

Final project bootstrapping

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1 Find the function $\langle r^a \rangle (E)$:

Using the recurrence relation

$$\langle r^{a-1} \rangle = \frac{1}{8mE} [(1-a)(a(a-2) - 4l(l+1))\langle r^{a-3} \rangle - 4(2a-1)\langle r^{a-2} \rangle]$$

$$\langle r^{-1} \rangle = -2E$$

$$\langle r^0 \rangle = 1$$

$$\langle r \rangle = -\frac{2El(l+1) + 3}{4E}$$

$$\langle r^2 \rangle = \frac{6El^2 + 6El - 2E + 5}{8E^2}$$

$$\langle r^3 \rangle = -\frac{70El(l+1) + 18E(l(l+1)-2)(2El(l+1)+3) + 14E(4l(l+1)-3) + 105}{192E^3}$$

$$\langle r^4 \rangle = \frac{60E^2l^4 + 120E^2l^3 - 140E^2l^2 - 200E^2l + 48E^2 + 140El^2 + 140El^2 + 140El - 210E + 63}{128E^4}$$

$$\langle r^5 \rangle = \frac{-40E^3l^6 - 120E^3l^5 + 200E^3l^4 + 600E^3l^3 - 160E^3l^2 - 480E^3l - 420E^2l^4 - 840E^2l^3 + 1680E^2l^2 + 2100E^2l - 1176E^2 - 630El^2 - 630El + 1470E - 231}{512E^5}$$

$$\langle r^6 \rangle = \frac{280E^3l^6 + 840E^3l^5 - 2240E^3l^4 - 5880E^3l^3 + 3976E^3l^2 + 7056E^3l - 1440E^3 + 1260E^2l^4 + 2520E^2l^3 - 7560E^2l^2 - 8820E^2l + 8484E^2 + 1386El^2 + 1386El - 4620E + 429}{1024E^6}$$

$$\langle r^7 \rangle = \frac{-560E^4l^8 - 2240E^4l^7 + 7840E^4l^6 + 31360E^4l^5 - 27440E^4l^4 - 109760E^4l^3 + 20160E^4l^2 + 80640E^4l - 10080E^3l^6 - 30240E^3l^5 + 115920E^3l^4 + 282240E^3l^2 - 328944E^3l^2 - 475104E^3l + 219168E^3 - 27720E^2l^4 - 55400E^2l^3 + 231000E^2l^2 + 258720E^2l - 374220E^2 - 24024El^2 - 24024El + 108108E - 6435}{16384E^7}$$

$$\langle r^8 \rangle = \frac{5040E^4l^8 + 20160E^4l^7 - 97440E^4l^6 - 362880E^4l^5 + 526512E^4l^4 + 1681344E^4l^3 - 848832E^4l^2 - 1753344E^4l + 322560E^4}{32768E^8}$$

$$+ \frac{36960E^3l^6 + 110880E^3l^5 - 572880E^3l^4 - 1330560E^3l^3 + 2310000E^3l^2 + 2993760E^3l - 2314400E^3 + 72072E^2l^4 + 144144E^2l^3}{32768E^8}$$

$$- \frac{792792E^2l^2 - 864864E^2l + 1741740E^2 + 51480El^2 + 51480El - 300300E + 12155}{32768E^8}$$

$$\langle r^9 \rangle = \frac{-2016E^5l^{10} - 10080E^5l^9 + 60480E^5l^8 + 302400E^5l^7 - 550368E^5l^6 - 2751840E^5l^5 + 1653120E^5l^4 + 8265600E^5l^3}{131072E^9}$$

$$- \frac{1161216E^5l^2 - 5806080E^5l - 55440E^4l^8 - 221760E^4l^7 + 1404480E^4l^6 + 4989600E^4l^5 - 10515120E^4l^4 - 29604960E^4l^3}{131072E^9}$$

$$+ \frac{25988160E^4l^2 + 41659200E^4l - 17005824E^4 - 240240E^3l^6 - 720720E^3l^5 + 4804800E^3l^4 + 10810800E^3l^3 - 25825800E^3l^2}{131072E^9}$$

$$- \frac{31351320E^3l + 35738560E^3 - 360360E^2l^4 - 720720E^2l^3 + 5045040E^2l^2 + 5405400E^2l - 14402388E^2 - 218790El^2 - 218790El}{131072E^9}$$

$$+ \frac{1604460E - 46189}{131072E^9}$$

the code is as following:

where $y[k]$ is the $\langle r^{k-1} \rangle$, using the python sympy library to calculate symbols operations.

```
import sympy as smp
import numpy as np
import matplotlib.pyplot as plt
x=smp.symbols('x')
E=smp.symbols('E')
l=smp.symbols('l')
N=12
i=2
rn_2=1
rn_3=-2*E
y=[rn_3, rn_2]
for i in range(2,N):
    x=1/(8*E*i)*((1-i)*(i*(i-2)-4*l*(l+1))*rn_3-4*(2*i-1)*rn_2)
    y.append(x)
    rn_3=rn_2
    rn_2=x
    i=i+1
smp.simplify(y[10])
```

Figure 1: calculating $\langle r^a \rangle$

2 Shifted Hankel matrix:

The nth level shifted Hankel matrix is as follow: $\begin{pmatrix} \langle r \rangle & \langle r^2 \rangle & \langle r^3 \rangle & \dots \\ \langle r^2 \rangle & \langle r^3 \rangle & \langle r^4 \rangle & \dots \\ \langle r^3 \rangle & \langle r^4 \rangle & \langle r^5 \rangle & \dots \\ \dots & \dots & \dots & \dots \end{pmatrix}$, i.e. $[H_s]_{ij} = \langle r^{i+j+1} \rangle_{0 \leq i,j \leq n}$

consider an operator

$$O = \sqrt{r} \sum_{i=0}^n c_i r^i$$

, when it acts on any state $|\alpha\rangle$, it's norm should be positive, hence

$$\langle \alpha | O^\dagger O | \alpha \rangle = \langle \alpha | \sum_{i,j} c_i^* c_j r^{i+j+1} | \alpha \rangle = \sum_{i,j} c_i^* c_j [H_s]_{ij} \geq 0$$

this proves that shifted Hankel matrix is positive semi-definite.

3 bootstrapping the hydrogen atom:

Using python code and scipy library, we have the following pattern.

The program first define a range of energies from -0.15 to near 0 with 100 points, then we define the recursion

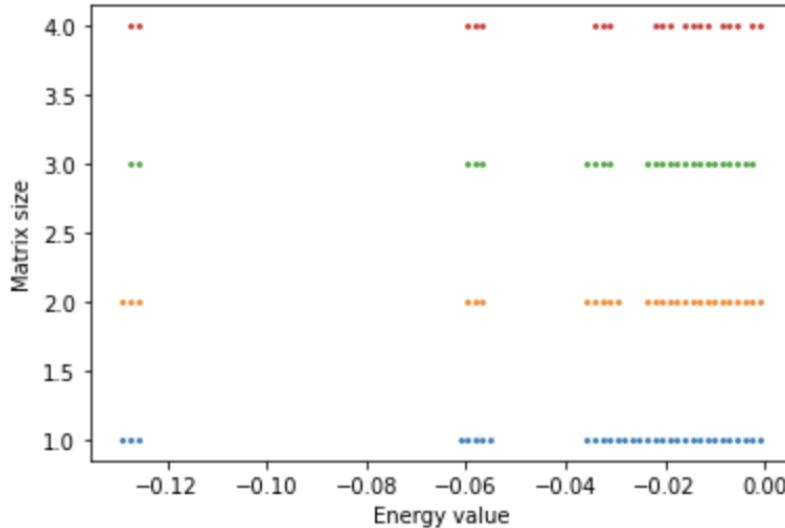


Figure 2: from size 12×12 to 15×15 , with $l=1$

relation of $\langle r^a \rangle$, and Hankel matrix as $Hankel(n, l, E)$, the for loop selects the allowed value of E_r and put it into E_sel , then we plot the allowed energy with vertical axis the size of matrix. Since the program may overflow, we use the `np.linalg.slogdet` instead of original `np.linalg.det`, also we do only the $n=12$ to $n=15$ spectrum since the program will overflow for larger sizes of Hankel matrix. From the figure we see that there are energy eigenvalues at around -0.125 , and at around -0.06 , also at around -0.03 , -0.02 and three

inside region $-0.02 \sim 0.00$. Which corresponds correctly with the spectrum of the exact hydrogen atom, $E_2 = -0.125$, $E_3 = -0.056$, $E_4 = -0.031$, $E_5 = -0.02$, $E_6 = -0.014$

```

import sympy as smp
import numpy as np
import matplotlib.pyplot as plt
from scipy.linalg import hankel
N,l=14,1
E_r=[]
E_sel=[]
y=[]
E_r=np.linspace(-0.15, -0.001, 100)
num=len(E_r)
def r(n,l,E):
    if n== -1: return -2*E
    if n==0: return 1
    else: return 1./(8.*E*(n+1))*((-n)*(n+1)*(n-1)-4*l*(l+1))*r(n-2,l,E)-4*(2*n+1)*r(n-1,l,E)
def Hankel(n,l,E):
    rk=[]
    for i in range(n+1):
        if i==0:
            rk.append(1)
        else:
            rk.append(r(i,l,E))
    return hankel(rk)[0:int(n/2)+1,0:int(n/2)+1]
for n in range(11,N+1):
    E=[]
    for j in range(num):
        M=Hankel(2*n,l,E_r[j])
        sgn, logdet=np.linalg.slogdet(M)
        if sgn>=0: E.append(E_r[j])
    E_sel.append(E)
x_val=[]
for i in range(N-10):
    x_val.append(np.full(len(E_sel[i]),i+1))
for i in range(N-10):
    plt.scatter(E_sel[i], x_val[i], 3)
plt.xlabel("Energy value")
plt.ylabel("Matrix size")

```

Figure 3: code for hydrogen atom

4 bootstrapping modified spectrum:

Theoretically speaking we can look at the r^{-2} term, if $\exists k$ such that $l(l+1) + 4 = k(k+1)$, where l is the angular momentum quantum number of the modified Hamiltonian, then it matches the state labeled by angular momentum quantum number k of the hydrogen atom. $(k-l)(k+l+1) = 4$ only gives us the $l=1$ value corresponding to hydrogen atom of $k=2$, hence we plot the following $l=1$ from $n=11$ to $n=15$.

We modify the recursion relation of $\langle r^a \rangle$ as function $r(n, l, E)$:

$$\langle r^{a-1} \rangle = \frac{1}{8aE} [(1-a)(a(a-2) - 4l(l+1) - 16)\langle r^{a-3} \rangle - 4(2a-1)\langle r^{a-2} \rangle]$$

Note that some spectrum disappear at bigger size matrix, this is because there is a precision limit of the 100 dots, if the precise value of energy is within the width of a single dot, it will disappear, hence we can just look at the trend of the figure and see that there are two energy eigenvalues at region $-0.02 \sim 0.00$ and one at about 0.02, also one between $-0.04 \sim -0.02$ and about two between $-0.06 \sim -0.04$ etc. The others especially for the energy lower than -0.06 are not obvious.

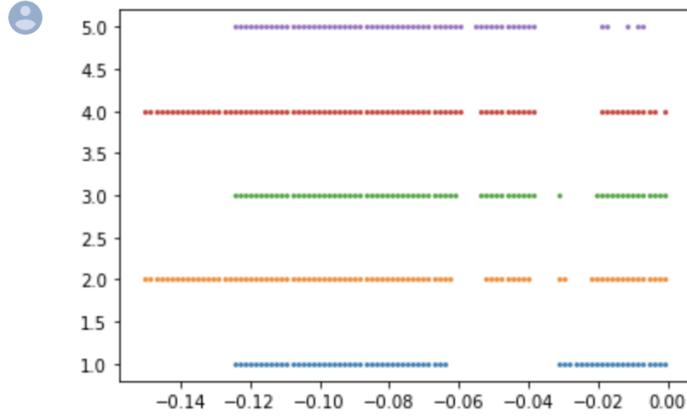


Figure 4: the dots are constrained energies and the labels 1 to 5 are the size of Hankel matrix from 11×11 to 15×15 , with $l=1$

```

import sympy as smp
import numpy as np
import matplotlib.pyplot as plt
from scipy.linalg import hankel
N,l=14,1
E_r=[]
E_sel=[]
y=[]
E_r=np.linspace(-0.15, -0.001, 100)
num=len(E_r)
def r(n,l,E):
    if n==l: return -2*E
    if n==0: return 1
    else: return 1./8./(n+1)/E*((-n)*((n+1)*(n-1)-4*l*(l+1)-16)*r(n-2,l,E)-4*(2*n+1)*r(n-1,l,E))
def Hankel(n,l,E):
    rk=[]
    for i in range(n+1):
        if i==0:
            rk.append(1)
        else:
            rk.append(r(i,l,E))
    return hankel(rk)[0:int(n/2)+1,0:int(n/2)+1]
for n in range(10,N+1):
    E=[]
    for j in range(num):
        M=Hankel(2*n,l,E_r[j])
        sgn, logdet=np.linalg.slogdet(M)
        if sgn>0: E.append(E_r[j])
    E_sel.append(E)
x_val=[]
for i in range(N-9):
    x_val.append(np.full(len(E_sel[i]),i+1))
for i in range(N-9):
    plt.scatter(E_sel[i], x_val[i], 3)

```

Figure 5: code for modified hydrogen

References

- [1] Bootstrapping Simple QM Systems, David Berenstein, George Hulsey