Table 1.1: Adsorption energies (E<sub>ads</sub>, eV) of intermediates in the CO<sub>2</sub> reduction reaction towards CH<sub>4</sub> and methanol on Cu clusters. Hollow sites are maked as either hollow, FCC, or HCP.

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Cu <sub>78,2</sub>	Site	top-top	bridge/bridge	top	HCP	top	bridge	bridge	top		bridge	bridge	4-fold/4-fold	4-fold/4-fold	:	bridge	hollow/4-fold	
	Eads	-0.03	$0.64/0.68^{b}$	0.84	1.89	0.78	1.18	-0.52	0.35		-0.53	0.63	$0.93/1.16^{b}$	$1.91/2.26^{b}$	(	89.0-	$0.38/0.50^{b}$	
Cu <sub>78,1</sub>	Site	top-top	hollow	bridge	hollow	bridge-top	bridge	hollow	top		bridge	hollow	4-fold	4-fold		hollow	hollow	
	Eads	0.38	09.0	1.09	1.78	0.71	1.50	-0.44	0.64		-0.36	99.0	1.17	2.08		-0.51	0.17	
Cu <sub>147</sub>	Site	top-top	top (corner)	top (C)	HCP	top (corner)	bridge	HCP	top (C)		top (corner)	bridge	dist. 4-fold	dist. 4-fold	100	HCP	HCP	
	Eads	0.47	0.94	1.37	2.33	1.07	1.71	-0.16	0.86		-0.14	1.01	1.74	2.83	•	-0.19	0.52	
Cuss	Site	top-top	top (corner)	top (C)	HCP	top (corner)	bridge	HCP	top (C)		bridge	HCP	dist. 4-fold	dist. 4-fold	5	HCP	HCP	
	Eads	0.49	0.71	1.29	2.02	0.00	1.67	-0.05	0.82		-0.05	0.92	1.35	2.49	•	-0.12	0.54	
Cu(111)	Site	1-				$DNB_c$					bridge	FCC	FCC	FCC	Ç	FCC	FCC	
	Eads	92.0	0.98	1.54	2.12	0.92	1.80	90.0	1.06		0.16	1.11	1.68	3.05	(	0.03	0.72	
Absorbate		COOH	00	CHO	COH	$CH_2O$	CHOH	$CH_3O$	$CH_2OH$		$ m CH_3$	$ m CH_2$	CH	C		ЮН	0	
		(P)	<u>်</u>	(p)	(e)	(£)	(g)	(p)	(i)		(E)	(k)	(3)	(m)		(n)	0	

a: -0.45 added to energy to account for poor description of C=O bonds with DFT. b: Two adsorption energies are given. The first is the adsorption energy for the strongest binding site. This binding site was located on the other side of the cluster from where the \*COOH moiety had initial adsorbed to. The second is the adsorption energy of a binding site near where the \*COOH moiety had initially adsorbed to. c: Did not bind.