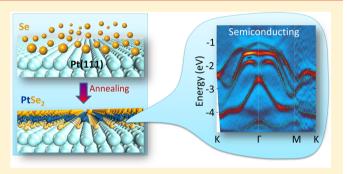


Monolayer PtSe₂, a New Semiconducting Transition-Metal-Dichalcogenide, Epitaxially Grown by Direct Selenization of Pt

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Supporting Information

ABSTRACT: Single-layer transition-metal dichalcogenides (TMDs) receive significant attention due to their intriguing physical properties for both fundamental research and potential applications in electronics, optoelectronics, spintronics, catalysis, and so on. Here, we demonstrate the epitaxial growth of high-quality single-crystal, monolayer platinum diselenide (PtSe₂), a new member of the layered TMDs family, by a single step of direct selenization of a Pt(111) substrate. A combination of atomic-resolution experimental characterizations and first-principle theoretic calculations reveals the atomic structure of the monolayer PtSe₂/Pt(111). Angle-resolved photoemission spectroscopy measurements



confirm for the first time the semiconducting electronic structure of monolayer PtSe2 (in contrast to its semimetallic bulk counterpart). The photocatalytic activity of monolayer PtSe₂ film is evaluated by a methylene-blue photodegradation experiment, demonstrating its practical application as a promising photocatalyst. Moreover, circular polarization calculations predict that monolayer PtSe2 has also potential applications in valleytronics.

KEYWORDS: two-dimensional materials, transition-metal dichalcogenides, PtSe2, epitaxial growth, monolayer, photocatalyst

Pollowing the discovery of many unique properties of monolayer graphene and graphene analogues, 1-5 attention has been paid to other two-dimensional crystalline materials with novel properties like sizable bandgap. Among them, layered TMDs with the general formula MX2, where M represents a transition metal from groups 4-10 and X is a chalcogen (S, Se, or Te), are particularly desirable not only for fundamental research⁶⁻¹⁶ but also for technological development.¹⁷⁻²² Recent investigations of MX₂ have resulted in discoveries of dramatically different electronic structures at the monolayer limit compared to the bulk materials due to quantum confinement effects. For example, while pushing from bulk to monolayer, MoS₂ and MoSe₂ show an indirect-to-

direct bandgap transition. 23-25 With these exciting findings, experimental research efforts so far have been mainly focused on semiconducting prototypical MX2 with group VIB transition metals (M = Mo, W). In the large family of layered TMDs, many other promising single-layer TMDs and related quantumdefined properties remain to be explored experimentally. However, the synthesis and fabrication of high-quality,

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monolayer TMDs materials other than group VIB ones remain a challenge.

In this Letter, we report epitaxial growth of monolayer PtSe₂—a heretofore-unexplored member of the single-layer TMDs family—on a Pt substrate by direct "selenization", an analog of direct oxidation. In contrast to conventional fabrication methods of MX2 by exfoliation or chemical vapor deposition, the present route toward a monolayer dichalcogenide is very straightforward: only one element, Se, is deposited on a Pt substrate. The atomic structure of epitaxial PtSe₂ film has been determined by a combination of low-energy electron diffraction (LEED), scanning tunneling microscopy (STM), scanning transmission electron microscopy (STEM) measurements, and density functional theory (DFT) calculations. Angle-resolved photoemission spectroscopy (ARPES) measurements together with band structure calculations directly reveal monolayer PtSe2 is a semiconductor, in contrast to a semimetal in its bulk form. In view of the simplicity of the production method and the advantages of the semiconducting nature, PtSe₂ film has potential for use in practical applications. Here, a photodegradation experiment based on as-prepared monolayer PtSe₂ film was performed, demonstrating its practical application as a visible light-driven photocatalyst. Moreover, circular polarization calculations indicate that monolayer PtSe₂ is a good candidate for valleytronics.

Monolayer $PtSe_2$ films were grown by a straightforward precursor-free method, that is, through a single step of direct selenization of a Pt(111) substrate (see schematic in Figure 1a).

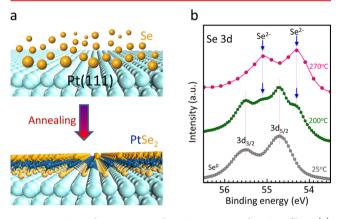


Figure 1. Growth process and XPS spectra of PtSe $_2$ film. (a) Schematic of the fabrication of PtSe $_2$ thin films by a single step of direct selenization of a Pt(111) substrate. PtSe $_2$ monolayer consists of one sublayer of Pt atoms sandwiched between two Se sublayers. The Pt spheres with different colors and sizes are used just to differentiate the Pt atoms in the Pt(111) substrate and in the PtSe $_2$ sublayer. (b) XPS measurements for the binding energies of Se during PtSe $_2$ growth demonstrating the formation of PtSe $_2$ at 270 °C. The blue arrows indicate the peak positions (55.19 and 54.39 eV) corresponding to the binding energy of Se $_2$. The Se $_2$ peaks (at 55.68 and 54.80 eV) are dominant at 25 °C, whereas at 200 °C the peaks in the curve indicate the coexistence of Se $_2$ and Se $_2$.

Selenium atoms were deposited on a Pt(111) substrate, and then the sample was annealed to $\sim\!200\,^{\circ}\text{C}$ to obtain epitaxial PtSe₂ films. The growth of PtSe₂ thin films was monitored by in situ X-ray photoemission spectroscopy (XPS). Figure 1b shows the XPS spectra of the Se 3d core level during PtSe₂ growth. When Se-deposited Pt(111) substrate is annealed to 200 $^{\circ}\text{C}$, two new peaks appear at binding energies of 55.19 and 54.39 eV (labeled by the blue arrows), which can be explained by a

change in the chemical state of Se from Se 0 to Se 2 -, corresponding to the selenization process of the sample. Further annealing of the sample to 270 °C results in the disappearance of Se 0 peaks (at 55.68 and 54.80 eV) and the dominance of Se 2 - peaks, indicating full crystallization and complete formation of PtSe $_2$ films.

To obtain the structural information on the as-grown epitaxial films, the samples were characterized by LEED. Figure 2a shows a LEED pattern. Hexagonal diffraction spots from

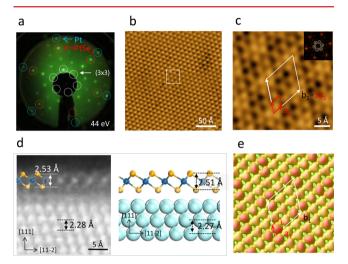


Figure 2. LEED pattern and atomic configuration of PtSe₂ film. (a) LEED pattern of a PtSe₂ film formed on the Pt(111) substrate. The blue, red, white circles indicate the diffraction spots from the Pt(111) lattice, PtSe₂ thin film and (3 × 3) superstructure with respect to PtSe₂, respectively. (b) Large-scale STM image (U = -1.9 V and I =0.12 nA) shows the moiré pattern of PtSe2 thin film on Pt(111). The white rectangle marks the size of the close-up image in (c). (c) Atomic-resolution STM image (U = -1.0 V and I = 0.12 nA) of singlelayer PtSe2 showing the hexagonal lattice of Se atoms in the topmost sublayer of the PtSe₂ sandwich-type structure. A (3 × 3) moiré superstructure is visible. The red and white rhombi denote the unit cell of the PtSe₂ lattice and (3×3) superlattice, respectively. The inset displays the FFT pattern corresponding to PtSe2 and the superstructure. (d) Side view of the atomic structure of the PtSe₂ monolayer on Pt(111). Left: atomic-resolution STEM Z-contrast image of the PtSe₂/Pt(111) interface along the [110] zone axis. A PtSe₂ single layer over the Pt(111) substrate is resolved at atomic scale with a model diagram overlaid for clarity. Right: the relaxed model. The blue and orange spheres represent Pt atoms and Se atoms, respectively. (e) Simulated STM image (U = -1.0 V) based on the calculated atomic structure in (d) is consistent with the experimental observation in (c).

PtSe $_2$ (red circles) are observed to have the same orientation as those from the Pt(111) substrate (blue circles), suggesting a rotational-domain-free growth. A (3 × 3) diffraction pattern of the epitaxial PtSe $_2$ film is clearly identified, which corresponds to a well-defined moiré superstructure arising from the lattice mismatch between the PtSe $_2$ film and Pt(111) substrate. Furthermore, identical LEED patterns were observed on the entire sample surface (4 mm × 4 mm in size), indicating the growth of a large-area, homogeneous and high-quality film.

To investigate the atomic structure of the $PtSe_2$ film, we performed STM studies. Figure 2b is a large-scale STM image with a well-ordered moiré pattern of $PtSe_2$ thin film on Pt(111). The periodicity of this moire pattern is about 11.1 Å, four times the lattice constant of Pt(111). Figure 2c shows an atomic-resolution image of the area indicated by the white

square in Figure 2b, revealing hexagonally arranged protrusions with an average lattice constant of $a_1 = 3.7$ Å, which agrees perfectly with the interatomic spacing of Se atoms in the (0001) basal plane of bulk PtSe₂. Therefore, we interpret the hexagonal protrusions in Figure 2c to be the Se atoms in the topmost Se plane of a PtSe₂ film. A regular (3 × 3) moiré superstructure with respect to the PtSe2 lattice is then established, with a periodicity of $b_1 = 3a_1 \cong 11.1$ Å (labeled by the white rhombus). The orientation of the moiré pattern is aligned with that of the PtSe2 lattice. This is in agreement with the LEED observation (Figure 2a), where the diffraction spots of the (3×3) superlattice are in line with those of the PtSe₂ lattice. On the basis of LEED and STM measurements, the moiré pattern can be explained as the (3×3) PtSe₂ supercells located on the (4×4) Pt(111) atoms (more discussion about the moiré pattern can be found in Figure S1 in the Supporting Information).

To gain further insight into interfacial features of the PtSe₂/ Pt(111) sample, we performed a cross-section high-angle annular-dark-field (HAADF) STEM study. A Z-contrast image of the PtSe₂/Pt(111) interface is shown in Figure 2d. One bright layer combined with two dark layers observed on the topmost surface suggests a Se-Pt-Se sandwich configuration (as indicated by a model diagram superimposed in Figure 2d). The atomically resolved bulk Pt substrate lattice with an experimentally measured interlayer spacing of 2.28 Å served as a reference for calibrating other spacing measurements. After the calibration, the spacing between the Se sublayers in the Se-Pt-Se sandwich is found to be 2.53 Å, which is in agreement with the calculated value in single-layer PtSe2. These atomicscale cross-section data obtained by STEM further verify that the fabricated structure is indeed a single-layer PtSe2 film on the Pt(111) substrate.

The combination of LEED, STM, and STEM studies indicates a (3×3) single-layer $PtSe_2$ on a (4×4) Pt(111) structure. We then carried out DFT calculations based on this structure. The simulated STM image is shown in Figure 2e, in which the overall features of the experimental STM image (Figure 2c) are well reproduced. Only the hexagonally arranged Se atoms in the topmost sublayer of monolayer $PtSe_2$ are imaged. The remarkable agreement between the STM simulation and experimental STM observation strongly supports our conclusions and thus demonstrates the successful growth of a highly crystalline $PtSe_2$ monolayer.

Having grown highly crystalline single-layer PtSe2, we investigated its electronic energy band structure by ARPES. Figure 3a shows ARPES data measured along the high symmetry direction K-\Gamma-M-K in the hexagonal Brillouin zone at a photon energy of 21.2 eV. Data taken at other photon energies show the same dispersion, confirming the 2D character of the monolayer PtSe2. Second-derivative spectra of raw experimental band structures (Figure 3a) are depicted in Figure 3b to enhance the visibility of the bands. Here, the top of the valence band is observed to be at -1.2 eV at the Γ point, and the conduction band is above the Fermi level, indicating that monolayer PtSe2 is a semiconductor. It is quite different from the bulk PtSe₂, which—according to calculations²⁶—is a semimetal. A direct comparison between the ARPES spectrum (Figure 3b) and the calculated band structure (green dotted lines in Figure 3b) shows excellent quantitative agreement. Combining the ARPES spectra with DFT calculations, we confirm that we have synthesized a single-layer PtSe2, and that the epitaxial PtSe₂ essentially has the same electronic properties

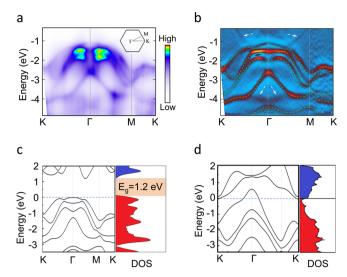


Figure 3. ARPES spectra and valence bands of single-layer $PtSe_2$. (a) ARPES spectra obtained on the monolayer $PtSe_2$ on Pt(111) along the $K-\Gamma-M-K$ directions in the hexagonal Brillouin zone (the high-symmetry directions are shown in the inset). (b) Second-derivative spectra of the raw ARPES data in (a). The calculated valence bands, superimposed as green dashed lines, are in excellent agreement with the experimental data. The bands marked by white arrows are from the Pt substrate. (ARPES result of clean Pt(111) surface is shown in Figure S2 in Supporting Information.) (c), (d) Theoretically calculated band structure and density of states (DOS) of monolayer $PtSe_2$ and bulk $PtSe_2$, respectively.

as the free-standing single-layer PtSe₂. For the first time, the band structure of monolayer PtSe₂ films has been determined experimentally.

The semimetal-to-semiconductor transition was revealed by DFT-LDA calculations. As shown in Figure 3c, the band structure and density of state (DOS) suggest single-layer PtSe₂ is a semiconductor with an energy gap of 1.20 eV (2.10 eV bandgap is predicted from GW calculation²⁷). As a comparison, the band structure and DOS of bulk PtSe₂ are plotted in Figure 3d, which confirm that bulk PtSe₂ is semimetallic.²⁶ When increasing the thickness, bilayer PtSe₂ remains a semiconductor, but the energy gap decreases to 0.21 eV. Starting from a trilayer, PtSe₂ becomes semimetallic (calculated band structures of bilayer and trilayer PtSe₂ are listed in Figure S3 in the Supporting Information). Therefore, only single-layer PtSe₂ is a semiconductor with a sizable bandgap.

The opening of a sizable bandgap within the range of visible light makes monolayer PtSe₂ potentially suitable for optoelectronics and photocatalysis. We explored the photocatalytic properties of monolayer PtSe₂ by the degradation of methylene blue (MB) aqueous solution, which serves as a typical indicator of photocatalytic reactivity. ^{28,29} The MB molecules adsorbed on the PtSe₂ films are degraded by electrons that are excited by visible light. The time-dependent degradation of MB with a single layer PtSe₂ catalyst was monitored by checking the decrease in the intensities of characteristic absorbance peaks of the MB molecules (shown in Figure 4). The photo degradation portion of MB molecules reached 38% after visible-light irradiation for 24 min. This rate is about four times faster than the rate obtained using PtSe₂ nanocrystals, ³⁰ putting single-layer PtSe₂ in the same class as nitrogen-doped TiO₂ nanoparticles for photocatalysis. ^{31,32}

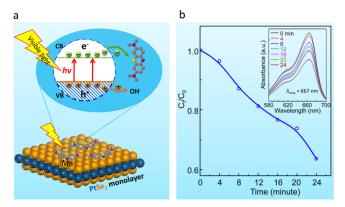


Figure 4. Photocatalytic activity of a single-layer PtSe₂ film. (a) Schematic diagram of the photocatalytic degradation of methylene blue (MB) molecules. Electrons and holes are excited by visible-light irradiation of epitaxial PtSe₂ monolayer films. The MB degradation by photoinduced electrons demonstrates the photocatalytic activity of PtSe₂ monolayer films. (b) Time trace of the normalized concentration (C_t/C_0) , where C_t and C_0 are the MB concentrations at time t min and 0 min, respectively) of the absorbance at a wavelength of 667 nm, the main absorbance peak of MB. The inset shows the UV—vis absorption spectra of MB, recorded at time intervals of 4 min.

With the existence of an energy gap in the single-layer PtSe₂, optical excitations can occur between the Γ point and the valley point along the Γ -M direction. This is reminiscent of the recently discovered valley-selective circular dichroism in MoS₂ by circularly polarized light.^{33–35} To explore this possibility, we calculated the degree of circular polarization of free-standing single-layer PtSe₂. The calculated circular polarization due to the direct interband transition between the top of the valence band (VB) to the bottom of the conduction band (CB) is shown in Figure 5a. It shows significant circular dichroism

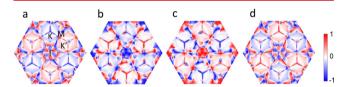


Figure 5. Momentum dependence of circular polarization of single-layer $PtSe_2$. (a–d) represent the calculated circular polarization due to the direct interband transition from the top of the valence band (VB1) to the bottom of the conduction band (CB1), from VB1 to the higher conduction band (CB2), from the lower valence band (VB2) to CB1, and from VB2 to CB2, respectively. Significant circular dichroism polarization exists along the M–K direction and near the Γ point.

polarization along the M–K direction and near the Γ point. In view of the indirect energy gap, this process can be assisted by lattice vibrations. It is noteworthy that the circular dichroism polarization not only exists in the transition between the top of the VB and the bottom of the CB, it also exists in transitions to higher energy levels. Circular dichroism polarization due to transitions from the vicinity of the VB to the vicinity of the CB can be clearly seen in Figure 5b, c, and d. Due to this significant circular polarization, in the presence of a nonvanishing in-plane electric field, the anomalous charge current driven by the Berry curvature would flow to the opposite edges, leading to a valley-polarized current and the resulting quantum valley Hall effect. 36,37

In summary, we have successfully fabricated high-quality, single-crystalline, monolayer PtSe₂ films, a new member of the TMDs family, through a single-step, direct selenization of a Pt(111) substrate at a relatively low temperature (\sim 270 °C). Characterizations by LEED, STM, STEM, and DFT calculations elucidated both in-plane and vertical monolayer structures with atomic resolution. The ARPES measurements and their agreement with calculations revealed the semiconducting electronic structure of the single-layer PtSe₂. Together with the photocatalytic performance observed experimentally, monolayer PtSe2 shows promise for potential applications in optoelectronics and photocatalysis. Moreover, the circular polarization of monolayer PtSe2 in momentum space indicates a promising potential for valleytronic devices. Thus, our studies are a significant step forward in expanding the family of single-layer semiconducting TMDs and exploring the application potentials of thin TMDs in photoelectronic and energy-harvesting devices.

Methods. Sample Preparation. Single-layer $PtSe_2$ films were grown in an ultrahigh vacuum chamber, with a base pressure of 3×10^{-10} mbar, equipped with standard MBE capabilities. The Pt(111) substrate was cleaned by several cycles of Ar^+ ion sputtering followed by annealing until sharp (1 \times 1) diffraction spots in the LEED pattern and clean surface terraces in the STM images were obtained. High-purity Se evaporated from a Knudsen cell was deposited onto the clean Pt(111) surface at room temperature. The sample was subsequently annealed up to 270 °C to achieve selenization and crystallization. The growth process was monitored by XPS and LEED. After growth, the sample was transferred to an STM chamber for imaging and to an ARPES chamber for measuring the electronic band structure.

STEM Characterization. The cross-section image shown in Figure 2d was obtained from a JEOL JEM-ARM 200 operating at 200 kV with a sample prepared by a JEOL Ion Slicer. Practically identical images were also obtained from a Nion UltraSTEM 200 operating at 200 kV using a sample prepared with a Zeiss dual beam Focused Ion Beam instrument and a Fischione Nanomill. For the UltraSTEM 200, the probeforming angle was ~30 mrad, and the inner angle of the HAADF detector was about 63 mrad. For the JEOL JEM-ARM 200, the probe-forming angle was 30 mrad, and the inner angle of the HAADF detector was about 90 mrad.

ARPES and XPS Measurements. The ARPES data were measured using both synchrotron radiation (BL-1 at HiSOR, Japan) and an ultraviolet lamp with He I α spectral line. For the synchrotron data the sample was held at 20 K. The energy analyzer was a VG Scienta 4000/8000 with total energy resolution set to less than 20 meV. The synchrotron XPS data were taken at a photon energy of 105 eV.

First-Principles Calculations. DFT calculations were performed within the local density approximation (LDA) by using the Vienna ab initio simulation package (VASP). The projector augmented wave method was employed. The periodic slab models included four layers of Pt substrate, one layer PtSe₂, and a vacuum layer of 15 Å. All atoms were fully relaxed except for the bottom two substrate layers until the net force on every atom was less than 0.01 eV/Å. The energy cutoff of the plane-wave basis sets was 400 eV, and the k-points sampling was 5 × 5 x 1 with the Monkhorst–Pack scheme. The momentum dependence of the degree of circular dichroism of 1T-PtSe₂ is given by $\eta(k) = (|P_+(k)|^2) - (|P_-(k)|^2))/(|P_+(k)|^2) + (|P_-(k)|^2))$, where P_\pm (k) = P_α^{ci} (k) \pm iP_α^{ci} (k). The

direct interband optical transition matrix $P_{\alpha}^{ci}(k) = \langle u_{c,k}|\hat{p}_{\alpha}|u_{i,k}\rangle$ was obtained by DFT calculations, including spin—orbit coupling, where $|u_{c,k}\rangle$ is the periodic part of the Bloch function. In the circular polarization calculations, the whole Brillouin zone was sampled with a very dense k-point grid of 14 400 points.⁴⁰

Photocatalytic Experiment. The photocatalytic activity of as-prepared monolayer PtSe2 films was tested by catalytic degradation of methylene blue serving as a standard model dye under visible-light irradiation at room temperature. The ultraviolet/visible-light source was a 150 W Xe lamp located at a distance of 15 cm above the solution. A set of appropriate cutoff filters was applied to determine illumination wavelength and ensured that the photocatalytic reaction took place just under visible light. The as-prepared monolayer PtSe2 films were exfoliated from the Pt(111) substrate by ultrasonication in aqueous solution. The Pt(111) crystal was picked out from the solution to avoid its possible influence on the photocatalytic activity. Then MB molecules and ethanol (0.01 mL) were added to the solution containing peeled PtSe2 monolayers. Before photocatalytic reaction, the solution was kept in darkness for 1 h in order to achieve an adsorption-desorption equilibrium between the PtSe2 film and MB molecules. After that, the suspension was exposed to visible-light irradiation. At time intervals of 4 min, solution samples were collected and their absorbance measured by a commercial ultraviolet-visible (UV-vis) spectrophotometer. Accordingly, the intensity changes of characteristic absorbance peaks of the MB molecules were recorded. The photocatalytic performance of PtSe2 monolayers was thus evaluated by the time-dependent degradation rate C_t/C_0 shown in Figure 4b.

ASSOCIATED CONTENT

S Supporting Information

Top view of the model of a PtSe₂ monolayer on a Pt(111), ARPES result of clean Pt(111), and calculated band structures of monolayer, bilayer, trilayer, and bulk PtSe₂. The Supporting Information is available free of charge on the ACS Publications website at DOI: 10.1021/acs.nanolett.5b00964.

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Notes

The authors declare no competing financial interest.

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