



## Vulnerability of the Bay of Bengal to Ocean Acidification.

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# Vulnerability of the Bay of Bengal to Ocean Acidification



INTERNATIONAL UNION FOR CONSERVATION OF NATURE



# Vulnerability of the Bay of Bengal to Ocean Acidification

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IUCN (International Union for Conservation of Nature)  
Bangladesh Country Office  
2015

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## Executive Summary

Fossil-fuel combustion releases carbon dioxide to the atmosphere, leading to a warmer climate. Increasing atmospheric CO<sub>2</sub> is changing the global ocean's chemistry, as one-fourth of the anthropogenic CO<sub>2</sub> is absorbed by the ocean. In addition, ocean absorbs CO<sub>2</sub> from the respiration and breakdown of dead organic matter. When CO<sub>2</sub> dissolves in seawater, it forms carbonic acid, decreasing both ocean pH and the concentration of the carbonate ion.

The historical trends analysis showed an increasing water temperature with a decreasing pH levels over the period which may lead substantial effect on the biodiversity of the Bay of Bengal. The Institute of Marine Sciences and Fisheries (IMSF) in Chittagong University have been contributed in research and data generation from the coastal and marine ecosystems of Bangladesh. In addition, Bangladesh Navy, Bangladesh Inland Water Transport Authority and Coast Guard have been significantly contributed in hydrographical data collection and monitoring of the shelf water of Bangladesh in the Bay of Bengal.

Ocean acidification could affect marine food chain and substantially change the marine biota which is huge threat to global protein supply and food security for millions of people, including the multi-billion dollar fishing industry. Acidification can affect marine organisms, especially to those that build their shells and skeletons materials from calcium carbonate, such as species of corals, oysters, clams, mussels, and snails. The molluscs and crustaceans support valuable commercial and recreational fisheries, where the coral reef ecosystems support a variety of subsistence, recreational, and commercial fisheries worldwide.

Due to the effect of ocean acidification, an estimated 19% of the world's coral reefs have been damaged and a further 35% are seriously threatened. As a result, one-third of all reef-building corals are at risk of extinction. The Saint Martin's Island is the only coral bearing island of Bangladesh located in the Bay of Bengal. Among the 66 coral species in 1997, only 40 species were recorded in 2008 and the remaining 26 coral species may be lost in the next 11 years, posing serious threat to the coral biodiversity in Bangladesh. The ocean acidification refers to the process of lowering the oceans' pH levels which results coral bleaching, slow growth and decrease coral species diversity. Moreover, feeding and spawning areas of reef inhabiting aquatic species may be reduced which ultimately lead to dramatic loss of fisheries biodiversity. The

global economic value of these reef services has been estimated to be \$30 billion per year based on a value of \$100,000 per square kilometer per year. Projections of the lost coral reef area in year 2100 due to acidification are 16-27% that translates to economic losses of as much as \$870 billion in year 2100.

The impacts of ocean acidification may be either direct, through alteration of different physiological processes of organisms (such as survival, reproduction and development, growth, metabolism, thermal tolerance, immune response, and behavioural responses), or indirect, through biophysicochemical interactions (such as predator-prey abundance, nutrient recycling or habitat changes). The global marine capture fisheries and aquaculture industries have provided 110 million metric tons of human food in 2006 with a commercial value of \$170 billion. Changes in fishery industries due to acidification are likely to disproportionately affect the developing nations that often rely very much on marine-related economic activities than developed nations.

An increased CO<sub>2</sub> causes reduction in water pH which also reducing soil pH and responsible for the damage of mangrove vegetations. The low soil pH may harm breathing roots of mangroves and causes mangrove mortality. Acidification can also cause physical damage and alter topography and hydrology. Declining of mangrove ecosystem will not only cause colossal loss of coastal habitat, aquatic resources and biodiversity, but also will increase soil erosion, changes in sedimentation pattern and shoreline configuration, vulnerability to cyclonic storms, tidal bore and lead to the loss of feeding, breeding and nursery ground for various marine, estuarine and fresh water fishery resources.

Coastal people inherit skill and knowledge on fishing, shrimping, crabbing, agriculture, livestock rising, trading and other occupations. But, their livelihoods from fisheries, farming and forestry are at risk in many regions of the world vulnerable to ocean acidification. Thus, impacts on food yields and livelihoods will therefore also affect mental health and health-related behaviours. If acidification damages marine habitats or reduces harvests, the resulting decline in revenues could mount job losses and additional indirect economic costs.

In Bangladesh, the Ministry of Environment and Forests and its agencies in collaboration with some other Ministries/Departments are responsible to develop the policy documents addressing climate change challenges, including ocean acidification. Each of the Ministry or Department has specific

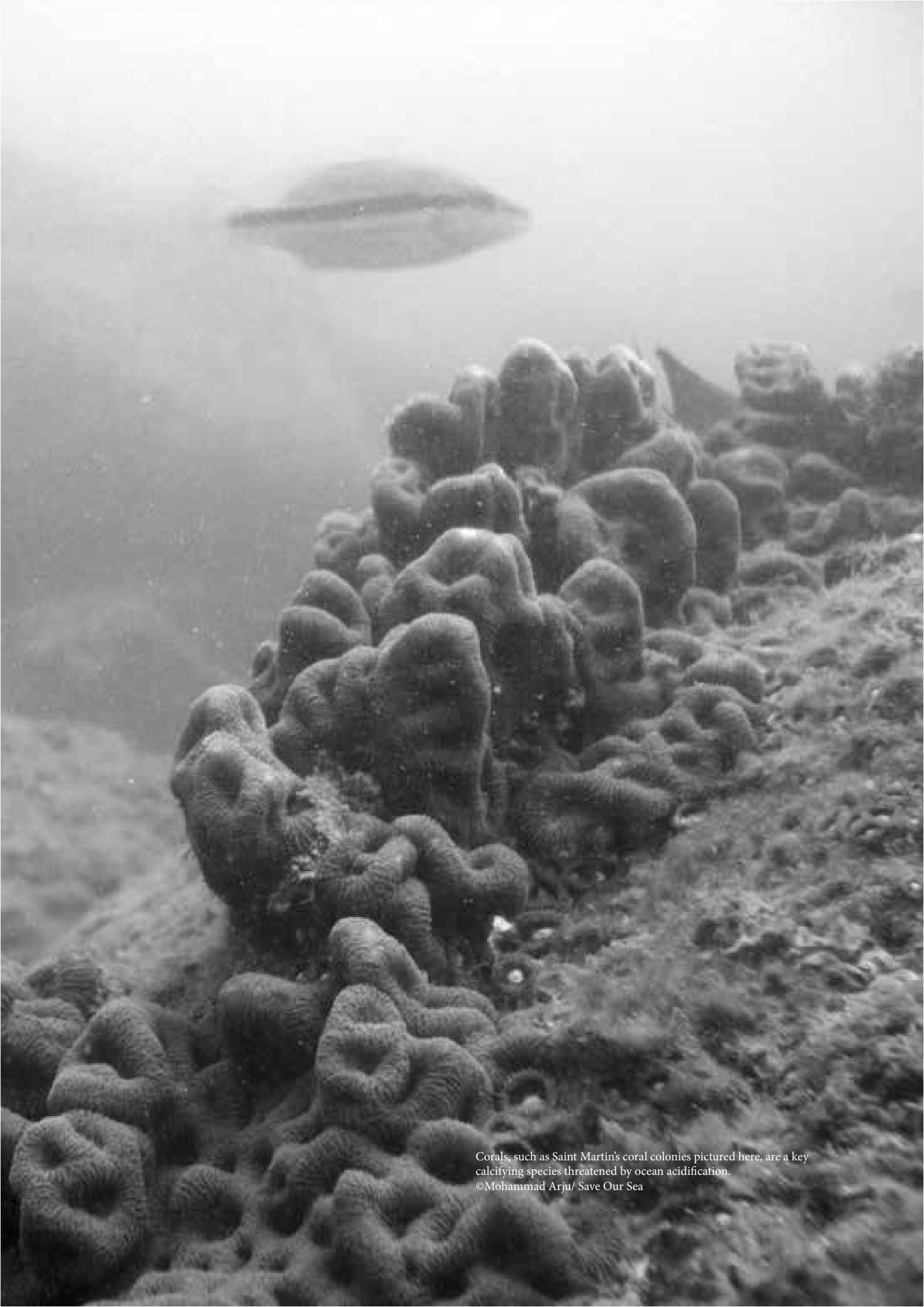
roles and responsibilities on mitigation and adaptation practices to enhance resilience of the community in the face of extreme natural events. Moreover, these wings are involved in advocacy, governance and negotiation in home and abroad. In addition, academic and research institutions work on climate change challenges, ocean acidification, carbon sequential, and aquatic food production issues to develop knowledgeable and skilled human resources. Given the economic and social importance of the oceans to human societies, government at the local, national, and international levels must begin to assess and implement adaptive approaches to acidification.

The UNFCCC agreed for National Adaptation Plans (NAPs) that can endow developing countries with a strong climate-smart strategic planning process and policy dialogue, embracing and integrating sector-wide and programmatic approaches as part of a coherent institutional, policy and regulatory framework. In addition, UNFCCC focused substantial further definition on loss and damage in the UNFCCC context. Bangladesh has prepared National Adaptation Programme of Action (NAPA, 2005) with 15 priority activities to combat climate change impacts, including awareness rising, technical capacity building, and implementation of projects in vulnerable regions with a special focus on agriculture and water resources. Bangladesh Climate Change Strategy and Action Plan (BCCSAP, 2009) was outlined to build capacity

and resilience within the country to meet climate change challenges over the next 20 – 25 years.

The development of adaptation capacity with resilient infrastructures (such as houses, roads, schools, health centres, community places) and income generating options are essential. As a potential initiative, coastal plantation, coral and oyster reefs development (i.e. introduction of living shoreline concept), high yielding varieties of rice/fish farming, raising livestock (for meat and milk) and poultry (for egg and meat) hold great promise. Thus, global and regional supports in the form of fund, technology, equipment and human resource development to promote adaptation capacity with resilient infrastructure are the prime requirements.

Moreover, funding is necessary to install in situ high-precision sensors around the territorial waters in the Bay of Bengal of Bangladesh to generate real-time monitoring of changing water conditions in sensitive areas. In this way, researchers will be able to analyse and integrate data for developing models to determine the current and future costs of ocean acidification. However, reductions in CO<sub>2</sub> emissions should be large enough to avoid the worst consequences of ocean acidification and climate change and at the same time it is important to introduce global economy-wide emissions policies and international agreements. Also, communicating of problems from ocean acidification to stakeholders (i.e. policy makers, donors, media and public) is utmost important.



Corals, such as Saint Martin's coral colonies pictured here, are a key calcifying species threatened by ocean acidification.  
©Mohammad Arju/ Save Our Sea

## 1. Introduction

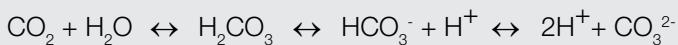
### 1.1. Background

Human emissions of too much carbon dioxide ( $\text{CO}_2$ ) are driving force of acidification of the ocean as well as global climate change. While research on various aspects of climate change has been widely studied, the problem of ocean acidification (OA) has only been recognized recently. Since the acidification of ocean water is primarily driven by the well-known law of chemical equilibrium between  $\text{CO}_2$  and water, the initial impact of ocean acidification is relatively clear (Caldeira and Wickett, 2005). However, the eventual impact depends on the complex interaction of many other variables.

air-sea gas exchange due to increasing anthropogenic  $\text{CO}_2$  is leading to increased hydrogen ion concentrations, and hence a fall in pH. Dissolved  $\text{CO}_2$ , carbonic acid and bicarbonate are also increasing; however, the concentration of carbonate ions is decreasing because of a reaction between  $\text{CO}_2$  and carbonate, further increasing the bicarbonate levels (Raven et al. 2005). The relative changes in bicarbonate, carbonate and hydrogen ion concentrations in the surface ocean arising from doubling, tripling and quadrupling of atmospheric  $\text{CO}_2$  (compared with pre-industrial values). Uptake of this additional  $\text{CO}_2$  has already increased the average acidity of the global ocean by ~30% (i.e. decreasing pH from 8.2 to 8.1, or by 0.1 units) since the beginning of the industrial revolution, and the increase in acidity is expected to be three fold (i.e. yielding a decrease in

#### Ocean Acidification – a simple primer

When  $\text{CO}_2$  dissolves in seawater it forms carbonic acid ( $\text{H}_2\text{CO}_3$ ), which dissociates to form equilibrium with hydrogen ions ( $\text{H}^+$ ), bicarbonate ions ( $\text{HCO}_3^-$ ) and carbonate ( $\text{CO}_3^{2-}$ ):



This equilibrium is dominated by bicarbonate, ( $\text{CO}_2 = 1\%$ ,  $\text{CO}_3^{2-} = 8\%$  and  $\text{HCO}_3^- = 91\%$ , of total dissolved inorganic carbon.

With continued uptake of  $\text{CO}_2$  by the ocean drivers, this equilibrium shift toward the centre, increasing the concentration of Hydrogen ions, and thereby reducing pH. This pH shift changes the equilibrium between bicarbonate and carbonate, driving that equilibrium also toward the centre, thereby depleting the available carbonate pool. This increases the solubility of  $\text{CaCO}_3$ .

Since the onset of the industrial revolution, an increasing partial pressure of atmospheric  $\text{CO}_2$  ( $\text{pCO}_2$ ) has already caused a ~30% increase in ocean acidity (= pH declines by 0.1 units). Therefore, a ‘business-as-usual’ type scenario will continue to raise atmospheric  $\text{CO}_2$  levels for the remainder of this century, leading to increased dissolution of  $\text{CO}_2$  in the oceans and further reduction in pH – this refers to ‘ocean acidification’.

Source: Havenhand 2012

The ocean reservoir of carbon is much greater than both of the terrestrial and atmospheric systems and provides an important net sink for carbon through exchange of  $\text{CO}_2$  across the air-sea. Over the past 200 years, atmospheric  $\text{CO}_2$  has increased from 280 ppm to a global average of nearly 390 ppm due to burning of fossil fuels, cement production and land-use changes (Hilmi et al. 2012). But, carbon uptake by the ocean has slowed down the atmospheric increase and its associated consequences for the Earth’s climate, and without such an uptake, it is estimated that the atmospheric  $\text{CO}_2$  would now be around 450 ppm (Sabine et al. 2004; Quere et al. 2009).

An increasing rate of addition of  $\text{CO}_2$  to seawater by

pH to 7.8) by the end of this century, if global  $\text{CO}_2$  emissions continue to rise at the current rates (Hilmi et al. 2012). Therefore, absorption of  $\text{CO}_2$  by the oceans at a rate of 25 million tons of  $\text{CO}_2$  per day may contribute to the mitigation of global warming, but evidently this will alter marine carbonate chemistry and cause ocean acidification.

### 1.2. Significance of ocean acidification in the context of climate change

Atmospheric  $\text{CO}_2$  concentrations are expected to reach 467-555 ppm by Year 2050 that would cause surface ocean pH to decline, on average, to 7.8 in Year 2050 (Cooley et al. 2009c). Models of future  $\text{CO}_2$  emissions and ocean uptake suggest that the

atmospheric level of CO<sub>2</sub> would peak shortly after the highest rate of fossil fuel combustion and then subside as the oceans absorb the CO<sub>2</sub>, resulting in an increase in ocean acidity (Ruttmann 2006).

### Past emissions of CO<sub>2</sub> in the atmosphere will continue the acidification process, even it is reduced substantially

The general characteristics of future chemical changes in the ocean from increasing atmospheric CO<sub>2</sub> are highly predictable. Across the range of scenarios used by the Intergovernmental Panel on Climate Change (IPCC), the surface ocean pH is projected to decrease about 0.4 ± 0.1 units by 2100 as compared to pre-industrial conditions (Meehl et al. 2007). Time series records of ocean carbon chemistry over the past 25 years also show clear trends of increasing ocean carbon and decreasing pH in lockstep with increasing atmospheric CO<sub>2</sub> (Bates, 2007; Dore et al. 2009). Over the past two decades, there have been measurable decreases in the weight of calcium carbonate (CaCO<sub>3</sub>) shells of pteropods (marine snail) (Roberts et al. 2008) and foraminifera (unicellular protists) (Moy et al. 2009) in the Southern Ocean, and corals of the Great Barrier Reef, suggesting a recent decline in calcification, a process in which body tissue is hardened by calcium salts or deposits (Cooper et al. 2008). However, more research is needed to confirm whether these issues are due to the effects of ocean acidification alone.

For several reasons, ocean acidification has serious implications for the type of policy interventions required to control climate change. First, since ocean acidification is exclusively driven by CO<sub>2</sub>, as opposed to climate change which is also caused by other greenhouse gases (such as methane, water vapor, nitrous oxide, ozone, chlorofluorocarbons, hydrofluorocarbons, perfluorocarbons and sulfur hexafluoride), the additional cost associated with CO<sub>2</sub> emissions due to ocean acidification changes the trade-offs between the reductions of greenhouse gases. Second, the absorption of CO<sub>2</sub> by the

oceans and the impact of ocean acidification occur over a short time scale, whereas the warming of the atmosphere substantially lags behind the build-up of greenhouse gases in the atmosphere. This changes the dynamics of optimal emission control. Third, the consideration of ocean acidification also has implications for the choice of policy instrument for the control of climate change. Climate change may be countered by geo-engineering, but ocean acidification would continue unabated and may even accelerate if sun-blocking sulphur dioxide particles are used to cool the planet.

### 1.3. Global state of ocean acidification

A continued global CO<sub>2</sub> emissions in line with current trends estimate atmospheric CO<sub>2</sub> levels of over 800 ppm by Year 2100 which may lead to ocean 150% more acidic than it was before industrialization (Makarow et al. 2009). Incidentally, with an emissions peak in 2100, the surface pH of the ocean would not begin to stabilize until 2750 (Ruttmann 2006). The pH level a kilometer below the ocean surface, in this scenario, is projected to continue to fall until 3000 (Ruttmann 2006). A general circulation model, simulating an atmospheric CO<sub>2</sub> concentration of 1,900 ppm in 2300, projects a maximum reduction of surface pH by 0.77 units (Caldeira and Wickett 2003).

Although the geological record shows that episodes of acidification occurred in the past and that the marine organisms adapted to an increased acidity, examples of acidification occurred before tens of thousands of years ago (Erba et al. 2010). Isotope studies of sediments indicate that the current observed and projected rates of acidification are as much as 100 times faster than what occurred in the earlier period (Ruttmann 2006). Furthermore, reviews of the geological record have found that over the past 300 million years, ocean pH was not more than 0.6 units lower than today (Caldeira and Wickett 2003). Thus under future projected levels, a continued acidification will present huge challenge to marine organisms adapted to current oceanic conditions.

## 2. Assessment of Ocean Acidification

### 2.1. Review of methods

Historical records and nearly real-time ocean observation and analysis systems indicate that the oceans of the world are changing (Bindoff et al. 2007; Roemich et al. 2012) with an increase in temperature and acidity, and changes in salinity in all ocean basins (Durack et al. 2012; Durack and Wijffels 2010). These changes are 'over and above' the seasonal, inter-annual, decadal and longer temporal-scale variability caused by climate oscillations such as the El Niño-Southern Oscillation (ENSO), Pacific Decadal Oscillation (PDO), Indian Ocean Dipole (IOD), North Atlantic Oscillation (NAO) and the Southern Annular Mode (SAM).

Multi-model projections, based on IPCC Special Report on Emissions SRES scenarios, indicated that a reduction of pH values between 0.14 and 0.35 units are likely to occur (IPCC 2007; Hobday and Lough 2011; Stock et al. 2011; Brown et al. 2013; Ganachaud et al. 2013). Although there is

uncertainty in the exact nature of these expected changes due to the limitations of the existing global climate models, climate change is likely to impact on the world's oceans by (1) increasing stratification (i.e. layering of water masses), causing reductions in the availability of nutrients to the photic zone; (2) affecting the strength of major ocean currents, causing changes in upwelling, mixing and transport of nutrients; (3) reducing dissolved oxygen levels and (4) causing sea levels to rise (Ganachaud et al. 2011, 2013). In turn, these changes are expected to reduce overall primary productivity (Polovina et al 2011; Howell et al. 2013) and alter the biomass and composition of lower trophic levels (Richardson 2008), leading to changes in the structure of food webs that support marine fisheries (Polovina 2005; Le Borgne et al. 2011). Studies of ocean acidification impacts on marine organisms from different climatological latitudes are shown in Table 1.

### 2.2 Parameters/indicators

Ocean acidification is the change in ocean chemistry driven by the oceanic uptake of chemical inputs

**Table 1.** Impacts of ocean acidification on marine organisms

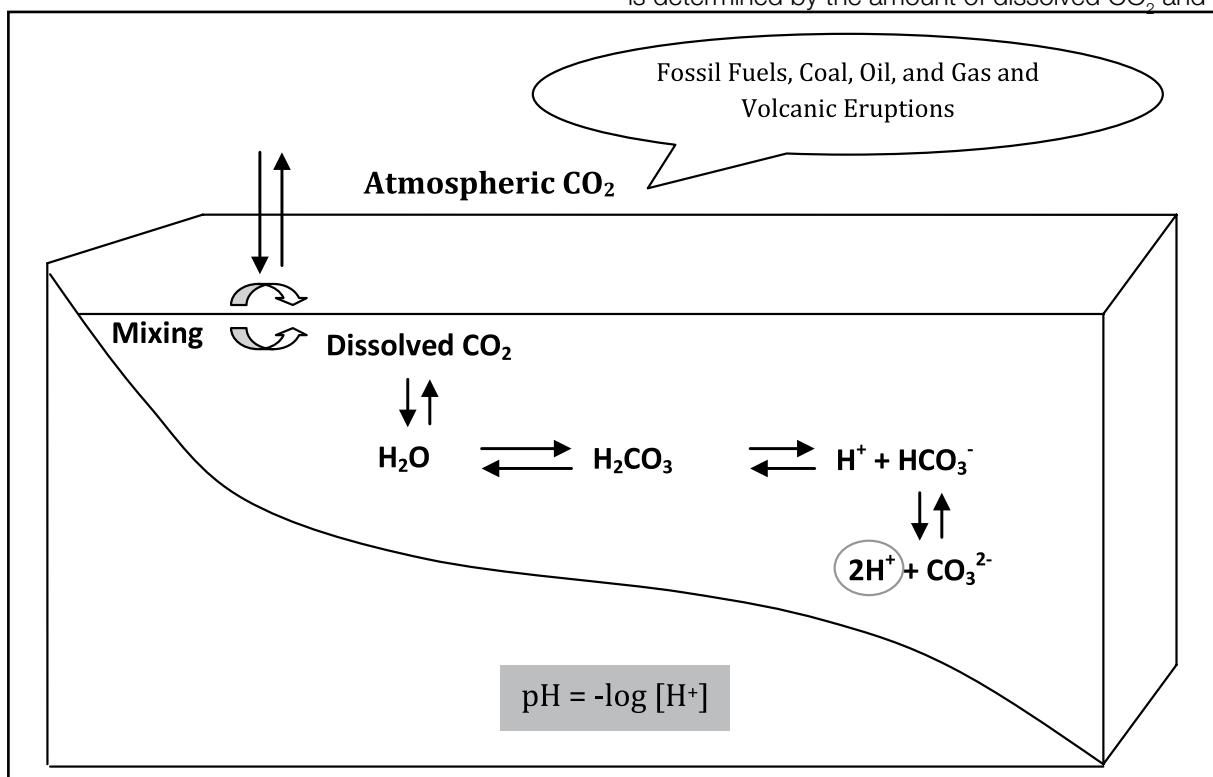
| Parameters   | Species   | References                 |
|--|---|----------------------------|
| pH, CO <sub>3</sub> , pCO <sub>2</sub> and sea surface temperature (SST)                                   | Over-wintering pteropods in the Arctic              | Lischka and Riebesell 2012 |
| Atm. CO <sub>2</sub> and pH  | Cellular and physiological impacts on coral         | Kaniewska et al. 2012      |
| SST, pH, Chl-a, NO <sub>2/3</sub> and PO <sub>4</sub>  | Effect on cyanobacteria                             | Lomas et al. 2012          |
| pCO <sub>2</sub> , CO <sub>2</sub> and pH  | Development of sea-urchin larvae                    | Brennand et al. 2010       |
| pH and CO <sub>2</sub>   | Impact on marine fauna and ecosystem                | Fabry et al. 2008          |
| pCO <sub>2</sub> and CO <sub>2</sub>   | Calcification in oyster larvae                      | Miller et al. 2009         |
| SST, pH, PCO <sub>2</sub> , Ω calcite, Ω aragonite, CO <sub>3</sub> , Alkalinity and Salinity              | Growth and survival of larval shell-fish            | Talmage and Gobler 2010    |
| pCO <sub>2</sub>   | Early life stages of Atlantic herring               | Franke and Clemmesen 2011  |
| CO <sub>2</sub> and pH   | Tropical marine fish                                | Munday et al. 2009         |
| CO <sub>2</sub>  | Death of coral reefs                                | Brander et al. 2009        |
| CO <sub>2</sub>  | Coral reef fishes                                   | Ferrari et al. 2012        |
| Temperature, salinity, DIN, POC, DIP, silicate, Chl-a, δ <sup>13</sup> C <sub>POM</sub> , C:N and Zea/Fuco | Phytoplankton                                       | Biswas et al. 2011         |
| CO <sub>2</sub> and pH   | Dogfish   | Hayashi et al. 2004        |
| CO <sub>2</sub> and pH   | Feed intake of seabass                              | Cecchini et al. 2001       |
| CO <sub>2</sub>  | Larval survivability of red sea-bream               | Ishimatsu et al. 2005      |
| pCO <sub>2</sub> and pH  | Shell dissolution and calcification rates of mussel | Gazeau et al. 2007         |
| pCO <sub>2</sub>   | Oyster calcification rate                           | Michaelidis et al. 2005    |
| Ωarag (the saturation state of aragonite)  | Clam shell dissolution and mortality                | Green et al. 2004          |
| pH   | Occurrence of Jelly fish                            | Attrill et al. 2007        |

from the atmosphere, including carbon, nitrogen, and sulfur compounds. The oceans currently absorb approximately half of the CO<sub>2</sub> produced by burning fossil fuel; put simply, climate change would be far worse if it were not for the oceans. When partial pressure of oceanic CO<sub>2</sub> is less than atmospheres, uptake is occur which reduce water pH and decrease alkalinity. Overall fate of atmospheric CO<sub>2</sub> in oceanic ecosystem is shown in Figure 1.

To a first-order approximation, annual changes in tropical surface ocean chemistry in response to ocean acidification can be estimated from the assumption that surface water CO<sub>2</sub> exists in equilibrium with atmospheric concentrations (Orr et al. 2005; Bates 2007). However, short-term and local variations in both temperature and salinity dramatically affect Ωarag on shorter temporal and

accurate means of tracking ocean acidification for the foreseeable future. However, these in situ measurements are inherently limited in space (time series, moored stations) and/or time (ship surveys). Although current satellite observations do not directly measure changes in ocean carbonate chemistry, they can provide synoptic observations of a range of physical and optical parameters that allow us to model changes in the distribution of carbonate chemistry within the surface ocean where no in situ observations are available. A host of satellite-derived products are becoming available that can offer important information regarding surface ocean thermodynamics, gas exchange, biological modification, and deepwater mixing process.

**pCO<sub>2</sub>:** pCO<sub>2</sub> is the partial pressure of CO<sub>2</sub> in the atmosphere and the ocean. In the ocean, the pCO<sub>2</sub> is determined by the amount of dissolved CO<sub>2</sub> and



**Figure 1.** The fate of atmospheric CO<sub>2</sub> in marine ecosystem

spatial scales. For example, increasing temperature (or decreasing salinity) will increase the carbonate ion concentration due to the temperature (and salinity) dependencies of the equilibrium constants governing the reactions shown in the following equation.



Direct measurements of ocean chemistry through ship surveys, long-term time-series stations, and a growing number of autonomous moored and underway platforms will provide the most

H<sub>2</sub>CO<sub>3</sub>. It varies with alkalinity, latitude, depth, and temperature. Biological processes in the ocean also exert an influence on the pCO<sub>2</sub> in the ocean. As the atmospheric CO<sub>2</sub> concentration increases, the difference in partial pressure between the atmosphere and the seawater results in absorption of anthropogenic CO<sub>2</sub> into the surface layer of the ocean (i.e. Henry's Law). The increase of pCO<sub>2</sub> in the surface ocean profoundly affects the seawater carbonate system. It lowers the pH, decreases the availability of carbonate (CO<sub>3</sub><sup>2-</sup>) ions and lowers the saturation state of the major shell-forming carbonate minerals.

**pH:** When  $\text{CO}_2$  dissolves in seawater, carbonic acid is produced via the reaction. This carbonic acid dissociates in the water, releasing hydrogen ions and bicarbonate. The increase in the hydrogen ion concentration causes an increase in acidity, since acidity is defined by the pH scale, where  $\text{pH} = -\log [\text{H}^+]$  (so, as hydrogen increases, the pH decreases). This log scale means that for every unit decrease on the pH scale (Figure 2), the hydrogen ion concentration has increased 10-fold. Monthly water pH variation in different rivers and estuaries of coastal Bangladesh are shown in Appendix-A.

'saturation horizons' in the oceans, which represent the transition depth between waters that are either under-saturated or over-saturated. Monthly temperature variation in different rivers and estuaries of coastal Bangladesh are shown in Appendix-B.

**Salinity:** Although seawater is naturally salty, but little variations in Sea Surface Salinity (SSS) can have dramatic effects on the water cycle and ocean circulation. SSS tells us about the concentration of dissolved salts in the ocean surface water (i.e. upper few centimetres). Throughout Earth's history, certain

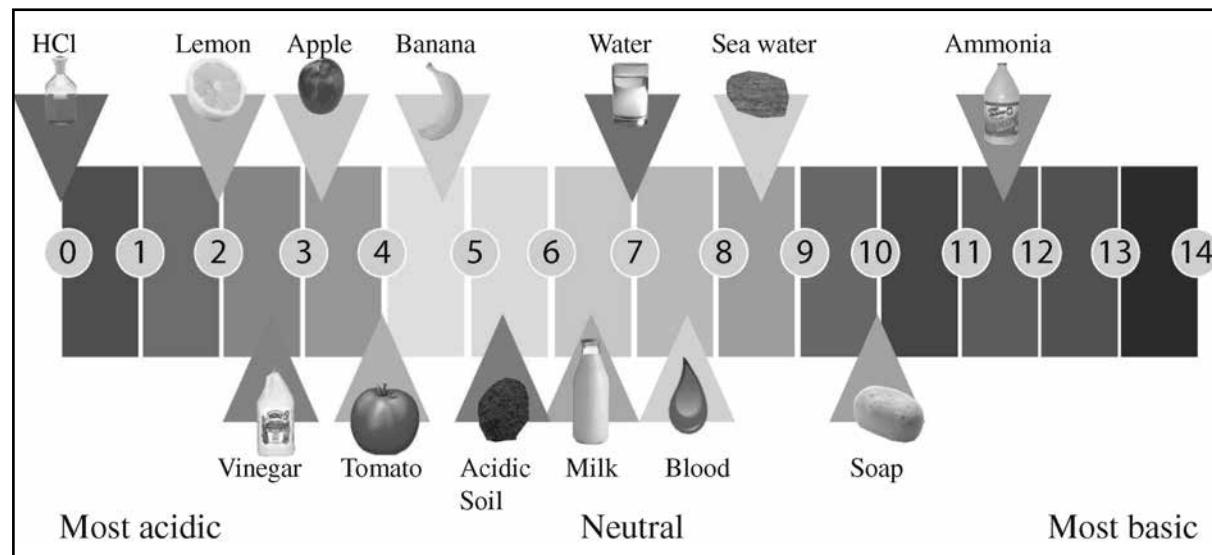


Figure 2. The pH scale with values of some common ingredients (modified from <http://asset.emsolf.com/>)

**Alkalinity:** Alkalinity of sea water measures the ability of water to neutralize acids to the equivalence point of carbonate or bicarbonate. The ability of the ocean to absorb atmospheric  $\text{CO}_2$  over century and longer timescales will depend on the amount of calcium carbonate ( $\text{CaCO}_3$ ) dissolution in the water column or sediments. Oceanographers define a quantity called the alkalinity [Alk] in terms of the concentrations of carbonate ion and bicarbonate ion as shown below:

$$[\text{Alk}] = [\text{HCO}_3^- \text{ (aq)}] + 2 \times [\text{CO}_3^{2-} \text{ (aq)}]$$

The alkalinity is a measure of the ability of seawater to neutralize acidity (protons). Each bicarbonate can neutralize one proton and each carbonate can neutralize two protons.

**Temperature:** The impacts of changes in seawater chemistry are complicated by ocean temperature. The solubility of both calcite and aragonite (which are different forms of calcium carbonate used by marine organisms) is affected by the amount of  $\text{CO}_2$  in seawater, which is partially determined by temperature, for example colder waters naturally hold more  $\text{CO}_2$  and are more acidic than warmer waters. The sum of these differences leads to

processes have served to make the ocean salty. The weathering of rocks delivers minerals, including salt, into the ocean. Evaporation of ocean water and formation of sea ice both increase the salinity of the ocean. However these 'salinity rising' factors are continually counterbalanced by processes that decrease salinity such as the continuous input of fresh water from rivers, precipitation of rain and snow, and melting of ice.

Since 86% of global evaporation and 78% of global precipitation occur over the ocean, SSS is the key variable for understanding how fresh water input and output affects ocean dynamics. By tracking SSS we can directly monitor variations in the water cycle, including land runoff into the ocean, sea ice freezing and melting, and evaporation and precipitation over the oceans. Monthly salinity variation in different rivers and estuaries of coastal Bangladesh are shown in Appendix-C.

**Dissolved oxygen:** The concentration of dissolved oxygen is an important indicator of the health of aquatic ecosystem. Persistently low dissolved oxygen will harm most aquatic life because there will not be enough oxygen for them to use. In some circumstances, water can contain too much oxygen

and is said to be supersaturated with oxygen, which is also dangerous for fish. Supersaturated conditions occur in highly turbulent waters because of aeration, and also on sunny days in waters experiencing algal blooms or with many aquatic plants, because of photosynthesis. The air is one source of dissolved oxygen and aquatic plants are another. The speed at which oxygen from the air enters and mixes through a water depends on the amount of agitation at the water surface, the depth of the water and the rate at which it mixes itself. As water temperature rises, oxygen diffuses out of the water into the atmosphere. Warm or saline water holds less dissolved oxygen than cold water or freshwater. Monthly dissolved oxygen variation in different rivers and estuaries of coastal Bangladesh are shown in Appendix-D.

### 2.3. Data sources and analysis

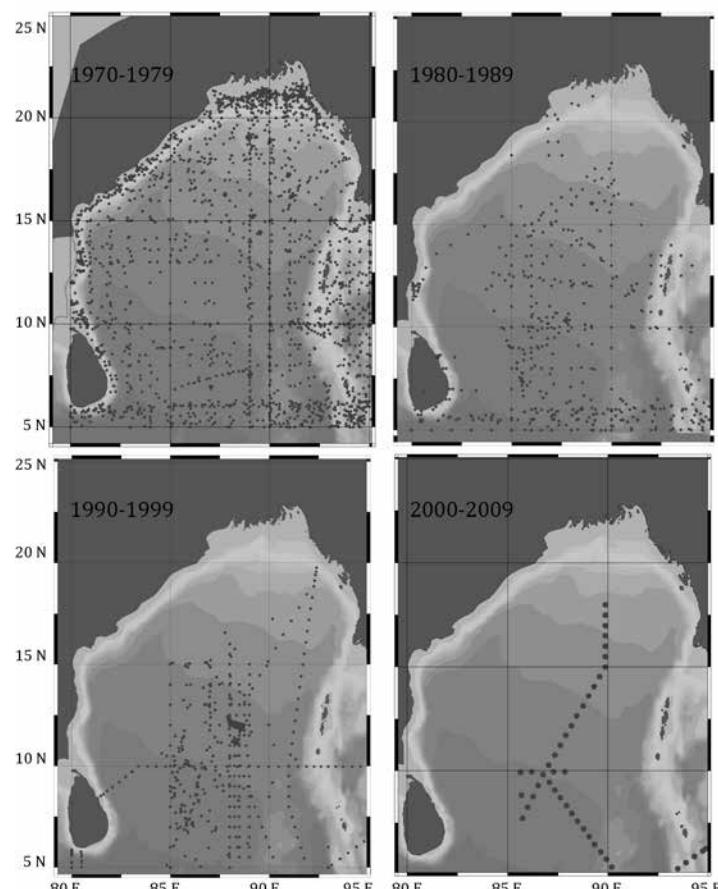
Present initiative aims to give an overview on the acidification of the Bay of Bengal with an emphasis on the territorial waters of Bangladesh. Data on ocean acidification in Bangladesh are still scarce.

The long term yearly average data on sea surface temperature (SST) and pH were collected from the National Oceanographic Data Center (NODC) to develop a scenario of ocean acidification in the Bay of Bengal of Bangladesh.

The NODC provides scientific and public stewardship for national and international marine environmental and ecosystem data. Moreover, the National Coastal Data Development Center (NCDDC) and NOAA Central Library, with its regional branches, are providing access to the world's most comprehensive sources of oceanographic data in an integrated way. The NODC maintains and updates a national ocean archive with environmental data acquired from domestic and foreign activities and produces research data which help monitor global environmental changes. These data include physical, biological and chemical measurements derived from *in situ* oceanographic observations, satellite remote sensing of the oceans, and ocean model simulations. The NODC manages and operates the World Data Center (WDC) for Oceanography. Also, the NODC represents NOAA/NESDIS to the general public,

**Table 2.** Oceanic environmental data of the Bay of Bengal from NODC, NOAA

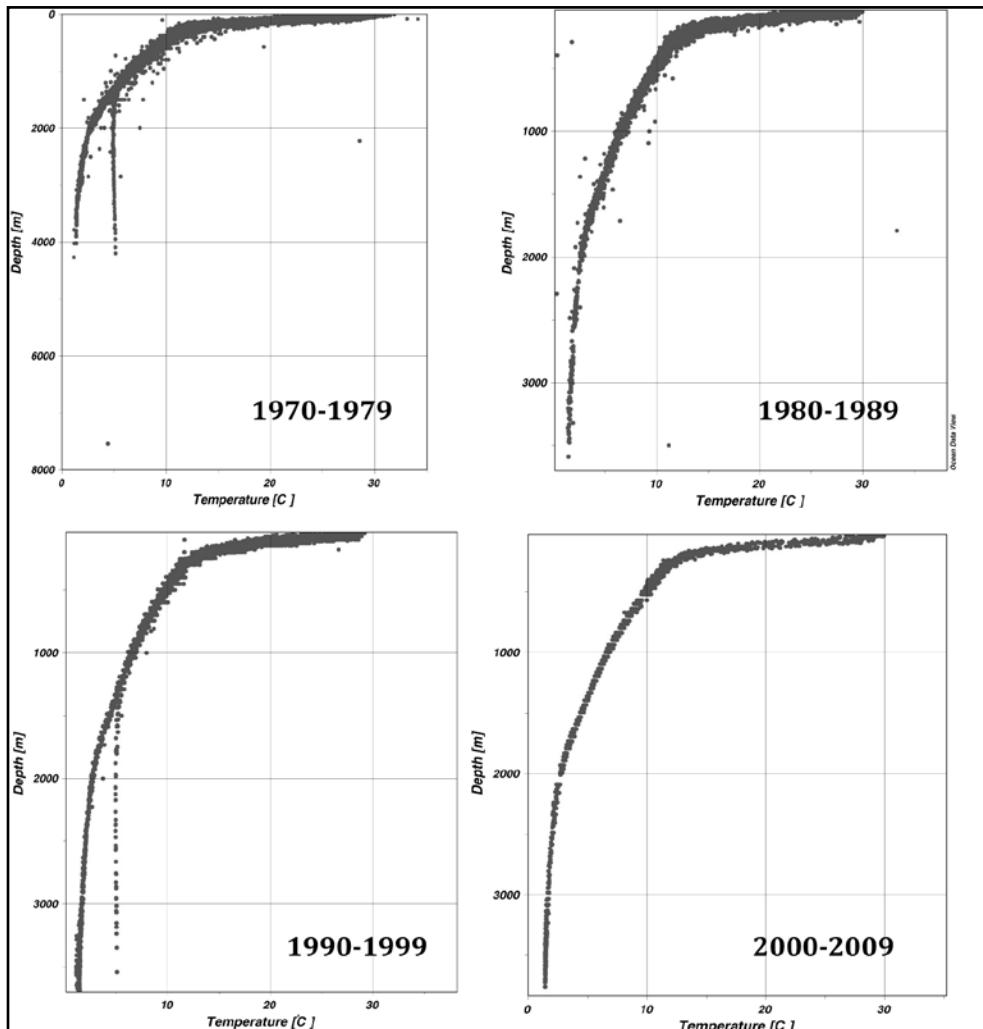
| Type           | Parameters | Archive | Temporal resolution | Spatial resolution | Format |
|----------------|------------|---------|---------------------|--------------------|--------|
| <i>In situ</i> | SST, pH    | WOD09   | 1963-1990           | 5°-25°N, 80°-95°E  | OSD    |



**Figure 3.** The geographical distribution of data points sampled in the Bay of Bengal between 1970-2009 (source: WOD09; <http://www.nodc.noaa.gov/OC5/WOA09/>)

government agencies, private institutions, foreign governments, international organizations and the private sector on matters involving oceanographic data (<http://www.nodc.noaa.gov/OC5/> WOA09).

The SST and pH for the year 1963-1990 were derived from NODC as OSD format (Table 2) which is then visualized in Ocean Data View (ODV, version 4) software. After visualization, data were exported to spreadsheet file with their temporal (Figure 3) and spatial distribution, and subsequently used in ArcGIS for further analysis.



**Figure 4.** The vertical distribution of temperature in the Bay of Bengal during 1970-2009.

#### 2.4. Trends analysis

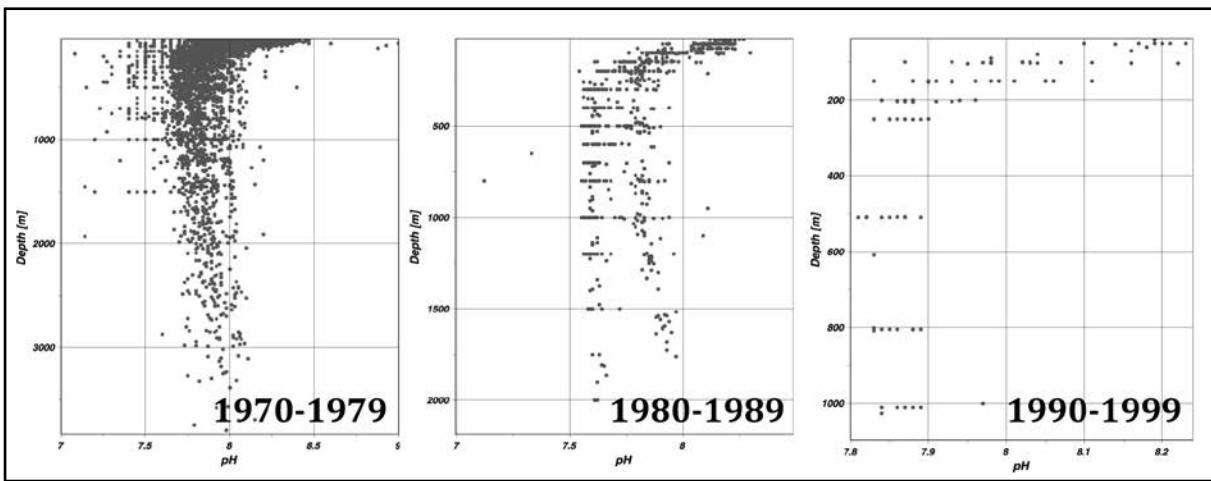
Ongoing changes in surface pH, pCO<sub>2</sub> and temperature saturation states are measurable at time-series stations, which tell us about trends as well as daily, seasonal, inter-annual and decadal variability. The measured surface ocean trends agree with what is expected from the measured atmospheric increase in carbon dioxide, assuming the air-sea CO<sub>2</sub> in equilibrium. Yet the high-precision

chemical measurements are needed over decadal time scales which are only available at a very limited number of stations confined mostly to the southern tropics of the Bay of Bengal.

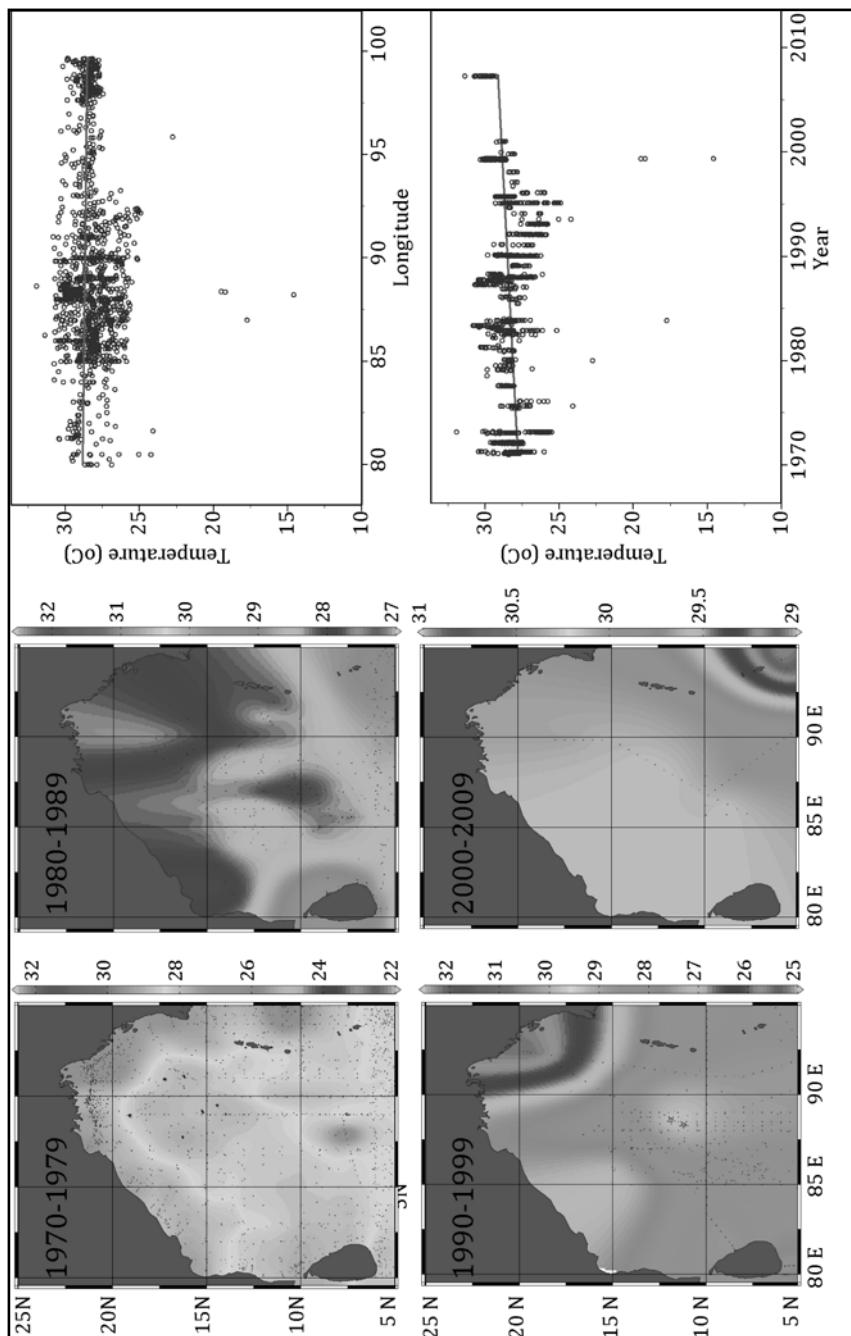
There is also a critical need for well-developed spatial and temporal models that give accurate present-day and future estimates of aragonite and calcite saturation states in the coastal zones. The shallow continental shelves are the most biologically productive areas in the Bay of Bengal and are home to the majority of the marine fisheries, but accurate

saturation state data are not available for analysis and prediction. Future initiative should focus more on measured variability, including oscillations on short-term biologically relevant to decadal time scales, to characterize driving mechanisms and to better evaluate models that are used to make future projections (Hofmann et al. 2011; Shaw et al. 2012; Doney et al. 2009). Time-series stations are complemented by large-scale sections in the Bay of Bengal.

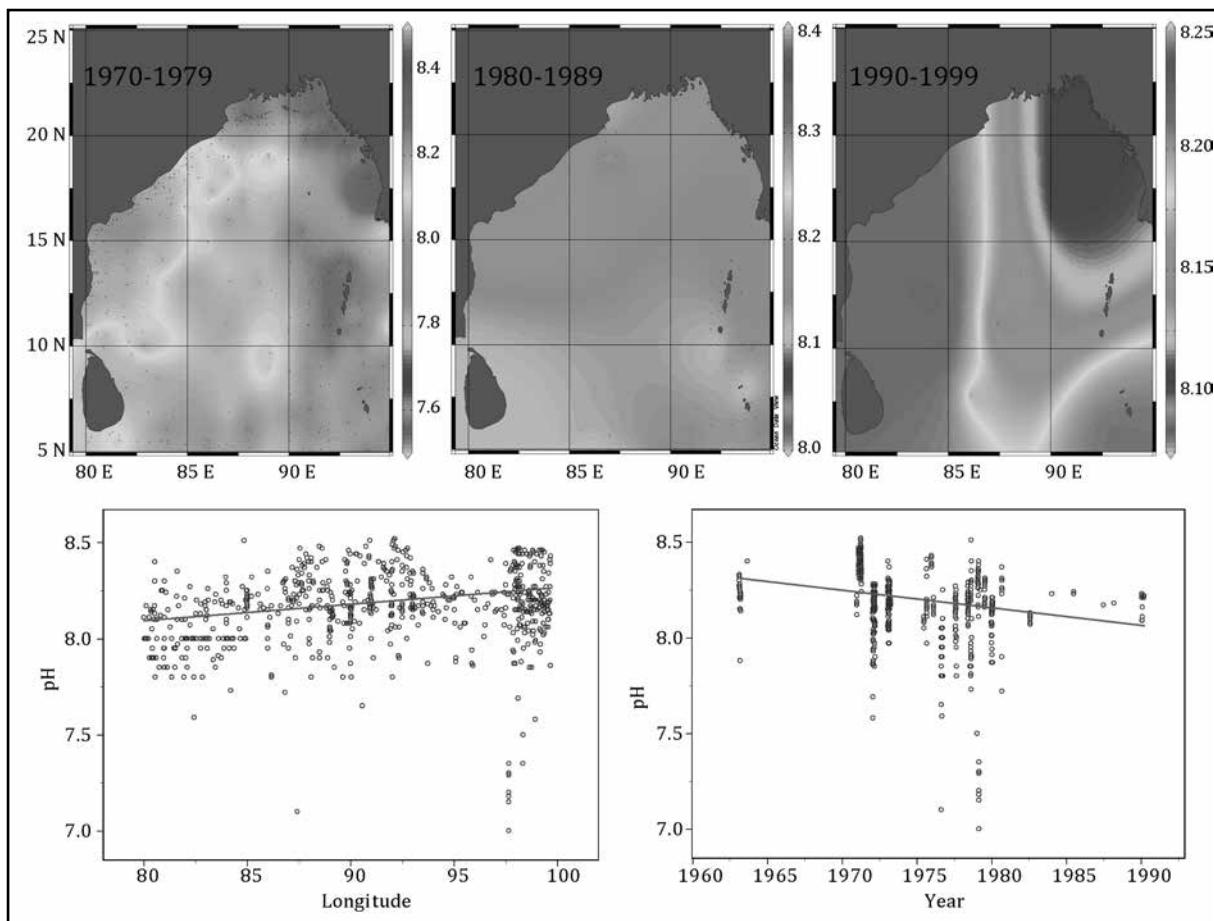
Vertical distribution of SST (Figure 4) showed that, temperature varies both with depth and time.



**Figure 5.** The vertical distribution of pH in the Bay of Bengal during 1970-1999



**Figure 6.** Showing the long-term sea surface temperature (SST) variation in the Bay of Bengal, presented with longitude and yearly trends



**Figure 7.** Showing the long-term water pH variation in the Bay of Bengal, presented with longitude and yearly trends

Average temperature was found  $28.39^{\circ}\text{C}$ ,  $28.86^{\circ}\text{C}$ ,  $29.31^{\circ}\text{C}$  and  $29.39^{\circ}\text{C}$  during 1970-1979, 1980-1989, 1990-1999 and 2000-2010, respectively. The vertical distribution of pH is shown in Figure 5, and the average pH was found 7.8, 7.6 and 7.3 during 1970-1979, 1980-1989 and 1990-1999, respectively.

The trends of forty years (1970-2010) of SST are shown in Figure 6. The maximum sea surface temperature was  $31.91^{\circ}\text{C}$  during 1973 (April) and Minimum  $14.35^{\circ}\text{C}$  during 1999 (December) with an average  $28.39^{\circ}\text{C}$ . Trends analysis showed that, SST is increasing over the period. There is also an increased trend found at higher latitude.

The trends of pH for the thirty years (1970-1999) are shown in Figure 7. The maximum pH was found 8.52 during 1971 and Minimum 7.0 during 1979 with an average 8.15. Trend analysis showed that, pH is decreasing over the period with an increased trends at higher latitude.

## 2.5. Facilities for measuring future ocean acidification in bangladesh

The Institute of Marine Sciences and Fisheries (IMSF) of Chittagong University is engaged in analyzing water, sediment and biological samples

from the river, estuarine and near shore ecosystems of the Bay of Bengal for academic and research purposes. The IMSF has covered several important coastal and marine environments such as the Saint Martin's Island, Naaf River, Moheshkhali Channel, Bakhkhali River, Kutubdia Channel, Karnafully River, Sandwip Channel, Meghna estuary and Sunderban mangrove ecosystems. Moreover, the laboratory facilities and qualified researchers of IMSF are active in primary data generation for the government organizations, donors and private entrepreneurs. Moreover, the Bangladesh Navy (BN), Bangladesh Inland Water Transport Authority (BIWTA), Bangladesh Coast Guard and Chittagong Port Authority are playing important role in collection and monitoring of oceanographic data from the shelf water of Bangladesh in the Bay of Bengal (Figure 8). Besides, the role of Space Research and Remote Sensing Organization (SPARRSO), Geological Survey of Bangladesh, Survey of Bangladesh and Meteorological Department are significant in this discipline. The Department of Fisheries and Marine Science at Noakhali Science and Technology University, Fisheries and Marine Resource Technology Discipline at Khulna University and Marine Fisheries Academy at Chittagong are the emerging organizations in ocean research. Geo-spatial distribution of water salinity and dissolved

oxygen in 10 m depth coastal water of Bangladesh have developed using IMSF research data (Figure 9).

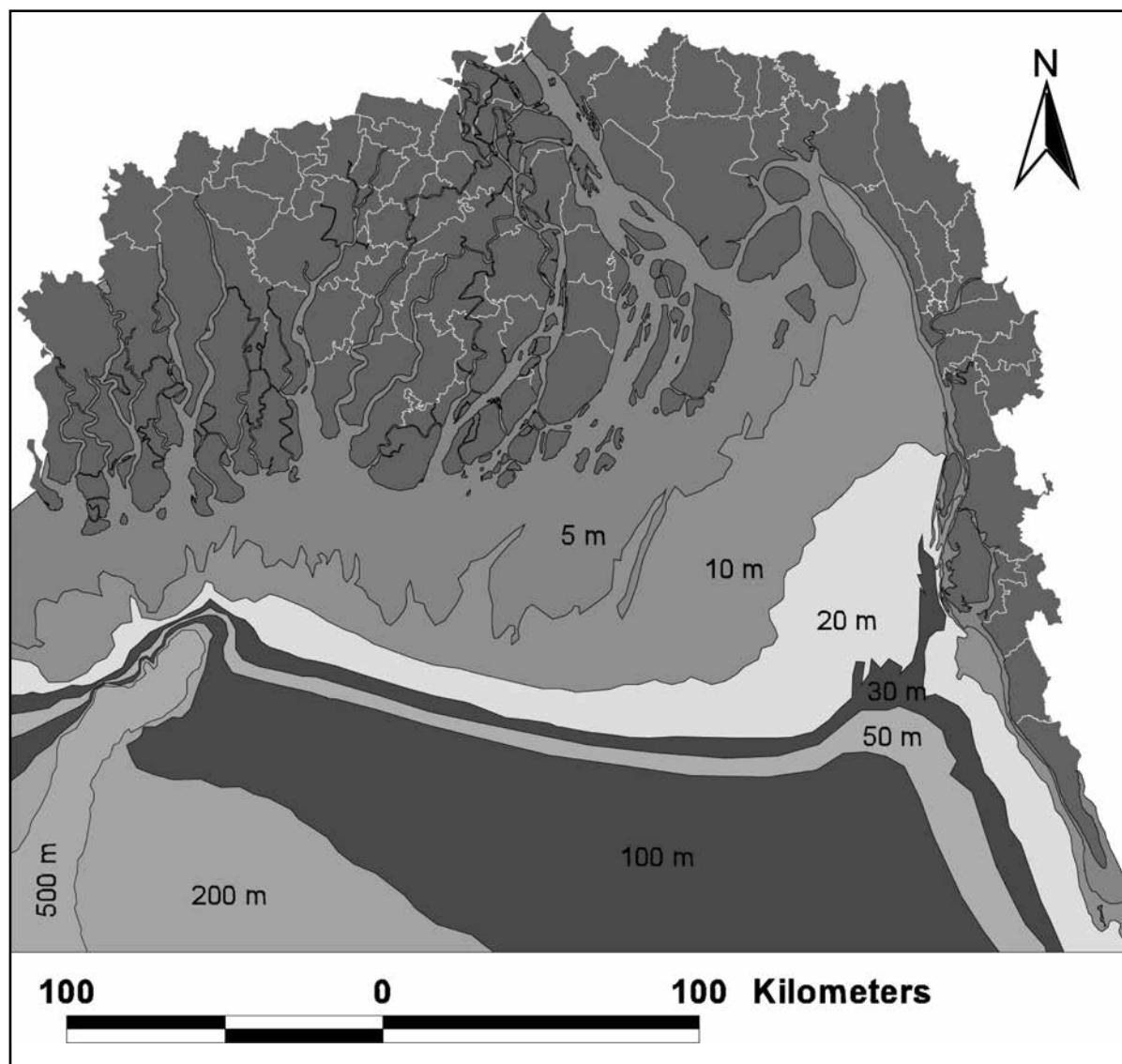


Figure 8. Coastal zone of Bangladesh with shelf water in the Bay of Bengal

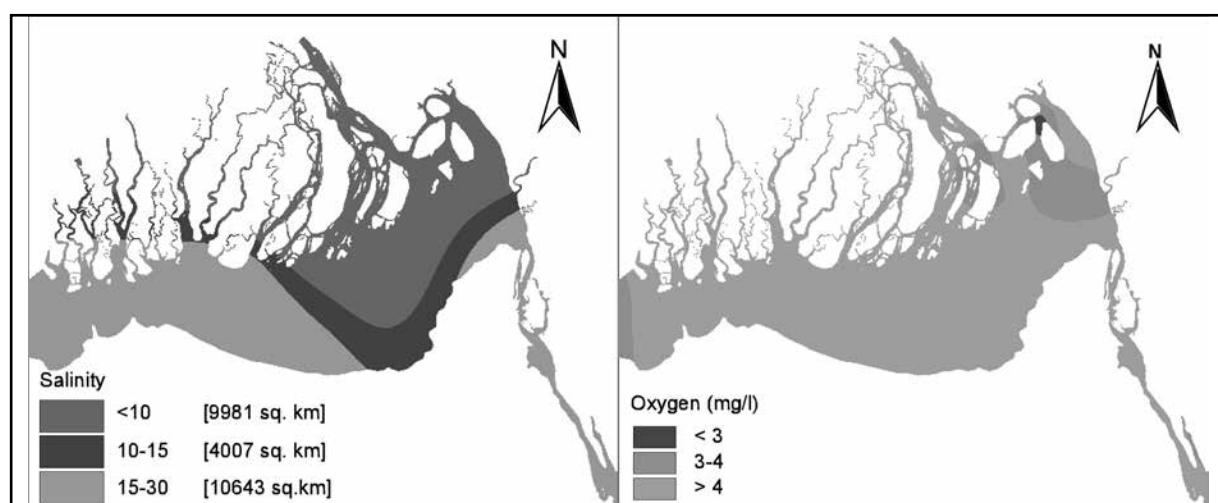


Figure 9. Distribution of water salinity and dissolved oxygen in 10 m depth coastal water of Bangladesh

### **3. Effects of Ocean Acidification on Marine Environment**

#### **3.1. Effects on biodiversity and food chain**

Ocean acidification could affect marine food chain and lead to substantial changes in oceanic biodiversity which may threaten protein supply and food security for millions of people as well as the multi-billion dollar fishing industry. Phytoplankton are the base of marine food chain and unlike land plants, most marine phytoplankton are thought to have mechanisms to actively concentrate CO<sub>2</sub> which will impact phytoplankton species diversity in the natural environment and primary production, food web structure and marine biogeochemical cycles. Islam (1981) found 107 species of phytoplankton in the Karnaphuli estuary. Sayedullah (2003) identified 55 genera of phytoplankton from the Meghna estuary, of which 23 genera belongs to 13 family of Bacillariophyta, 19 genera belongs to 8 family of Chlorophyta and the remaining 13 genera belongs to 5 family of Cynophyta. Incidentally, the jellyfish blooms have increased over the last several decades (Purcell et al. 2007). Attrill et al. (2007) reported a significant correlation of jellyfish occurrence in the North Sea from 1971 to 1995 with a decreased pH (from 8.3 to 8.1) of surface waters. The jellyfish are both predators and potential competitors of fish and may substantially affect pelagic and coastal ecosystems. Therefore, substantial changes to species diversity and abundances, food-web dynamics, and other fundamental ecological processes could occur in the marine environment.

Ocean acidification may directly affect some marine species that supply many services to humans. For instance, molluscs and crustaceans support valuable commercial and recreational fisheries (Cooley and Doney, 2009a), and coral reef ecosystems support a variety of subsistence, recreational, and commercial fisheries worldwide (Bryant et al. 1998). Calcifiers are also provide pearls, shells, and coral pieces that are used for jewellery making. However, the declining populations of small calcifiers, like planktonic pteropods, and molluscs and corals larvae may increase competition among aquatic predators and at the same time may decrease the abundance/diversity of many commercial species. If reduced calcification decreases the fitness or survivorship of calcifying organisms, then some planktonic calcareous species may undergo shifts in their distributions as the inorganic carbon chemistry of seawater changes. The calcifying species are CO<sub>2</sub> sensitive and could potentially be replaced by non-calcifying species and/or by those species not sensitive to elevated pCO<sub>2</sub>. Acidification can

affect many other marine organisms, but especially impacts those that build their shells and skeletons from calcium carbonate, such as corals, oysters, clams, mussels, snails, and the phytoplankton and zooplankton which form the base of the marine food web.

The effects of ocean acidification in Bangladesh are analyzed in this report by using the data of Islam (2005), Kamal (2000), Chowdhury et al. (2009), Samsuzzaman (2009), Talukder (2010), Nabi et al. (2011) and Hossain et al. (2013). Historical trends analysis of water temperature and pH (Figure 10) showed that, water temperature has an increasing trend, whereas water pH has a decreasing trend, although these changes are not statistically significant. The effects of historical changes in water temperature and pH on fisheries biodiversity is shown in Figure 11. A reduction of pH value has negative effects, and a positive higher pH value has positive effects on fisheries biodiversity. Whereas positive temperature implies increase in water temperature which has negative impact and a negative value indicate reverse effect on fisheries biodiversity. Hussain (1971) reported 475 finfish species from the coastal and marine waters of the Bay of Bengal, but Islam et al. (1993) reported 185 species of finfish and shellfish. Chwdhury et al. (2011) recorded 98 finfish, 23 species of shrimps and prawns, 13 crabs, 11 species of molluscs, 3 echinoderms and 4 other crustaceans from the Naaf River Estuary. Hossain et al. (2013) identified 53 finfish species from the Meghna River estuary.



A bleached Gorgonian sea fan in the Bay of Bengal.  
©Mohammad Arju/ Save Our Sea

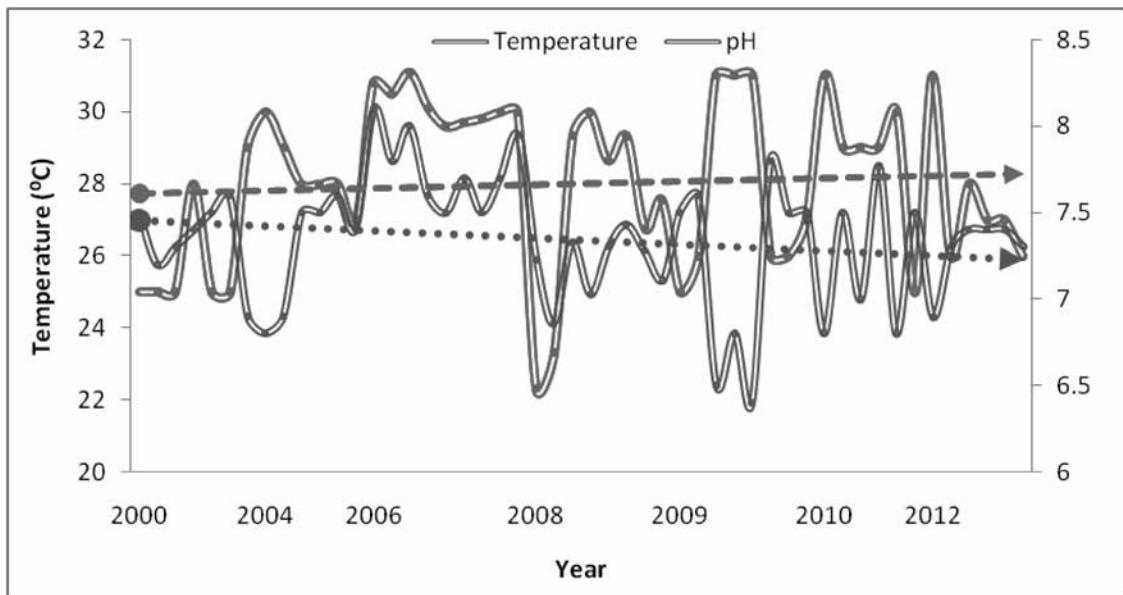


Figure 10. The historical trends of water temperature and pH in the coastal waters of Bangladesh

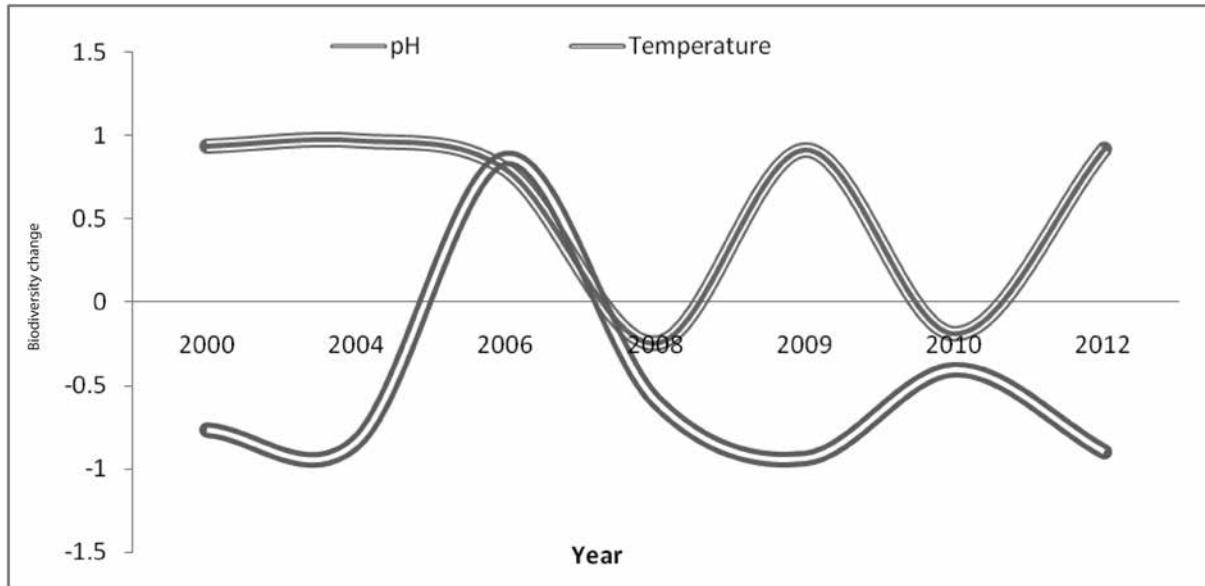


Figure 11. The effects of water temperature and pH on the fisheries biodiversity

### 3.2. Effects on coral reefs

Although coral reefs occupy only 0.2% area of the marine environment, they are the most productive and biologically diverse marine ecosystems (Reaka-Kudla 2001) and are thought to provide important habitat, feeding grounds and nursery supports for many deep-water organisms, including several commercially important fish species (Mortensen 2000; Fossa et al. 2002; Husebø et al. 2002; Roberts et al. 2006). In the coming decades, the net effect of ocean acidification on coral reef ecosystems will probably be negative as many reef-building marine calcifiers will be heavily impacted by the combined effects of increasing sea-surface temperatures (i.e. coral bleaching) and decreasing carbonate saturation states of surface waters (Guinotte et al.

2003; Buddemeier et al. 2004). Already an estimated 19% of the world's coral reefs have been lost and a further 35% are seriously threatened (Wilkinson 2008). As a result, one-third of all reef-building corals are considered to be at risk of extinction (Carpenter et al. 2008). A small increases in sea temperature (i.e. 1-2°C) will breakdown the relationship between host corals and their symbiotic dinoflagellate algae (such as zooxanthellae), on which they rely for energy and growth (Muscatine 1973; Trench 1979). The vulnerability of different taxa to acidification depends on the form of carbonate that they secrete. The coralline algae, which are essential for cementing coral rubble into solid reef, forms a critical habitat for the early life stages of many organisms, including corals, and secrete high magnesium calcite and thus are particularly vulnerable to acidification (Kuffner et

al. 2007). Mass bleaching of corals (i.e. bleaching of multiple species on an ecologically significant scale) was first observed in the late 1970s and was correlated with abnormally high sea temperatures, especially induced by *El Niño* events recurring every 4-7 years (Glynn, 1991) superimposed on generally elevated sea temperatures due to global warming (Hoegh-Guldberg, 1999).

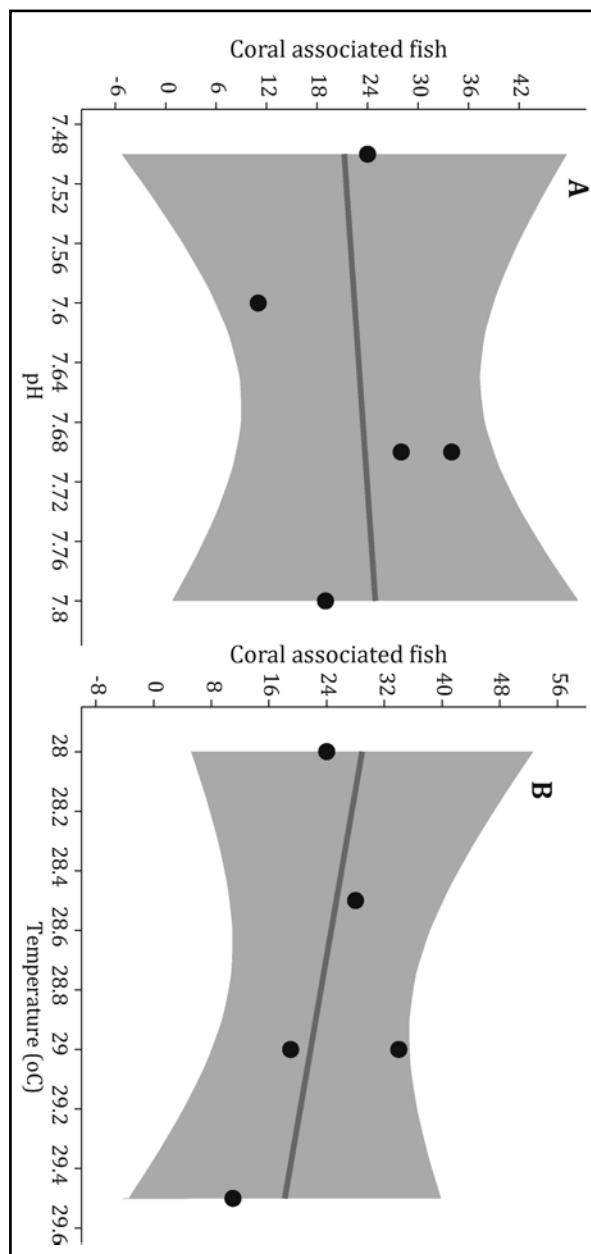
The Saint Martin's Island is the only coral bearing island of Bangladesh in the Bay of Bengal with a total area of about 7.5 km<sup>2</sup>. Haider (1992) recorded four coral species of the genus *Acropora* (*A. pulchra*, *A. horrida*, *A. humilis* and *A. variabilis*) from neritic waters of the island. The reef is subject to greater scouring and supports rich stands of macro algae, sometime seasonally, as well as some well-developed coral communities. The hard substrates and reef are important for artisanal fisheries and they can act as important foci for commercially important species. The island is an important nesting area for marine turtles, and a wintering area for migratory shore birds (Hossain 2001). Ocean acidification due increased CO<sub>2</sub> level will decrease pH, which will results coral bleaching, and down growth and decrease coral species diversity. Moreover, when CO<sub>2</sub> levels go up, the calcification of coralline algae will probably be completely inhibited. The feeding and spawning areas of reef inhabiting fishes will be reduced which ultimately lead to dramatic loss of fisheries biodiversity.

**Table 3.** The records of coral species at Saint Martin's Island, Bangladesh

| Year | Coral species | Temperature (°C) | pH      | Reference           |
|------|---------------|------------------|---------|---------------------|
| 1997 | 66            | 28-29            | -       | Tomasik 1997        |
| 2008 | 40            | 27.5-29.5        | 7.5-7.8 | Sultana et al. 2008 |

Tomasik (1997) recorded 66 coral species from the Saint Martin's Island and found the water temperature between 28 and 29°C. Sultana et al. (2008) found only 40 coral species in the region, and the water temperature was 27.5-29.5°C with pH values of 7.5-7.8. These two comparisons indicate declination of corals by 26 species in 11 years which is alarming for the coral biodiversity in Bangladesh (Table 3). The specific effect of water temperature and pH changes on coral associated fishes is shown in Figure 12. The water temperature and pH data were reproduced from Sultana et al. (2008). All data were tested for normality and effects were calculated at 95% confidence intervals. The dashed line shows the regression line, whereas the shaded lines represent the 95% confidence intervals. Findings show that, with an increase in pH, the

coral associated fish species number is increasing, alternatively a decrease in pH will inversely affect the fish diversity (Figure 12A). Also, an increase water temperature decrease the number of fish species (Figure 12B).



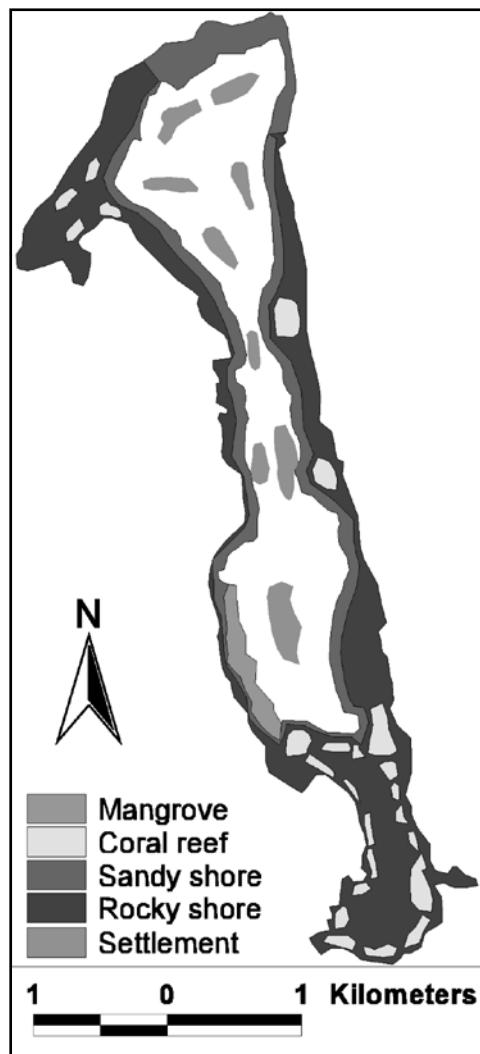
**Figure 12.** The effects of water pH and temperature on coral associated fish diversity in the Saint Martin's Island, Bangladesh

The northeaster part of the Saint Martin's Island is a sandy beach extending for about 5 km in a single stretch and averaging about 100 m in width at high tide. During low tide the rocky zone extend 0.8-1.0 km in the south-western coast, while 2.0-2.5 km in the south-eastern coast with 200-500 m width. Most of the sessile mollusc species, particularly the oysters, green mussels and barnacles are abundant in the rocky zone. The living corals are present in the rock pools that remain submerged during high tide.

Large coral fauna were observed in the rock pools of about 0.5-2.0 m depth surrounding the island (Hossain et al. 2007). Dense coral populations were observed in the east coast of Galachipa, southeast coast of Daskhinpara, rocky area of Siradia and southwest coast of Daskhinpara (Figure 13). Macro algae are produced naturally in the intertidal and subtidal zone surrounding the island. Huge quantity of uprooted and torn parts of algae used to be on the beach of east, west and south coast of the island due to tide, current and wave action of the Bay of Bengal.

Coral reefs function as habitat for commercial fish stocks, natural barriers protecting coastlines, and catalysts for recreation and tourism (Makarow et al. 2009). The value of coral reefs include direct use values such as coral mining and recreational opportunities, as well as indirect use values including coastal protection, and habitat and nursery functions for commercial and recreational fisheries (Brander et al. 2009). In addition, there is welfare value of preserving a unique diverse natural ecosystem (Brander et al. 2009). The global economic value of these reef services has been estimated to be \$30 billion per year based on a value of \$100,000 per km<sup>2</sup> per year (Cooley et al. 2009). Given their open access nature and role as a public good, reefs are usually undervalued in decisions that relate to their use and conservation (Brander et al. 2009).

The global economic value for only the shoreline protection provided by coral reefs is estimated to be \$9 billion per year (Cesar et al. 2003) (Table 4). If reefs are compromised by acidification, the ecosystem services of coastline protection and shoreline



**Figure 13.** Extent and distribution of rocky shore, sandy shore and coral reef in the saint Martin's Island (Source: Hossain et al. 2008)

**Table 4.** Examples of ecosystem services provided by coral reefs (source: Cooley et al. 2009)

| Nation  | Year     | Value (US \$)                | Source             |
|---|----------|------------------------------|--------------------|
| Shoreline protection by coral reefs             |          |                              |                    |
| St. Lucia                                       | 2006     | \$28–50 million              | Burke et al. 2008  |
| Tobago  | 2006     | \$18–33 million              | Burke et al. 2008  |
| Belize  | 2007     | \$120–180 million            | Cooper et al. 2008 |
| Shoreline protection by mangroves               |          |                              |                    |
| Belize  | 2007     | \$120–180 million            | Cooper et al. 2008 |
| Direct economic impacts of coral reef tourism   |          |                              |                    |
| Tobago  | 2006     | \$43.5 million (15% of GDP ) | Burke et al. 2008  |
| St. Lucia                                       | 2006     | \$91.6 million (11% of GDP ) | Burke et al. 2008  |
| Indirect economic impacts of coral reef tourism |          |                              |                    |
| Trinidad & Tobago                               | 2006     | \$58–86 million              | Burke et al. 2008  |
| St. Lucia                                       | 2006     | \$68–102 million             | Burke et al. 2008  |
| Belize  | Annually | \$26–69 million              | Cooper et al. 2008 |
| Reef tourism direct and indirect impacts        |          |                              |                    |
| Belize  | 2007     | \$150–196 million            | Cooper et al. 2008 |

stabilization would be lost (Cooley et al. 2009). This would result in greater economic losses from storms and require greater investment in seawalls and other fortifications to protect property and human lives (Cooley et al. 2009). Recent estimates found that projected damage to coral reefs due to acidification would result in a 0.18% loss in global gross domestic production in Year 2100 (Makarow et al. 2009). In 2000, the total area of coral reefs was 307,000 km<sup>2</sup>, globally (Brander et al. 2009). Projections of lost coral reef area in Year 2100 due to acidification only, excluding degradation due to warming, sea level rise, and other pollution, range from 16% (30,000 km<sup>2</sup>) to 27% (65,000 km<sup>2</sup>) (Brander et al. 2009). These projected losses in coral reef area translate to economic losses of as much as \$870 billion in Year 2100 (Brander et al. 2009).

### 3.3. Effects on shellfish

Increased acidity in ocean water is affecting the size and weight of shells and skeletons, and similar sort of problem is widespread across marine species. Different species-specific responses among molluscs have been observed, but majority of mollusc due to ocean acidification ate neutral to negative (Cooley et al. 2012). By examining four different types of marine animals, i.e. clams, sea snails, lamp shells and sea urchins, scientist found that as the availability of calcium carbonate decreases due to increased ocean acidification, thus skeletons get lighter and account for a smaller part of the animal's weight. The effect occurs consistently among the mentioned species suggesting that the effect is widespread across marine organisms, and

that increasing ocean acidification will progressively reduce the availability of calcium carbonate.

Gazeau et al. 2007; Miller et al. 2009 also found decrease shell thickness and rate of calcification in shellfish as a resultant of ocean acidification. The rise in acidity makes seawater corrosive to many marine organisms, especially those that require calcium in their 'formative' growth stages, such as shellfish larvae. The shellfish industry, therefore, is the first to bear the burden of the chemical changes taking place in the oceans' pH levels, as they require calcium carbonate for external shell formation. Molluscs are important aquaculture species, and provide a small but significant protein source for the humans (Turley and Boot 2010). Recently, molluscs production has experienced a sharp drop due decrease in ocean pH values (Berge et al. 2006). Physiological impacts induced by lowered pH have the potential to affect a shellfish at any stage in its life cycle; however, adults tend to have more protection as well as better mechanisms to deal with a fluctuating environment with early life stages tending to be more vulnerable. In nature, large numbers of larvae are often produced by the organisms which have high rates of larval mortality, for example, coastal estuarine bivalves experience more than 98% mortality during settlement (Turley et al. 2006). The oyster (Ridgwell and Schmidat 2010), echinoderm (Fabry et al. 2008) and fish larvae (Guittuso and Hansson 2011) as well as barnacle, tube worm and copepod eggs have all been found to either be increasingly malformed or have slower rates of development at higher CO<sub>2</sub>. Moreover, barnacle settlement has also been affected due to elevated CO<sub>2</sub> (Hendriks et al.

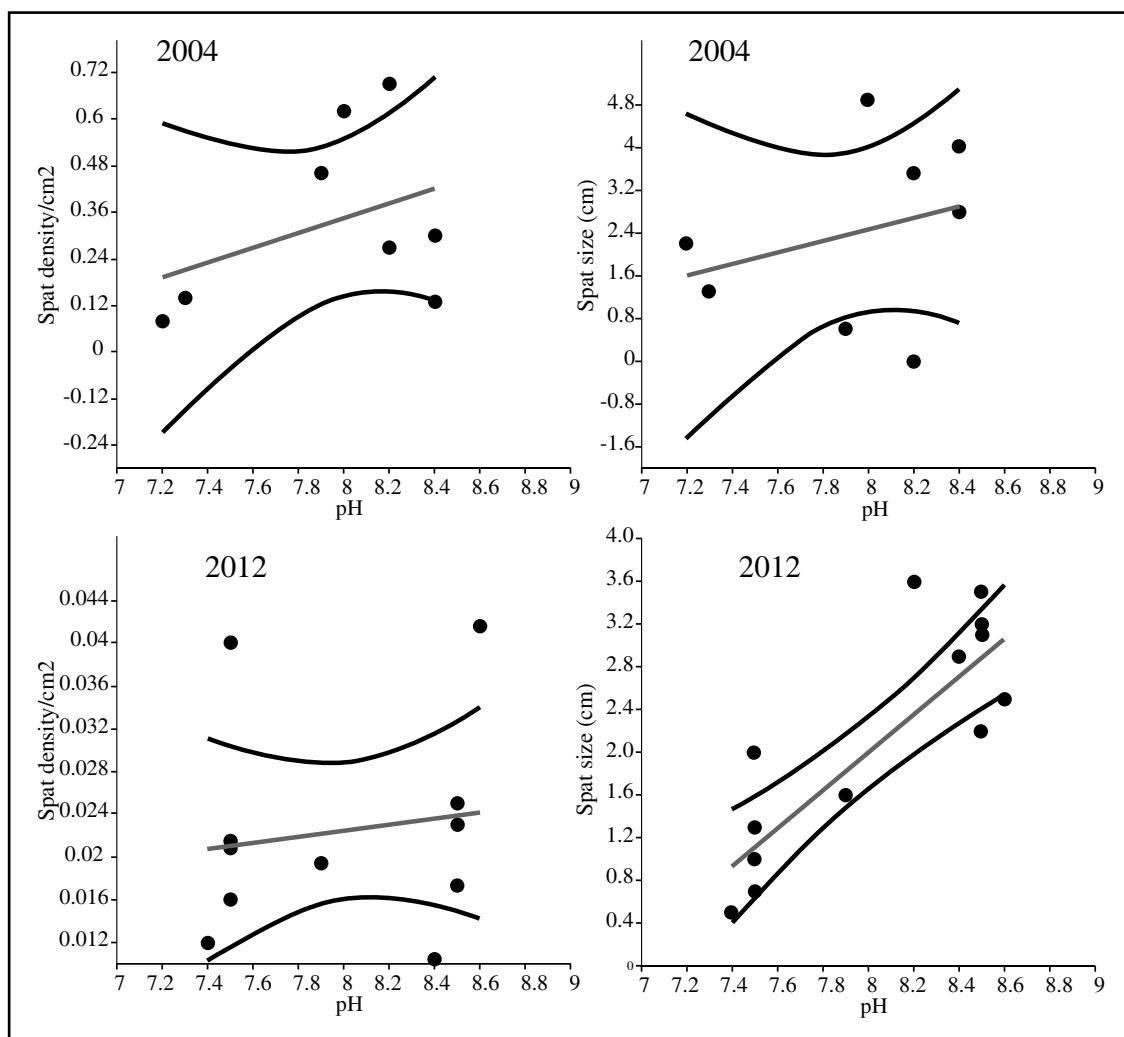
**Table 5.** The effects of ocean acidification on different mollusc species, negative effects (decreasing trend) are denoted by minus sign

| Species                          | Experimental organ/stage          | References                   |
|----------------------------------|-----------------------------------|------------------------------|
| <i>Mercenaria mercenaria</i>     |                                   | Green et al. 2004            |
| <i>Pinctada fucada</i>           | Shell (-)                         | Knutzen 1981                 |
| <i>Tivela stultorum</i>          | Fertilization rate (-)            | Alvarado-Alvarez et al. 1996 |
| <i>Placopecten magellanicus</i>  | Fertilization (-), Embryo (-)     | Desrosiers et al. 1996       |
| <i>Mytilus galloprovincialis</i> | Metabolism (-)                    | Michaelidis et al. 2005      |
| <i>Mytilus edulis</i>            | Calcification (-)                 | Lindinger et al. 1984        |
| <i>Haliotis laevigata</i>        |                                   | Harris et al. 1999           |
| <i>Crassostrea virginica</i>     | Calcification (-)                 | Miller et al. 2009           |
| <i>Mercenaria mercenaria</i>     |                                   | Talmage and Gobler 2009      |
| <i>Crassostrea virginica</i>     | Calcification (-), Metabolism (-) | Ries et al. 2009             |
| <i>Mercenaria mercenaria</i>     | Calcification (-)                 | Beesley et al. 2008          |
| <i>Mya arenaria</i>              | Calcification (-)                 | Gazeau et al. 2007           |
| <i>Mytilus galloprovincialis</i> | Shell (-)                         | Michaelidis et al. 2005      |
| <i>Saccostrea glomerata</i>      |                                   | Parker et al. 2010           |
| <i>Strombus alatus</i>           | Calcification (-)                 | Ries et al. 2009             |

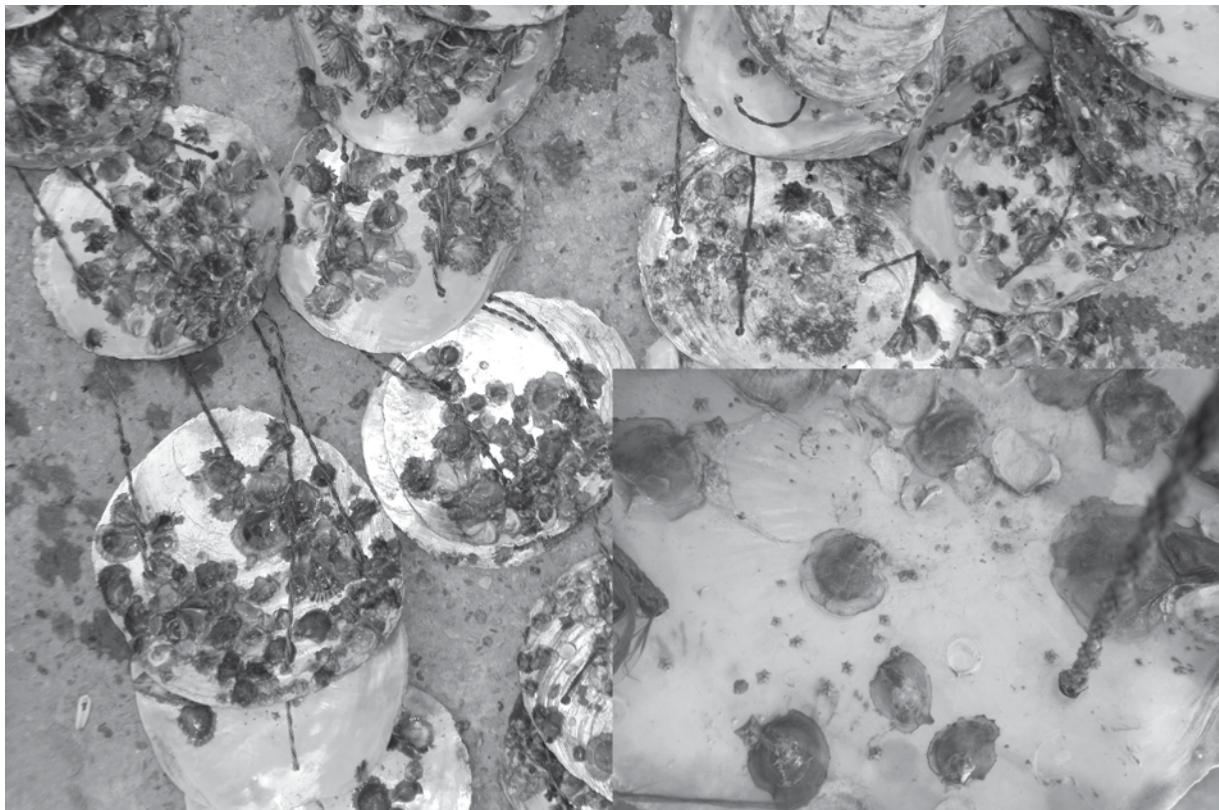
2010). Finally, nutrition and income that come from shellfish is going towards a vulnerable state due to ocean acidification. The effect of ocean acidification on mollusc species is shown in Table 5.

The effects of water pH and temperature on the settlement and growth of oyster spats were assessed by linear regression analysis from two different temporal spectrum (2004 and 2012) but in same temporal scale in the Cox's Bazar coast, Bangladesh (Figure 14). The water temperature and pH data were used from Hossain et al. (2004) and ECOBAS (2012). All data were tested for normality and the effects were calculated at 95% confidence intervals. Data points are value of spat density and pH and spat size and pH for 2004 and 2012. The dashed line shows the regression line, whereas the shaded lines represent the 95% confidence intervals. Findings show that, with increase of pH, spat settlement and growth is decreasing which indicate decrease in pH will inversely effect the spat settlement and growth rate at Cox's Bazar coast, Bangladesh (Figure 15).

ECOBAS (2012) is the integrated efforts of the IMARES Wageningen UR, Royal Haskoning, LEI and Chittagong University addressing the "Building and Farming with Nature" can easily be embedded in the multi-annual strategic plan for the Dutch-Bangladesh future bilateral cooperation as it is being developed in the context of the CIWK (Information Chain for Water, Food and Climate Services). Some shellfish species in the coastal waters of Bangladesh such as oyster (*Crassostrea madrasensis*), green mussel (*Perna viridis*) and clam (*Meretrix meretrix*) can be potentially used as ecosystem engineers and grown naturally to adapt ecosystem-based solutions for coastal defence. These organisms are capable of forming conspicuous habitats that influence tidal flow, wave action and sediment dynamics in the coastal ecosystem and, in doing so, reduce hydrodynamic stress and modify the patterns of local sediment transport, deposition, consolidation, and stabilization processes (Figure 16). Besides, the bivalve reefs also support habitat for numerous species of fish, crustaceans and shellfish, can provide food/protein security, and contribute to a healthier ecosystem with multiple benefits and functions.



**Figure 14.** Effect of pH and temperature on oyster spat density and growth at Cox's Bazar coast, Bangladesh (Hossain et al 2004; ECOBAS 2012)



**Figure 15.** Settled oyster spats on the windowpane shell in the Cox's Bazar coast, Bangladesh (ECOBAS 2012)



**Figure 16.** The green mussel population in the Moheshkhali Island, Cox's Bazar (ECOBAS 2012)

### 3.4. Effects on commercial fisheries

Oceanic ecosystem supports rich and diverse fisheries that form an integral component of the ecosystem and are associated with economic and other societal benefits. The acidity of the oceans is one important parameter that influences fisheries production; others include ocean temperature, availability of nutrients and other factors affecting primary productivity, and the state of fish stocks. The state of fish stocks is determined by ‘natural’ factors influencing recruitment and mortality, and also, importantly by fishing pressure, which in turn is determined by fisheries management. Adult marine fish are generally tolerant of high CO<sub>2</sub> conditions (Melzner et al. 2009) and juvenile stages are most sensitive to changes in acidity, coupled with changes in seawater temperature. As these stages are also those affected most by over fishing of mature animals, it is particularly important that we understand the potential additional stress that ocean acidification may induce on these organisms and those that support them. Responses by juveniles and larvae to ocean acidification include diminished olfactory and auditory ability that

affecting predator detection and homing ability in coral reef fish (Munday et al. 2010), reduced aerobic scope (Munday et al. 2009) and enhanced otolith growth in seabass (Checkley et al. 2009), however, the relative insensitivity of crustaceans to acidification (Ries et al. 2009) has been ascribed to well-developed ion transport regulation and high protein content of their exoskeletons (Krocker et al. 2010). Nevertheless, spider crabs show a narrowing of their range of thermal tolerance by approximately 2°C under high CO<sub>2</sub> conditions (Walther et al. 2009). Mortality occurred in fish species only at very high CO<sub>2</sub> levels (50,000 ppmv), although mortality due to anthropogenic CO<sub>2</sub> is never expected in marine environments (Hayashi et al. 2004). The hatchling stages of some species appeared fairly sensitive to pH decreases on the order of 0.5 or greater, but high CO<sub>2</sub> tolerance developed within a few days of hatching (Ishimatsu et al. 2004) and also causes depress metabolism in some species (Hand 1991; Pörtner and Reipschläger 1996; Guppy and Withers 1999). The effects of ocean acidification on some fish species are summarized in Table 6. Examples of the response of marine fauna to ocean acidification are shown in Appendix-E.

**Table 6.** The effects of ocean acidification on different mollusc species, negative effects are denoted by minus sign, positive effects plus sign and no effect 0

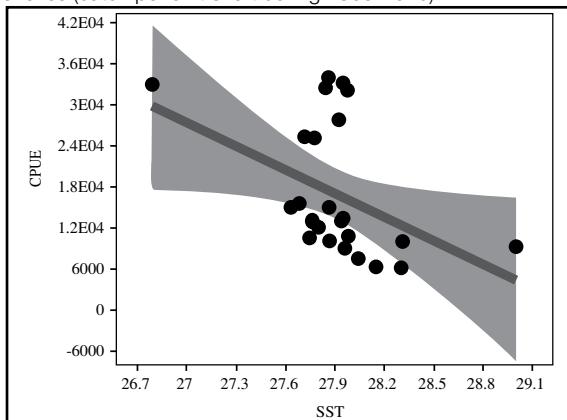
| Species                            | Growth | Survival | Others                             | References                |
|------------------------------------|--------|----------|------------------------------------|---------------------------|
| <i>Amphiprion percula</i>          | 0      | 0        | Embryonic duration (0)             | Munday et al. 2009        |
| <i>Rainbow trout</i>               | +      |          | Energy conversion efficiency (+)   | Dockray et al. 1998       |
| <i>Sillago japonica</i>            |        | -        |                                    | Kikkawa et al. 2006       |
| <i>Pagrus major</i>                |        | -        |                                    | Ishimatsu et al. 2005     |
| <i>Euthynnus affinis</i>           |        | -        |                                    | Kikkawa et al. 2003       |
| <i>Paralichthys olivaceus</i>      |        | -        |                                    | Hayashi et al. 2004       |
| <i>Clupea harengus</i>             | -      |          | Metabolism (-)                     | Franke and Clemmesen 2011 |
| <i>Scyliorhinus canicula</i>       |        |          | Embryogenesis (-)                  | Davenport 1987            |
| <i>Gadus morhua</i>                |        | -        |                                    | Frommel et al. 2011       |
| <i>Anarhichas minor</i>            | -      |          | Metabolism (-)                     | Foss et al. 2003          |
| <i>Sparus aurata</i>               | -      |          | Metabolism (-)                     | Michaelidis et al. 2007   |
| <i>Acanthochromis polyacanthus</i> |        |          | Sagittal otoliths (-)              | Munday et al. 2011        |
| <i>Oncorhynchus mykiss</i>         |        |          | Sagittal and lapellar otoliths (-) | Payan et al. 1999         |
| <i>Psetta maxima</i>               |        |          | Sagittal and lapellar otoliths (-) | Payan et al. 1999         |
| <i>Dicentrarchus labrax</i>        |        |          | Feeding (-)                        | Small et al. 2010         |
| <i>S. canicula</i>                 |        | -        |                                    | Small et al. 2010         |
| <i>Gadus morhua</i>                |        |          | Stock (-)                          | Toresen and Østvedt 2000  |
| <i>Pomacentrus amboinensis</i>     |        |          | Sensory organ (-)                  | Ferrari et al. 2012       |
| <i>Sparus aurata</i>               |        |          | Metabolic capacity (-)             | Michaelidis et al. 2007   |
| <i>Seriola quinqueradiata</i>      |        | -        | Cardiac output (-)                 | Ishimatsu et al. 2004     |

## In acidic seas, it is required to domesticate resistant species for successful aquaculture operations

Marine fishes are playing the central role in meeting the demand of animal protein supply in Bangladesh. The total production of fisheries from 1950-2009 is shown in Table 7. Data are discussed at 10 years intervals emphasizing capture fisheries, culture fisheries and total fisheries production. Marine capture fisheries contributed 5.5% of the total capture during 1950-1959 which increased to 28.8% in 2000-2009. Similarly, coastal culture fisheries contributed 5.9% during 1950-1959 which increased to 32.1% in 2009-2009. Trade and export of fisheries commodity is also showing an increasing trend (Figure 17).

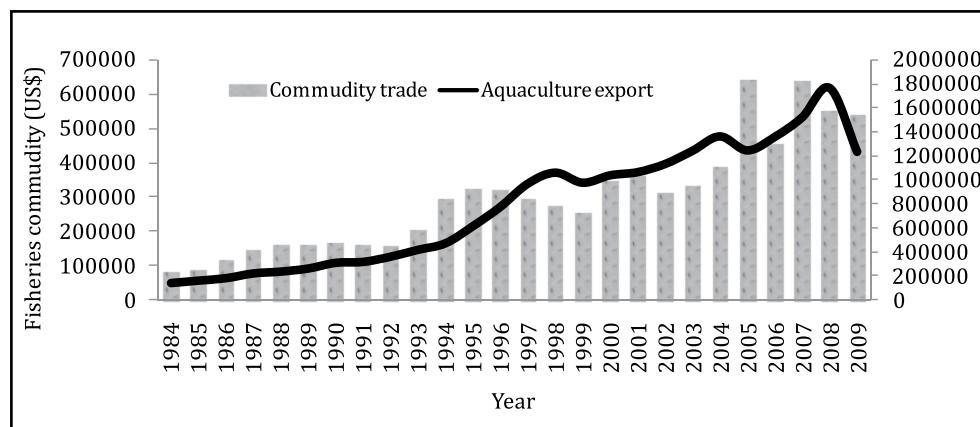
intervals. Findings show that, with increase of SST, CPUE is decreasing ( $R^2 = 0.79$ ) which is indicating a lower production of marine capture fisheries.

**Figure 18.** The effects of increasing SST on marine capture fisheries (catch per unit effort during 1986-2010)



**Table 7.** Annual average catch of marine fisheries in Bangladesh from 1950-2009 (FAO, 2012)

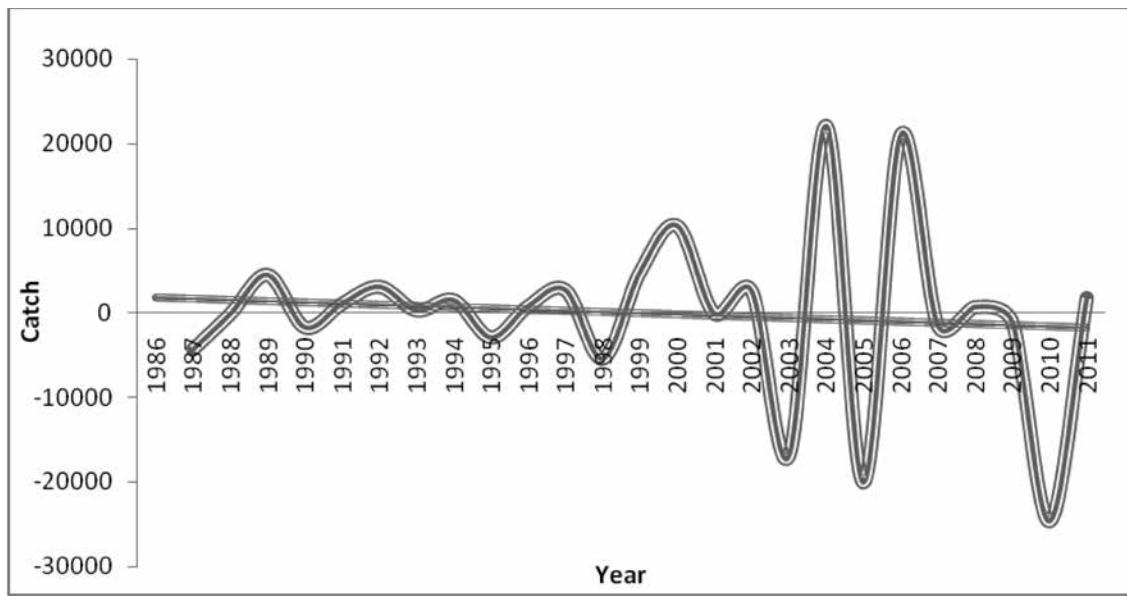
| Year      | Marine capture fisheries | % total capture | Marine culture fisheries | % of total culture | Total marine fisheries | % of total |
|-----------|--------------------------|-----------------|--------------------------|--------------------|------------------------|------------|
| 1950-1959 | 335000                   | 5.5             | 2426103                  | 5.9                | 2761103                | 5.9        |
| 1960-1969 | 511600                   | 8.4             | 4952471                  | 12.1               | 5464071                | 11.6       |
| 1970-1979 | 701209                   | 11.6            | 6369497                  | 15.6               | 7070706                | 15.1       |
| 1980-1989 | 980811                   | 16.2            | 6192072                  | 15.1               | 7172883                | 15.3       |
| 1990-1999 | 1784722                  | 29.5            | 7823183                  | 19.1               | 9607905                | 20.5       |
| 2000-2009 | 1744897                  | 28.8            | 13148945                 | 32.1               | 14893842               | 31.7       |
| Total     | 6058239                  | 100             | 40912271                 | 100                | 46970510               | 100        |



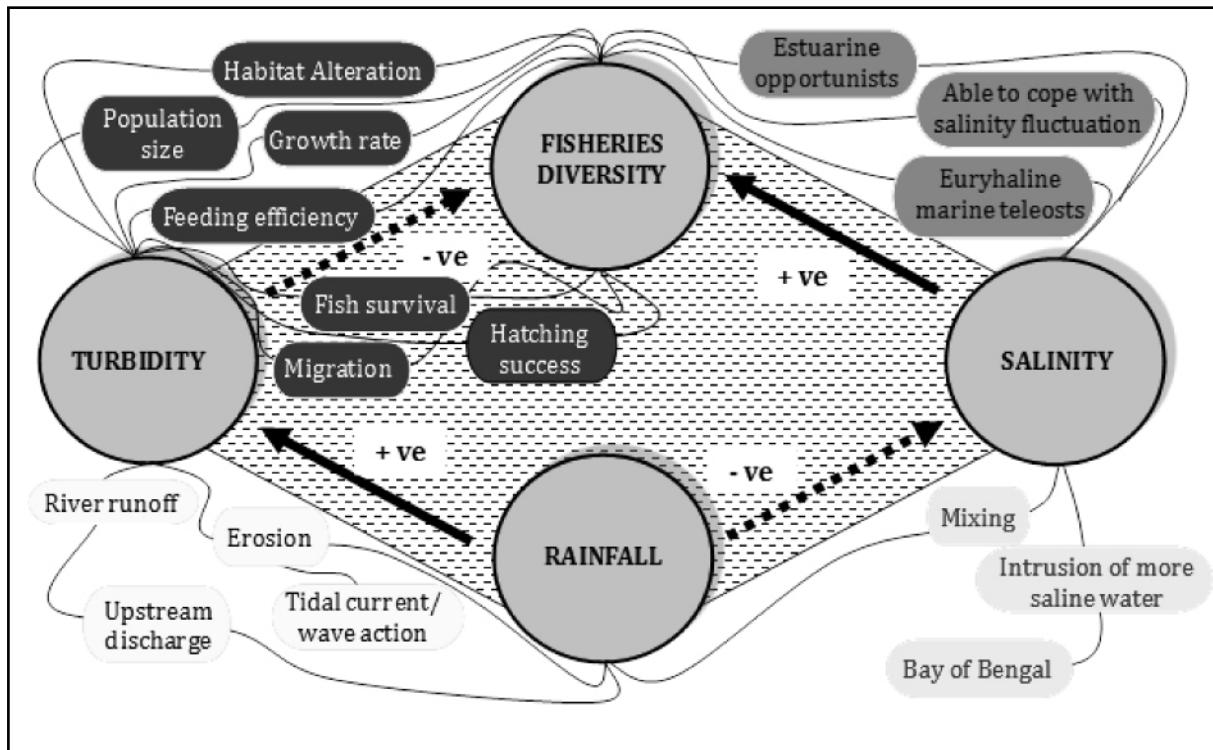
**Figure 17.** The trends in international fish trade of Bangladesh from 1984-2009 (FAO 2012)

The effects of SST on catch per unit effort (CPUE) were assessed by linear regression analysis (Figure 18). CPUE data were calculated from total yearly catch data of 1986-2010 and SST data were derived from WOD09 for the same time period. All data were tested for normality and the effects were calculated at 95% confidence intervals. Data points are value of SST and CPUE from the 1986-2010 time periods. The dashed line shows the regression line, whereas the shaded lines represent the 95% confidence

Though the total fisheries commodity trade and aquaculture product export increased over the period (Figure 17), the CPUE is decreasing (Figure 18). Negative value in figure 19 shows a decline in CPUE from the previous year and positive value indicate increase in CPUE from the previous year. Accumulating the decline and increased rate of CPUE for each year from 1986-2010 time period, a declining trend is observed for the net CPUE.



**Figure 19.** The changes in CPUE from 1986-2010 in the Bay of Bengal

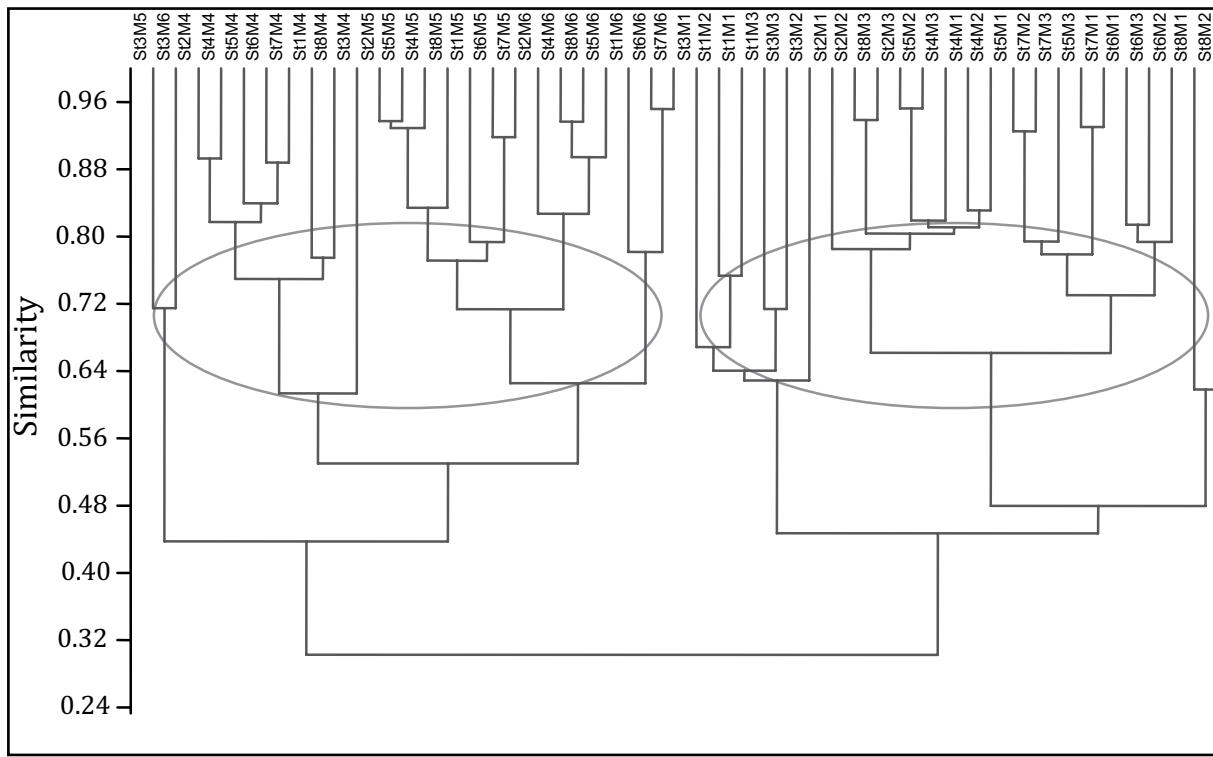


**Figure 20.** Interactions among the major environmental variables and their effects on the fisheries diversity of Naaf River, Cox's Bazar (Chowdhury et al. 2011)

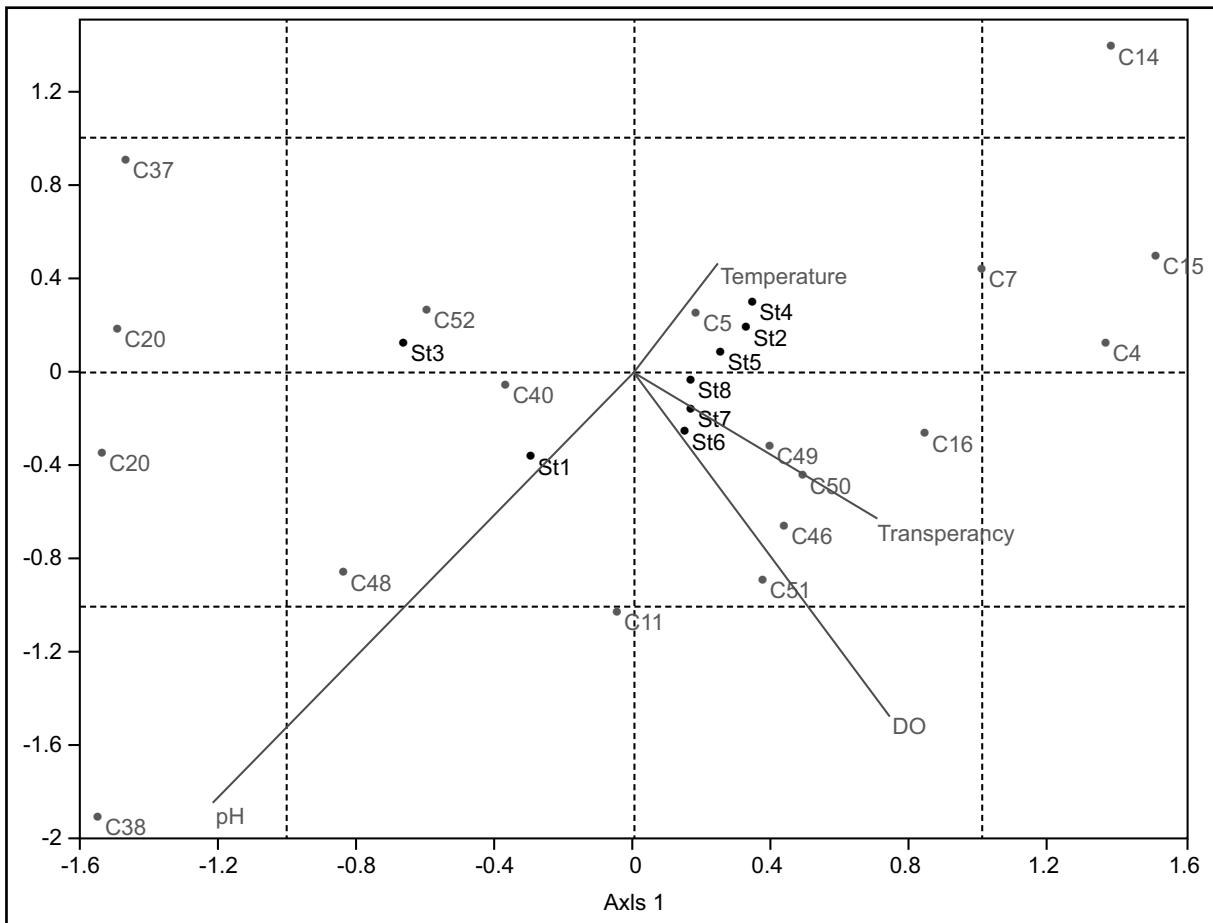
Chowdhury et al. (2011) reported the major environmental variables and their effects on fisheries diversity at the Naaf River estuary (Figure 20). The Naaf River is a highly diverse fisheries ecosystem with favourable water quality than that of many other river/estuarine systems in Bangladesh.

The environmental aspects i.e. hydrological and meteorological parameters act as a driving force for fisheries distribution at the Meghna river estuary. Regarding spatial and temporal fish assemblage

structure, two major groups were indicated by cluster analysis in the Meghna river estuary (Hossain et al. 2013). Group 1 comprises the sample of the February, March and April of 2003. On the other hand Group 2 is formed by November, December and January of 2003 and showed 32% similarity with Group 1 (Figure 21). Water pH was the most important hydrological factors followed by dissolved oxygen, transparency and water temperature on species distribution at the Meghna river estuary (Figure 22).



**Figure 21.** The spatial and temporal cluster of fish assemblage (a Bray-Curtis similarity matrix) in the Meghna estuary (Hossain et al. 2013)



**Figure 22.** Canonical correspondence analysis of fish species abundance with hydrological parameters in the Meghna estuary (Hossain et al. 2013)

### 3.5. Effects on mangrove forest

Mangroves play an essential role in maintaining a healthy coastal environment by providing protection for a myriad of juvenile aquatic species, functioning as a habitat for a variety of terrestrial fauna, operating in improving coastal protection and acting as a source of nutrients that sustain many complex food chains (Kovacs 1999; Hossain 2013). The protective benefits of mangrove forest against tropical cyclone and wave action is important and well-recognized (Hossain et al. 2003; Barbier et al. 2008). The importance of mangroves as nursery grounds of larval and juvenile stages of finfishes, shrimps, crabs, and cockles has highlighted by many researchers around the world (Mahmood et al. 1985; Khan and Hossain 1996; Hossain 2001; Glaser 2003; Lee 2004; Hossain et al. 2009) (Appendix-F). The annual economic value of mangroves, estimated by the cost of the products and services they provide, is about US\$ 200,000-900,000 per ha (Wells 2006). Important goods and services of mangroves upon which local community can ensure their daily demand are shown in Figure 23. Fishing, shrimping and crabbing in the mangrove forest and adjacent areas is an important source of income and livelihood option in the islands of Ganges basin (Hossain 2013).



The environmental impacts of ocean acidification on mangrove ecosystem are not well documented. Increased  $CO_2$  causes reduction in water pH which also reducing soil pH and responsible for damaging mangrove vegetations. Low soil pH may damage breathing roots of mangroves and causes mortality. Acidification can also cause physical damage and altered topography and hydrology. Van Breemen et al. (1973), while working in humid tropical mangrove in Thailand found that acidification causes dying from the root upwards. At the site of vegetation damage, fish usually die a year later at the onset of the rainy season. The fiddler crab (*Uca tangeri*) and mudskippers (*Periophthalmodon koelreuteri*) that used to be abundant in these areas are becoming less common.

The Bangladesh coast supports about 587,380 ha of natural mangroves and a further 100,000 ha of planted mangroves. Declining of mangrove ecosystem due to acidification will not only causing colossal loss of coastal habitat, aquatic resources and biodiversity, but also will increase soil erosion, changes in sedimentation pattern and shoreline configuration, vulnerability to cyclonic storms, tidal bore and denudation of feeding, breeding and nursery ground for various marine, estuarine and fresh water fishery resources. As a result, natural shrimp fry as well as shrimp brood, which supports exportable shrimp farming in Bangladesh; availability will be greatly reduced in the Bay of Bengal (Hossain 2001).

Figure 23. Ecosystem services of mangrove forests in the islands of Ganges basin, Bangladesh (Hossain 2013)

### 3.6. Effects on coastal ecosystems

Coastal ecosystems, like most other ocean ecosystems, are feeling the effects of anthropogenic problems such as climate change, ocean acidification and eutrophication due to excessive nutrient runoff. Many of the physiological changes from ocean acidification are expected to affect key functional groups of organisms that play a disproportionately important role in ecosystems. These include expected effects on phytoplankton, which serve as the base of food webs. Such changes may lead to indiscriminate shifts in composition, structure and function of ecosystem and ultimately affect the goods and services provided to society. Mangrove and coral reef ecosystems are providing services and goods such as coastline protection from storm waves and shoreline stabilization. Declining of these ecosystems due to ocean acidification will make coastal development expensive (Cesar et al. 2003).

Moreover, without coral reefs and mangrove forest, coastal ecosystems would lose valuable habitat and diversity. Oysters and mussels rely on  $\text{CaCO}_3$  for the formation of their protective exoskeleton. Likewise many other organisms rely on these molluscs for services such as providing habitat and protection. Thus the effects of ocean acidification will not only affect those organisms dependent on  $\text{CaCO}_3$  for skeletal construction, but it will also severely affect all those smaller organisms that depend on those larger ones for protection and habitat as well as those organisms (including humans) that rely on them as sources of food. Thus, evidences demonstrating that ocean acidification can damage coastal ecosystems, including species extinction and insecure community livelihoods. However, these impacts are highly species-specific and even population-specific. Probable effects of ocean acidification on coastal ecosystem of Bangladesh are shown in Figure 24.

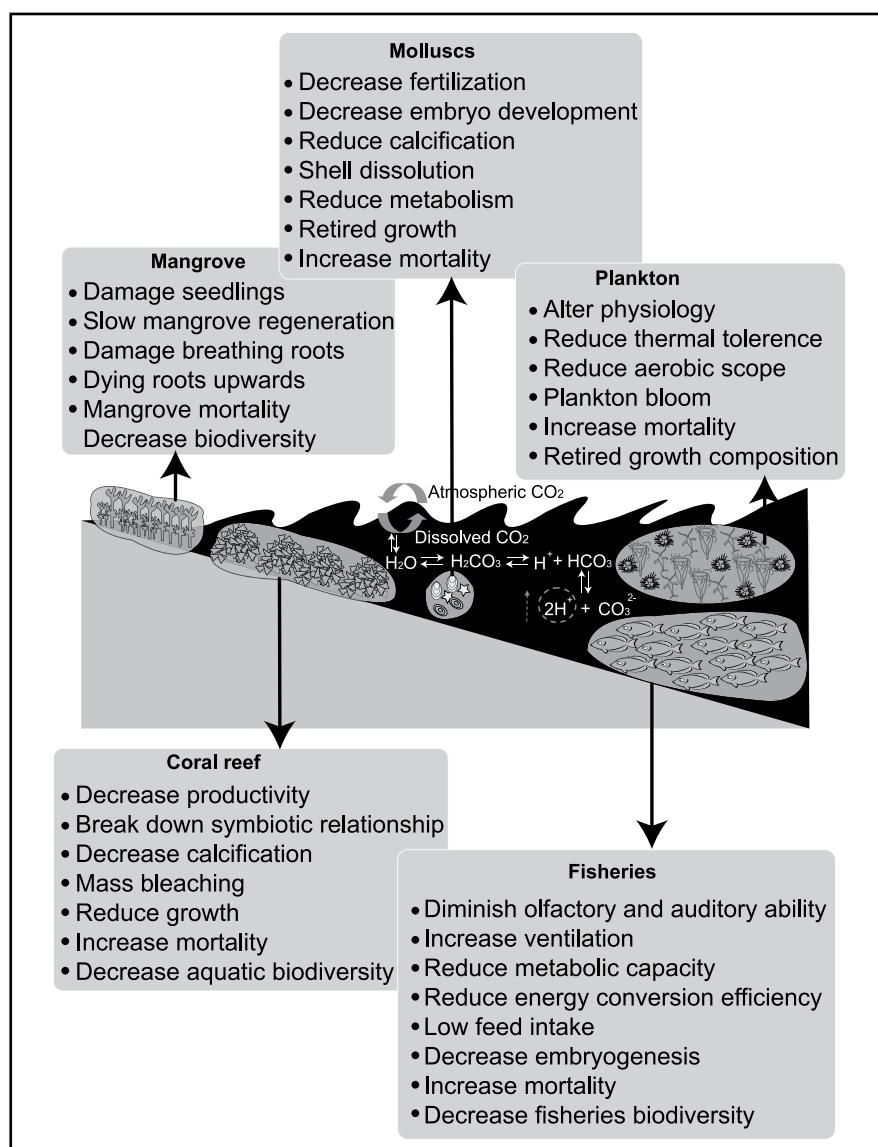


Figure 24. Probable effects of ocean acidification on the coastal ecosystems of Bangladesh



Ocean acidification will badly reduced coastal community's resilience by affecting fish habitats..  
© Sumon Karmokar/ Save Our Sea

## 4. Effects of Ocean Acidification on Coastal Livelihoods

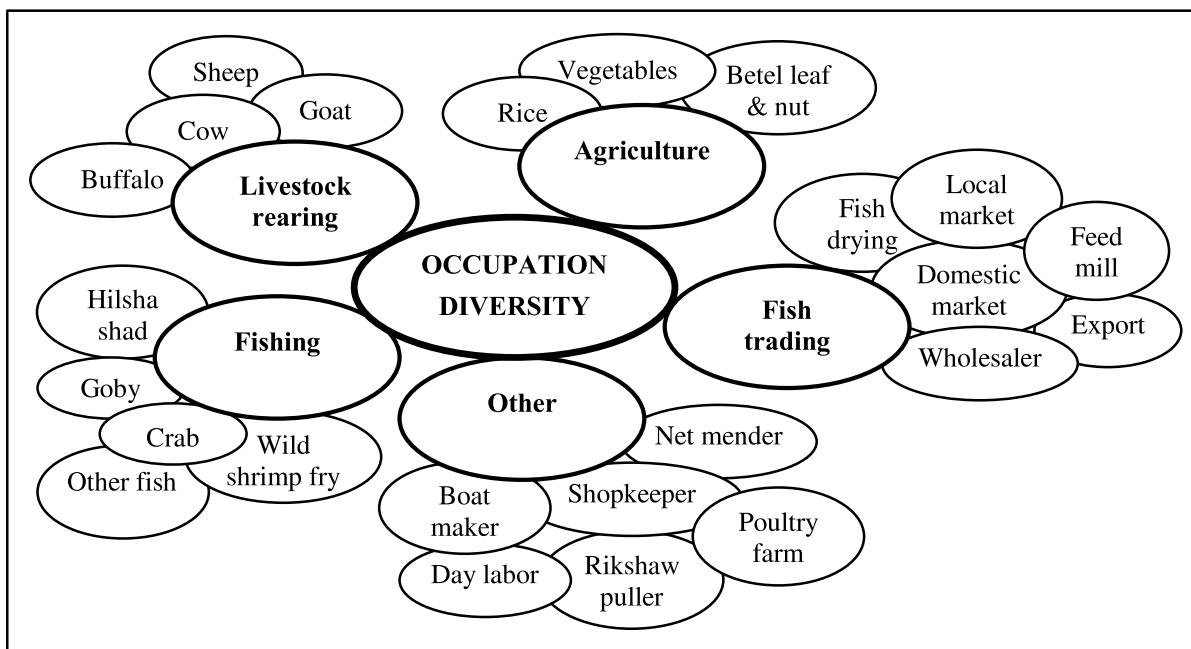
Ocean acidification is likely to affect organisms that support subsistence, commercial, and recreational fisheries, thereby impacting the ability of those ecosystems to provide services upon which humans have been dependent (Cooley et al. 2009). Coastal people naturally have the skills and knowledge on fishing, shrimping, crabbing, agriculture, livestock raising, trading and other occupations in the locality. A typical example of islands occupation diversity from Nijhum Dwip is given in Figure 25. These people, by their skills, are using the natural resources in a responsible way to achieve a higher income and food security.

Changes in climatic conditions will have many health effects, via stresses on human biology, environmentally mediated exposures (infectious diseases, food yields etc.) and the physical risks of injury (McMichael et al. 2006). These effects will occur via direct and indirect pathways and across

traditional living, hunting and eating patterns among the Inuit communities of northern Canada. This has caused a decline in physical activity and a greater reliance on imported energy-dense processed foods, resulting in a much greater probability of adverse health consequences, especially obesity, cardiovascular disease and the occurrence of diabetes (McMichael et al. 2006).

### 4.1. Fishing communities and aquafarmers

The global marine capture fisheries and aquaculture industries provided 110 million metric tons of human food in 2006 with a commercial value of \$170 billion (FAO 2009). The value of global fisheries associated with coral reefs alone is estimated to be \$5.7 billion per year (Conservation International 2008). Changes in these industries due to acidification are likely to disproportionately affect developing nations who often rely more on marine-related economic activities than the developed nations. Projections of population increase and acidification rates suggest that by 2050, low-latitude regions will be most affected, and also the nations that rely upon tropical



**Figure 25.** A Venn diagram illustrating the occupation diversity of coastal community at Nijhum Dwip (Hossain et al. 2013)

different timescales. Livelihoods from fisheries, farming and forestry are at risk in many vulnerable regions of the world. The impacts of climate change on food yields and livelihoods will therefore also affect mental health and health-related behaviours. The indirect impacts on health are well illustrated by the consequences of the recent warming of the Arctic region, which is occurring more rapidly than that for the world at large. The resultant loss of ice (both sea-ice and permafrost) has begun to disturb

marine ecosystems, which would adversely affect by acidification (Cooley et al. 2009). The total economic value generated by fisheries, including wild fish and aquaculture, is estimated to be \$150 billion per year (Makarow et al. 2009). In 2000, recreational saltwater fishing produced \$12 billion income, supported 350,000 jobs, and generated a total economic benefit of \$43 billion in the United States (Cooley et al. 2009). If acidification damages marine habitats or reduces harvests, the resulting decline in revenues could produce job losses with additional indirect economic costs (Cooley and Doney 2009b).

#### 4.2. Livestock raiser

Ocean acidification effects on a single organism is not always negative; phytoplankton photosynthesis is generally enhanced (Riebesell and Tortell 2011), and some seagrass species even thrive in low pH environments (Hall-Spencer and Rodolfo-Metalpa 2008), but scientific information on the existence of seagrass beds is lacking. In the coastal region generally sea fronts of newly formed islands and some low-lying coastal areas are found to be carpeted with seagrasses (Hossain 2001). Cow and buffalo are reared on grass (Figure 26) for milk and meat. Milk indigenously preserves for natural yogurt, ghol (indigenous yogurt drink) and ghee (butter for cooking) preparation to supply in city markets.



**Figure 26.** Grass beds with mangroves and buffalo grazing at the Sonadia Island, Cox's Bazar (August 2012)



Sheeps graze in nearshore coastal grasslands of Hatiya island  
© IUCN/Enamul Mazid Khan Siddique

## 5. Mitigation of Ocean Acidification Impact

### 5.1. National institutional framework

#### Government Ministries and Agencies

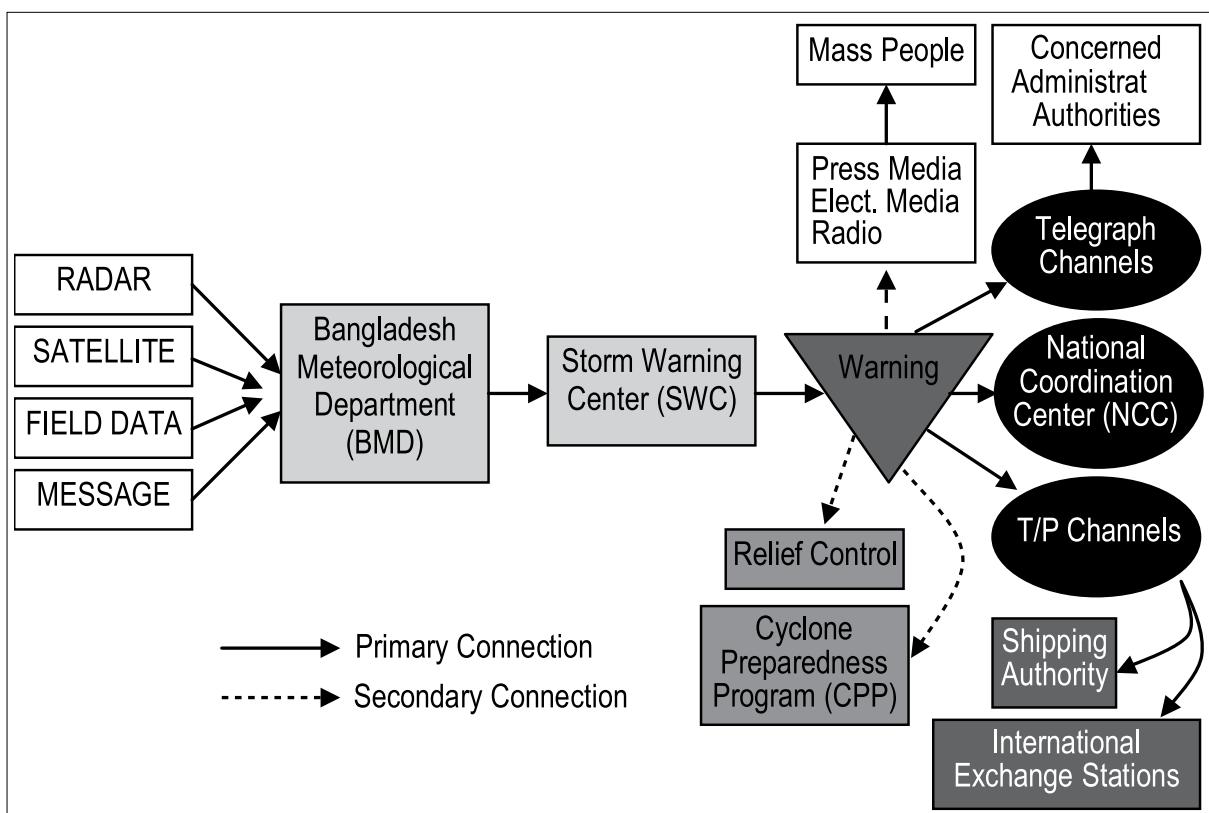
The main ministries of the Government of Bangladesh involved in climate change are the Ministry of Environment and Forests and its agencies (e.g. the Department of Environment – DoE and Department of Forests – DoF); Food and Disaster Management (MoFDM), which includes the Disaster Management Bureau (DMB) and the Comprehensive Disaster Management Programme (CDMP); Water Resources, which includes the Bangladesh Water Development Board and other

research and forecasting organizations; Local Government Engineering Department (LGED) and the Department of Public Health Engineering (DPHE); Agriculture, including the National Agricultural Research System, which develops new crops and practices suited to different climatic and salinity conditions in the country; Livestock and Fisheries; Power, Energy and Mineral Resources; Health and Family Welfare; Roads and Railway Division, Ministry of Communication; Ministry of Foreign Affairs and the Planning Commission which is charged with the framing of development plans as well as approval of programmes and projects. Key roles and responsibilities of some of these organizations are given in Table 8.

**Table 8.** Examples of entity actions to promote, implement, and support adaptation at all scales

| Entity  | Key roles and responsibilities  |
|---|---|
| The Ministry of Environment and Forests   | The Ministry of Environment and Forests is the focal ministry for all work on climate change, including international negotiations. It provides the Unit for the National Environment Committee, which ensures a strategic overview of environmental issues and is chaired by the Prime Minister. Immediately after the Bali Conference (COP 13), the Government formed the National Steering Committee on Climate Change. It is headed by the Minister, the Ministry of Environment and Forests, and comprises secretaries of all relevant ministries and civil society representatives. It is tasked with developing and overseeing implementation of the National Climate Change Strategy and Action Plan. Five technical working groups were also constituted on adaptation, mitigation, technology transfer, financing and public awareness.                               |
| The National Disaster Management Council (NDMC)   | The National Disaster Management Council (NDMC), headed by the Prime Minister, is the highest-level forum for the formulation and review of disaster management policies. The Inter-Ministerial Disaster Management coordination Committee is in charge of implementing disaster management policies and the decisions of the NDMC, assisted by the National Disaster Management Advisory Committee.  |
| The Ministry of Food and Disaster Management  | The Ministry of Food and Disaster Management is the focal ministry for disaster management. Its Disaster Management Bureau (DMB) is the apex organization responsible for co-ordinating national disaster management interventions across all agencies. It is a technical arm of the Ministry of Food and Disaster Management. It oversees and coordinates all activities related to disaster management at national and local levels. In 2000, the Government published Standing orders on Disasters, which provide a detailed institutional framework for disaster risk reduction and emergency management and defines the roles and responsibilities of different actors. The Comprehensive Disaster Management Programme (CDMP), a donor funded programme, aims to strengthen the DMB and shifts the emphasis away from relief of disaster preparedness and risk reduction. |
| Bangladesh Meteorological Department (BMD) and Space Research and Remote Sensing Organization (SPARRSO) | BMD and SPARRSO under the Ministry of Defence; the Flood Forecasting and Early Warning Centre (FFEWC) and Bangladesh Water Development Board (BWDB) under the Ministry of Water Resources are two of the key institutions in this field. BMD takes responsibility for preparing all weather forecasts and disaster warnings through interconnected subdivisions (Figure 27). As the coastal people are used to facing multiple hazards each year, their responses to warnings depend on the intensity of wind speed, experience of hazards, local belief in the probability of dangerous climatic events, or the presence of a cyclone signal hoisted by the Bangladesh Meteorological Department.  |
| The Planning Commission   | The Planning Commission being the final approving authority under the Ministry of Planning which is preparing the Sixth Five Year Plan will look into the over-all integration of climate change issues within the planning framework as well as be the final approving authority for programmes and projects to be carried out as part of the implementation of the Action Plan.   |
| Others  | There are 35 more other ministries also responsible for sectors that are vulnerable to the effects of climate change, including agencies responsible for water resources, health, agriculture, urban planning, roads and transport.   |

| Entity                              | Key roles and responsibilities  |
|-------------------------------------|---|
| Non-government Organizations (NGOs) | Many non-government entities have been significant actors in the national effort to prepare for climate change by providing assistance that includes planning guidance, implementation tools, contextualized climate information, best practice exchange, and help with bridging the science-policy divide to a wide array of stakeholders. Bangladesh has lots of non-government organizations like BRAC, Grameen Bank, BCAS, ASA, TMSS, CODEC, NACOM, COAST, etc. NGOs have made a major contribution to poverty reduction and has increased the resilience of poor people to natural disasters, through microfinance, income generation, health and education programmes. Their experience and capacity will be used to develop innovative approaches to adaptation. Moreover, international organizations such as IUCN, IIED, WWF, OXFAM, CARITAS, YMCA, etc have been contributed in overall social development. |
| Universities                        | Different universities have been addressed climate change issues in the curriculum and conduct research.  |
| Donors                              | DFID, DANIDA, USAID, EU UNION, GIZ, WB, ADB   |



**Figure 27.** Flow diagram of national warning systems by the Bangladesh Meteorological Department (source: Haque, 1997; Chowdhury et al. 2012).

## 5.2. International institutional framework

The international distributional challenges of acidification are significant. Developed nations are responsible for the largest portion of historic emissions and therefore for current acidification. They also possess the most diverse economies and food systems. Transfer schemes to compensate developing nations particularly affected by acidification may be warranted. Mechanisms to achieve this include the Adaptation Fund, established by the parties to the Kyoto Protocol (UNFCCC 2010a). The Fund supports adaptation activities in developing countries with Clean Development Mechanism

(CDM) funding from developed countries. However, with the expiration of the Kyoto Protocol in 2012, there is uncertainty about the future of the CDM and hence, the Adaptation Fund. The Copenhagen Accord, agreed to by 114 parties in 2009, recommitted developed countries to providing financial resources to support adaptation measures in developing countries (UNFCCC 2010b).

Ocean acidification is an unambiguous consequence of increasing atmospheric CO<sub>2</sub> concentrations. While reducing CO<sub>2</sub> emissions will be necessary to limit the most disruptive effects, the process of acidification has begun and will continue in the

future. Policy instruments are available to reduce the vulnerability of marine ecosystems and coastal communities to changes in the composition and productivity of the oceans' biological systems. Given the economic and social importance of the oceans to human societies, governments at the local, national, and international levels must begin to assess and implement adaptive approaches to acidification.

### 5.3. Policy recommendations

Policy makers considering ways to address acidification have a menu of options. The most obvious, and thus far most politically difficult, is to reduce CO<sub>2</sub> emissions in order to avoid the most severe impacts of acidification. Past missions CO<sub>2</sub> in the atmosphere will continue to contribute to future acidification. While limiting the extent of acidification depends upon curtailing future emissions, efforts at mitigation have thus far been unsuccessful. The lack of comprehensive policies to reduce emissions in the U.S. and other nations with significant emissions, as well as the inability of the international community to adopt binding reduction targets to replace the expiring Kyoto Protocol, make meaningful emissions reductions in the near term unlikely. In the absence of significant reductions in emissions, acidification will continue, requiring adaptation based measures. While mitigation will require a global commitment, adaptation actions can be adopted at the local, national, and international levels.

**As a potential initiative, coastal plantation, coral/oyster reefs, raising high yielding varieties of rice/fish/livestock hold great promise.**

One set of adaptation strategies focuses on limiting the vulnerability of ecosystems to the stresses of acidification. Therefore, policies to limit marine pollution and curtail over fishing may have a positive effect on the ability of marine ecosystems to adapt to acidifying conditions (Cooley et al. 2009). Reducing harvest limits in the near term may result in short term revenue losses, but larger fish stocks and higher revenues in the long term (Cooley et al. 2009). Policy instruments to reduce pressure on fisheries include license or vessel buyouts and regional fishery closures (Cooley et al. 2009). Other strategies include increasing aquaculture operations in order to compensate for losses in wild harvests. In acidifying seas, aquaculture operations may have to use selective breeding techniques to rear species able to withstand these conditions (Cooley et al. 2009). In addition, freshwater aquaculture

operations, isolated from acidifying oceans, may become more attractive as a means of supplying protein to consumers.

Within nations, extractive users, such as fishermen and aquaculture operators, face additional costs and economic losses if harvests decline (Cooley et al. 2009). A number of adaptation measures are available to support fishing dependent communities. Possible actions include diverting fishing effort to new or underutilized species, protecting key functional groups, investing in monitoring and research, and decoupling the local economy from fishing (MacNeil et al. 2010). This last action entails the greatest disruption to existing economic and social structures. Therefore, the social costs of lost employment and dislocation may require subsidies from less affected segments of the national population. Social policies such as job retraining and relocation assistance may be required to support formerly fishery-dependent communities (Cooley et al. 2009). For example, the U.S. National Oceanic and Atmospheric Administration (NOAA) requested \$8 million to create a National Working Waterfronts grant program to aid fishing-dependent communities diversify their economies (NOAA 2011).

### Key recent initiatives

**Ocean Acidification Network** – International initiative developed by the International Geosphere-Biosphere Programme (IGBP), the Scientific Committee on Oceanic Research (SCOR), the UNESCO's Intergovernmental Oceanographic Commission (IOC) and the IAEA's Marine Environment Laboratories (MEL) as a follow-up of an IOC-SCOR symposium in 2004. This information network community provides a central source of information for ocean scientists on ocean acidification research activities. [www.ocean-acidification.net](http://www.ocean-acidification.net).

**Ocean Acidification blog** – an information outlet on ocean acidification- [oceanacidification.wordpress.com](http://oceanacidification.wordpress.com)

**United Nations Environment Programme (UNEP) World Conservation Monitoring Centre (WCMC)**, Ocean Acidification info page — [www.unep-wcmc.org/resources/ocean\\_acid\\_promo.aspx](http://www.unep-wcmc.org/resources/ocean_acid_promo.aspx)

**International Union for Conservation of Nature (IUCN) Coral Reef Resilience to Climate Change** — [www.iucn.org/cccr/resilience\\_to\\_climate\\_change/#2](http://www.iucn.org/cccr/resilience_to_climate_change/#2)

**International Coral Reef Initiative (ICRI) Recommendation on Acidification and Coral Reefs (2007).** [www.icriforum.org/secretariat/japan/grm/docs/Reco\\_acidification\\_2007.pdf](http://www.icriforum.org/secretariat/japan/grm/docs/Reco_acidification_2007.pdf)

**Position Analysis: CO<sub>2</sub> emissions and climate change: Ocean Impacts and Adaptation Issues.** The Australian Antarctic Climate & Ecosystems—Cooperative Research Centre. June 2008—[www.acecrc.org.au/drawpage.cgi?pid=publications&aid=797037](http://www.acecrc.org.au/drawpage.cgi?pid=publications&aid=797037)

**Honolulu Declaration on Ocean Acidification and Reef Management,** August 2008, Hawaii, prepared by the IUCN and TNC, August 2008 — [cmsdata.iucn.org/downloads/honolulu\\_declaration\\_with\\_appendices.pdf](http://cmsdata.iucn.org/downloads/honolulu_declaration_with_appendices.pdf)

**Monaco Declaration** (2008) calling for immediate action by policy makers to reduce CO<sub>2</sub> emissions sharply to avoid possible widespread and severe damage to marine ecosystems from ocean acidification, presented at the end of the Second Symposium on “The Ocean in a High-CO<sub>2</sub> World”, October 2008, Monaco — [ioc3.unesco.org/oanet/Symposium2008/MonacoDeclaration.pdf](http://ioc3.unesco.org/oanet/Symposium2008/MonacoDeclaration.pdf)

**Ocean Acidification – Recommended Strategy for a U.S. National Research Program,** recently prepared by the US Ocean Carbon and Biogeochemistry Program, Subcommittee on Ocean Acidification, March 2009 — [www.us-ocb.org/OCB\\_OA\\_Whitepaper.pdf](http://www.us-ocb.org/OCB_OA_Whitepaper.pdf)

**Research Priorities for Ocean Acidification** prepared by Orr, J.C., Caldeira, K., Fabry, V., Gattuso, J.-P., Haugan, P., Lehodey, P., Pantoja, S., Portner, H.-O., Riebesell, U., Trull, T., Hood, M., Urban, E., and Broadgate, W. (January 2009) — [ioc3.unesco.org/oanet/Symposium2008/ResearchPrioritiesReport\\_OceanHighCO2WorldII.pdf](http://ioc3.unesco.org/oanet/Symposium2008/ResearchPrioritiesReport_OceanHighCO2WorldII.pdf)

**Inter Academy Panel (IAP) statement on Ocean Acidification** calls for world leaders to explicitly recognize the direct threats posed by increasing atmospheric CO<sub>2</sub> emissions to the oceans and its profound impact on the environment and society. June 2009 — [www.interacademies.net/?id=9075](http://www.interacademies.net/?id=9075)

**European Integrated Maritime Policy,** In October 2007, the European Commission presented its vision for an Integrated Maritime Policy (IMP) for the European Union in a Blue Book, providing the overall framework for policy areas related to the sustainable management of European marine waters, including environment and research. The IMP is the central policy instrument to develop a thriving maritime economy in an environmentally sustainable manner. The Action Plan that comes with the Blue Book introduces several actions which are directly or indirectly related to the issue of ocean acidification and greenhouse gas emissions, including actions on reduction of air pollution from ships, marine-

based energy infrastructures and resources, and on mitigation and adaptation to climate change. It also considers Carbon Capture and Storage as a crucial element of European actions needed to meet the Community's climate change objectives to reduce impacts of things such as ocean acidification. To this end, the Commission proposed a Directive in January 2008 to enable environmentally-safe capture and geological storage of CO<sub>2</sub> in the EU as part of a major legislative package. The Directive was adopted by the Parliament and the Council in March 2009.

**European Climate Change Policy,** The EU has been trying to address its own greenhouse gas emissions since the early 1990s with the first (2000) and second (2005) European Climate Change Programme (ECCP) and more recently the Climate Action and Renewable Energy Package (2008) to fight climate change and promote renewable energy up to 2020 and beyond. This package sets out the contribution expected from each Member State to meet these targets of reducing overall emissions to at least 20% below 1990 levels by 2020 and increasing the share of renewables in energy use to 20% by 2020. It also proposes a series of measures to help achieve targets. Central to the strategy is a strengthening and expansion of the Emissions Trading System (EU ETS), the EU's key tool for cutting emissions cost-effectively. Emissions from the sectors covered by the trading system will be cut by 21% by 2020 compared with levels in 2005. A single EU-wide cap on ETS emissions will be set, and free allocation of emission allowances will be progressively replaced by auctioning of allowances by 2020. The climate and renewable energy package also seeks to promote the development and safe use of Carbon Capture and Storage (CCS), a suite of technologies that allows the CO<sub>2</sub> emitted by industrial processes to be captured and stored underground where it cannot contribute to global warming.

In April 2009, the European Commission presented a policy paper known as a White Paper, “Adapting to climate change: Towards a European framework for action”, which presents the framework for adaptation measures and policies to reduce the European Union's vulnerability to the impacts of climate change. It emphasises that decisions on how best to adapt to climate change must be based on solid scientific and economic analysis. The White Paper is accompanied by several working documents, one of which touches upon water, coasts and marine issues including ocean acidification.

## **6. Policy and International Negotiation**

### **6.1. Ocean acidification issues in Bangladesh**

The ocean has absorbed about 430 billion tons of CO<sub>2</sub> from the atmosphere, or about one-third of anthropogenic carbon emissions. This absorption has benefited humankind by significantly reducing greenhouse gas levels in the atmosphere, thereby minimizing global warming. However, the pH (the measure of acidity) of ocean surface waters has already decreased by about 0.1 units, from an average of about 8.2 to 8.1 since the beginning of the industrial revolution. By the middle of this century atmospheric CO<sub>2</sub> levels could reach more than 500 ppm, and by 2100 they could be well over 800 ppm (IPCC scenario B2, 2007). This would result in an additional mean surface water pH decrease of approximately 0.3-0.4 pH units by 2100, implying that the ocean would be about 100-150% more acidic than at the beginning of the industrial revolution.

Oceanic uptake of atmospheric CO<sub>2</sub> released by humans is altering global seawater chemistry in ways that will affect marine biota, ecosystems, and biogeochemistry. Forecasting these impacts requires an integrated understanding of the linkages among ecosystem components and feedbacks to climate. Ocean acidification could alter the ocean carbon, nitrogen, phosphorus, and metal (e.g., iron and copper) cycles, and marine biota from individual organisms to ecosystems could be negatively influenced. Such changes could alter marine biodiversity and food webs. Consequently, fisher's livelihoods, animal protein supply and export earnings may raise red flags. Ocean acidification is recognized as a high priority research topic in Bangladesh. The important issues on ocean acidification in Bangladesh are as follows:

- OA is the global challenge of climate change but affecting the local places including the poor people of Bangladesh. It is clear evidence of "climate injustice" to the innocent people by the developed nation's activities (e.g. CO<sub>2</sub> emission)
- Establishment of ocean data recording stations and laboratories for precise pH measurement and trend analysis is another issue in Bangladesh. Science recognized changes in oceanic ecosystem i.e. about 0.002 pH units changes to see per year
- Detrimental effects of OA on the ecosystem niche of marine organisms e.g. fishes, shrimps, crabs, molluscs and coral reefs may change and migrate in neighbouring country's territory. Recent decreasing trends of capture fisheries from the Bay of Bengal may be the impact of ocean acidification.

The consequences are insecure millions of fisher's livelihood and failure investment on fishing crafts and gears as well as reducing export earning from fisheries commodity

- Development of adaptation capacity with resilient infrastructures (houses, roads, schools, health centers, community places) and income generating options are essential. Coastal plantation, oyster reefs development, high yielding varieties of rice/fish farming, raising livestock (for meat and milk) and poultry (for egg and meat) may be the potential initiatives

### **6.2. National ocean acidification policy of Bangladesh**

#### ***National Adaptation Plans (NAPs)***

In Durban, South Africa in December 2011 the guidelines for National Adaptation Plans (NAPs) were agreed. Further discussions focused NAPs in Bonn, Germany in May 2012 under the United Nations Framework Convention on Climate Change (UNFCCC). Financial support at least for LDCs is necessary for the preparation and implementation of these plans in a participatory and transparent manner. National Adaptation Plans can endow developing countries with a strong climate smart strategic planning process and policy dialogue, embracing and integrating sector-wide and programmatic approaches as part of a coherent institutional, policy and regulatory framework. It is expected that NAPs and associated activities will receive funding from newly established institutions, such as the Green Climate Fund (GCF), but also from the Kyoto Protocol Adaptation Fund and others. Since NAPs should reach out to the wider development agenda, it is also important to target bilateral and plurilateral adaptation and development funding streams.

#### ***Loss and damage***

The Conference of Parties (CoP) in Durban (CoP 17, 2011) took an important step forward by agreeing a work programme on loss and damage through a series of expert workshops. The first expert workshop in Tokyo, Japan in March 2011 focused on the assessment of risks of loss and damage. The discussions showed that there is still a substantial need to further define what should be understood as loss and damage in the UNFCCC context. It is also necessary to address impacts related to extreme weather events and slow onset events.

The Bangladesh-specific component of the Loss and Damage in Vulnerable Countries Initiative is being carried out with three objectives in mind. The first objective is to help GoB, stakeholders and LDC negotiators in the UNFCCC process better

understand the concept of loss and damage. Secondly, stakeholder-based activities to move loss and damage forward in Bangladesh will be identified. Finally, the work in Bangladesh to identify possible approaches to loss and damage will inform other LDCs in developing their own loss and damage strategies.

#### ***National Adaptation Programme of Action (NAPA, 2005)***

The National Adaptation Programme of Action (NAPA) is prepared by the Ministry of Environment and Forest (MoEF), Government of the People's Republic of Bangladesh as a response to the decision of the Seventh Session of the Conference of the Parties (CoP7) of the United Nations Framework Convention on Climate Change (UNFCCC). The National Adaptation Programme of Action (NAPA) has been prepared the adaptation measures as a set of actions complementary to national goals and objectives of other multilateral environmental agreements to which Bangladesh is one of the signatories. The NAPA is also clearly closely related to other environment related policies and programmes in particular the National Action Plan on Biodiversity as well as the earlier implemented National Environmental Management Action Plan (NEMAP). NAPA identified 15 priority activities to combat climate change impacts in Bangladesh, including general awareness raising, technical capacity building, and implementation of projects in vulnerable regions with a special focus on agriculture and water resources.

The basic approach to NAPA preparation was along with the sustainable development goals and objectives of the country where it has recognized necessity of addressing environmental issue and natural resource management with the participation of stakeholders in bargaining over resource use, allocation and distribution. Therefore, involvement of different stakeholders was an integral part of the preparation process for assessing impacts, vulnerabilities, adaptation measures keeping urgency and immediacy principle of the NAPA. Policy makers of Government, local representatives of the Government (Union Parishad Chairman and Members), scientific community members of the various research institutes, researchers, academicians, teachers (ranging from primary to tertiary levels), lawyers, doctors, ethnic groups, media, NGO and CBO representatives and indigenous women contributed to the development of the NAPA for Bangladesh.

At the highest level NAPA has a Project Steering Committee (PSC) headed by Secretary Ministry of Environment and Forest. The Project Steering

Committee is represented by high level officials and experts from different government and non-government organizations to provide guidance. In addition to the Ministry of Environment and Forest, other noteworthy government ministries and agencies involved in the PSC are Ministry of Planning, Economic Relation Division, Ministry of Agriculture, Ministry of Food and Disaster Management, Water Resource Planning Organization, Ministry of Fisheries and Livestock, Ministry of Land, Department of Environment etc. The Project Steering Committee is represented by government, non-government and international research institutes including Bangladesh Institute of Development Studies (BIDS), Bangladesh Forestry Research Institute (BFRI) Bangladesh Centre for Advanced Studies (BCAS), IUCN Bangladesh etc.

#### ***Bangladesh Climate Change Strategy and Action Plan (BCCSAP, 2009)***

In 2009, the Government outlined the BCCSAP, a 10-year program to build capacity and resilience within the country to meet climate change challenges over the next 20–25 years, with six thematic areas: (i) food security, social protection, and health; (ii) comprehensive disaster management; (iii) infrastructural development; (iv) research and knowledge management; (v) mitigation and low-carbon development; and (vi) capacity building and institutional strengthening.

The implementation of BCCSAP will require significant strengthening of the coordination and facilitation capacity of the Ministry of Environment and Forests (MoEF) with relevant line ministries and agencies.

#### **6.3. Capacity building options for mitigating future ocean acidification in Bangladesh**

- Develop university curricula addressing climate change challenges, ocean acidification, carbon sequential, aquatic food production, and the likes for quality higher education.
- Establish research collaboration with developed country academic institutions to undertake need-based action research involving
- Arrange short training programmes for managers, extension workers and policy makers to enhance their knowledge on ocean acidification, sea water chemistry and oceanic uptake of atmospheric CO<sub>2</sub> as well as impact on oceanic organisms and ecosystem.
- Funding is necessary to build networks of ocean acidification buoys around the state territorial water in the Bay of Bengal to generate real-time monitoring of changing conditions throughout the most sensitive

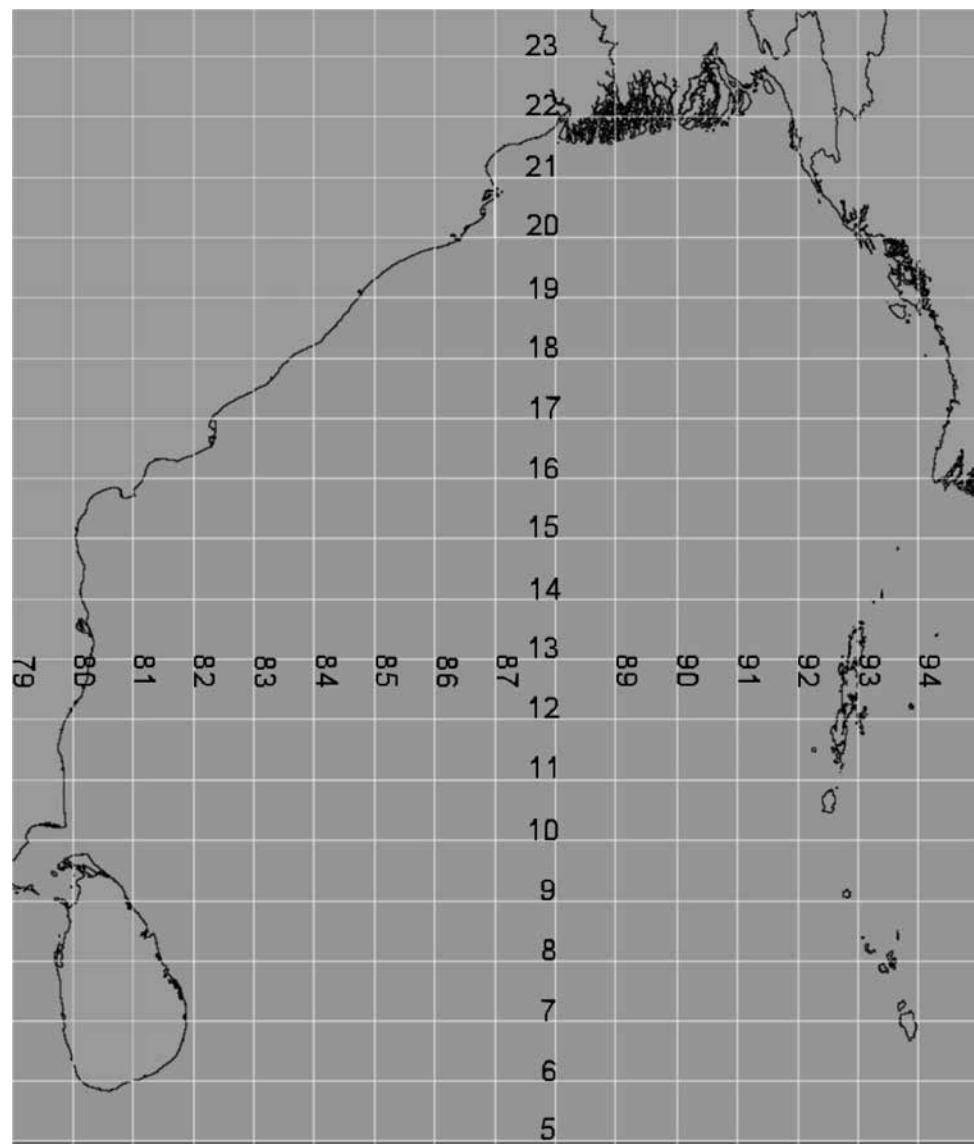
coastal areas i.e. Saint Martin's Island, Cox's Bazar beach, Chittagong port, Meghna deltaic gateway, Sunderbans mangrove, Swatch of No Ground, South Patches, etc.

It is our responsibility to invest resources in areas where stakeholders are at risk due to the consequences of ocean acidification. When everything is in place, we will be capable to measure ocean acidification from our oceanic area. Scientists will combine collected information with data from other research to develop a model to determine the current and future costs of ocean acidification. The data will be made available to the community in a number of different locations through a partnership with universities and research organizations.

#### ***Develop observational and predictive technology***

The greatest opportunity for a leap in understanding variation in ocean chemistry would come from represented real-time data covering coastal and

marine territories. Installation of in situ high-precision sensors for pH and related biogeochemical variables, including oxygen, nitrate, partial pressure of  $\text{CO}_2$  ( $\text{pCO}_2$ ), and particulate inorganic carbon are necessary. More effort is needed to monitor precision and stability of these sensors, develop others especially for key carbonate system variables (e.g. dissolved inorganic carbon and alkalinity), and deploy them not only on moorings but in a new networks of gliders and floats. The Bay of Bengal can be divided into 207 quadrates with  $1^{\circ}\text{N}$  latitude and  $1^{\circ}\text{E}$  longitude (Figure 28). Each quadrate may contain 2 moorings or floats with sensor to measure temperature and salinity in the upper 2 km of the ocean every 7 days, has revolutionized our ability to characterize the ocean's circulation and warming. Likewise, equipping the moorings or floats with multiple biogeochemical sensors would revolutionize our ability to characterize the ocean carbon cycle and quantify how it is being affected by OA.



**Figure 28.** The Bay of Bengal with  $1^{\circ}$  interval quadrates and surrounding countries

### **Perform long-term and multigenerational experiments**

To date, the majority of published OA research involves short-term experiments (days to weeks) that acutely stress organisms using predicted future ocean conditions. Although such experiments are useful for studying physiological processes and identifying major biological effects, they may not provide a true assessment of the longer-term impacts of OA. First, some effects of exposure to acidified conditions may take months to appear. For example, increased costs associated with high CO<sub>2</sub> and low pH may only become evident in whole-organism traits such as growth and survival after several months of continuous exposure (e.g. Langenbuch and Pörtner 2004; Shirayama and Thornton 2005; Kurihara et al. 2008). Alternatively, organisms might acclimate to acidification over time. For example, the crab *Necora puber* used bicarbonate ions acquired from shell dissolution to regulate extra-cellular pH during a short-term (14 days) exposure to elevated CO<sub>2</sub> (Spicer et al. 2007); however, bicarbonate was acquired from the surrounding seawater to regulate pH after longer-term exposure (30 days) (Small et al. 2010). One of the key questions is to determine what is the relevant time of exposure. In a study on the green sea urchin *Strongylocentrotus droebachiensis*, Dupont et al. (2012) show that female fecundity was decreased 4.5-fold when acclimated to elevated pCO<sub>2</sub> for 4 months during reproductive conditioning, while no difference was observed in females acclimated for 16 months. Therefore, short-term experiments can both over- and underestimate the impact of OA.

Experiments should also include subsequent life history stages (e.g. –gametes–embryos–larvae–juveniles–adults) that may be critical in understanding the effects of environmental stressors on marine species (Marsahill and Morgan 2011). Carry-over effects between life stages can significantly affect the outcome of environmental stress in subsequent life stages.

### **6.4. Role of Bangladesh**

Decision makers need to take ocean acidification threat seriously, and to carefully consider this environmental challenge in the context of sustainable development. We recommend to take actions that promote:

- Campaign and advocacy at global and regional levels to reduce current rate of ocean acidification by significantly lowering the absolute rate of anthropogenic carbon dioxide emissions
- Improvement of the current state of knowledge

of this problem by commitment to research and long-term monitoring of the chemical, biological, and ecosystem impacts of ocean acidification and development of a global ocean acidification observation network

- Continued research to inform ocean acidification adaptation strategies for marine ecosystems and human communities
- Global and regional supports in the form of fund, technology, equipment and human development to promote adaptation capacity with resilient infrastructure.
- Develop area-specific land use models addressing settlement, livelihood options and community places.

Engineering and social consensus on adaptation is a prerequisite for successful adaptation planning, and reaching out to all stakeholders especially those that suffer dis-proportionally. Successful national adaptation planning approaches have to have meaningful engagement and consultation approaches at their centre.

At least three well-equipped laboratories at southeast, central and southwest coastal regions of Bangladesh need to establish to make the precise pH measurements and trend analysis. In this connection, Institute of Marine Sciences and Fisheries from Chittagong University, Department of Marine Science from Noakhali University and Marine Resources Technology Discipline from Khulna University would be the best choice. The existing laboratory facilities of these universities can be renovated with required equipments. It is expected the changes to see are about 0.002 pH units per year.

### **Monitoring networks**

It is highly recommended that Bay of Bengal region countries need to develop oceanic data collection facilities. The “Bay of Bengal Large Marine Ecosystems” project (BOBLEM, [www.boblem/fao.org](http://www.boblem/fao.org)), established under an FAO programme should also be involved. Existing long-time series should be maintained and new sensors and sampling/collecting devices should be developed. The biological pump should be monitored for changes in organic carbon and calcium carbonate production and export, as it has the potential to decrease oceanic natural carbon sequestration. Such data should then be kept access to academic and research organizations of the surrounding countries for further analysis and interpretation.

### **Dissemination, outreach and capacity building**

Communicating on ocean acidification to stakeholders (policy makers, funders, media, public, etc.) is important. Clear messages should be defined

and used consistently across the sector, including confidence limits. A system should be put in place to share best practices and success stories.

- **Communication to research and potential industry**

Many research funding organizations are already aware of ocean acidification and are funding research in this area. However, they are not necessarily aware of the full spectrum of European activities in this area or the gaps in research. Therefore, European research relevant to ocean acidification is too fragmented and poorly coordinated. A strong effort by Europe to coordinate its own ocean acidification research would stimulate activities while enabling resources to be optimized by eliminating gaps and duplications. Equally important will be the social and economic consequences of ocean acidification, meaning that communication with the general public and all potential funders is crucial.

- **Communication to the climate change research community**

Surprisingly, many researchers in climate change, including those working on mitigation and adaptation strategies, are unaware of many of the known implications of ocean acidification. The ocean acidification community has made inroads to reach members of the global change community, who is not always fully aware of the impacts of ocean acidification (e.g., the IPCC fourth Assessment Report on Climate Change in 2007, the Stern Report on the Economics of Climate Change, several scientific statements and declarations on ocean acidification). Impacts of ocean acidification should be included in all future climate change assessments. The certainty associated with ocean acidification, due to the simplicity of the chemical reaction of CO<sub>2</sub> and water, provides a powerful supplementary reason adding to the certainty of climate change.

- **Communication and Interactions with the international research community**

It is recommended that the European ocean acidification community strengthens links with the international research community in order to stimulate research, share resources, and avoid needless redundancy; for instance, taking full advantage of PAGES, SCOR, GLOBEC, OCCC, the International Council for the Exploration of the Sea (ICES), the Marine Environmental Simulation Study for Future Projection of Marine Ecosystems (funded by the Japan Science and Technology Agency), the US Ocean Carbon and Biogeochemistry program (OCB) as well as some

of the ongoing NSF, NASA and NOAA-funded, scientific activities, and a few research projects in Australia and New Zealand.

- **Communication to Stakeholders**

In order to further address socio-economic implications, there is an essential need to establish two-way communication with stakeholders to better understand their problems and questions so that research can be properly designed and information conveyed in a useful manner. These stakeholders include fisheries and marine habitat managers, fishermen, community leaders, conservation groups, etc. Similarly communication with the media and the general public is also important.

### ***Coordination***

Coordination of ocean acidification research needs to address:

- Coordination and collaboration of research programmes relevant at national level and at the regional level, taking into account the very different marine environments in Europe (Arctic, Baltic, North Sea, Mediterranean, Black Sea, North-East Atlantic) and of EU projects related to ocean acidification;
- Coordination and collaboration at the international level of European activities with those outside Europe (e.g., United States, Japan, Korea, Australia, China). While coordination within Europe is underway, it will be important to integrate European activities with the global effort on ocean acidification research. European science (and the science of other countries) will benefit from such coordination, including stimulation of new ideas by bringing together scientific perspectives from around the world, sharing of physical resources (e.g., ships, mesocosms, new equipment designs), standardization of protocols and instrumentation, possibilities for coordination of related national efforts to identify overlaps and gaps, and access to parts of the world where European scientists might benefit from other expertise and past experiences.

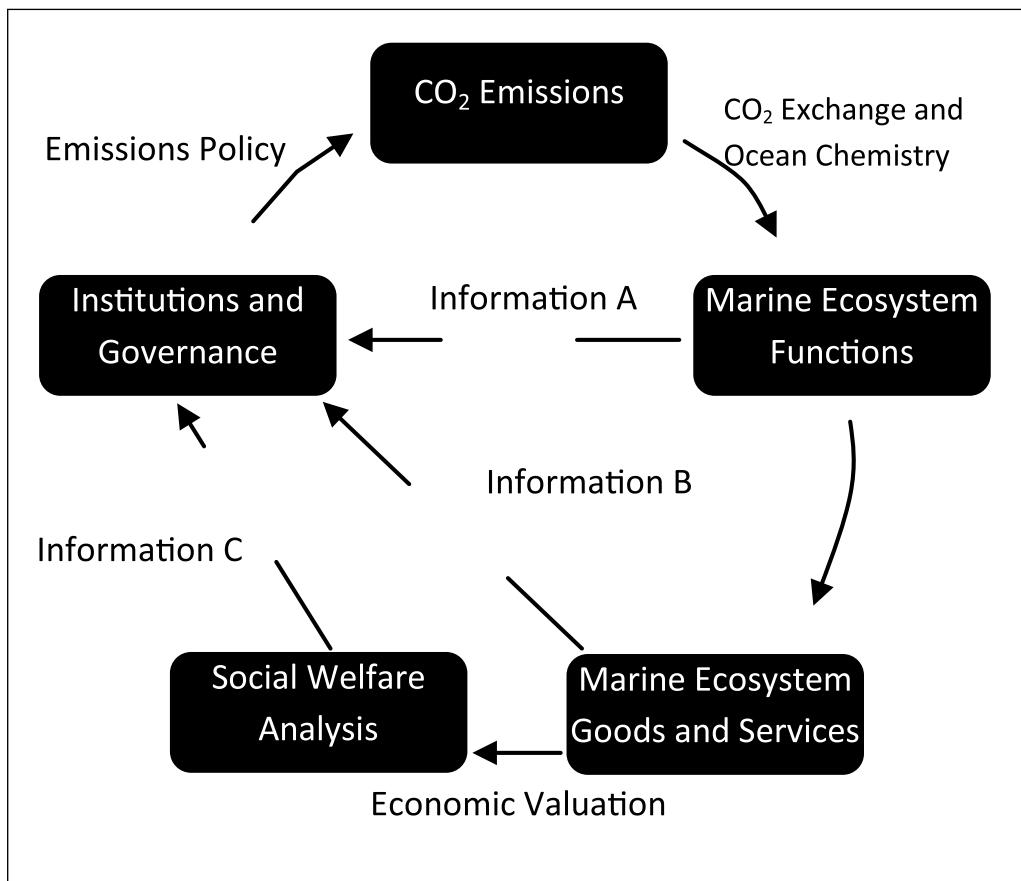
## **6.5. International policies and institutional frameworks**

The international distributional challenges of acidification are significant. Developed nations are responsible for the largest portion of historic emissions and therefore for current acidification. They also possess the most diverse economies and food systems. Transfer schemes to compensate developing nations particularly affected by acidification may be warranted. Mechanisms to achieve this include the Adaptation Fund, established

by the parties to the Kyoto Protocol (UNFCCC, 2010a). The Fund supports adaptation activities in developing countries with Clean Development Mechanism (CDM) funding from developed countries. However, with the expiration of the Kyoto Protocol in 2012, there is uncertainty about the future of the CDM and hence, the Adaptation Fund. The Copenhagen Accord, agreed to by 114 parties in 2009, recommitted developed countries to providing financial resources to support adaptation measures in developing countries (UNFCCC, 2010b). Unfortunately, the Obama administration's proposed funding for international adaptation aid has not found support in the current U.S. Congress (Friedman, 2011). This lack of support by the U.S. may deter other nations from meeting their commitments. However, if the impacts of acidification in developing nations intensify, developed nations may face increased pressure to meet these commitments.

other sectors of the economy. Higher energy prices, for example, may slow economic growth, meaning a lower standard of living for many people in the near term. So when policy makers consider emissions policies, they need to weigh the benefits that will be realized far into the future with the social costs that will begin to accrue almost immediately. Whether emission reductions are achieved through economic instruments, such as a carbon tax or cap and trade programme, or through technology mandates, those policies are likely to carry with them significant costs (e.g. OECD 2008; Paltsev et al. 2009; Tol 2010). At the same time, effects of OA and climate change will likely play out on individual marine resources, disproportionately affecting certain groups of people with high dependence on affected ecosystems.

Figure 29 is a conceptual schematic showing how policy decisions, ecological impacts and social



**Figure 29.** CO<sub>2</sub> impact pathway and flow of information (source: Hilmi et al. 2012)

When designing efficient policies to address CO<sub>2</sub>-linked environmental change, issues spanning multiple temporal and spatial scales must be taken into consideration. Reductions in CO<sub>2</sub> emissions large enough to avoid the worst consequences of OA and climate change will require global economy-wide emissions policies and international agreements (Turley and Gattuso 2012). These are likely to be costly and necessitate trade-offs within

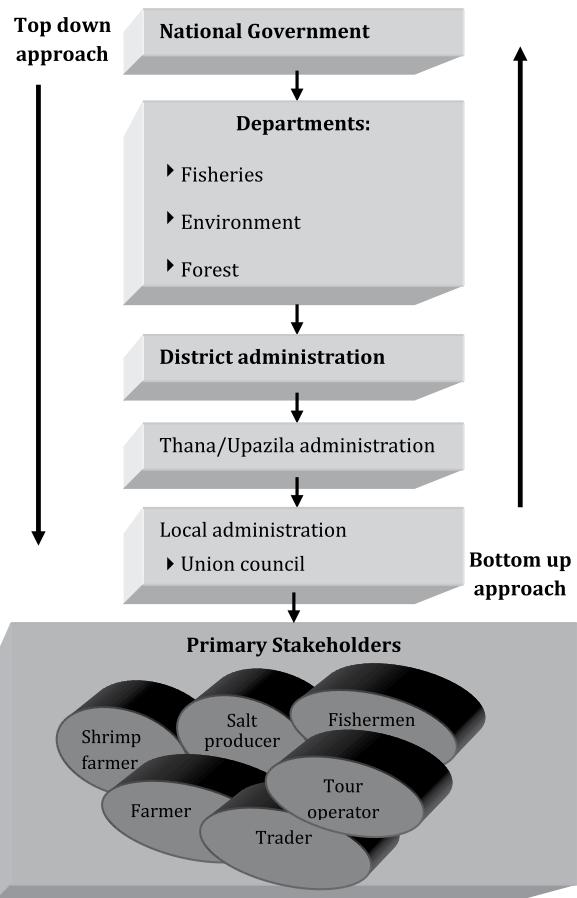
welfare are related in the context of CO<sub>2</sub> emissions-linked ocean changes such as ocean acidification. The solid arrows represent pathways through which CO<sub>2</sub> emissions impact marine ecosystems and the human systems that rely on them. The dashed lines are flows of information to policy makers whose charge is to maximize social welfare. Policy makers can base decisions on observations of the natural system (information A and B) or on

economic analyses that express costs and benefits of emissions policy in monetary terms (information C). The latter provides the most comprehensive set of information but also requires knowledge embodied in information A and B. This figure also demonstrates the relevance of economic valuation and shows how knowledge of welfare impacts can be used directly by the policy maker, or the natural resource manager, to support policy action since the absence of action will often be associated with high welfare damages (as quantified by the information flow C).

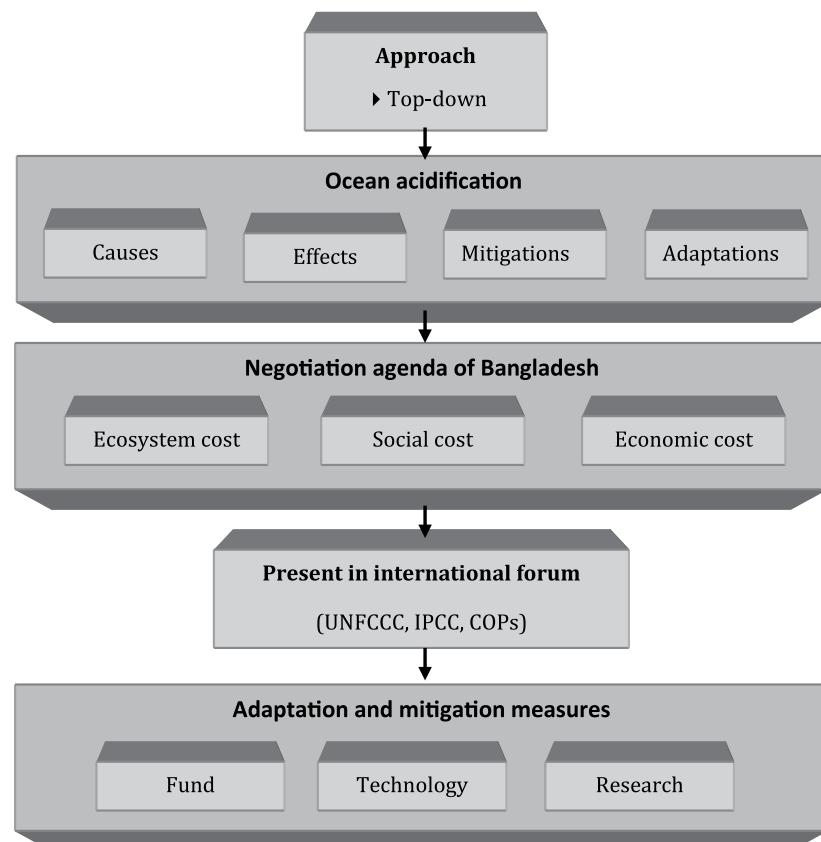
## 6.6. Negotiating strategies for Bangladesh at the international level

It is necessary to analyze the impacts and adaptation process of ocean acidification following the ‘top-down and bottom-up’ strategy allowing participations of relevant stakeholder groups (Figure 30). The vertical integration refers to integration between local level and national or international activities and policies. The horizontal integration refers to integration between different sectors (such as fisheries and forestry).

The impacts and adaptation of ocean acidification with ecosystem, social and economic costs need to formally present in international forum. Bangladesh needs adequate advocacy initiatives at the international level to establish the roles of developed nations on climate justice, vulnerable



**Figure 30.** The ‘top-down and bottom-up’ approach for deploying ocean acidification impacts and adaptation process (after Hossain and Shamsuddoha, 2006)



livelihoods and inadequate adaptation options of developing nations. In this connection, negotiation options can be in the form of fund, technology and collaborative research works (Figure 31). Problem tree analysis for ocean acidification indicates relationships of fossil fuel combustion, water pH, ocean ecosystem and people livelihood (Figure 32).

**Figure 31.** The ocean acidification mitigation and adaptation strategies for Bangladesh at the international level

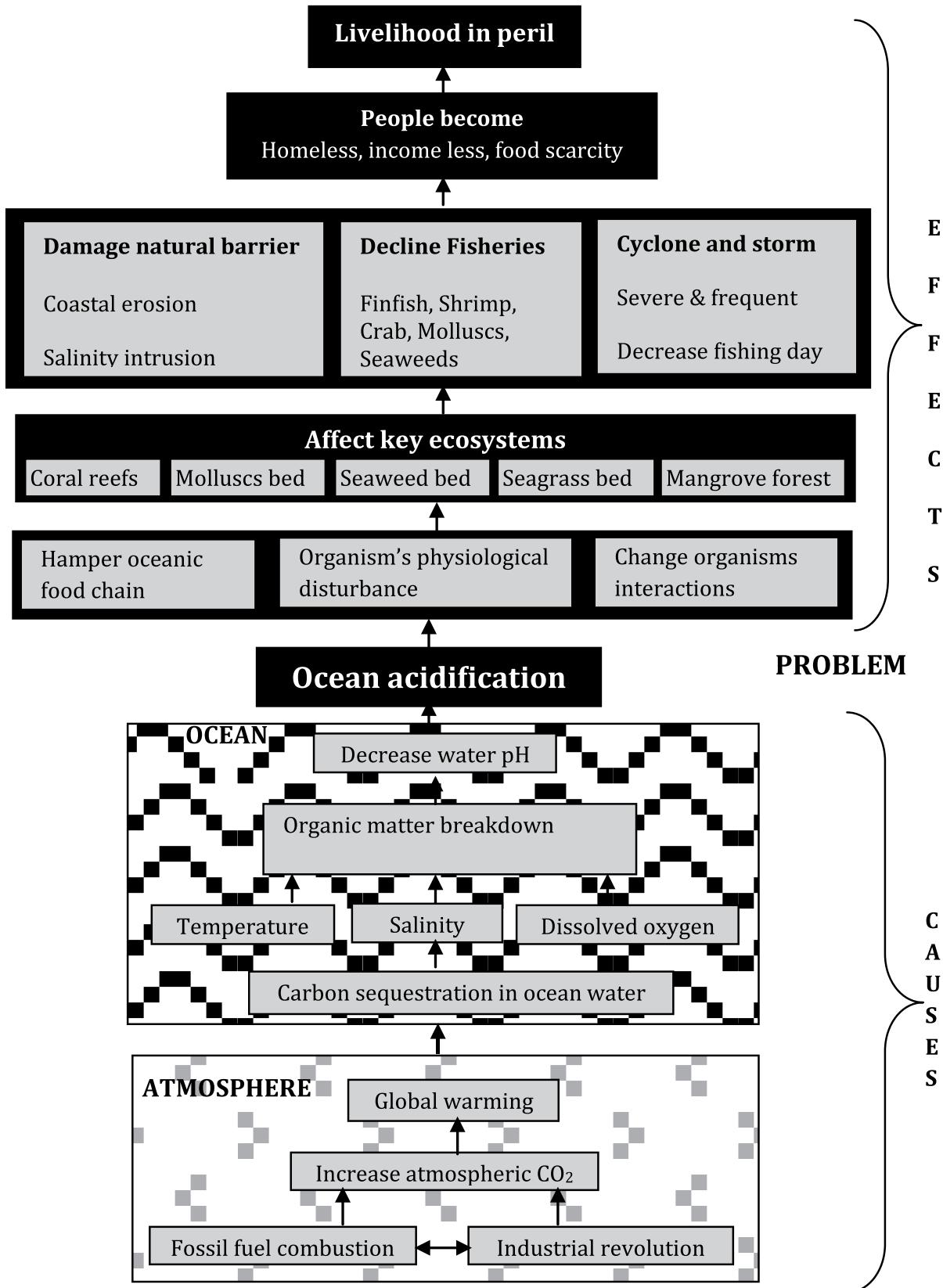


Figure 32. Problem tree analysis shows the causes and effects for ocean acidification

## 7. Conclusions and Recommendations

Although the changes in seawater chemistry due to oceanic uptake of anthropogenic CO<sub>2</sub> are well characterized, the impacts of ocean acidification on the biological compartment of ocean, i.e. on marine flora and fauna are still poorly understood. New technologies and skills, as well as integrated, multidisciplinary efforts among biologists, chemists and mathematicians are much important to assess and quantify the effects of ocean acidification on marine fauna and the changes in ecosystem structure and function.

Evidence suggests that ocean acidification is happening now and probably will continue to intensify, if global CO<sub>2</sub> emissions are not reduced. Given the potential threat to marine ecosystems and the subsequent impact on human society and economy, especially as it acts in conjunction with anthropogenic global warming, there is an urgent need for immediate action. This 'double trouble' of climate change and ocean acidification is arguably the most critical environmental issue facing humans at the present time. The key recommendations in order to understand and respond to the impacts of ocean acidification are as follows:

- A robust observing network to establish a baseline, and to detect and predict biochemical changes (i.e. from organism to ecosystem level) attributable to acidification. To develop the observing network, it is necessary to identify the appropriate chemical and biological parameters to be measured by the network, ensuring data quality and consistency across space and time.
- Conduct research to fulfil critical information needs. Capacity building (i.e. education and training) of ocean acidification researchers.
- Integrate the researchers of natural and social sciences to help mitigate ocean acidification and to develop adaptation strategies, taking into account the potential impacts on both natural resources and human, including socio-economics.
- Total costs of abating CO<sub>2</sub> emissions, and of carbon capture and storage/sequestration should be considered and evaluated.
- Ensure adequate monitoring of key marine ecosystems and environmental services.
- Facilitate dissemination, communication and capacity building to help deliver scientific knowledge-based advice to relevant stakeholders, and to share success stories with the general public.

- Coordinate and strengthen the regional and national research on the impacts of ocean acidification, including sharing of research infrastructure, resources and knowledgebase.

### *Need for development of best practices for ocean acidification research*

Many of the approaches for ocean acidification research are new and there are currently no common guidelines for best practices. Therefore, it is important to develop common standards so that research carried out by different investigators can be easily compared. Dickson et al. (2007) provided an internationally agreed-upon guide to best practices for carbon measurements, but other measurements relevant to ocean acidification are also need to be standardized.

### *Data management*

Data management and dissemination for ocean acidification research must be planned before new data are collected. Data must be reported and archived in ways that make them readily accessible now and in the future. Likewise, data mining and archiving of historical data may provide useful insights into the evolution of ocean acidification over time.

### *Education, outreach, and engagement of policymakers*

It has recognized the need for interdisciplinary training for policy makers, researchers, and managers to improve research, observations, modeling, and data management. Meetings should be held with coral reef managers, fisheries managers, other stakeholders, and policymakers to engage them in formulating research questions that will enable scientists to provide results relevant for decision making. Readily accessible presentations and fact sheets on ocean acidification and its effects on marine life should be created for the public and schools, and made available via posters, stickers, holding boards and websites.

### *Communication within the oceanography community*

The research community needs to hear about 'mistakes' and conditions that produced no significant effects, so that public can learn from them. Dissemination of results and conclusions should be accelerated. Communication among disciplines relevant to ocean acidification research should be extended.

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**Appendix-A.** Monthly water pH variation in rivers and estuaries of coastal Bangladesh

| Location                  | Jan  | Feb  | Mar  | Apr | May  | Jun  | Jul  | Aug | Sep | Oct  | Nov | Dec                    | Ref |
|---------------------------|------|------|------|-----|------|------|------|-----|-----|------|-----|------------------------|-----|
| Meghna estuary            |      |      |      |     |      |      |      |     |     |      |     |                        |     |
| Meghna estuary            | 7.3  | 7.4  | 7.3  | -   | -    | -    | -    | -   | -   | 7.5  | 7.5 | Hossain et al. 2013    |     |
| Karnafully river estuary  | 7.3  | 7.5  | 7.3  | 7.2 | 7.1  | 7.2  | 7.1  | 6.9 | 7.2 | 7.5  | 7.7 | Mahmood and Khan 1976  |     |
| Karnafully river estuary  | -    | -    | 2.51 | 2.7 | 2.01 | 2.71 | -    | -   | -   | -    | -   | Alam 2000              |     |
| Kutubdia channel          | 8.22 | 7.95 | 7.14 | 7.9 | 7    | 7.5  | 7.14 | 8   | 7.1 | 7.16 | 8.3 | Zafar and Mahmood 1997 |     |
| Kutubdia channel          | -    | -    | -    | 8.5 | 8.9  | 8.7  | 7.5  | 8.1 | 7.8 | 7.6  | -   | ECOBAS 2012            |     |
| Moheshkhali channel       | 8.3  | 8.1  | 7.5  | 7.5 | 8.2  | 7.9  | 8.2  | 7   | 7.7 | 8.1  | -   | Mahmood et al. 1985    |     |
| Moheshkhali channel       | 8.2  | 8.4  | 8    | 7.5 | 7.3  | 7.1  | 7.2  | 7.3 | 7.1 | 7.6  | 7.9 | Hossain et al. 2003    |     |
| Moheshkhali channel       | -    | -    | -    | 8.6 | 8.5  | 8.5  | 7.5  | 7.5 | 7.5 | 7.5  | -   | ECOBAS 2012            |     |
| Bakkhali river estuary    | 7.6  | 7.4  | 7.3  | 7.2 | 7.3  | 7.4  | 7.3  | 7.6 | 7.7 | 7.6  | 7.7 | Mahmood and Khan 1980  |     |
| Bakkhali river estuary    | 6.2  | 6.1  | 6.2  | 7.1 | 7.9  | 7.8  | 7.4  | 7.9 | 7.4 | 6.7  | 6.2 | Rashid 1999            |     |
| Naaf river estuary        | -    | -    | 8.1  | 7.8 | 8    | 7.6  | 7.5  | 7.7 | 7.5 | -    | -   | Chowdhury et al. 2009  |     |
| Mathamuhuri river estuary |      |      |      |     |      |      |      |     |     |      |     |                        |     |

**Appendix-B.** Monthly water temperature variation in rivers and estuaries of coastal Bangladesh

| Location                  | Jan   | Feb   | Mar   | Apr   | May   | Jun   | Jul   | Aug   | Sep   | Oct   | Nov   | Dec                 | Ref                    |
|---------------------------|-------|-------|-------|-------|-------|-------|-------|-------|-------|-------|-------|---------------------|------------------------|
| Meghna estuary            | -     | 0     | 0     | 0     | 0     | -     | -     | -     | -     | 0     | 0     | Hossain et al. 2013 |                        |
| Karnafully river estuary  | 3.8   | 5.58  | 6.24  | 4.34  | 2.95  | 1.82  | 0.4   | 1.44  | 0.69  | 1.3   | 1.32  | 1.25                | Mahmood and Khan 1976  |
| Karnafully river estuary  | -     | -     | 0.3   | 0.3   | 1.6   | 1     | -     | -     | -     | -     | -     | -                   | Alam 2000              |
| Kutubdia channel          | 30    | 29.45 | 25    | 26    | 25    | 18    | 9.72  | 10.79 | 11    | 20    | 25.57 | 27.53               | Zafar and Mahmood 1997 |
| Kutubdia channel          | -     | -     | -     | 13    | 15    | 15    | 11    | 10    | 14    | -     | -     | -                   | ECOBAS 2012            |
| Moheshkhali channel       | 28    | 27    | 25.9  | 25    | 22.7  | 21    | 21    | 20    | 28    | 27    | -     | -                   | Mahmood et al. 1985    |
| Moheshkhali channel       | 29.1  | 28.8  | 28.05 | 29.42 | 25.09 | 22.7  | 22    | 20.9  | 23.85 | 27.3  | 28.55 | 29.8                | Hossain et al. 2003    |
| Moheshkhali channel       | -     | -     | -     | 14    | 15    | 16    | 12    | 15    | 18    | -     | -     | -                   | ECOBAS, 2012           |
| Bakkhali river estuary    | 22.5  | 23.9  | 26.7  | 30.0  | 30.2  | 29.6  | 28.9  | 27.7  | 28.8  | 28.6  | 26.5  | 24.8                | Mahmood and Khan 1980  |
| Bakkhali river estuary    | 32.18 | 34.71 | 33.47 | 22.01 | 12.81 | 5.81  | 16.07 | 14.9  | 15.58 | 18.51 | 22.13 | 28.08               | Rashid 1999            |
| Naaf river estuary        | -     | -     | 29.1  | 21.2  | 2.2   | 5.5   | 2.3   | 1.2   | 3     | 10.1  | -     | -                   | Chowdhury et al. 2009  |
| Mathamuhuri river estuary | 19.5  | 26.24 | 32.56 | 34.13 | 16.49 | 10.74 | 2     | 2.06  | 1.92  | 3.34  | 11.89 | 13.8                | Mahmood 1990           |

**Appendix-C. Monthly water salinity variation in rivers and estuaries of coastal Bangladesh**

| Location                  | Jan   | Feb   | Mar   | Apr   | May   | Jun   | Jul   | Aug   | Sep   | Oct   | Nov   | Dec   | Ref                    |
|---------------------------|-------|-------|-------|-------|-------|-------|-------|-------|-------|-------|-------|-------|------------------------|
| Meghna estuary            | -     | 0     | 0     | 0     |       |       |       |       |       |       |       |       | Hossain et al. 2013    |
| Karnafully river estuary  | 3.8   | 5.58  | 6.24  | 4.34  | 2.95  | 1.82  | 0.4   | 1.44  | 0.69  | 1.3   | 1.32  | 1.25  | Mahmood and Khan 1976  |
| Karnafully river estuary  | -     | -     | 0.3   | 0.3   | 1.6   | 1     | -     | -     | -     | -     | -     | -     | Alam 2000              |
| Kutubdia channel          | 30    | 29.45 | 25    | 26    | 25    | 18    | 9.72  | 10.79 | 11    | 20    | 25.57 | 27.53 | Zafar and Mahmood 1997 |
| Kutubdia channel          | -     | -     | -     | 13    | 15    | 15    | 11    | 10    | 14    | -     | -     | -     | ECOBAS 2012            |
| Moheshkhali channel       | 28    | 27    | 25.9  | 25    | 22.7  | 21    | 21    | 20    | 28    | 27    | -     | -     | Mahmood et al. 1985    |
| Moheshkhali channel       | 29.1  | 28.8  | 28.05 | 29.42 | 25.09 | 22.7  | 22    | 20.9  | 23.85 | 27.3  | 28.55 | 29.8  | Hossain et al. 2003    |
| Moheshkhali channel       | -     | -     | -     | 14    | 15    | 16    | 12    | 15    | 18    | -     | -     | -     | ECOBAS 2012            |
| Bakkhali river estuary    | 26.8  | 24.4  | 23.6  | 23.3  | 23.8  | 21.3  | 21.0  | 20.5  | 23.2  | 24.7  | 25.2  | 25.3  | Mahmood and Khan 1980  |
| Bakkhali river estuary    | 32.18 | 34.71 | 33.47 | 22.01 | 12.81 | 5.81  | 16.07 | 14.9  | 15.58 | 18.51 | 22.13 | 28.08 | Rashid 1999            |
| Naaf river estuary        | -     | -     | 29.1  | 21.2  | 2.2   | 5.5   | 2.3   | 1.2   | 3     | 10.1  | -     | -     | Chowdhury et al. 2009  |
| Mathamuhuri river estuary | 19.5  | 26.24 | 32.56 | 34.13 | 16.49 | 10.74 | 2     | 2.06  | 1.92  | 3.34  | 11.89 | 13.8  | Mahmood 1990           |

**Appendix-D.** Monthly dissolved oxygen variation in rivers and estuaries of coastal Bangladesh

| Location                  | Jan  | Feb  | Mar  | Apr  | May  | Jun  | Jul  | Aug  | Sep  | Oct  | Nov  | Dec          | Ref                    |
|---------------------------|------|------|------|------|------|------|------|------|------|------|------|--------------|------------------------|
| Meghna estuary            | -    | 4.9  | 5    | 5.5  | 5.6  | -    | -    | -    | -    | -    | 5    | 5            | Hossain et al. 2013    |
| Karnafully river estuary  | 4.97 | 4.16 | 3.37 | 3.36 | 4.38 | 4.23 | 4.34 | 3.93 | 3.62 | 4.43 | 4.35 | 4.96         | Mahmood and Khan 1976  |
| Karnafully river estuary  | -    | -    | 7.6  | 7.7  | 8.1  | 7.2  | -    | -    | -    | -    | -    | -            | Alam 2000              |
| Kutubdia channel          | 8.34 | 5.35 | 3.82 | 4.97 | 5.35 | 3.51 | 4.71 | 6.97 | 6.99 | 5.28 | 4.63 | 8.35         | Zafar and Mahmood 1997 |
| Kutubdia channel          | -    | -    | -    | 4.5  | 4.4  | 4.2  | 4.6  | 5.1  | 4.7  | -    | -    | -            | ECOBAS 2012            |
| Moheshkhali channel       | 8.1  | 7.5  | 7    | 6.5  | 7.6  | 7.8  | 7.5  | 6.5  | 7    | 7.9  | -    | -            | Mahmood et al. 1985    |
| Moheshkhali channel       | -    | -    | -    | 4.6  | 4.9  | 4.0  | 5.2  | 5.3  | 5.2  | -    | -    | -            | ECOBAS 2012            |
| Bakkhali river estuary    | 4.15 | 3.98 | 3.65 | 3.30 | 3.30 | 3.80 | 3.65 | 4.20 | 4.35 | 4.25 | 4.40 | 4.30         | Mahmood and Khan 1980  |
| Bakkhali river estuary    | 3.15 | 2.87 | 1.91 | 2.36 | 2.51 | 3.71 | 3.24 | 3.89 | 4.3  | 3.25 | 3.61 | 5.44         | Rashid 1999            |
| Naaf river estuary        |      |      |      |      |      |      |      |      |      |      |      |              |                        |
| Naaf river estuary        |      |      |      |      |      |      |      |      |      |      |      |              |                        |
| Mathamuhuri river estuary | 5.42 | 5.13 | 4.83 | 4.03 | 3.85 | 4.63 | 5.28 | 5.45 | 5.34 | 5.41 | 5.74 | Mahmood 1990 |                        |

**Appendix-E.** Examples of the response of marine fauna to ocean acidification (source: Victoria et al., 2008).

| Species                           | Description           | CO <sub>2</sub> system parameters          | Sensitivity   | Reference                            |
|-----------------------------------|-----------------------|--|---|--------------------------------------|
| <b>Planktonic Foraminifera</b>    |                       |  |   |                                      |
| <i>Orbulina universa</i>          | Symbiont-bearing      | pCO <sub>2</sub> 560–780 ppmv              | 8–14% reduction in shell mass   | Spero et al. 1997; Bijma et al. 2002 |
| <i>Globigerinoides sacculifer</i> | Symbiont-bearing      | pCO <sub>2</sub> 560–780 ppmv              | 4–8% reduction in shell mass  | Bijma et al. 2002                    |
| <b>Cnidaria</b>                   |                       |  |   |                                      |
| <i>Scyphozoa Hydrozoa</i>         | Jellyfish             | North Sea seawater pH drop from 8.3 to 8.1 | Increase in frequency as measured by CPR from 1958 to 2000            | Attrill et al. 2007                  |
| <b>Mollusca</b>                   |                       |  |   |                                      |
| <i>Clio pyramidata</i>            | Shelled pteropod      | Ωarag<1                                    | Shell dissolution   | Orr et al. 2005                      |
| <i>Mitilus edulis</i>             | Mussel                | pH 7.1 / 10 000 ppmv                       | Shell dissolution   | Lindner et al. 1984                  |
| <i>Crassostrea gigas</i>          | Oyster                | pCO <sub>2</sub> 740 ppmv                  | 10% decrease in calcification rate                                    | Gazeau et al. 2007                   |
| <i>Mitilus galloprovincialis</i>  | Mediterranean mussel  | pH 7.3, ~5000 ppmv                         | Reduced metabolism, growth rate                                       | Michaelidis et al. 2005              |
| <i>Piacopecten magellanicus</i>   | Giant scallop         | pH <8.0                                    | Decrease in fertilization and embryo development                      | Desrosiers et al. 1996               |
| <i>Pinctada fucata martensi</i>   | Japanese pearl oyster | pH 7.7                                     | Shell dissolution, reduced growth, increasing mortality               | Reviewed in Knutzen 1981             |
| <b>Arthropoda</b>                 |                       |  |   |                                      |
| <i>Acartia steueri</i>            | Copepod               | 0.2–1%CO <sub>2</sub>                      | Decrease in egg hatching success; increase in nauplius mortality rate | Kurihara et al. 2004                 |
| <i>Euphausia pacifica</i>         | Krill                 | pH , 7.6                                   | Mortality increased with increasing exposure time and decreasing pH   | Yamada and Ikeda 1999                |
| <i>Cancer pagurus</i>             | Crab                  | 1% CO <sub>2</sub> , ~10 000 ppmv          | Reduced thermal tolerance, aerobic scope                              | Metzger et al. 2007                  |
| <b>Chaetognatha</b>               |                       |  |   |                                      |
| <i>Sagitta elegans</i>            | Chaetognath           | pH<7.6                                     | Mortality increased with increasing exposure time and decreasing pH   | Yamada and Ikeda 1999                |
| <i>Echinodermata</i>              |                       |  |   |                                      |

| Species                              | Description         | CO <sub>2</sub> system parameters                                | Sensitivity  | Reference                         |
|--------------------------------------|---------------------|--|--|-----------------------------------|
| <i>Strongylocentrotus purpuratus</i> | Sea urchin          | pH ~6.2–7.3  | High sensitivity inferred from lack of pH regulation and passive buffering via test dissolution during emersion. | Burnett et al. 2002               |
| <i>Cystechinus</i> sp.               | Deep-sea urchin     | pH 7.8   | 80% mortality under simulated CO <sub>2</sub> sequestration  | Barry et al. 2002                 |
| <b>Vertebrata</b>                    |                     |  |  |                                   |
| <i>Scyllorhinus canicula</i>         | Dog fish            | pH 7.7 / 0.13% CO <sub>2</sub> 7% CO <sub>2</sub> , ~70 000 ppmv | Increased ventilation, 100% mortality after 72 h   | Truchot 1987; Hayashi et al. 2004 |
| <i>Sillago japonica</i>              | Japanese whiting    | 7% CO <sub>2</sub> , ~70 000 ppmv                                | Rapid mortality in 1-step exposure   | Kikkawa et al. 2006               |
| <i>Paralichthys olivaceus</i>        | Japanese flounder   | 5% CO <sub>2</sub> , ~50 000 ppmv                                | 100% mortality within 48 h   | Hayashi et al. 2004               |
| <i>Euthynnus affinis</i>             | Eastern little tuna | 15% CO <sub>2</sub> , ~150 000 ppmv                              | 100% mortality of eggs after 24 h  | Kikkawa et al. 2003               |
| <i>Pagrus major</i>                  | Red sea bream       | 5% CO <sub>2</sub> , ~50 000 ppmv                                | >60% larval mortality after 24 h   | Ishimatsu et al. 2005             |
| <i>Seriola quinqueradiata</i>        | Yellowtail          | 5% CO <sub>2</sub> , 50 000 ppmv                                 | Reduced cardiac output; 100% mortality after 8 h   | Ishimatsu et al. 2004             |
| <i>Dicentrarchus labrax</i>          | Sea bass            | pH 7.25, 24 mg l <sup>-1</sup> CO <sub>2</sub>                   | Reduced feed intake  | Cecchini et al. 2001              |

**Appendix-F.** Monthly occurrence of different larval fish in Sunderban mangrove ecosystem of Satkhira, Bangladesh (source: Mahmood et al. 1985)

| Species recorded                   | Months (1982-1983) |   |   |   |   |   |   |   |   |   |   |   |
|------------------------------------|--------------------|---|---|---|---|---|---|---|---|---|---|---|
|                                    | J                  | J | A | S | O | N | D | J | F | M | A | M |
| <i>Colia ramcarati</i>             |                    |   |   |   |   |   |   |   |   |   |   |   |
| <i>Setipinna phasa</i>             |                    |   |   |   |   |   |   |   |   |   |   |   |
| <i>Harpodon nehereus</i>           |                    |   |   |   |   |   |   |   |   |   |   |   |
| <i>Puntius bimaculatus</i>         |                    |   |   |   |   |   |   |   |   |   |   |   |
| <i>Ophichthys boro</i>             |                    |   |   |   |   |   |   |   |   |   |   |   |
| <i>Hemirrhampus ectuncio</i>       |                    |   |   |   |   |   |   |   |   |   |   |   |
| <i>Ichthyocarpus carce</i>         |                    |   |   |   |   |   |   |   |   |   |   |   |
| <i>Panchax melastigma</i>          |                    |   |   |   |   |   |   |   |   |   |   |   |
| <i>Mugil spp.</i>                  |                    |   |   |   |   |   |   |   |   |   |   |   |
| <i>Lates calcarifer</i>            |                    |   |   |   |   |   |   |   |   |   |   |   |
| <i>Sillago domina</i>              |                    |   |   |   |   |   |   |   |   |   |   |   |
| <i>Leiognathus ruconius</i>        |                    |   |   |   |   |   |   |   |   |   |   |   |
| <i>Leiognathus equulus</i>         |                    |   |   |   |   |   |   |   |   |   |   |   |
| <i>Pseudosciaena sp.</i>           |                    |   |   |   |   |   |   |   |   |   |   |   |
| <i>Otolithoides pama</i>           |                    |   |   |   |   |   |   |   |   |   |   |   |
| <i>Sicyopterus qymnauchen</i>      |                    |   |   |   |   |   |   |   |   |   |   |   |
| <i>Glossogobius giuris</i>         |                    |   |   |   |   |   |   |   |   |   |   |   |
| <i>Pseudopocryptes lanceolatus</i> |                    |   |   |   |   |   |   |   |   |   |   |   |
| <i>Odontarbyopus rubicandus</i>    |                    |   |   |   |   |   |   |   |   |   |   |   |
| <i>Trypauchen vaqina</i>           |                    |   |   |   |   |   |   |   |   |   |   |   |
| <i>Platycephalus scaber</i>        |                    |   |   |   |   |   |   |   |   |   |   |   |
| <i>Cynoglossus cynoglossus</i>     |                    |   |   |   |   |   |   |   |   |   |   |   |
| <i>Cynoglossus lida</i>            |                    |   |   |   |   |   |   |   |   |   |   |   |
| <i>Macrognathus aculeatus</i>      |                    |   |   |   |   |   |   |   |   |   |   |   |
| <i>Batrachus grunniens</i>         |                    |   |   |   |   |   |   |   |   |   |   |   |
| Unidentified                       |                    |   |   |   |   |   |   |   |   |   |   |   |

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