

Unit 1: Electromagnetic Radiation

Unit 2: Atomic Structure

Unit 3: The Periodic Table

Unit 4: Chemical Reactions

Unit 5: Acids and Bases

Unit 6: The Earth and the Atmosphere

Unit 7: The Solar System

Unit 8: The Universe

Unit 9: The Human Body

Unit 10: The Environment

Unit 11: The History of Science

Unit 12: The Future of Science

Unit 13: The Science of the Future

Unit 14: The Science of the Past

Unit 15: The Science of the Present

Unit 16: The Science of the Future

Unit 17: The Science of the Past

Unit 18: The Science of the Present

Unit 19: The Science of the Future

Unit 20: The Science of the Past

Unit 21: The Science of the Present

Unit 22: The Science of the Future

Unit 23: The Science of the Past

Unit 24: The Science of the Present

Unit 25: The Science of the Future

Unit 26: The Science of the Past

Unit 27: The Science of the Present

Unit 28: The Science of the Future

Unit 29: The Science of the Past

Unit 30: The Science of the Present

Unit 31: The Science of the Future

Unit 32: The Science of the Past

Unit 33: The Science of the Present

Unit 34: The Science of the Future

Unit 35: The Science of the Past

Unit 36: The Science of the Present

Unit 37: The Science of the Future

Unit 38: The Science of the Past

Unit 39: The Science of the Present

Unit 40: The Science of the Future

Unit 41: The Science of the Past

Unit 42: The Science of the Present

Unit 43: The Science of the Future

Unit 44: The Science of the Past

Unit 45: The Science of the Present

Unit 46: The Science of the Future

Unit 47: The Science of the Past

Unit 48: The Science of the Present

Unit 49: The Science of the Future

Unit 50: The Science of the Past

Unit 51: The Science of the Present

Unit 52: The Science of the Future

Unit 53: The Science of the Past

Unit 54: The Science of the Present

Unit 55: The Science of the Future

Unit 56: The Science of the Past

Unit 57: The Science of the Present

Unit 58: The Science of the Future

Unit 59: The Science of the Past

Unit 60: The Science of the Present

Unit 61: The Science of the Future

Unit 62: The Science of the Past

Unit 63: The Science of the Present

Unit 64: The Science of the Future

Unit 65: The Science of the Past

Unit 66: The Science of the Present

Unit 67: The Science of the Future

Unit 68: The Science of the Past

Unit 69: The Science of the Present

Unit 70: The Science of the Future

Unit 71: The Science of the Past

Unit 72: The Science of the Present

Unit 73: The Science of the Future

Unit 74: The Science of the Past

Unit 75: The Science of the Present

Unit 76: The Science of the Future

Unit 77: The Science of the Past

Unit 78: The Science of the Present

Unit 79: The Science of the Future

Unit 80: The Science of the Past

Unit 81: The Science of the Present

Unit 82: The Science of the Future

Unit 83: The Science of the Past

Unit 84: The Science of the Present

Unit 85: The Science of the Future

Unit 86: The Science of the Past

Unit 87: The Science of the Present

Unit 88: The Science of the Future

Unit 89: The Science of the Past

Unit 90: The Science of the Present

Unit 91: The Science of the Future

Unit 92: The Science of the Past

Unit 93: The Science of the Present

Unit 94: The Science of the Future

Unit 95: The Science of the Past

Unit 96: The Science of the Present

Unit 97: The Science of the Future

Unit 98: The Science of the Past

Unit 99: The Science of the Present

Unit 100: The Science of the Future

Sep 29-15:54

Modelling 2017

21 (a) Fig. 21.1 shows some of the energy levels of electrons in hydrogen gas atoms. The energy levels are labelled A, B, C and D.

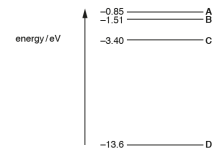


Fig. 21.1 (not to scale)

(i) Explain why the energy levels are negative.

[1]

(ii) An electron makes a transition (jump) from level C to level A.

1 Calculate the energy gained by this electron.

energy = 2.55 eV [1]

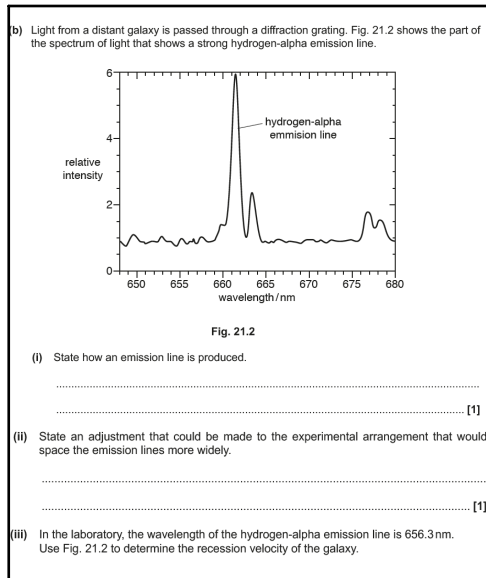
2 Calculate the wavelength in nm of the photon absorbed by this electron.

$$hc/\lambda = E$$

$$\lambda = 488 \text{ nm}$$

wavelength = 488 nm [3]

Feb 5-14:09



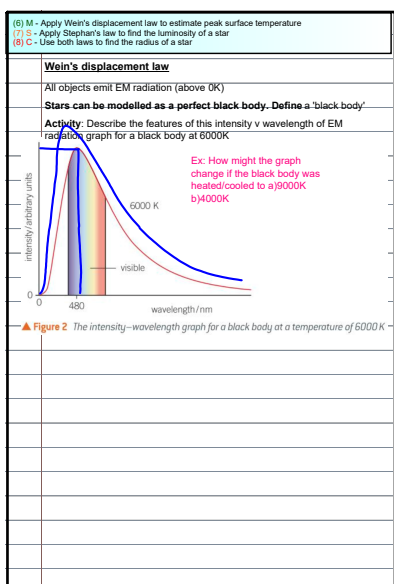
Feb 5-14:09

(6) M - Apply Wein's displacement law to estimate peak surface temperature
(7) S - Apply Stefan's law to find the luminosity of a star
(8) C - Use both laws to find the radius of a star

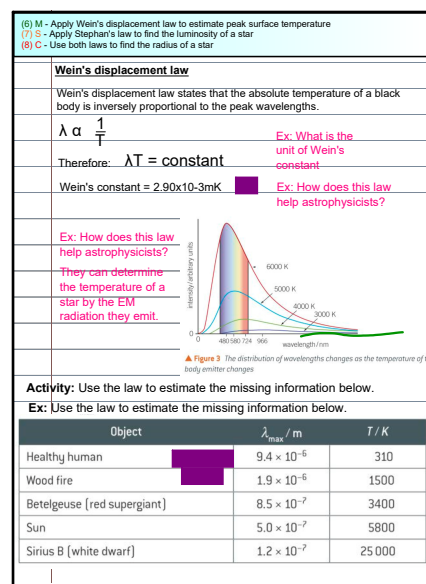
Stellar luminosity

Discuss, what happens when you heat a material - though a range of temperatures?

Sep 29-15:54



Sep 29-15:54



Sep 29-15:54

(6) M - Apply Wein's displacement law to estimate peak surface temperature
(7) S - Apply Stefan's law to find the luminosity of a star
(8) C - Use both laws to find the radius of a star

Stephan's law 1m 12 10W

Total power radiated per unit surface area of black body is directly proportional to the 4th power of the absolute temperature.

Ex: Derive the formula. (lower case sigma is 'stephan's constant')

Quantity /unit

$L = 4\pi r^2 \sigma T^4$

- L- luminosity / W
- r- radius / m
- σ - Stephan constant /
- T- Absolute temp / K. works

Activity: Try the summary questions 4 and 5

4 Using Stefan's law, compare the luminosity of a star with another that has:
a double the surface temperature and the same radius;
b double the radius and half the surface temperature;
c half the mass, the same density, and three times the surface temperature. (6 marks)

5 The Sun has a radius of approximately 700 000 km and a surface temperature of 5800 K. Calculate the energy radiated by the Sun during one year. (3 marks)

Sep 29-15:54

(6) M - Apply Wein's displacement law to estimate peak surface temperature
(7) S - Apply Stefan's law to find the luminosity of a star
(8) C - Use both laws to find the radius of a star

Finding the radius of a star (could LO)

Worked example: Radius of a star

The peak wavelength of radiation emitted by our Sun is about 500 nm, its surface temperature is 5800 K, and its luminosity is 3.85×10^{26} W. The peak wavelength emitted by a nearby star with a luminosity 10 times that of our Sun is 310 nm. Show that the radius of this star is approximately 840 000 km.

Step 1: To determine the surface temperature of the star use Wien's displacement law.
 $\lambda_{\text{max}} T = \text{constant}$
 $500 \times 10^{-9} \times 5800 = 310 \times 10^{-9} \times T_{\text{star}}$
Sun Star
Therefore
 $T_{\text{star}} = \frac{500 \times 10^{-9} \times 5800}{310 \times 10^{-9}} = 9355 \text{ K}$

Step 2: Use Stefan's law to determine the radius of the star.
 $L = 4\pi r^2 \sigma T^4$, therefore $r = \sqrt{\frac{L}{4\pi \sigma T^4}}$
The luminosity L of the star is $10 \times 3.85 \times 10^{26} \text{ W}$
 $r = \sqrt{\frac{10 \times 3.85 \times 10^{26}}{4\pi \times 5.67 \times 10^{-8} \times 9355^4}}$
 $r = 8.399 \dots \times 10^8 \text{ m} = 840\,000 \text{ km}$ (2 s.f.)

Sep 29-15:54

(6) M - Apply Wein's displacement law to estimate peak surface temperature
(7) S - Apply Stefan's law to find the luminosity of a star
(8) C - Use both laws to find the radius of a star

Finding the radius of a star (could LO)

8 The peak wavelength emitted by a star is 500 nm and its luminosity is $3.85 \times 10^{26} \text{ W}$.
a Use Stefan's law to find the luminosity of a star.
b Use Stefan's law to find the radius of a star. (3 marks)

Ex:

a A star emits light with a peak wavelength of 500 nm and has a luminosity of $3.85 \times 10^{26} \text{ W}$. What is the star's radius? (4 marks)

b A scientific model suggests a star with a radius of $8.39 \times 10^8 \text{ m}$ that has a luminosity of $3.85 \times 10^{26} \text{ W}$. If these values are correct, what will the peak wavelength be? (4 marks)

The first of these calculations follows the example closely. The second needs a slightly different approach.

a

$L = 4\pi r^2 \sigma T^4$
 $r = \sqrt{\frac{L}{4\pi \sigma T^4}}$
 $r = \sqrt{\frac{3.85 \times 10^{26}}{4\pi \times 5.67 \times 10^{-8} \times 5800^4}}$
 $r = 6.96 \times 10^8 \text{ m}$
 $r = 696\,000 \text{ km}$ (3 marks)

b

$L = 4\pi r^2 \sigma T^4$
 $T = \sqrt[4]{\frac{L}{4\pi r^2 \sigma}}$
 $T = \sqrt[4]{\frac{3.85 \times 10^{26}}{4\pi \times (8.39 \times 10^8)^2 \times 5.67 \times 10^{-8}}}$
 $T = 5800 \text{ K}$ (3 marks)

Sep 29-15:54

(6) M - Describe energy levels in isolated atoms
(7) S - Explain emission spectra in terms of transition of electrons between energy levels
(8) C - Explain how spectral lines can be used to identify elements within a star

Plenary circle

Simple key terms Challenging key terms

Stars

Use 1-3 key terms to make a sentence that shows knowledge and understanding towards the HLOS

Sep 29-15:54

Feb 5-15:31