Research Question and Aim

Broader Context:

Green hydrogen, produced from water, is a promising clean energy-rich fuel that enables a sustainable, affordable, and secure global energy system.

Specific Problem:

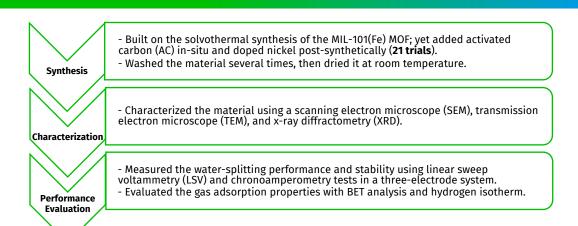
- Currently used materials to split water and produce H₂ have limitations such as corrosion, high cost, and dissolution.
- High pressure and low temperature are needed in conventional hydrogen storage, posing safety concerns and requiring complicated infrastructure.

Scheme 1. Illustration of the reaction pathway for HER (green arrows) and OER (blue arrows) in alkaline media [1]

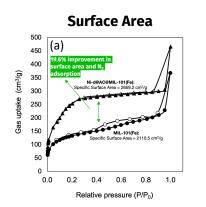
• Aim: to synthesize and engineer a novel, cost-effective, and efficient metal-organic framework (MOF) to produce green H₂ from water and store it at relative minimal pressure and temperatures.

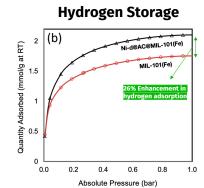
[1] Min Yang at Applied Materials Today. 2020.

Procedures



Data Analysis & Results





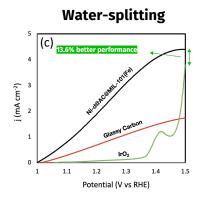


Figure 1. (a) Nitrogen adsorption-desorption BET isotherm of MIL-101(Fe) and Ni-d@AC@MIL-101(Fe). Filled (adsorption) and empty (desorption) symbol, (b) Adsorption isotherms of hydrogen on the MIL-101(Fe) and Ni-d@AC@MIL-101(Fe), and (c) Linear sweep voltammetry (LSV) of the Ni-d@AC@MIL-101(Fe), glassy carbon, and IrO₂ (RHE).

Conclusion

This novel metal-organic framework:

- 1. Enhanced the water-splitting and hydrogen production after the modification, which proved to be 13.6% better than the benchmark, IrO₂ at 1.5 V.
- 2. Built and engineered the MIL-101(Fe) without the collapse of the framework, which improved the surface area by 19.6% and the hydrogen uptake by 26% at 1 bar.
- 3. Could store hydrogen in off-grid systems (houses, research facilities, etc.) at relative minimal pressure and temperatures, and produce H₂ at a fraction of IrO₂ cost and higher efficiency.

This work has the potential to help decarbonize the transportation and power sectors once deployed.



All figures, photos, and diagrams were created by the student research, unless otherwise noted