Nitrile hydratase

In <u>enzymology</u>, **nitrile hydratases** (NHases; <u>EC</u> <u>4.2.1.84</u> (https://enzyme.expasy.org/EC/4.2.1.84)) are mononuclear <u>iron</u> or non-corrinoid <u>cobalt</u> enzymes that catalyse the hydration of diverse nitriles to their corresponding amides

 $R-C\equiv N + H_2O \rightarrow R-C(O)NH_2$

Contents

Metal cofactor

Metabolic pathway
Industrial applications
Structure
Assembly
Mechanism
References
Further reading

Metal cofactor

In biochemistry, <u>cobalt</u> is in general found in a <u>corrin</u> ring, such as in <u>vitamin B₁₂</u>. Nitrile hydratase is one of the rare enzyme types that use cobalt in a non-corrinoid manner. The mechanism by which the cobalt is transported to NHase without causing toxicity is unclear, although a cobalt <u>permease</u> has been identified, which transports cobalt across the cell membrane. The identity of the metal in the active site of a nitrile hydratase can be predicted by analysis of the sequence data of the alpha subunit in the region where the metal is bound. The presence of the amino acid sequence VCTLC indicates a Co-centred NHase and the presence of VCSLC indicates Fe-centred NHase.

Metabolic pathway

Nitrile hydratase and amidase are two hydrating and hydrolytic enzymes responsible for the sequential metabolism of <u>nitriles</u> in bacteria that are capable of utilising nitriles as their sole source of nitrogen and carbon, and in concert act as an alternative to <u>nitrilase</u> activity, which performs nitrile <u>hydrolysis</u> without formation of an intermediate primary amide. A sequence in genome of the choanoflagellate *Monosiga brevicollis* was suggested to encode for a nitrile hydratase. The *M. brevicollis* gene consisted of both the alpha and beta subunits fused into a single gene. Similar nitrile hydratase genes consisting of a fusion of the beta and alpha subunits have since been identified in several eukaryotic supergroups, suggesting that such nitrile hydratases were present in the last common ancestor of all eukaryotes.

Industrial applications

NHases have been efficiently used for the industrial production of <u>acrylamide</u> from <u>acrylonitrile^[3]</u> on a scale of 600 000 tons per annum, and for removal of nitriles from wastewater. Photosensitive NHases intrinsically possess <u>nitric oxide</u> (NO) bound to the iron centre, and its <u>photodissociation</u> activates the enzyme. <u>Nicotinamide</u> is produced industrially by the hydrolysis of <u>3-cyanopyridine</u> catalysed by the nitrile hydratase from <u>Rhodococcus rhodochrous</u> $J1, \frac{[5][6]}{}$ producing 3500 tons per annum of nicotinamide for use in animal feed.

Structure

NHases are composed of two types of subunits, α and β , which are not related in amino acid sequence. NHases exist as $\alpha\beta$ dimers or $\alpha_2\beta_2$ tetramers and bind one metal atom per $\alpha\beta$ unit. The 3-D structures of a number of NHases have been determined. The α subunit consists of a long extended N-terminal "arm", containing two α -helices, and a C terminal domain with an unusual four-layered structure (α - β - β - α). The β subunit consists of a long N-terminal loop that wraps around the α subunit, a helical domain that packs with N-terminal domain of the α subunit, and a C-terminal domain consisting of a β -roll and one short helix.

Nitrile hydratase, alpha chain		Nitrile hydratase beta subunit	
	Identifiers		Identifiers
Symbol	NHase_alpha	Symbol	NHase_beta
Pfam	PF02979 (http://pfam.xfam. org/family?acc=PF02979)	Pfam	PF02211 (http://pfam.xfam. org/family?acc=PF02211)
InterPro	IPR004232 (https://www.eb i.ac.uk/interpro/entry/IPR00 4232)	InterPro	IPR003168 (https://www.eb i.ac.uk/interpro/entry/IPR00 3168)
SCOP2	2ahj (http://scop2.mrc-lmb.c am.ac.uk/search?t=txt;q=2a hj) / SCOPe (https://scop.be rkeley.edu/pdb/code=2ahj) / SUPFAM (http://supfam.or g/SUPERFAMILY/cgi-bin/se arch.cgi?search_field=2ahj)	SCOP2	2ahj (http://scop2.mrc-lmb.co am.ac.uk/search?t=txt;q=2a hj) / SCOPe (https://scop.be rkeley.edu/pdb/code=2ahj) / SUPFAM (http://supfam.or g/SUPERFAMILY/cgi-bin/se arch.cgi?search_field=2ahj)
Ava	ilable protein structures:	Ava	ilable protein structures:
Pfam	structures (http://pfam.xfam.or g/family/PF02979?tab=pdbBl ock) / ECOD (http://prodata.s wmed.edu/ecod/complete/sea rch?kw=PF02979)	Pfam	structures (http://pfam.xfam.or g/family/PF02211?tab=pdbBlo ck) / ECOD (http://prodata.sw med.edu/ecod/complete/searc h?kw=PF02211)
PDB	RCSB PDB (https://www.rcsb.org/search?q=rcsb_polymer_entity_annotation.annotation_id:PF02979%20AND%20rcsb_polymer_entity_annotation.type:Pfam); PDBe (https://www.ebi.ac.uk/pdbe/entry/search/index?pfam_accession:PF02979); PDBj (https://pdbj.org/searchFor?query=PF02979)	PDB	RCSB PDB (https://www.rcsb.org/search?q=rcsb_polymer_entity_annotation.annotation_id:PF02211%20AND%20rcsb_polymer_entity_annotation.type:Pfam); PDBe (https://www.ebi.ac.uk/pdbe/entry/search/index?pfam_accession:PF02211); PDBj (https://pdbj.org/searchFor?query=PF02211)
PDBsum	structure summary (https://www.ebi.ac.uk/thornton-srv/databases/cgi-bin/pdbsum/GetPfamStr.pl?pfam_id=PF02979)	PDBsum	structure summary (https://www.ebi.ac.uk/thornton-srv/databases/cgi-bin/pdbsum/GetPfamStr.pl?pfam_id=PF02211)
PDB	1ahj (https://www.rcsb.org/structure/1ahj), 1ire (https://www.rcsb.org/structure/1ire), 1ugp (https://www.rcsb.org/structure/1ugp), 1ugq (https://www.rcsb.org/structure/1ugq), 1ugr (https://www.rcsb.org/structure/1ugr), 1ugs (https://wwww.rcsb.org/structure/1ugs), 11/20 (https://www.rcsb.org/structure/1ugs),	PDB	1ahj (https://www.rcsb.org/structure/1ahj), 1ire (https://www.rcsb.org/structure/1ire), 1ugp (https://www.rcsb.org/structure/1ugp), 1ugq (https://www.rcsb.org/structure/1ugq), 1ugr (https://www.rcsb.org/structure/1ugr), 1ugs (https://wwww.rcsb.org/structure/1ugs), 2chi (https://www.rcsb.org/structure/1ugs), 2chi (https://www.rcs

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2cyz (https://www.rcsb.org/str

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2cz1 (https://www.rcsb.org/str

w.rcsb.org/structure/2ahj),

w.rcsb.org/structure/2cz0),

	Databases		
IntEnz	IntEnz view (https://www.ebi.ac.uk/intenz/query?comd=SearchEC&ec=4.2.2		
BRENDA	BRENDA entry (http://www.brenda-enzymes.org/enzyme.php?ecno=4.2.1.4)		
ExPASy	NiceZyme view (https://enzyme.expasy.org/EC/42.1.84)		
KEGG	KEGG entry (https://www.genome.jp/dbget-bin/www.w_bget?enzyme+4.2.1.84)		
MetaCyc	metabolic pathway (http s://biocyc.org/META/sub tring-search?type=NIL&d bject=4.2.1.84)		
PRIAM	profile (http://priam.prabi r/cgi-bin/PRIAM_profiles CurrentRelease.pl?EC=4 2.1.84)		
PDB structure	RCSB PDB (https://www.rcsb.org/search?q=rcsb.polymer_entity.rcsb_ec_neage.id:4.2.1.84) PDB (https://www.ebi.ac.uk/pbe/entry/search/index?e_number:4.2.1.84) PDBsum (https://www.ebi.ac.uk/thornton-srv/dataases/cgi-bin/enzymes/GtPage.pl?ec_number=42.1.84)		
Gene Ontology	AmiGO (http://amigo.ger eontology.org/amigo/ter m/GO:0018822) / QuickGO (https://www.el i.ac.uk/QuickGO/term/G O:0018822)		
	Search		
PMC	articles (https://www.ncbi.nlm.nih.gov/entrez/query.fcgi?db=pubmed&term=4.2.1.84%5BE C/RN%20Number%5D%20AN D%20pubmed%20pmc%20local%5Bsb%5D)		
PubMed	articles (https://www.ncbi.nlm. nih.gov/entrez/query.fcgi?db=p ubmed&term=4.2.1.84%5BE C/RN%20Number%5D)		
NCBI	oroteins (https://www.ncbi.nli nih.gov/protein?term=4.2.1.8 4%5BEC/RN%20Number%5 D)		

nitrile hydratase

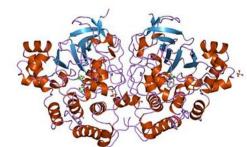
Identifiers

4.2.1.84 (https://www.qm ul.ac.uk/sbcs/iubmb/enzy me/EC4/2/1/84.html)

82391-37-5 (http://www.c ommonchemistry.org/Che

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Structure of nitrile hydratase.[7]

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2cz7 (https://www.rcsb.org/structure/2cz7), 2d0q (https://wwww.rcsb.org/structure/2d0q),
2qdy (https://www.rcsb.org/structure/2qdy)

ucture/2cz6), 2cz7 (https://ww w.rcsb.org/structure/2cz7), 2d0q (https://www.rcsb.org/str ucture/2d0q), 2dpp (https://w ww.rcsb.org/structure/2dpp), 2qdy (https://www.rcsb.org/str ucture/2qdy), 2zcf (https://ww w.rcsb.org/structure/2zcf), 2zpb (https://www.rcsb.org/str ucture/2zpb), 2zpe (https://ww w.rcsb.org/structure/2zpe), 2zpf (https://www.rcsb.org/str ucture/2zpf), 2zpg (https://ww w.rcsb.org/structure/2zpg), 2zph (https://www.rcsb.org/str ucture/2zph), 2zpi (https://ww w.rcsb.org/structure/2zpi)

Assembly

An assembly pathway for nitrile hydratase was first proposed when gel filtration experiments found that the complex exists in both $\alpha\beta$ and $\alpha2\beta2$ forms. [8] In vitro experiments using mass spectrometry further revealed that the α and β subunits first assemble to form the $\alpha\beta$ dimer. The dimers can then subsequently interact to form a tetramer. [9]

Mechanism

The metal centre is located in the central cavity at the interface between two subunits. All protein ligands to the metal atom are provided by the α subunit. The protein ligands to the iron are the sidechains of the three <u>cysteine</u> (Cys) residues and two mainchain amide nitrogens. The metal ion is octahedrally coordinated, with the protein ligands at the five vertices of an octahedron. The sixth position, accessible to the active site cleft, is occupied either by NO or by a solvent-exchangeable ligand (hydroxide or water). The two Cys residues coordinated to the metal are post-translationally modified to Cys-<u>sulfinic</u> (Cys-SO₂H) and -<u>sulfenic</u> (Cys-SOH) acids.

Quantum chemical studies predicted that the Cys-SOH residue might play a role as either a base (activating a nucleophilic water molecule)^[10] or as a nucleophile.^[11] Subsequently, the functional role of the SOH center as nucleophile has obtained experimental support.^[12]

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