



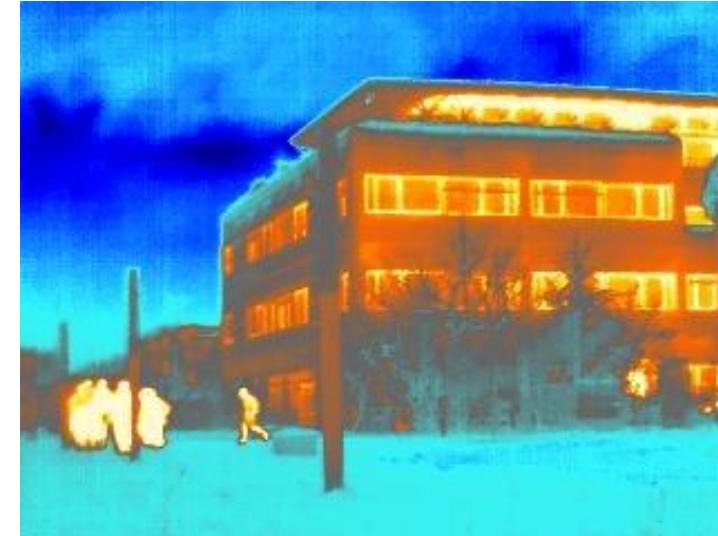
Photo: A. Black, UBC

## 07 Long-wave radiation.

## Learning objectives

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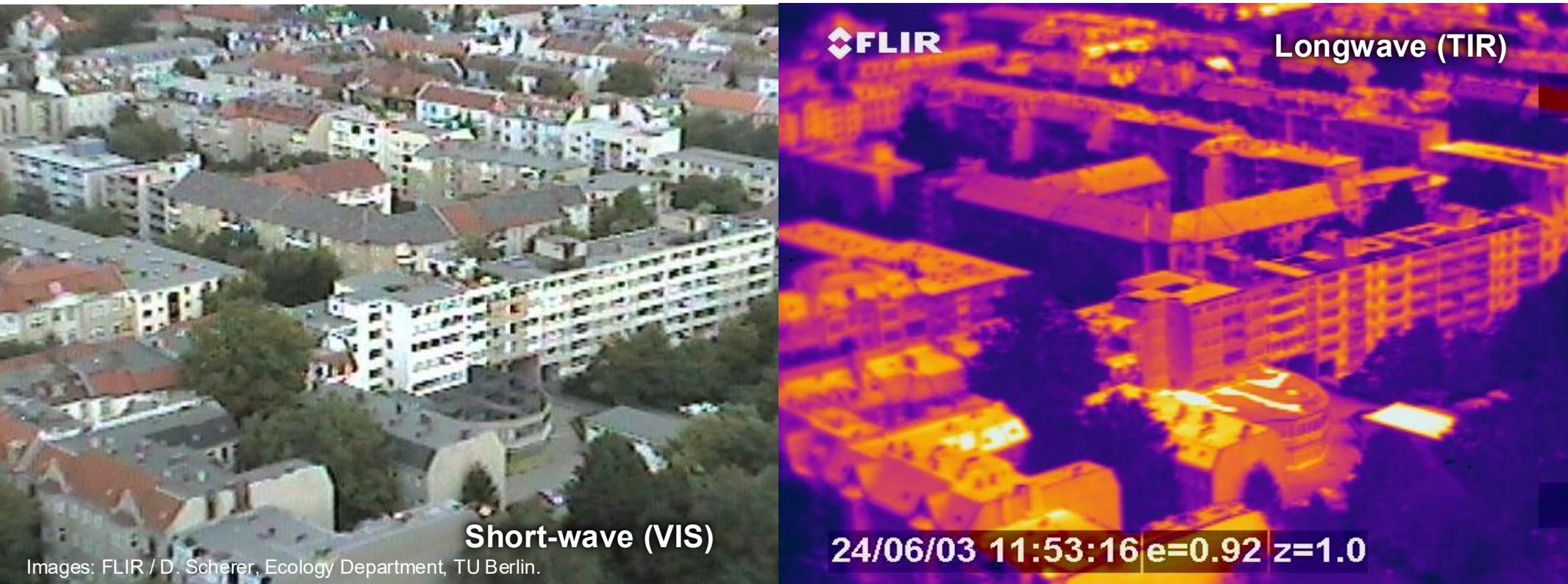
- Explain how radiation laws apply to long-wave radiative exchange as well.
- Describe how long-wave radiation can interact with surfaces.
- Explain how we can calculate longwave outgoing radiation, and how it relates to surface emissivity.



People in front of a heated building in winter as seen in the thermal infrared by a thermal camera (Source: A. Christen)

# What is ‘Longwave’ radiation?

- Wavelength range: 3  $\mu\text{m}$  to 100  $\mu\text{m}$
- Longwave = far-infrared = thermal infrared radiation (TIR)



# Measuring long-wave radiation

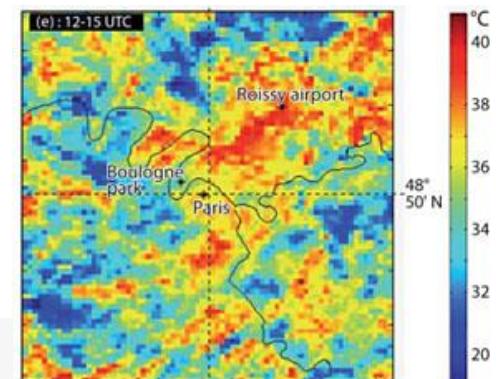
There is a range of instruments available that measure long-wave radiation received in a particular **field of view** (FOV), within a particular **band**, and/or with a particular spatial **resolution**.



Pyrgeometer



Thermal camera



Thermal satellite channel

## Review: Stefan-Boltzmann law: grey body

Natural objects are not full radiators.

The emittance from these objects (called grey bodies) is given by

$$E_g = \varepsilon \sigma T^4 \quad \star$$

where  $\varepsilon$  is their **surface emissivity**.

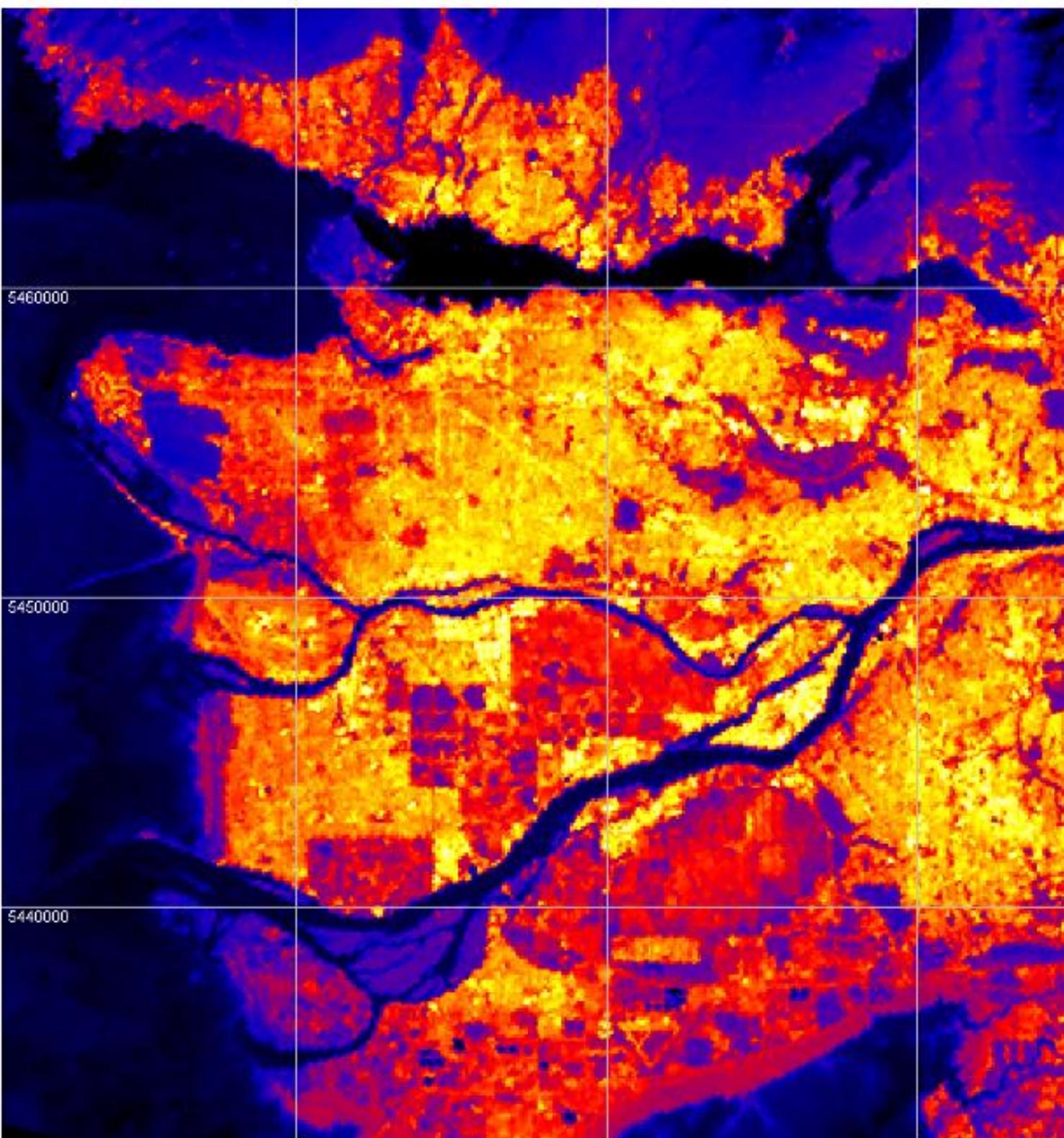
Emissivity is the ratio of the actual emission to that of a blackbody (i.e.  $\varepsilon = 1.0$ ).

This law is the basis of remote sensing in the TIR incl. satellite sensors.

Surface                          Emissivity  $\varepsilon^*$

Soil	0.90 – 0.98
Grass	0.90 – 0.95
Crops	0.90 – 0.99
Forests	0.97 – 0.99
Water	0.92 – 0.97
Iron	0.13 – 0.28

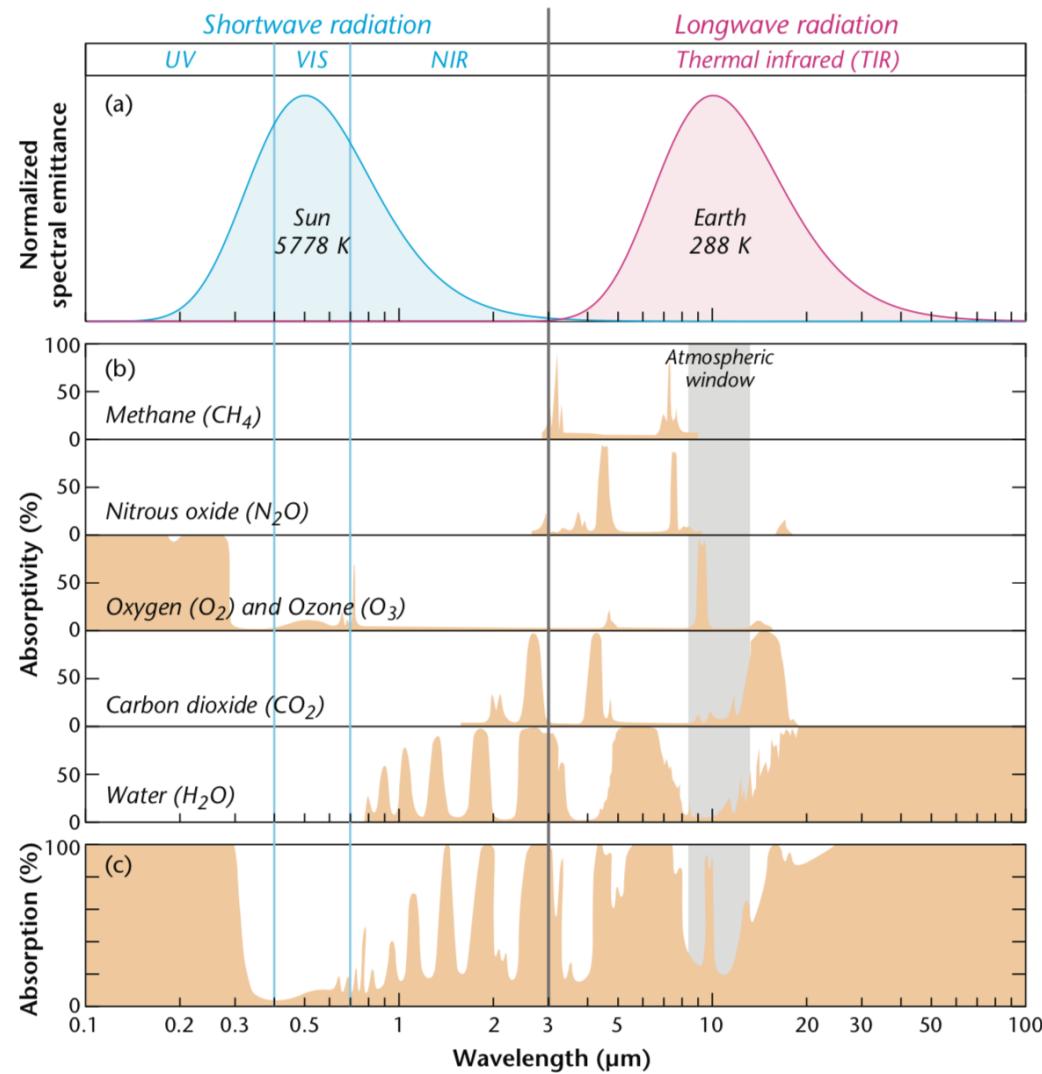
\* in the long-wave range of the spectrum (relevant for microclimates)



Sept 3 2010  
12:24 PDT

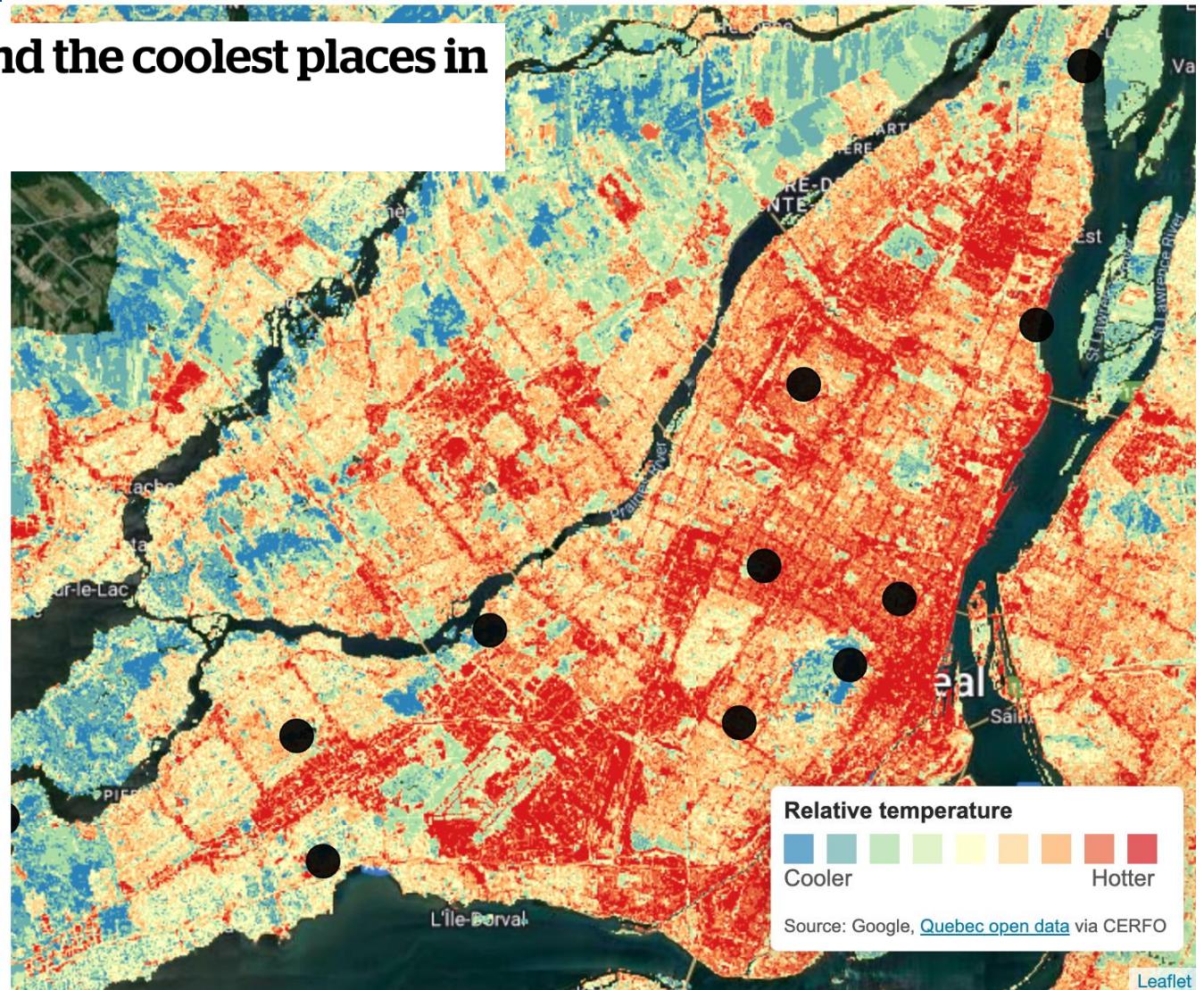
**Emittance of**  
**Vancouver**  
**seen from**  
**ASTER**  
**satellite**

# Atmospheric window & remote sensing in the TIR



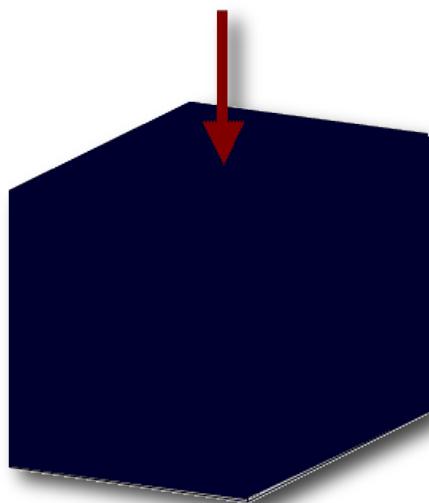
# From emittance we can estimate surface temperature

Where are the warmest and the coolest places in Montreal?

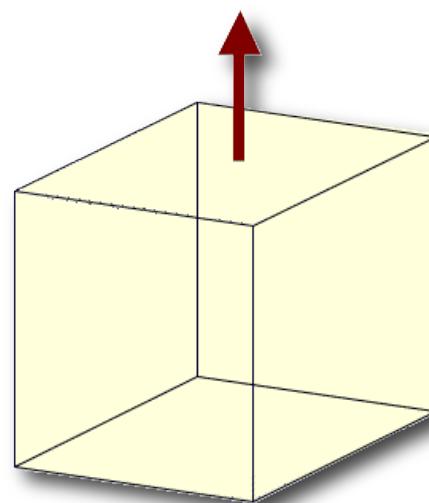


## Review: Kirchhoff's law

Absorptivity



Emissivity



Assuming no transmission  
the absorptivity of a body ( $\zeta_\lambda$ )  
equals its emissivity ( $\varepsilon_\lambda$ ) at a  
given wavelength.

*A good absorber is a good  
emitter*

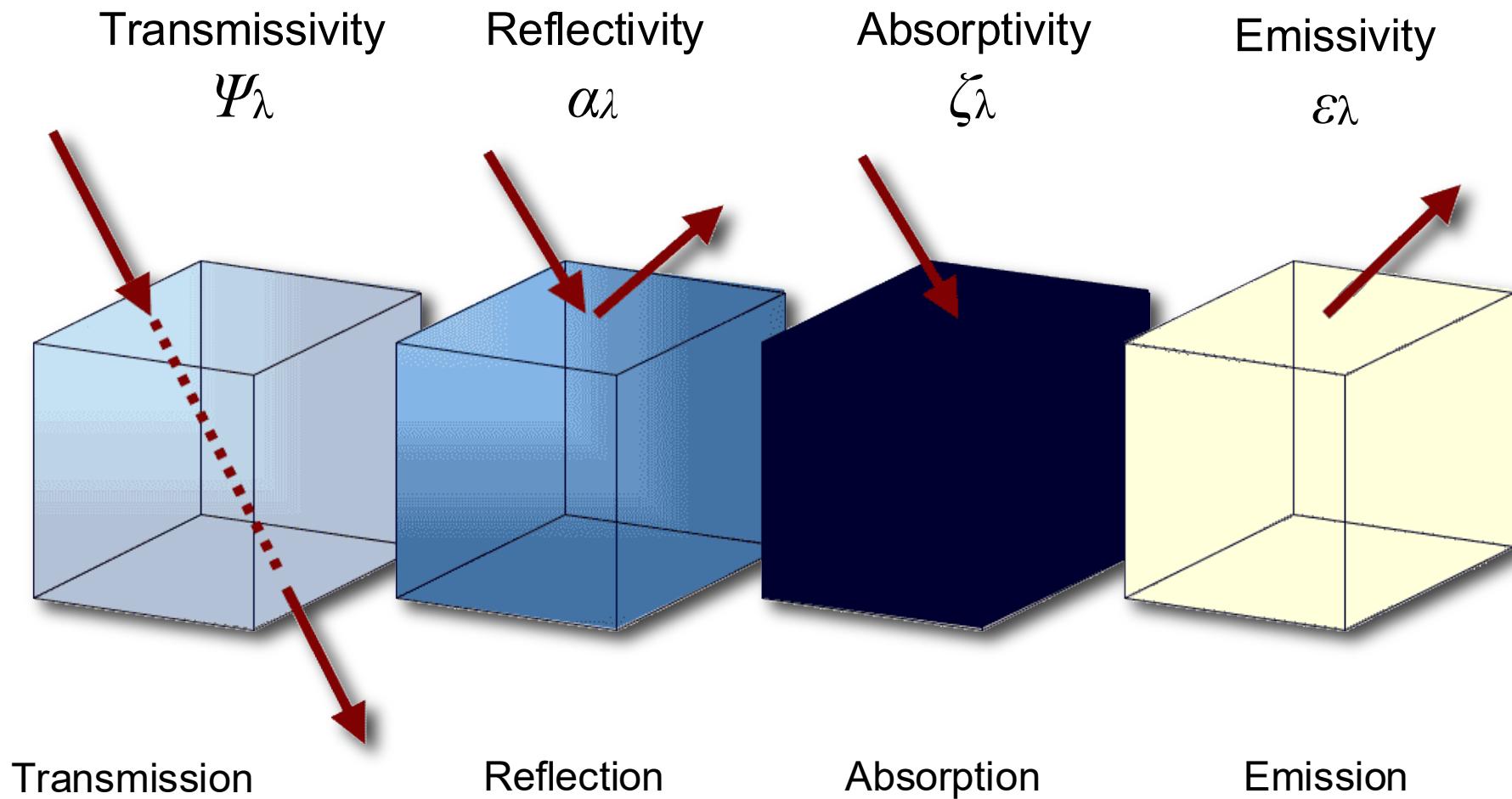
$$\zeta_\lambda = \varepsilon_\lambda \quad \star$$

## Kirchhoff's law of thermal radiation

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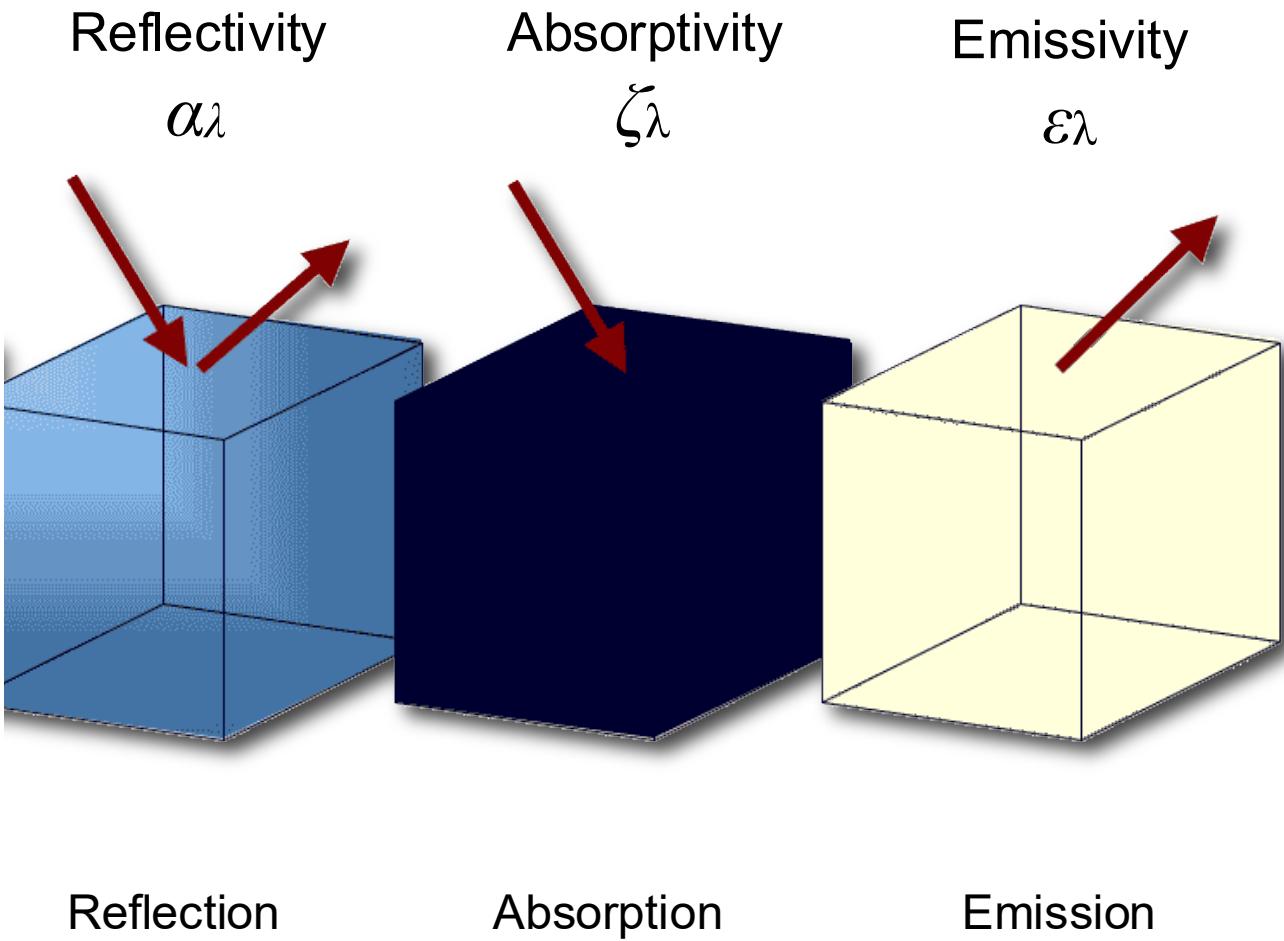
- Kirchhoff's law only applies if the wavelength considered is the same – do not mix them together.
- Kirchhoff's law only has relevance to **long-wave exchange** in climatology.
- The law does not apply to fluorescent objects, which can absorb energy at a given a wavelength and release it at another one.

# Mass-radiation interactions



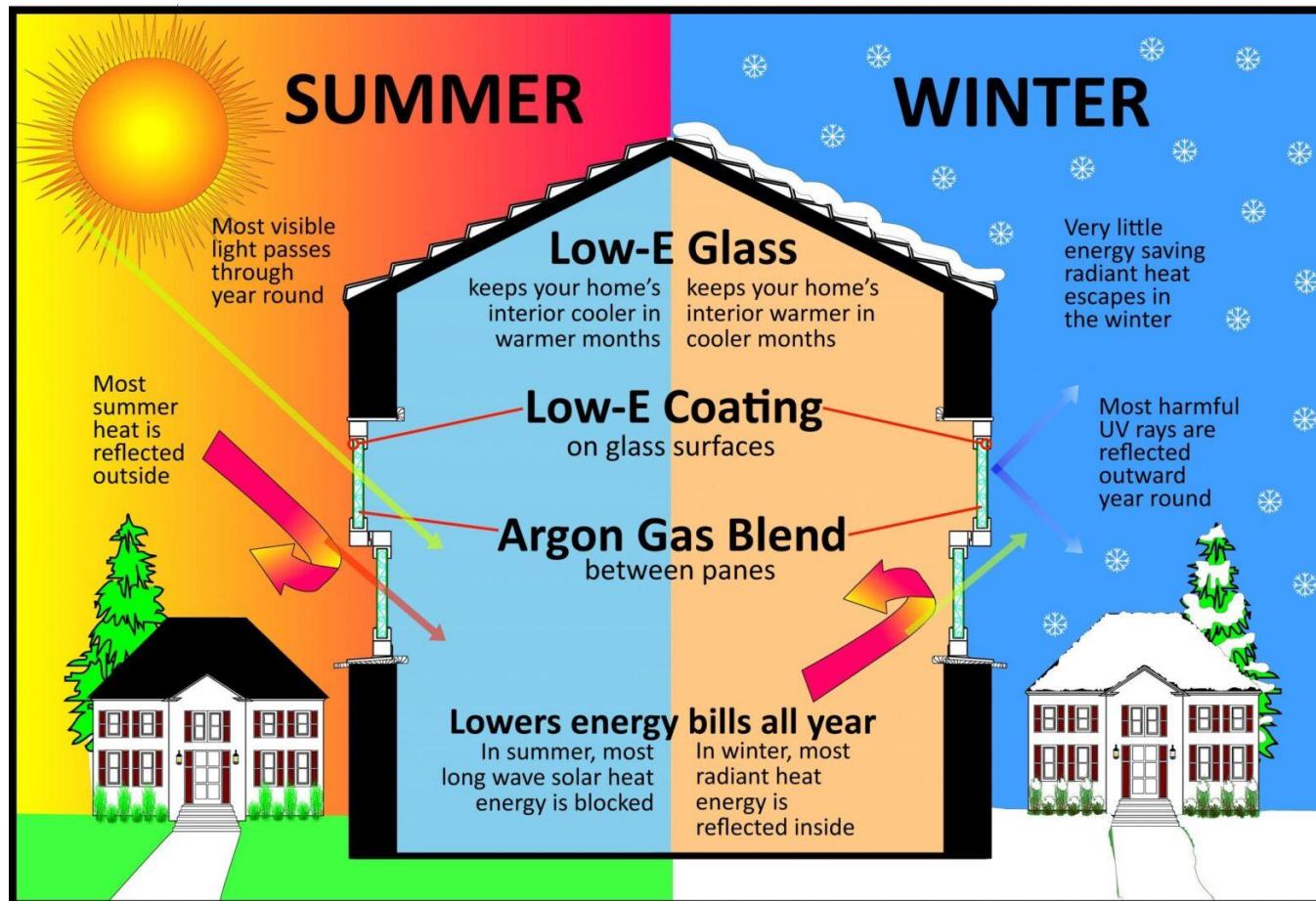
**most surfaces are opaque to long-wave radiation therefore absorptivity typically = emissivity**

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# Low emissivity = reflects much of the radiant thermal energy

Low-e glass as one practical example of a low emissivity surface



# Is long-wave reflection important?

Energy  
conservation

Transmissivity

Absorptivity

Reflectivity

$$\psi_{o,LW} + \zeta_{o,LW} + \alpha_{o,LW} = 1.0$$



However, most surfaces are opaque to long-wave (i.e.,  $\psi_{o,LW} \sim 0$ ) so

$$\zeta_{o,LW} + \alpha_{o,LW} = 1.0$$

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and since  $\zeta_{o,LW} = \epsilon_{o,LW}$  (according to Kirchhoff's Law)

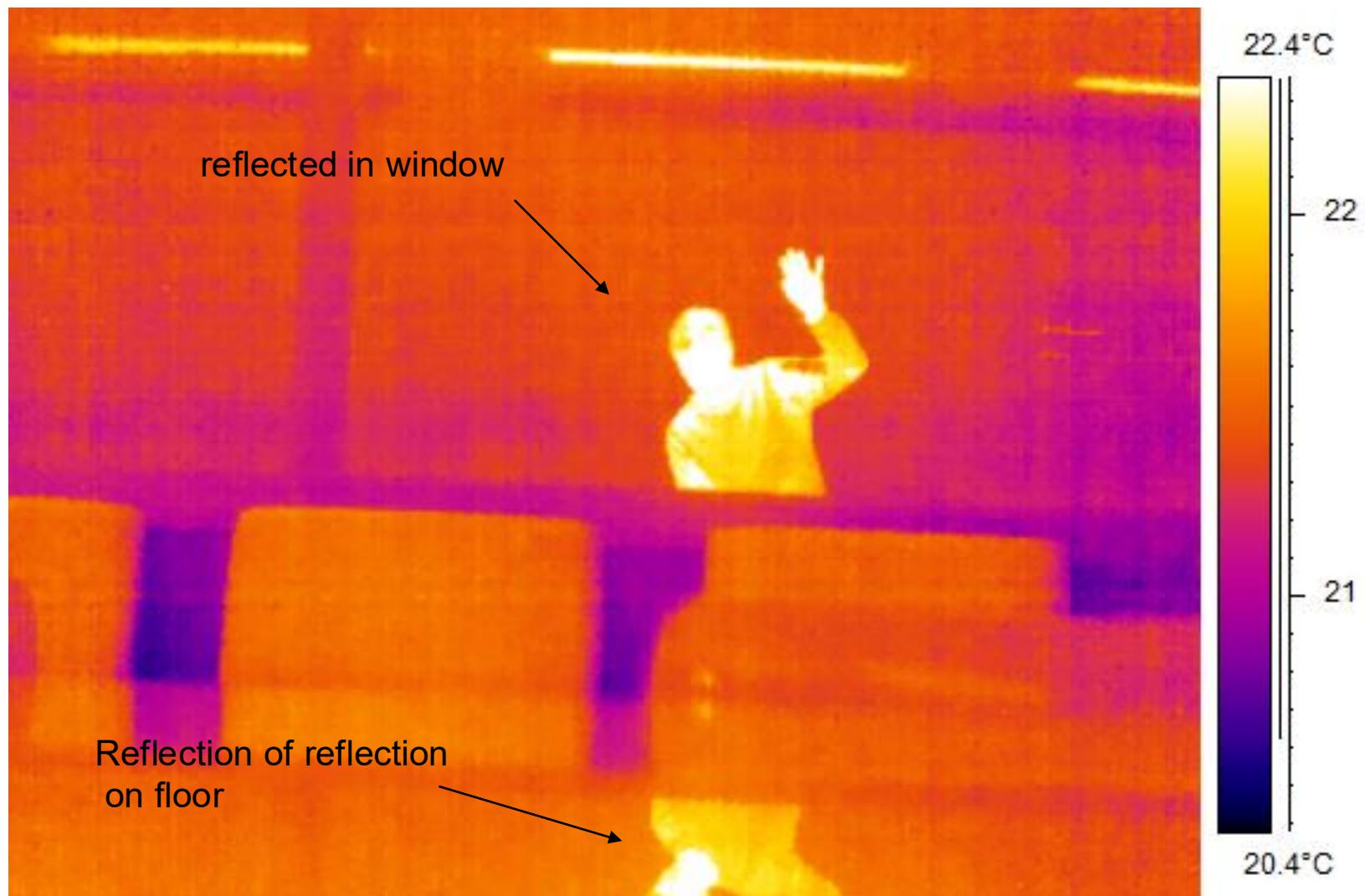
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$$\alpha_{o,LW} =$$

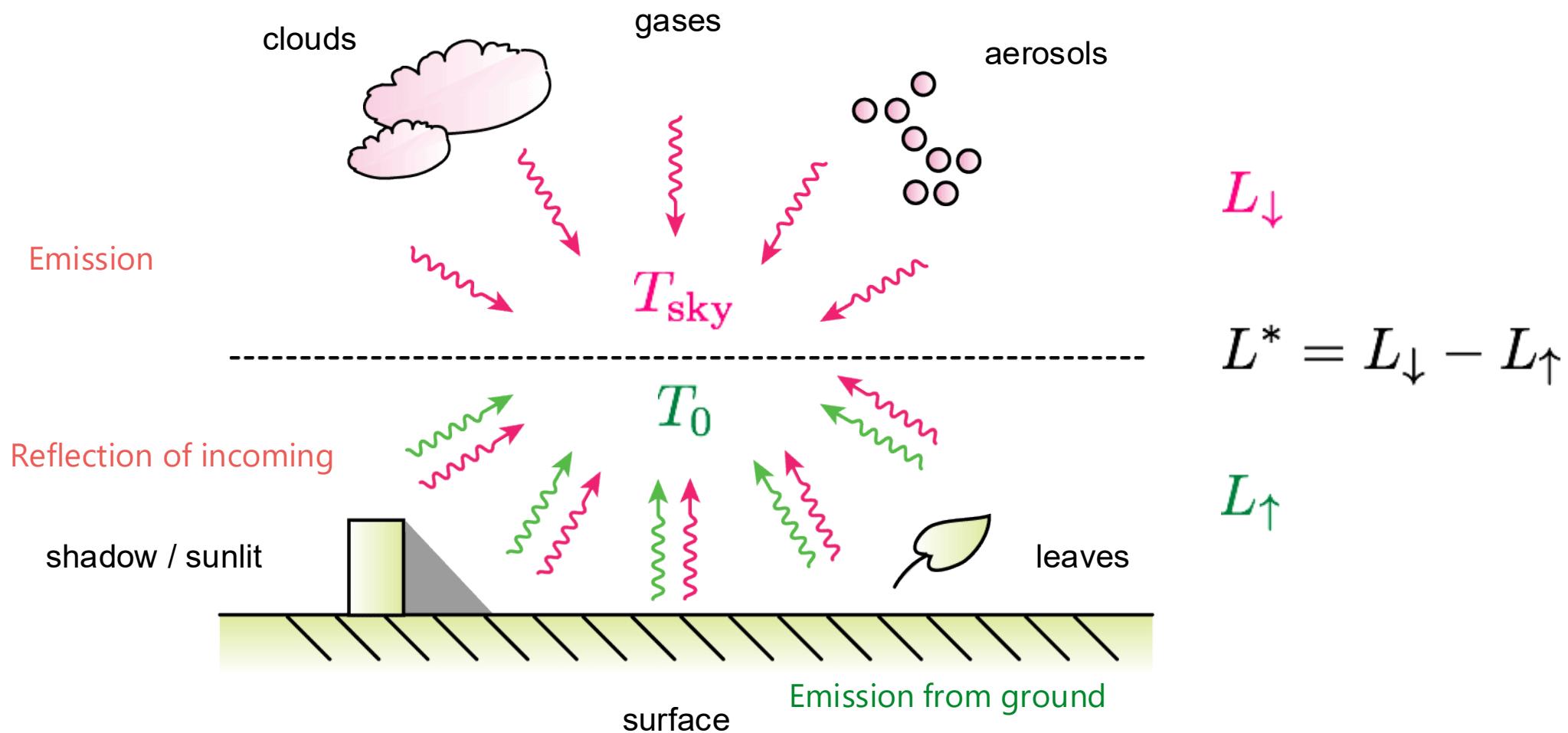


The subscript '0' refers to values of the land surface (in contrast to atmosphere with subscript *a*). In the upcoming slides we will implicitly assume values are in the long-wave band, and omit the subscript 'LW'

# Long-wave reflection



# Long-wave radiation exchange of a surface



## Calculating $L_{\uparrow}$

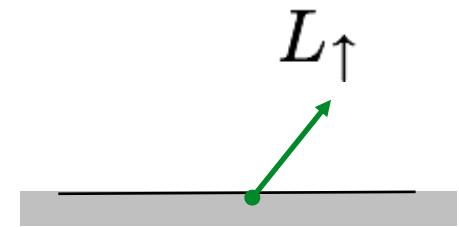
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$$\varepsilon = \frac{\text{radiative flux density emitted by a body}}{\text{radiative flux density emitted by a blackbody}}$$

defines blackbody as  $\varepsilon = 1.0$ , and grey bodies as  $\varepsilon < 1.0$ .

$$L_{\uparrow} = \varepsilon_o \sigma T_o^4$$

emission

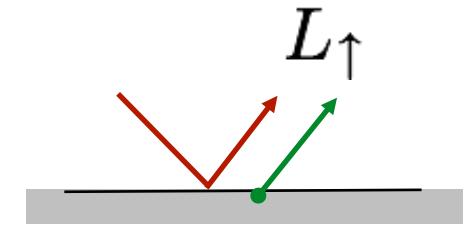


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$$L_{\uparrow} = \varepsilon_o \sigma T_o^4 + \underbrace{(1 - \zeta_o)}_{\alpha_o} L_{\downarrow}$$

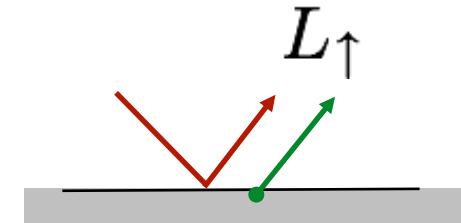


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where  $\zeta_o$  - absorptivity of surface in the long-wave,  $\alpha_o$  reflectivity of surface in the long-wave and from Kirchhoff's law ( $\zeta_{\lambda} = \varepsilon_{\lambda}$ ):

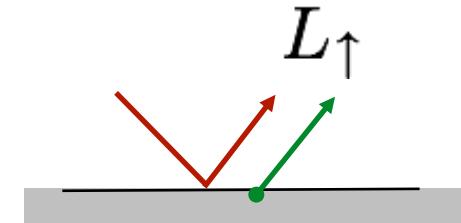
where subscript '0' indicates surface value.

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where subscript '0' indicates surface value.

## Net long-wave radiation flux density $L^*$

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The long-wave net radiation  $L^*$  at the surface is the difference between the input from above  $L_{\downarrow}$  and the output from emission and reflected  $L_{\uparrow}$ :

$$L^* = L_{\downarrow} - L_{\uparrow}$$

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$$L^* = L_{\downarrow} - (\varepsilon_o \sigma T_o^4 + \overbrace{(1 - \zeta_o) L_{\downarrow}}^{\alpha_o})$$

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$$L^* = L_{\downarrow} - L_{\uparrow}$$

$$L^* = L_{\downarrow} - (\varepsilon_o \sigma T_o^4 + \overbrace{(1 - \zeta_o) L_{\downarrow}}^{\alpha_o})$$

$$L^* = L_{\downarrow} - (\varepsilon_o \sigma T_o^4 + (1 - \varepsilon_o) L_{\downarrow})$$

emitted

reflected

Kirchhoff's  
Law

## Question for discussion

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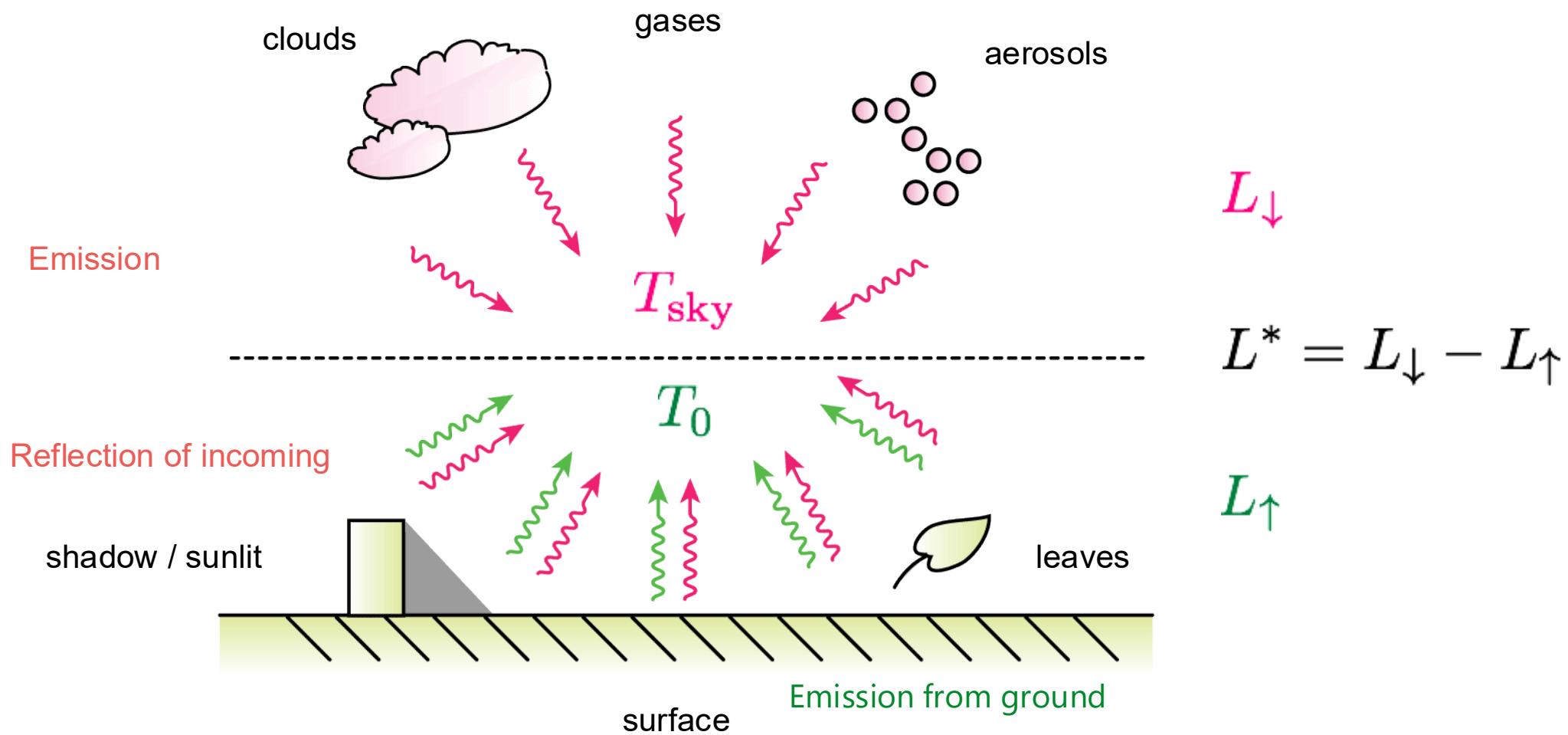
What influences  $L_{\downarrow}$ ?

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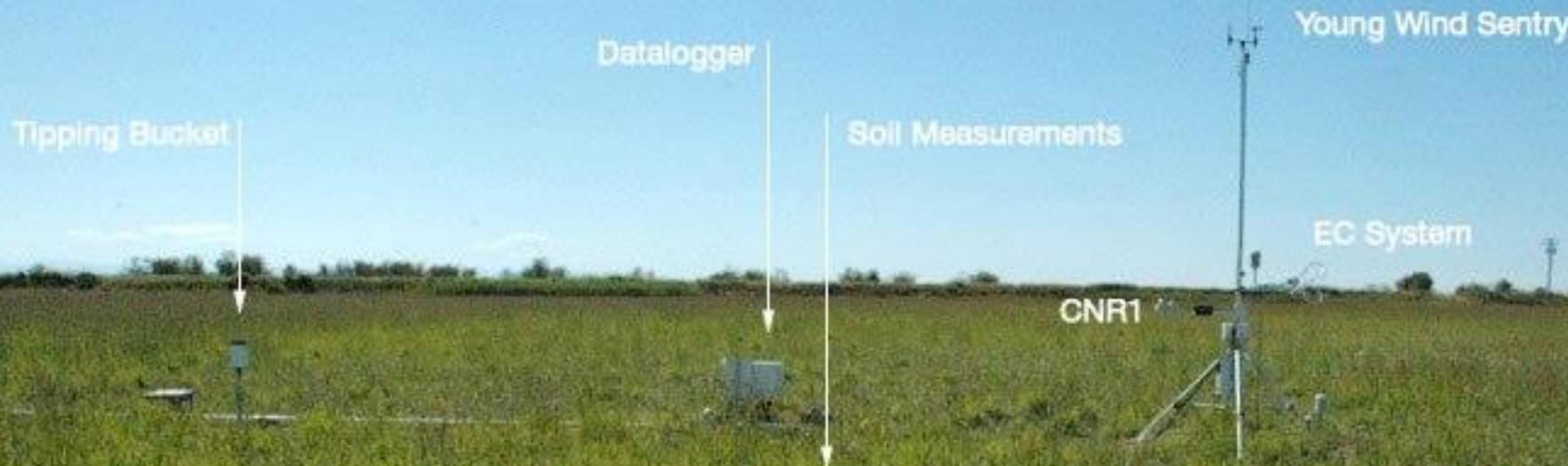
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# Long-wave radiation exchange of a surface



# Long-wave radiation - example of measurements



UBC Energy Balance Station “Westham Island”

Photo: A. Christen

# Long-wave radiation - example of measurements

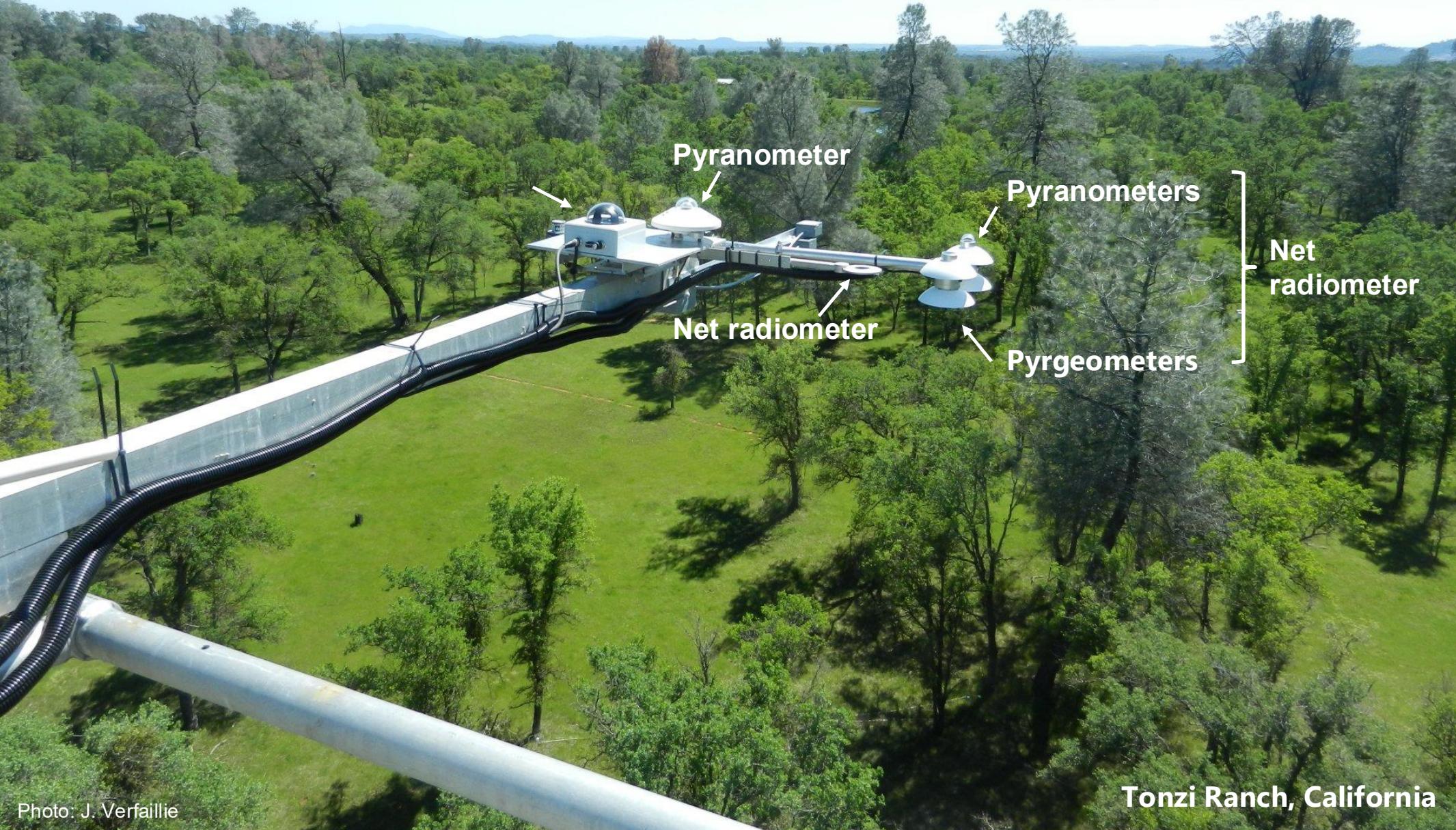
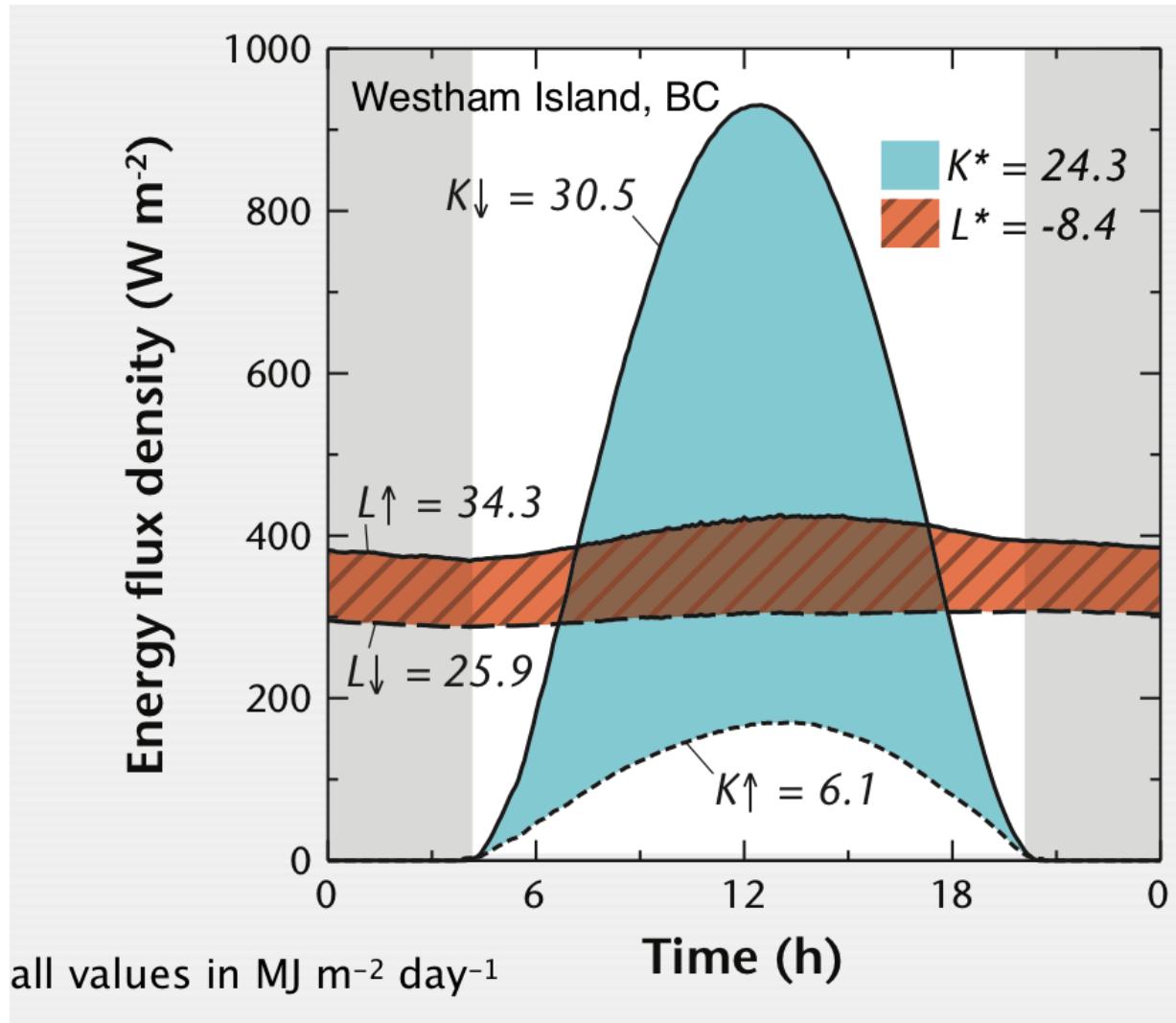


Photo: J. Verfaillie

Tonzi Ranch, California

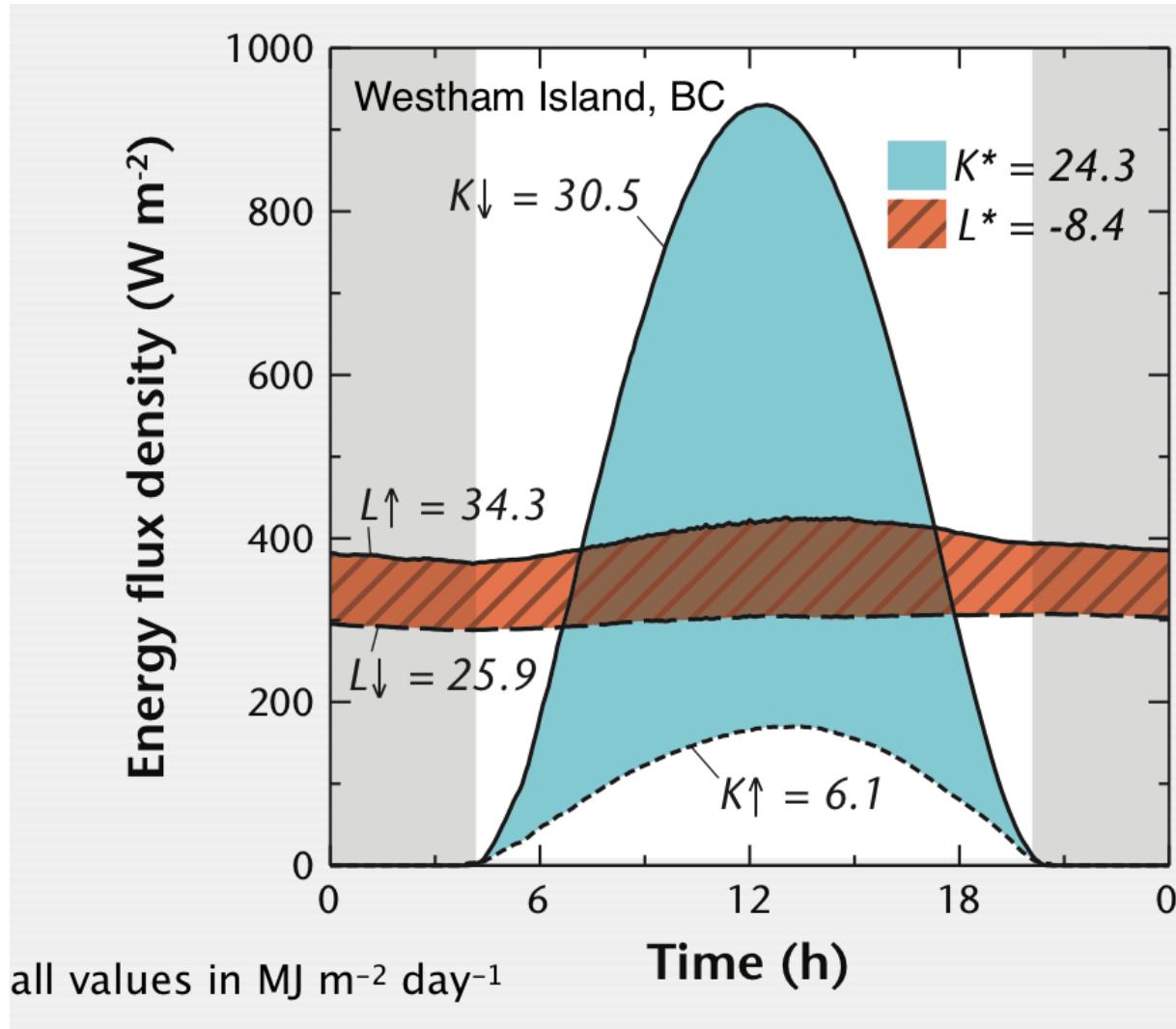
# Diurnal course of radiation - clear sky day



Images from Katkam in Vancouver

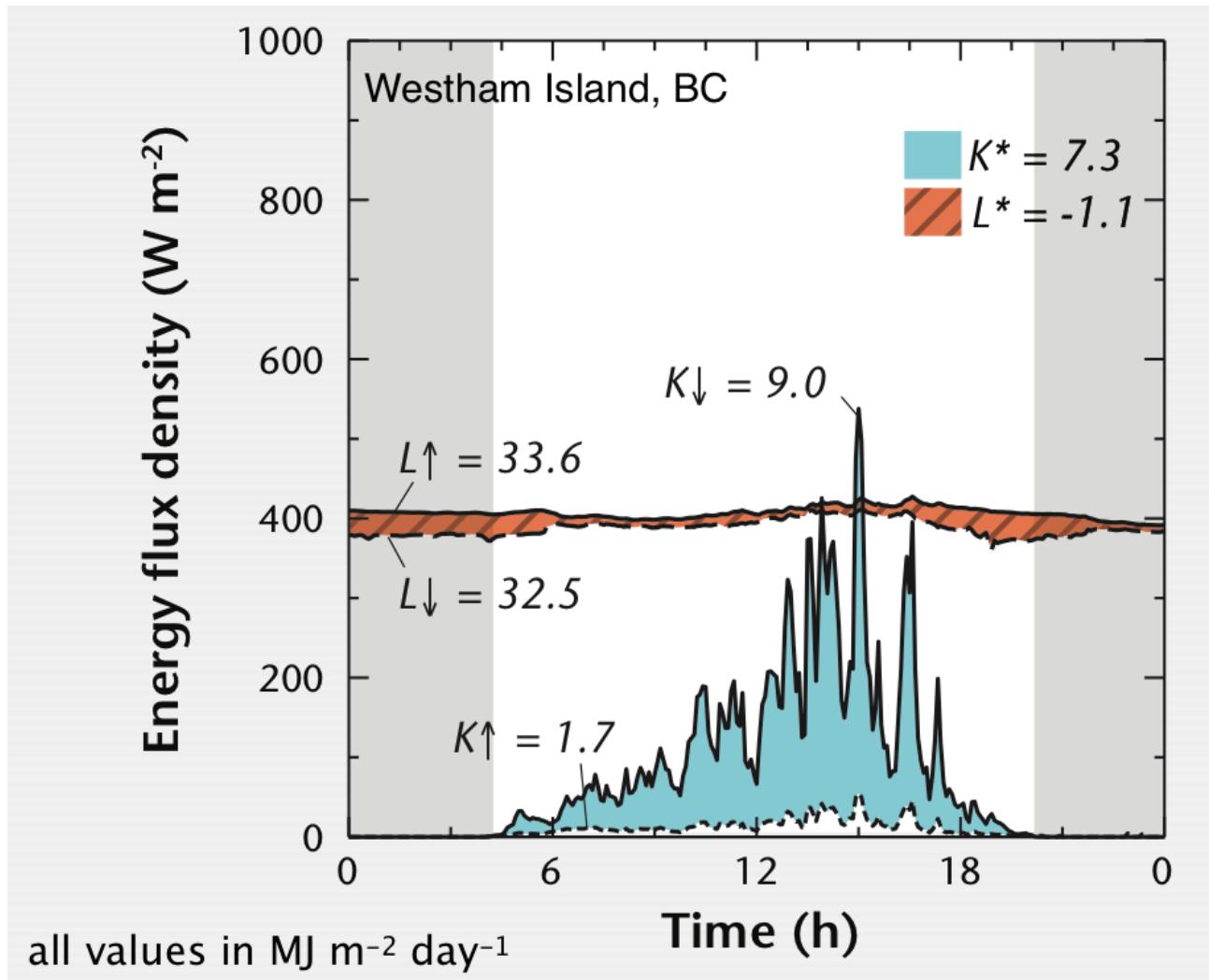


## Class activity - Why is $L_{\text{up}} > L_{\text{down}}$ ?



Images from Katkam in Vancouver

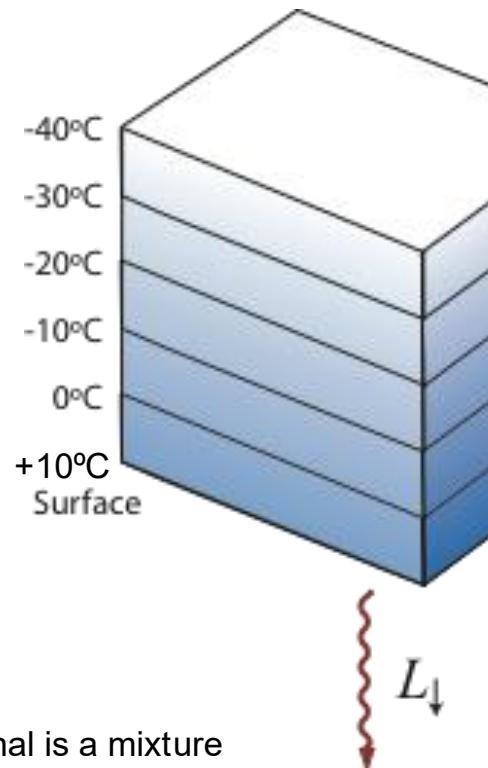
## Diurnal course of radiation - overcast / broken sky day



Images from Katkam in Vancouver

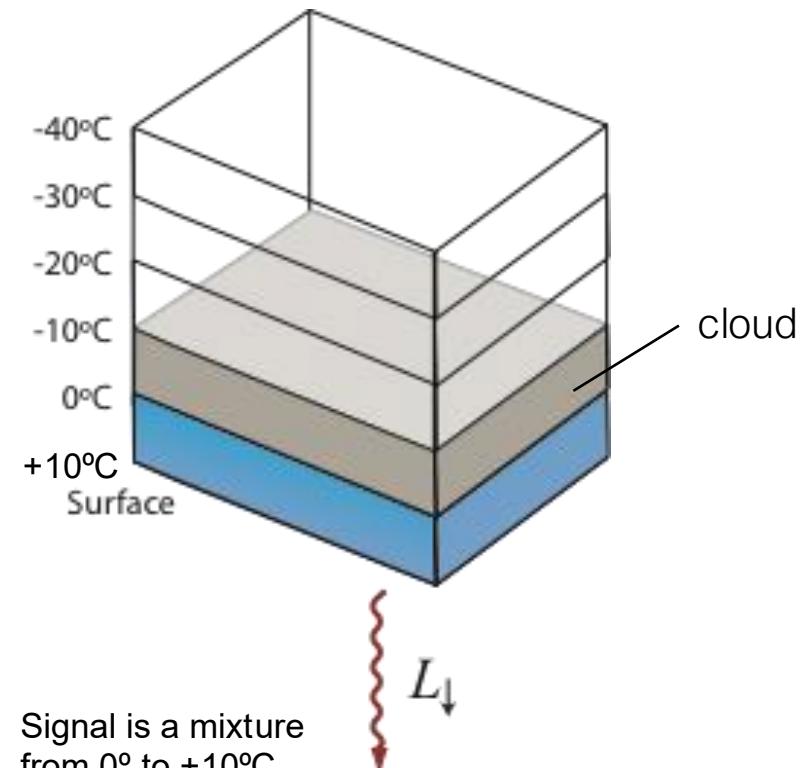
# Effect of clouds on longwave irradiance

With a clear-sky,  $L_{\downarrow}$  originates from **all layers** of the atmosphere, because the atmosphere is **partly transparent** ('atmospheric window' open)



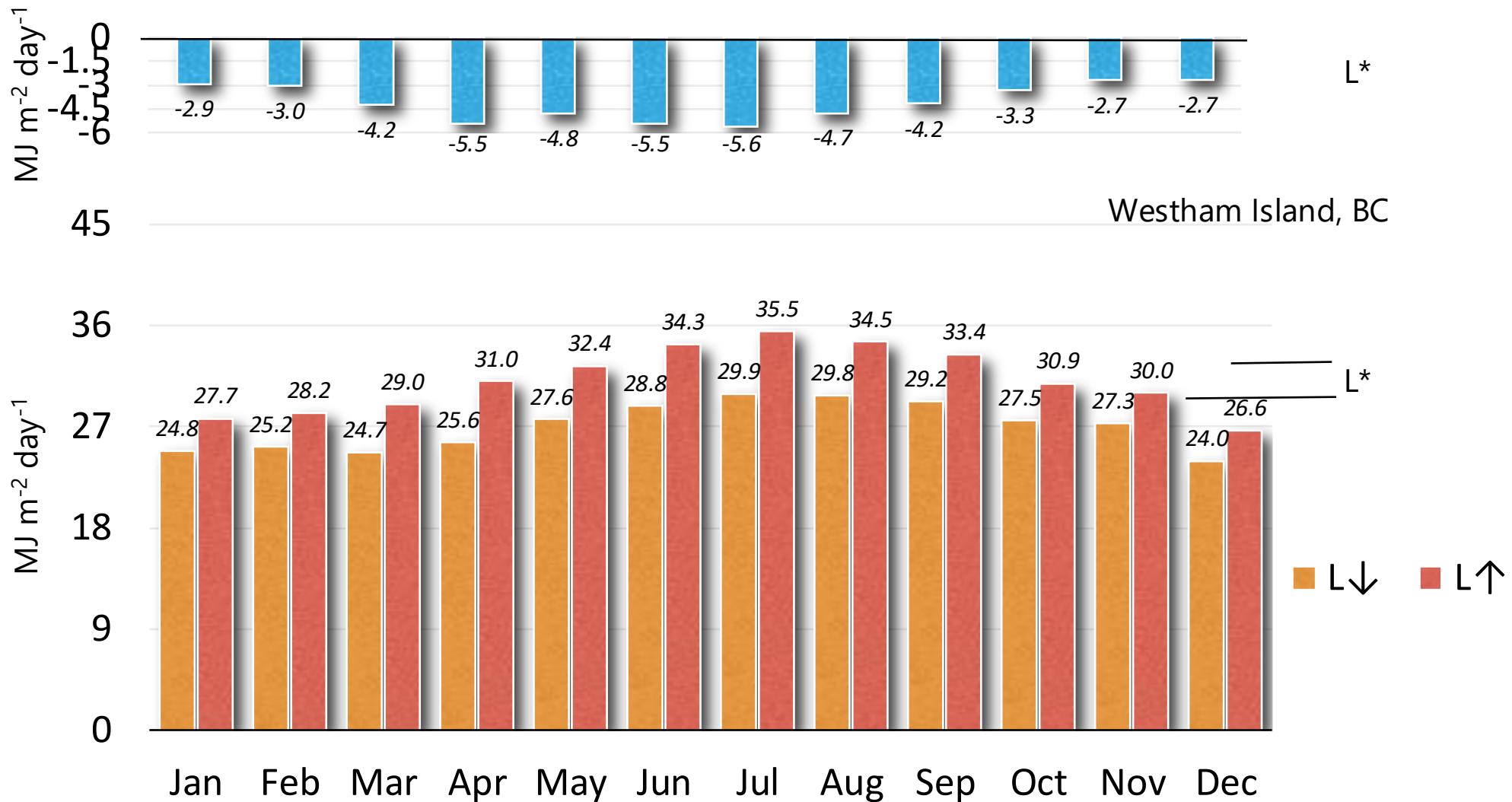
Signal is a mixture  
from -40°C to +10°C

With clouds  $L_{\downarrow}$  originates from **the cloud base** and the atmosphere below the cloud, because the cloud is **opaque to long wave** ('atmospheric window' closed)



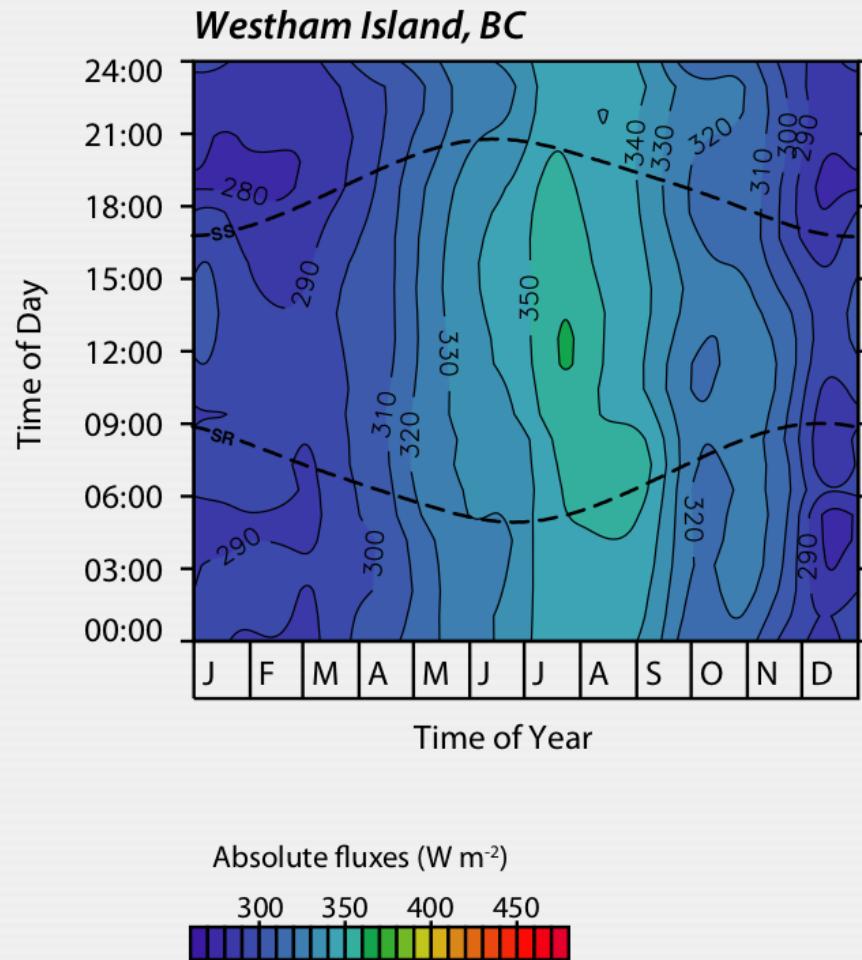
Signal is a mixture  
from 0° to +10°C

## Monthly totals of $L\downarrow$ , $L\uparrow$ and $L^*$ in Vancouver

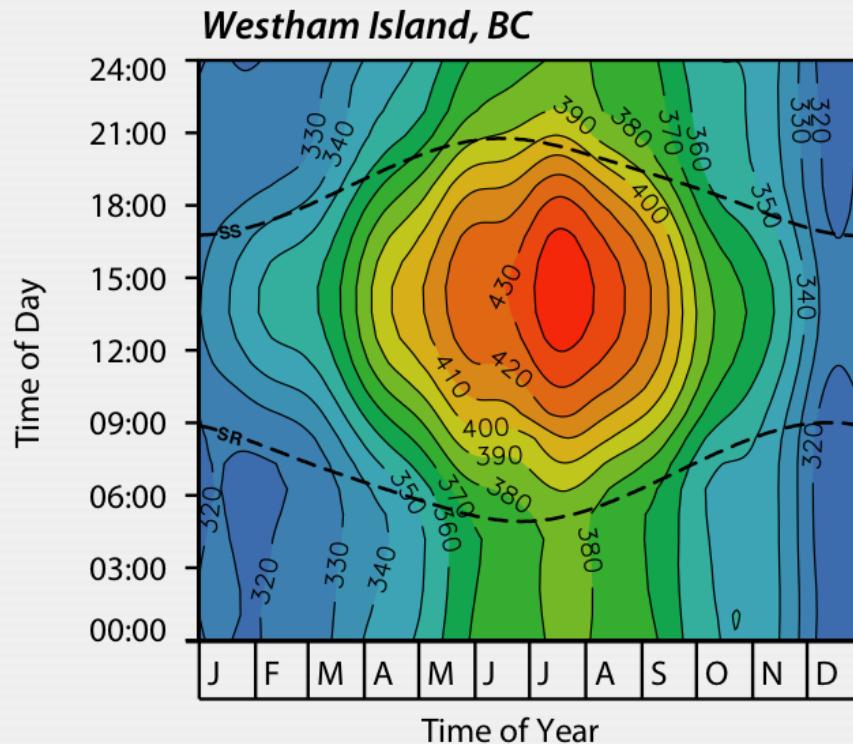


# 'Fingerprint' of $L\downarrow$ and $L\uparrow$ measured in Vancouver

(a)  $L\downarrow$



(b)  $L\uparrow$



## Take home points

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- Radiation laws apply the same way to the longwave part of the spectrum - but **Kirchhoff's law** and the concept of **emissivity** become relevant.
- The net-long wave radiation is driven by the difference in apparent sky and surface temperatures and hence **clouds and thermal surface properties** are controlling radiative exchange in the long-wave.