.
 - b. Hot days.

GENERAL EVALUATION

SELF EVALUATION	STRENGTH	WEAKNESS	WAY FORWARD

LESSON 5

			No. of pupils		No. of pupils			Teacher's name
Date	class	subject				Time		
			В	G	T			

EFFECTS OF HEAT ON ICE.

What happens to ice when heated?

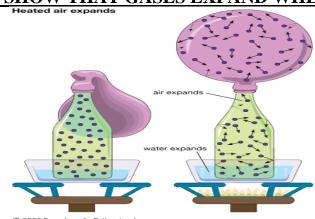
- Ice melts
- The volume decreases, the density increases and the mass remains the same.

EFFECTS OF HEAT ON GASES

What happens to gases when:-

a). Heated - Gases expandb). Cooled - Gases contract

EXPERIMENT TO SHOW THAT GASES EXPAND WHEN HEATED.



QN:1. Why does the balloon in diagram A expand?

Due to expansion of air inside the plastic bottle.

3. What happens to the balloon when the bottle is removed from the hot water?

The balloon collapses due to contraction of the air inside the plastic bottle.

Effects of heat on liquids

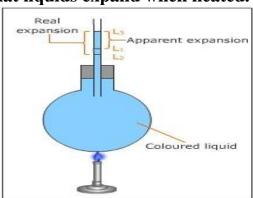
- Liquids expand when heated.

- Liquids evaporate when heated.
- Liquids contract when cooled

Effects of freezing of liquids

- Liquids increase in volume
- Liquids reduce in density
- Mass of liquids remains the same.

Diagram to show that liquids expand when heated.



Qn: 1. why is space left while bottling drinks like soda?

To leave room for increase in volume of drinks when frozen.

2. Why is carbon dioxide packed in bottled drinks like soda? To preserve the soda.

A BOTTLE WITH THE SPACE LEFT



Space left to give room for increase in volume on freezing.

Effects of heat on sublimates

NB: Sublimates are substances which can directly change from solid to gas.

Examples of sublimates

- i. Iodine granules (crystals)
- ii. Ammonium chloride(salts)
- iii. Solid carbon dioxide.

GENERAL EVALUATION

SELF EVALUATION	STRENGTH	WEAKNESS	WAY FORWARD

WEEK 5

LESSON 1&2

Date	class	subject	No. of pupils		ils	Time	Teacher's name
		-	В	G	T		

TEMPERATURE

Temperature is the degree of hotness or coldness of an object.

Standard units for measuring temperature.

Temperature **Degrees**

<u>Instrument used to measure temperature.</u>

Temperature Thermometer

Thermometer

Thermometer is an instrument used to measure temperature.

The following thermometer measure the following:-

- Lowest temperature of the day $\boldsymbol{Minimum\ thermometer}$
- Highest temperature of the day $\boldsymbol{Maximum\ thermometer}$
- Room temperature or temperature of the air **Wall thermometer**
- Highest and lowest temperature of the day Six's thermometer
- Human body temperature **Clinical thermometer**

Places where we find clinical thermometer in daily life

- In clinics
- In hospitals
- In dispensaries

Places or common sites on our bodies where a clinical thermometer can be placed while measuring the human body temperature

- Under the arm pits
- In the anus
- In the mouths / under the tongue

Diagram of clinical thermometer

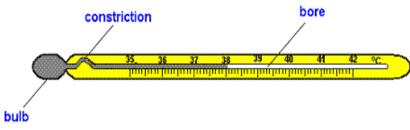


Fig. 10.5

Importance of each part

Stem: Protects the inside parts of a thermometer

Kink: Prevents the back flow of mercury to the bulb before the actual temperature is taken.

Bore: The bore has a regular scale.

Bulb: The bulb store mercury.

<u>Liquids commonly used in thermometers.</u>

- Alcohol
- Mercury

ACTIVITY

- 1. What is temperature?
- 2. State the difference between heat and temperature.
- 3. Name the any one type of thermometer.
- 4. Which liquids are used in the thermometers

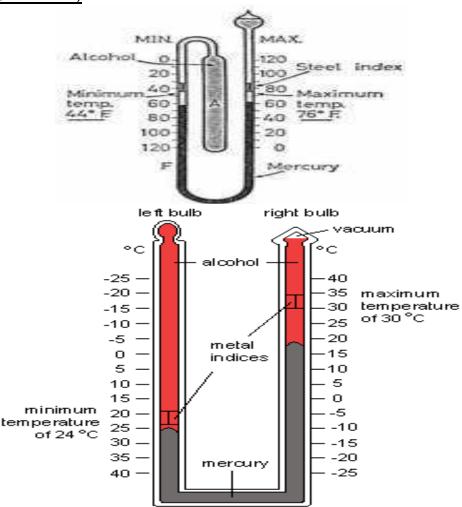
GENERAL EVALUATION

SELF EVALUATION	STRENGTH	WEAKNESS	WAY FORWARD				

LESSON 3

Date	class	subject	No. of pupils		ils	Time	Teacher's name
			В	G	T		

SIX'S THERMOMETER (MINIMUM AND MAXMUM THERMOMETER)



Reasons why mercury is used in a thermometer

- Mercury is a good conductor of heat
- Mercury does not stick on the walls of the bulb / bore.
- Mercury can easily be seen in the glass.
- Mercury has even and regular expansion.

Reasons why water is not used in a clinical thermometer

- Water is not easily seen
- Water is a bad conductor of heat
- Water needs a lot of heat to expand.
- Water's does not expand uniformly.

Why do doctors shake a clinical thermometer before using it on other patients?

- To draw the mercury back to the bulb

Reason why a clinical thermometer is sterilized using surgical spirit

- Boiling it will make the stem expand and burst.

The normal human body temperature - Celsius 37°c.

- 98.6/98.4degrees Fahrenheit.

GENERAL EVALUATION

SELF	STRENGTH	WEAKNESS	WAY					
EVALUATION			FORWARD					

LESSON 4

 $F^0 = (5 \times 9) + 32$

			No. of pupils		No. of pupils			Teacher's name
Date	class	subject				Time		
			В	G	T			

Changing from Celsius to Fahrenheit	
a. 20^{0} c	5
$F = (c \times 9) + 32^0$	$F^0 = (1 \times 9) + 32$
5	$F^0 = 9 + 32$
$F^0 = (20 \times 9) + 32$	${}^{0}F = 41^{0}$
5	
$F^0 = (4 \times 9) + 32$	
$F^0 = 36 + 32$	
${}^{0}F = 68^{0}$	
${b.}$ 25 0 c	$F = (c \times \underline{9}) + 32^0$
$F = (c \times 9) + 32^0$	5
5	$F^0 = (\theta \times \underline{9}) + 32$
$F^0 = (25 \times 9) + 32$	5
5	$F^0 = (0 \times 9) + 32$
$F^0 = (5 \times 9) + 32$	$F^0 = 0 + 32$
$F^0 = 45 + 32$	${}^{0}F = 32^{0}$
${}^{0}F = 77^{0}$	
$\overline{\mathrm{c.}}$ 5 $^{0}\mathrm{c}$	d. 80°
$F = (c \times \underline{9}) + 32^0$	$F = (c \times \underline{9}) + 32^0$

$$F^{0} = (16 \times 9) +32$$

$$F^{0} = 144 + 32$$

$${}^{0}F = 176^{0c}$$

e.
$$100^{0}$$
c
 $F = (c \times 9) + 32^{0}$
 5
 $F^{0} = (100 \times 9) + 32$
 5
 $F^{0} = (20 \times 9) + 32$
 $F^{0} = 180 + 32$
 ${}^{0}F = 212^{0}$

Change the following from degrees Celsius to degrees Fahrenheit

- 1. 5^{oc}
- 2. 10^{0c}
- $3. 100^{0c}$
- $4. \ 0^{0c}$
- 5. 8^{0c}

ACTIVITY:

LESSON 5

	_		No. of pupils				ils		Teacher's name
Date	class	subject				Time			
			В	G	T				

Changing temperature from Fahrenheit to Celsius

Example
$$41^{0}$$
F to C^{0}
 $C^{0} = (F - 32) \times \frac{5}{9}$
 $C^{0} = (41 - 32) \times \frac{5}{9}$
 $C^{0} = (41^{0} - 32) \times \frac{5}{9}$
 $C^{0} = 9 \times \frac{5}{9}$
 $C^{0} = 1 \times 5$
 $C^{0} = 5^{0}$

Change
$$68^{0}$$
F to C^{0}
 $C^{0} = (F - 32) \times \frac{5}{9}$
 $C^{0} = (68^{0} - 32) \times \frac{5}{9}$
 $C^{0} = 36 \times \frac{5}{9}$
 $C^{0} = 4 \times 5$
 $C^{0} = 20^{0}$

Change
$$32^{0}$$
F to C^{0}
 $C^{0} = (F - 32) \times \frac{5}{9}$
 $C^{0} = (32^{0} - 32) \times \frac{5}{9}$

$$C^{0} = 0 \times \frac{5}{9}$$

 $C^{0} = 0 \times 5$
 $C^{0} = 0^{0}$
Change 59^{0} F to C^{0}
 $C^{0} = (F - 32) \times \frac{5}{9}$
 $C^{0} = (59^{0} - 32) \times \frac{5}{9}$
 $C^{0} = 27 \times \frac{5}{9}$
 $C^{0} = 3 \times 5$
 $C^{0} = 15^{0}$

Change
$$95^{0}$$
F to C^{0}
 $C^{0} = (F - 32) \times \frac{5}{9}$
 $C^{0} = (95^{0} - 32) \times \frac{5}{9}$
 $C^{0} = 63 \times \frac{5}{9}$
 $C^{0} = 7 \times 5$
 $C^{0} = 35^{0}$

Change
$$77^{0}$$
F to C^{0}
 $C^{0} = (F - 32) \times \frac{5}{9}$
 $C^{0} = (77^{0} - 32) \times \frac{5}{9}$

$$C^{0} = 45 \text{ x}^{5}/_{9}$$

 $C^{0} = 5 \text{ x} 5$
 $C^{0} = 25^{0}$

ACTIVITY

Change from Fahrenheit to Celsius

- 1. 68^{0c}
- $2. 32^{0c}$
- $3. 41^{0c}$
- 4. 77^{0c}

GENERAL EVALUATION

SELF EVALUATION	STRENGTH	WEAKNESS	WAY FORWARD

WEEK 6 LESSON 1

			No.	No. of pupils			Teacher's name
Date	class	subject				Time	
			В	G	T		

HEAT TRANSFER

How heat travels through the states of matter

- 1. Solids by Conduction
- 2. Liquids by Convection
- 3. Gases by Convection
- Vacuum by Radiation

Vacuum

A vacuum is a space without molecules.

Importance of molecules in heat transfer

• Molecule act as a medium of heat transfer.

Qn: In which state of matter does heat travel?

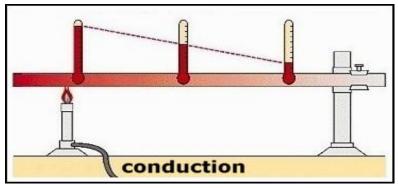
A) **Fastest** – gases

Why? The molecules in gaseous state move freely than in any other state of matter.

Slowest – Solids

Why? The molecules do not move freely.

An illustration of heat transfer in solids



Qn: 1. how does heat move from point B to point A? By conduction

2. Which of the above wax will melt first?

Wax 1.

Reason: Wax 1 is nearest to the flame

3. Which of the above wax will melt last?

Wax 4

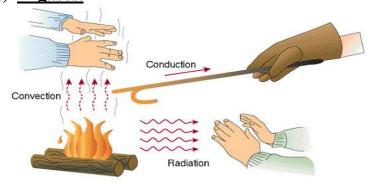
Reason: wax 4 is furthest from the flame.

Importance of heat transfer by conduction

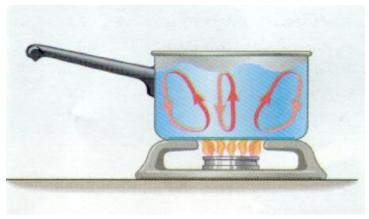
- Ironing our clothes using hot bodies like flat irons, iron boxes.
- Cooking food.
- Boiling water in a kettle.
- Welding or smelting metallic objects.
- Roasting meat using an iron rod.

Diagram to show convection of heat

a) in gases.



b) <u>liquids</u>



Importance of convection in our daily life

- helps smoke to move out of the kitchen through the chimney.
- Convection current helps in free circulation of fresh air in our houses.
- Convection currents help in formation of breezes.
 - Taking of bad smell through the vent pipes of a VIP latrine.

Importance of each of the following on a house.

Ventilators: let out warm air
 Windows: let in fresh air

3. Doors: let in fresh air

ACTIVITY

- 1. How does heat travel through the following
 - a) Solids?
 - b) Liquids?
 - c) Gases?
 - d) Vacuum?
 - 3. in which state of matter does heat travel
 - a) fastest
 - b) slowest

GENERAL EVALUATION

OBIVERURE EVIRENTION							
SELF EVALUATION	STRENGTH	WEAKNESS	WAY FORWARD				

LESSON 2

Date	class	subject	No. of pupils		ils	Time	Teacher's name
		3	В	G	T		

RADIATION.

It is the process by which heat passes through a vacuum.

QN: How does a person standing in Namboole on sunny day receive heat from the sun?

By radiation.

Importance of heat transfer by radiation in the environment

- Radiation is used while roasting meat, fish or chicken in an oven.
- Warming our bodies using warmers or heaters.
- Dries harvested crops / wet clothes on wires.

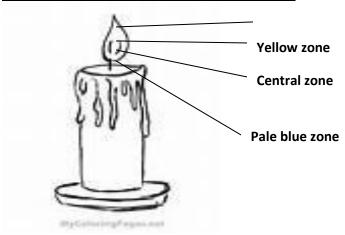
BURNING

Burning is a chemical reaction in which heat and light are produced.

Oxygen gas is needed to support burning.

Carbon dioxide gas is given off during the process.

ZONES OF A BURNING CANDLE.



Blue zone

• Complete burning takes place since it gets enough oxygen.

Yellow zone

- It has partly burnt gas.
- It gives a lot of light.
- It is the brightest part of the flame.

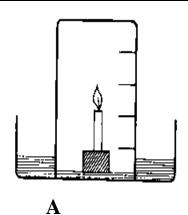
Central zone

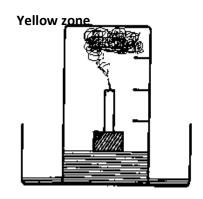
- It gives no light.
- There is no burning since oxygen does not reach it.

Pale blue zone

It is very hot since it gets enough oxygen.

EXPERIMENT ABOUT BURNING





B

Water increased in B to occupy the space of oxygen.

WAYS OF PUTTING OUT FIRE.

- Using a fire extinguisher.
- Using sand.
- Use of water.
- Use of fire blankets.

NOTE

It is not advisable to put out fire caused by **petrol using water because petrol floats on water and burning continues**.

GENERAL EVALUATION

SELF EVALUATION	STRENGTH	WEAKNESS	WAY FORWARD

LESSON 3

			No. of pupils		ils		Teacher's name
Date	class	subject				Time	
			В	G	T		

Reflectors and absorbers or heat.

Reflectors: are shiny surface that reflect heat and light

Absorbers: are dull / black surface that absorb heat and light.

Examples of reflectors

- Mirrors

- Glasses

Absorbers

- Black clothes
- Black cars

Why are most houses, vehicles and fridges in most tropical countries like Uganda painted white? **To reflect heat**

If John washed a black and a white shirt and spread under sunshine;

a. Which shirt would dry first? **The black shirt.**

Reason: Black absorbs a lot of heat.

Last? The white shirt

Reason: The white shirt would reflect heat.

CONDUCTORS AND INSULATORS OF HEAT

Conductors of heat

Conductors of heat are materials which allow heat to pass through them easily.

Examples of good conductors of heat

- i. Iron
- ii. Mercury
- iii. Aluminium
- iv. Brass
- v. Zinc
- vi. Copper
- vii. Silver
- viii. Lead

- Aluminium metal is used to make saucepans since it is light to carry and cannot easily rust.
- Copper is commonly used conductor because it is cheap.
- Silver is the best conductor though it is expensive.

GENERAL EVALUATION

SELF EVALUATION	STRENGTH	WEAKNESS	WAY FORWARD

LESSON 4

Date	class	subject	No. of pupils				No. of pupils		oils	Time	Teacher's name
			В	G	T						

Use of conductors

- Used to make saucepans
- Used to make kettles
- Used to make bottoms of iron boxes/ flat irons.

Insulators of heat

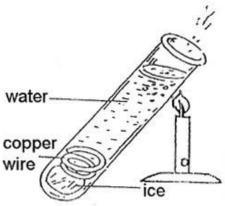
Insulators are the materials which don't allow heat to pass through them easily.

Examples of bad conductors of heat

- Distilled Water
- Rubber
- Plastic
- Paper
- Cotton wool
- Cloth

Sponges.

Experiment to show that water is a poor conductor of heat



Observation

- Ice did not melt
- The water will boil at the area being heated while the ice cubes at the bottom will remain unmelted.

Uses of Insulators

- Paper is used to make cards
- Cloth keeps us warm
- Cork prevents heat loss in a vacuum flask.

Why are handles of iron boxes, frying pans, flat irons made of wood, rubber / plastics?

To prevent the user from getting burnt.

GENERAL EVALUATION

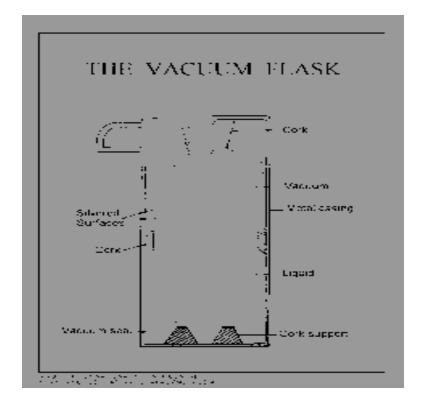
STRENGTH	WEAKNESS	WAY							
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	STRENGTH								

LESSON 5

			No. of pupils		Teacher's name
Date	class	subject		Time	

	В	G	T

A FLASK **VACUUM**



(THERMOMETER FLASK)

A vacuum flask keeps hot things hot and cold things cold.

A DIAGRAM OF A VACUUM FLASK

Uses of each part of a vacuum flask

Cork: Prevents heat loss or gain by conduction.

Silvered surfaces: prevents heat loss or gain by radiation (a good reflect of heat).

Vacuum: Prevents heat loss or gain by both conduction and convection.

Felt (cork base): Absorbs shocks to prevent the glass from breaking.

Felt are also poor conductors of heat.

Vacuum seal: Prevents matter form entering the vacuum.

ACTIVITY

- 1. Of what importance is the thermos flask at home?
- 2. Why is a vacuum seal important in thermos flask?
- 3. What is the use of the cork on the vacuum flask?
- 4. Why are the walls of a vacuum flask double silvered?

MIXTURE

A mixture is a combination of two or more substances.

Example

When we mix sand and cement we get concrete blocks

Dissolving substance/ solutes.

These are substances that dissolve when put in water/ any liquid after stirring. So, solutes are soluble in water or the liquid

Examples of solutes.

- Sugar
- Salt
- Milk powder

Insoluble substances.

These are substances which don't completely dissolve in any liquid or water. **e.g.** sand, maize flour, stones etc

A suspension: is a substance where the solute does not completely dissolve in the solvent.

Solvent.

A solvent is a substance in which a solute dissolves.

Examples.

- water
- petrol
- methylated spirit

NB: Water is universal solvent because it dissolves almost all solutes.

GENERAL EVALUATION

SELF	STRENGTH	WEAKNESS	WAY
EVALUATION			FORWARD

WEEK 7 LESSON 1

			No. of pupils		ils		Teacher's name
Date	class	subject				Time	
			В	G	T		

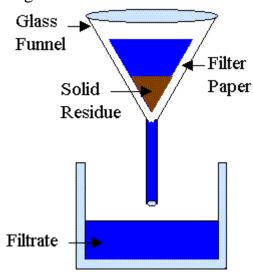
SOLUTION.

A solution is a uniform mixture of a solute and solvent. **saturated solution** is a solution which dissolves more solutes after heating it. **Super saturated solution** is a solution which cannot dissolve any more solute even after heating it.

SEPARATION OF MIXTURES.

a. Filtration.

It is a process of separating solid particles from a liquid using a filter. The solid particles that remain in the filter are called **residues**. The liquid that passes through is called **a filtrate**.



How filtration method can be applied in our homes.

- b. When separating passion, orange, etc fruit juice from the seeds.
- c. When separating clear water from muddy water.

GENERAL EVALUATION

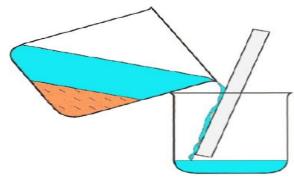
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SELF	STRENGTH	WEAKNESS	WAY						
EVALUATION			FORWARD						

LESSON 2

Date	class	No. of pupils		No. of pupils		Time	Teacher's name
			В	G	T		

2. Decantation.

It is when solids particles are allowed to settle at the bottom of a liquid then a liquid is carefully poured off to separate it from the solid particles.



The solid particles that remain in the one container is called a **residue**. The liquid that is collected in another container is called a **decantee**.

How decantation method can be applied in our homes.

- 1. When separating passion, orange, etc fruit juice from the seeds.
 - d. When separating clear water from muddy water.

Qn: Why is it important for one to first shake a syrup before taking it? For the medicine to mix with the liquid that has settled at the top.

4. Distillation.

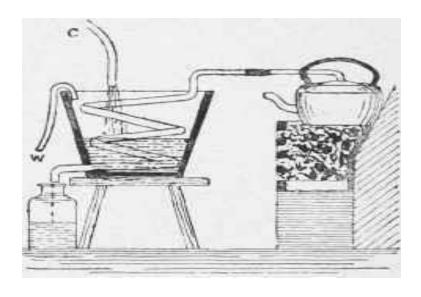
It is the making of dirty water pure by evaporation and condensation.

Nb: Distilled water is not good for drinking because it lacks mineral salts.

The substance that remains in the container is called a residue.

The substance that is collected in the second container is called a distillate.

Distilled water is used to mix drugs in hospitals.



SEPARATION OF MIXTURES OF SOLIDS.

1. Floatation.

It is a method of separation of mixtures where one sinks and the other floats.

- 2. Using magnets to separate metallic objects from non metallic objects.
- 3. Hand picking / sorting.

GENERAL EVALUATION

SELF EVALUATION	STRENGTH	WEAKNESS	WAY FORWARD

LESSON 3

Date	class	subject	No. of pupils		ils	Time	Teacher's name
			В	G	T		

CROP GROWING

ROOT CROPS

Root crops are crops that store their food in the swollen underground roots.

Examples of root crops

- Cassava
- Sweet potatoes
- Carrots
- Yams

STEM TUBERS

Are crops which store their food in swollen underground stems.

Examples of stem tubers

- Coco yams
- Irish potatoes

Which food values do we get from eating mostly root crops.

We get carbohydrates.

How are the following crops propagated?

Crop	Propagation
Cassava	By planting stem cuttings
Sweet potatoes	By planting vines / stem cuttings
Carrots	By planting carrot seeds
Yams	By planting the roots
Pineapples/sisal/banana	By planting suckers

CARING FOR ROOT CROPS

1. By weeding the root crops (removing unwanted crops from the garden)

Importance of weeding

- To reduce competition for sunlight, water and mineral salts between weeds and crops.
- To prevent easy spread of pests and diseases from the weeds to crops
- Improve the crop yields
- 2. Pruning the crops (cutting off excess or poorly growing plants)

Importance of pruning

- To reduce over weight on the plants
- To prevent competition for sunlight carbon dioxide and oxygen
- Pruned materials can be used as mulches in the garden

Garden tools used for pruning crops

- Pruning saw
- Secateurs

3. **Thinning** (removal of excess or poorly growing seedlings / crops in a nursery bed /garden)

Importance of thinning

- Creates space for the crops to grow well.
- Improves on the yields.
- There is less competition for nutrients from the soil
- Reduces hiding places for the crop pests
- The crops are easily sprayed with pesticides.

How is thinning done in the garden

- By uprooting unwanted crops manually
- Watering the crops (providing water for the plants to prevent withering wilting /drying up)
- The garden tool used for carrying water and watering the crops is called the watering can.

NB: We can also use an over head sprinkler or dig channels to allow the flow of water to the crops in the garden.

4. Plant training /staking.

Staking is the providing of support to plants with weak stems.

Examples of crops that can be trained /staked

- Tomatoes
- Passion fruits
- Vanilla

Why staking?

- ❖ To control ground pests.
- ❖ To ease weeding.
- ❖ To ease harvesting.
- ❖ To improve access to sunlight.

ACTIVITY

- 1. What are root crops?
- 2. Give three examples of root crops.
- 3. What are stem tubers?
- 4. Why is a sugar cane not grouped under stem tubers?
- 5. How are the following crops propagated

- a) cassava
- b) sweet potatoes
- c) bananas

GENERAL EVALUATION

SELF EVALUATION	STRENGTH	WEAKNESS	WAY FORWARD

LESSON 4

			No. of pupils		ils		Teacher's name
Date	class	subject				Time	
			В	G	T		

ROOT CROP PESTS AND DISEASES

Pests:

Pests are living organisms that destroy crops e.g. birds, worms, insects, rodents etc **Diseases:**

Diseases are illness /sickness in living organisms i.e. plants and animals.

A TABLE SHOWING THE PESTS AND DISEASE THAT AFFECT VARIOUS ROOT CROPS

No	Root crops	Pests	Diseases
1.	Cassava	Whitefly, monkeys,	Cassava mosaic, leaf
		cassava scale, green	spot, bacterial blight
		cassava mites,	
		caterpillars, wild pigs,	
		rodents e.g. moles	
2.	Sweet	Eel worms, caterpillars,	Potatoes blight,
	potatoes	sweet potatoes, weevils,	bacterial wilt sweet
		wild pigs, monkeys,	potato mosaics
		rodents	
3.	Carrots	Aphids, moles, eel	Bacterial wilt, leaf
		worms, cut worms, root	spot, a leaf rust

		knot, nematodes	
4.	Yams	Termites, Locusts, Mole,	Leaf spot and leaf rust
		Yam beetles	(fungi)

CHARACTERISTICS OF ROOT CROP PESTS

- Pests which damage the shoot system (leaves and stem /braches) have strong mouth parts to cut and chew the leaves.
- Pests which destroy the tubers have sharp claws which help them to dig the soil.
- They have sharp incisors which bite or cut the roots /tubers.
- Other pests that damage root crops have fingers which they use to uproot the root a crop e.g. apes and monkeys.

Examples

- Locusts
- Caterpillars
- Army worms
- Sweet potatoes weevils
- Variegated grasshoppers

HOW TO CONTROL CROP PESTS

- By spraying the crops with pesticides.

Dangers of pests to crop farmers

- Reduce on the crop yields
- Cause decaying root crops
- Wastes money to control

Uses of pests to farmers

- Some pests are a source of food to man e.g. grasshoppers and locusts.
- Some are eaten by farmer's poultry e.g. caterpillars

ACTIVITY

- 1. What is thinning?
- 2. Of what importance is thinning to a crop farmer?
- 3. Identify any one garden tool for pruning crops.
- 4. What are pests?
- 5. Give one way of controlling pests.

GENERAL EVALUATION

SELF EVALUATION	STRENGTH	WEAKNESS	WAY FORWARD	

LESSON 5

Date	class	subject	No. of pupils		ils	Time	Teacher's name
			В	G	T		

FOUR MAJOR CONTROL METHODS OF PESTS

1. Mechanical control method

E.g.

- physical guarding (Fencing the garden)
- Silting traps /scares
- Staying scary crows

The above methods can control pests like wild pigs, moles, birds, rodents etc

2. Biological pest control

This is where a predator is used to control the pests e.g. you can tame a cat to kill rats.

3. Cultural methods

- By practicing crop rotation
- The available pests of a particular crop
- Early planting and harvesting
- Practicing resident species /varieties
- Through proper control of weeds
- By planting disease free cuttings /vines
- Chemical control method; is a method where a farmer sprays pesticides /insecticides to kill the pests

HARVESTING AND STORAGE OF ROOT CROPS

Harvesting

Harvesting is the removal of mature and ready crops from the garden.

Storage

Storage is the keeping of harvested crops safely for future use

Sweet potatoes

- Mature within the first six months after planting.
- However, first growing varieties mature within three months
- Potato tubers do not mature at the same time, but keep growing as you harvest the ready /mature tubers.
- Ready sweet potatoes make cracks in the soil
- Farmers spot the cracks and harvest the ready tubers using sharp sticks or sharp pointed metal rods.
- Sweet potatoes tubes can as well be harvested at once using hoes mostly for scale.
- Harvested tubers can be peeled, cooked and eaten immediately
- The surplus can be sold in market or sliced into tiny pieces, dried and stored in cool /dry places.

Cassava

- Some varieties take 6 months to mature others take up to 1 year.
- During harvesting the whole plant is dug out using a hoe and tubers removed
- While preparing cassava for eating, you peel using a knife, wash, cook and eat.
- Surplus cassava can be sold off in markets or slices and dried before storage or dried slices can be pounded to make cassava bread or pan cakes, local beer (kwete), local glue.

Methods of storing root crops

- Temporary storage e.g. burying the tubes under wet soil
- Long time storage (after drying the slices) e.g. storing in granaries /sacks /slices (modern stores) silos

ACTIVITY

- 1. What is harvesting?
- 2. Why is dry season important to crop farmers?
- 3. Suggest one biological method of controlling crop pests.
- 4. Mention one method of harvesting crops.
- 5. Name the modern method of storing harvested crops.

GENERAL EVALUATION

SELF EVALUATION	STRENGTH	WEAKNESS	WAY FORWARD

WEEK 8 LESSON 1

			No. of pupils		ils		Teacher's name
Date	class	subject				Time	
			В	G	T		

KEEPING AND USING FARM RECORD

Farm records

Farm records are written documents about all farm activities.

Example of farm records used on root crops

- Farm inventory e.g. farm equipment size of lands.
- Cash record e.g. money spend or received when carrying out the project
- Non-cash records unpaid for labour may be family members
- Production records e.g. number of acres planted /map of the farm

Importance of keeping farm records

- To know the progress of the farm
- Proper records can be used to get loans in the bank
- To plan for the farm for future use
- To know whether the farmer is making profits or losses.

SPECIMEN OF FARM RECORDS SHEET

Date	Crop planted	Date of weed control	Date of manuring	Date of harvest	Date of sale
2/2/2005	cassava	3/3/2005	3/4/2005	7/7/2005	13/5/2005

SCIENCE CLUBS /SOCIETIES IN THE SCHOOL The science club

Involve learners in science related activities

Examples of Science related societies.

- Wildlife clubs
- Red cross clubs
- Young farmers clubs
- Science contest
- Science exhibition
- Science quizzes
- Science projects
- Science paper presentation etc

GENERAL EVALUATION

SELF EVALUATION	STRENGTH	WEAKNESS	WAY FORWARD

LESSON 2

Date	class	subject	No. of pupils		ils	Time	Teacher's name
		J	В	G	T		

Importance of science activities to learners

- Develop a positive attitude in learners towards science.
- Expose learners to the kind of work that can lead to their career
- Shape learners for their future career in the science field i.e. doctors, engineers, surgeons, dentists, electricians, and agriculturalist.

Wildlife club

Enables learners to learn more about uncultivated plants and wild animals that exist in the natural environment.

Roles of the wild life club

- Protecting the environment.
- Conserving the environment.

- Teaching or sensitizing others to protect or conserve the environment.

Activities done by the wildlife clubs in school

- Maintaining a free nursery in the school
- Tree labeling
- Establishing wood / tree projects
- Monitoring wildlife abusers e.g. poachers
- Bird watching
- Preventing water, air and soil pollution
- Setting up a botanical garden
- Construction of an aquarium

Aims of the science oriented clubs in school

- To promote /boost children's interests in the science subject
- To enable children to know how scientists work.
- To equip learners with knowledge and skills for their future career.
- To promote learning of science in the school.

The young farmers clubs

The young farmers clubs include young boys and girls in and out of school who are interested in farming.

Roles of the young farmers clubs

- To keep animals and grow crops.
- To teach other farmers better farming methods.

Qn: When are the activities of young farmer's clubs done?

- After school time

Qn: Under which department / ministry in Uganda are the young farmers clubs

- Department of agriculture

Role of the department of agriculture in young farmers clubs

- i. To unite all the young farmers' clubs in the country.
- ii. To set up competitions in school where prizes are given to winners.
- iii. Train and send technical people to teach and answer questions of the young farmers.
- iv. To teach young farmers better farming methods.
- v. To start money making and savings projects for self support.

Topical questions

1. How can root crops be cared by crop farmers

- 2. Give any two examples of stem tubers
- 3. State the importance of farm records on a crop farm
- 4. Identify the main role of the wildlife club in a school like greenhill
- 5. Which is the best season for crop growing
- 6. How are the following crops propagated

Sweet potatoes

Cassava

Carrots

- 7. Which food value is mostly obtained from root crops
- 8. Give the difference between a pest and a parasite.

GENERAL EVALUATION

SELF EVALUATION	STRENGTH	WEAKNESS	WAY FORWARD

LESSON 3

Date	class	subject	No. of pupils		ils	Time	Teacher's name
			В	G	T		

BACTERIA AND FUNGI

Bacteria are tiny organisms made up of one cell.

They are viewed using a microscope.

That is why they are called microscopic.

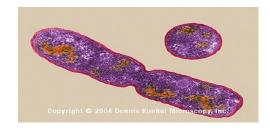
Places where bacteria are mostly found

- Animal bodies
- Contaminated water
- Soil
- Nodules of legumes
- Latrines
- Rubbish pits
- Air
- Contaminated food

How do bacteria reproduce

1) By cell division (binary fission)

Diagram



Types of bacteria

Bacteria are grouped according to their shapes.

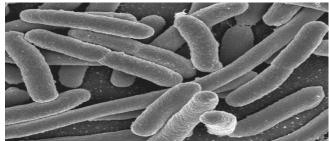
- Bacilli
- Cocci
- Vibrio
- Spirilla

Bacilli

Bacilli bacteria are rod shaped bacteria.

Examples

- ❖ Salmonella typhi ----- typhoid
- ❖ Bacillus anthracis-----anthrax

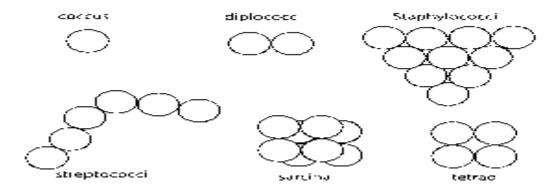


Cocci

Cocci bacteria are spherical bacteria.

Examples

- staphylococci --- boils
- streptococcus --- sore throat
- * diplococcus-----pneumonia
- ❖ Gonococcus-----gonorrhoea



Spirilla/spirochaete

Spirilla bacteria are bacteria that are strongly curved.

Example

Treponema pallidum -----Syphilis



Vibrio

Vibrio bacteria are comma shaped bacteria.

Example

❖ Vibrio cholerae -----Cholera



Respiration of bacteria

- ❖ Bacteria which require oxygen to respire are called **Aerobic bacteria**.
- ❖ Bacteria which do not need oxygen to respire are called **Anaerobic** bacteria.

ACTIVITY

- 1. What are bacteria?
- 2. How do bacteria reproduce?
- 3. Identify one place where bacteria can be found.
- 4. In which way is reproduction in bacteria similar to that in fungi?
- 5. which type of bacteria course the following disease
- a) Gonorrhea?
- b) Syphilis?
- c) Cholera?

GENERAL EVALUATION

SELF EVALUATION	STRENGTH	WEAKNESS	WAY FORWARD

LESSON 4

			No. of pupils		ils		Teacher's name
Date	class	subject				Time	
			В	G	T		

Importance of useful bacteria

- 1. Nitrogen fixing bacteria fix nitrogen back into the soil to improve soil fertility
- 2. Bacteria break or digest fibre food/roughage / cellulose in the caeca of birds
- 3. Bacteria help in decomposition / rotting of dead plants and animals
- 4. Bacteria help too reduce on the amount of faeces in pit latrines as well as sewage tanks
- 5. Bacteria help in fermentation of beer, ghee, yoghurt and cheese.
- 6. Bacteria help in making humus (manure) in compost pits)
- 7. Bacteria help to reduce on the amount of garbage as they rot.

Dangers of harmful bacteria

- 1. Harmful bacteria cause diseases to both animals and plants.
- 2. Bacteria cause poor yields to crops
 - 4. Bacteria cause food poisoning
 - 5. contaminates food and makes it poisonous to human health
 - 6. Some cause diseases.

How to control dangers caused by harmful bacteria

- i. by preserving food(keeping food free from bacteria)
- ii. by observing proper food hygiene i.e. ensuring that food is kept in a clean environment and clean containers
- iii. by using antiseptics(drugs that kill germs on cuts and wounds)e.g. detol, saloon, hydrogen peroxide tincture of iodine etc
- iv. by sing disinfectants to kill bacteria in places like bacteria and reduce the bad smell e.g. harpic, jeyz, etc
- v. by using antibiotic (drugs used to kill bacteria in our bodies) inform of injections, tablets, capsules, syrups e.g. penicillin from penicillium) that was discovered by Alexander Fleming, etc
- vi. through sterilization of medical instruments e/g springs
- vii. through pasteurization (involves) heating the food stuffs to a high temperature and sealing it before suddenly cooling it)

it was discovered by Dr. Louis Pasteur to preserve milk. He also discovered the vaccines against anthrax and rabies.

GENERAL EVALUATION

SELF	STRENGTH	WEAKNESS	WAY
EVALUATION			FORWARD

LESSON 5

Date	class	subject	No. of pupils		oils	Time	Teacher's name
	- Class	susject	В	G	Т		

FUNGI

Fungi are simple organisms that obtain their food from decaying plants and animal matter.

Fungi are saprophytes because they feed saprophytically (obtain soluble food from dead organic matter)

Examples of fungi

- Mushrooms
- Moulds
- Yeasts
- Puffball
- Toad stools (poisonous fungi)

Characteristics of fungi

- Fungi exist in both as single celled (unicellular) or multicellular organisms.
- Fungi lack chlorophyll (they can make their own food)
- Fungi feed saprophytically or parasitically
- They have nuclei in their cells

Importance of fungi

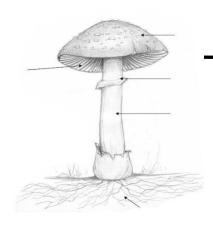
- Some fungi like yeast are used to bake bread and cakes
- Some fungi like mushrooms are a source of food to man
- Some fungi like moulds (penicillium rotatum) are used to make medicine penicillium
- Yeast is used to brew local beer or ferment fruit juices to make wine
- Fungi help in decomposition of rubbish in rubbish pits to make humus
- Yeast is used to flavors cheese
- Yeast is a source of vitamin B that prevents beriberi

MUSHROOMS

Cap

Ring

Stalk /stipe



- They are
- Some

purpose.

IMPORTANCE OF MUSHROOMS.

Hyphae

eaten as food.

mushrooms are sold to get money.

- Mushrooms are used for study

Activity

- 1. What are fungi?
- 2. How do fungi reproduce?
- 3. Give examples of fungi.
- 4. Draw a mushroom and name the cap, gills and stipe.

GENERAL EVALUATION

SELF EVALUATION	STRENGTH	WEAKNESS	WAY FORWARD

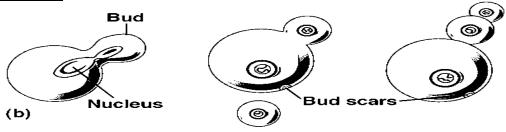
WEEK 9 LESSON 1

_			No. of pupils		oils		Teacher's name
Date	class	subject				Time	
			В	G	T		

How do fungi reproduce?

Most fungi reproduce by means of **spores** unlike Yeast which reproduces by **budding**

Diagram



Danger of harmful fungi

- Harmful fungi cause food poisoning
- Harmful fungi cause a number of diseases to both plants and animals

Examples of fungal diseases in animals

- Ring worms (cause round patches on the skin)
- Athletes foot (attacks the skin between the hoes)
- Thrush caused by candida
- Finger nail deformation

Examples of fungal diseases in plants

- Panama e.g. banana
- Cigarand rot
- Blast wilt
- Coffee berry disease (rust) coffee plant
- Root rot in tea plants.

GENERAL EVALUATION

SELF	STRENGTH	WEAKNESS	WAY
EVALUATION			FORWARD

		1

LESSON 2

Date	class	subject	No.	No. of pupils		Time	Teacher's name
			В	G	T		

How to control the dangers caused by harmful fungi

- Boiling milk and water before drinking.
- Reheating and warming cold food before eating it.
- Salting the food.
- Pickling putting vinegar in edible vegetables and other foods.
- Avoid eating uncovered food.
- Spray plants with fungicides.
- Get early treatment for any fungal infection /disease.
- Sterilize all surgical instruments to kill all the germs.
- Irradiation in canned foods.
- Avoid eating or catching poisonous fungi.
- Proper management of house refuse and rubbish
- Proper use of the latrine.
- Avoid sharing towels, socks, under wears etc.
- Use medicated soap to bathe e.g. detol, protex, etc.
- Use disinfectants in cleaning toilets e.g. jeyz.

Similarities between bacteria and fungi

- Both feed saprophytic ally.
- Both cause fermentation.
- Both can cause rotting /decomposition.
- Some of them cause diseases while others are useful to man.

Differences between bacteria and fungi

- Bacteria reproduce by means of cell division (binary fission) while fungi reproduce by means of budding and spores.
- All bacteria are very tiny while some fungi like mushrooms are big.

EVALUATIO		RENGTH	WEAK	NESS	WAY FORWARD
LESSON 3			No. of pupils		Teacher's name
Date	class	subject	B G T	Time	
Topical question 1. How are back 2. Where are b	eteria differ				
3. Name any o	ne poisonoi	ıs fungi			
4. Give the thre	e types of t	acteria			

c) Cause typhoid			
d) Cause cholera			
6 Identify any two in	nmunisable disease ca	used by bacteria	
7 How do the followa a) Yeast			
b) Bacteria			
8 Give the importanc			
9 How are bacteria u	seful to people?		
10 In the space below	v, draw a mushroom a	nd label all the parts	
GENERAL EVALU		WEARNEGG	XX7.4.X7
SELF EVALUATION	STRENGTH	WEAKNESS	WAY FORWARD

GEN.WEAKNESSES	GEN. STRENGTH	AREAS MISSED OI NEED MORE REVISION	

