熱力学 4回目

[1] (1) 0-1;定客変化的, dv=0, v,=v。

熱力学第一法則的, dq = du+pdv

de=du=mcvdT

Q=mCy(T,-To)

TI = R + To

理想気体の状態方程式的, PN=MRT→ N=MRT/p=Const M,

T/p = const

Ti/p, = To/po

in Pr = TI Po = (Q +1) Po

(2) dg= Tds +1,

= Sor = m Cv In (Ti) = m Cv In (a mCv To)

(3) 1-2;断熟变化的, d8=0

Pyrk = const fy,

PoMk = PiNK

- V2 = (P1) K V0 = V0 (R +1) K

PN = mRT -> P= mRT/v, PUK = const F),

TMK-1 = const

TI NOK-1 = T2/12/K-1

 $\frac{1}{\sqrt{2}} = \left(\frac{\sqrt{2}}{\sqrt{2}}\right)^{K-1} T_1 = \left(\frac{\partial}{\partial x_1} + 1\right)^{-\frac{1}{2}} \left(\frac{\partial}{\partial x_2} + 1\right) T_0$

= To (Q +1) K

(4) dq = Tds = 0 =1,

: 512 = 0

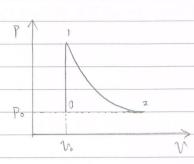
(5) de = du+pdn = 0 +),

W12 = - m(v(T2-T1)

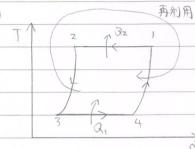
= mCv { mCv + To - To (Q +1) k}

= mCnTo { Q (Q +1) R}





2 (1)



(2) 2 → 3 ; 等容変化 もり, dv = 0

$$= Q_{23} = U_3 - U_2 = m(v(T_3 - T_2))$$

(3) 3-74;等温爱化却, dT=0

 $= \frac{4}{3} P dv = P3V3 = \frac{4}{3} P dv = P3V3 = \frac{V4}{V3} = mRT3 ln(\frac{V_4}{V_3}) = mRT3 ln(\frac{V_4}{V_4}) = mRT3 ln($

de=du+por +1, de=por

$$Q_{12} = \int_{1}^{2} Pohv = P_{1}V_{1} h_{1}\left(\frac{V_{2}}{V_{1}}\right) = -mRT_{1}\left(\frac{V_{1}}{V_{2}}\right) = -mRT_{1}h_{2}$$

$$\frac{1}{N_2} = \frac{N_1 \times P_1}{N_2 \times P_1}$$

$$T/p^{\frac{K-1}{K}} = const$$

$$\frac{1}{1} = P_1 \left(\frac{T_C}{T_1} \right)^{\frac{K}{K-1}} = P_1 \left(\frac{2}{K+1} \right)^{\frac{K}{K-1}} \left(\frac{T_C}{T_1} - \frac{2}{K+1} \right)$$

$$M(h_1 + \frac{1}{2}W_1^2) = M(h_2 + \frac{1}{2}W_2^2)$$

$$W_2 = 2(h_1 - h_2)$$

$$=$$
 $\left[2Cp\left(T_1-T_2\right)\right]$

$$= \sqrt{2 \frac{K}{K-1} R(T_1 - T_2)}$$

$$= \frac{2K}{K-1} P_1 V_1 \left\{ \left| -\frac{P_2}{P_1} \cdot \frac{V_2}{V_1} \right\} \right.$$

$$W_{2} = \begin{cases} 2K & P_{1}N_{1} \\ 1 - (N_{1})^{K-1} \end{cases} - 0$$

No.

| | | • | | |
|----|-----|--------|--------|--------|
| 臨界 | 流れで | t, P = | -PC, T | 2 - Tc |

$$TN^{k-1} = const F_{j},$$

$$T_{i}N_{i}^{k-1} = T_{c}N_{i}^{k-1}$$

$$\frac{\mathcal{N}_1}{\mathcal{N}_2} = \left(\frac{\mathsf{T}_C}{\mathsf{T}_1}\right)^{\frac{1}{\mathsf{K}_{-1}}} = \left(\frac{2}{\mathsf{K}_{+1}}\right)^{\frac{1}{\mathsf{K}_{-1}}}$$

$$P_1 \mathcal{N}_1 = P_2 \frac{\mathcal{N}_2^K}{\mathcal{N}_1^{K-1}} = P_2 \mathcal{N}_2 \left(\frac{\mathcal{N}_2}{\mathcal{N}_1}\right)^{K-1}$$
 3

31二代入

(5) 酷界压力比; Pc 17,