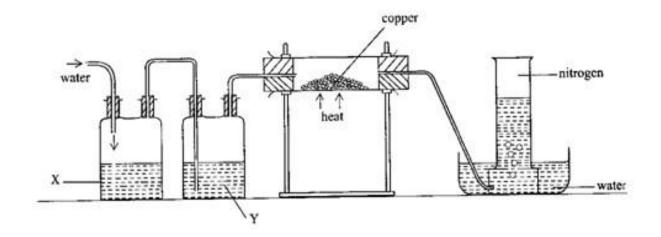
CHEMISTRY DEPARTMENT 2023 S.4 BRAINSTORMING TEST

TOPIC; <u>NITROGEN AND ITS COMPOUNDS</u> SUBTOPIC; <u>NITROGEN</u>

NAME	INDEX number	
Signature		
 Although nitrogen is generally unreactive, i magnesium ribbon 		
(a) State why nitrogen is generally inert	(01 mark)	
(b) Burning magnesium reacts with nitrogen		
(i) Give a reason for the reaction	(1½ marks)	
(ii) Write equation for the reaction	(1½ marks)	
(c) Water was added to the product in (b) and a co		
(i) Name T	(0½ mark)	
(ii) Write equation for the reaction leading	to the formation of T $(1\frac{1}{2} \text{ marks})$	

(d) A piece of litmus paper was held over the reaction tube (i) State what was observed	e in (b). (0½ mark)
(ii) Explain your answer in (c) (i)	(1½ marks)
(iii) Name one other substance that can react with nitrogen the sa magnesium (0)	ame way as I mark)
(a) (i) Name two major components of air and state their percentage composition by volume.	corresponding (01 mark)
(ii) Mention the method that can be employed to separate the connamed in a(i) . Give a reason for your answer.	nponents of air (01 mark)
	••••••

(b) Study the setup of diagram below and use it to answer the questions.



	(i)identify solutions \mathbf{X} and \mathbf{Y}	(01 mark)
	X	
	Y	
	(ii)Write the equation for the reaction that took place in a wash containing solution \mathbf{Y}	(1½ marks)
•••••	(c) (i) State what is observed in the combustion tube.	(01 mark)
•••••	(ii) Write the chemical equation for the reaction that took place	
•••••		 END.