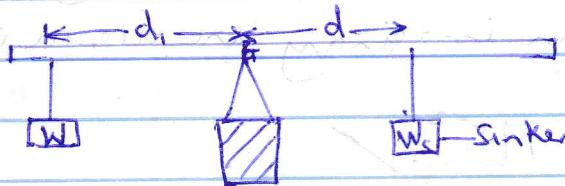


S6 MOCK 1 2017 - PHYSICS PAPER ONE

MARKING GUIDE

1. (a) (i) Relative density - ratio of the mass of any volume of substance to the mass of an equal volume of water.
 or $R.D = \frac{\text{density of substance}}{\text{density of water}}$
- (ii) A metre rule is balanced horizontally on a knife edge and the position, G , of balance noted.
- The knife edge is now fixed at G . A sinker is attached at a distance d from the pivot, and the distance kept constant.
 - A weight, W , of any suitable mass, is hung on the opposite side and adjusted until the metre rule balances horizontally. The distance of W is noted as d_1 from G .

Using moments, $W_s \times d = Wd_1$; $W_s = \text{weight of sinker in air}$
 ~~$W_s = Wd_1 / d$~~



- With the knife edge still at G , the sinker is immersed in water; the weight W is moved until the metre rule balances horizontally. The distance d_2 of W from the pivot is read and noted.



Using moments and letting, $W'_s = \text{weight of sinker in water}$
 $W \times d_2 = W'_s \times d$

$$W'_s = \frac{Wd_2}{d} \quad \text{(ii)}$$

- With the knife edge maintained at G , the sinker is then immersed in a liquid. The weight W is adjusted until the metre rule balances horizontally. The distance d_3 of W from

any depth d_3 - $F = \rho g A d_3$

Using moments, $W \times d_3 = W_s'' \times d$; $W_s'' = \text{wt/g similar to liquid}$.

mass of water at depth d $W_s'' = \frac{W d_3}{d}$ ~~in liquid~~ (by \propto)

surface tension at d \propto mass at d \propto water density ρ

upthrust in water $= W_s - W_s''$

$$\text{upthrust in water} = \frac{W d_1}{d} - \frac{W d_2}{d} = \frac{W (d_1 - d_2)}{d}$$

upthrust in liquid $= W - W_s''$ when $d_3 \rightarrow \infty$

$$\text{upthrust in water} = W d_1 - \frac{W d_2}{d} \approx \frac{W (d_1 - d_2)}{d}$$

Relative density \propto the calculated form

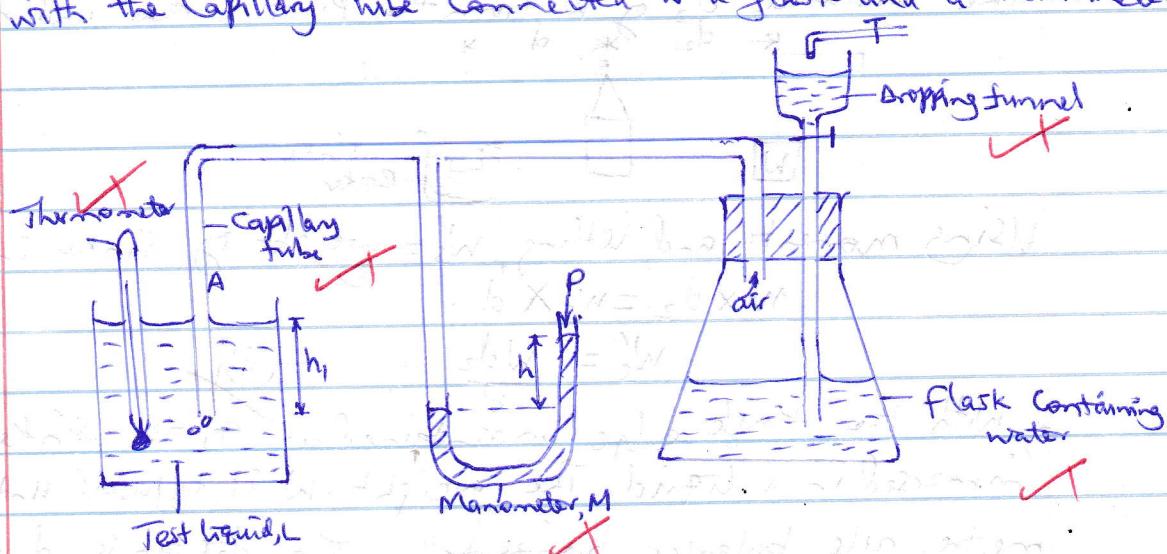
$$\text{relative density} = \frac{\text{upthrust in water}}{\text{upthrust in liquid}} = \frac{W (d_1 - d_2)}{W (d_1 - d_3)} = \frac{d_1 - d_2}{d_1 - d_3} \quad \checkmark$$

b) Surface tension - tangential force acting perpendicular to one side of an imaginary line drawn in the liquid surface, or Surface tension - force per unit length acting normally to one side of an imaginary line drawn in the liquid surface.

$$[S] = \frac{[F]}{[L]} = MT^{-2}$$

02

(ii) The experiment is set up as shown in the figure below with the capillary tube connected to a flask and a manometer M .



(3)

The temperature, θ , of L is noted on the thermometer and recorded.

Drops of water are slowly dropped from the funnel into the flask while observing the manometer. This makes a bubble to form slowly at the end of the tube A.

The maximum pressure difference, h, is noted on the manometer.

Now, let P = atmospheric pressure, f = density of liquid in manometer, σ = density of the test liquid and r = radius of the orifice of A.

When the radius of the orifice of A equals the radius of the bubble formed, then the pressure is a maximum.

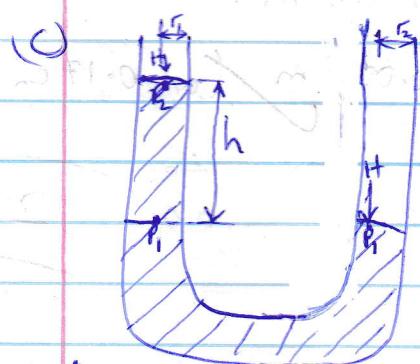
$$\text{Excess pressure inside the bubble, } \frac{2\gamma}{r} = (P + h_f g) - (P + h_i \sigma g) \quad (05)$$

$$\gamma = \frac{r g (h_f - h_i \sigma)}{2}, \text{ and } \gamma \text{ is calculated.}$$

The experiment is repeated for various temperatures, θ , of the test liquid and each time calculating the value of the surface tension, γ .

(iii) Variation of surface tension with temperature:

When the temperature of a liquid is increased, the average kinetic energy of its molecules increases. The intermolecular forces of attraction between the molecules decrease. Hence surface tension of a liquid decreases as the temperature increases and vanishes at the critical temperature.



$$r_2 = \frac{3}{2} \text{ mm} = 1.5 \times 10^{-3} \text{ m}$$

$$r_1 = 10 \text{ mm} = 1.0 \times 10^{-2} \text{ m}$$

$$\theta = 120^\circ$$

$$\sigma = 1.36 \times 10^4 \text{ kg m}^{-3}; \gamma = 0.4 \text{ N m}^{-1}$$

Let P_1 = pressure just below the surface of mercury in smaller tube.

P_2 = pressure just below the surface of mercury in larger tube

H = atmospheric pressure.

i) $F_g = 2\pi R \gamma$ between $r = R$ & $r = R \cos \theta$ Radius of curvature of the surface.

But $r = R \cos \theta$

Centrifugal force acting on a particle at radius r is zero

∴ 2nd system is stationary at radius $r = R \cos \theta$ ✓

∴ Resultant effect of all forces on particle must be zero

∴ $P_1 - H = 2\pi C \cos \theta$, taking components along \vec{r} . (Ans)

~~∴ $P_1 = 2\pi C \cos \theta + H$~~ ✓ (Ans)

∴ $P_1 = 2\pi C \cos \theta + H$ ✓ (Ans)

Also $P_1 = P_2 + h g g = H + 2\pi C \cos \theta + h g g$ ✓ (Ans)

∴ $h g g = 2\pi C \cos \theta (\frac{1}{R} - \frac{1}{r})$ ✓ (Ans)

~~∴ $h g g = 2\pi C \cos \theta (\frac{1}{R} - \frac{1}{r})$~~ ✓ (Ans)

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~~∴ $h g g = 2\pi C \cos \theta (\frac{1}{R} - \frac{1}{r})$~~ ✓ (Ans)

$= -1.699 \times 10^{-3} \text{ m}$.

Difference in mercury levels = 0.0017 m ✓ (Ans)

$m \cdot 0.1 \times 10^3 = 0.017$

$100 \cdot 10^3 = 0.017 \times 10^3 \times 10^{-3}$

✓ (Ans)

∴ Net additional tension in string attached to wall = 9 N .

∴ Net upward tension in string attached to wall = 9 N .

∴ Net upward tension = 4 N .

- 2(a) - Every body remains in its state of rest or of uniform motion in a straight line unless an external force acts otherwise.
- The rate of change of momentum of a body is directly proportional to the applied force and takes place in the direction of the force.
 - For every action there is an equal and opposite reaction.

(b) The rocket exerts a strong force on the burning gases, expelling them downward. The exhaust gases exert an equal and opposite force on the rocket, which propels the rocket to move forward.

(c) A bullet of mass $m = 0.02 \text{ kg}$ is fired from a gun of mass $M = 980 \text{ g}$.



Let V be the common velocity of the bullet and the block.

KE gained by block + bullet = Electrostatic PE.

$$\frac{1}{2}(m+M)V^2 = \frac{1}{2}Kx^2$$

$$\frac{1}{2}(0.02 + 0.98)V^2 = \frac{1}{2} \times 100 \times (4.8 \times 10^{-2})^2$$

$$V = \sqrt{0.2304} = 0.48 \text{ ms}^{-1}$$

Conserving momentum initially

$$m_u + Mu_2 = (m+M)V$$

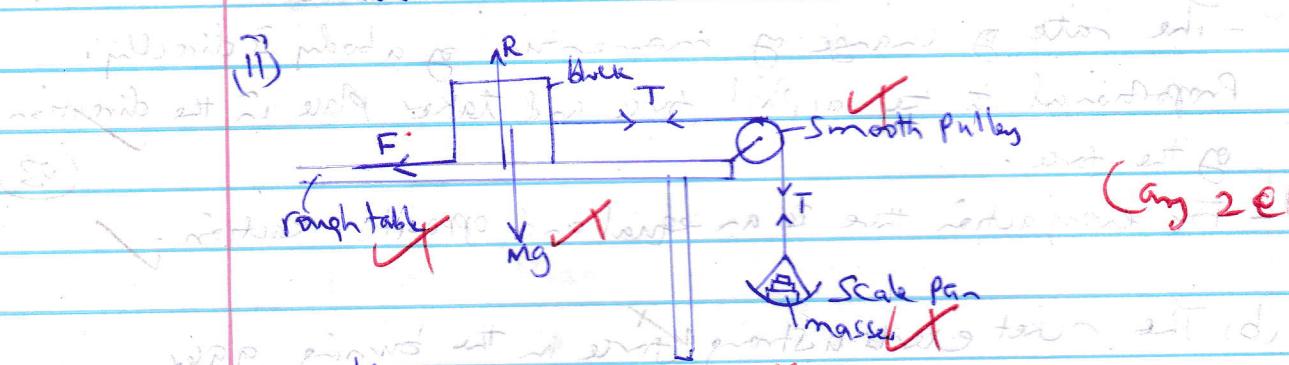
$$0.02u + 0.98 \times 0 = (0.02 + 0.98) \times 0.48$$

$$u = 24 \text{ ms}^{-1}$$

(d) i) Frictional force between two surfaces oppose their relative or attempted motion.

- Frictional force is independent of the area of contact provided

- Frictional force is directly proportional to the normal reaction force \checkmark (b3)



The sphere is weighed and its mass recorded as m .

- A light string is attached to the block resting on the table and passes over a pulley to support a scale pan at its free end.

- Small weights are carefully added into the scale pan until the block just begins to move. The total weight of the pan and its contents is recorded as W .

- Under these conditions, the limiting frictional force, $F = W$.

The coefficient of friction, μ is then calculated $\frac{W}{mg}$ (b3)

$$\mu = \frac{W}{mg} = \frac{Mg}{mg} = \frac{M}{m}$$

- The experiment may be repeated for more values of W and the average value of μ calculated. (e.g. 0.01)

NB- You may follow through up to if the graphical approach is used. $\mu = \text{slope of graph of } W \text{ against } m$.

ALT: (Follow through)

$$F_r = \mu N \quad \text{and} \quad F_r = mg - T$$

$$mg - T = \mu Mg + mg$$

$$2mg - \mu Mg = Mg$$

first weight will be measured and substituted (b20)

in next iteration μ will be calculated

What friction is one set of readings? μ not found

Qn 3(a) Elasticity - ability of a material to change in length when a force is applied and to regain its original shape and size when the deforming force is removed.

(i) Young's modulus - ratio of the tensile stress to the tensile strain. \rightarrow ~~to stretch a material~~ ~~initially~~ \rightarrow ~~specify~~ (03)

(ii) Plastic deformation - region in which a material stretched to it cannot regain its original shape and size when the stretching force is removed. \checkmark

(b) Consider a material stretched through a length, Δl when a force F is applied. $l_0 = \text{original length of the material}$.



$$\text{Work done by force, } EP_E = Fx = F\Delta l \quad \checkmark$$

$$\text{Total work done, } W = \int_0^e Fdx = \int_0^e Kx dx$$

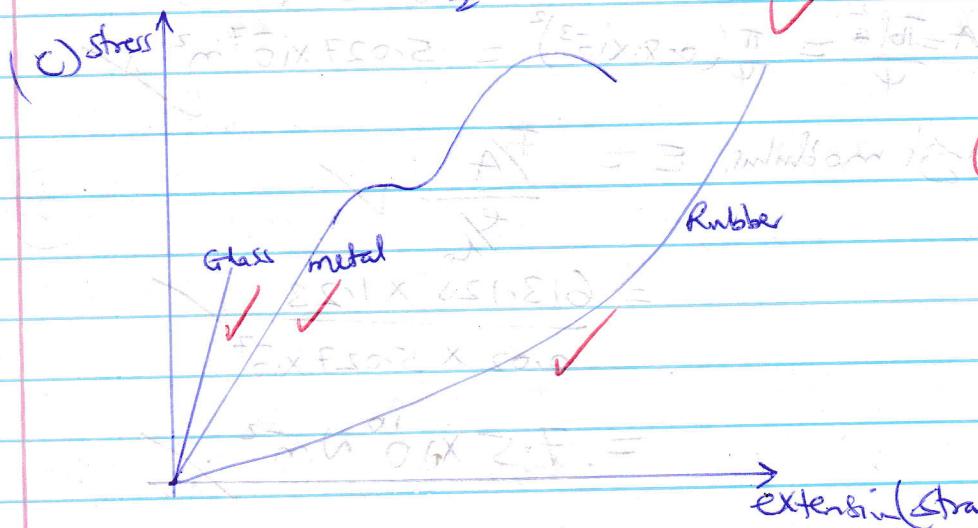
$$W_{\text{total}} = \frac{K e^2}{2} \quad \checkmark$$

$$EP_E \text{ per unit volume} = \frac{K e^2}{2} / A l_0 \quad \checkmark$$

$$= \frac{1}{2} \left(\frac{K e^2}{A l_0} \right) \quad \checkmark$$

$$= \frac{1}{2} \times \frac{F}{A} \times \frac{e}{l_0} \quad \checkmark$$

$$= \frac{1}{2} \times \text{Stress} \times \text{Strain} \quad \checkmark$$

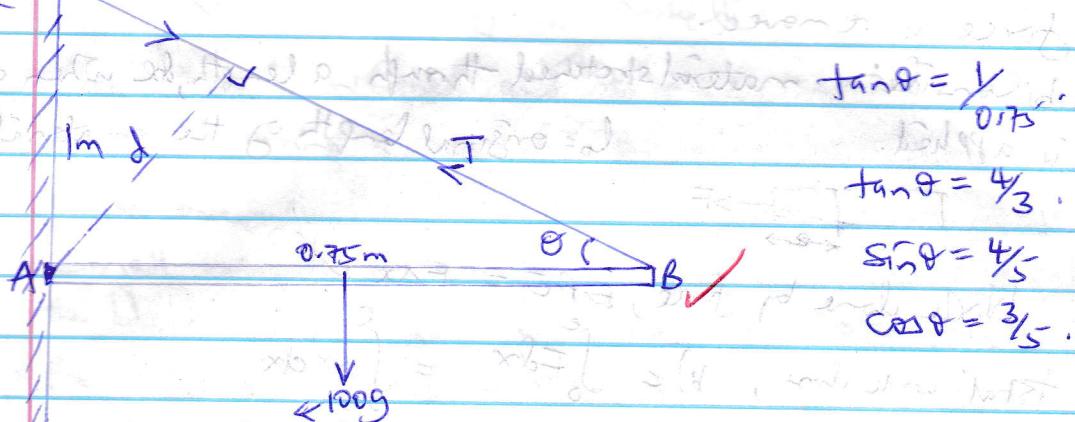


(03) (Provided the axes are labelled)

Metal - metal undergoes both elastic and plastic deformation. Before reaching the proportional limit, the stress is directly proportional to the strain.

Rubber - Rubber has a greater range of elasticity and does not undergo plastic deformation. Unstretched rubber has coiled molecules which when stretched, they unwind and become straight.

(Q8)



i) Taking moment about A;

$$100 \times 9.81 \times 0.75 = T \times 0.75 \sin \theta \quad \checkmark \quad \text{(04)}$$

$$T = \frac{100 \times 9.81 \times 0.375}{0.75 \times \frac{4}{5}}$$

$$= 613.125 \text{ N.} \quad \checkmark$$

ii) $AC = \sqrt{1^2 + 0.75^2} = 1.25 \text{ m.}$

extension, $\epsilon = 1.25 - 1.23 = 0.02 \text{ m}$ \checkmark

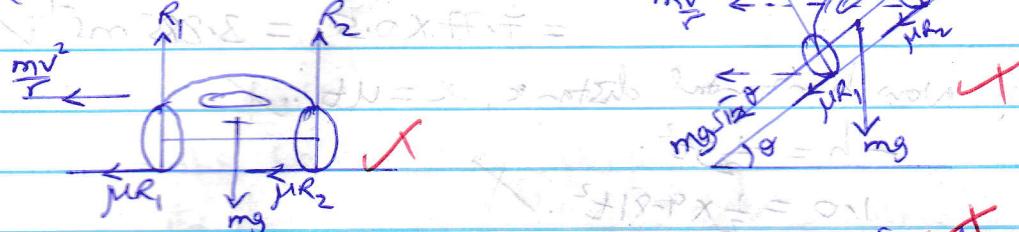
$$A = \frac{\pi d^2}{4} = \frac{\pi (0.8 \times 10^{-3})^2}{4} = 5.027 \times 10^{-7} \text{ m}^2 \quad \checkmark \quad \text{(04)}$$

Young's modulus, $E = \frac{F/A}{\epsilon}$ \checkmark

$$= \frac{613.125 \times 1.23}{0.02 \times 5.027 \times 10^{-7}} \quad \checkmark$$

$$= 7.5 \times 10^{10} \text{ N m}^{-2} \quad \checkmark$$

4. (a)



On a flat track, the centripetal force is provided for by only the frictional force. Thus $\frac{mv^2}{r} = \mu(R_1 + R_2)$ X (i).

On a banked track, the centripetal force is provided for by both the component of the normal reactions acting horizontally and the horizontal component of the frictional force X.

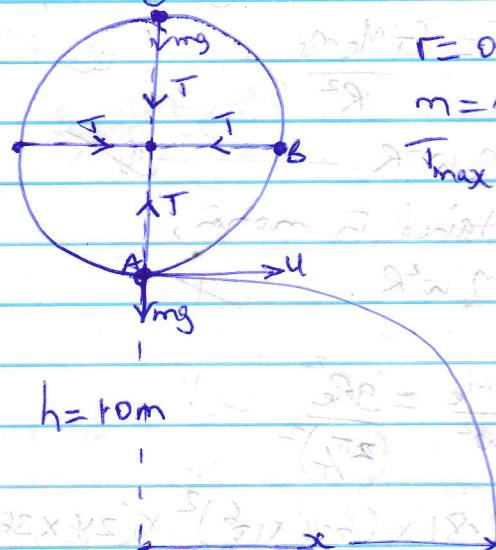
$$\text{Thus } \frac{mv^2}{r} = (R_1 + R_2) \sin \theta + \mu(R_1 + R_2) \cos \theta$$

$$\frac{mv^2}{r} = (R_1 + R_2) [\sin \theta + \mu \cos \theta]$$

(04)

Hence a car travels faster on a banked track than on a flat track of the same radius. X

(b)



$$m = 0.5 \text{ kg}$$

$$T_{\max} = 20 \text{ N}$$

i) At A (directly below the centre).

$$T - mg = \frac{mv^2}{r}$$

$$T = mg + \frac{mv^2}{r} \quad \text{(i)}$$

$$\text{At B, } T = \frac{mv^2}{r} \quad \text{(ii)}$$

$$\text{At C; } T + mg = \frac{mv^2}{r} \quad \text{(iii)}$$

$$T = \frac{mv^2}{r} - mg \quad \text{(iv)}$$

Now T is greater at A and the string will break there. X

$$\text{i) } 20 = 0.5 [9.81 + w^2 r] \quad \checkmark$$

$$40 = 9.81 + 0.5w^2$$

(03)

(iii) Horizontal velocity, $v = \omega r$.

$$= 7.77 \times 0.5 = 3.885 \text{ m}^{-1} \checkmark$$

Now, horizontal distance, $x = vt \checkmark$

$$\therefore h = -\frac{1}{2}gt^2.$$

$$1.0 = \frac{1}{2} \times 9.81 t^2 \checkmark$$

$$\therefore \text{Time taken, } t = \sqrt{\frac{2h}{g}} = 0.452 \text{ s. } \checkmark \quad (03)$$

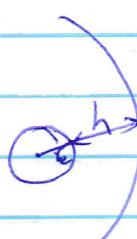
$$\therefore x = 3.885 \times 0.452 \checkmark$$

Horizontal distance, $= 1.756 \text{ m from vertical through the centre of rotation}$

Distance at θ , the angular position of the circle.

(d) (i) Parking orbit — Path in space in which a satellite appears to be in the same position relative to a point on the earth's surface as the earth rotates about its axis. $\checkmark \quad (01)$

(iv)



Let M_e = mass of the earth,

m_s = mass of the satellite

By Newton's law of gravitation,

$$m_s F_c = \frac{G M_e m_s}{R^2} \checkmark$$

Centrifugal force, $F_c = m_s \omega^2 R \checkmark$

For satellite to be maintained in motion,

$$\frac{G M_e m_s}{R^2} = m_s \omega^2 R. \checkmark$$

$$R^3 = \frac{G M_e}{\omega^2} = \frac{g R_e^2}{(2\pi/\text{T})^2} \checkmark$$

$$R^3 = \frac{9.81 \times (6.4 \times 10^6)^2 \times (24 \times 3600)^2}{4\pi^2} \checkmark \quad (04)$$

$$R = 42.35 \times 10^6 \text{ m} \checkmark$$

$$h = 42.35 \times 10^6 - 6.4 \times 10^6$$

$$= 35.95 \times 10^6 \text{ m} \checkmark$$

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5. (a) i) - Molecules of a gas are randomly moving about, continuously colliding with each other and the walls of the container at all times.

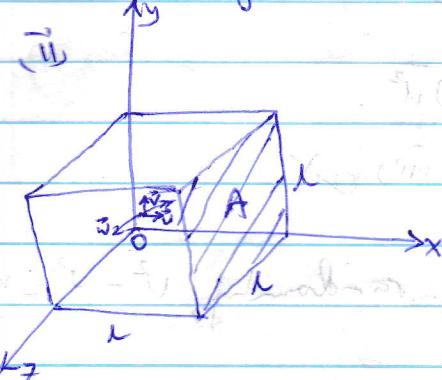
- The collisions are perfectly elastic ✓

- The duration of a collision is negligible compared to the time spent between collisions ✓

- Inter-molecular forces of attraction are negligible ✓

- Volume occupied by molecules themselves is negligible compared to the volume occupied by the gas, ✓

(iv)



Consider N molecules of a gas contained

In a cube of side, l , each molecule having a mass, m . The molecules are considered to be in a state of continuous random motion.

Consider one molecule having a velocity \vec{c} at any instant where \vec{c} has components u , v and w in the ox , oy and oz -directions respectively. Then

$$c^2 = |\vec{c}|^2 = |u\hat{i} + v\hat{j} + w\hat{k}|^2 = u^2 + v^2 + w^2$$

Let $|u| = u$; $|v| = v$ and $|w| = w$

$$\Rightarrow c^2 = u^2 + v^2 + w^2$$

Consider the speed u of a molecule perpendicular to the face A of the cube.

Change in momentum of the molecule $= mu - (-mu) = 2mu$ ✓

Let t be the time taken for the molecule to move from one face to another and back,

Then $t = \frac{2l}{u}$. ✓

The rate of change of momentum, $F = \frac{2mu}{2l} = \frac{mu}{l}$

$$F = \frac{mu^2}{l}$$
 ✓ (iii)

Given $P = \frac{mu^2}{l} = \frac{mu^2}{l^3} \cdot l^3$ (iv) ✓

For N molecules of He gas, the total pressure, P , exerted will be

$$P = \frac{m u_1^2}{l^3} + \frac{m u_2^2}{l^3} + \dots + \frac{m u_N^2}{l^3} \quad \text{✓}$$

$$= \frac{m (u_1^2 + u_2^2 + \dots + u_N^2)}{l^3}$$

Let \bar{u}^2 be the mean of the squares of the speeds in the OX -direction.

$$\bar{u}^2 = \frac{u_1^2 + u_2^2 + \dots + u_N^2}{N} \quad \text{✓}$$

Substituting $P \Rightarrow u_1^2 + u_2^2 + u_3^2 + \dots + u_N^2 = N \bar{u}^2$.

∴ $P = \frac{m N \bar{u}^2}{l^3}$ (iv) ✓

Since the molecules are moving randomly, $\bar{u}^2 = \bar{v}^2 = \bar{w}^2$.

$$\Rightarrow \bar{u}^2 = \bar{v}^2 = \bar{w}^2$$

Now $\bar{c}^2 = \bar{u}^2 + \bar{v}^2 + \bar{w}^2$

$$\therefore \bar{c}^2 = \bar{u}^2 + \bar{v}^2 + \bar{w}^2 = 3\bar{u}^2$$

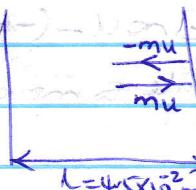
$$\bar{u}^2 = \frac{1}{3} \bar{c}^2 \quad \text{✓}$$

$$\therefore P = \frac{m N}{l^3} \cdot \frac{1}{3} \bar{c}^2$$

But density, $f = \frac{m N}{l^3}$, $\therefore f N = \frac{f}{3} \bar{c}^2$

(b)

$\therefore P = \frac{1}{3} f \bar{c}^2$ ✓

(b) 

$m = 2.32 \times 10^{-26} \text{ kg}$; mean speed, $\bar{u} = 500 \text{ m/s}$

$N = 2 \times 10^{22} \text{ molecules}$

Change in momentum of each molecule = $m \bar{u} - (-m \bar{u}) = 2m \bar{u}$.

Time taken to move from one end to another and back

$$\therefore \text{Time} = \frac{l}{\bar{u}} \quad t = \frac{2l}{\bar{u}} \quad \text{✓}$$

Pressure exerted by one molecule, $P = \frac{mv^2}{l/l^2} = \frac{mv^2}{l^3}$ ✓

Now total pressure exerted by N molecules is not

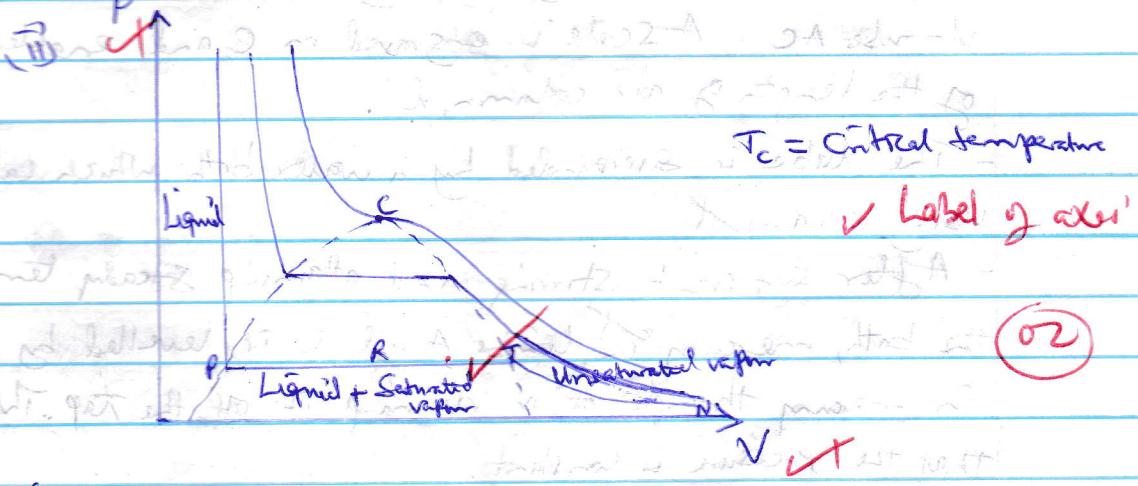
$$P_f = \frac{Nm v^2}{l^3}$$

$$= \frac{2 \times 10^{22} \times 2.32 \times 10^{-26} \times 500^2}{(4.5 \times 10^{-2})^3}$$

(04)

$$= 1.273 \times 10^6 \text{ Nm}^{-2}$$

- (Q) i) - The rate of evaporation equals the rate of condensation
 - They do not obey gas laws.
 - The vapour pressure is independent of the volume.



(02)

- iii) In the region TN, there is unsaturated vapor which approximately obeys Boyle's law.

At T, droplets begin to form. ✓

In the region TP, the vapor is saturated and the pressure remains constant. The volume of the gas reduces as more of it becomes liquid.

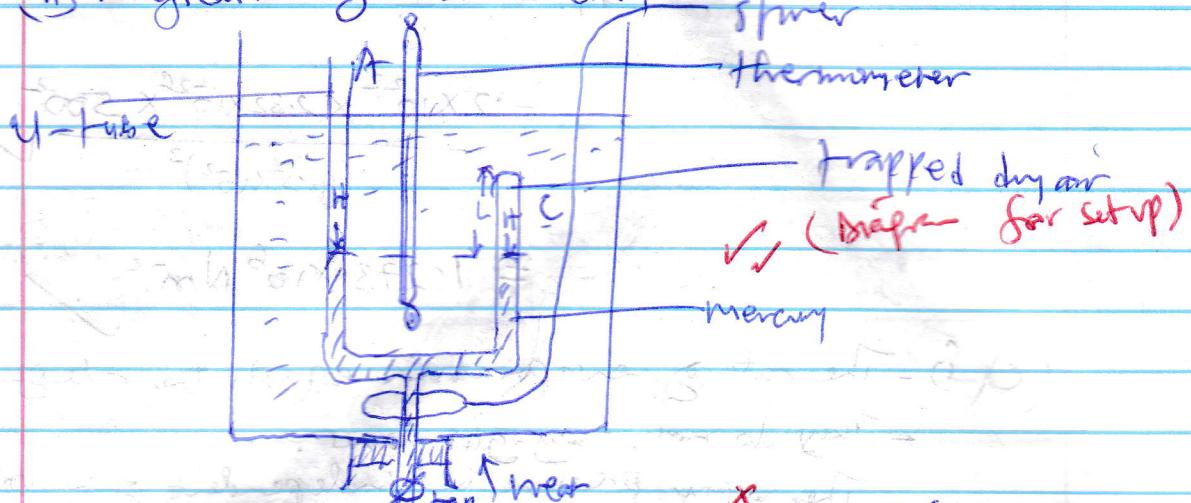
(02)

At P, all the gas becomes liquid and the pressure increases rapidly since liquids are almost incompressible.

✓ 20

6. (a) i) Charles' law - the volume of a fixed mass of gas at constant pressure is directly proportional to the absolute temperature.

ii) Verification of Charles' law:



- Dry air is trapped by mercury in closed limb C of the U-tube AC. A scale is engraved on C and enables measurement of the length of air column, l .

- The tube is surrounded by a water bath which can be heated using steam.

- After sufficient stirring and attaining steady temperature of the bath, mercury in limbs A and C is levelled by either pouring in mercury through A or running it off the top. This ensures that the pressure is constant.

- For each value of temperature, the corresponding value of the length of the air column, l , is obtained in $(V \propto l)$

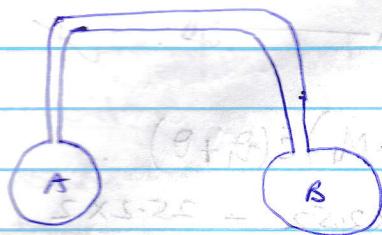
- A graph of l against absolute temperature T is plotted.

Graph of l vs T is a straight line with positive slope.

A straight line through the origin is obtained, hence verifying Charles' law.

br (i) Dalton's law of Partial Pressure - In a mixture of gases which do not interact chemically the total pressure equals the sum of the partial pressures of the constituent gases. ✓ 07

(ii)



Let n_A and n_B be the number of moles of the gas in the bulbs respectively.

$$\text{Initially; } n_A = \frac{P_1 V_A}{R T_1} \quad ; \quad n_B = \frac{P_1 V_B}{R T_1}$$

Total number of moles,

$$n_T = \frac{P_1 V_A}{R T_1} + \frac{P_1 V_B}{R T_1} \quad ; \quad V_A = V_B = V.$$

~~$$n_T = \frac{2 P_1 V}{R T_1}$$~~

After immersing one bulb;

$$n_A = \frac{P_2 V}{R T_1} ; \quad n_B = \frac{P_2 V}{R T_2} \quad ; \quad n = n_A + n_B$$

$$2 P_1 V = P_2 V + \frac{P_2 V}{R T_2}$$

$$\frac{V}{R} \left[\frac{2 P_1}{T_1} \right] = \frac{V}{R} \left[\frac{P_2}{T_1} + \frac{P_2}{T_2} \right]$$

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$$\frac{P_2}{T_2} = \frac{2 P_1 - P_2}{T_1}$$

$$\frac{P_2}{T_2} = \frac{2 P_1 - P_2}{T_1}$$

$$T_2 = \frac{P_2 T_1}{2 P_1 - P_2}$$

(c) Specific heat capacity - Amount of heat required to raise

(d) $C_w = 4150 \text{ J kg}^{-1} \text{ K}^{-1}$, $\theta_1 = 15^\circ\text{C}$, initial state (i)

let the out flow temperature be θ . \rightarrow adiabatic condition

(i) $V_1 I_1 = M_1 c(\theta_1 - \theta) + h_{\text{outflow}}$ at flow rate

$$V_1 I_1 = M_1 c(\theta_1 - \theta) + h \quad \text{Q. } \checkmark \quad (i)$$

$$V_2 I_2 = M_2 c(\theta_1 - \theta) + h \quad \text{Q. } \checkmark \quad (ii)$$

(iii) \rightarrow (i) $-$ (ii)

$$V_2 I_2 - V_1 I_1 = (M_2 - M_1) c(\theta_1 + \theta)$$

$$-15 + \theta = 30 \times 2.52 - 25.2 \times 2$$

$$\frac{(115.9 \times 10^3 - 75 \times 10^3)}{60} \times 4150 \quad \text{Q3}$$

$$-15 + \theta = 8.91$$

$$23.91^\circ\text{C}$$

$$\theta = 6.09^\circ\text{C}$$

(iv) $25.2 \times 2 - \frac{75 \times 10^3}{60} \times 4150(15 - 6.09) = h$ \checkmark

$$h = 50.4 - 46.22$$

$$h = 4.18 \text{ JS}^{-1} (\text{or W})$$

(e) (i) Measure the ^{resistance} R_0 at the unknown temperature, θ \checkmark

- Realise the resistance R_{ref} at the triple point of water

- The unknown temperature is given by $\theta = \frac{R_0 \times 273.16 \text{ K}}{R_{\text{ref}}} \quad \text{Q3}$ \checkmark

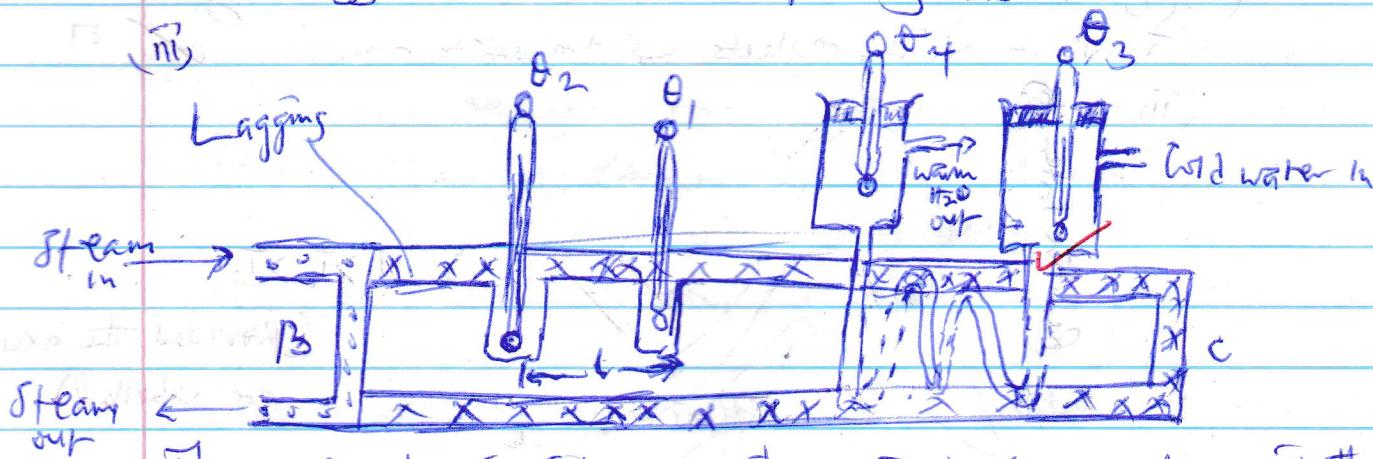
(ii) Thermocouple can measure temp at a point where as the resistance thermometer cannot. \times Q1

- Thermocouple has a small heat capacity, hence can measure rapidly changing temperatures where as the resistance thermometer cannot. \checkmark

7.(a)(i) Thermal Conductivity - rate of heat flow across opposite faces of unit of a parallel sided slab when the temperature difference across the faces is one Kelvin. (Q1)

Or; K - heat flow rate normal to one square metre area of a sample whose faces are maintained at a temperature gradient of one Kelvin per metre.

(ii) Conduction in metals is mainly due to free moving electrons. When one end is heated, the free electrons gain thermal energy and move with increased ~~velocity~~ ^(kinetic energy). They dissipate the energy upon collision with the ions in the lattice. Also energy can be transmitted through molecular vibrations where heat energy is transmitted to neighbouring molecule.



The apparatus is set up as shown in the figure above, with the specimen metal rod AB long compared to its diameter.

A heater is placed at one end of the bar while the bar (rod) is cooled by circulating water at the end B . ✓

Two holes are drilled and thermometers placed there at a measured distance, l . at some point it is written in bracket 'not at the required distance' (Q2)

The holes are filled with mercury to ensure good thermal contact.

The whole apparatus is left running until the temperatures here become steady. The circulating water is collected over a measured time interval is recorded.

Let m = mass of water collected per second and θ_1 and θ_2 be the respective thermometer readings at C and D .

The diameter, d , of the metal bar is obtained at atleast three positions of the bar, and the area, A , calculated ($A = \frac{\pi d^2}{4}$).

The rate of heat flow is also given by $\frac{dQ}{dt} = mc_w(\theta_2 - \theta_3)$.

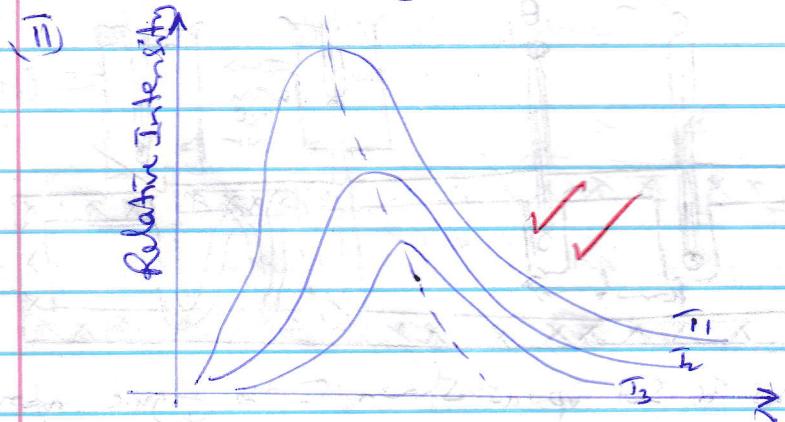
$$\text{and from above } \frac{dQ}{dt} = KA(\theta_2 - \theta_1) \quad \checkmark$$

$$\therefore \frac{KA(\theta_2 - \theta_1)}{l} = mc_w(\theta_2 - \theta_3) \quad \checkmark$$

$$K = \frac{mc_w(\theta_2 - \theta_3) \cdot l}{A(\theta_2 - \theta_1)} \quad \checkmark, \text{ from which } K \text{ can be calculated.}$$

θ_3, θ_2 are the entrance and exit temp. water.

(b) (i) Black body - one that absorbs all radiations of all wavelengths incident on it, reflects and transmits none. \checkmark



$$T_1 > T_2 > T_3$$

(provided the axes are labelled)

(ii) At first, the ball is invisible. \checkmark

- It becomes dull red, then bright red and finally less red tending to white. \checkmark

- This is because as the temp. rises, the intensity of the shorter wavelengths increases more rapidly. So the peak intensity shifts from the red of the spectrum into the visible spectrum with green and blue colours coming into play. \checkmark

(c) At equilibrium, Power radiated by filament = Power generated.

$$A \cdot 1870 = 85 \cdot A \cdot 4 \quad \text{or } 1870 = 85 \cdot 4 \quad \text{or } 1870 = 340$$

$$\text{But } A = 2\pi r l = \pi d l$$

$$= \pi \times 1.5 \times 10^{-2} \times 30 \times 10^{-2}$$

$$= 0.01414 \text{ m}^2$$

$$\therefore 1800 = 85 \times 5.7 \times 10^{-8} \times 0.01414 T^4$$

$$T = 2.628 \times 10^{12}$$

$$\therefore T = 1273.2 \text{ K}$$

(04)

20

8.(a).i) X-rays

- Carry no charge
- Travel at the speed of light
- Have a higher penetration power
- Have a relatively low ionization power

Cathode rays

- Carry a negative charge
- Have a relatively low speed
- Have a relatively low penetration power
- Have a relatively high ionization power.

- Deflected by electric & magnetic fields

- Deflected by electric & magnetic fields

(02)

ii) The anode is a high melting point material so that it does not melt easily when struck by fast moving electrons. At one end of it are cooling fins that help to get rid of the heat gained by the anode (or there is circulating water).

(b).i) $2d \sin \theta = n\lambda$ Where d = interplanar spacing

θ = glancing angle

n = order of diffraction

λ = wavelength of the beam

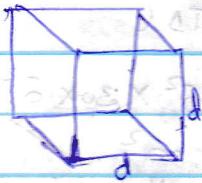
(07)

ii) The wavelength of the incident radiation must be of the same order as the interplanar spacing.

$$\text{iii) } \lambda = 1.10 \times 10^{-10} \text{ m}, n=1, \theta = 19^\circ$$

(01)

$$\text{R.M.M.} = 75.5$$



let d = interplanar spacing

$$2d \sin \theta = n\lambda$$

$$d = 1 \times 1.10 \times 10^{-10}$$

$$= 2 \sin 19^\circ$$

$$= 1.69 \times 10^{-10} \text{ m} \checkmark$$

\Rightarrow Volume associated with one atom $= d \times d \times d = d^3$.

Volume associated with KCl crystal, $V = 2d^3 \checkmark$

Now mass of one molecule, $m = \frac{M}{N_A}$:

$$\text{Density, } \rho = \frac{m}{V} = \frac{M}{N_A / 2d^3} \checkmark$$

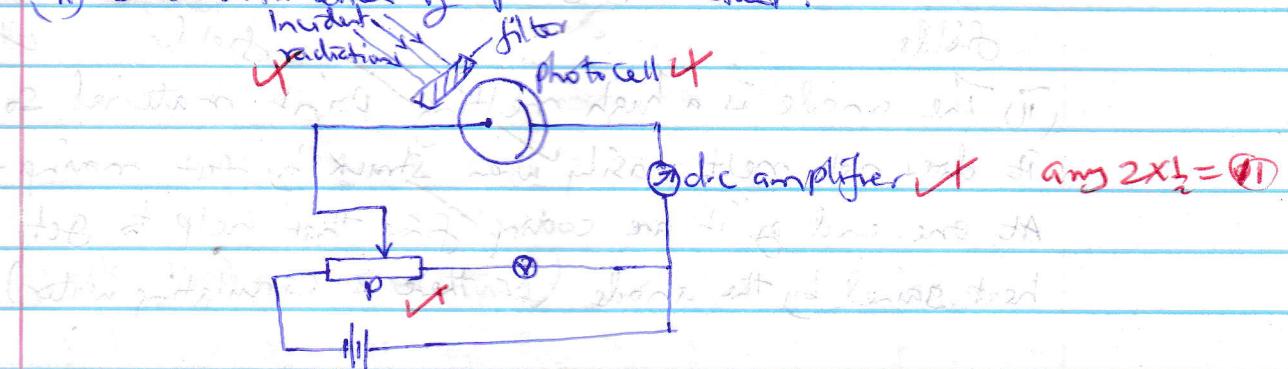
⑩

$$\rho = \frac{75.5 \times 10^{-3}}{6.02 \times 10^{23} \times 2 \times (1.69 \times 10^{-10})^3} \checkmark$$

$$= 1.299 \times 10^4 \text{ kg m}^{-3} \checkmark$$

(Ch. i) Work function - minimum energy required to eject an electron from a metal surface. ✓ ⑪

(ii) Determination of Planck's Constant.



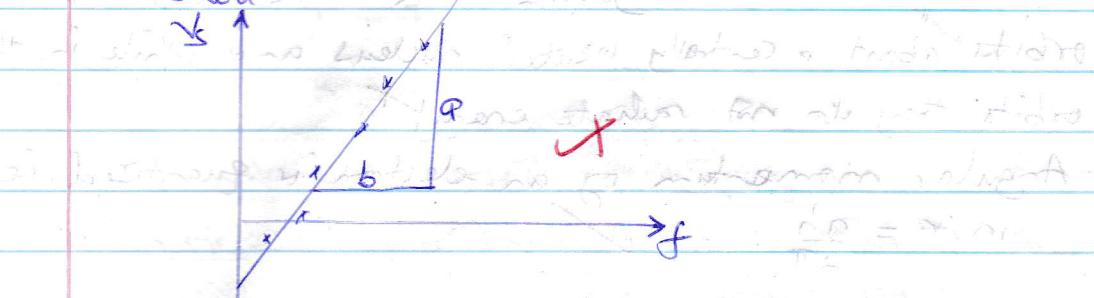
- An incident beam of electromagnetic radiation is passed through colour filter onto a photocell \checkmark

- The potential divider P is used to vary the pd V between the anode and the cathode and the d.c. amplifier is used for measuring small currents. ✓ ⑫

The pd. V is increased negatively until the current becomes zero, \checkmark and the stopping potential, V_s , is read and recorded from a Voltmeter. ✓

- A graph of V against f is now plotted and its slope, s ,

calculated from steps 1 and 2 -



(Q4)

- Planck's constant, h , is then calculated from part (g)

using $s = \text{constant} + \frac{h}{e \times \text{slope}}$, where e is electronic charge.

$$\text{iii) } f_i = 6.0 \times 10^{14} \text{ Hz}, V_i = 0 \text{ v} \quad \text{using method 2}$$

~~$$f_f = 1.0 \times 10^{15} \text{ Hz}, V_f = 2.2 \text{ V}$$~~

From Einstein's equation

$$eV = hf - W_0 \quad \checkmark$$

$$W_0 = 6.64 \times 10^{-34} \times 6.0 \times 10^{14} - 0.6 \times 1.6 \times 10^{-19} \quad \checkmark$$

$$= 3.024 \times 10^{-19} \text{ J} \quad \checkmark$$

$$\frac{hc}{\lambda_0} = 3.024 \times 10^{-19} \text{ J} \quad \text{Method 2}$$

$$\lambda_0 = \frac{6.64 \times 10^{-34} \times 3.0 \times 10^8}{3.024 \times 10^{-19}} \quad \checkmark$$

$$= 6.59 \times 10^{-7} \text{ m} \quad \checkmark$$

(Q4)

No graphing of graphing is mentioned in part (b) so

no graphing of graphing is mentioned in part (b) so

there is no method or graphing of graphing at part (b)

part (b) uses direct calculation from formula - method

Q. (a) Bohr's postulates;

- Electrons in the hydrogen atom move in circular orbits ~~✓~~ about a centrally located nucleus and while in these orbits they do not radiate energy ~~✓~~ (02)

- Angular momentum of an electron is quantized. i.e.

$$mVr = \frac{nh}{2\pi}$$

(b) Ionisation energy - energy required to remove an electron from an atom in its ground state to infinity so that it is completely lost. (01)

Excitation energy - energy required to remove an electron from an atom in its ground state to a higher energy level. (01)

(c) (i) Ionization energy = 0 - 10.4 eV.

$$= 10.4 \text{ eV.}$$

$$= 10.4 \times 1.6 \times 10^{-19} \text{ J}$$

(02)

$$= 1.664 \times 10^{-18} \text{ J}$$

(ii) For 4.0 eV.

$$\Delta E = -10.4 + 4.0$$

$$= -6.4 \text{ eV}$$

The atom remains unexcited. (01)

For 11.0 eV.

$$\Delta E = -10.4 + 11.0$$

(04)

$$= +0.6 \text{ eV.}$$

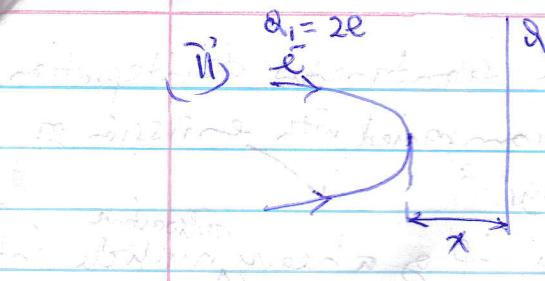
The atom will become ionised. (01)

(d) (i) Successes:

- It accounts for the observations of alpha-particle scattering by a metal foil (01)

- Using the model, the radius of the atom can be estimated (01)

Failure - It fails to account for existence of stable atoms (03)



$$Q_2 = Ze$$

$$Q_1 = 2e$$

At distance x , potential energy of interaction between two charges is

At distance x , kinetic energy of α -particle = Electrostatic Potential energy.

$$\frac{1}{2}mv^2 = -\frac{Q_1Q_2}{4\pi\epsilon_0 x}$$

$$\frac{1}{2}mv^2 = \frac{2e \times Ze}{4\pi\epsilon_0 x}$$

$$\text{and } x = 4ze^2 / 4\pi\epsilon_0 mv^2$$

03

$$\therefore x = \frac{ze^2}{\pi\epsilon_0 mv^2}$$

(e) Emission spectrum of hydrogen

- Continuous spectrum - It is the one in which all the wavelengths of the radiation emitted are spread over a wide range. It occurs in稀薄 and dense gases.

- Line spectrum - This consists of a number of well defined lines each with a particular wavelength. It is common with excited gases which emit radiation of only certain wavelengths.

- Band spectrum - Is the one in which the band consists of a series of lines very close together at one end and far apart at the other end.

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10. (a) i) Radioactivity - Is the random spontaneous disintegration of an unstable radioactive nuclei accompanied with emission of any of α , β - particles or gamma rays ✓ (02)
- While nuclear fission is the splitting of a heavy ^{radioactive} nuclide into lighter nuclei. ✓
- ii) B.E - Is the work done to take all the nucleon of an atom apart so that they are completely separated. ✓ (01)
(or Is the energy required to bring the constituent nucleon together to form a nucleus.)
- (b) i) Half life - Is the time taken for ^{the mass of} a radioactive nuclide to reduce to half its original value. ✓ (01)
- ii) Let N be the number of atoms present at a time t .

$$-\frac{dN}{dt} \propto N \quad \text{(decay law)}$$

$$-\frac{dN}{dt} = \lambda N$$

$$\int \frac{dN}{N} = -\lambda dt$$

$$\ln N = -\lambda t + C$$

At $t=0$, $N=N_0$ (initial condition)

$$\ln N_0 = C$$

$$\ln N = -\lambda t + \ln N_0$$

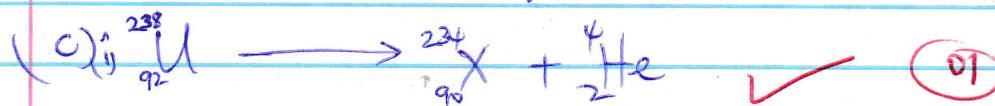
Let $t=T$ (half life, $N=\frac{N_0}{2}$) (03)

$$\ln \left(\frac{N_0}{2}\right) = -\lambda T_{1/2} + \ln N_0$$

$$\ln \left(\frac{N_0}{2}\right) = -\lambda T_{1/2}$$

$$T_{1/2} = \ln 2$$

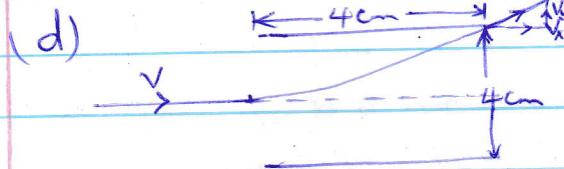
$$T_{1/2} = \frac{\ln 2}{\lambda} \quad \text{or} \quad T_{1/2} = \frac{0.693}{\lambda}$$



$$\begin{aligned}
 (\text{c}) \text{ii) Mass defect} &= 238.12492 - (234.11650 + 4.00387) \checkmark \\
 &= 0.00455 \text{ u} \\
 &= 0.00455 \times 1.66 \times 10^{-27} \text{ kg} \\
 &= 7.553 \times 10^{-30} \text{ kg} \checkmark
 \end{aligned}$$

$$\begin{aligned}
 \text{Energy released} &= (\Delta m)c^2 \\
 &= 7.553 \times 10^{-30} \times (3.0 \times 10^8)^2 \checkmark \\
 &= 6.798 \times 10^{-13} \text{ J.} \\
 &= \frac{6.798 \times 10^{-13}}{1.6 \times 10^{-19}} \times \frac{1}{10^6} \checkmark \\
 &= 4.25 \text{ MeV.} \checkmark
 \end{aligned}$$

(05)



$$\begin{aligned}
 (\text{i}) \frac{1}{2}mv^2 &= eV \\
 v^2 &= \frac{2eV}{m_e} \checkmark
 \end{aligned}$$

$$v = \sqrt{\frac{2 \times 1.6 \times 10^{-19} \times 1800}{9.11 \times 10^{-31}}} \checkmark$$

(03)

$$= 2.51 \times 10^7 \text{ ms}^{-1} \checkmark$$

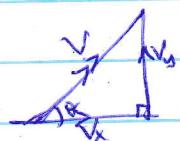
(ii) The electron beam describes a parabolic path. (01)

$$(\text{iii}) v_x = 2.51 \times 10^7 \text{ ms}^{-1}$$

$$v_y = a_y t = \left(\frac{F_e}{m_e}\right) \frac{L}{v} \cdot v \checkmark$$

$$v_y = \frac{F_e}{m_e} \cdot \frac{L}{v} \cdot v$$

$$= \frac{90}{4 \times 10^{-2}} \times \frac{1.6 \times 10^{-19}}{9.11 \times 10^{-31}} \times \frac{4 \times 10^{-2}}{2.51 \times 10^7} = 6.30 \times 10^5 \text{ ms}^{-1} \checkmark$$



$$\tan \theta = \frac{v_y}{v_x} \cdot \cancel{v} \checkmark$$

$$\tan \theta = \frac{6.30 \times 10^5}{2.51 \times 10^7} \checkmark$$

(03)