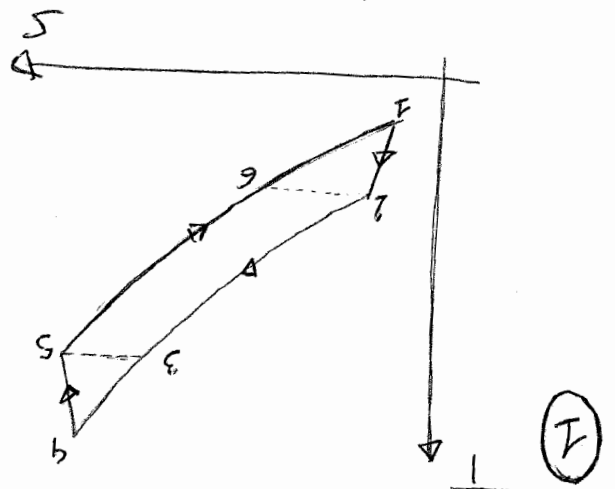


11/09/2018



Compression 1 → 2  
 $\frac{T_2}{T_1} = \left(\frac{P_2}{P_1}\right)^{\frac{\gamma-1}{\gamma}}$   
 $\Rightarrow T_{2,15} = T_1 P_{2,15}^{\frac{\gamma-1}{\gamma}}$

$T_1 = 200^\circ\text{C}$   $P_1 = 1 \text{ bar}$   
 $\frac{P_2}{P_1} = P_{2,15} = 2,8$   
 $T_2 = 1200^\circ\text{C}$   
 $P_3 - P_4 = 0,05 \text{ bar} = \Delta P_{\text{max}}$

$\gamma = 5/3$  (Gas Monatomic)

$\frac{T_{2,15} - T_1}{T_2 - T_1} = \eta_{15, \text{carn}} \Rightarrow T_2 = \frac{T_{2,15} - T_1}{\eta_{15, \text{carn}}} + T_1 = 501^\circ\text{C}$

$P_2 = P_1 P_{2,15} = 2,8 \text{ bar}$

$S_2 = C_p \ln \frac{T_2}{T_1} - R^* \ln \frac{P_2}{P_1} = 42,1 \frac{\text{J}}{\text{K}}$

Expansion 4 → 5  
 $\frac{T_5}{T_4} = \left(\frac{P_5}{P_4}\right)^{\frac{\gamma-1}{\gamma}}$   
 $P_{5,15} = P_4 = 2,667$

$T_{5,15} = 722^\circ\text{C}$

$\frac{T_4 - T_5}{T_4 - T_{5,15}} = \eta_{15, \text{turb}} \Rightarrow T_5 = T_4 - \eta_{15, \text{turb}} (T_4 - T_{5,15}) + T_4 = 880^\circ\text{C}$

Ciclo refrigerativo locale

$$T_6 = T_2 \quad T_5 = T_3$$

$$P_2 = P_1 + \Delta P_{can} \quad P_6 = P_1 + \Delta P_{can} = 1,05 \text{ bar}$$

Per il calcolo dell'entropia dei vari punti si utilizza la relazione

riportata per il calcolo di \$S\_2\$

$$Q_{netto} = \left[ \overbrace{C_p (T_4 - T_5)}^{\text{ESPANSIONE}} - \overbrace{C_p (T_2 - T_1)}^{\text{COMPRESSIONE}} \right] \cdot m$$

$$m_{Aron} = \frac{Q_{netto}}{f_{EL}} = \frac{1,2 \text{ MWel}}{26,54 \text{ W/s}} = 26,54 \text{ kg/s}$$

$$Q_{in,ciclo} = m_{Aron} C_p (T_3 - T_2) = 5,407 \text{ MW}$$

$$\eta_I = \frac{P_{EL}}{Q_{in,ciclo}} = 0,2219$$

$$Q_{netto} = m_{Aron} C_p (T_6 - T_1) = 4,16 \text{ MW}$$

$$\textcircled{2} \quad g z_A + \frac{P_A}{\rho} + \frac{V_A^2}{2} = g z_B + \frac{P_B}{\rho} + \frac{V_B^2}{2} + K_{AO} \frac{V_{AO}^2}{2} + K_{DB} \frac{V_{DB}^2}{2}$$

$$V_A = V_B = 0 \text{ m/s (SERBATOIO } \rightarrow \infty) \quad P_A = P_B = 1 \text{ bar}$$

$$V_{AO} = V_{DB} = V_{netto}$$

$$V_{netto} = \sqrt{\frac{g(z_A - z_B)}{\left( \frac{K_{AO} + K_{DB}}{2} \right)}} = 1,98 \text{ m/s}$$

$$\dot{m}_{AO} = \int V_{netto} \cdot \rho \, dA = 1,05 \text{ kg/s}$$

②

$$m_{ob} = m_{oc} = \frac{m_{ao}}{2} = 0,52516/s$$

$$D_{ob} = \sqrt{\frac{m_{ob} \cdot 4}{\pi}} = D_{oc} = 0,02121 \text{ m}$$

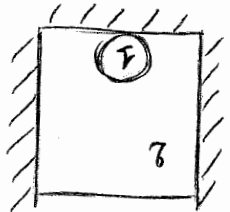
$$\frac{\rho}{2} + g z + \frac{V^2}{2} = \frac{\rho}{2} + g z_c + \frac{V_c^2}{2} + K_{ao} \frac{V_{tubo}^2}{2} + K_{oc} \frac{V_{tubo}^2}{2}$$

$$(z_A - z_c) = \left( (K_{ao} + K_{oc}) \frac{V_{tubo}^2}{2} \right) / g = -1,8 \text{ m}$$

$$z_B - z_c = -0,2 \text{ m}$$

3) • Calcolo Temperatura di Equilibrio

La grana cede calore al liquido contenuto nel recipiente fino al raggiungimento della Temperatura di Equilibrio

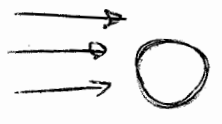


$$M_1 C_1 (T_{in}^{STEM} - T_{eq}) = M_2 C_2 (T_{eq} - T_{in}^{LIQ})$$

$$\frac{M_1 C_1 T_{in}^{STEM}}{M_2 C_2} - \frac{M_1 C_1 T_{eq}}{M_2 C_2} = T_{eq} - T_{in}^{LIQ}$$

$$T_{eq} = \left( \frac{M_1 C_1 T_{in}^{STEM}}{M_2 C_2} + T_{in}^{LIQ} \right) / \left( 1 + \frac{M_1 C_1}{M_2 C_2} \right) = 108,1^\circ \text{C}$$

• Processo di Raffreddamento della grana per convezione forzata.  $T_{\infty} = 15^\circ \text{C}$



$$L_c = \frac{\frac{4}{3} \pi R^3}{K_{STEM}} = \frac{4 \pi R^3}{3 K_{STEM}} \quad \alpha = \frac{\int_{STEM} C_{STEM}}{K_{STEM}} = 5,05 \cdot 10^{-5} \frac{1}{m} \quad T_{av} = 607,29$$

$$\frac{\theta_{t=0}}{\theta_{t=s_{min}}} = 0 \quad -Bi \cdot F_{av} - \frac{h L_c}{K_{STEM} L_c^2} \alpha t = 0$$

$$Bi = \ln \left( \frac{T_{\infty} - T_{min}}{T_{\infty} - T_{eq}} \right) / \frac{F_{av}}{K_{STEM}} = 2,16 \cdot 10^{-3} < 0,1 \quad (\text{ARC Approccio a Parametri Concentrati})$$

$$h = \frac{Q_c}{L_c} = 52,12 \frac{W}{m^2K}$$

ENERGIA SEPARADA DURANTE IL PROCESSO ISOTERMICO (COSTA DI ANNO)

$$E_{sew} = C_{sew} \int_{T_{sew}}^{T_{cp}} (T_{cp} - T_{sew}) = 2,378 KJ$$

④

$T_1 = 460 K$   $X_1 = 0$  (acqua calda)

$h_1(T_1, X_1) = 793,406 \frac{KJ}{kg}$

LAUREAZIONE ANONIMA  $\Rightarrow h_2 = h_1$  ;  $T_2 = 99,60^\circ C$   $X_2 = 0,1665$

SEPARAZIONE

$$m_2 = m_{3vs} + m_{32s}$$

$$m_{3vs} = m_2 \cdot X_2 = 1,998 \frac{kg}{s}$$

ESPRESSIONE ISOTERMICA

$$h_4(T_4, s_4 = s_{3vs}) = 2338,89 \frac{KJ}{kg}$$

$$s_4(T_4, s_4 = s_{3vs}) = 12,51 \frac{m^3}{kg}$$

FORMA TERMICA GLOBALE SEPARAZIONE = 0

FORMA MECCANICA =  $m_{3vs} (h_{3vs} - h_4)$

$$V_4 = m_{3vs} \cdot V_4 = 25 \frac{m^3}{s}$$

