P525/3 Inst.

Sch.

### CHEMISTRY

Practical

Instructions

August 2023



## ENTEBBE JOINT EXAMINATION BUREAU

# Uganda Advanced Certificate of Education

CHEMISTRY PRACTICAL INSTRUCTIONS

Paper 525/3 Inst. Sch

August 2023

#### CONFIDENTIAL:

Great care should be taken that the information given below DOES NOT reach the candidates either directly or indirectly.

#### INSTRUCTIONS FOR PREPARING APPARATUS

NB: The head teacher **must** ensure that the teacher responsible for preparing the apparatus hands his/her trial results properly sealed in a separate envelope and **firmly** fastened (attached) to the candidates' scripts envelope(s).

In addition to the fittings and substances ordinarily contained in a chemical laboratory, each candidate will require:

- **1** burette  $(50 \text{ cm}^3)$
- **1** pipette (25.0cm³ or 20.0cm³)
- 1 volumetric flask (250 cm<sup>3</sup>)
- 2 conical flasks

Phenolphthalein indicator

- 8 test tubes
- 100 cm $^3$  of **FA1** (1MH $_2$ SO $_4$ )
- 100 cm<sup>3</sup> of **FA2** (0.25M NaOH)
- 100 cm<sup>3</sup> of **FA3** (0.041M NaOH)
- 3.0 g of  $\boldsymbol{W}$  (Anhydrous Sodium hydrogen carbonate)
- 3.0g of substance  $\mathbf{F}$  is made by mixing zinc carbonate  $(ZnCO_3)$  + Barium carbonate  $(BaCO_3)$  + Potassium iodide (KI) in the ratio of 3g: 2g: 1g respectively.

10cm<sup>3</sup> of substance **R.** (Acetone or propanone)

2 END