NAME:

S.5
PCB/SM
BCM/ICT
PCM/ICT
BCA/SM
BCG & FCM

Tick your subject combination

FINAL MARKS

%

For Examiner's Use Only	
01	
02	
03	
04	
05	
TOTAL	

UGANDA ADVANCED CERTIFICATE OF EDUCATION
INTERNAL EXAMINATION PROGRESS –, 2024
CHEMISTRY PAPER 2
(Principal Subject)   **DURATION:** 2 Hours 30 minutes
INSTRUCTIONS TO CANDIDATES:
✓ Attempt *any* **FIVE** questions **a**nd **ALL** the questions carry equal marks.
✓ Incorrect symbols, formulae and spellings of especially technical terms will lead to loss of marks. And **good hand writing** is **paramount** please Boys.
✓ The **periodic table**, with **relative atomic masses**, is **attached at the end** of the paper and **graph paper** will be provided however need it.
✓ Begin answering **each** question on a **fresh page**, Answer **booklets are enough**.
✓ Illustrate your answers with **equations where applicable**, **Non-programmable scientific** calculators may be used.
✓ Assume where necessary, use the following values.
— Molar gas constant, R   = $8.314 \ JK^{-1} \ mol^{-1}$.
— Molar volume of gas at s.t.p   =   22.4 litres.
— Standard temperature, T   =   273.15 K.
— Standard pressure, P   = $101325 \ Nm^{-2}$.
— Molar volume of gas at r.t.p   =   24.0 litres.

This paper consists of 04 pages. You should check the question paper to ensure that all pages are printed as indicated and no questions are missing.