545/2 Inst. Sch. CHEMISTRY PRACTICAL INSTRUCTIONS



MEPSA RESOURCEFUL ASSESSMENT 2024

Uganda Certificate of Education CHEMISTRY PRACTICAL INSTRUCTIONS 545/2 Inst. Sch.

CONFIDENTIAL

This information is given only to facilitate preparation of examination Great care should be taken that the information given below does not reach the candidates either directly or indirectly.

INSTRUCTIONS FOR PREPARING CHEMICALS AND APPARATUS

The head teacher **must** ensure that the teacher responsible for preparing the chemicals and apparatus hands in his/her trail results properly sealed in a separate envelope and **firmly** fastened (attached) to the candidates' scripts envelop(s).

- 1. The description of the reagents and chemicals specified below does not necessarily correspond
- 2. Candidates must **not** be informed of the differences.
- 3. Candidates are **not** allowed to use reference books (i.e. text books, booklets on qualitative analysis etc.) during examination.
- 4. In addition to the fittings and substances ordinarily contained in a chemistry laboratory, each candidate will require:
 - 1 thermometer.
 - 1 measuring cylinder of 50cm³ or 100 cm³.
 - 1 burette.
 - 2 conical flasks.
 - 6 test tubes.
 - 1 pipette of 25.0cm³ or 20.0cm³.
 - 1 stop clock.
 - 5 labels.
 - 100 cm³ of distilled water.
 - 1**50 cm³ of BA1**.
 - **150cm**³ of BA2
 - 130cm³ BA³
 - 3 plastic beakers (250cm³)

Easy access to heat source

Phenolphthalein and methyl orange indicators.

BA1 is prepared by diluting 108.7 cm3 of concentrated sulphuric acid (1.84 g/cm³, 98%) acid with distilled water to make one litre of solution.

BA2 is prepared by diluting 117.65 cm^3 of ethanoic acid $(1.048 - 1.051 \text{ g/cm}^3)$ with distilled water to make one litre.

BA3 is prepared by dissolving 80g of Z, to make one litre with distilled water.

Z will be provided by **school.**'

Z is sodium hydroxide