Name:			
			- 70
School:	9 x	Signature:	

545/2
CHEMISTRY
Paper 2
(Practical)
November 2024
2 Hours



WAKISSHA JOINT EXAMINATIONS

Uganda Certificate of Education

End of Year Assessment

SENIOR THREE

CHEMISTRY

Paper 2

(Practical)

2 Hours

INSTRUCTIONS TO CANDIDATES;

This paper consists of one compulsory item.

Answers to this item are to be written in the spaces provided in this booklet. Use blue or black ink.

All working must be clearly shown. Graph paper will be provided.

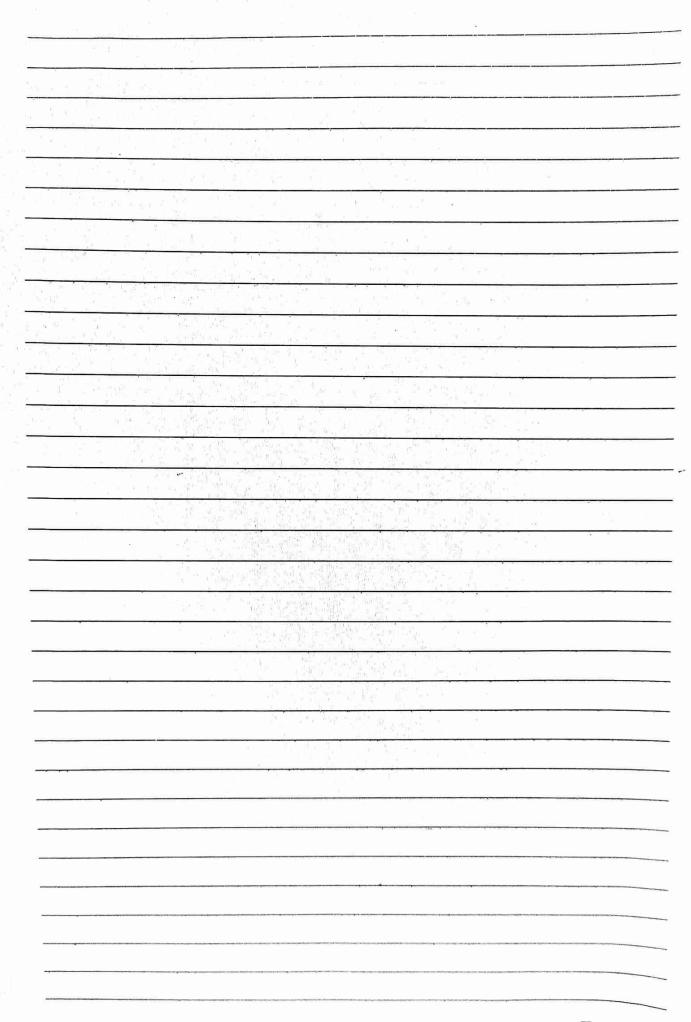
Mathematical table and silent non-programmable scientific calculators may be used.

You are not allowed to use reference books (i.e. text books, booklets on qualitative analysis etc.)

Candidates are advised to carefully read the item, make sure they have all the apparatus and chemicals they may need and then plan appropriately before starting.

Mr. Brown a farmer dealing in cotton has been doing well but of recent his detoriated due to the soils lacking Epsom salt. Episom salt contains magnesium sulphate an important ingredient in this fertilizer root growth and development. Brown is intending to start a local factory to prepare on a small scale by reacting magnesium metal and dilute sulphuric acid but is not the rate of the reaction can be improved at room temperature. Magnesium and sulphuric acid react according to the equation Mg(s) + H ₂ SO ₄ (aq) MgSO ₄ (aq) + H ₂ (g) You have been approached to help Brown and you are provided with the following and which is sulphuric acid solution Which is magnesium	s to stimulate e the fertilizer aware on how
As a chemistry learner, Design an experiment you will use to determine the rate of the reaction for address their concerns (Your answer should include aim, hypothesis, variab and materials, procedure, risks and their mitigations)	he factory to les, apparatus
	Fi Name of the Control of the Contro

Item 1



	× ·
7 <u></u>	
	Y Y Y
(4	
1	
	B TO THE RESERVE OF T

	sults and inform Brown accordingly.
-4	
F3 2	
t v	
	The state of the s
*	
	The second secon

