

Data Analysis Practice

1. Give the number of significant digits in each of the following measurements.

- a) 12.04 m 4 b) 156.00 kg 5 c) 0.065 cm 2 d) 0.00320 ms 3
 e) 129 g 3 f) 2.5×10^5 s 2 g) 6.300×10^5 km 4 h) 1070 kg 3

2. a) State the "weakest link rule" for determining the number of significant digits in the resultant when multiplying or dividing measured values:

- Keep same number of significant digits as measurement with lowest significant digit count

State the "weakest link rule" for determining the precision of the resultant when adding or subtracting measured values:

- Keep same number of decimal places as measurement with lowest precision

b) Perform the following calculations, rounding off the answers to the appropriate number of significant digits. Include the correct units in your answer.

i) $136.06 \text{ g} + 2.1 \text{ g} = 138.2 \text{ g}$

ii) $1560.35 \text{ kg} + 242 \text{ kg} = 1802 \text{ Kg}$

iii) $8.5 \text{ m} + 2.194 \text{ m} - 3.45 \text{ m} = 7.2 \text{ m}$

iv) $2.55 \text{ m} \div 0.38 \text{ s} = 6.7 \text{ m/s}$

v) $4.550 \text{ N} \times 1.2 \text{ m} = 5.5 \text{ N.m}$

vi) $3.2 \text{ m} \times 1.455 \text{ m} \times 0.55 \text{ m} = 2.6 \text{ m}^3$

vii) $13.25 \text{ cm} + 42.985 \text{ cm} + 26.4 \text{ cm} - 695 \text{ mm} =$

13.1 cm

viii) $467.28 \text{ g} \div (10.6 \text{ cm} \times 3.7 \text{ cm} \times 2.75 \text{ cm}) =$

4.3 g/cm³

3. Give the units that would result from each of the following calculations.

a) $15 \text{ m/s} \div 5.0 \text{ s}$

$\frac{\text{m}}{\text{s}} \times \frac{1}{\text{s}} = \frac{\text{m}}{\text{s}^2}$

b) $4 \text{ kgm} \div 6 \text{ m/s}$

$\text{Kgm} \times \frac{1}{\text{m}} = \text{Kg.s}$

c) $5 \text{ m/s}^2 \times 4 \text{ s}^2$

$\frac{\text{m}}{\text{s}^2} \times \text{s}^2 = \text{m}$

d) $5 \text{ kgm/s}^2 \div 3 \text{ s}$

$\frac{\text{Kgm}}{\text{s}^2} \times \frac{1}{\text{s}} = \frac{\text{Kgm}}{\text{s}^3}$

e) $3 \text{ N} \div 2 \text{ m}^2$

$\frac{\text{N}}{\text{m}^2}$

f) $16 \text{ kg/m}^2 \div 4 \text{ m}$

$\frac{\text{Kg}}{\text{m}^2} \times \frac{1}{\text{m}} = \frac{\text{Kg}}{\text{m}^3}$

4. Make the following unit conversions, using the "multiply by one" method. Keep the same number of significant digits after the conversion. Express the result in scientific notation if necessary! Show your work.

a) $3360 \text{ pm} = ? \text{ nm}$ $1 \text{ nm} = 10^3 \text{ pm}$

$$3360 \text{ pm} \times \frac{1 \text{ nm}}{10^3 \text{ pm}} = 3.36 \text{ nm}$$

c) $0.00785 \text{ mm} = ? \mu\text{m}$ $1 \text{ mm} = 10^3 \mu\text{m}$

$$7.85 \times 10^{-3} \text{ mm} \times \frac{10^3 \mu\text{m}}{1 \text{ mm}} = 7.85 \mu\text{m}$$

e) $79 \text{ km/h} = ? \text{ m/s}$ $1 \text{ km} = 1000 \text{ m}$
 $1 \text{ h} = 3600 \text{ s}$

$$\frac{79 \text{ km}}{\text{h}} \times \frac{1 \text{ h}}{3600 \text{ s}} \times \frac{1000 \text{ m}}{1 \text{ km}} = 21.94 \frac{\text{m}}{\text{s}} \sim 22 \text{ m/s}$$

g) $494 \text{ Mg} = ? \text{ Gg}$ $1 \text{ Gg} = 10^3 \text{ Mg}$

$$494 \text{ Mg} \times \frac{1 \text{ Gg}}{10^3 \text{ Mg}} = 0.494 \text{ Gg}$$

b) $678.2 \text{ kg} = ? \text{ Mg}$ $1 \text{ Mg} = 10^3 \text{ kg}$ ✓

$$\textcircled{2} \quad \frac{678.2 \text{ kg}}{1} \times \frac{1 \text{ Mg}}{10^3 \text{ kg}} = 0.6782 \text{ Mg} \textcircled{3}$$

d) $1.5 \text{ year} = ? \text{ seconds}$

$$1.5 \text{ y} \times \frac{365 \text{ day}}{\text{y}} \times \frac{24 \text{ h}}{\text{day}} \times \frac{3600 \text{ s}}{\text{h}} = 4.7 \times 10^7 \text{ s}$$

f) $6.3 \text{ m/s}^2 = ? \text{ km/h}^2$ $1 \text{ Km} = 1000 \text{ m}$
 $1 \text{ h}^2 = (3600 \text{ s})^2$

$$\frac{6.3 \text{ m}}{\text{s}^2} \times \frac{1 \text{ km}}{1000 \text{ m}} \times \frac{(3600 \text{ s})^2}{1 \text{ h}^2} = 8.2 \times 10^4 \frac{\text{Km}}{\text{h}^2}$$

h) $0.000056 \text{ m}^3 = ? \text{ cm}^3$ $1 \text{ m}^3 = (10^2 \text{ cm})^3$
 $= 10^6 \text{ cm}^3$

$$5.6 \times 10^{-5} \text{ m}^3 \times \frac{10^6 \text{ cm}^3}{1 \text{ m}^3} = 5.6 \times 10^1 \text{ cm}^3$$

$$= 56 \text{ cm}^3$$

5. a) State the formula for finding percent uncertainty in a measurement given the absolute uncertainty:

$$\% \text{ uncertainty} = \frac{\text{abs uncertainty}}{\text{measurement}} \times 100 \%$$

- b) Convert the following measurement uncertainties between absolute and percent uncertainty as indicated:

Quantity with absolute uncertainty	Quantity with percent uncertainty
$24.0 \pm 0.5 \text{ kg}$	$= 24.0 \text{ kg} \pm 2 \%$
$23.65 \pm 0.05 \text{ s}$	$23.65 \text{ s} \pm 0.2\%$
$8.135 \text{ m} \pm 0.005 \text{ m}$	$8.135 \text{ m} \pm 0.06\%$
$0.128 \pm 0.005 \text{ g}$	$0.128 \text{ g} \pm 3.9\%$
$4.8 \pm 0.5 \text{ km}$	$4.8 \text{ km} \pm 10\%$

Quantity with percent uncertainty	Quantity with absolute uncertainty
$37.5 \text{ kg} \pm 2 \%$	$37.5 \pm 0.8 \text{ kg}$
$156 \text{ m} \pm 1 \%$	$156 \pm 2 \text{ m}$
$6.3 \text{ s} \pm 5 \%$	$6.3 \pm 0.3 \text{ s}$
$23.6 \text{ cm} \pm 0.5 \%$	$23.6 \pm 0.1 \text{ cm}$
$18 \text{ g} \pm 3 \%$	$18 \pm 0.5 \text{ g}$

- 6.a) State the following rules for combining measurement uncertainties:

When adding or subtracting measured values we ADD the ABSOLUTE uncertainties.

When multiplying or dividing measured values we ADD the RELATIVE uncertainties.

- b) Find the resultant for each calculation below and determine the absolute uncertainty.

In cases involving multiplication or division, you may need to convert to relative or percent uncertainties first.

i) $L_{\text{total}} = L_1 + L_2$, $L_1 = 5.35 \text{ m} \pm 0.02 \text{ m}$, $9.87 \text{ m} \pm 0.02 \text{ m}$

$$\begin{aligned} L_{\text{TOTAL}} &= 5.35 \text{ m} + 9.87 \text{ m} \pm (0.02 + 0.02) \text{ m} \\ &= 15.22 \pm 0.04 \text{ m} \end{aligned}$$

ii) $m_{\text{total}} = m_1 + m_2 + m_3$, $m_1 = 14.3 \pm 0.2 \text{ kg}$, $m_2 = 2.6 \pm 0.1 \text{ kg}$, $m_3 = 7.4 \pm 0.1 \text{ kg}$

$$\begin{aligned} m_{\text{TOTAL}} &= 14.3 \text{ kg} + 2.6 \text{ kg} + 7.4 \text{ kg} \pm (0.2 + 0.1 + 0.1) \text{ kg} \\ &= 24.3 \pm 0.4 \text{ kg} \end{aligned}$$

iii) $\Delta t = t_2 - t_1$, $t_1 = 18.6 \pm 0.1 \text{ s}$, $t_2 = 39.5 \pm 0.1 \text{ s}$

$$\begin{aligned} \Delta t &= (39.5 - 18.6) \text{ s} \pm (0.1 + 0.1) \text{ s} = 20.9 \pm 0.2 \text{ s} \end{aligned}$$

iv) $A = L \times W$, $L = 1.64 \pm 0.05 \text{ m}$, $W = 0.65 \pm 0.05 \text{ m}$

$$\begin{aligned} A &= 1.64 \text{ m} \times 0.65 \text{ m} \quad \% \text{ unc} = \left(\frac{0.05}{1.64}\right)100\% + \left(\frac{0.05}{0.65}\right)100\% \\ &\approx 1.0725 \text{ m}^2 \\ &\sim 1.1 \text{ m}^2 \end{aligned}$$

$$A = 1.1 \pm 0.1 \text{ m}^2$$

v) $D = m / V$, $m = 67.4 \text{ g} \pm 1\%$, $V = 18.2 \text{ cm}^3 \pm 2\%$

$$\begin{aligned} D &= \frac{67.4 \text{ g}}{18.2 \text{ cm}^3} \pm (1+2)\% \\ &= 3.7 \pm 3\% \end{aligned}$$

$$D = 3.7 \pm 0.1 \text{ g/cm}^3$$

vi) $v = \Delta d / \Delta t$, $\Delta d = 25.8 \pm 0.2 \text{ m}$, $\Delta t = 6.3 \text{ s} \pm 0.2 \text{ s}$

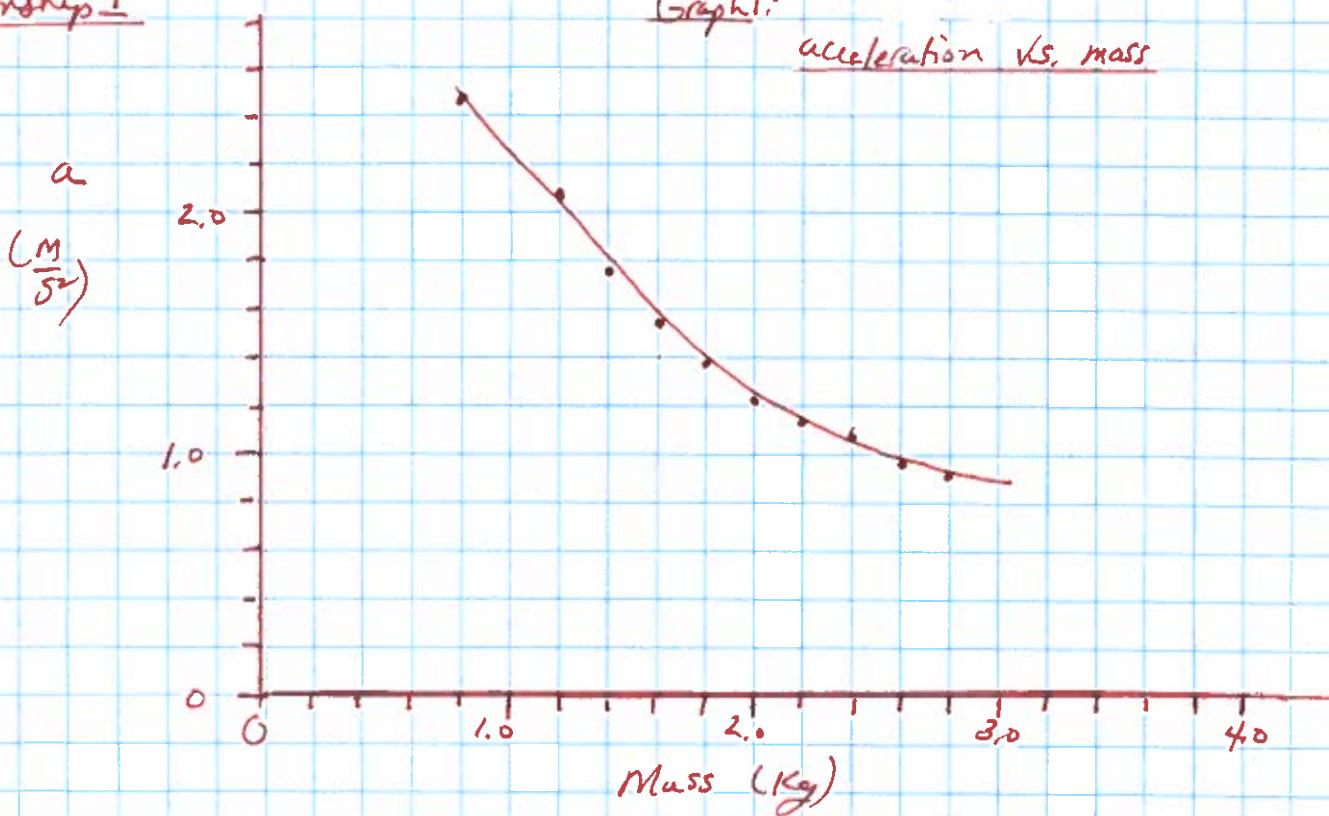
$$\begin{aligned} v &= \frac{25.8 \text{ m}}{6.3 \text{ s}} \quad \% \text{ unc} = \left(\frac{0.2}{25.8}\right)100\% + \left(\frac{0.2}{6.3}\right)100\% \\ &= 4.075 \text{ m/s} \\ &= 4.1 \pm 0.2 \text{ m/s} \end{aligned}$$

$$v = 4.1 \pm 0.2 \text{ m/s}$$

GRAPHICAL ANALYSIS

①

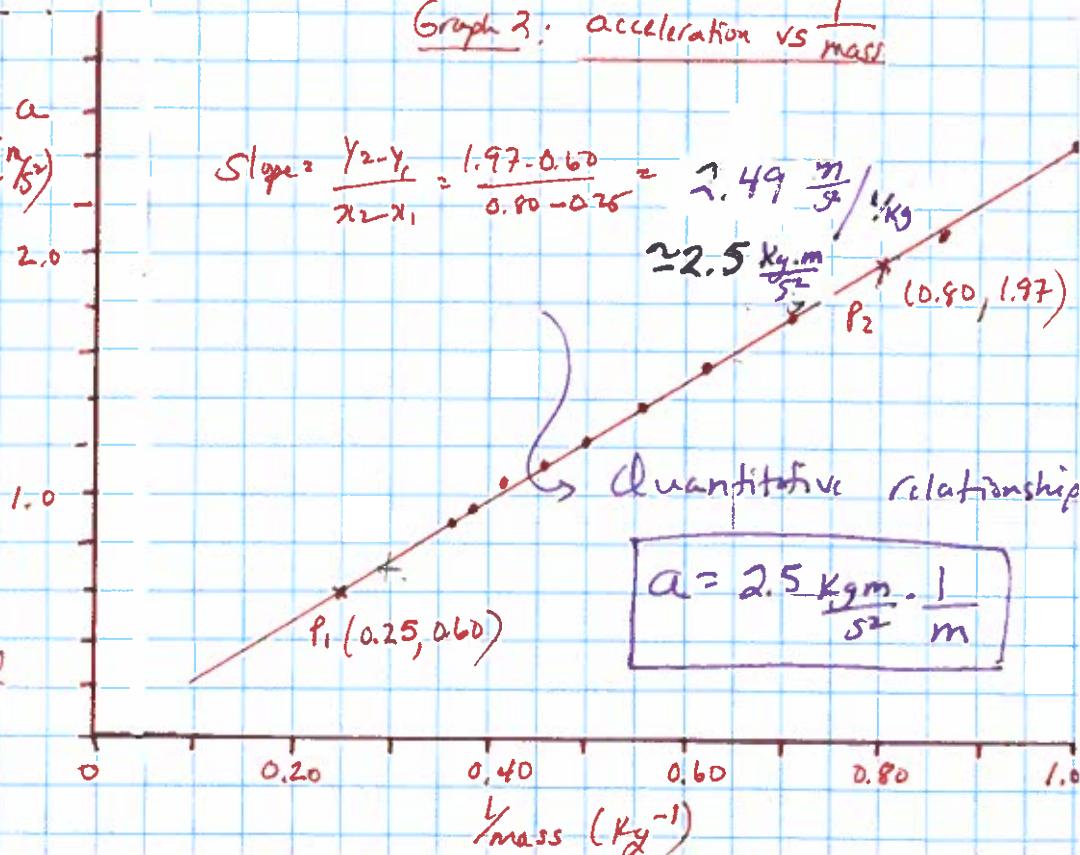
Relationship 1



Analysis: - The curve appears to demonstrate an inverse relationship: $a \propto \frac{1}{m}$

Table 2: Test first if $a \propto \frac{1}{m}$ ($n=1$); Plot a graph of a vs $\frac{1}{m}$.

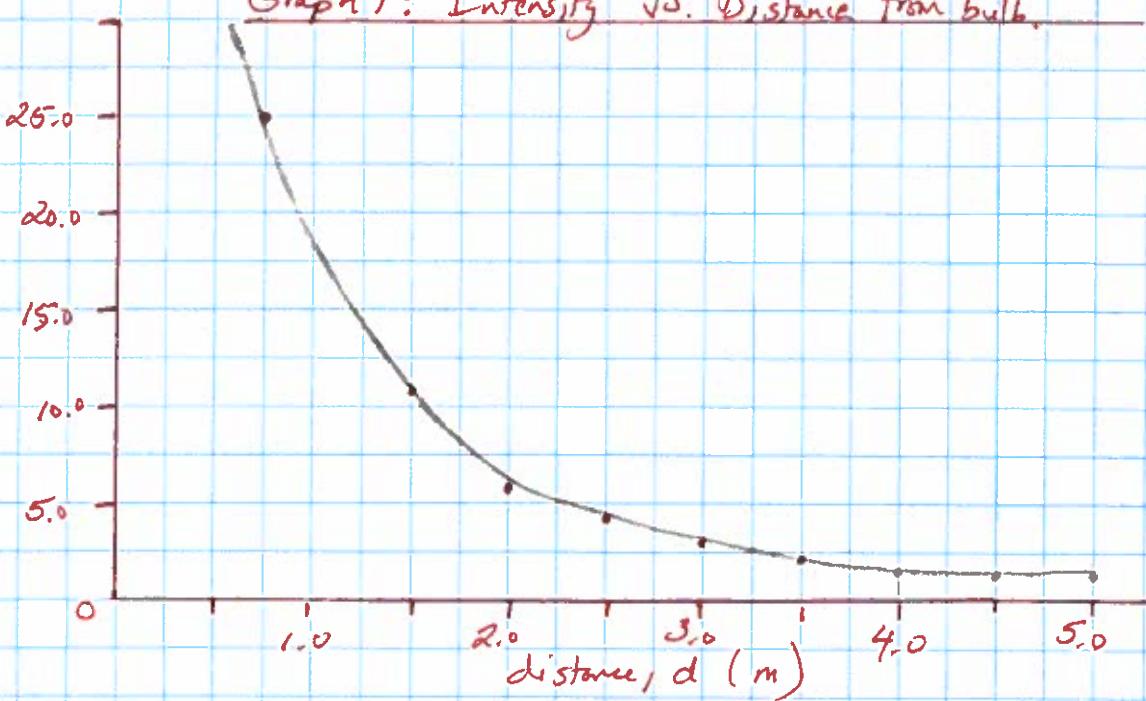
Mass M (kg)	Inverse Mass $\frac{1}{m}$ (kg⁻¹)	Acceleration a (m/s²)
1.00	1.00	2.46
1.20	0.833	2.08
1.40	0.714	1.76
1.60	0.625	1.56
1.80	0.556	1.39
2.00	0.500	1.22
2.20	0.455	1.13
2.40	0.417	1.05
2.60	0.385	0.94
2.80	0.357	0.89



∴ Acceleration is directly proportional to inverse mass
 $a \propto \frac{1}{m}$

2)

Intensity
 I
($\frac{W}{m^2}$)

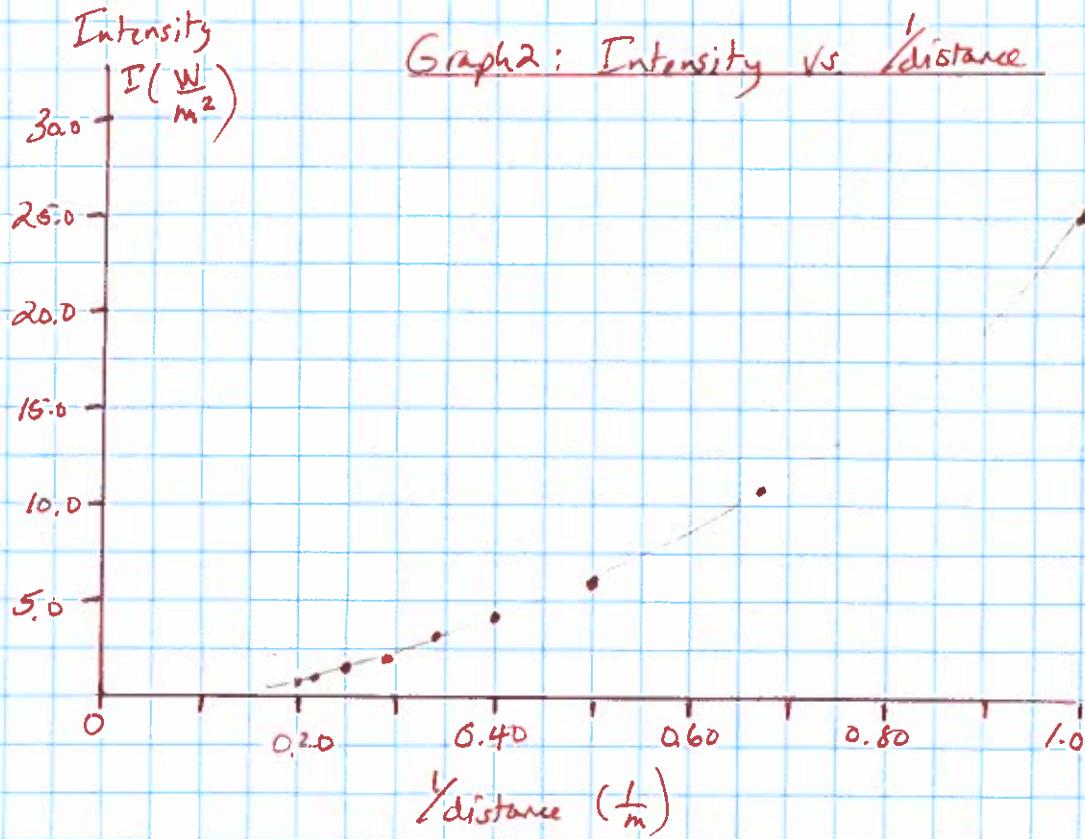


(2)

Analysis: The curve appears to demonstrate an inverse relationship ($I \propto \frac{1}{d}$).

Table 2: Test first if $I \propto \frac{1}{d}$, where $n=1$. Plot graph of I vs $\frac{1}{d}$.

Distance d (m)	Inverse distance $\frac{1}{d}$ (1/m)	Intensity I ($\frac{W}{m^2}$)
1.0	1.0	25.0
1.5	0.67	11.1
2.0	0.50	6.3
2.5	0.40	4.0
3.0	0.33	2.8
3.5	0.29	2.0
4.0	0.25	1.6
4.5	0.22	1.2
5.0	0.20	1.0



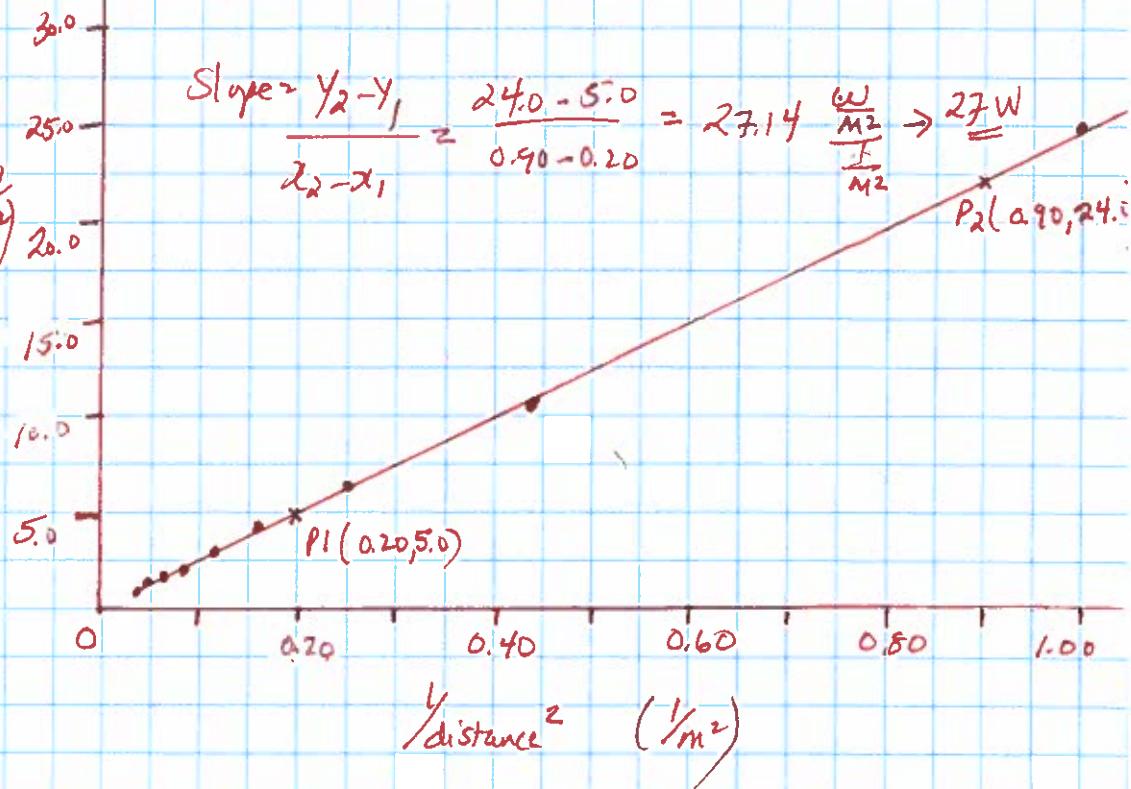
Analysis: Although this curve appears closer to a linear trend, there is still a significant deviation from a straight line.

We will ~~try~~ try the next largest integer value of n ($n=2$) to test if this is an inverse square relationship.

(3)

Graph 3: Intensity vs $\frac{1}{\text{distance squared}}$

Distance d (m)	$\frac{1}{d^2}$ (m^{-2})	Intensity I (W/m^2)
1.0	1.0	25.0
1.5	0.44	11.1
2.0	0.25	6.3
2.5	0.16	4.0
3.0	0.11	2.8
3.5	0.082	2.0
4.0	0.062	1.6
4.5	0.049	1.2
5.0	0.040	1.0



Analysis: The third graph appears linear. This verifies that Intensity is proportional to the inverse square of distance.

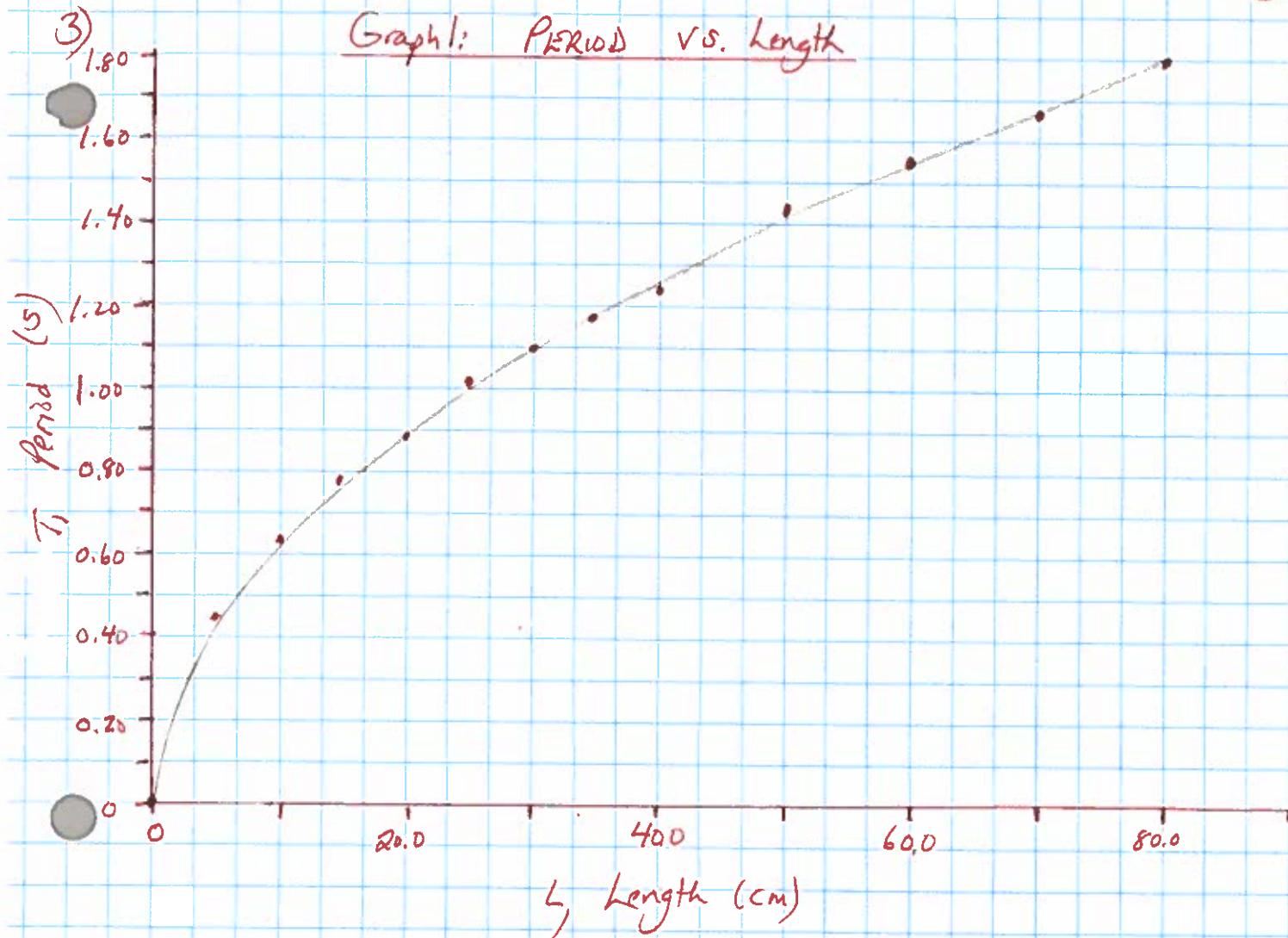
$$I \propto \frac{1}{d^2}$$

The slope provides the constant of proportionality

$$I = (27 \text{ W}) \cdot \frac{1}{d^2}$$

(4)

Graph: Period vs. Length



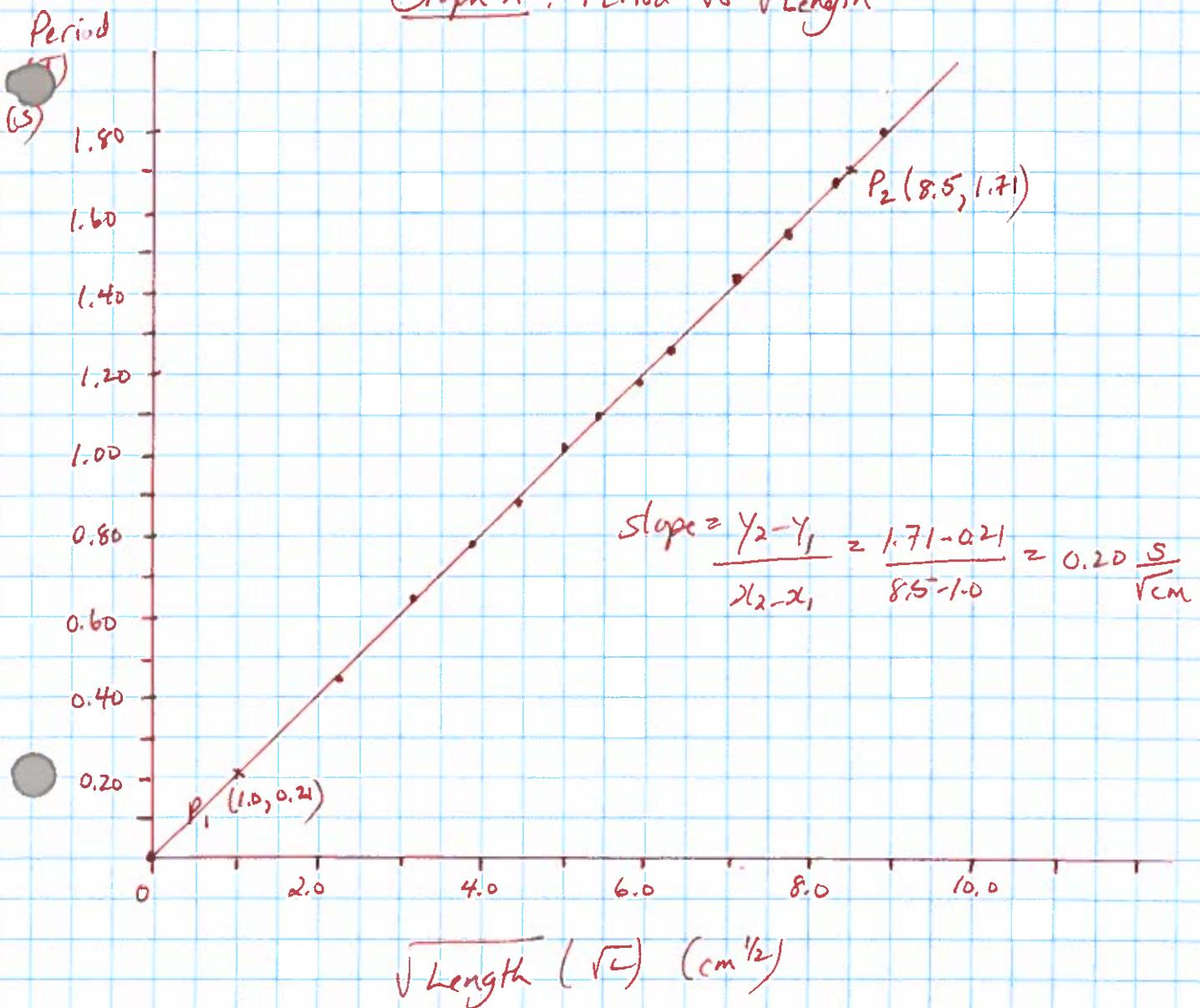
Analysis: The curve appears to demonstrate a root relationship. ($T \propto \sqrt{L}$), $n=1$

* Test first if $T \propto \sqrt{L}$ ($n=1$). Plot a graph of T vs \sqrt{L} .

Length (L), cm	Length, $\sqrt{\text{cm}}$	Period (T) s
0.0	0.0	0.0
5.0	2.24	0.45
10.0	3.16	0.63
15.0	3.87	0.78
20.0	4.47	0.89
25.0	5.00	1.02
30.0	5.48	1.10
35.0	5.92	1.18
40.0	6.32	1.25
50.0	7.07	1.43
60.0	7.75	1.58
70.0	8.37	1.68
80.0	8.94	1.80

(5)

Graph 2 : Period Vs $\sqrt{\text{Length}}$



Analysis: This graph is linear which verifies that period is directly proportional to the square root of pendulum length:

$$T \propto \sqrt{L}$$

The constant of proportionality is determined from the slope which gives the complete relationship:

$$T = 0.20 \frac{\text{s}}{\sqrt{\text{cm}}} \sqrt{L}$$

