

## Exercise sheet

In this practical you will:

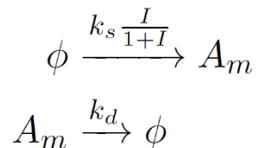
- Look at the effect of transcriptional leakiness on dose-response curves;
- Study feedback and autoregulation of transcriptional processes, and
- Observe time scale separation (as we did for the derivation of Michaelis-Menten) in another context.

Solutions will be made available on Brightspace.

### Part 1: TRANSCRIPTION LEAKINESS

Leakiness is a term often used in synthetic biology to describe background levels of transcription that takes place unregulated by an activating or inhibiting transcription factor. This mechanism allows for a measure of control. For example, you want an inhibiting transcription factor to completely shut down transcription of a lethal compound. Another example is the background transcription of an enzyme that can deal with a certain substrate to allow for a quick response to when that substrate becomes available.

Let us take a look at the effect of leakiness on dose-response curves. We first consider a system in the absence of background transcription. Consider an inducer  $I$  that promotes transcription of  $A_m$ . In yesterday's lecture we saw the following reactions:



**Q1:** Write the above chemical reactions as one ODE.

**Result:**  $\frac{dA_m}{dt} = k_s \frac{I}{1+I} - k_d A_m$

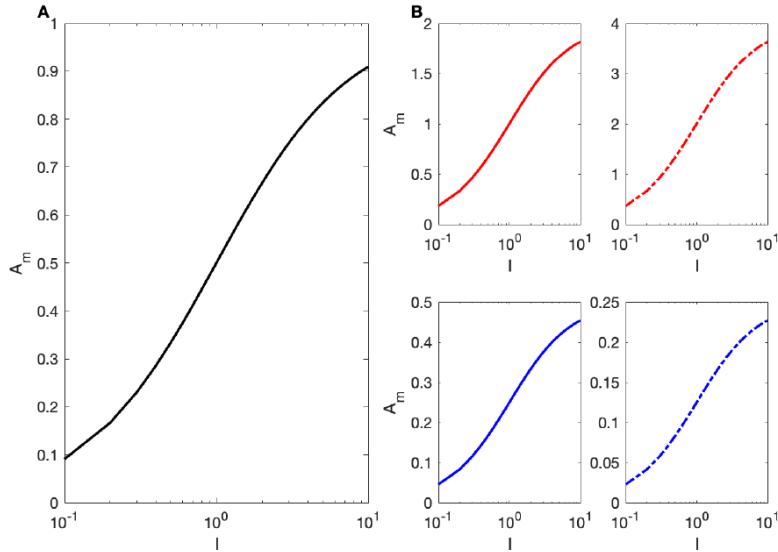
**Q2:** Solve this ODE to find the steady state of  $A_m$ , and explain from the expression why this steady state can ever be negative.

**Steady state:**  $k_s \frac{I}{1+I} - k_d A_m = 0 \Rightarrow k_s \frac{I}{1+I} = k_d A_m \Rightarrow A_m = \frac{k_s}{k_d} \cdot \frac{I}{1+I}$

Because  $I \geq 0$ , and the reaction rates  $k_s \geq 0$ ,  $k_d \geq 0$ ,  $A_m \geq 0$  always.

**Q3:** Set  $k_s = k_d = 1$ . Implement the function in Jupyter and plot  $A_m = f(I)$  for  $I = 0, I = 0.1$ , and  $I = 10$ . Re-scale the  $x$ -axis using a logarithmic scale. (*Hint: The resulting curve should be S-shaped.*)

Resulting graph(s):



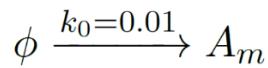
**Figure 1:** Panel A shows the solution with a logarithmic x-axis where  $k_s = k_d = 1$ . Panel B shows how the solution changes if  $k_s$  and  $k_d$  are changed. Top left:  $(k_s, k_d) = (2, 1)$  or  $(k_s, k_d) = (1, 0.5)$ . Top right:  $(k_s, k_d) = (2, 0.5)$ . Bottom left:  $(k_s, k_d) = (1, 2)$  or  $(k_s, k_d) = (0.5, 1)$ . Bottom right:  $(k_s, k_d) = (0.5, 2)$ .

Look at panel A.

**Q4:** Systematically vary  $k_s$  and  $k_d$  by dividing or multiplying each parameter by 2, either individually or both at the same time. What happens to the resulting dose-response curve?

Look at panel B. This should not be surprising, given  $\frac{k_s}{k_d}$ .

Yesterday we also added background translation – leakiness – to the chemical equations. This was depicted as:



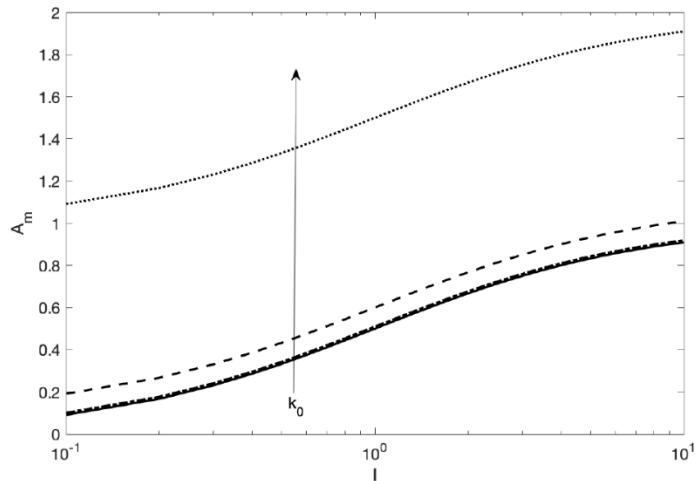
**Q5:** Modify the ODE from Q1 to include this reaction, and calculate the steady state.

Result:  $\frac{dA_m}{dt} = k_0 + k_s \frac{I}{1+I} - k_d A_m$

Steady state:  $k_0 + k_s \frac{I}{1+I} - k_d A_m = 0 \Rightarrow k_0 + k_s \frac{I}{1+I} = k_d A_m \Rightarrow A_m = \frac{k_0}{k_d} + \frac{k_s}{k_d} \cdot \frac{I}{1+I}$

**Q6:** Plot the dose-response curve for this modified model, and plot the curves for  $k_0 = 0.1$  and  $k_0 = 1$ .

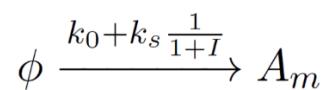
**Result:**



**Figure 2:** Dose-response curve of  $A_m$  against  $I$  for varying levels of background transcription  $k_0$ .

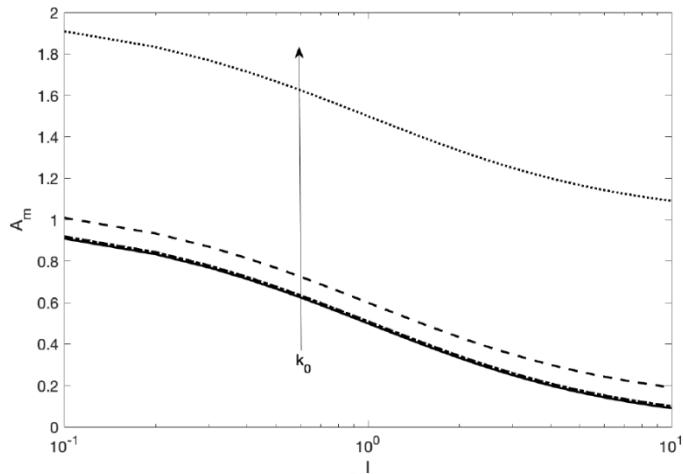
I.e., with increasing background transcription the curve is shifted upward. This is likely what you would expect: with  $I = 0$  the production is fully determined by the leaky transcription.

Yesterday we also looked at how to assume inhibition rather than activation. Assume the added reaction is now:



**Q7:** How does the dose-response curve change now? (*Hint: Modify the ODE, calculate the steady state, and implement the model in Jupiter like before.*)

**Result:**



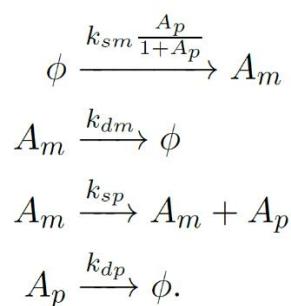
**Figure 3:** Dose-response curve of  $A_m$  against  $I$  for varying levels of background transcription  $k_0$  with an inhibitory inducer.

It is essentially the mirrored image of the previous model result.

## Part 2: AUTOREGULATION AND FEEDBACK

Now, we will replace the inducer of the previous exercise with a transcription factor. This way we can construct autoregulatory feedback loops on transcription. This is commonly referred to as a network motive; its biological mechanism is often found in nature, for instance, roughly half of the ca. 200 transcription factors in *E. coli* are autoregulated (Gao & Stock, 2018, DOI:10.1016/j.celrep.2018.08.023).

Given are the following reactions for positive autoregulation:



**Q8:** Give the two ODEs for the above chemical equations for positive autoregulation.

First equation:  $\frac{dA_m}{dt} = k_{sm} \frac{A_p}{1+A_p} - k_{dm} A_m - k_{sp} A_m + k_{sp} A_m = k_{sm} \frac{A_p}{1+A_p} - k_{dm} A_m$

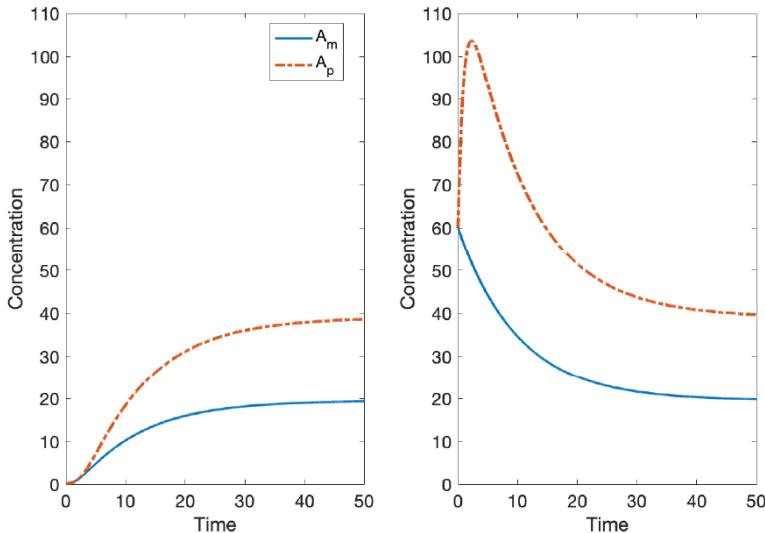
Second equation:  $\frac{dA_p}{dt} = k_{sp} A_m - k_{dp} A_p$

**Q9:** Assume  $k_{sm} = k_{sp} = 2$ ,  $k_{dm} = 0.1$ , and  $k_{dp} = 1$ . Simulate the model in Jupyter with initial conditions  $A_m(0) = A_p(0) = 0.1$  for a time span from  $t_0 = 0$  to  $t_0 = 50$ . Does the system reach a steady state, and if yes, what is it? Plot the time series to show.

See the results of Q10 below.

**Q10:** Consider the same model with the same settings as in Q9, but with initial conditions  $A_m(0) = A_p(0) = 60$ . Does the system reach a steady state now, and if yes, what is it? Plot the time series to show.

Result:



**Figure 4:** Time-series of  $A_m$  and  $A_p$  for (left)  $A_m(0) = A_p(0) = 0.1$  and (right)  $A_m(0) = A_p(0) = 60$ .

So yes, with the other initial conditions the same steady state is reached.

**Q11:** Now create a plot in which the steady state value of  $A_p$  is a function of  $k_{dm}$ . Consider increasing values of  $k_{dm}$ , for instance, 0.15, 0.2, 0.25, 0.5, and 1. Each time restart the simulations at  $A_m(0) = A_p(0) = 60$ . What happens to the steady state value of  $A_p$  as  $k_{dm}$  increases?

Result:

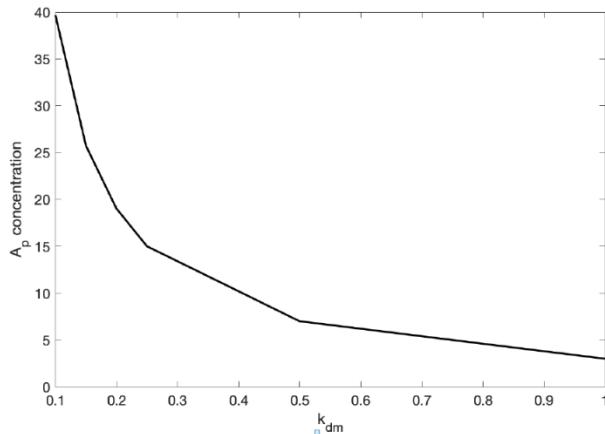


Figure 6: Plot of  $A_p$  concentration against mRNA degradation rate  $k_{dm}$ . The plot shows that as  $k_{dm}$  increases, the final concentration of  $A_p$  decreases.

Hence,  $A_p$  decreases as  $k_{dm}$  increases.

**Q12:** Show algebraically that the steady state of  $A_p$  is a function of  $k_{dm}$ . (*Hint: Follow the next steps: Substitute the numerical values of all parameters, except  $k_{dm}$ ; Solve  $\frac{dA_m}{dt} = 0$  to get  $A_m = f(A_p)$ ; Solve  $\frac{dA_p}{dt} = 0$  to get  $A_p = g(A_m)$ ; Substitute  $f(A_p)$  into  $A_p = g(A_m)$  to get  $A_p = h(A_p)$ , then rearrange and solve.*)

Substitute parameter values:  $\frac{dA_m}{dt} = 2 \frac{A_p}{1+A_p} - k_{dm}A_m$ ;  $\frac{dA_p}{dt} = 2A_m - A_p$

Second hint:  $2 \frac{A_p}{1+A_p} - k_{dm}A_m = 0 \Rightarrow 2 \frac{A_p}{1+A_p} = k_{dm}A_m \Rightarrow A_m = \frac{2}{k_{dm}} \cdot \frac{A_p}{1+A_p}$

Third hint:  $2A_m - A_p = 0 \Rightarrow A_p = 2A_m$

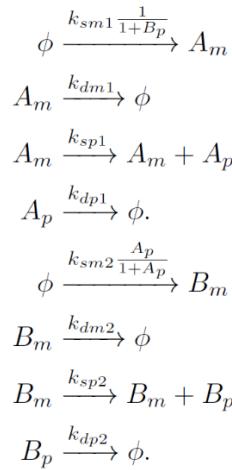
Fourth hint:  $A_p = 2A_m = 2 \left( \frac{2}{k_{dm}} \cdot \frac{A_p}{1+A_p} \right) = \frac{4}{k_{dm}} \cdot \frac{A_p}{1+A_p}$

Rearranging:  $A_p = \frac{4}{k_{dm}} \cdot \frac{A_p}{1+A_p} \Rightarrow A_p(1 + A_p) = \frac{4A_p}{k_{dm}} \Rightarrow A_p(1 + A_p) - \frac{4A_p}{k_{dm}} = A_p \left( (1 + A_p) - \frac{4}{k_{dm}} \right) = 0$

This gives  $A_p = 0$  (biologically possible) or  $(1 + A_p) - \frac{4}{k_{dm}} = 0 \Rightarrow 1 + A_p = \frac{4}{k_{dm}} \Rightarrow A_p = \frac{4}{k_{dm}} - 1$

As  $k_{dm}$  is in the denominator, increasing it will decrease  $A_p$ , i.e., the steady state.

While the previous was an example with feedback involving a single component, many feedback loops in biological systems involve two or more components. Consider the following system with the reactions:



**Q13:** Give the ODEs of this system and simulate the system from  $t_0 = 0$  to  $t_0 = 100$ , setting  $k_{sm1} = k_{sm2} 1$ ,  $k_{dm1} = k_{dm2} = 0.1$ ,  $k_{sp1} = k_{sp2} 0.25$ ,  $k_{dp1} = k_{dp2} = 0.5$ , and as initial conditions all variables set to 1. Describe the type of behaviour you see (Remark: We will discuss this behaviour more in-depth in the coming weeks.)

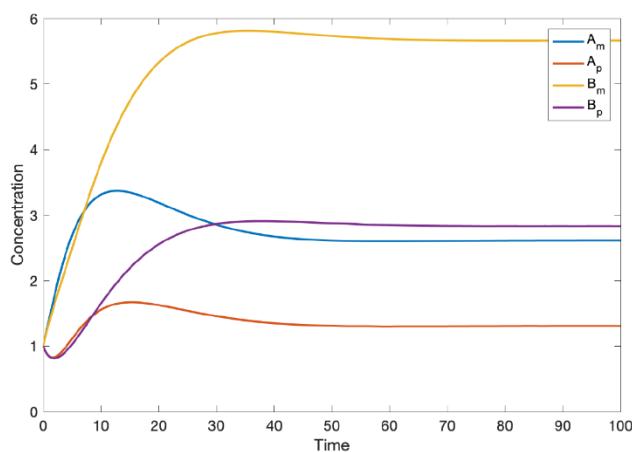
First equation:  $\frac{dA_m}{dt} = k_{sm1} \frac{1}{1+B_p} - k_{dm1} A_m - k_{sp1} A_m + k_{sp1} A_m = k_{sm1} \frac{1}{1+B_p} - k_{dm1} A_m$

Second equation:  $\frac{dA_p}{dt} = k_{sp1} A_m - k_{dp1} A_p$

Third equation:  $\frac{dB_m}{dt} = k_{sm2} \frac{A_p}{1+A_p} - k_{dm2} B_m - k_{sp2} B_m + k_{sp2} B_m = k_{sm2} \frac{A_p}{1+A_p} - k_{dm2} B_m$

Fourth equation:  $\frac{dB_p}{dt} = k_{sp2} B_m - k_{dp2} B_p$

The students should see damped oscillations:



This will come back in the next weeks.

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