

Dual Nature of Matter

$$h = 6.624 \times 10^{-34} \text{ Js} = 4.14 \times 10^{-15} \text{ eV/sec}, 1 \text{ eV} = 1.6 \times 10^{-19} \text{ J}$$

$$E = \frac{hc}{\lambda} = h\nu = \frac{12400}{\lambda(\text{in nm})} \text{ eV}$$

Properties of photons:

1. Energy radiated is in form of photons, speed of photon $= 3 \times 10^8 \text{ m/s}$
E depends on ν . (st. line)
2. Photons are electrically neutral, neither deflected by electric or magnetic fields.
3. During propagation of photons, speed, λ of photons changes but $\nu = \text{constant}$.
4. Rest mass of photon = zero.
5. Momentum

$$\text{Mass of photon (m)} = \frac{h\nu}{c^2} = \frac{h}{\lambda c} \quad P = mc = \frac{h\nu}{c} = \frac{h}{\lambda}$$

$$\rightarrow \text{Intensity (I)} = \frac{\text{Energy}}{\text{Area} \times \text{time}} = \frac{E}{At} \left(\frac{W}{m^2} \right) \quad I \propto \frac{1}{r^2}$$

$$\text{no of photons emitted (n)} = \frac{IA\lambda}{hc}$$

$$\text{absorption coefficient (a)} = \frac{\text{absorbed energy}}{\text{total energy}} \rightarrow 0 \leq a \leq 1 \quad \left. \begin{array}{l} \\ \end{array} \right\} a + r = 1$$

$$\text{reflected coefficient (r)} = \frac{\text{reflected energy}}{\text{total energy}} \rightarrow 0 \leq r \leq 1$$

\rightarrow Radiation force: force acting on the surface of the body due to radiations in the form of photons is known as radiation force. Corresponding pressure is radiation pressure.

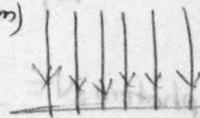
Case 1: falls normally

$$a=1, r=0$$

$$\text{initial momentum (i.m)} = h/\lambda \quad \text{final momentum (f.m)} = 0$$

$$\text{force acting on photon} = \frac{IA}{c} \uparrow \downarrow$$

$$\text{Radiation pressure} = \frac{I}{c}$$



Case 2: The amount of radiation falls on the surface completely reflected.

$$a=0, r=1 \quad i.m = h/\lambda \quad f.m = h/\lambda \quad \Delta m = 2h/\lambda$$

$$n = \frac{IA\lambda}{hc}, \text{ force of photons} = \frac{2IA}{c}$$

$$\text{Radiation pressure} = \frac{2I}{c}$$

Case 3: partial absorption, partial reflection

$$0 < a < 1, \quad 0 < r < 1$$

$$\Delta p = \frac{2h}{\lambda} \quad F = \frac{IA}{c} [1+r] \quad P = \frac{I}{c} [1+r]$$

$$\text{no. of photons incident } (N_i) = \frac{I(A \cos \theta) \lambda}{hc}$$

$$\text{If made } \theta \text{ with surface } \parallel \parallel \text{ absorbed } (N_a) = \frac{I(A \cos \theta) \lambda}{hc} (1-r)$$

$$\text{During absorption } F_{R\perp} = \frac{IA \cos^2 \theta}{c} (1-r) \quad F_{a\parallel} = \frac{IA (\cos \theta) \sin \theta (1-r)}{c}$$

$$\text{Total radiation } F_{\perp} = \frac{IA \cos^2 \theta}{c} [1+r] \quad F_{\parallel} = \frac{IA \cos \theta \sin \theta}{c} (1-r)$$

$$\text{Resultant force } F = \frac{IA \cos \theta}{c} \sqrt{1+r^2+2r \cos 2\theta}$$

$$F_{\text{radi}} = \frac{2P_{\text{inc}}}{c}$$

$$\text{Radiation pressure } P = \frac{I \cos^2 \theta (1+r)}{c}$$

Case 1 if θ made by radiation

$$\Delta p = \frac{IA \cos^2 \theta}{c} \text{ along surface}$$

$$\Delta p = \frac{IA \cos \theta \sin \theta}{c}$$

$$\text{Radiation pressure} = \frac{I \cos^2 \theta}{c}$$

Case 2: if θ made by photon

$$\Delta p = 0 \text{ along surface}$$

$$\Delta p = \frac{2h}{\lambda} \cos \theta \text{ along}$$

$$\text{No. of photons } N = \frac{I(A \cos \theta) \lambda}{hc}$$

$$\text{Total } \Delta p_{\parallel \& \perp} = \frac{2IA \cos^2 \theta}{c}$$

$$\text{Radiation pressure} = \frac{2I \cos^2 \theta}{c}$$

Photoelectric effect:

- Min energy required to eject e^- from metal is work function.
- Min frequency of photon required to eject e^- from metal is threshold frequency.
- Max. wavelength of photon required to eject e^- is threshold wavelength.

$$\text{Work function } (W) = h\nu_0 = \frac{hc}{\lambda_0} \text{ (J)} \Rightarrow W = \frac{12400}{\lambda(\text{\AA})} \text{ eV}$$

if $\nu < \nu_0$, $E < W$, $\lambda > \lambda_0$ no emission of photoelectron.

if $\nu \geq \nu_0$, $E \geq W$, $\lambda \leq \lambda_0$ e^- ejected from metal.

$$\rightarrow E = W + KE$$

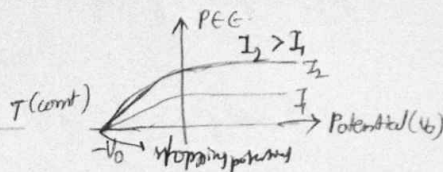
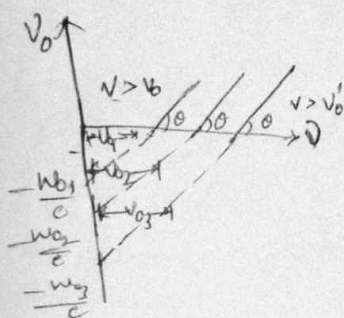
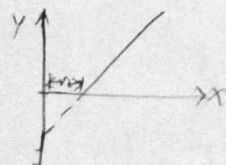
$$\eta = \frac{I A \lambda}{hc} \propto \text{photo } e^- \propto \text{photo current}$$

$$E = \frac{hc}{\lambda} = h\nu$$

$$KE_{\text{max}} = eV \quad (V = \text{stopping potential})$$

$$KE = h\nu - W_0 \Rightarrow eV_0 = h\nu_0 - W_0 \Rightarrow V_0 = \frac{h}{e} \nu_0 - \frac{W_0}{e}$$

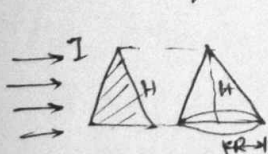
$$y = mx + c$$



→ Franck-Hertz experiment - discrete energy levels of atom

Photo-electric experiment - particle nature of atom

Davison-Germer experiment - wave nature of e^- . (49V to 68V) at 54° max. I $\theta \approx 50^\circ$



$$F = \frac{I(RH)}{c}$$

$$I$$



$$F = \frac{I(RH)}{c}$$

Recoil velocity

$$v_R = \frac{h}{m\lambda}$$