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\mathcal{O}_{2}^{+} [Pt f_{6}]	
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- / / / The state of the state	Oz [Pt+6] Oz EXE has same I-P values (1175kJ/mol 1170kg/m

2/5/23 Xenon form Compounds with fluxure Xef2, Xefy, Xef6 Neo3, Xeo Pereparation in 8 1 of 1 poil 1 15% Xe +F2 673k XeF2
Excen) XeF2 Xe +2F1 8732 XeFy Xe+3F₂ 573k, XeF6 CE +2Xe +2OF2 - 2xeF2+O2 Koomtemp Xe + O2F2 low term (XeF2+02 stoss) 4 Xe + 80f2 = 2 Ne F4 + 2 Xe O F4 + 302 Xefy +02 F2 XeF6 +02 XeF2, Xefy Xef (latourless) deliquentes solids. density order XeF2 > XeF4 > XeF4 XeF2, XeFy, XeF6 good fluoringting agents cotochtext jos XefgXefy xef2XH HOT Stax XeF6+8NH3 6NH4F+N2+Xe Xefy +Sfy SF6 + Xe Oxiduring power Xefy+Pt -> Ptfy+Xe Xefy>Xefy>Xefy>Xefy>Xefy Xef2+Hg→ Hgf2+Xe Xef 2+ (2+4) - (2+4) +2+ Xe 34 1000 2NO+ Xefz -> Xe+2NOF XeF2, XeFy, XeF6 good oxiding agent

Xefo +6H(1 -> 3Cl2+6HF+ Xe Xefy +4KI → 212 +4KF + Xe XeB+2H(1 -> C12+2HF+Xe Neff + Brog + H20 - Brog + Hf + Xe XeF2 exiderses (e+3 to Cety) Xef2 + SOy2+ Ce2(SOy)3 -> 2 (e(SOy2+Xe+F2 Hydecolyss: xe+ 3/2 513kg Xef6 2 XeF2+2120 -> 2xe+44frt02 hydrolysis of XeFy is dispropositionerion seasion 6 XeFy +12 Hz0 - 4XeO3+2 Xe +24 HF +302 Xefy H20 XeOf2 H20 XeO3 XeFy+120 -> XeOFy+2HF Xe Fo+2H20 -> XeO2 F2 + 4HF Sold xefe+ 3420 > XeO3+ 16HF Xe F₆ poential Xe OF 4 partial Xe of 2 Complete Xe 03 Xefo tOH -> HX e Oy Xeo TXet H20 +02 extent There & Kenate ion powerate Xefo cannot be stored in Iglan: 12+ 110x KeF6 + SIO2 -> SIFY+XeOF4 - SIO2 SIFY+XeO2 F2 9×+ AgH ~ Siky+ Xeo, ← Sig form the ions with fluoride ion alleptor Xefy + Sbfs -> [xef] (Sbf6] sets , x ety, xefe good oxiditing

 $XeF_2 + PF_5 \rightarrow [xeF]^+[PF_6]^$ form -ve sons with fluoride ion and done Xefo +MF MT [xefy] Total 3 by M- Na, K, Rb, Cs Xefo + Cst -> cs+[xef] Steructure

XeF2 Xe-sp3d Loobsdoto betselfel $Xe - 5s^2 5p^6$ 1 ME state -552 5p 5 5d' sp3d 31.P XCOFL 3rd Estate 552 5p3 THEN INION LINEAR I dA - PB bond Aprose pylamidal Xe= 16 Xe-5525p6 10 and 1 grad Estate - 552 5p45d2 3101332 1p3d2 pt-9X square planar

Kets+ 19ts -> | Xell 1[PF] Xe F6 Xe=5525p6 word downly down most we smoot 3rd E. Mate 1+ 552 15p3 7145d3 12.p +10 - Fp3d3,11X Distarted octahedral Will 348 XeOfy TILP Xe=+6 $5s^2 5p^3 5d^3$ 3rd Estate M and ahh 1 dy - pa bon1 XeQF Xe =+6 Xe=sp3d= 1-LP See saw Memehore I doll por bonds XeOF2 Ke-p3d 2.49 $\chi e - \eta ^3 d^2$ Octahederal

XeO3 F2 100X Terigonal bipy camidal Xe+8 93 -0/ 3dTI-pTI Boods Ne-1p3d XeF5 pentagonal planar 3/5/23 XeO3 46 O.S XeF6+340 - XEO34BHFILL 6 XeFy +12H20 - 2xe03 + 4xe +24 HF +302 XeO3 Colonules, deliquercent's explosive solid Brutable Ne and O2 . moderned lake 03 - explorive solid rouiding agent 3Pu+2 + XeO3 +6+1 3Pu+4 + Xe+3120 (XeO2 + XeOF4 - XxeO2F2 while all bruged pritambrowsque / X & 782300 mother Sec- 552 5p6 MN wespour to treet Oppy and Marker - 22 562 29 xe pyamidal m 2nd Es - 5s² 5p4 5d² 31d 6,1 - 512 5p3 5d3 Miller of board 3 df-pubords of houter all 111 of its very low solubility in blood +80.3 Step I: XeF6 +Ba(OH)2 -> BaxeO6 + BaF2 + Xe+H20 gosdon and in given tower Barcier percunate Step II: Ba2 XeO6 + 2 H2 soy - 2Basoy + XeO4 + 2 H2O Level Conce of mapper in pish Xe Oy Coloweles, explosive gas Les stable than XeO3 exploded at low temp . 233K

XeF6+ Keoy -> NeO3F2+ XeOF4 Tengonal by from tal XeOy TIDE Xe-sp3 4th Es = 5s1 5p3 5d4 produgered planes 4 dTITHATE bonds - OCHEH, 42% EXCE + 12 H) 0 - 9 14 CM + 4xc + 2410 + 300 D'DON selevators deliquerent proposine nolid Uses of He; Unstable Xe and O2 -> Filling are balloom for meterclogical observations. - 4 Gas-cooled nuclear deceators stugE -> Liquid He finds use as conjugent agent for caverying out vacious explainents at low temperature. -> To produce and nestain powerful superconducting magnet which from an essential part of modern NAME Medermeters and Magnette Revinance Imaging (MRI) As a diducnt for onygen in modern diving apparatus because of its very low solubility in blood Meon: Used in doutrouge tubes and perform bulk. Used in botanical goodons and in green houses. Augon: Augon is used mainly to bround an inext atmorphere in high temperatures metalliogical Lecces - for electeur bulk in lab, for handling rebrances that air_ semitive.

to reparete the Hobbe gas from rubb gas Dewards method is used Noble gares mixture He, He, Ar, Kr, Xe belought in contact with coconut Charcoal at 173k Unadvolbed Asy, Kr, Xe advoested Contact with harcoal other charcoal at at 180'c (93 K) lig. aire temp (77k) He unadioned Fiert charcoal Ne "Ar diffused KrEXe attoribed on Terms eines to (90°C) سەھەرىي fresh chascoal Ke in advolved Ne evolved state warm Xe evolved

-> Quinnot, H2S, SO2, CH3OH, CH3CN, etc form clatherates "Hyperth or soul others by sage stock others" -> Intermolecular spaces of molecular compounds is Occupied by noble gans at Certain applied pecemiar (10-40 atm). does not form clatherates. \rightarrow He fale Ari Kr, Xe Quinol Ar, Xe, Kr form clatherates with above compounde