In which of the following is vanadium in the lowest oxidation state? 2.

- (a)
- (b)
- $\begin{array}{c} \mathsf{V_2O_5} \\ \mathsf{VO_2} \\ \mathsf{VO^+} \\ \mathsf{V_4O_9^{2-}} \end{array}$ (c) (d)