Name	Centre/Index No
Signature	

545/2

CHEMISTRY
Paper 2
July/August 2024
2 Hours



TORORO ARCHDIOCESE EXAMINATION BOARD

Uganda Certificate of Education MOCK EXAMINATIONS 2024 CHEMISTRY PRACTICAL

PAPER 2

2 Hours

INSTRUCTIONS TO CANDIDATES

This paper consists of one compulsory examination item. Answers to the item are to be written in the spaces provided in the booklet. Use blue or black ink.

All working must be clearly shown. Graph paper will be provided.

Mathematical tables and silent non-programmable scientific calculators may be used.

You are not allowed to use reference books (i.e text books, books on qualitative analysis etc)

Candidates are advised to carefully read the item, make sure they have all the apparatus and chemicals they may need and then plan appropriately before starting.

Turn Over

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Item 1

according to the equation: produced by the reaction between sodium thiosulphate and dilute hydrochloric acid various health centres has increased the demand of sulphur. substances for the treatment of skin conditions. The large number of patients in the Most health centres in Uganda use Sulphur when mixed with other chemical Sulphur can be

$$S_2O_3^{2-}_{3-}_{(aq)} + 2H^+_{(aq)} \longrightarrow S_s + SO_{2(g)} + H_2O_{(1)}$$

rate of the reaction. produce sulphur must be adjusted for a better yield of sulphur because it affects the The pharmacists have advised that concentration of the above chemicals used to

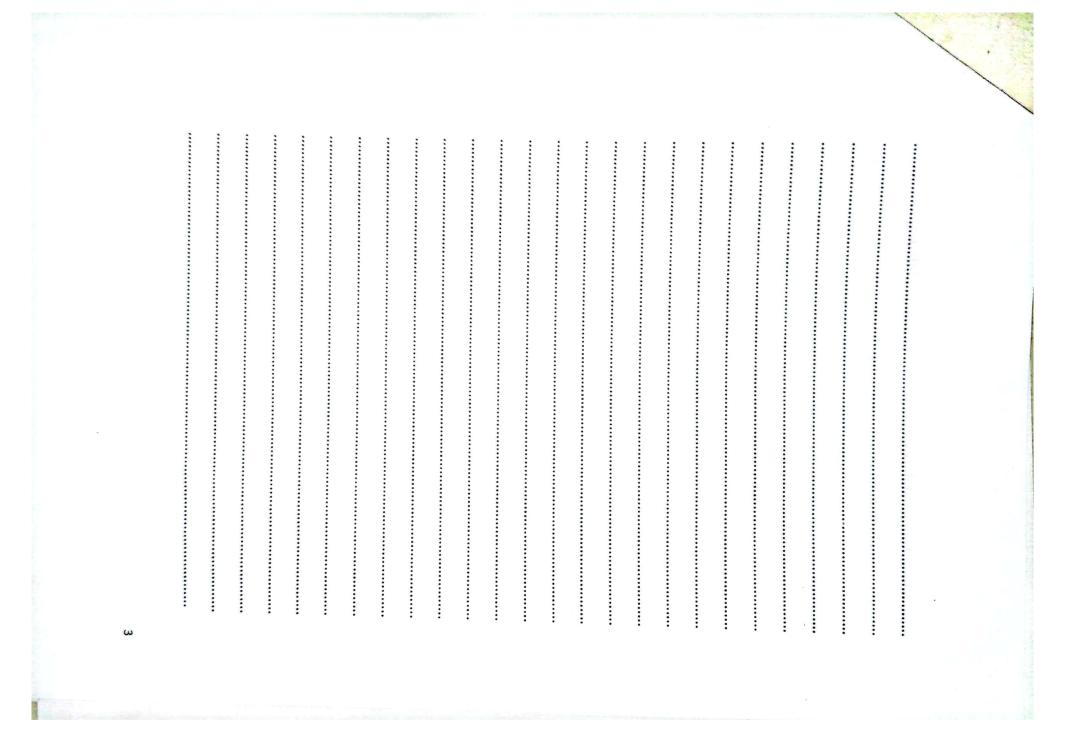
You are provided with:

BA1 which is sodium thiosulphate solution and BA2 which is dilute hydrochloric

Task

As a learner of chemistry;

 \odot concentration on the rate of reaction between BA1 and BA2. Design an experiment you will carry out to determine the effect



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4										ord your findings.		

END		:						(iii) What conclusion can you make from your findings?	