

CAPABILITY ENHANCEMENT OF ELECTRONICS AND SOFTWARE

Module 2 - Exhaust After Treatment system Basic architecture and functions (Part 1)

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- III. Basic architecture
 - a) Compact S Type
 - b) Compact liner type

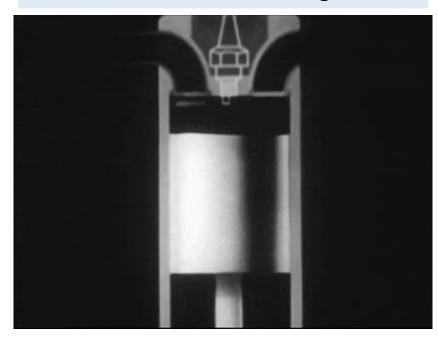
IV. Functions of Exhaust after treatment system

- a) Diesel Oxidation Catalyst- Functions and Major diagnostics
- b) Diesel Particulate Filter- Function and Types of Regeneration
- c) Selective catalyst Reduction- Function and Major diagnostics

Why exhaust after treatment system?



4 Stroke combustion engine





Air pollution









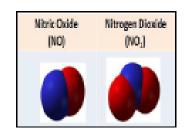




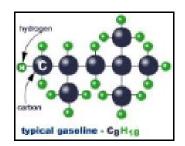
Emission elements and Norms



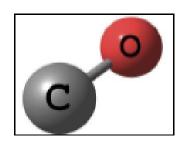
 No_x – Oxides of Nitrogen



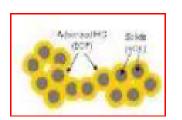
HC – Unburnt Hydro carbon

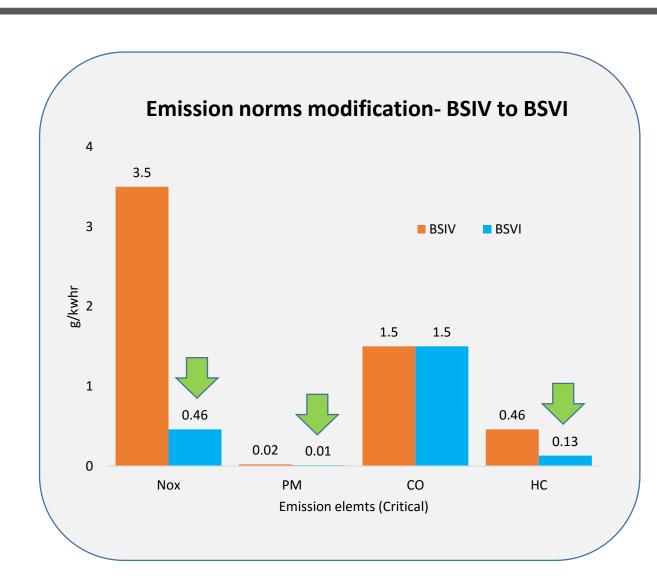


CO - Carbon Monoxide



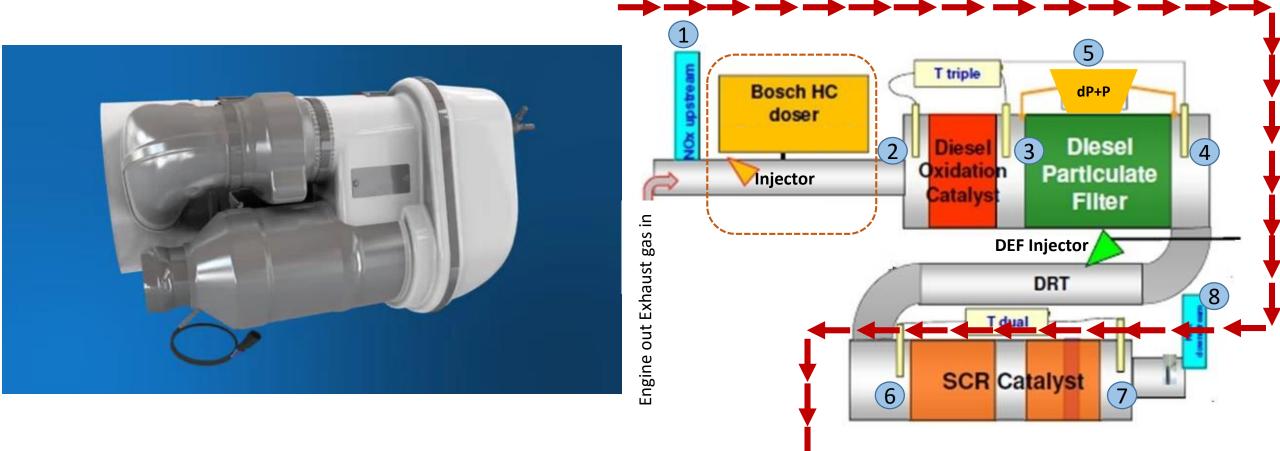
PM - Particulate Matter





Compact S type architecture- H6, A4 and A6





DEF : Diesel Exhaust fluid (Urea/AdBlue)

DRT : Decomposition Reactant Tube

dP : Delta pressure

SCR : Selective Catalyst Reduction

HC: HydroCarbon

1 Nox Sensor- Engine out

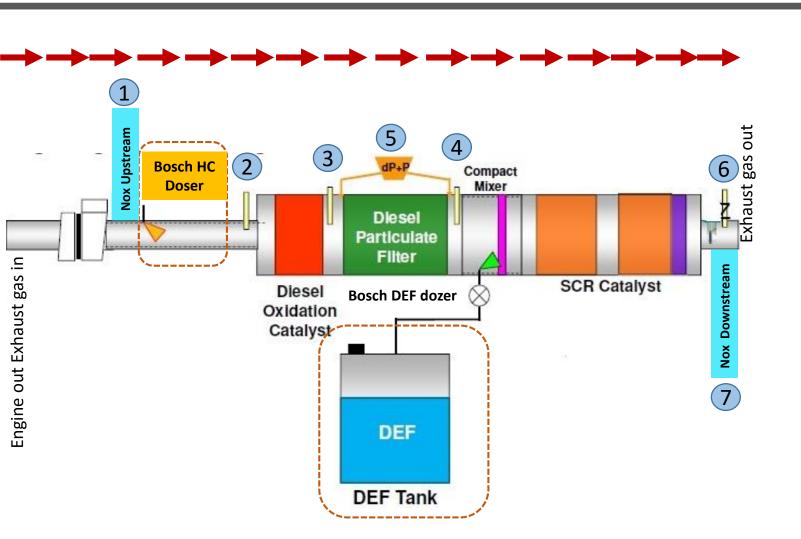
8 Nox Sensor- Tailpipe out

2 3 4 6 7 Exhaust Gas temperature sensors

5 Delta pressure sensor

Compact Linear type architecture- H4





- 1 NOx (Nitrogen Oxide) sensor
 To measure the NOx coming out Engine(ppm)
- 2 Temperature sensor- T1

 To measure DOC inlet temperature or Engine outlet temperature
- 3 Temperature sensor-T2
 To measure DOC outlet temperature or DPF inlet temperature
- 4 Temperature sensor- T3

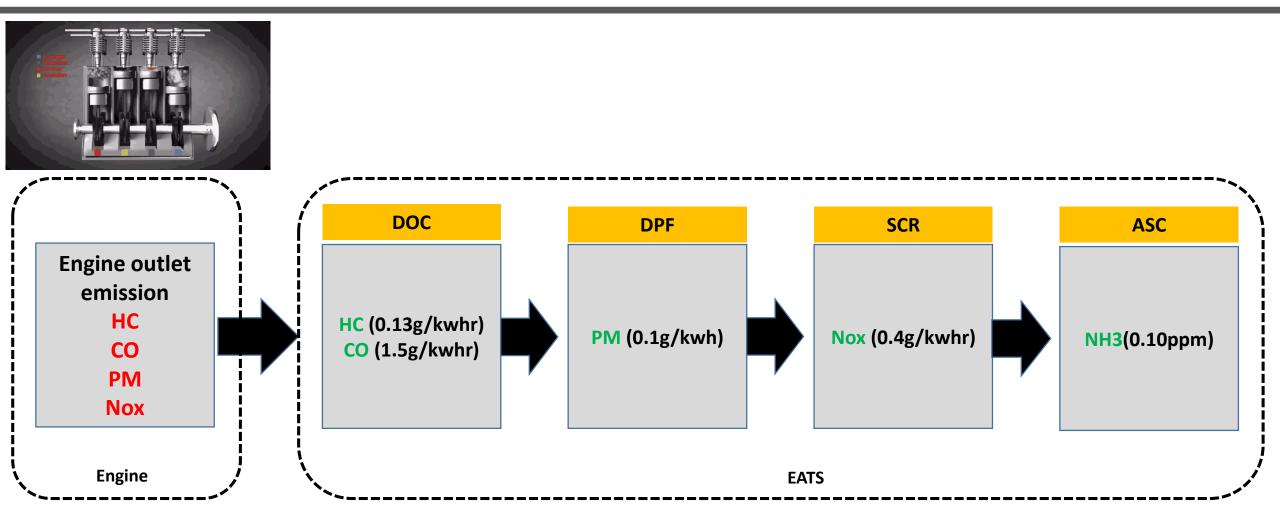
 To measure DPF outlet temperature or SCR inlet temperature
- 5 Delta pressure sensor

 Measure Delta pressure across the DPF to estimate Soot load collected in DPF
- 6 Temperature sensor-T4
 To estimate SCR bed temperature
- 7 NOx sensor- Tail Out

 To measure the tail out NOx after reduction from SCR

Basic function of Exhaust After Treatment system





HC : Hydrocarbons DOC : Diesel Oxidation Catalyst CO : Carbon monoxide DPF : Diesel Particular Filter PM : Particular Matter SCR : Selective Catalyst Reduction NoX : Nitric oxides **ASC** : Ammonia Slip Catalyst

NH3 : Ammonia

Diesel Oxidation Catalyst- DOC



The diesel oxidation catalyst (DOC) promote oxidation of exhaust gas components by oxygen.



Functions of Diesel Oxidation Catalyst

Emission control of Hydrocarbons (HC) and Carbon monoxide(CO)

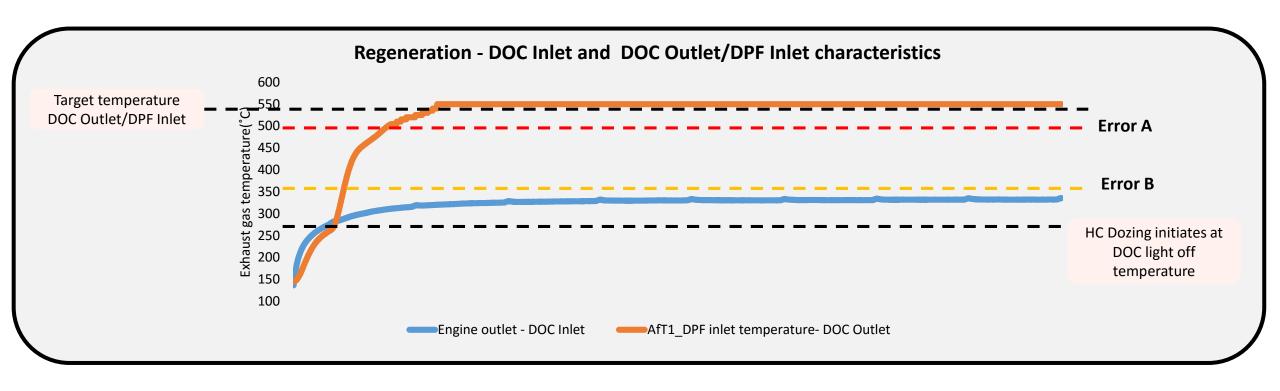
[Hydrocarbons] +
$$O_2$$
 = CO_2 + H_2O
2 CO + O_2 = $2CO_2$

- Active regeneration: Oxidization of hydrocarbons—introduced via diesel fuel injection into the exhaust system or post-injection in the engine—to increase the exhaust gas temperature and facilitate active DPF regeneration
- Passive regeneration: An increased NO₂/NOx ratio promotes passive DPF regeneration.
- Noxidation: an important function of the DOC is to oxidize nitric oxide (NO) to nitrogen dioxide (NO₂)—a gas needed to support the performance of diesel particulate filters and SCR catalysts used for NOx reduction

$$2NO + O_2 = 2NO_2$$

DOC – Exothermic process for regen and Performance related diagnostics





DOC Performance related diagnostics

3 Byte Pcode	Fault Code Description	Thresholds for triggering error
P0421-00- Error A	DOC Conversion efficiency	10% drop in efficiency
P0420-00- Error B	DOC Missing	90% drop in efficiency

Diesel Particulate Filter- DPF



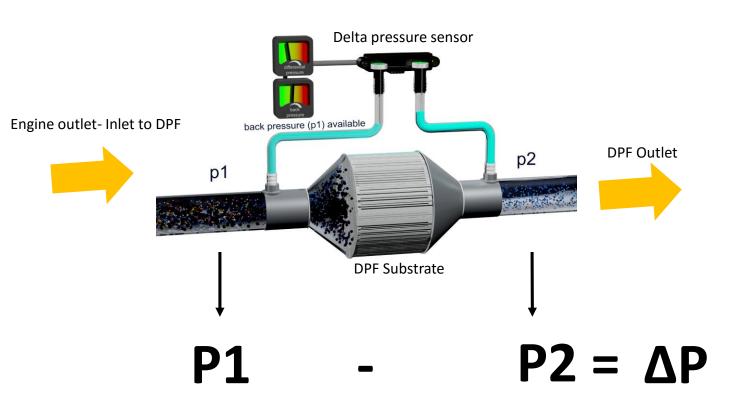
What is DPF?

- DPF is an exhaust after treatment component that traps particulate matter such as soot and ash.
- A DPF typically uses a substrate made of a ceramic material that is formed into a honeycomb structure.

DPF Substrate



DPF Soot load estimation- Delta pressure sensor based

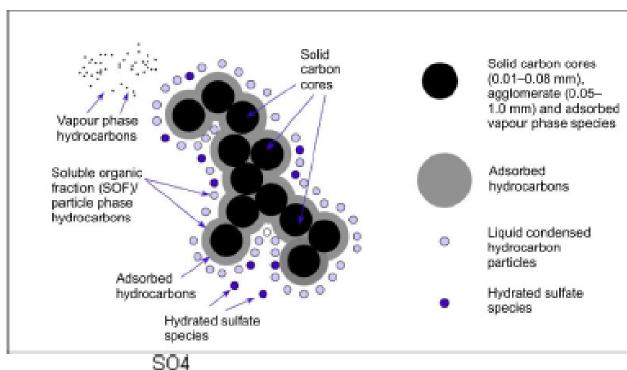


ΔP Increases with increase in Soot load and vice

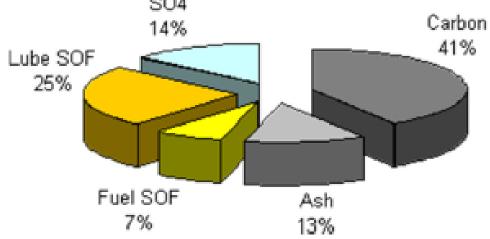
The ΔP signal is then used by the ECU to estimate soot mass inside the filter using calibration maps.

Soot - Composition





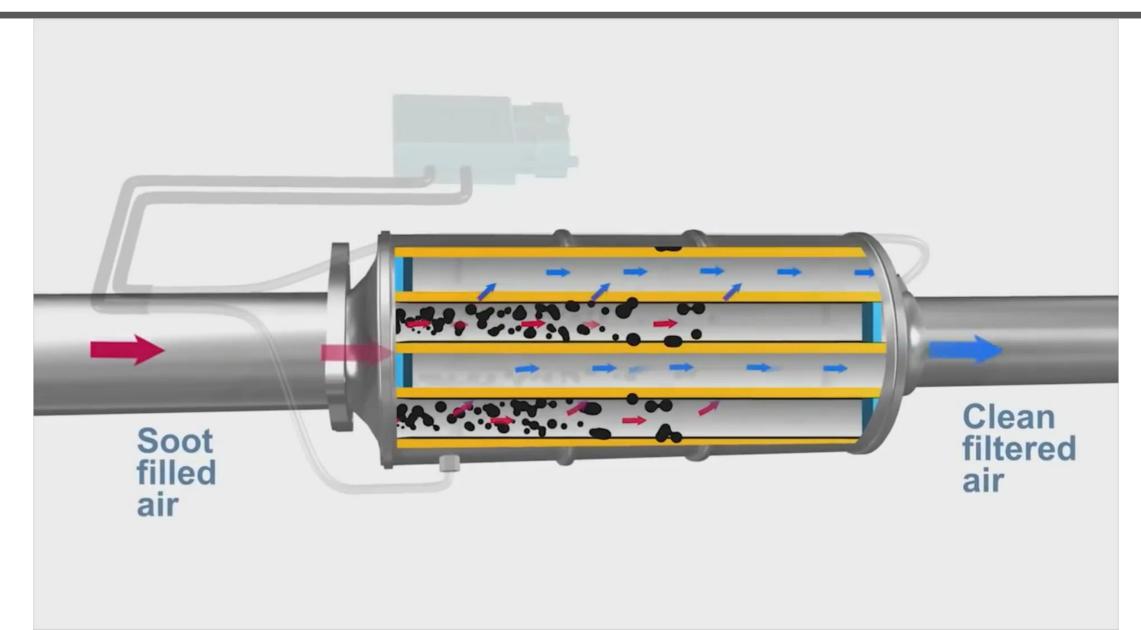
- Adsorbed HCs (soluble organic fractions, or SOF)
- "Dry carbon"
 - Primary particles: ~20-40 nm
 - Agglomerated to ~0.1 1 μm
 - From fuel or lube oil
- Inorganic materials
 - Ash/precursors, H₂SO₄, HNO₃, H₂O



Solids (carbon +ash) - 54% SOF (fuel + lube) - 32% SO4 (sulfate + water) - 14% Note: Particulate composition depends on engine condition, operating point, dilution conditions and fuel type

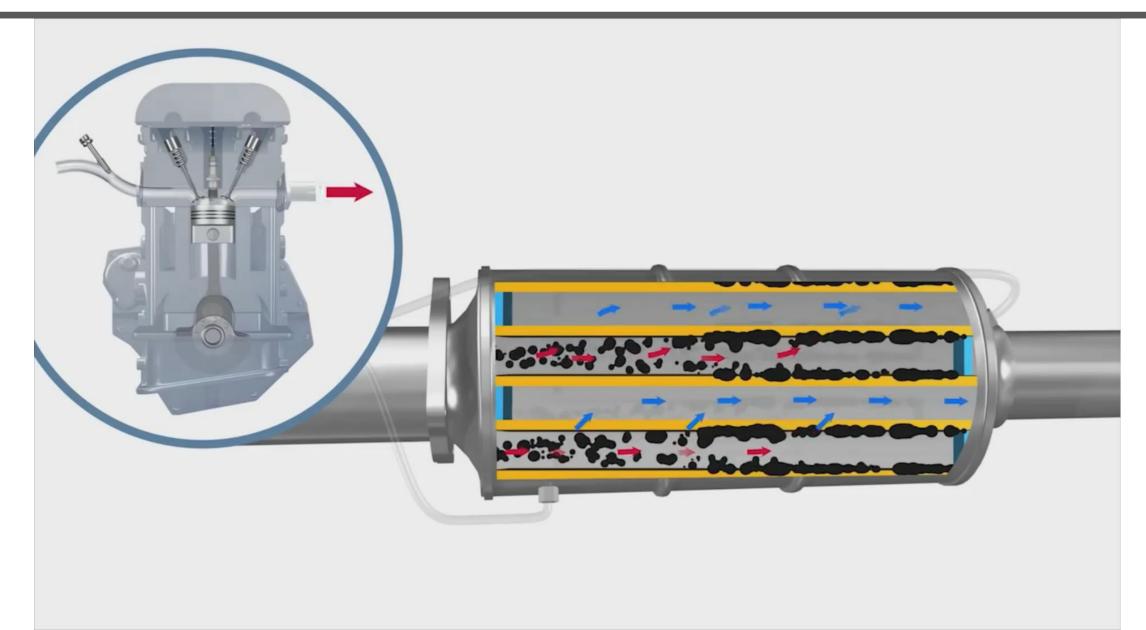
DPF Soot accumulation





DPF Regeneration

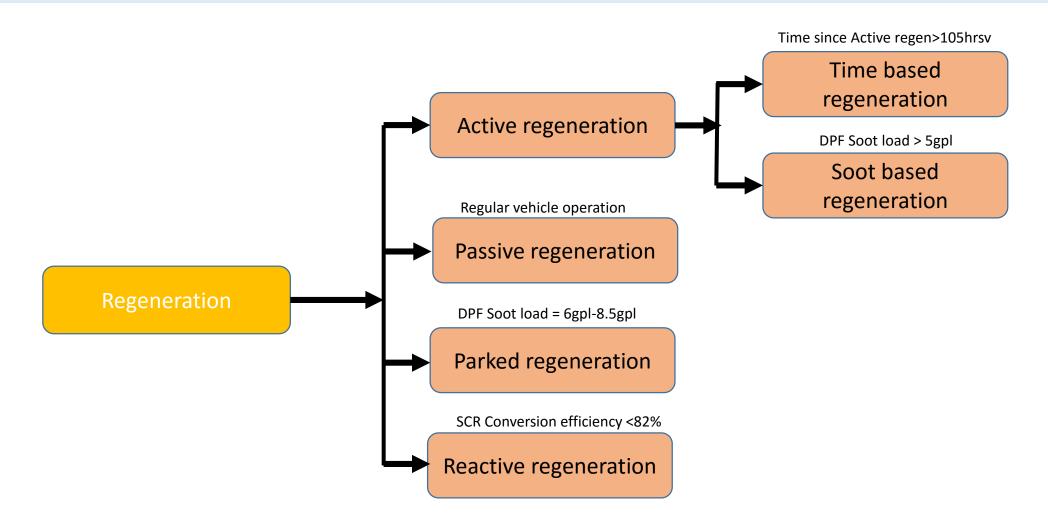




Regeneration-Types



Regeneration is the process of restoring the performance of DPF and SCR Catalysts. Various types of Regenerations features are available in the system which triggers based on the Time, Soot and SCR Conversion efficiency conditions



1. Active regeneration



Active Regeneration

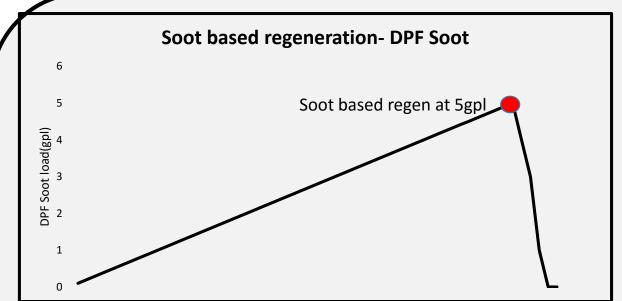
Hydro Carbon injects in the Exhaust gas channels and subsequent oxidation of these hydrocarbons takes place at DOC. High temperature from the DOC outlet is using for Soot burning.

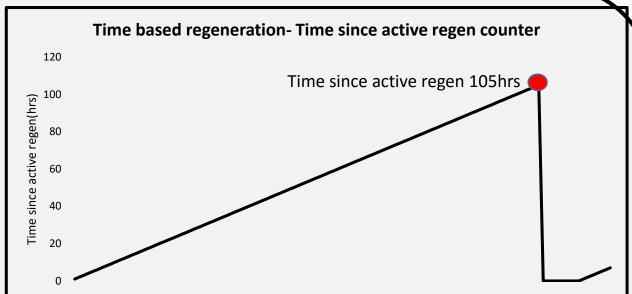


Soot can be burnt in presence of O₂ at very high temperature. The required temperatures are achieved through HC dosing

Active regeneration- Type and Logic







Soot based regeneration: An Active regeneration triggers at 5gpl of soot load to burn the accumulated soot.

Necessity: To remove excessive soot from the DPF before filtration ability is hampered and risk of uncontrolled exothermal increases

Time based regeneration: A time based regeneration triggers every 105hours since last active regeneration.

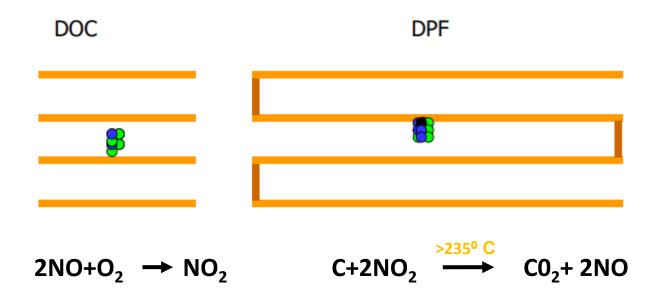
Necessity: As time without regen increases, soot tends to get caked on the periphery of the DPF. This is hard to remove later since temperature at the periphery may not be high enough to burn all the soot collected.

2. Passive regeneration



Passive Regeneration

Takes place in presence of High exhaust gas temperature and Nox from the engine outlet. An external hydrocarbon dosing is not required for the phenomenon to occur. Passive regeneration happens at the favorable conditions.

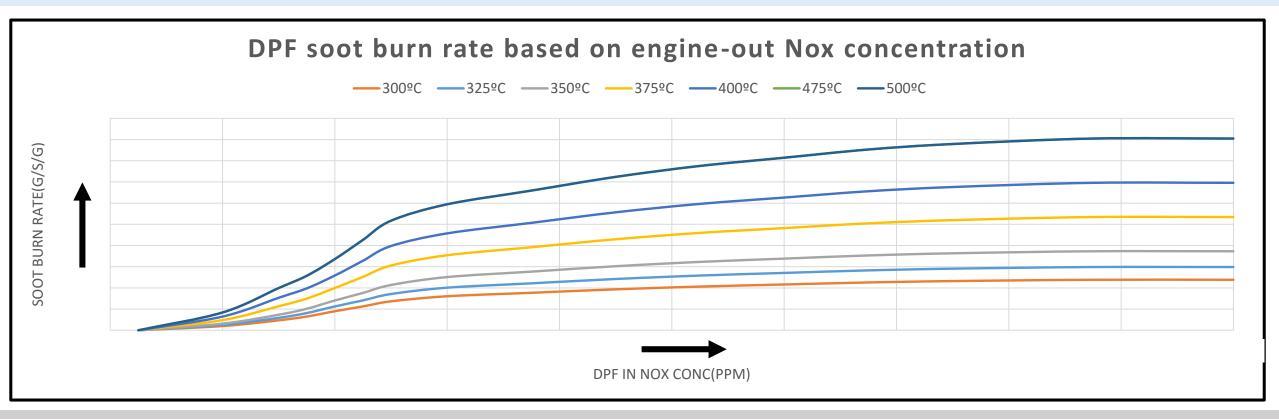


NO₂ being a better oxidizing agent assists soot burning and can burn soot at relatively lower temperatures.

2. Passive regen - Characteristics



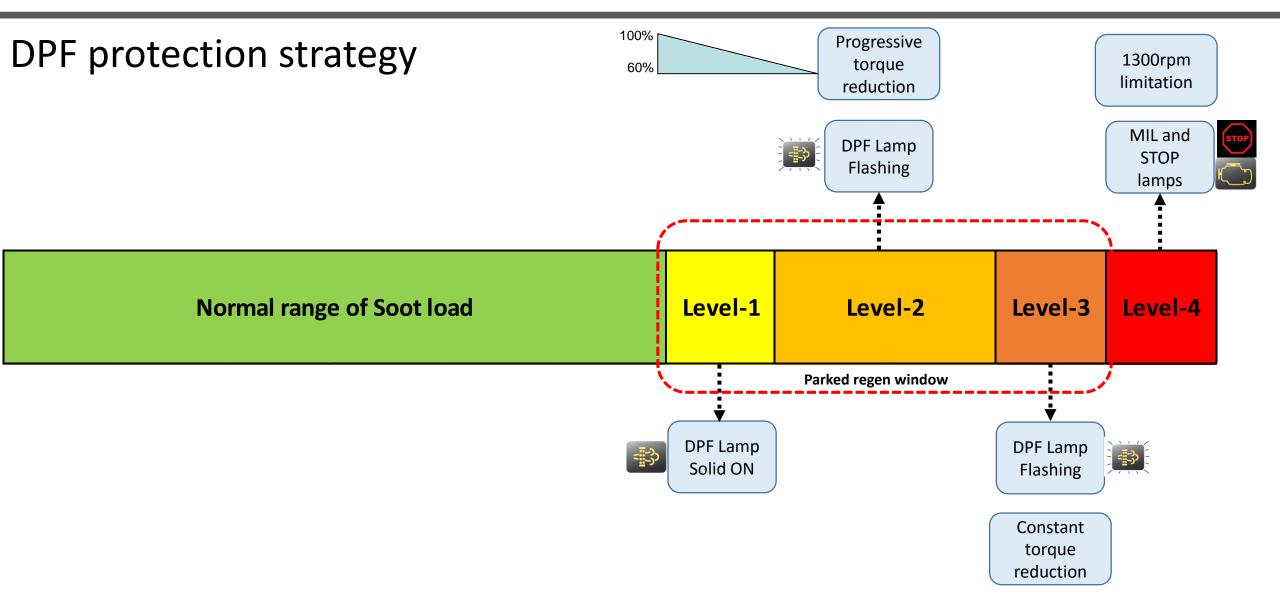
Passive regeneration- Correlation with Engine outlet Nox concentration is explained in the below graph



- ☐ Soot burn rate increases with increase in IN Nox Concentration
- ☐ Soot burn rate increases with increase in temperature
- □ Soot burn rate is higher at higher soot load as soot oxidation process is an exothermic process releasing heat which in turn provides the temperature to burn more soot. Hence soot burn rate is indicated in **Grams per second/ Gram of soot present in DPF**

3. Parked regeneration





3. Parking regeneration

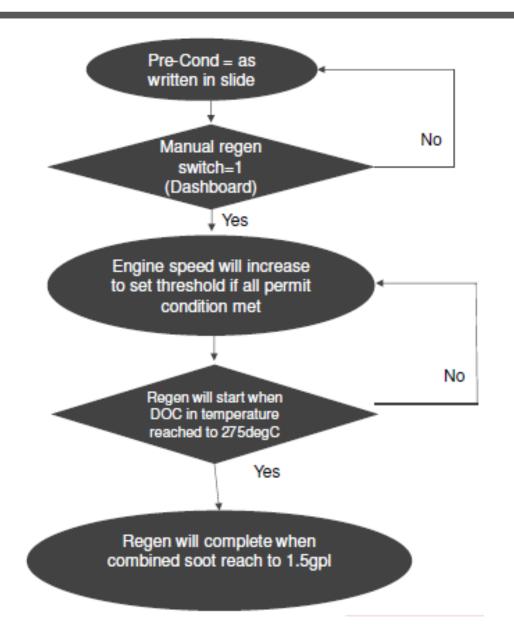


Necessity:

If Active and Passive regenerations are not effective then Parked regen will trigger to clean aftertreatment system at controlled engine Idle condition.

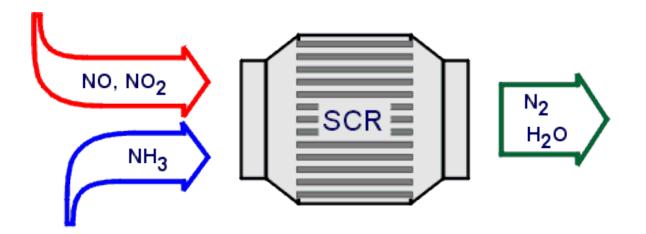
Enabled condition:

- DPF Regen lamp=ON,
- Park the vehicle at safe place
- Inhibit Switch=OFF
- Parking brake=ON
- Acceleration pedal position=0
- Clutch and Brake Switch= Off
- Vehicle Speed=0
- PTO= Off



Selective catalytic Reduction(SCR)





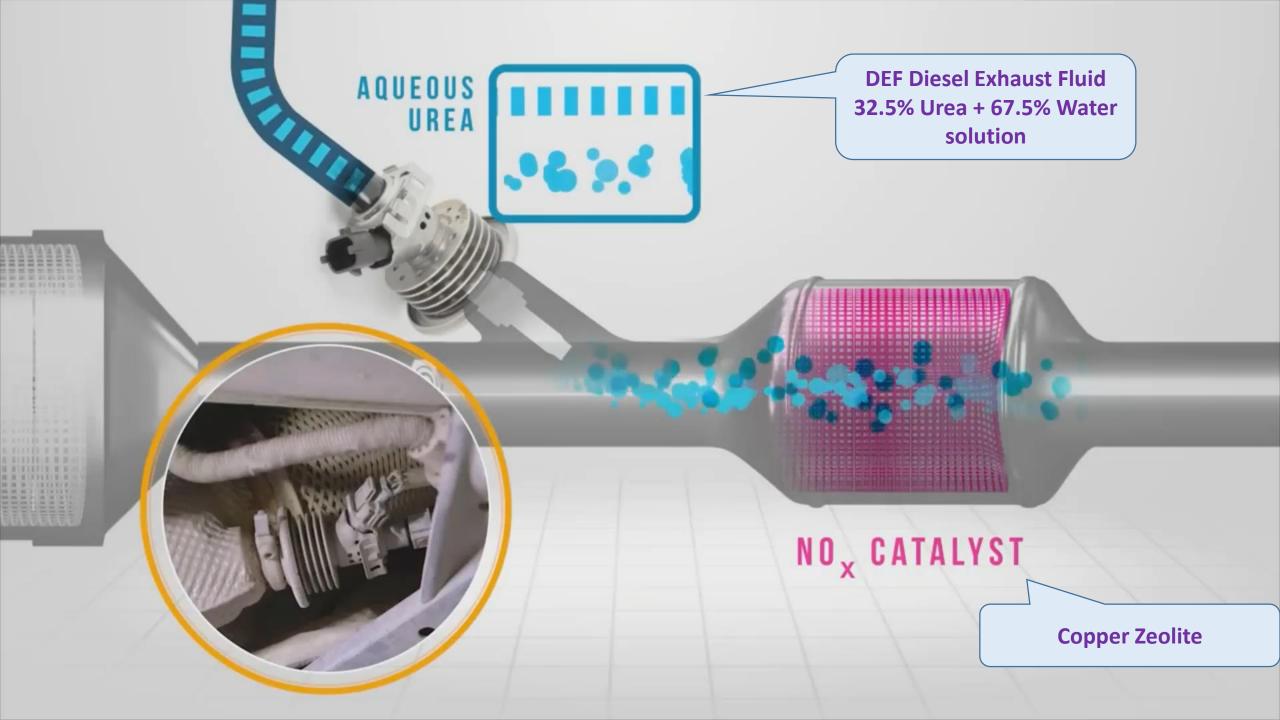
The name Selective Catalytic Reduction (SCR) catalyst is derived from the ability of the catalyst to selectively reduce NOx to N_2 and H_2O in the presence of Reducing agent NH_3 (Product of hydrolysis of Ad-blue in DRT)

- Reduction: The function of the SCR is to reduce Nitrogen oxides (NOx) to Nitrogen (N₂) and Water(H₂O)— Ammonia (NH₃) is used as a reducing agent
- Ammonia (NH₃) is obtained from the hydrolysis of Ad-blue in presence of Carbon-di-oxide (CO2)

$$4 \text{ NO} + 4 \text{ NH}_3 + \text{O}_2 ----> 4\text{N}_2 + 6\text{H}_2\text{O}$$

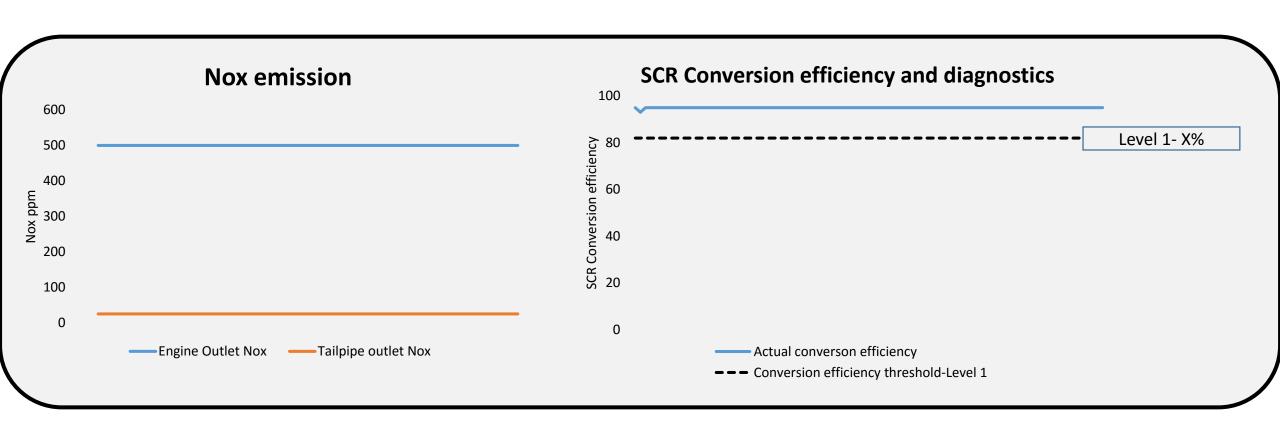
 $2\text{NO}_2 + 4 \text{ NH}_3 + \text{O}_2 ----> 3\text{N}_2 + 6\text{H}_2\text{O}$

• Catalyst is a combination copper and zeolite oxides



SCR Conversion efficiency diagnostics





4. Reactive regeneration



What is Reactive regeneration?

Reactive regeneration is the process of cleaning SCR catalyst through high temperature using HC dosing.

When?

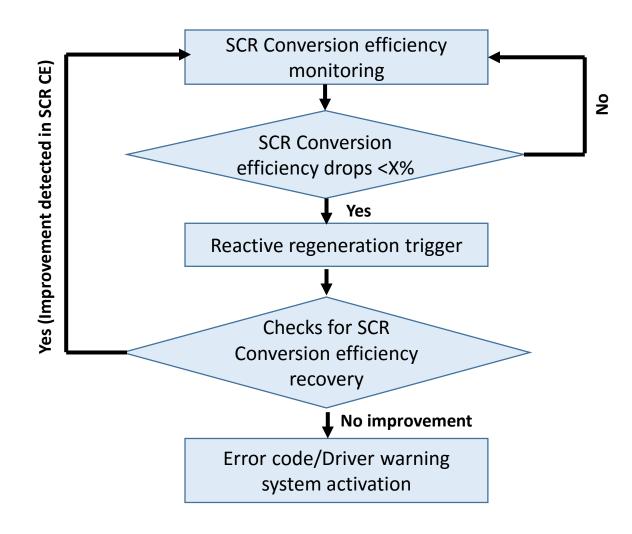
Sulphur masking due to high Sulphur diesel

Excessive urea deposits over SCR catalyst

Any other chemical masking over SCR catalyst

Deteriorated SCR catalyst

Logic





Thank you