

NAME: \_\_\_\_\_

P525/1

CHEMISTRY  
PAPER 1  
JUNE 2024  
2 HRS 15 MINS



S5

PCB/SM

BCM/ICT

PCM/ICT

BCA/SM

BCG/SM & FCM

FINAL MARKS



Tick your  
subject  
combination

**UGANDA ADVANCED CERTIFICATE OF EDUCATION  
INTERNAL EXAMINATION PROGRESS –, 2024  
CHEMISTRY PAPER 1**

(Principal Subject)

DURATION: 2 Hours 15 minutes

**INSTRUCTIONS TO CANDIDATES:**

- ✓ This paper consists of two sections A and B, Attempt **ALL** questions.
- ✓ Incorrect symbols, formulae and spellings of especially technical terms will lead to loss of marks. And **good hand writing** is paramount please Boys.
- ✓ The **periodic table**, with relative atomic masses, is attached at the end of the paper.
- ✓ Illustrate your answers with equations where applicable, *Non-programmable scientific calculators may be used.*
- ✓ Assume where necessary, use the following values.
  - Molar gas constant, R = 8.314 JK<sup>-1</sup> mol<sup>-1</sup>.
  - Molar volume of gas at s.t.p = 22.4 litres.
  - Standard temperature, T = 273.15 K.
  - Standard pressure, P = 101325 Nm<sup>-2</sup>.
  - Molar volume of gas at r.t.p = 24.0 litres.

*This paper consists of 11 pages. You should check the question paper to ensure that all pages are printed as indicated and no questions are missing.*

1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	TOTAL

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