

EAGLE'S NEST SECONDARY SCHOOL KAMPALA Uganda Certificate of Education

COMPETENCE BASED ASSESSMENT EXAMINATIONS 2023 CHEMISTRY

S.2

TIME: 2RS

NAME:STREAM......

INSTRUCTIONS:

- Answer **all** questions in the spaces provided.
- Illustrations in form of drawings should be made where necessary, with a sharp pencil.

SECTION A

1. (a) The figure below shows some examples of important chemical reactions in our daily life.

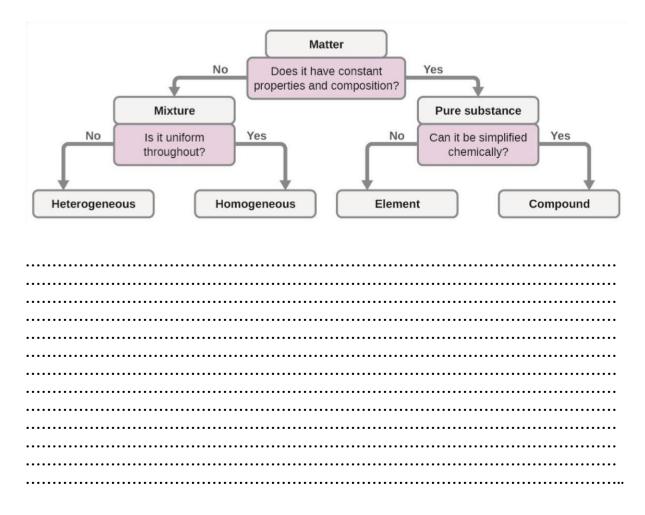


| (i) | Briefly explain the meaning of the chemical reactions sl | hown above (04 marks) |
|-------|--|---|
| Coml | oustion | |
| | ••••• | • |
| Rust. | | |
| | ••••• | |
| /ii\ | State one condition of each process to occur | (02 marks) |

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| | pustion | | |
|-------|--|---|---|
| (iii) | How can you prevent your metallic Equipments from rusting? | • | |
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2. As a senior two chemistry student, make a write up giving meanings and examples of the bolded words in the boxes. (10 marks)



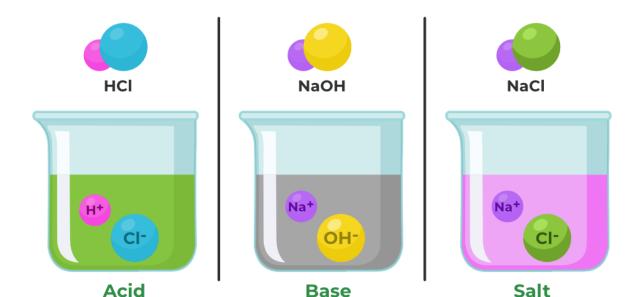
4(a). Kinetic theory of matter states that "matter consists of very smallest invisible particles in the state of continuous random motion" A teacher instructed S.2 students to come up with every day examples demonstrating the existence of particles in matter.

Owen; a S.2 student presented his example before the whole class. His presentation was "when someone closes him or herself in a dark room with closed windows and doors and looks into a

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ray of light penetrating through one simple hole in one piece of iron sheet, dust particles are seen moving in a zig zag pattern "the teacher confirmed Owen's findings.

| (i) State why the dust particles are seen moving in zig zag pattern.? (01 mark) |
|---|
| (ii) Give a reason for your answer in (a) (01mark) |
| (iii) What conclusion can be drawn from Owen's simple experiment? (01 mark) |
| (iv) State what would be observed within the ray of light when the room gets hotter? (01 mark |
| (v) Explain your observations in (c) above. (02 mark) |
| |



Answer the questions below

5. From the support material below.

(a)Using the support material above, with relevant examples, Define;

i). An acid(2mks)

| ••••• | | | | | | | | | | | |
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| | | | | | | | | | | | |
| Examples(3mks) | | | | | | | | | | | |
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| | | | | | | | | | | | |
| •••••• | | | | | | | | | | | |
| ii). A bo | ii). A base. (2mks) | | | | | | | | | | |
| ••••• | | | | | | | | | | | |
| ••••• | | | | | | | | | | | |
| Exampl | Examples. (2mks) | | | | | | | | | | |
| ••••• | | | | • | • | ••••• | | | | | |
| ••••• | | | | | ••••• | ••••• | | | | | |
| b). Out | line any three | indicators foun | ıd in the labora | atory. (3mks) | | | | | | | |
| • | b). Outline any three indicators found in the laboratory. (3mks) | | | | | | | | | | |
| • | | | | | | | | | | | |
| ••••• | | | | | | | | | | | |
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| /b)\\/rita | (b)Write down the symbols of the following elements (04 marks) | | | | | | | | | | |
| 1 | <u> </u> | | 1 | , | | | | | | | |
| Element | Bromine | Silver | Potassium | Nitrogen | Aluminium | Silicon | | | | | |
| Symbol | | | | | | | | | | | |

END

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