

# ICF IC Calibration Report (v1)

20250909 BLIZZARD NORTH: Anion 44 & Cation 38

Generated 2025-11-26 16:08:00 by MHarris (HUTL21335)

## Contents

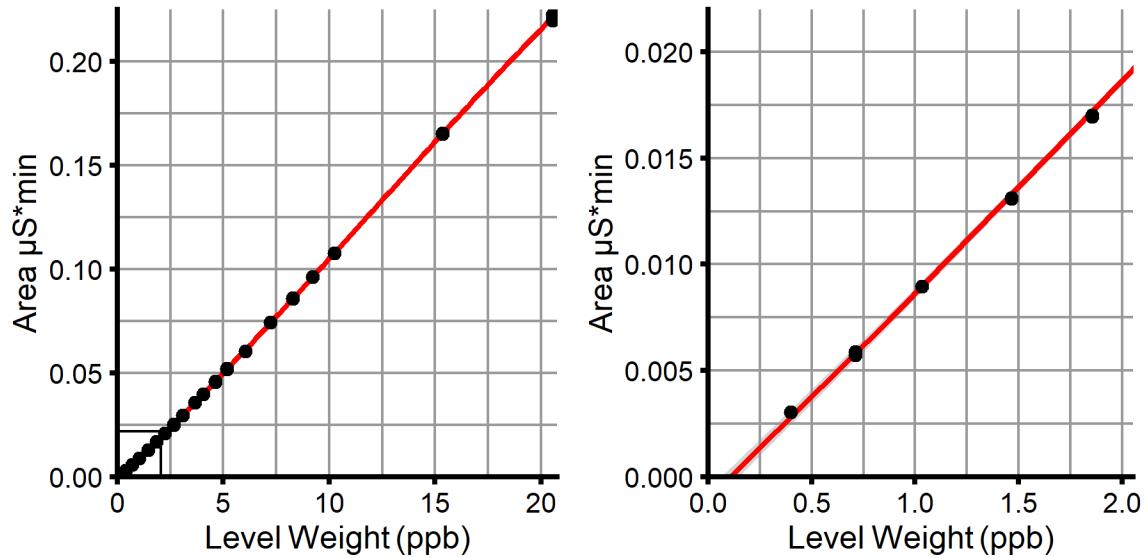
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Cations . . . . .	15

This is an automatically generated report for the following calibration sequence:

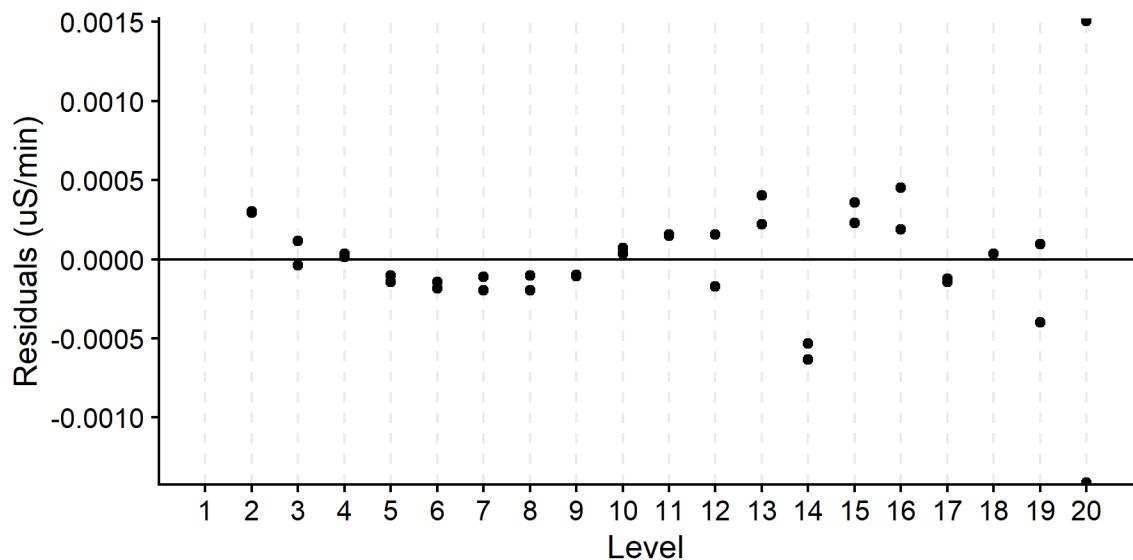
20250909\_BLIZZARD\_NORTH\_Calibration\_Anion\_44\_Cation\_38.xls

## Anions

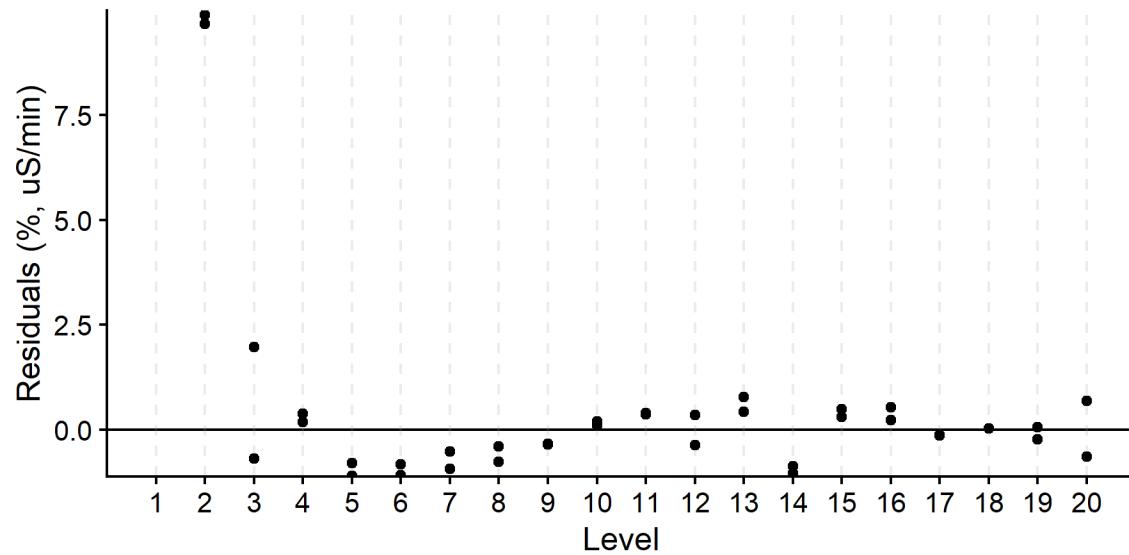
Fluoride, valid n = 38, Cubic, WithOffset  
BLIZZARD\_NORTH, Anion 44, 09/09/2025  
 $y = -4.179E-06*x^3 + 1.47E-04*x^2 + 9.571E-03*x - 1.106E-03$   
 $R^2 = 0.99995$



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[1] 2
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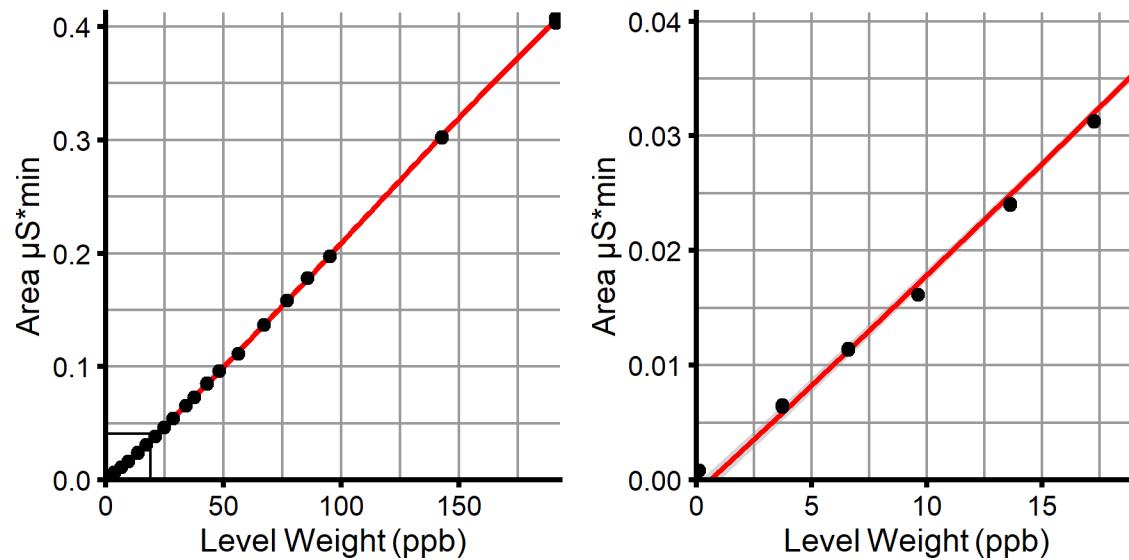
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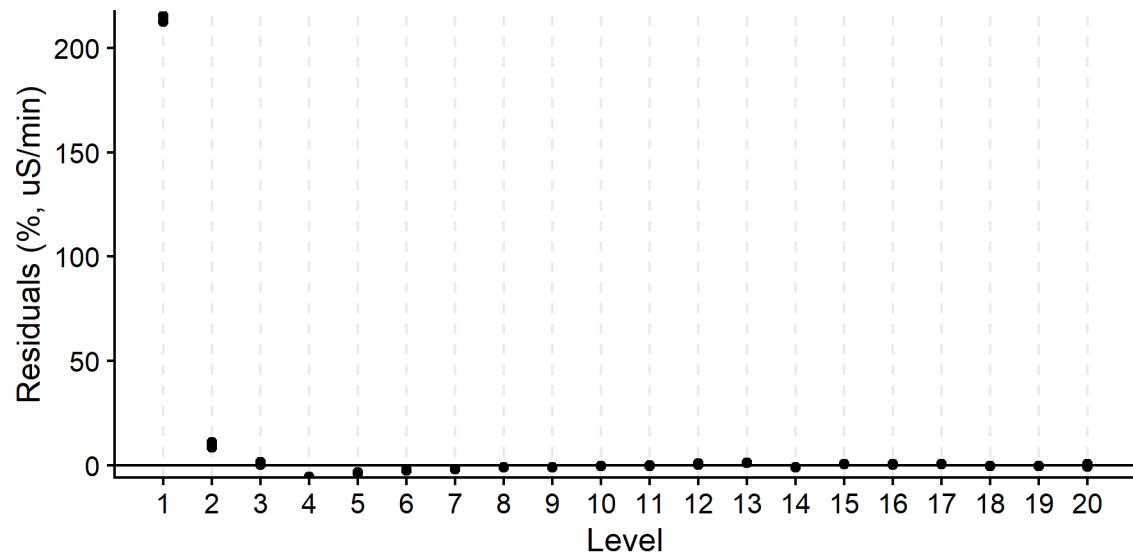
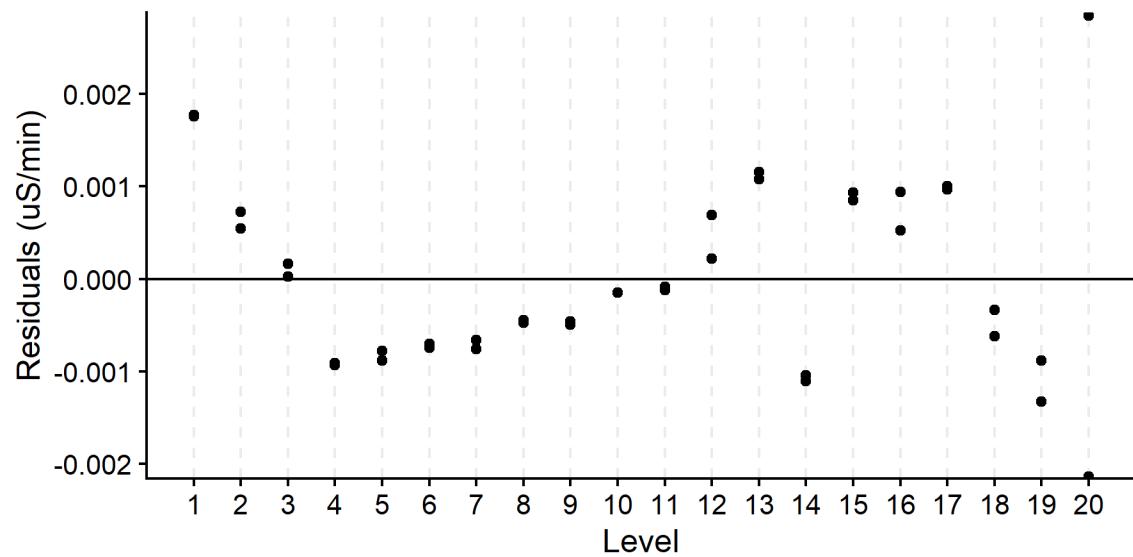


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MSA, valid n = 40, Cubic, WithOffset  
 BLIZZARD\_NORTH, Anion 44, 09/09/2025  
 $y = -1.056E-08*x^3 + 3.396E-06*x^2 + 1.866E-03*x - 1.139E-03$   
 $R^2 = 0.99991$

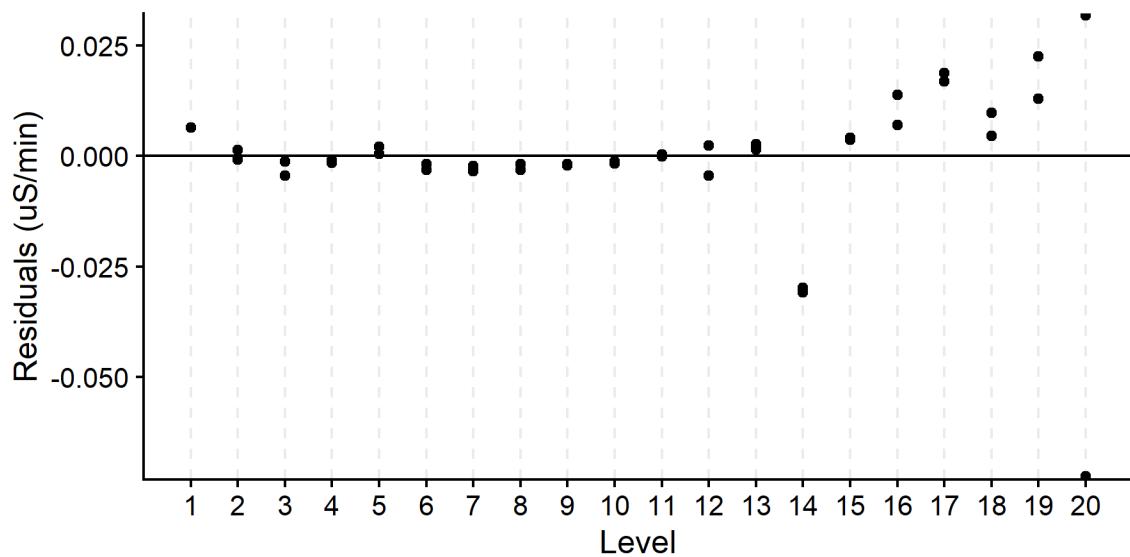
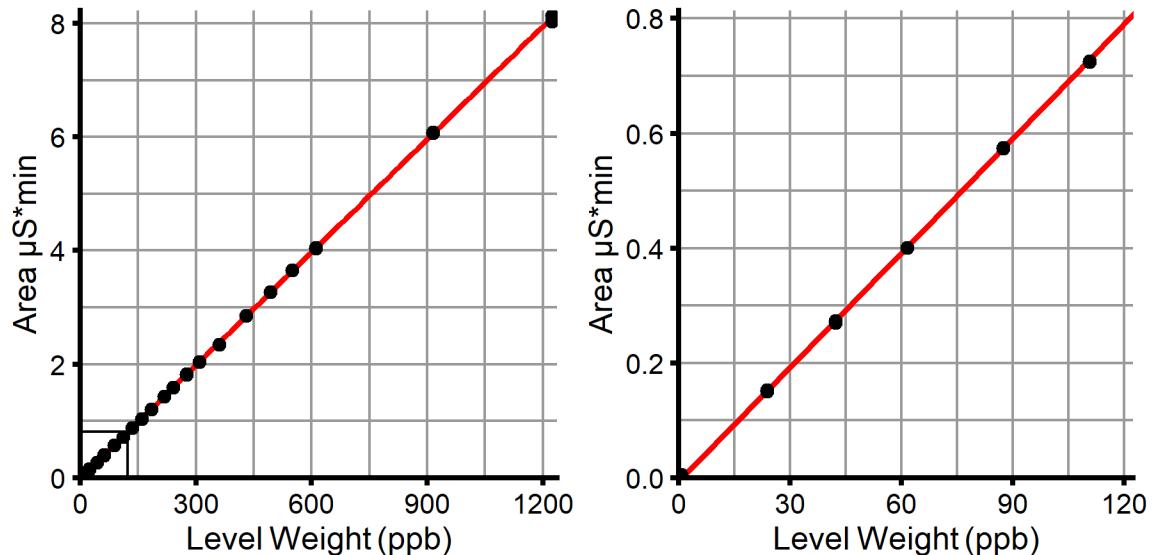


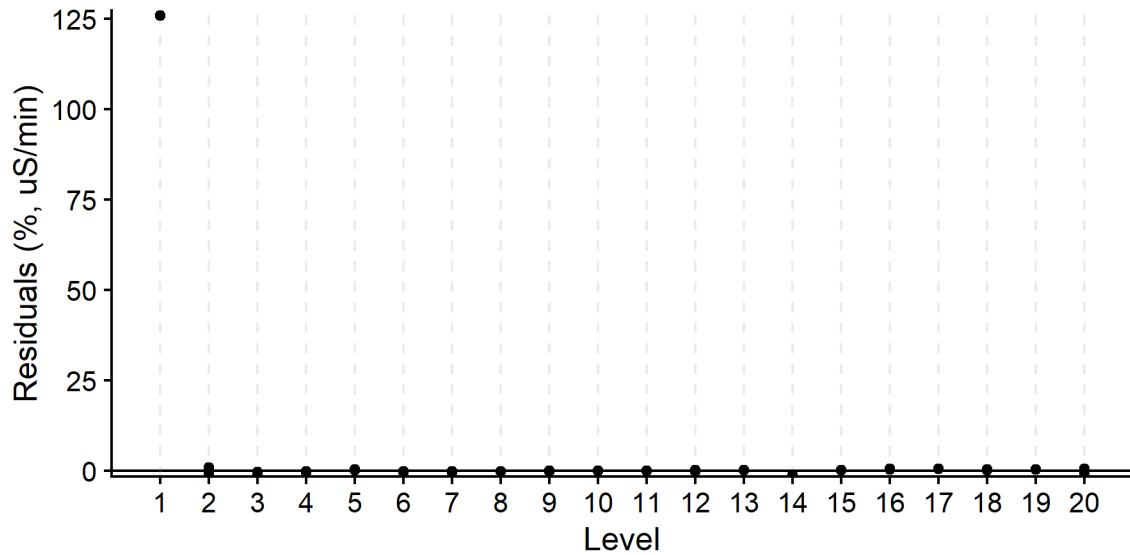


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Chloride, valid n = 40, Lin, WithOffset  
BLIZZARD\_NORTH, Anion 44, 09/09/2025  
 $y = 6.627\text{E}-03*x - 5.88\text{E}-03$   
 $R^2 = 0.99994$

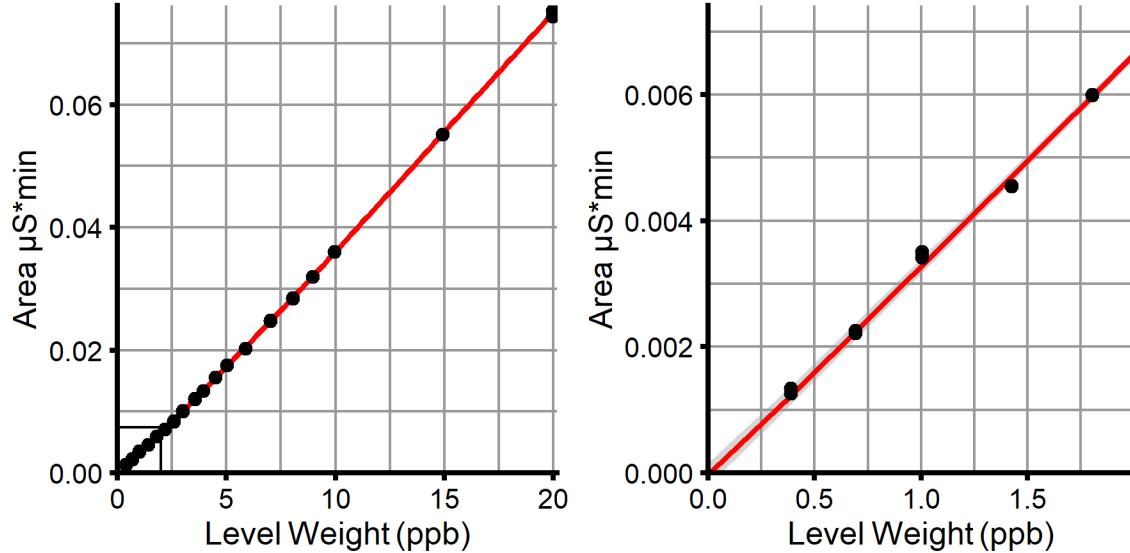




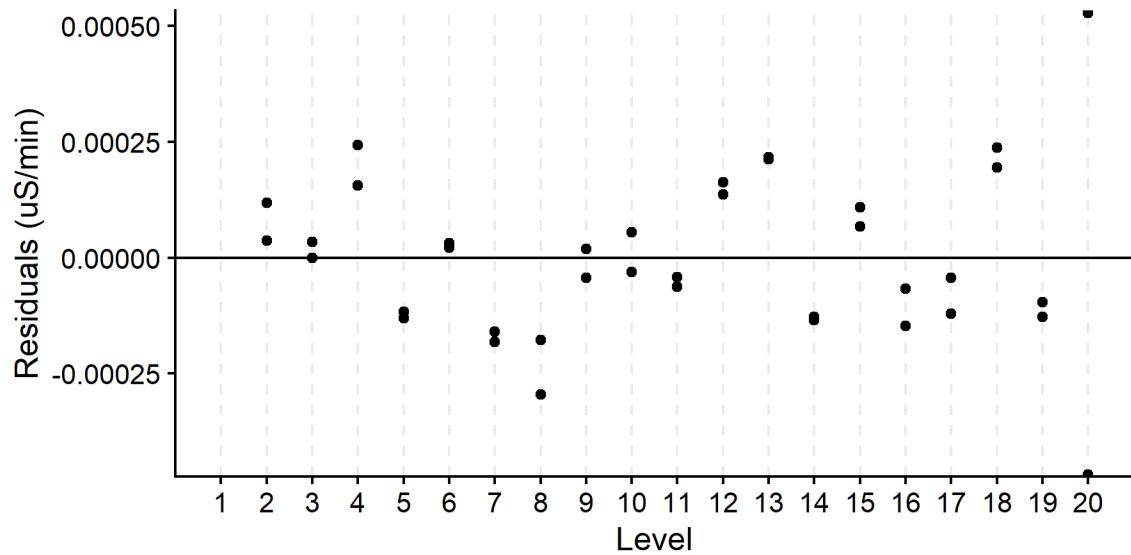
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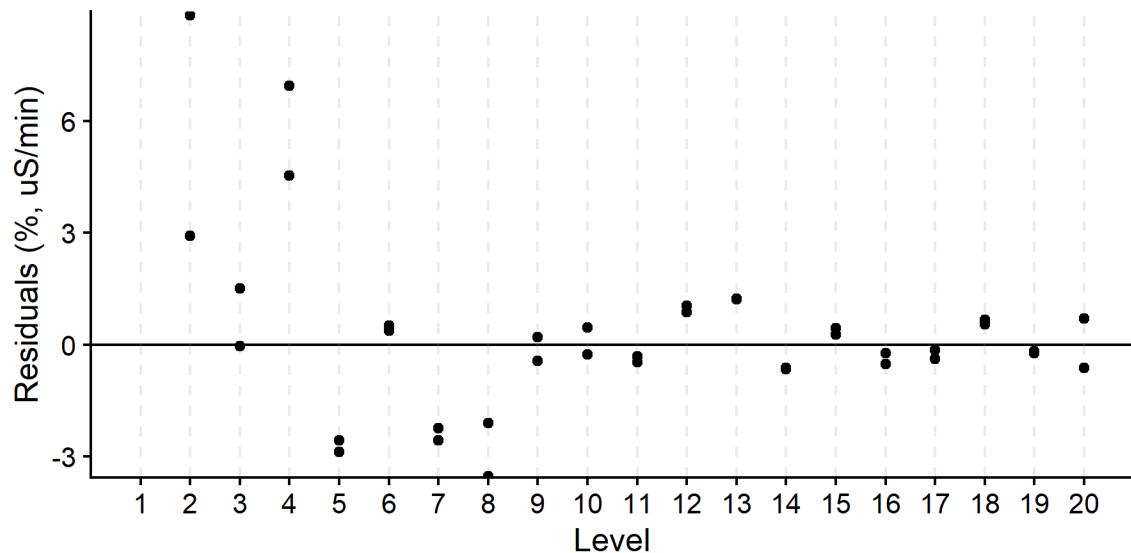
Nitrite, valid n = 38, Cubic, WithOffset  
 BLIZZARD\_NORTH, Anion 44, 09/09/2025  
 $y = -1.067E-06*x^3 + 4.633E-05*x^2 + 3.251E-03*x - 4.463E-05$   
 $R^2 = 0.99991$



[1] 1  
[1] 2



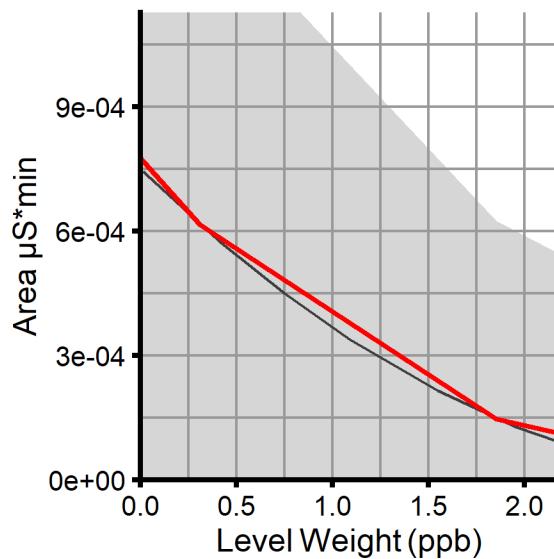
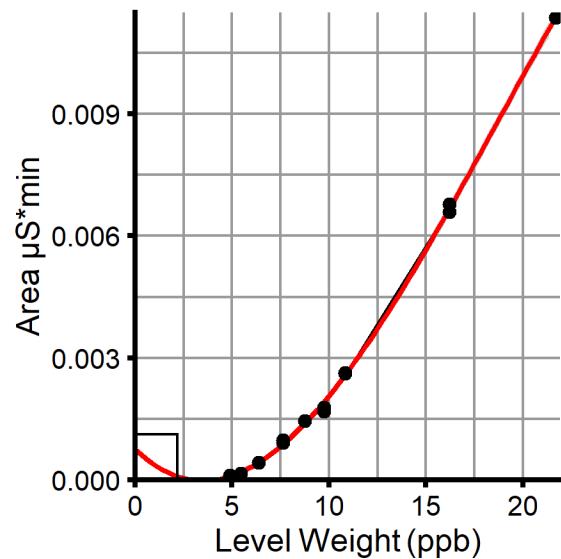
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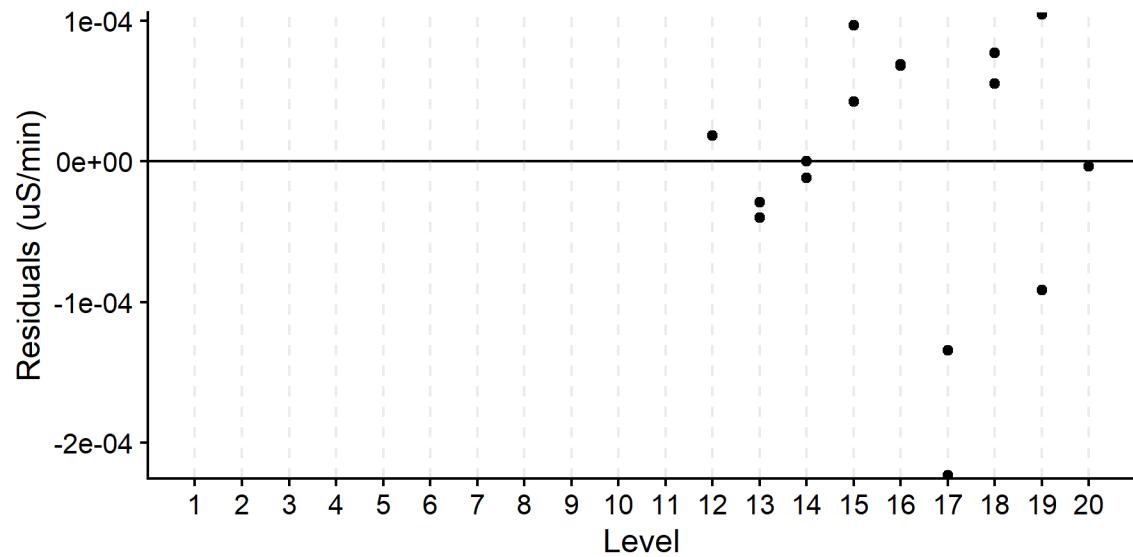
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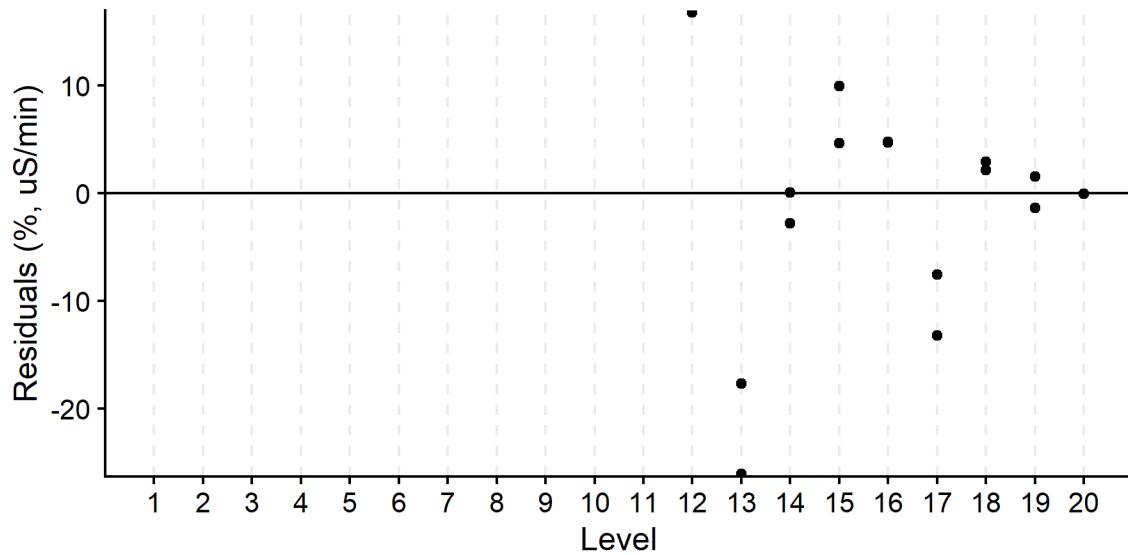
Bromide, valid n = 16, Cubic, WithOffset  
BLIZZARD\_NORTH, Anion 44, 09/09/2025  
 $y = -1.273E-06x^3 + 7.098E-05x^2 - 4.513E-04x + 7.485E-04$   
 $R^2 = 0.99919$



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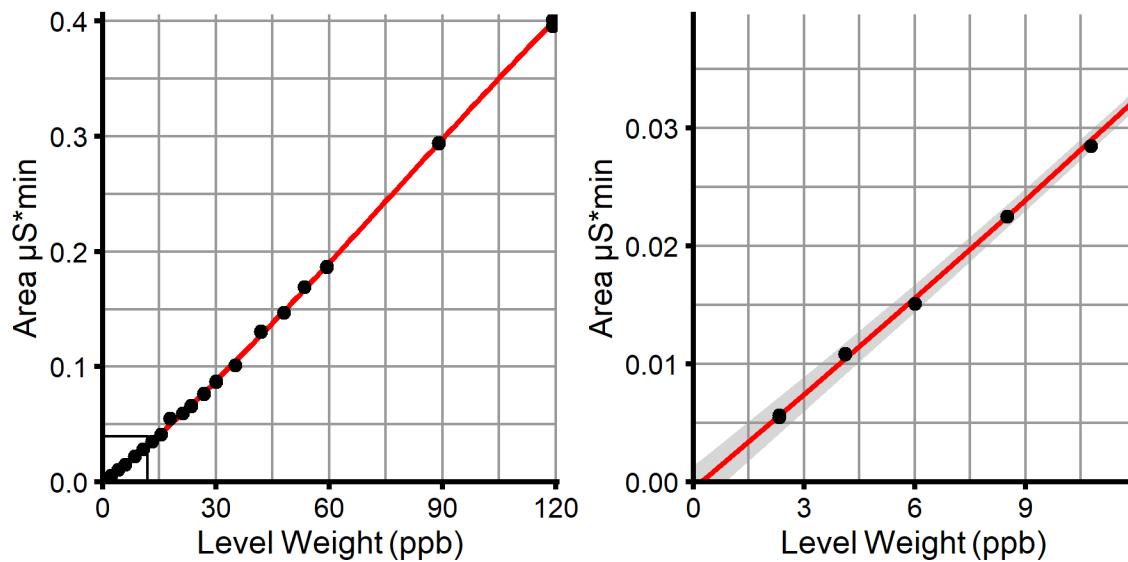
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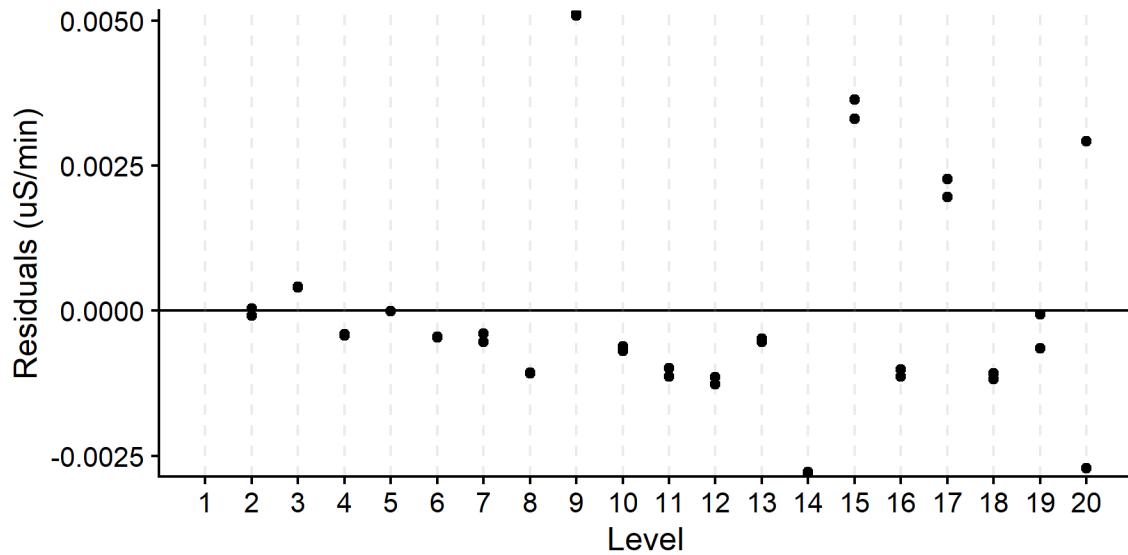
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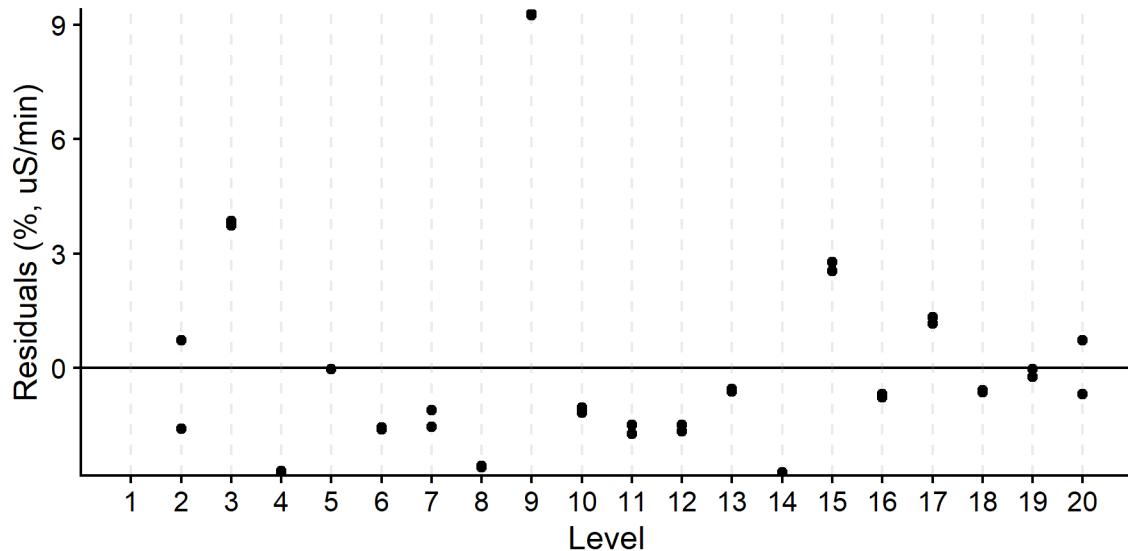
Nitrate, valid n = 38, Cubic, WithOffset  
 BLIZZARD\_NORTH, Anion 44, 09/09/2025  
 $y = -5.546E-08*x^3 + 1.288E-05*x^2 + 2.602E-03*x - 5.049E-04$   
 $R^2 = 0.99965$



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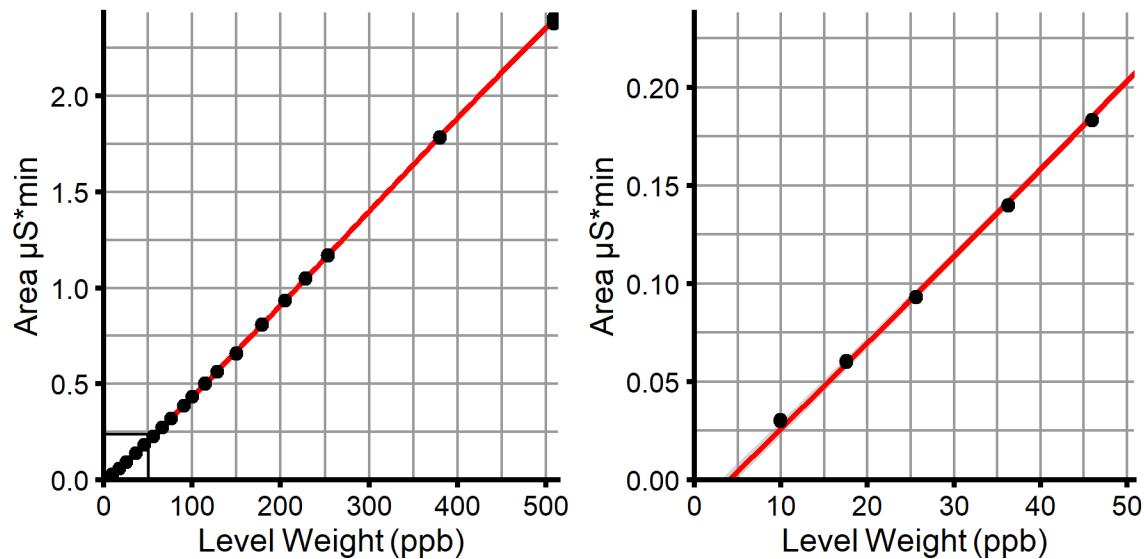
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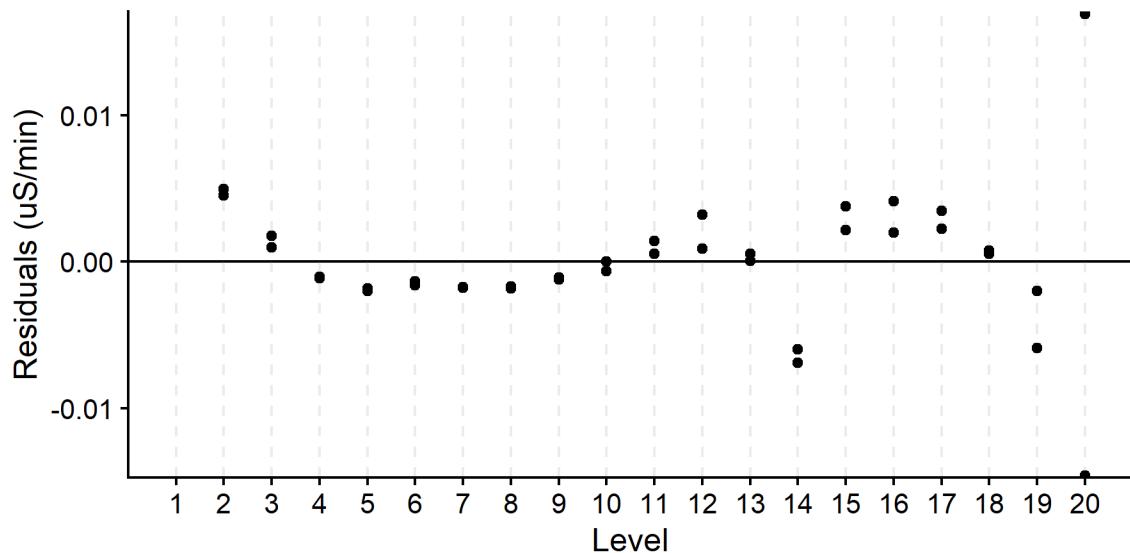
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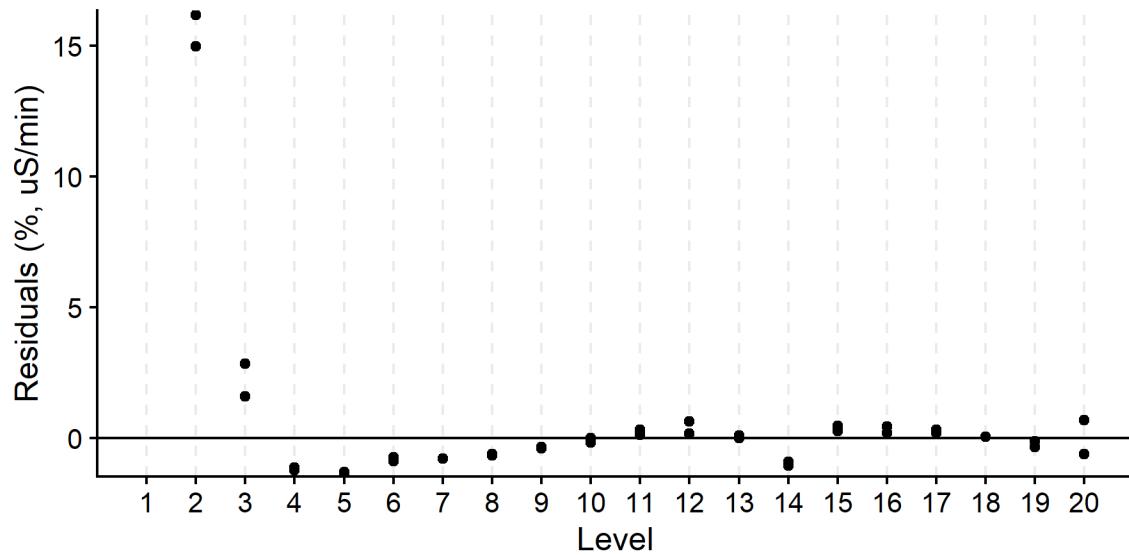
Sulphate, valid n = 38, Cubic, WithOffset  
BLIZZARD\_NORTH, Anion 44, 09/09/2025  
 $y = -2.511E-09*x^3 + 2.131E-06*x^2 + 4.309E-03*x - 1.715E-02$   
 $R^2 = 0.99994$



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[1] 1
[1] 2
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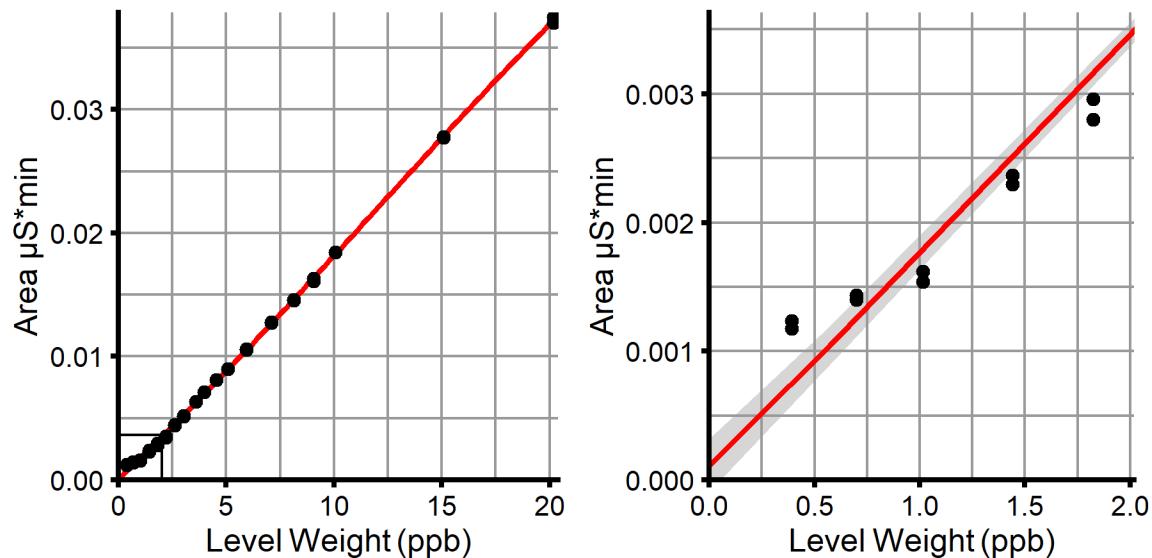
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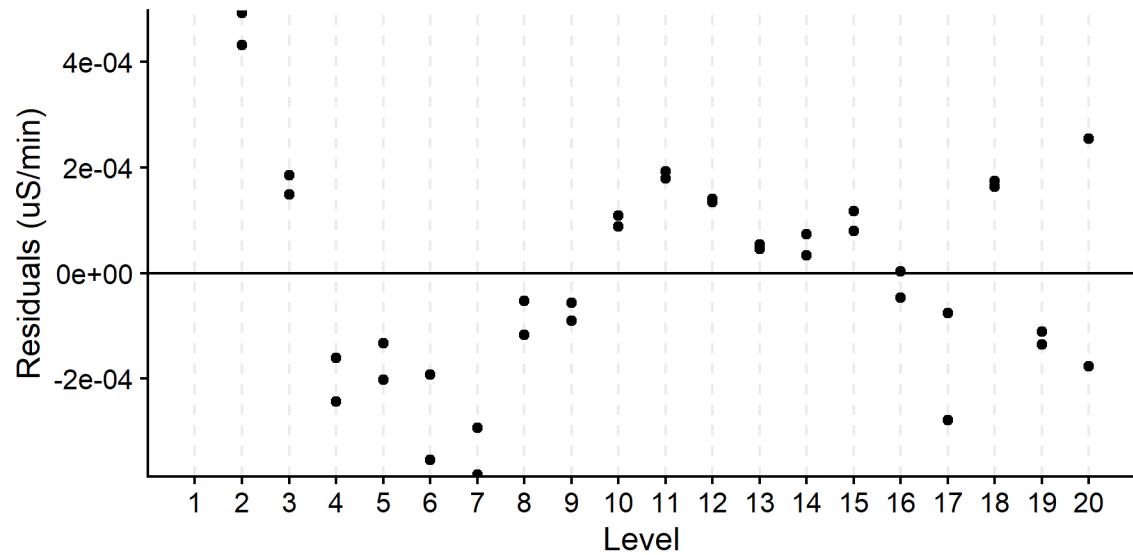
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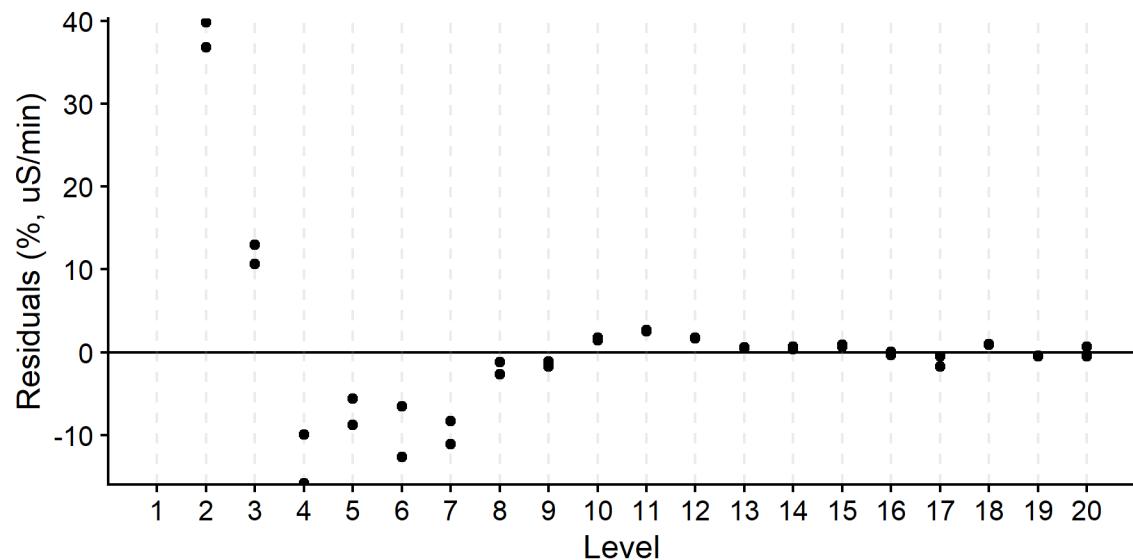
Phosphate, valid n = 38, Cubic, WithOffset  
 BLIZZARD\_NORTH, Anion 44, 09/09/2025  
 $y = -6.948E-07*x^3 + 2.422E-05*x^2 + 1.633E-03*x + 9.845E-05$   
 $R^2 = 0.99955$



[1] 1  
[1] 2

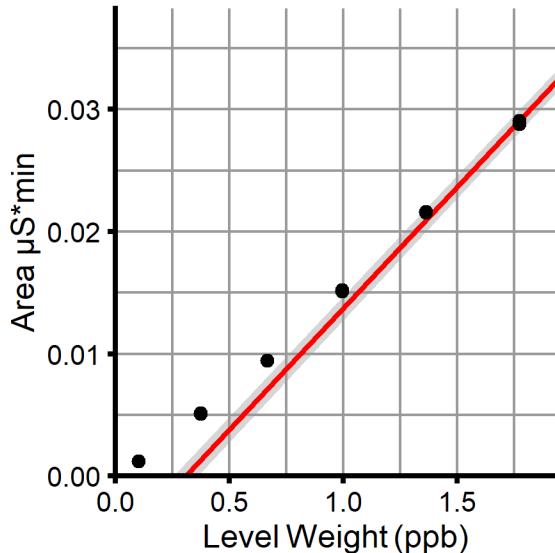
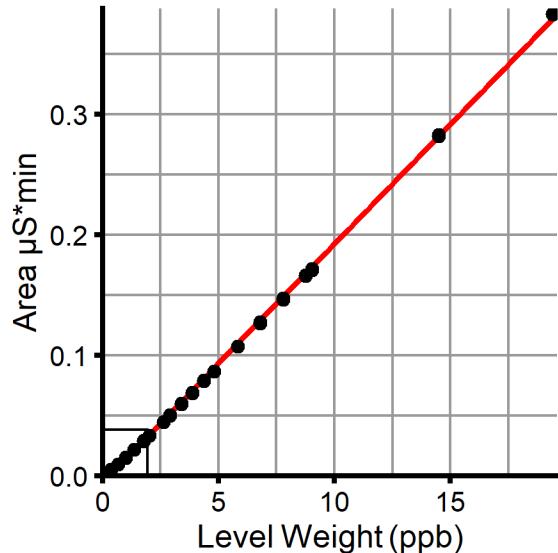


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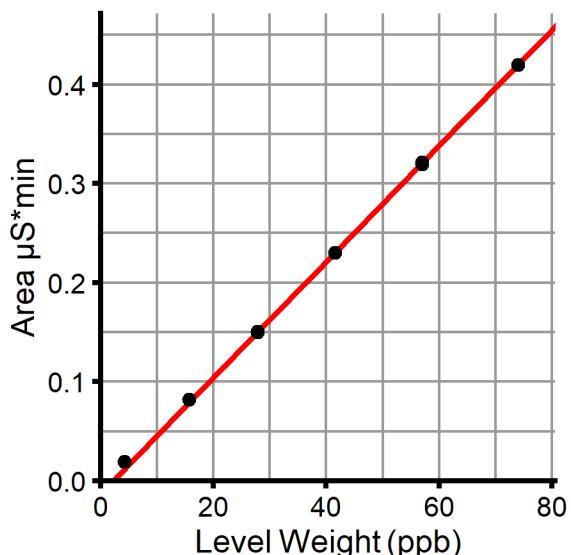
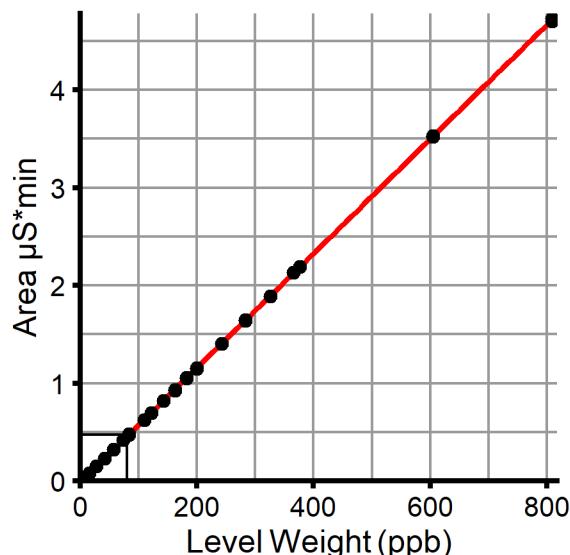


## Cations

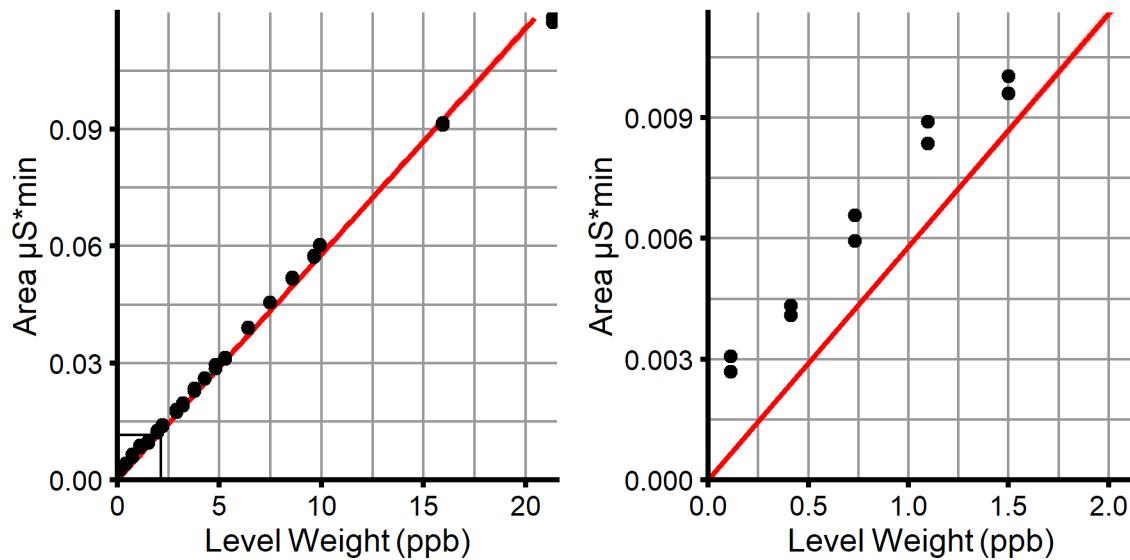
Lithium, valid n = 40, Lin, WithOffset  
BLIZZARD\_NORTH, Cation 38, 09/09/2025  
 $y = 1.985E-02*x - 6.14E-03$   
 $R^2 = 0.99943$



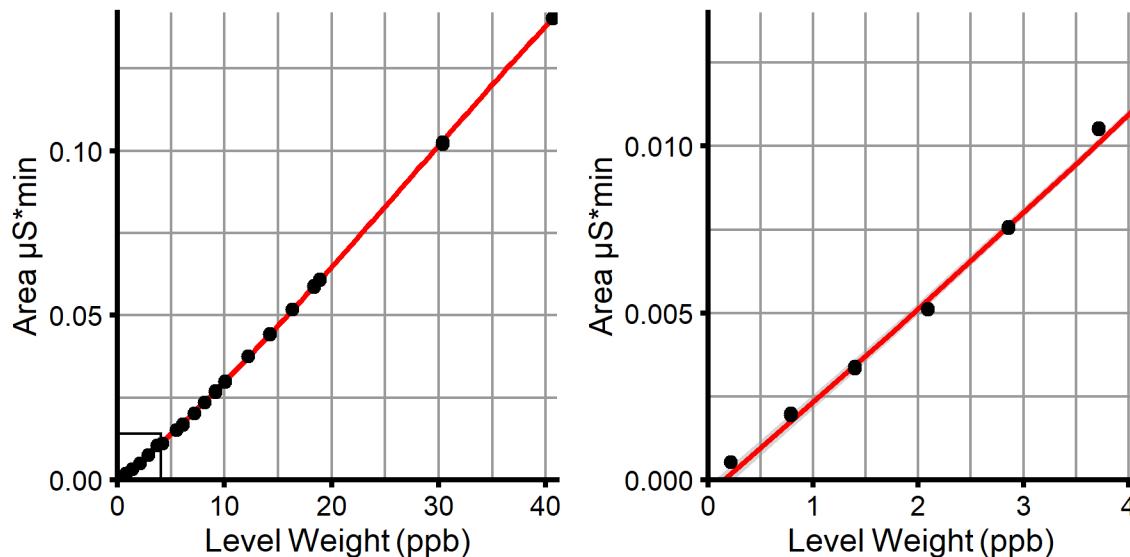
Sodium, valid n = 40, Lin, WithOffset  
BLIZZARD\_NORTH, Cation 38, 09/09/2025  
 $y = 5.848E-03*x - 1.275E-02$   
 $R^2 = 0.99998$



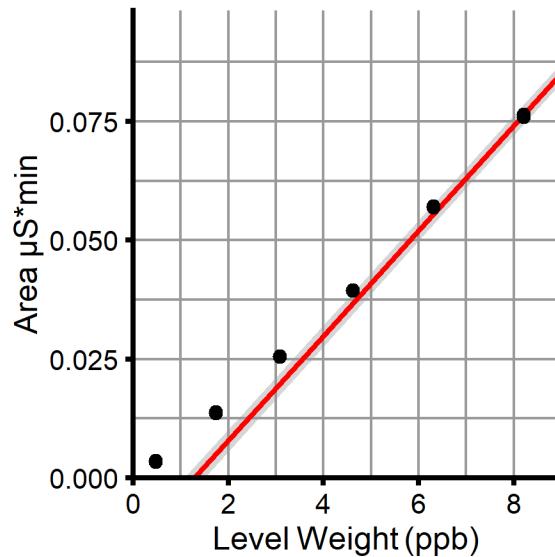
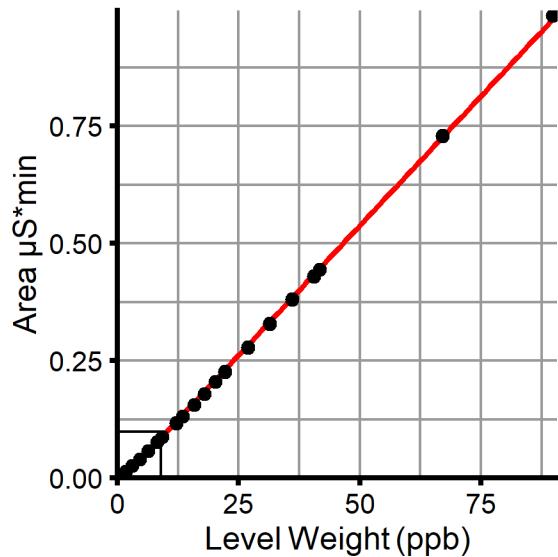
Ammonium, valid n = 40, Lin  
 BLIZZARD\_NORTH, Cation 38, 09/09/2025  
 $y = 5.797E-03*x$   
 $R^2 = 0.99782$



Potassium, valid n = 40, Cubic, WithOffset  
 BLIZZARD\_NORTH, Cation 38, 09/09/2025  
 $y = -4.388E-07*x^3 + 3.653E-05*x^2 + 2.7E-03*x - 4.216E-04$   
 $R^2 = 0.99992$



Magnesium, valid n = 40, Lin, WithOffset  
 BLIZZARD\_NORTH, Cation 38, 09/09/2025  
 $y = 1.104E-02*x - 1.425E-02$   
 $R^2 = 0.99955$




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Calcium, valid n = 40, Cubic, WithOffset  
 BLIZZARD\_NORTH, Cation 38, 09/09/2025  
 $y = -2.654\text{E-}07*x^3 + 3.485\text{E-}05*x^2 + 5.276\text{E-}03*x - 4.839\text{E-}04$   
 $R^2 = 0.99988$

