

# Statistical Thermodynamics in Chemistry and Biology

## Introduction

Per-Olof Åstrand

D3-119 Realfagsbygget, Department of Chemistry,  
Norwegian University of Science and Technology,  
*per-olof.aastrand@ntnu.no*

January 5, 2018

# Course in statistical thermodynamics and thermodynamics

- ▶ **Lecturers:** Per-Olof Åstrand (coordinator): per-olof.aastrand@ntnu.no  
Raffaella Cabriolu: raffaella.cabriolu@ntnu.no
- ▶ **Lectures** Mon, 15-17, R3; Fri, 14-16, R3
- ▶ **Exercises:** Raffaella Cabriolu (coordinator), Christopher Fjeldstad  
Mon, 17-19, R10; Thu, 16-18, R3  
The first exercise session: Mon 15.01
- ▶ Lectures will this year be in English. Teaching assistants speak Norwegian and English.
- ▶ Last date for teaching activities: Mon 23.04

# Literature

- ▶ **Book:** Ken A. Dill and Sarina Bromberg, *Molecular Driving Forces: Statistical Thermodynamics in Biology, Chemistry, Physics, and Nanoscience*, 2nd ed., Garland Science, 2010.
- ▶ Some exercises are taken from the book. Solutions will not be handed out. A list of answers is provided.
- ▶ A second set of exercises (including previous exams) are available with solutions.
- ▶ A basic goal is to learn how to **solve** exercises.

# Exam

- ▶ Exam date: 16.05, 15.00-19.00 in KJL1
- ▶ Written exam (code **A**: it is allowed to bring books and notes)
- ▶ Learn to use the book and own notes!

# Learning outcome

- ▶ *The course introduces the basis of statistical thermodynamics with examples from chemistry and biology*
- ▶ Introductory course
- ▶ General course - knowledge applicable in “all” fields
- ▶ Fundamental course - generic phenomena
- ▶ Theoretical course
- ▶ Pilot project: Python-based exercises (obligatory from 2017).  
6 exercises in total

# Learning outcome

From the study handbook

After the course, the student is expected to be able to:

- ▶ Explain the basic concepts and principles in statistical thermodynamics.
- ▶ Use lattice models to study basic phenomena in chemistry and nanoscience.
- ▶ **Construct new models** based on the basic principles in statistical thermodynamics.

# Learning methods

- ▶ One overview lecture on most chapters
- ▶ Some chapters (mathematics) are given only as exercises
- ▶ Around 3-5 suggested exercises per chapter
- ▶ Focus this year: to get the Python exercises integrated in the course

# Communication

- ▶ Reference group - three persons
- ▶ Blackboard for information/messages from teachers
- ▶ Ask questions! Both on new and old topics.
- ▶ Ok to ask the teaching assistants on "old" exercises.



# Some advice

- ▶ Everything is connected in the course. We use all the previous material all the time.
- ▶ Thus:
  - ▶ Work continuously: not everything is easy to understand from the start, but it is the same concepts coming over and over again.
  - ▶ Do not lag behind. Drop some of the recommended exercises on the previous chapter, and return to them later.
  - ▶ Work together in small groups. Both discussing and explaining the text to each other, and solving the exercises.

# Contents

- ▶ What is statistical thermodynamics (statistical mechanics)?
  - ▶ Connection between microscopic and macroscopic theory
- ▶ What is thermodynamics?
  - ▶ Macroscopic theory
- ▶ Nanoscience is at the mesoscopic level. What does that mean?

# Questions to you

- ▶ What is temperature?
- ▶ What is entropy?
- ▶ What is the chemical potential?
- ▶ What is the hydrophobic effect?
- ▶ Why is water an unusual liquid?