## Batch to fed batch system

Haiting Wang

February 2024

## 1 Introduction

This is the description of how to derive the mass balance of a fed-batch bioreactor, given the mass balance of the batch system. A fed-batch system of biodiesel production using microalgae is demonstrated as an example[?]

## 2 Batch system

The mass balance of the batch system is shown as below:

$$\frac{dX}{dt} = \mu_0 X - \mu_d X \tag{1}$$

$$\frac{dN}{dt} = -\mu N \cdot \frac{N}{N + K_N} \cdot X \tag{2}$$

$$\frac{dq}{dt} = \mu_N \cdot \frac{N}{N + K_N} - (\mu_0 - \mu_d) \cdot q \tag{3}$$

$$\frac{df}{dt} = \mu_0(\theta \cdot q - \epsilon \cdot f) - \gamma \cdot \mu_N \frac{N}{N + K_N} + \mu_d \cdot \epsilon \cdot f \tag{4}$$

$$\mu_0 = g(X, q, I_0) \tag{5}$$

The system above is the structure of hybrid model that contains 4 state variables: X, N, q, f. Eq. 5 represents the data-driven equations of the hybrid model, which takes the concentration of X, q and the control variable  $I_0$  as inputs. The goal of Eq. 5 is to predict  $\mu$  as a time-varying variable through the whole process.

## 3 Fed-batch system

To start with, for all state variables in the fed-batch system without direct feeding streams, the mass balance can be represented as the following form:

$$\frac{d(CV)}{dt} = Vr_c(C, \mu) \tag{6}$$

Where C is the concentration of the state variable (g/L), V is the volume of the feed stream that being added to the bioreactor (L),  $r_c$  is the reaction rate of state variable C  $(g/(L \cdot h))$ , which is the function of the state variable C and the control variable  $\mu$ .

On the other hand, the general mass balance of the state variable which is fed to the biosystem directly, the mass balance can be represented as below:

$$\frac{d(SV)}{dt} = F_{in}S_{in} + Vr_s(S,\mu) \tag{7}$$

Where  $F_{in}$  (L/h) is the volumetric flowrate that being fed to the system. Therefore, the change of the volume is shown as below:

$$F_{in} = \frac{dV}{dt} \tag{8}$$

Therefore, Eq. ?? and Eq. 7 can be further derived through the chain rule of the differential calculus:

$$\frac{d(CV)}{dt} = C\frac{dV}{dt} + V\frac{dC}{dt} = V\frac{dC}{dt} + F_{in}C$$
(9)

$$\frac{d(SV)}{dt} = S\frac{dV}{dt} + V\frac{dS}{dt} = V\frac{dS}{dt} + F_{in}S$$
 (10)

Furthermore, by substituting Eq. 6 to Eq. 9 and Eq. ?? to Eq. 10, we can obtain the following mass balance equations:

$$\frac{dC}{dt} = -\frac{F_{in}}{V}C + r_c(C, \mu) \tag{11}$$

$$\frac{dS}{dt} = -\frac{F_{in}}{V}(S - S_{in}) + r_s(S, \mu) \tag{12}$$

Now, we can start to drive the mass balance of the fed-batch system given Eq 1 to Eq. 4.

$$\begin{split} \frac{d(XV)}{dt} &= Vr_X(X) \\ \frac{dX}{dt}V + \frac{dV}{dt}X &= Vr_X(X) \\ \frac{dX}{dt}V &= -\frac{dV}{dt}X + Vr_X(X) \\ \frac{dX}{dt} &= -\frac{F_{in}}{V}X + r_X(X) \\ \frac{dX}{dt} &= -\frac{F_{in}}{V}X + (\mu_0 X - \mu_d X) \end{split}$$

$$\begin{split} \frac{d(NV)}{dt} &= F_{in}N_{in} + Vr_N(N) \\ \frac{dN}{dt}V + \frac{dV}{dt}N &= F_{in}N_{in} + Vr_N(N) \\ \frac{dN}{dt}V + F_{in}N &= F_{in}N_{in} + Vr_N(N) \\ \frac{dN}{dt}V &= F_{in}(N_{in} - N) + Vr_N(N) \\ \frac{dN}{dt} &= \frac{F_{in}(N_{in} - N)}{V} + r_N(N) \\ \frac{dN}{dt} &= \frac{F_{in}(N_{in} - N)}{V} - \mu N \cdot \frac{N}{N + K_N} \cdot X \end{split}$$

The derivation of the mass balance of q needs to be modified since q is the intracellular concentration of nitrate. Therefore, the concentration is not directly impacted by the feeding of nitrate. Thus, the mass balance of q is derived considering the mass balance between q and N. As pointed out from the literature, the nitrate consumption rate must be equal to the accumulation of the intracellular nitrogen:

$$\frac{d(X \cdot q)}{dt} = -\frac{dN}{dt} \tag{13}$$

Therefore, the following equations can be derived:

$$\begin{split} \frac{d(X \cdot q)}{dt} &= -\frac{dN}{dt} \\ \frac{dq}{dt}X + \frac{dX}{dt}q &= -\frac{dN}{dt} \\ \frac{dq}{dt}X &= -\frac{dN}{dt} - \frac{dX}{dt}q \\ \frac{dq}{dt} &= -\frac{\frac{dN}{dt}}{X} - \frac{dX}{dt}\frac{q}{X} \\ \frac{dq}{dt} &= -[\frac{F_{in}(N_{in} - N)}{V} - \mu_N \cdot \frac{N}{N + K_N} \cdot X] \cdot \frac{1}{X} - [-\frac{F_{in}}{V}X + (\mu_0 - \mu_d)X]\frac{q}{X} \\ \frac{dq}{dt} &= -\frac{F_{in}(N_{in} - N)}{VX} + \mu_N \cdot \frac{N}{N + K_N} + \frac{F_{in}q}{V} - (\mu_0 - \mu_d)q \end{split}$$

The derivation of product f is derived following different procedure. As pointed out by literature, the reaction rate of f is derived as the difference between biolipid synthesis and consumption, respectively. Thus, the mass balance of f keeps the same as the batch system:

$$\frac{df}{dt} = \mu_0(\theta \cdot q - \epsilon \cdot f) - \gamma \cdot \mu_N \frac{N}{N + K_N} + \mu_d \cdot \epsilon \cdot f \tag{14}$$

In summary, the mass balance of the fed batch bioreactor of the hybrid model can be illustrated as below:

$$\frac{dX}{dt} = -\frac{F_{in}}{V}X + (\mu_0 X - \mu_d X) \tag{15}$$

$$\frac{dN}{dt} = \frac{F_{in}(N_{in} - N)}{V} - \mu N \cdot \frac{N}{N + K_N} \cdot X \tag{16}$$

$$\frac{dq}{dt} = -\frac{F_{in}(N_{in} - N)}{VX} + \mu_N \cdot \frac{N}{N + K_N} + \frac{F_{in}q}{V} - (\mu_0 - \mu_d)q$$
 (17)

$$\frac{df}{dt} = \mu_0(\theta \cdot q - \epsilon \cdot f) - \gamma \cdot \mu_N \frac{N}{N + K_N} + \mu_d \cdot \epsilon \cdot f \tag{18}$$

$$F_{in} = \frac{dV}{dt} \tag{19}$$

$$\mu_0 = g(X, q, I_0) \tag{20}$$