

# Ammonia emission abatement does not fully control reduced forms of nitrogen deposition

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Contributed by John H. Seinfeld, February 28, 2020 (sent for review November 15, 2019; reviewed by Jeffrey Lee Collett Jr and Maria Kanakidou)

Human activities and population growth have increased the natural burden of reactive nitrogen (N) in the environment. Excessive N deposition on Earth's surface leads to adverse feedbacks on ecosystems and humans. Similar to that of air pollution, emission control is recognized as an efficient means to control acid deposition. Control of nitrogen oxides ( $\text{NO}_x = \text{NO} + \text{NO}_2$ ) emissions has led to reduction in deposition of oxidized nitrogen ( $\text{NO}_y$ , the sum of all oxidized nitrogen species, except nitrous oxide [ $\text{N}_2\text{O}$ ]). Reduced forms of nitrogen ( $\text{NH}_x = \text{ammonia} [\text{NH}_3] + \text{ammonium} [\text{NH}_4^+]$ ) deposition have, otherwise, increased, offsetting the benefit of reduction in  $\text{NO}_y$  deposition. Stringent control of  $\text{NH}_3$  emissions is being considered. In this study, we assess the response of N deposition to N emission control on continental regions. We show that significant reduction of  $\text{NH}_x$  deposition is unlikely to be achieved at the early stages of implementing  $\text{NH}_3$  emission abatement. Per-unit  $\text{NH}_3$  emission abatement is shown to result in only 60–80% reduction in  $\text{NH}_x$  deposition, which is significantly lower than the demonstrated 80–120% benefit of controlling  $\text{NO}_x$  emissions on  $\text{NO}_y$  deposition. This 60–80% effectiveness of  $\text{NH}_x$  deposition reduction per unit  $\text{NH}_3$  emission abatement reflects, in part, the effects of simultaneous reductions in  $\text{NO}_x$  and  $\text{SO}_2$  emissions.

nitrogen deposition | reduced forms of nitrogen | ammonia emission | emission control strategy

Atmospheric deposition is the principal pathway for the exchange of nitrogen (N) between Earth's compartments. Elevated N deposition on Earth's surface has occurred (1–5), due to increasing human consumption of energy, with estimates of continuing growth in the near future (1, 3, 4, 6, 7). However, future increases of N deposition depend largely on the projection of future N emissions, which in some cases show slowdown or even stabilization trends (1). Response to excessive N deposition is exhibited mainly by eutrophication and acidification of terrestrial and aquatic ecosystems (8–11). The well-studied influences are damage and toxicity to vegetation and soil systems (12), impact on natural dominant species (13, 14), and even loss of ecosystem biodiversity (15). Therefore, prevention of excessive atmospheric N deposition is of vital importance in the protection of a sustainable ecosystem and in avoiding irreversible future damages.

Implementation of emission controls is an effective avenue to reduce acid deposition (3, 16–18). In the United States, oxidized nitrogen ( $\text{NO}_y$ , sum of all oxidized nitrogen species except nitrous oxide [ $\text{N}_2\text{O}$ ]) deposition, a result of nitrogen oxides ( $\text{NO}_x = \text{NO} + \text{NO}_2$ ) emissions from combustion, has been reduced since implementation of the Clean Air Act (CAA) in the 1970s. Nonetheless, rapid growth of the deposition of reduced forms of nitrogen ( $\text{NH}_x = \text{ammonia} [\text{NH}_3] + \text{ammonium} [\text{NH}_4^+]$ ) has taken place (19) and offsets the benefits of  $\text{NO}_y$  deposition reduction.  $\text{NH}_x$  deposition results from wet deposition (precipitation) and dry deposition (driven by turbulent and molecular diffusion processes that bring  $\text{NH}_3$  into contact with surfaces) of gaseous  $\text{NH}_3$  and particulate  $\text{NH}_4^+$ . Relatively few studies have

examined the effect of reducing  $\text{NH}_3$  emissions on  $\text{NH}_x$  deposition. However, the effectiveness of  $\text{NH}_3$  emission abatement on reducing airborne fine-particle levels has proven lower than expected (20–23), since conversion between  $\text{NH}_3$  and  $\text{NH}_4^+$  particulate matter is highly dependent on acidic aerosol components such as sulfate ( $\text{SO}_4^{2-}$ ) and nitrate ( $\text{NO}_3^-$ ). A similar situation is likely to arise associated with the control of  $\text{NH}_x$  deposition.

Here, we address the impact of adding  $\text{NH}_3$  emissions control to  $\text{NO}_x$  and  $\text{SO}_2$  emission control as a means to reduce  $\text{NH}_x$  deposition over continental regions, as well as a comparison with the benefit of  $\text{NO}_x$  emission abatement on reducing  $\text{NO}_y$  deposition. We first review the consistency between the spatio-temporal trends of  $\text{NH}_3$  emissions and  $\text{NH}_x$  deposition over the conterminous United States (CONUS) based on both national data and site measurement data. Then, we examine the response of  $\text{NH}_x$  deposition to controlling  $\text{NH}_3$  emissions via modeling analysis of emission control scenarios.

## Results and Discussion

**Temporal Changes of N Emissions and N Deposition.** We analyzed the temporal trends of N emissions, N wet deposition, and  $\text{NO}_2$  ambient concentrations with measured data in nine regions of CONUS from 2000 to 2017 (Fig. 1). Site measurements of  $\text{NH}_3$  ambient concentrations are available after 2007, but the numbers of sites in regions (*SI Appendix, Table S1*) are considerably lower than those of  $\text{NO}_2$  ambient concentration and N deposition (*SI Appendix, Table S2*), especially before 2011. Thus, temporal trends of  $\text{NH}_3$  ambient concentration are not included in Fig. 1,

## Significance

Control of nitrogen oxides emissions has successfully reduced the deposition of oxidized nitrogen over the United States. Recent studies have called attention to the rapid growth of deposition of reduced forms of nitrogen, which would offset the benefits of reduction in oxidized nitrogen deposition. Our analysis, based on both monitoring and modeling approaches, illustrates that implementing ammonia emission abatement may not lead to full response in nitrogen deposition over continental regions. Here we provide important information to the community and policy makers concerning nitrogen deposition, given the urgent needs for developing control strategies on ammonia emissions.

Author contributions: J.T. and J.S.F. designed research; J.T. performed research; J.T. analyzed data; and J.T., J.S.F., and J.H.S. wrote the paper.

Reviewers: J.L.C., Colorado State University; and M.K., University of Crete.

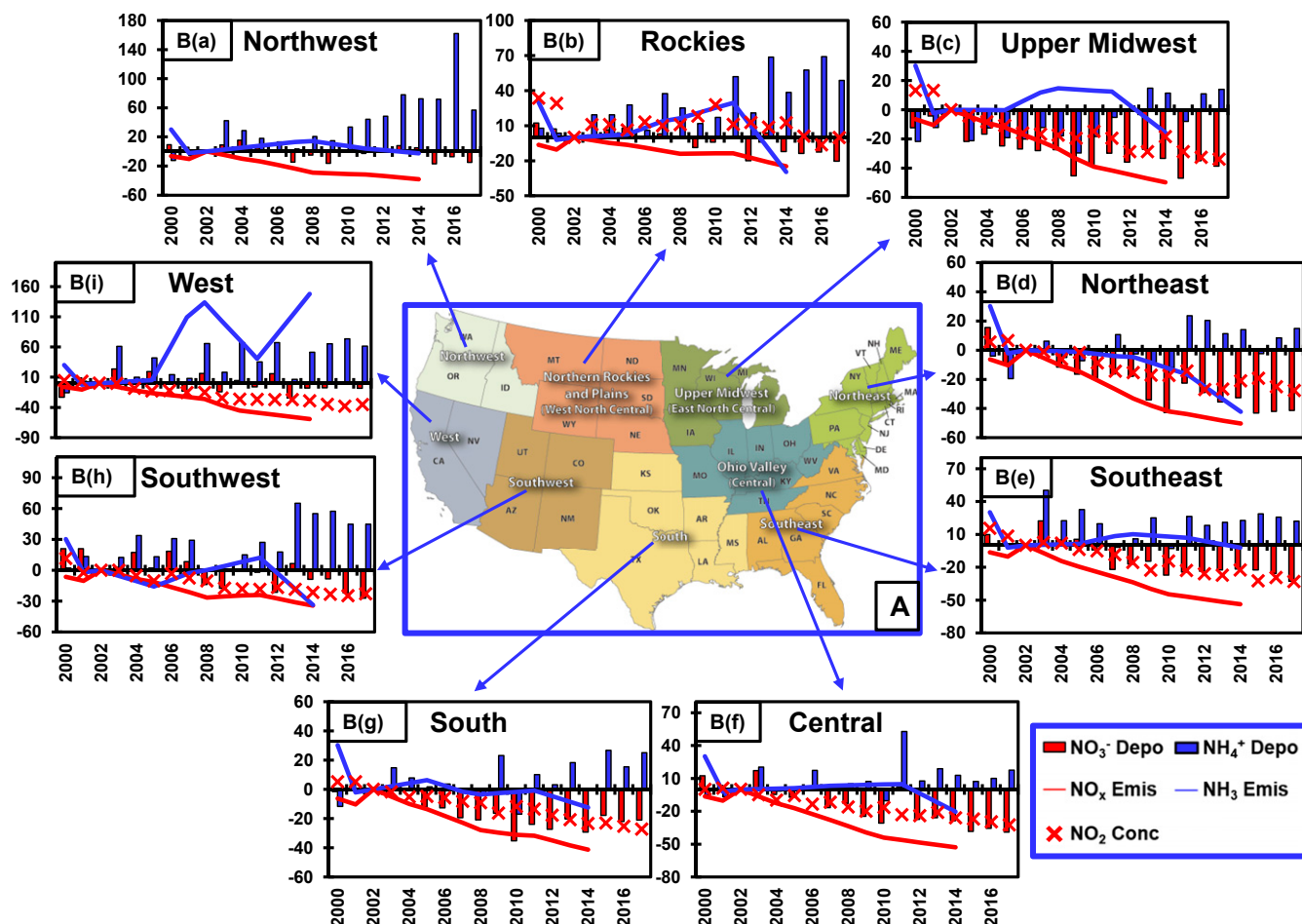
The authors declare no competing interest.

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This article contains supporting information online at <https://www.pnas.org/lookup/suppl/doi:10.1073/pnas.1920068117/-DCSupplemental>.

First published April 20, 2020.



**Fig. 1.** Trends of annual N emissions and N deposition over nine CONUS regions from 2000 to 2017. (A) Definition of the nine CONUS regions in this study. (B, a–i) Percentage changes of N emission,  $\text{NO}_2$  concentration, and N deposition in regions (unit: %). Airborne concentration and deposition values were averages over all sites in the regions. The percentage changes were calculated based on the 2002 level, calculated as  $100 \times (\text{Emission of target year} - \text{Emission of 2002}) / (\text{Emission of 2002})$ .

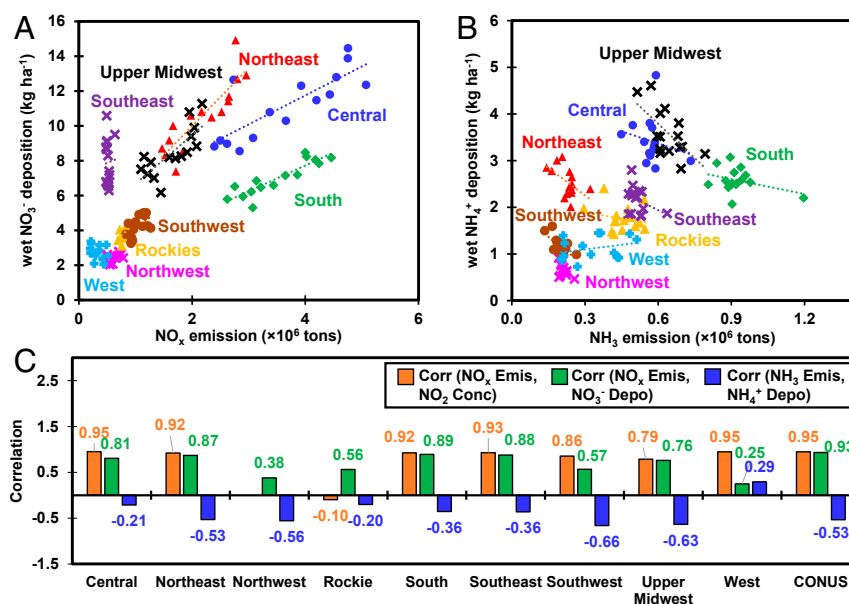
but are shown with the trends of N emission and deposition in *SI Appendix, Fig. S1*.  $\text{NO}_3^-$  and  $\text{NH}_4^+$  wet deposition and ambient  $\text{NO}_2$  concentration are annual average values over all sites in the regions.  $\text{NO}_x$  and  $\text{NH}_3$  emissions are state-level values derived from the National Emissions Inventory (NEI) (see *Materials and Methods* for details).

$\text{NO}_x$  emissions (Fig. 1B, red line) peaked in 2002 in all CONUS regions, with amounts dropping by  $\sim 10\%$  each year following for most regions, except the Northern Rockies and Plains (hereafter denoted Rockies) and the Southwest. This reduction of  $\text{NO}_x$  emissions, associated with implementation of regulations and policies subsequent to the 1970 CAA, is reflected in the reduction of  $\text{NO}_2$  levels (Fig. 1B, red X's).  $\text{NO}_3^-$  wet deposition (Fig. 1B, red bars) also started to decrease after 2002 in all regions, although levels in all regions did not decrease at the same rate. For instance, 7 y were required for the Rockies to show a significant decrease of  $\text{NO}_3^-$  wet deposition, while clear and rapid (1–2 y after 2002) responses were found in the Upper Midwest, Northeast, South, and Central United States. Response times between decreases of  $\text{NO}_x$  emissions and  $\text{NO}_y$  wet deposition were associated with the reduction rates of  $\text{NO}_x$  emissions, and the atmospheric acidity and humidity (23–25), which determine the gas-aerosol partitioning of  $\text{NO}_3^-$ .

Overall, clear consistency exists between changes of  $\text{NO}_x$  emissions and  $\text{NO}_y$  wet deposition in the temporal trends in all regions. The relationship between  $\text{NH}_3$  emissions and  $\text{NH}_x$  wet

deposition, on the other hand, is considerably more complex than that between  $\text{NO}_x$  emissions and  $\text{NO}_y$  deposition. For example, a decrease in  $\text{NH}_4^+$  wet deposition occurred during an increase of  $\text{NH}_3$  emissions during 2004–2012 in the Upper Midwest (Fig. 1B, c). As well, with continuously decreasing  $\text{NH}_3$  emissions after 2006 (Fig. 1B, d), the Northeast continued to experience increases in  $\text{NH}_4^+$  wet deposition after 2011. Increases occurred in  $\text{NH}_3$  airborne concentration in the Central and Northeast regions during 2011–2014 when  $\text{NO}_x$  and  $\text{NH}_3$  emissions and  $\text{NO}_3^-$  and  $\text{NH}_4^+$  wet deposition were decreased (*SI Appendix, Fig. S1*). The increased  $\text{NH}_3$  air concentration is likely caused by release of  $\text{NH}_4^+$  from the aerosol phase as  $\text{SO}_4^{2-}$  and  $\text{NO}_3^-$  concentrations and associated aerosol water decreased. Generally, it is difficult to establish a clear relationship between the trends of  $\text{NH}_3$  emissions and  $\text{NH}_x$  wet deposition.

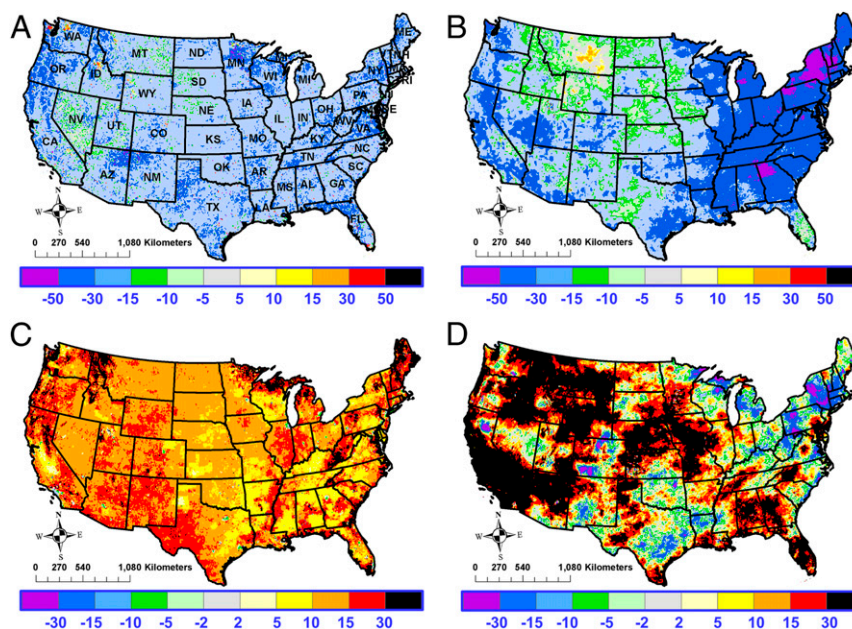
Temporal changes of  $\text{NO}_3^-$  wet deposition exhibited strongly positive linear correlations with those of  $\text{NO}_x$  emissions for several regions, such as Northeast, Central, and South United States (Fig. 24), with correlations (R values) between these trends of  $\sim 0.8$  (Fig. 2C, green bars). On the other hand, correlations between  $\text{NH}_3$  emissions and  $\text{NH}_4^+$  wet deposition were either unclear (Central, Rockies, South, Southeast, and West) or weakly negative (Northeast, Northwest, Southwest, and Upper Midwest) (Fig. 2B). Moreover, R values between the trends were low (Fig. 2C, blue bars).



**Fig. 2.** Relationships between N emissions and N deposition. Scatter plots of regional emissions with deposition for (A)  $\text{NO}_x$  emission and  $\text{NO}_3^-$  wet deposition, and (B)  $\text{NH}_3$  emission and  $\text{NH}_4^+$  wet deposition from 2000 to 2014 for nine CONUS regions. (C) Correlations between  $\text{NO}_x$  emissions and  $\text{NO}_2$  concentration (orange bars), between  $\text{NO}_x$  emissions and  $\text{NO}_3^-$  wet deposition (green bars), and between  $\text{NH}_3$  emissions and  $\text{NH}_4^+$  wet deposition (blue bars). Northwest lacked sufficient  $\text{NO}_2$  observations.

**Spatial Changes of N Emissions and N Deposition.** Spatial distributions of percentage changes in N emissions and N deposition (wet + dry) from 2001 to 2010 are shown in Fig. 3. Annual maps of N deposition were developed by combining the site observation data with deposition fields simulated by a chemical transport model provided by the Total Deposition Science Committee (TDEP) of the National Atmospheric Deposition Program (NADP) (*Materials and Methods*). Over this period,  $\text{NO}_x$  emissions (Fig. 3A) decreased by 15–30% over CONUS, with slight increases in some regions, such as Montana (MT) and Nevada

(NV).  $\text{NO}_y$  deposition responses (Fig. 3B) were generally consistent with the changes in  $\text{NO}_x$  emissions in both amount and distribution.  $\text{NH}_3$  emissions (Fig. 3C) increased by 10–30% over CONUS, whereas  $\text{NH}_x$  deposition (Fig. 3D) exhibited mixed responses in different states. We found that the locations of areas with increased  $\text{NH}_x$  deposition agreed well with areas of increased  $\text{NO}_y$  deposition (Fig. 3B, contours with yellow and orange) and increased S deposition (*SI Appendix, Fig. S2*) in large extended regions around the states of MT, California (CA), Nebraska (NE), and Texas (TX). Changes in  $\text{NH}_4^+$  wet



**Fig. 3.** Spatial distributions of the percentage changes of N emissions (derived from NEI) and deposition (derived from TDEP) from 2001 to 2010. Percentage changes of (A)  $\text{NO}_x$  emissions, (B)  $\text{NO}_y$  deposition, (C)  $\text{NH}_3$  emissions, and (D)  $\text{NH}_x$  deposition. The percentage changes were calculated as  $100 \times (\text{Emission of 2010} - \text{Emission of 2001}) / (\text{Emission of 2001})$ .



deposition were strongly influenced by the acidic components of wet deposition (mainly  $\text{SO}_4^{2-}$  and  $\text{NO}_3^-$ ) since  $\text{NH}_4^+$  is the main alkali neutralizer. Thus, in regions with decreased  $\text{NO}_y$  deposition, only a portion of the increased  $\text{NH}_3$  emissions partitioned to the aerosol phase ( $\text{NH}_4^+$ ). Once  $\text{NH}_3$  emissions are controlled, the remaining  $\text{NH}_3$  in the atmosphere might partition to particulate  $\text{NH}_4^+$  as the aerosol becomes more acidic (26).

**Response of N Deposition to N Emission Abatement.** We calculated the response (Res) of deposition of  $\text{NO}_y$  or  $\text{NH}_x$  to emission control over continental regions with Eq. 1, using results from 11 global climate models under base case (5) and emission perturbation scenarios (27) (*Materials and Methods*).

$$\text{Res} = \frac{\text{Predicted \% change of } \text{NO}_y \text{ or } \text{NH}_x \text{ deposition under emission changes over continental regions}}{\text{Predicted \% change of } \text{NO}_x \text{ or } \text{NH}_3 \text{ emissions under emission changes over continental regions}} \quad [1]$$

The response of  $\text{NO}_y$  deposition to controlling  $\text{NO}_x$  emissions (Fig. 4A) generally ranged from 80 to 120% over most CONUS regions. The  $\pm 20\%$  variation around 100% could be a result of meteorological factors such as precipitation and long-range transport, in agreement with findings from Hemispheric Transport of Air Pollution, second phase (HTAP II) simulations (25), and could be viewed as impacts of long-term climate change. Some areas in MT, South Dakota (SD), and NE exhibited extraordinary values ( $\sim 200\%$ ) (Fig. 4A, rust color), owing to very low ( $<5\%$ ) reductions in  $\text{NO}_x$  emissions (*SI Appendix, Fig. S3A*). Overall, the average Res value of  $\text{NO}_y$  deposition approached the ideal value of 100%.

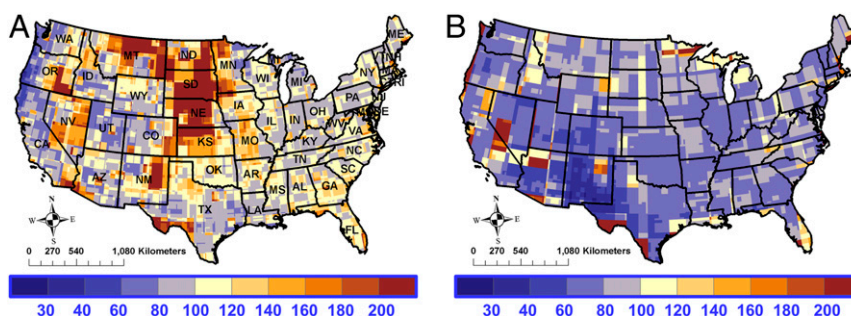
The response of  $\text{NH}_x$  deposition to controlling  $\text{NH}_3$  emissions ranged from 60 to 80% (Fig. 4B) in most CONUS regions. Spatial distributions of the Res values were more consistent among nearby regions for  $\text{NH}_x$  deposition (Fig. 4B) than  $\text{NO}_y$  deposition (Fig. 4A). A possible explanation is that  $\text{NO}_x$  can be transported over relatively long distances via reservoir species (27). Gaseous  $\text{NH}_3$  is removed mainly by wet and dry deposition, with only minor consumption in slow oxidation of  $\text{NH}_3$  by OH (24). Importantly, the 20–40% shortfall to “full” effects indicates that additional reduction of  $\text{NH}_3$  emissions is needed to reach the target for  $\text{NH}_x$  deposition.

**Implications and Summary.** Despite a tight connection between  $\text{NO}_y$  deposition and  $\text{NO}_x$  emissions, such a strong relationship

does not exist between  $\text{NH}_3$  emissions and  $\text{NH}_x$  deposition. In this study we controlled the emissions of  $\text{SO}_2$ ,  $\text{NO}_x$ , and  $\text{NH}_3$  simultaneously. It mimicked the situation that we are likely to face in the future, in which  $\text{NO}_x$  and  $\text{SO}_2$  emission control strategies remain in effect. As  $\text{NH}_x$  deposition increased,  $\text{NH}_3$  emission control was added to the existing control policies. We find that  $\text{NH}_x$  deposition would resist decrease after  $\text{NH}_3$  emissions are reduced. Reduction in  $\text{SO}_2$  and  $\text{NO}_x$  emissions leads to reduction of aerosol  $\text{SO}_4^{2-}$  and  $\text{NO}_3^-$  but also to reduction in the associated aerosol water; thus, the level of acidity does not change appreciably (24). The overall result is promotion of the release of  $\text{NH}_4^+$  to the gas phase, which increases the dry deposition of  $\text{NH}_x$ .

The sensitivity of  $\text{NH}_x$  dry deposition to  $\text{NH}_3$  emission reduction is not included in the present study. Vegetation can serve as either a sink or source of atmospheric  $\text{NH}_3$  depending on the balance between the  $\text{NH}_3$  concentrations in the atmosphere and those in the stomata of leaves, which is referred to as the compensation point (28). This bidirectional  $\text{NH}_3$  flux can increase the model prediction of  $\text{NH}_3$  emission flux by 0–70% (29). It also increased the model prediction of  $\text{NH}_3$  air concentration by 0–75%, with high influences during June–October and low impacts during December–February (30). A modeling study employing the Long Term Ozone Simulation–European Operational Smog (LOTOS-EUROS) model (31) with compensation point found that  $\text{NH}_3$  dry deposition flux would only decrease about 10–15% (seasonal variations) when  $\text{NH}_3$  emissions were decreased by 30%, and would increase about 15–25% when  $\text{NH}_3$  emissions were increased by 30%. The compensation can be viewed as a buffering effect that weakens the sensitivity of  $\text{NH}_x$  dry deposition to  $\text{NH}_3$  emission change.

It is concluded that significant reduction of  $\text{NH}_x$  deposition is not likely to emerge in the early stage of  $\text{NH}_3$  emission control. Our modeling results estimate that a 60–80% reduction of  $\text{NH}_x$  deposition can be achieved per unit of  $\text{NH}_3$  emission control, considerably lower than the full benefits of controlling  $\text{NO}_x$  emissions to reduce  $\text{NO}_y$  deposition. The management of  $\text{NH}_x$  deposition is likely to be a focus for many regions in the near future. This “weakened” sensitivity needs to be taken into consideration



**Fig. 4.** Predicted responses of N deposition (wet plus dry) to N emissions abatement in 2010. (A) Response of  $\text{NO}_y$  deposition to  $\text{NO}_x$  emission control. (B) Response of  $\text{NH}_x$  deposition to  $\text{NH}_3$  emissions control.

when developing control strategies for  $\text{NH}_3$  emissions. The results of the present study for the United States are likely applicable to regions with intensive  $\text{NH}_x$  deposition rates, such as India, and regions in which the dominant deposition components are in transition from  $\text{NO}_y$  to  $\text{NH}_x$ , such as Europe.

## Materials and Methods

Site observations of wet  $\text{NO}_3^-$  deposition and wet  $\text{NH}_4^+$  deposition were provided by the National Trends Network (NTN) of the NADP (<http://nadp.slh.wisc.edu/NTN/>). NTN measured the wet deposition at 373 sites located throughout the entire United States (*SI Appendix, Fig. S4*) from 1980s until present day. This study used the annual accumulated wet deposition data from 244 sites with available data in the study period (*SI Appendix, Table S2*). Site observations of airborne  $\text{NO}_2$  concentrations were provided by Air Quality System Monitoring Network of US Environmental Protection Agency (US EPA) (<https://www.epa.gov/air-trends/nitrogen-dioxide-trends>) (*SI Appendix, Fig. S5 and Table S2*). Spatial distributions of  $\text{NO}_y$  and  $\text{NH}_x$  deposition were developed by the TDEP by spatial interpolation of quality-controlled observation sites with model simulations (<http://nadp.slh.wisc.edu/NTN/maps.aspx>). Annual amounts of  $\text{NO}_x$  and  $\text{NH}_3$  emissions were provided by US EPA's NEI (<https://www.epa.gov/air-emissions-inventories/national-emissions-inventory-nei>). Spatial distributions of emissions were obtained from the Emissions Database for Global Atmospheric Research (EDGARv4.3.2) (<https://edgar.jrc.ec.europa.eu/overview.php?v=432>) at  $0.1 \times 0.1^\circ$  resolution ( $\sim 11.1 \text{ km} \times 11.1 \text{ km}$ ). The EDGAR emissions over CONUS were developed based on information from US EPA's NEI (32).

The responses of deposition to emission changes were determined from simulated  $\text{NO}_y$  and  $\text{NH}_x$  deposition from multimodel mean (MMM) results of 11 global climate models from HTAP II, directed by the United Nations Economic Commission for Europe (<http://www.htap.org/>) (33). The models used were CAM-Chem, CHASER\_rel, CHASER\_t106, EMEP\_rv48, GEMMACH, GEOS5, GEOSSCHEMAJOINT, OsloCTM3v2, GOCARTv5, SPRINTARS, and C-IFS\_v2. Simulations were conducted for 2010, with an additional 6 mo as spin-up. The modeled  $\text{NO}_y$  deposition comprises deposition of  $\text{NO}_2$ ,  $\text{HNO}_3$ , aerosol  $\text{NO}_3^-$ , peroxyacyl nitrate (PAN), and organic nitrates other than PAN. MMM performance on wet deposition of  $\text{NO}_y$  and  $\text{NH}_x$  was evaluated with the NADP observation network (ref. 5 and *SI Appendix, Fig. S6*). Simulation of dry deposition is subject to uncertainty, owing to lack of comprehensive observation data (5, 34). The base case utilized the HTAP v2.2 emissions inventory (35). In the emission control scenario, a modified emissions inventory was used with 20% reduction of all anthropogenic emissions, including  $\text{SO}_2$ ,  $\text{NO}_x$ , and  $\text{NH}_3$  over North America. Since the meteorological fields and model parameters remained unchanged in the base case and control scenarios, emissions change is the main factor contributing to the variation of deposition.

**Data Availability.** All data are publicly available. Details about measurement data, emission, and modeling datasets are given in *SI Appendix*.

**ACKNOWLEDGMENTS.** We thank NADP and all participating modeling groups in HTAP II. We acknowledge the computational resources of the Oak Ridge National Laboratory, supported by the Office of Science of the US Department of Energy (Contract DE-AC05-00OR22725).

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