



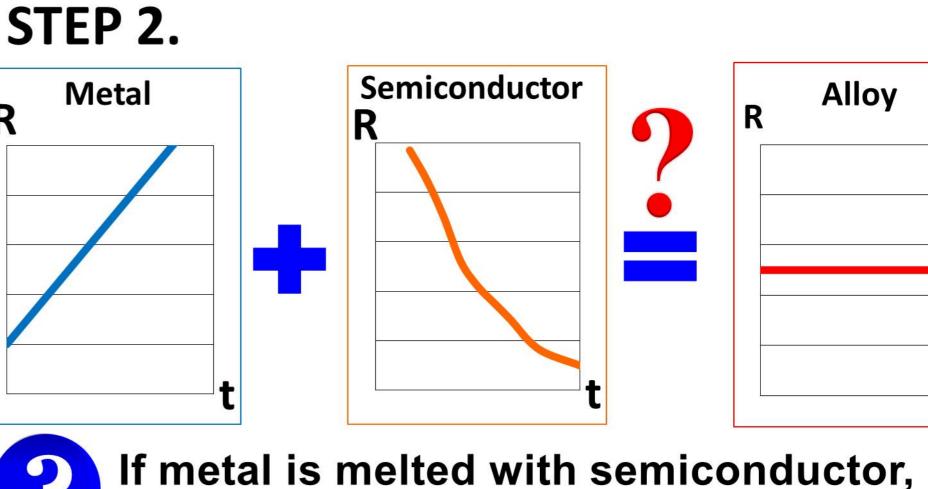
STEP 1.

Table 1. Properties of thermostable materials.

Properties	Manganin	Constantan
Density, kg/m³	8400	8800
Specific heat, J/(kg*K)	418	415
Resistivity, Ω*m*10 ⁻⁸	45	48
Temperature coefficient of resistance, K ⁻¹	0.00002	0.00003
Minimum price per 1 kg	28 \$	22 \$

These materials are **pretty** expensive, they have low specific heat and high density.

How can a cheap, lightweight, thermostable material with high specific heat be produced?



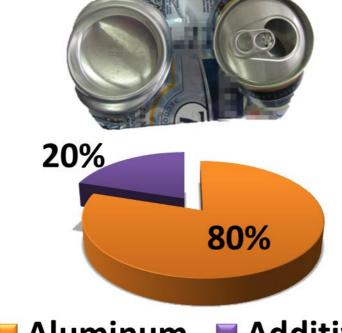
could a thermostable alloy be produced?

What metal and semiconductor to choose for the production of a new material?

STEP 3.

Table 2. Properties of aluminum.





■ Aluminum
■ Additives Figure 1. Aluminum can and its composition.

> its conductivity depends on the impurities; it becomes a semiconductor when melted with Al; > cullet contains it.

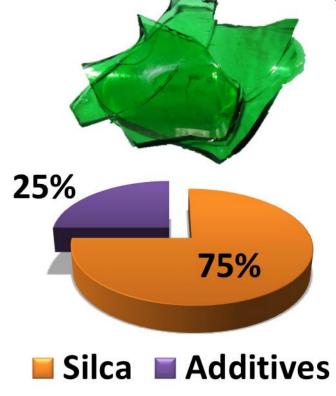


Figure 2. Cullet and its composition.

Aluminum is the suitable metal for the production of the new material: its price is low, it has high specific heat and low density.

Silicon dioxide is a suitable semiconductor for the production of the new material.

Aluminum and silicon dioxide alloy is silumin. Are the required characteristics to be found in silumin?



per 1 kg;

Figure 3. Silumin.

is based on aluminum and quartz sand alloying $(4-22\% SiO_2);$ > its cost is about 3 \$

its electrical properties haven't been deeply studied by scientists yet.

So I tried to study them...

source of energy millivoltmeter amperemeter R_{box} t, °C infrared thermometer resistance pattern spirit stove box

Figure 4. Installation scheme.

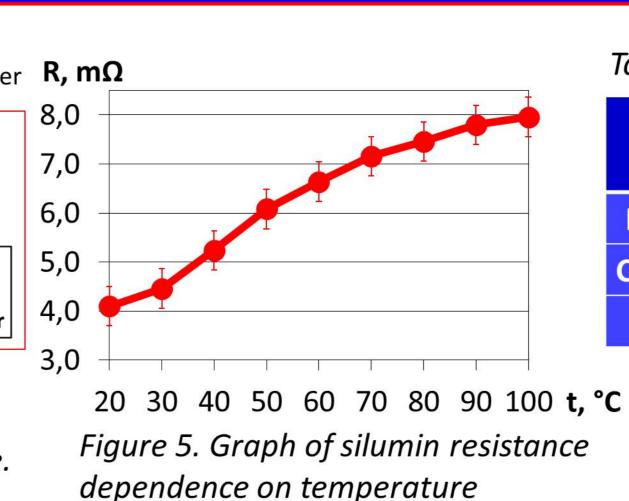


Table 3. Properties of aluminum.

	Temperature coefficient	
	of resistance, K ⁻¹	
Manganin	0.00002	
Constantan	0.00003	
Silumin	0.017	



Silumin is **NOT** a thermostable material.

What is the optimal mass ratio

of glass and aluminum for thermostable material?

STEP 5.

Stages of the production of new material with different mass ratio of glass and aluminum.







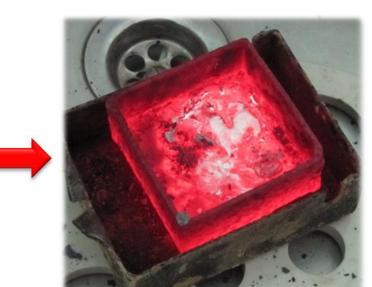
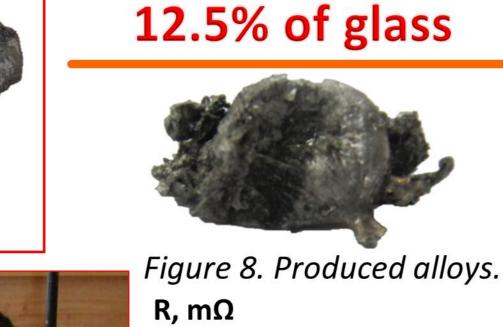


Figure 6. Production of new material in school lab.



Figure 7. Failed patterns.



12

11

10



100 t, °c

90

→12,5% of glass

◆25% of glass

◆37,5% of glass

37.5% of glass





Resistance dependence of glass and aluminum alloy with 37.5% of glass on temperature is **insignificant**.

I named this material AluGlass 1



How to increase the thermal stability of AluGlass?

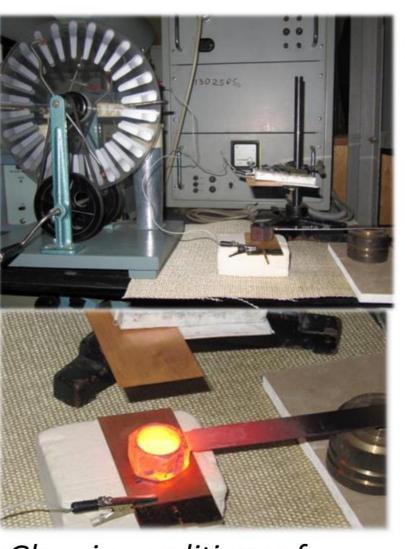
Figure 10. Graphs of glass and aluminum alloys

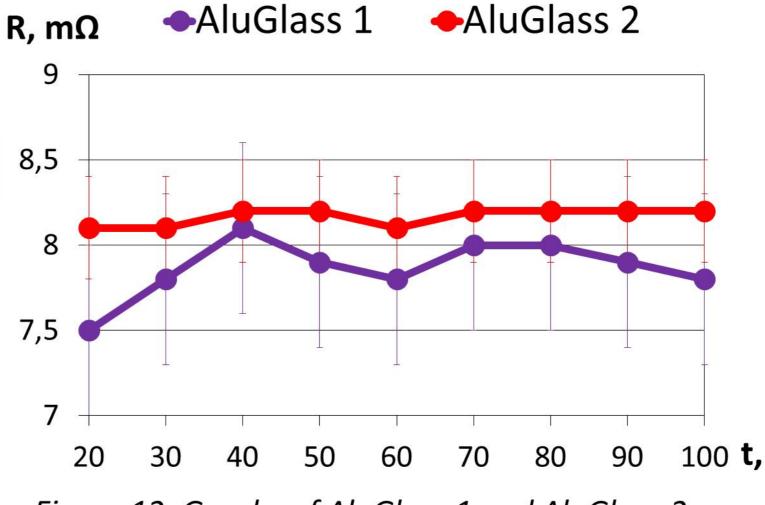
resistance dependence on temperature.

STEP 6. metal plate fusion electrostatic

metal tray

machine





100 **t, °C** Figure 12. Graphs of AluGlass 1 and AluGlass 2 resistance dependence on temperature

AluGlass 2



Figure 13. AluGlass 2 was crystallized in uniform electric field.



Crystallization in conditions of uniform electric field increased thermal stability of AluGlass.

Figure 11. Crystallization of AluGlass in conditions of uniform electric field.