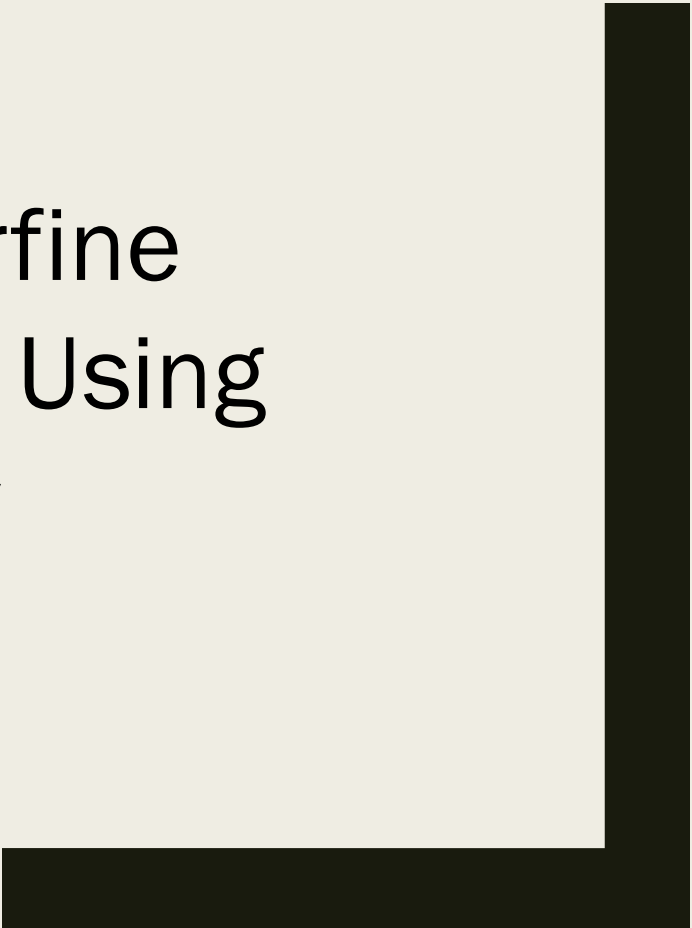
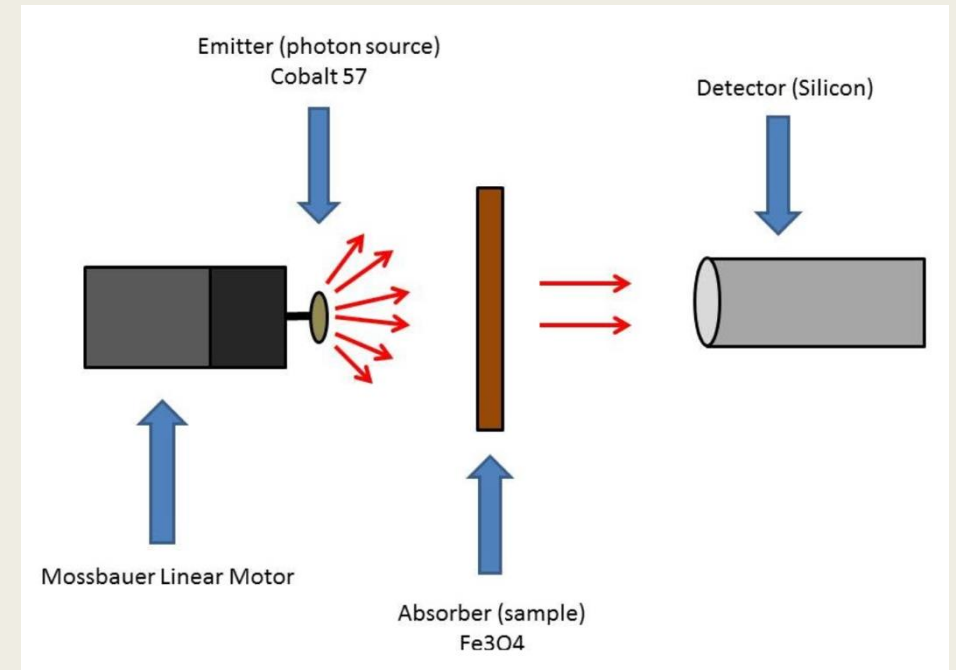


Characterization of Hyperfine Interactions in Magnetite Using Mössbauer Spectroscopy



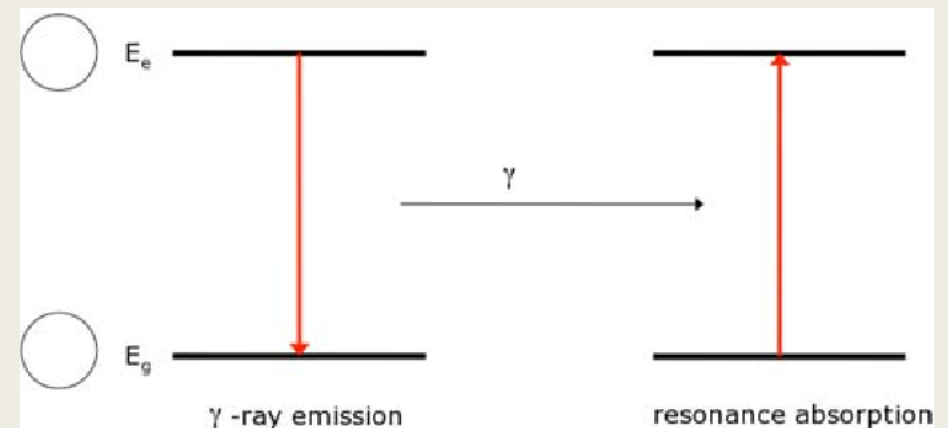
Objectives and Outline

- Goal: Characterize Three Hyperfine Interactions
 - *Isomer Shifts*
 - *Quadrupole Splitting*
 - *Magnetic Hyperfine Splitting*
- Talk Outline
 - *Overview of Mössbauer spectroscopy*
 - *What are the Hyperfine Interactions?*
 - *Experimental Design*
 - *Data Analysis*
 - *Schedule*



Mössbauer Spectroscopy – Overview

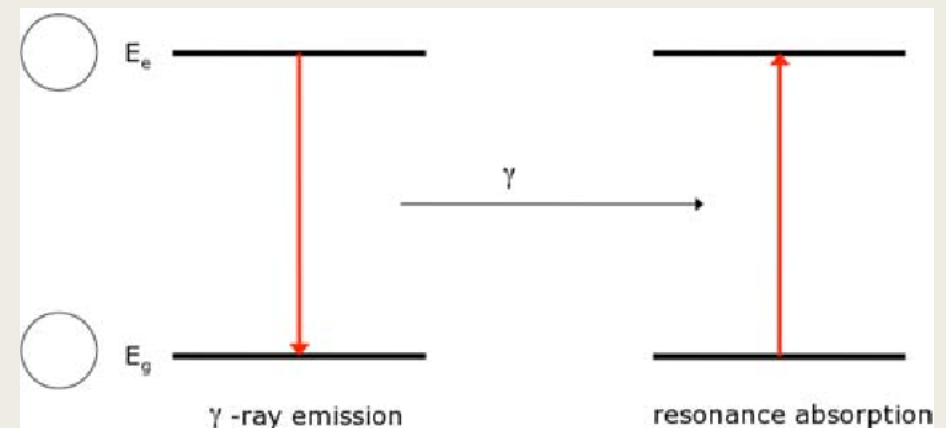
- What is Mössbauer Spectroscopy?
 - *Recoil free emission and absorption of gamma rays*



Kurian, R. (2011). First principles theoretical modeling of the isomer shift of Mossbauer spectra.

Mössbauer Spectroscopy – Overview

- What is Mössbauer Spectroscopy?
 - *Recoil free emission and absorption of gamma rays*
- What are hyperfine interactions?
 - *Very small energy shifts in the atomic energy levels of a material*
- What do they tell us?
 - *Oxidation states*
 - *Electron density*
 - *Electron symmetry*
 - *Spin states*
 - *Magnetic properties*

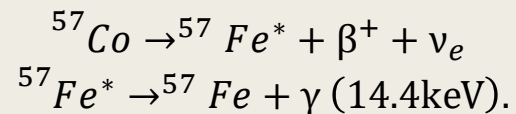


Kurian, R. (2011). First principles theoretical modeling of the isomer shift of Mossbauer spectra.

Mössbauer Spectroscopy – Source and Target

- Why Magnetite (Fe_3O_4)?
 - Contains Iron 57
 - Lattice structure
 - Fe^{2+} and Fe^{3+} Ions \rightarrow interesting hyperfine interactions

- Why Cobalt 57?
 - \rightarrow Because it decays into Iron 57!



* Excited State

PERIODIC TABLE OF ELEMENTS

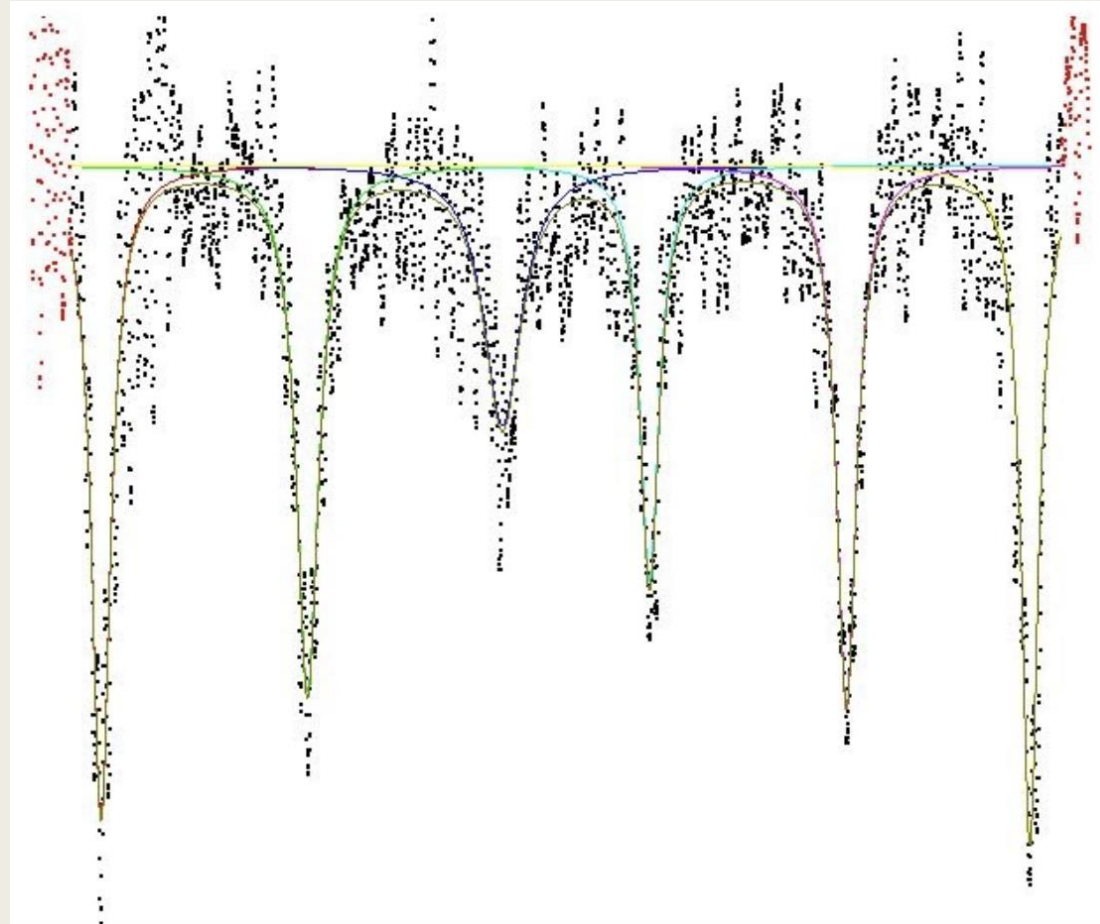
Chemical Group Block

PubChem

The image shows a standard periodic table of elements. Two boxes are highlighted with red borders: one for Iron (Fe, atomic number 26, atomic mass 55.84) and one for Cobalt (Co, atomic number 27, atomic mass 58.93319). A red arrow points from the Cobalt box to the Iron box, indicating the decay of Cobalt-57 into Iron-57. The table includes element symbols, names, atomic numbers, and atomic masses. The transition metal block is highlighted in blue.

U.S. National Library of Medicine. (n.d.). Periodic Table of elements

Mössbauer Spectroscopy – The Data

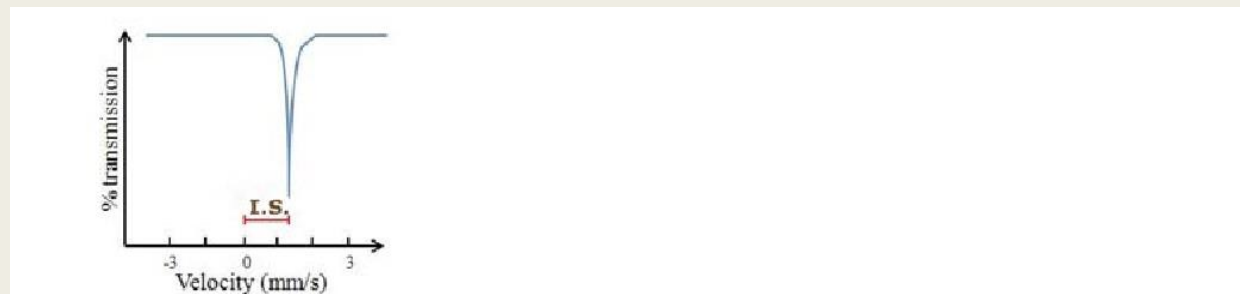


Methods of Experimental Physics (MXP) Website - Mössbauer Effect Lab.

Isomer Shift

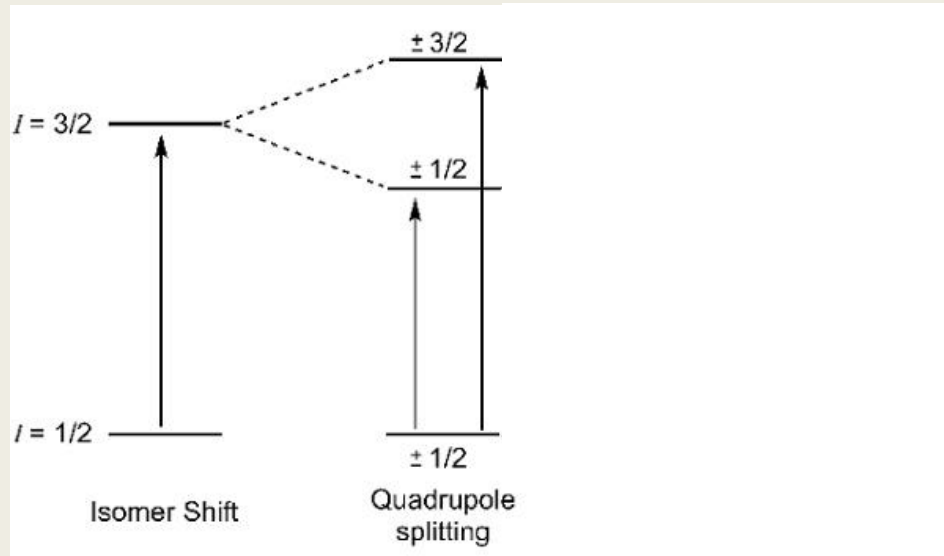


- A measure of nuclear charge distribution due to electron density at the nucleus
 - Distinguish between Fe^{2+} and Fe^{3+} ions
 - $Fe^{2+} \rightarrow$ higher oxidation state \rightarrow larger shift
- How does it affect our data?
 - Shifts peaks

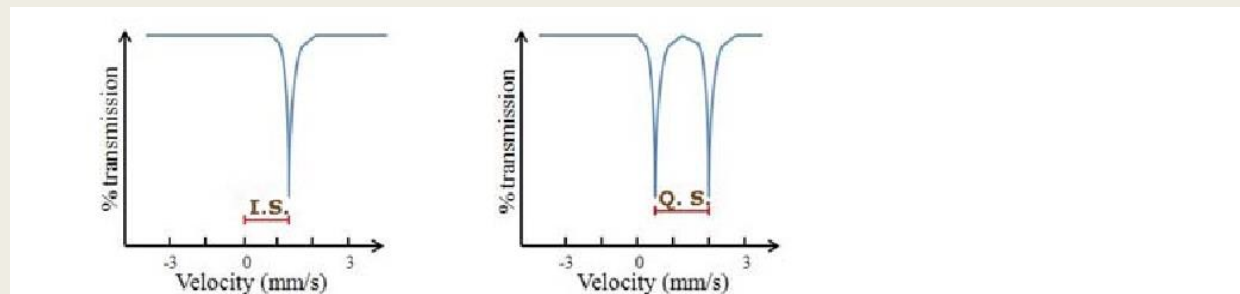


$$\delta E = K[\Psi(0)_s]^2 R^2$$

Quadrupole Splitting

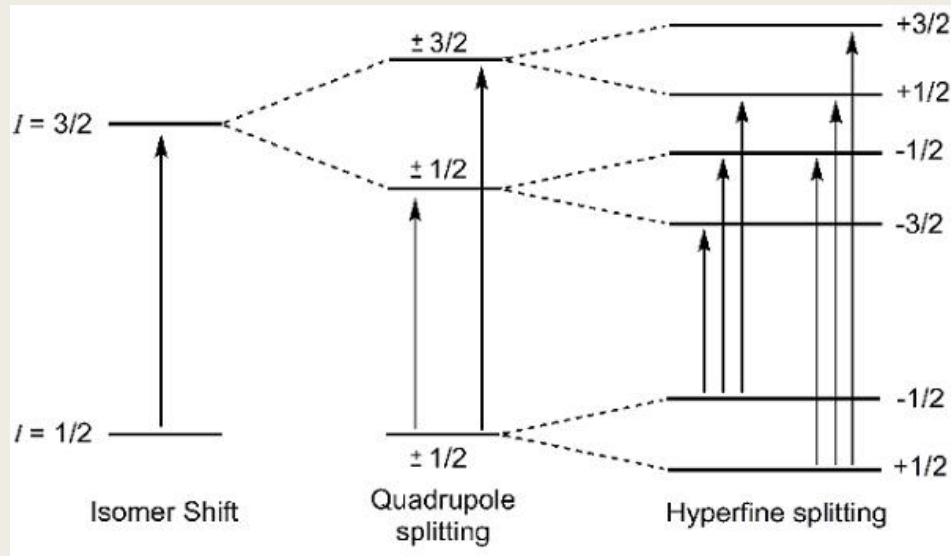


- A radially asymmetric nucleus interacts with a surrounding electric field gradient (EFG)
 - *Quadrupole moment*
 - *Excited state splits into two levels*
 - *A measure of nuclear asymmetry*
- How does it affect our data?
 - *Splitting of peaks*

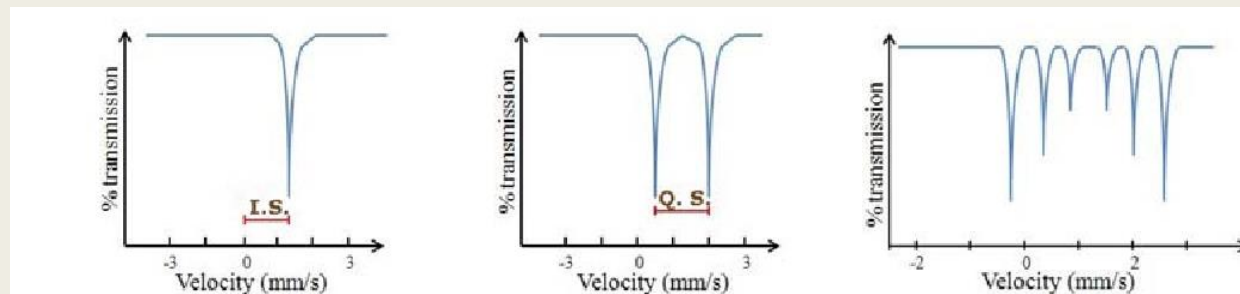


Magnetic Hyperfine Splitting

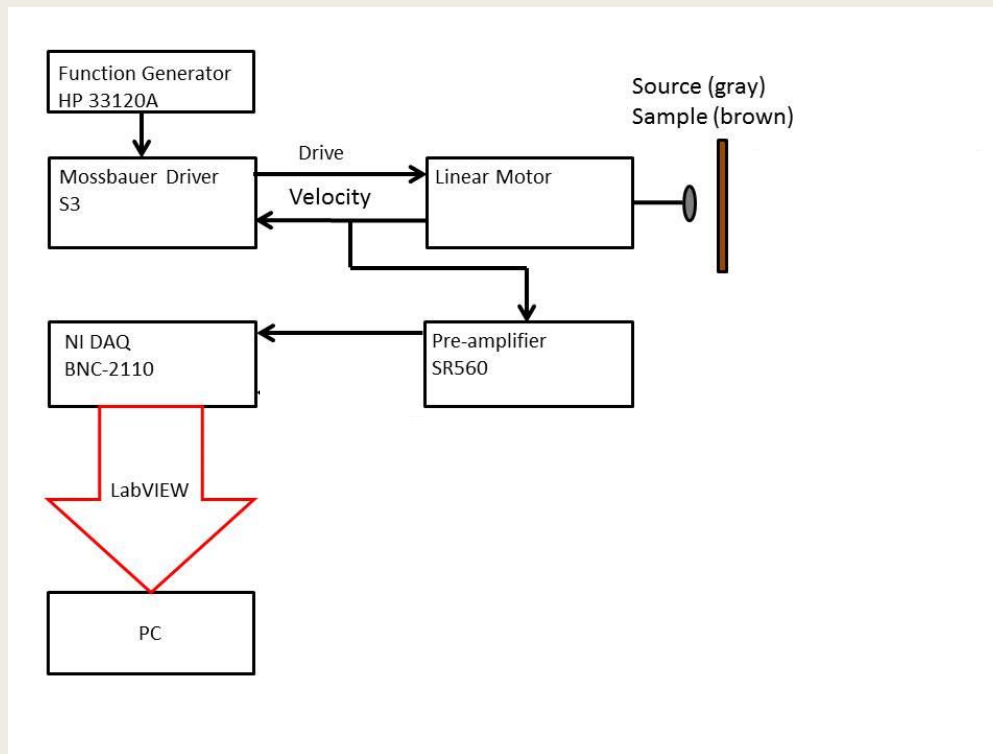
a.k.a. Zeeman Splitting or MHS



- Interaction of the nuclear magnetic dipole moment with an internal magnetic field.
 - *Magnetic ordering*
 - *Spin states*
- How does it affect our data?
 - *Characteristic splitting of peaks*

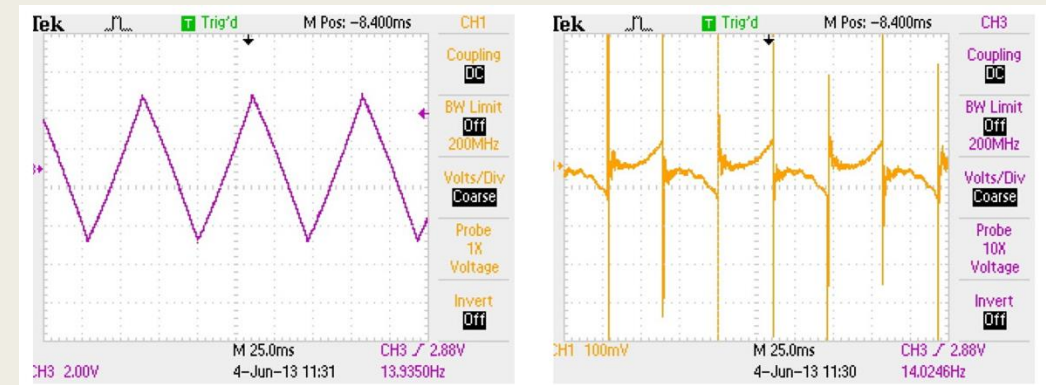


Experimental Setup – *Driver*

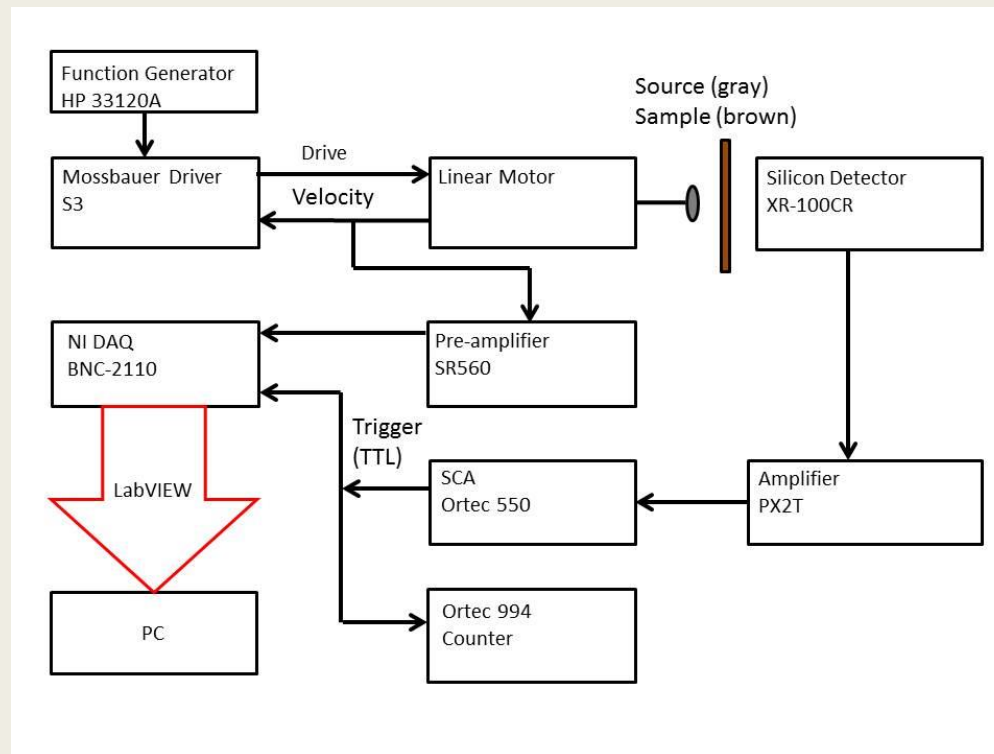


Methods of Experimental Physics (MXP) Website - Mössbauer Effect Lab.

- Source is attached to a linear motor to exploit the doppler effect
 - *Driven by a 14 Hz signal passed through a driver apparatus*
- Feedback loop between motor and driver
 - *Drive = square wave trigger signal*
 - *Velocity = triangle wave*
 - Uniform in shape



Experimental Setup – *Detector*

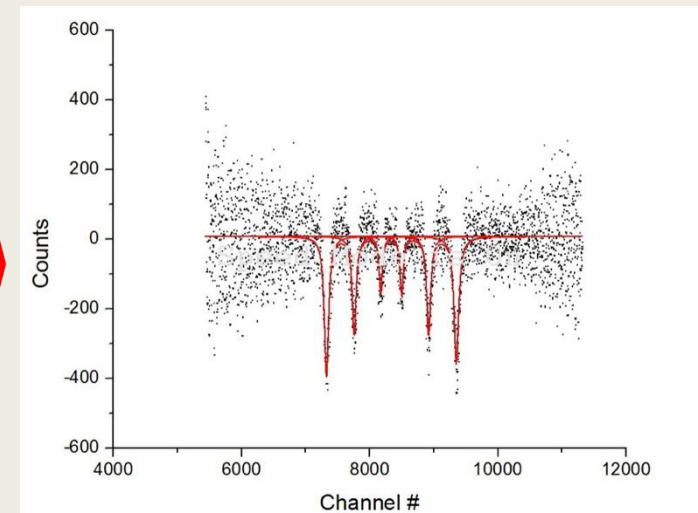
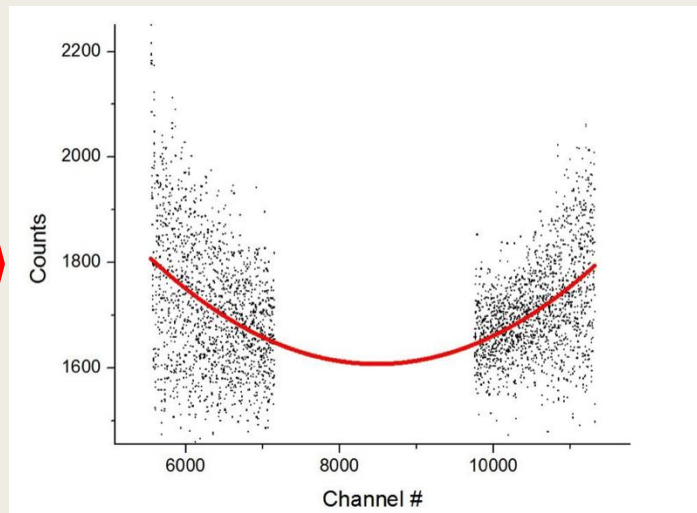
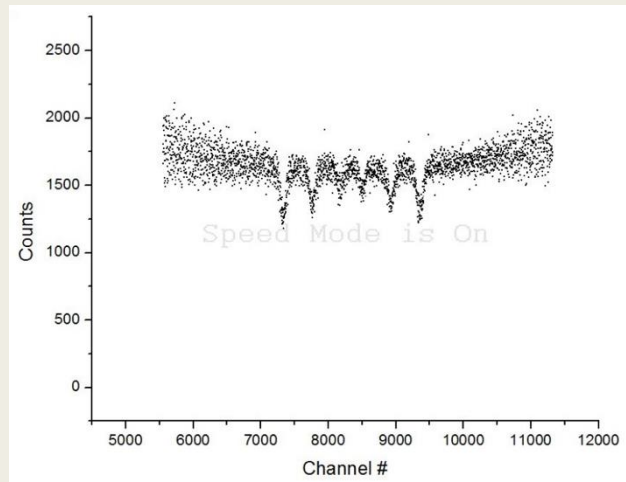


Methods of Experimental Physics (MXP) Website - Mössbauer Effect Lab.

- Silicon detector
 - *Very efficient in desired range*
 - *Photoelectric effect dominates
photon → photoelectron*
 - *Very small pulses emitted, then amplified*
- Pass to Single Channel Amplifier (SCA)
 - *Acts as a “window”*
 - *If criteria is met, emits small square wave pulse*
- Ortec counter → crude approximator

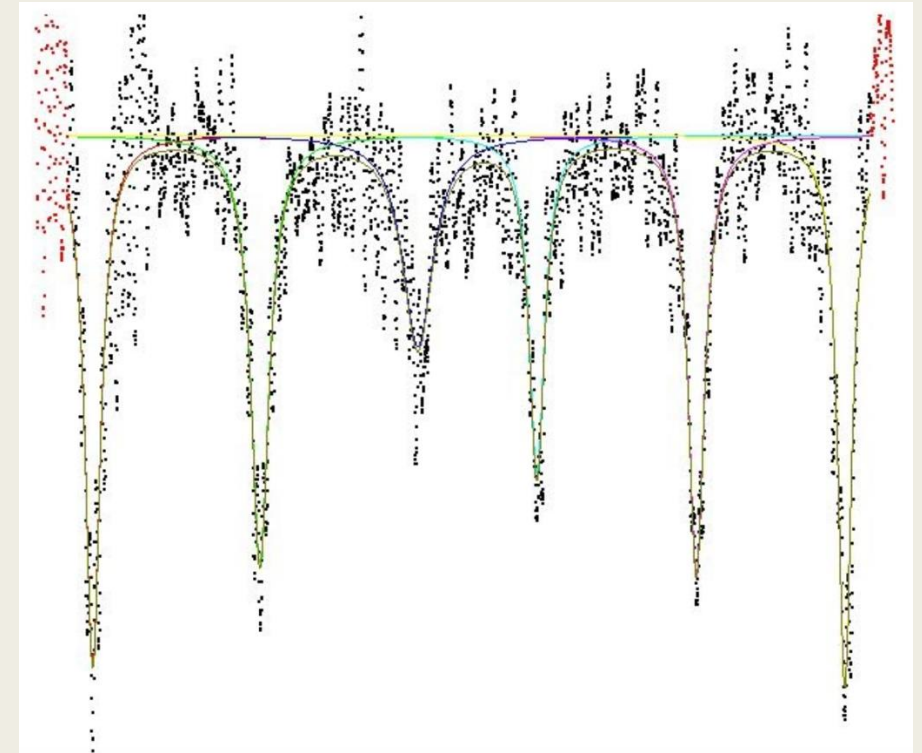
Data Analysis – *Background Subtraction*

- Edges data are trimmed to focus on important data
- Any impurities in the Velocity signal create a quadratic trend in the data
 - *Removed by fitting data without peaks*
- Convert to physical units – 10.657 mm/s
 - *Associate each channel number with a velocity and thus gamma ray energy*



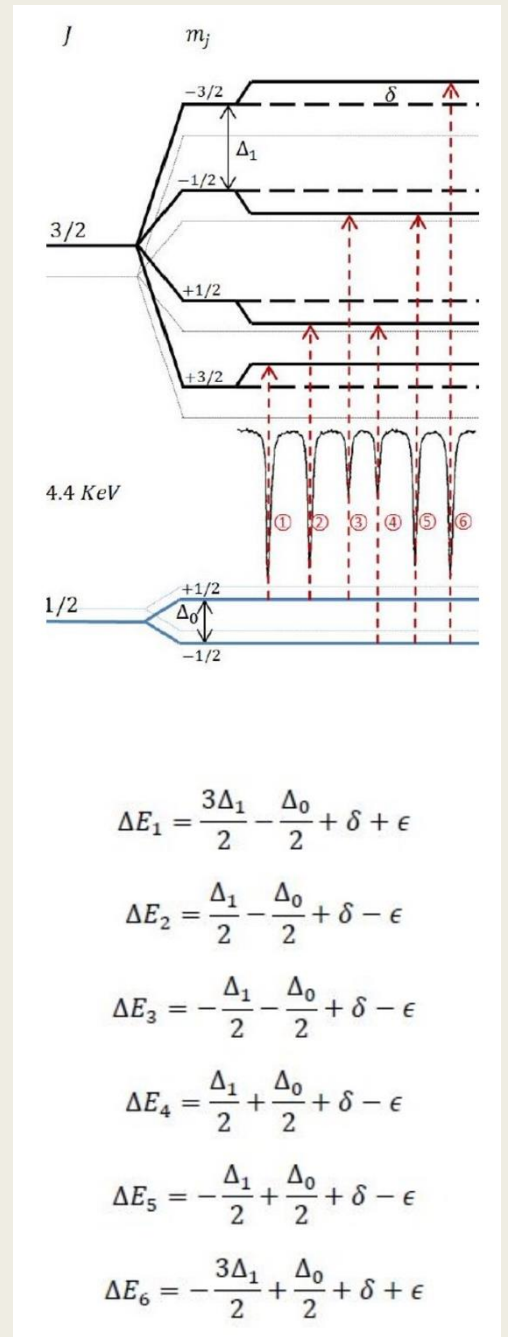
Data Analysis – Multiple Peak Fitting

- Each peak can be represented by a Lorentzian distribution
- Use Origin to:
 - *Fit peaks*
 - *Extract peak width and x axis position*
 - *Extract uncertainties*



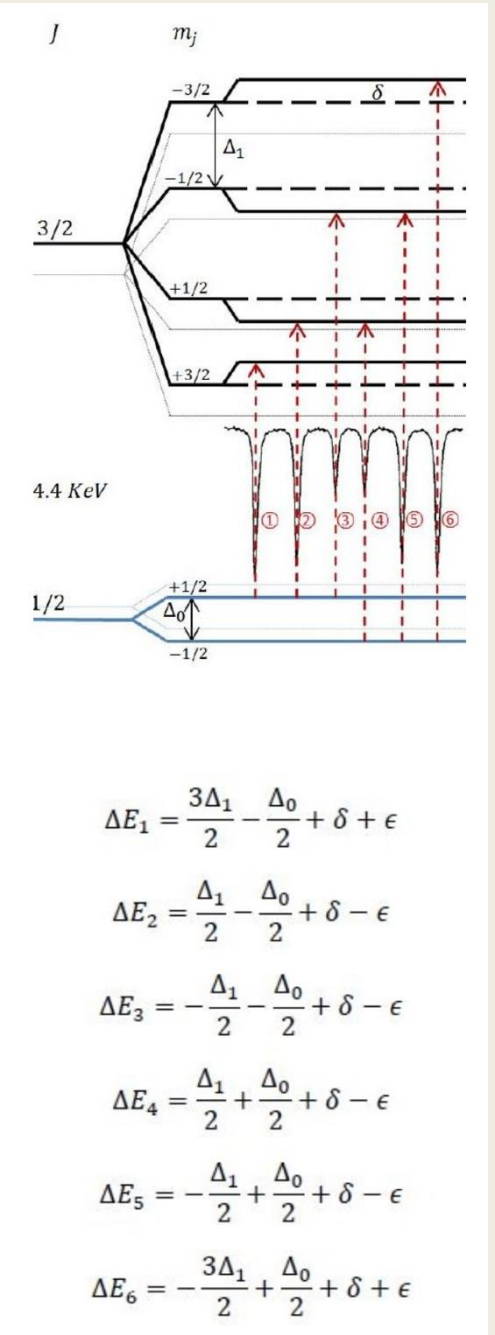
Data Analysis – Extraction of Parameters

- Peak centers tell us about hyperfine interactions!
 - **Isomer shift:** *Compare with True Center*
 - True Center: obtained from calibration data (iron foil target which has no isomer shift)



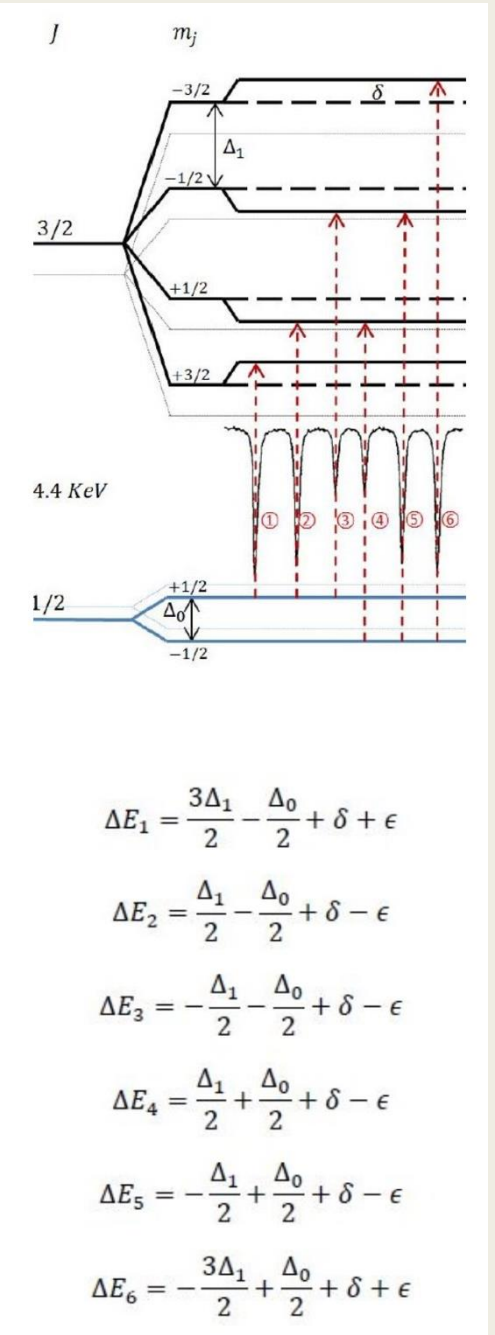
Data Analysis – Extraction of Parameters

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 - **Quadrupole splitting:** Subtract Peak 5 from 6 and 1 from 2



Data Analysis – Extraction of Parameters

- Peak centers tell us about hyperfine interactions!
 - **Isomer shift:** *Compare with True Center*
 - True Center: obtained from calibration data (iron foil target which has no isomer shift)
 - **Quadrupole splitting:** *Subtract Peak 5 from 6 and 1 from 2*
 - **MHS:** *Subtract peak 3 from peak five to get Δ_o and peak 4 from peak 5 to get Δ_1*



Schedule

- Literature Review ✓
- Initial calibration of detector and SCA using MPL americium source ✓
- Calibration using weak cobalt source ✓
- Data collection (one week)
- Data analysis
- Presentation of results

References

- U.S. National Library of Medicine. (n.d.). *Periodic Table of elements - pubchem*. National Center for Biotechnology Information. PubChem Compound Database. <https://pubchem.ncbi.nlm.nih.gov/periodic-table/>
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- Westerdale, S. "Mossbauer Spectroscopy of ^{57}Fe ", 2010
- Preston, R. S. Hanna, S. S. "Mössbauer Effect in Metallic Iron", 1962