



**MATIGO MOCK EXAMINATIONS 2022**

**Uganda Advanced Certificate of Education**

**CHEMISTRY**

**Paper 3**

**CONFIDENTIAL**

Each candidate should be provided with the following;

- Stop clock
- Thermometer
- Two conical flasks
- Three beakers
- 20cm<sup>3</sup> or 25cm<sup>3</sup> pipette
- 50cm<sup>3</sup> burette
- 250cm<sup>3</sup> volumetric flask
- Six test tubes and one boiling tube
- Access to heat, weighing balance and distilled water
- 100cm<sup>3</sup> of FA1
- 10.0g of Y
- 2.0g of W
- 2.0g of T

**Note;**

FA<sub>1</sub> is 0.1M hydrochloric acid

Y is anhydrous sodium carbonate

W is a mixture of aluminium potassium sulphate + ammonium ferrous sulphate in the ratio of 1:2

T is Tannic acid