

Module 7 and 8

Step 1:

$$4\pi R^2 D \left[ \frac{\partial c}{\partial r} \right] = v(t)$$

$r=R$

$$v(t) = c(R, t)$$

$$R=1, D=1$$

Step 2:

Fick's Law

$$J = -D \frac{\partial c}{\partial x}$$

$$\frac{\partial c}{\partial x} = D \frac{\partial^2 c}{\partial x^2}$$

Fick's law for ideal

spherical medium

$$\frac{\partial c}{\partial r} = D \frac{1}{r^2} \frac{\partial}{\partial r} \left[ r^2 \frac{\partial c}{\partial r} \right]$$

$$\frac{\partial \sigma}{\partial t} + r_2 \frac{\partial \sigma}{\partial r} + \frac{\partial^2 \sigma}{\partial r^2}$$

$$= C(S, r)$$

$$+ \frac{\partial^2 \sigma}{\partial r^2} C(S, r)$$

Boundary Conditions:

$$\left. \frac{\partial \sigma}{\partial r} \right|_{r=R} = 0$$

$$\sigma(S, r=R)$$

$$\Rightarrow U(t) = \frac{4\pi R^2 D}{\epsilon_r} \left[ \frac{\partial C}{\partial r} \right]$$

Or

$$r=R$$

$$\frac{\partial C}{\partial r}(t, r=0) = G$$

$$\frac{\partial C}{\partial r}(t, r=R) = U(t)$$

$$= \frac{4\pi R^2 D}{\epsilon_r} \left[ \frac{\partial C}{\partial r} \right]_{r=R}$$

### Step 3: Padé Approximation

Joel C. Forman et al 2011 J. Electrochem. Soc. 158 A93-  
 Reduction of an Electrochemistry-Based Li-Ion Battery  
 Model via Quasi-  
 Linearization and Padé Approximation

$$\text{Here, } \frac{\partial C(t, R)}{\partial r} = -mU(t)$$

$$\frac{C(S_{1,n})}{C(S_{0,n})} = \frac{(2VFD_{n-1})^{1/2}}{1 + D_{n-1} + (2VFD_{n-1})^{1/2}}$$

$$1 + \frac{R}{\omega} s + \dots + \left( \frac{R}{\omega} s - 1 \right) n$$

∴ Second Order

$$\frac{f(s)}{U(s)} = \frac{a_0 + a_1 s}{s(1 + b_2 s)}$$

$$(1 - \frac{3D_m}{sP} - \frac{2mP}{sT}) S$$

$$\times \frac{\omega^2 D}{\omega^2 D + \omega^2}$$

$$U(s) = \frac{1}{4\pi c^2 D} ((s, r)) S$$

$$\frac{f(s)}{U(s)} = \frac{3D_m}{sP} - \frac{2mP}{sT} S$$

$$\times \frac{\omega^2 D}{\omega^2 D + \omega^2}$$

$$\tau = -1 / \sqrt{4\pi c^2 D}$$

$$\begin{aligned}
 & \text{Discretization, Finite Differences} \\
 & \frac{\partial^2 F}{\partial t^2} = D \left[ \frac{\partial^2 F}{\partial r^2} + r^2 \frac{\partial^2 F}{\partial r^2} \right] \\
 & = \frac{\partial^2 F}{\partial t^2} \left[ \frac{\partial^2 F}{\partial r^2} \right] + r^2 \frac{\partial^2 F}{\partial r^2} \left[ \frac{\partial^2 F}{\partial r^2} \right]
 \end{aligned}$$

$$= \frac{1}{r^2} \left[ \frac{\partial^2}{\partial r^2} + \frac{1}{r} \frac{\partial}{\partial r} + \frac{1}{r^2} \frac{\partial^2}{\partial \theta^2} \right] C$$

$$\begin{aligned} & r = 0: \quad C_{1,2,3,4} = 0 \\ & r = R: \quad C_{1,2,3,4} = 2R/y \\ & r = 2R/y: \quad C_{1,2,3,4} = 0 \end{aligned}$$

$$C_{1,2,3,4}(t) = C(r_{1,2,3,4}, t)$$

$$\frac{\partial^2}{\partial r^2} C_{1,2,3,4} = C_{1,2,3,4}$$

$$\frac{\partial^2}{\partial r^2} C_{1,2,3,4} = 0$$

$$\begin{aligned} & \frac{\partial^2}{\partial r^2} C_{1,2,3,4} = 0 \\ & \quad \Rightarrow C_{1,2,3,4} = 0 \end{aligned}$$

$$C_2 = C_0 + C_P(\omega)$$

$$C_0 @ \pi = -$$

$$C_P(\omega) = D \left[ \frac{2}{R/3} \frac{C_0}{2R/3} \right]$$

$$+ \left[ \frac{(2-2C_1+C_0)}{(R/3)^2} \right] \rightarrow C_0 = C_1$$

$$= D \left[ \frac{2}{R/3} \frac{C_2-C_1}{R/3} \right]$$

$$+ \left[ \frac{(2-2C_1+C_0)}{R^2/9} \right]$$

$$= D \left[ 27((C_2-C_1)) \right]$$

$$C_1 @ \pi = 2 \\ C_P(\omega) = D \left[ \frac{2}{2R/3} \frac{C_0-C_1}{2R/3} \right]$$

$$+ \left[ \frac{C_2 - 2C_2 + C_1}{R_1 R_2} \right] C_3 = C_2 + \frac{C_1}{R_2}$$

$$= 0 \left[ \frac{2C_2 + 2\frac{C_1}{R_2} - 2C_1}{R_2} \right]$$

$$+ C_2 + \left[ \frac{C_1}{R_2} - 2C_2 + C_1 \right]$$

$$= 0 \left[ -0.5C_2 + 0.5R_2 + 0.5C_1 \right]$$

$$C_1 = \frac{2R_2}{R_1 + R_2} \left[ C_1 - C_2 + UR \right]$$

$$C_1 = \frac{2R_2}{R_1 + R_2} \left[ S - C_1 \right]$$

$$C_2 = \frac{2R_1}{R_1 + R_2} \left[ C_1 - C_2 + UR \right]$$

$$C_2 = C_2 + UR$$

$$G_3 = C_2 + \frac{R}{j\omega}$$

LaPlace

$$= G_2(\omega) = C_2(\omega) + \frac{R}{j\omega} U(\omega)$$

$$+ \frac{R}{j\omega} \left[ \frac{R^2}{2T} s + \frac{1}{j\omega} \right] U(\omega)$$

$$\left[ \frac{R^2}{2T} s + \frac{1}{j\omega} \right] U(\omega) = C_2(\omega)$$

$$\left[ \frac{S^2 R^2}{2T} + \frac{1}{j\omega} \right] U(\omega) = C_2(\omega) + R U(\omega)$$

$$= \left[ \frac{S^2 R^2}{2T} s + \left( \frac{R^2}{2T} s + \frac{1}{j\omega} \right) \right] U(\omega)$$

$$= \frac{2R}{2T} C_1(\omega) + \frac{R}{j\omega} U(\omega)$$

$$+ \left[ \frac{2R^2}{2T} s^2 + \left( \frac{R^2}{2T} s + \frac{1}{j\omega} \right)^* \right]$$

$$= C \begin{bmatrix} 2Hg \\ g \\ 2R^2 \end{bmatrix} = C \begin{pmatrix} g \\ g \\ g \end{pmatrix}$$

$$= C \begin{bmatrix} 2Hg \\ g \\ 2R^2 \end{bmatrix} = C \begin{pmatrix} g \\ g \\ g \end{pmatrix}$$

$$= C \begin{bmatrix} 2Hg \\ g \\ 2R^2 \end{bmatrix} = C \begin{pmatrix} g \\ g \\ g \end{pmatrix}$$

$$= C \begin{pmatrix} g \\ g \\ g \end{pmatrix} + R$$

$$= C \begin{pmatrix} g \\ g \\ g \end{pmatrix} + R \begin{bmatrix} 2Hg \\ g \\ 2R^2 \end{bmatrix}$$

$$= C \begin{pmatrix} g \\ g \\ g \end{pmatrix} + R \begin{bmatrix} 2Hg \\ g \\ 2R^2 \end{bmatrix}$$

$$C \begin{pmatrix} g \\ g \\ g \end{pmatrix} = \begin{bmatrix} g \\ g \\ g \end{pmatrix} + R \begin{bmatrix} 2Hg \\ g \\ 2R^2 \end{bmatrix}$$

$$C \begin{pmatrix} g \\ g \\ g \end{pmatrix} = \begin{bmatrix} g \\ g \\ g \end{pmatrix} + R \begin{bmatrix} 2Hg \\ g \\ 2R^2 \end{bmatrix}$$

$$U = \frac{2R_1}{2R_1 + R_2} \frac{C_2}{C_2 + \frac{2R_2}{2R_1 + R_2}} C$$

Step 2:

$$\frac{a+s}{sC+Cs} = \frac{a+s}{sC(1+C)}$$

$$\frac{a+s}{sC(1+C)} = \frac{A}{s} + \frac{B}{1+C}$$

$$a+s = A(1+C) + Bs$$

$$s=0$$

$$A=a$$

$$a - b = a - b$$

$$B = -C(a-b) = b - Ca$$

$$C(s) = \frac{a}{s} + \frac{b-Ca}{1+C}$$

$\text{G}(s) = \text{"storage"} + \text{"RC"}$

$$\text{V}_1(s) = \frac{1}{C} + \text{V}_2(s)$$

$$\text{V}_1(s) = \frac{Q}{C} + \text{V}_2(s)$$

$$2\text{V}_2 = \frac{(b - ca)}{c} \text{V}_2$$

$$\text{V}_2 = \frac{ca}{b - ca} \text{V}_2$$

$$\text{V}_2 = \frac{(b - ca)}{c} \text{V}_2$$

$$- \quad \downarrow \quad \frac{1}{2}$$

For 1 electrode

$$\text{V}_2 = \frac{ca}{b - ca} \text{V}_2$$

$$C = V_1 + V_2$$

$$C = V_1 + V_2$$

Advanced \*

$$\text{let } \alpha = \frac{b - C}{q}$$

$$\text{let } z_1 = V_1, C$$

$$\text{let } z_2 = \underline{V_2}$$

$$\alpha^2$$

Intercalation Current

Density  $\gamma \rightarrow V(t)$

Surface Concentration  $\rightarrow Y(t)$

Reference Potential  $\rightarrow U(t)$

$$z_1 = \alpha V(t)$$

$$z_2 = U(t) - \frac{1}{C} z_1$$

$$Y = z_1 + \alpha z_2$$

$$U = U_{\text{ref}}(Y) + R^* V(t)$$

$$U = U_{\text{ref}}(z_1 + \alpha z_2) + R^* V(t)$$

For Anode + Cathode

$$z_1 = \alpha V(t)$$

$$z_2 = V - \frac{1}{C} z_2$$

$$C_{S,A} = z_1 + \frac{1}{C} z_2$$

$$z_3 = -Q_C V(t)$$

$$z_4 = -V - \frac{1}{C} z_4$$

$$C_{S,C} = z_3 + \frac{1}{C} z_4$$

However, eliminate

redundancies

$$R_{tot} = R_c + R_a + R_{bol}$$

$$V(t) = E_{elec} - E_{elec}$$

$$\begin{aligned} & \left[ -n_c \cdot z_a - R_{tot} V(t) \right] \} \text{discharge current } V(t) \\ & \left[ +n_c \cdot z_a + R_{tot} V(t) \right] \} \text{charge current } V(t) \end{aligned}$$

$$z_1 = Q_C V(t)$$

$$z_2 = V - \frac{1}{C} z_2$$

$$C_{S,A} = z_1 + \frac{1}{C} z_2$$

$$z_3 = -V - \frac{1}{C} z_3$$

$$C_{SC} = \alpha C_2 z_3 + \\ C_{SC, \text{initial}} - \frac{\alpha}{\alpha} (z_1(t) \\ - z_1(0))$$

Assume, overpotential  
 $\gamma_{C12a}$ , determined by  
 the Butler-Volmer  
 equation is neglected  
 $\therefore SPM$

Intercalation Current  
 Density  $\rightarrow V(t)$

Surface Concentration  $\rightarrow Y(t)$

$$\dot{z}_1 = Q_a V(t)$$

$$\dot{z}_2 = V - V_{Z2}$$

$$\dot{z}_3 = -V - f_a z_3$$

$$V(t) = R_{tot} + U(t)$$

$$+ E_{ac} (x_1 z_3 + C_{SC, \text{initial}})$$

$$+ F_{\text{air}}(x_3 t_3 + \text{constant}) \\ - \frac{\alpha_c}{\alpha_a} (x_1(t) - x_1(0))$$

$$+ F_{\text{air}}(x_a t_2 + x_1)$$