

**545/3**  
**CHEMSITRY**  
**PRACTICAL**  
**Paper 3**  
**INSTRUCTIONS**  
**Jul/Aug 2022**



## **BUSOGA REGION JOINT EXAMINATION BOARD**

### **Uganda Certificate of Education**

#### **CHEMISTRY PRACTICAL INSTRUCTIONS**

##### **Paper 3**

#### **CONFIDENTIAL**

Great care should be taken that the information given below does not reach the candidates either directly or indirectly.

#### **INSTRUCTIONS FOR PREPARING APPARATUS**

NB. The head teacher must ensure that the teacher responsible for preparing the apparatus hands in his/her trial results properly sealed with in a separate envelope and firmly fastened/attached to the candidates' scripts envelope(s)

1. The description of the reagents and chemicals specified below does not necessarily correspond with description in the question paper. Candidates must not be informed of the difference.
2. Candidates are not allowed to use reference books (i.e. textbooks, booklets on qualitative analysis etc) during the examination.
3. In addition to the fittings and substances ordinarily contained in a chemistry laboratory, each candidate will require.
  - 1 burette (50ml)
  - 1 pipette (20 or 25ml)
  - 2 conical flasks (250ml)
  - 1 measuring cylinder (50ml)
  - 1 filter paper
  - 6 test tubes
  - 100cm<sup>3</sup> of BA<sub>1</sub>
  - 100cm<sup>3</sup> of BA<sub>2</sub>
  - 5g of **F**
  - Methyl orange indicatorAccess to;
  - heat source
  - reagents for identifying cations, anions and gases.
  - **BA<sub>1</sub>** is made by diluting **9.0cm<sup>3</sup>** of concentrated Hydrochloric acid to make one litre of solution
  - **BA<sub>2</sub>** is made by dissolving **19.3g** of Borax to make one litre of solutionSubstance **F** will be provided by **BURJEB**

**END**