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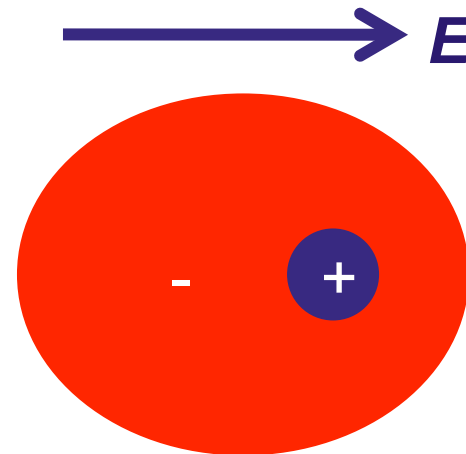
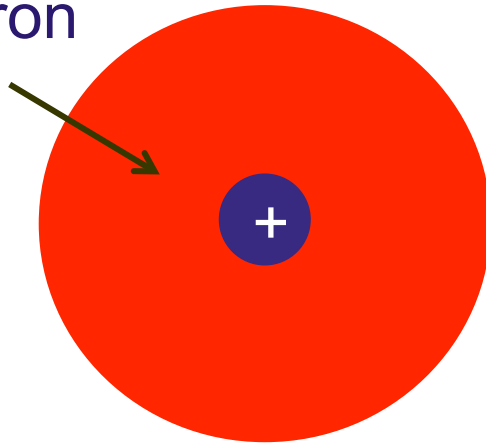
EEE6212

“Semiconductor Materials” -Conduction Processes

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Polarization

-ve electron
“cloud”



- When a system is subject to an electric field, \mathbf{E} , there is a tendency of the +ve and -ve charge to displace relative to one another so the system has an electric dipole moment. The dipole moment per unit volume is the polarization \mathbf{P} . See EEE101.

$$\mathbf{P} = \epsilon_0 \chi_e \mathbf{E}$$

ϵ_0 = permittivity of free space, χ_e = electric susceptibility

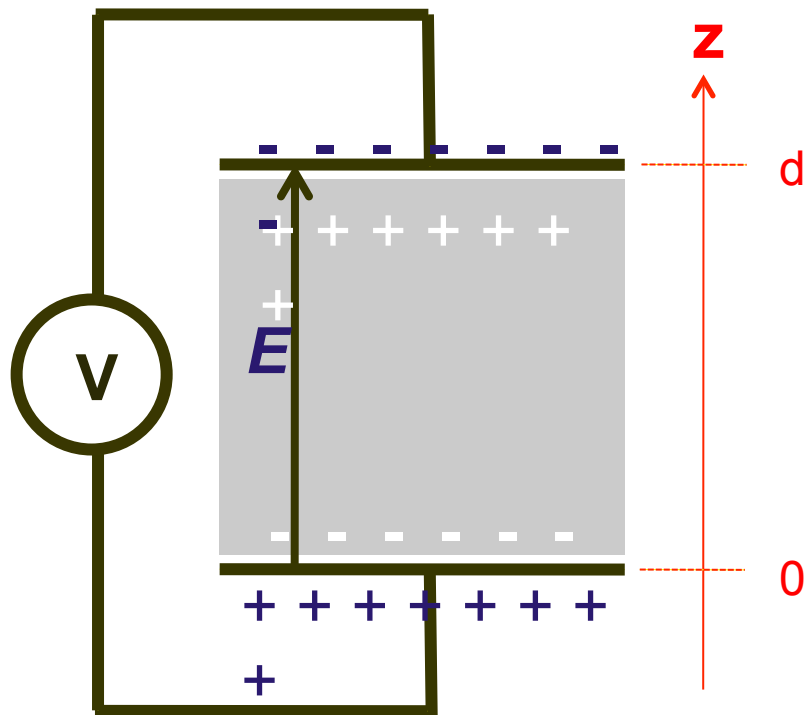
Permittivity and Susceptibility

- **Electric susceptibility** χ_e - measure of how easy it is to polarize a dielectric in response to an electric field
- **Permittivity**, ϵ , is a physical property of a solid (dielectric medium). Measure of the ability of the material to polarize in response to the field, and thereby reduce the total electric field inside the material. It is a measure of how easily the material “permits” the electric field to propagate.
- We usually compare the permittivity of a material to free space (i.e. vacuum) ϵ_0 ($= 8.8 \times 10^{-12}$ F/m) through the relative permittivity ϵ_r
- The permittivity of air \sim permittivity of free space so $\epsilon_r = 1$ for air

Polarization Mechanisms

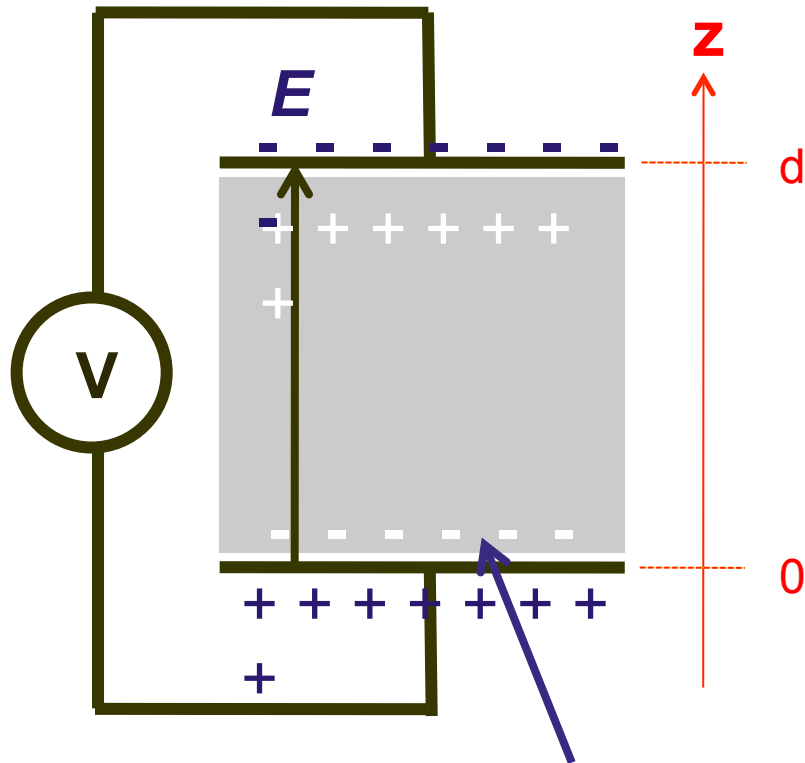
- **Electronic or Induced Polarization** (previous slide). All dielectric materials – fast frequency response $\sim 10^{15} \text{ s}^{-1}$
- **Ionic Polarization** – In ionic crystals e.g. NaCl E field can shift sub lattice of Na^+ and Cl^- ions. Moderate frequency response $\sim 10^9 \text{ s}^{-1}$
- **Orientational Polarization** – can have a dipole within a molecule (polar molecule). E field aligns randomly oriented dipoles causing net polarization. Important for liquids & gases. Slow frequency response $\sim 10^4 \text{ s}^{-1}$
- Moving charge and molecules takes energy – loss appears as resistive component to impedance

Dielectric Capacitor



- In response to an E -field a dielectric produces a polarization
- Displaced surface charge opposite the capacitor plates
- Charge on plates to maintain E -field
- Capacitance, $C=Q/V$ where Q = charge and V = voltage

Dielectric Capacitor



Charge per unit area = ρ

$$V = \int_0^d E \, dz = \int_0^d \frac{\rho}{\epsilon} \, dz$$

$$= \frac{\rho d}{\epsilon} = \frac{Qd}{\epsilon a}$$

inserting $C = \frac{Q}{V}$

$$C = \frac{\epsilon A}{d}$$

where $\epsilon = \epsilon_0 \epsilon_r$

ϵ_r = relative permittivity

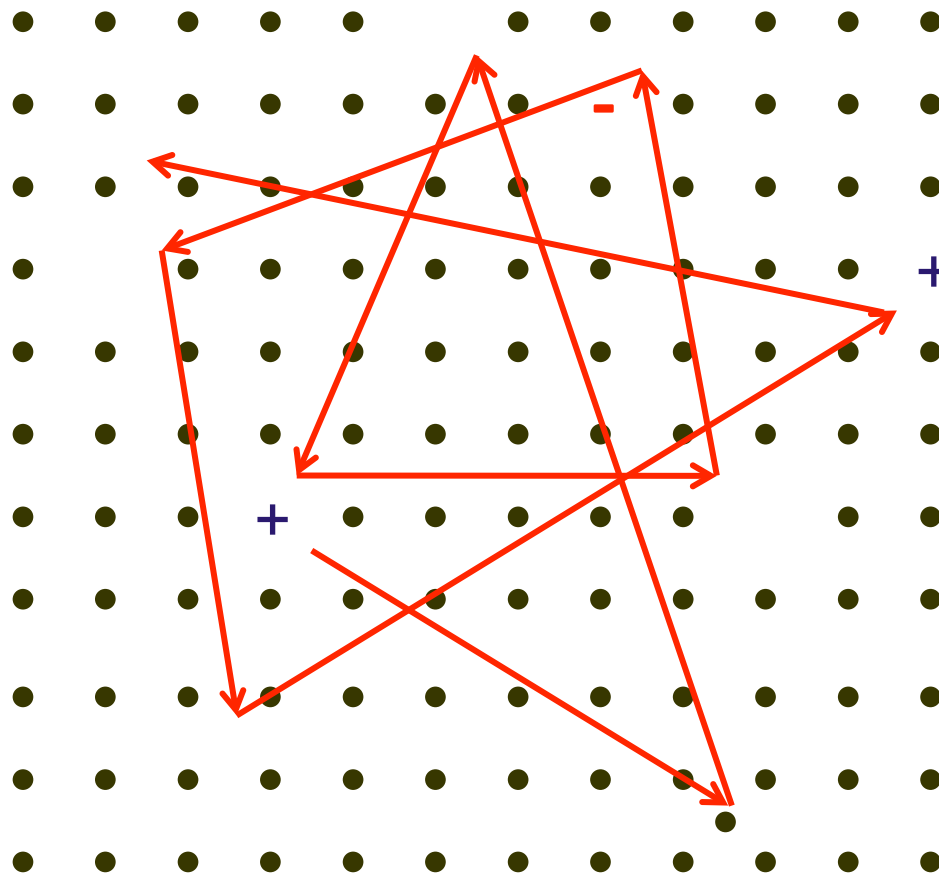
Dielectric Breakdown

- Sudden increase in current above a critical electric field
- Limitation to dielectric – capacitor or insulator become a ~short circuit
- Reversible & non-reversible (=catastrophic)
- Breakdown E-field can be $\sim 10^9 \text{ Vm}^{-1}$
- Ideally as high as possible

Ideal Capacitor Dielectric

- High ϵ_r
- Breakdown only at very high fields
- Low cost
- Manufacturability of thin films
- Reliability

Free Electron - Scattering



Imagine we can visualise the motion of electrons...

Free electrons -thermal energy–
c.f. Brownian motion

Observe enough carriers - no
net movement of charge – no
net current. Collisions from
imperfections to crystal lattice

- Ionised impurities
- Interstitial defects
- Vacancy defects
- Phonons

What causes a current?

Three causes of net flow of current

- An electric potential gradient dV/dx (i.e. an E-field)
- An electron density gradient dn/dx
- A temperature gradient dT/dx

Application of E-field



$$\text{Force} = -e E$$

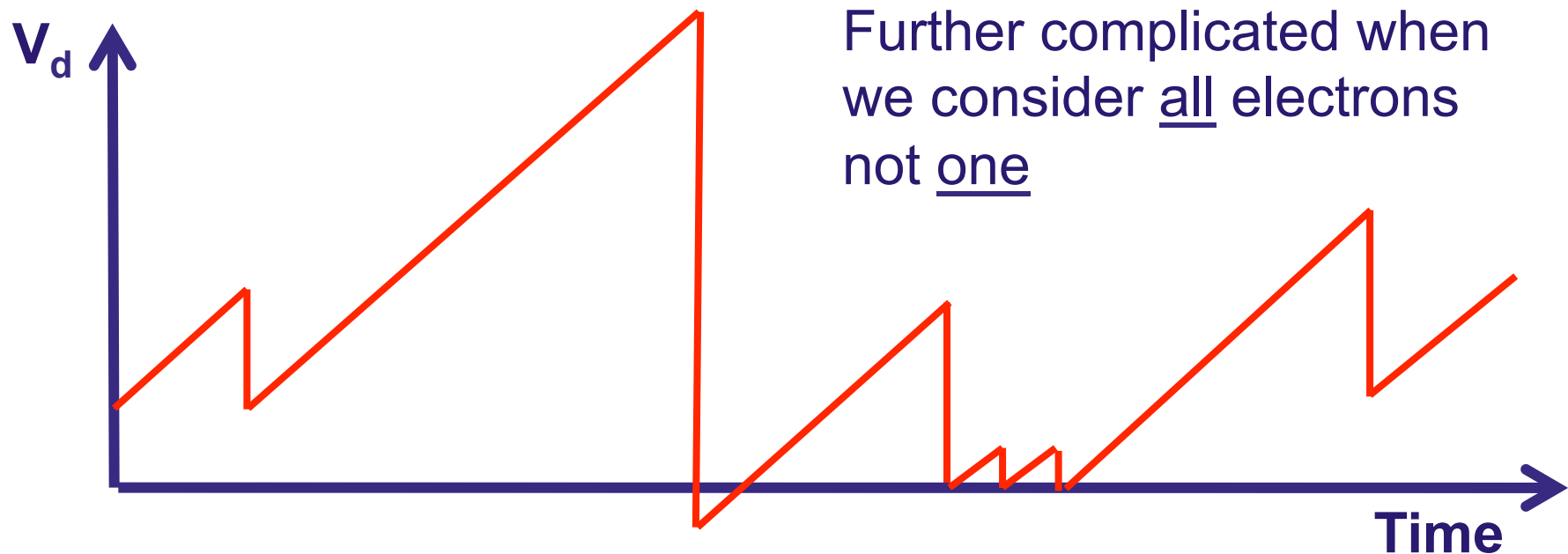

electron

Electron acceleration?

- In vacuum we use Newton's laws $F = m_e A$
- In semiconductor just apply effective mass $F = m^* A$
- n.b. you need to be careful in using m^* - some authors assume you know $m^* = m^* m_e$ (just like I do)

Drift Velocity

- Electron gains velocity as it is accelerated in E-field
- Electron can loose velocity when they are scattered



Derivation

- Solid with free electron concentration, $n \text{ m}^{-3}$
- E-field applied, electrons will accelerate $a = \frac{-qE}{m^*}$
- Acceleration so velocity and momentum are changing.
Rate of change of momentum with time for each electron

$$\frac{dp_e}{dt} = \frac{d(m^* v_d)}{dt} = m^* \frac{dv_d}{dt} = m^* a = F = -qE$$

- Rate of change of momentum, p , of all n electrons

$$\left(\frac{dp}{dt} \right)_{\text{drift}} = -qEn$$

Derivation (2)

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- Electrons will accelerate until scattered
- Chance (or probability) that a particular electron will be scattered in unit time is a number between 0 and 1. For a large number of electrons this fraction will be scattered.

$$\text{Number scattered per unit time} = \frac{n}{\tau}$$

- Where τ is a time constant (= average time between scattering events)
- Assume that on average each scattering event causes the electron to loose all momentum (momentum at this instant = $m \cdot v_d$)
- Total change in momentum due to scattering events is;

$$\left(\frac{dp}{dt} \right)_{\text{scatter}} = \frac{\text{number scattering events}}{\text{time}} \times \text{momentum change}$$

Derivation (3)

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- Where v_d is the average drift velocity of the electron population
- At equilibrium the total momentum change of the population is zero

$$\left(\frac{dp}{dt}\right)_{\text{drift}} + \left(\frac{dp}{dt}\right)_{\text{scatter}} = 0$$

- Hence

$$-qEn - \frac{n}{\tau} m^* \langle v_d \rangle = 0 \quad \text{and} \quad \langle v_d \rangle = -\frac{qE\tau}{m^*}$$

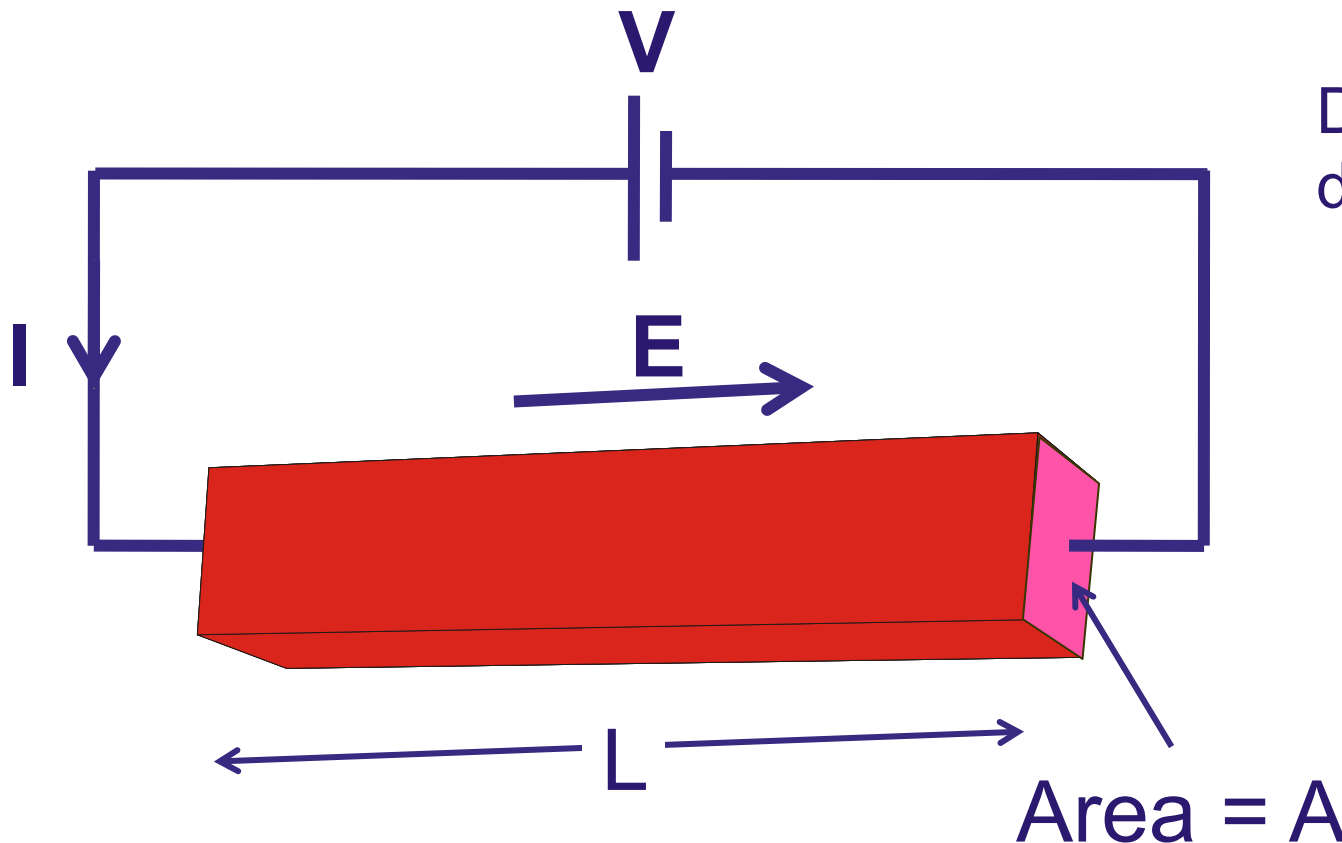
- n.b. Some text books derivations assume regular scattering events and obtain the approximate result in slide 8. This derivation is better.

Mobility, μ

- Drift velocity given by $\langle v_d \rangle = -\frac{q \tau E}{m^*}$
- Important parameter is average time between scattering events, τ
- Governed by impurity concentration, phonons, defects
- Effective mass also important – can simplify to one (easily measureable) material parameter the mobility, μ , to give;

$$\langle v_d \rangle = -\mu E \quad \text{Where} \quad \mu = \frac{q \tau}{m^*}$$

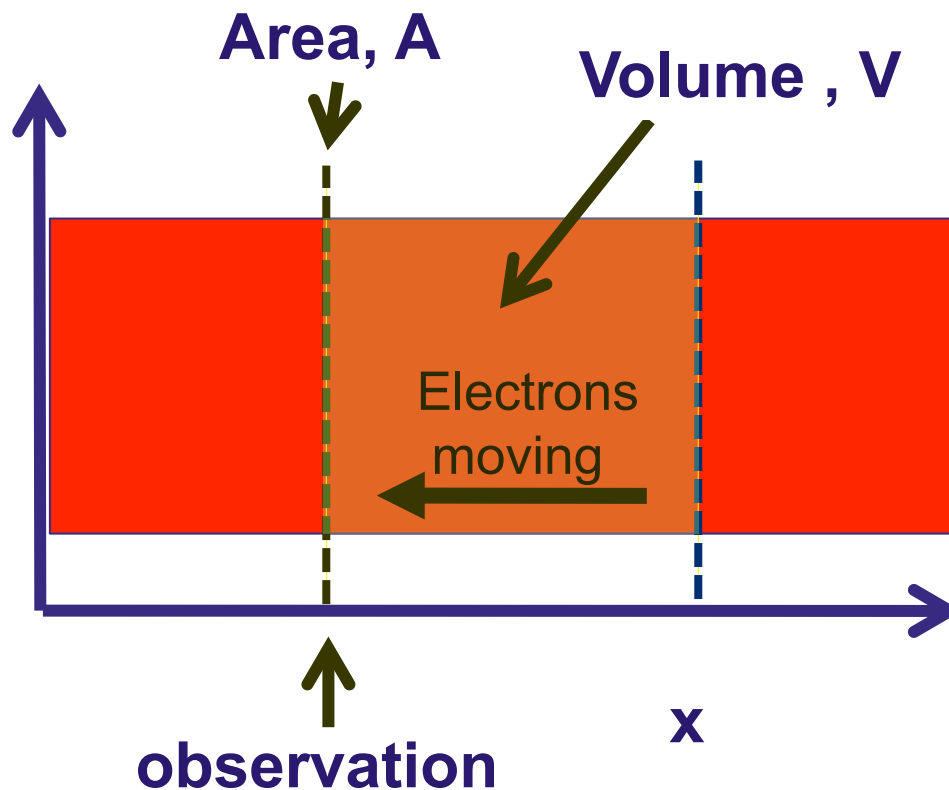
Solid with Free Electrons



Define a current density, J ,

$$J = \frac{I}{A}$$

Longitudinal Slice



Have average velocity ,
 v_d of electrons and density of
electrons n

In time t , all electrons in
shaded region will move past
observation point

$$\rightarrow x = v_d t$$

Number of electrons in this
volume is

$$n V = n A x = n A v_d t$$

Continued

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- Charge on electron $= -q$ ($q = 1.6 \times 10^{-19}$ C) so in unit time (a second) the amount of charge flowing past our observation point is the current $I = -n A q v_d$

The current density is given by; $J = I/A = -n q v_d$

n.b.

J and I in opposite direction to electron flow as expected

Sometimes drift velocity written as v or v_d

Sometimes charge on electron written as e or q

Often we use centimetres instead of metres be careful!

Eliminate V_d

- The previous equation is only useful if we know the drift velocity, which we have derived $v_d = -\mu E$
- Which gives $J = n q \mu E$
- So the current density in our solid depends upon
 - Carrier concentration - how many carriers
 - E-field – magnitude dv/dx
 - Mobility - how easy the carriers can move
 - (The charge on an electron not negotiable!)

Ohm's Law

$$J = n q \mu E \quad \text{Can be simplified to} \quad J = \sigma E$$

Where the conductivity $\sigma = nq\mu$

Conductivity is inverse of resistivity $\rho = \frac{1}{\sigma}$

This is the general form of ohm's law



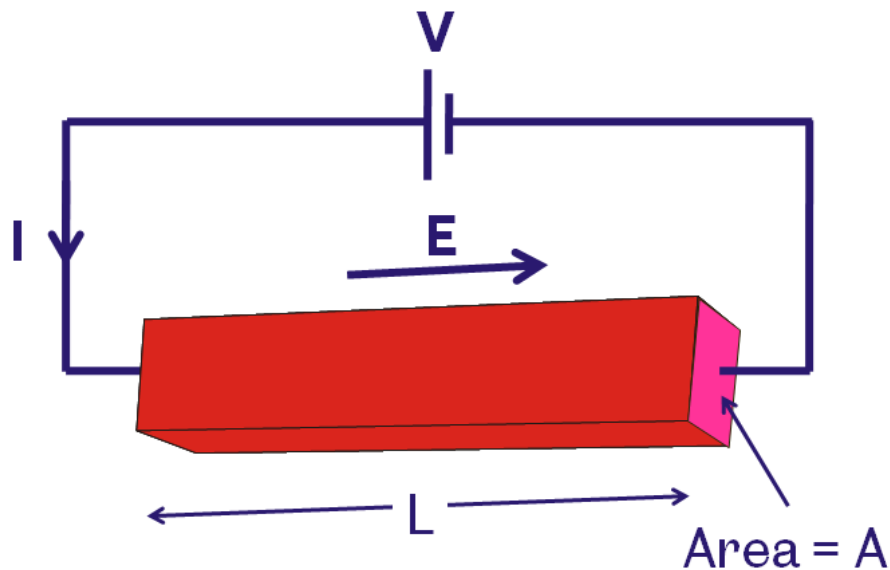
Ohm's Law (2)

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$$J = \frac{I}{A} \quad E = \frac{V}{L} \quad J = \sigma E$$

$$\frac{I}{A} = \frac{\sigma V}{L}$$

$$I = \frac{\sigma A V}{L}$$



(Ohm's law) is true if $R = \frac{L}{\sigma A}$

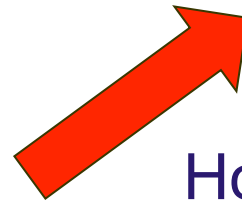
Extrinsic Semiconductor -Drift

Extrinsic Si
– p-doped with B to give

$$p = 10^{21} \text{ m}^{-3}$$
$$n \sim n_i = 10^{16} \text{ m}^{-3}$$

$$\mu_e = 0.12 \text{ m}^2\text{V}^{-1}\text{s}^{-1}$$
$$\mu_h = 0.05 \text{ m}^2\text{V}^{-1}\text{s}^{-1}$$

$$\sigma = nq\mu_e + pq\mu_h$$



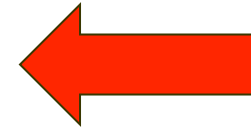
Hole drift current $> 10^4 \times$
electron drift current

If doping is high – ignore
minority carrier drift current

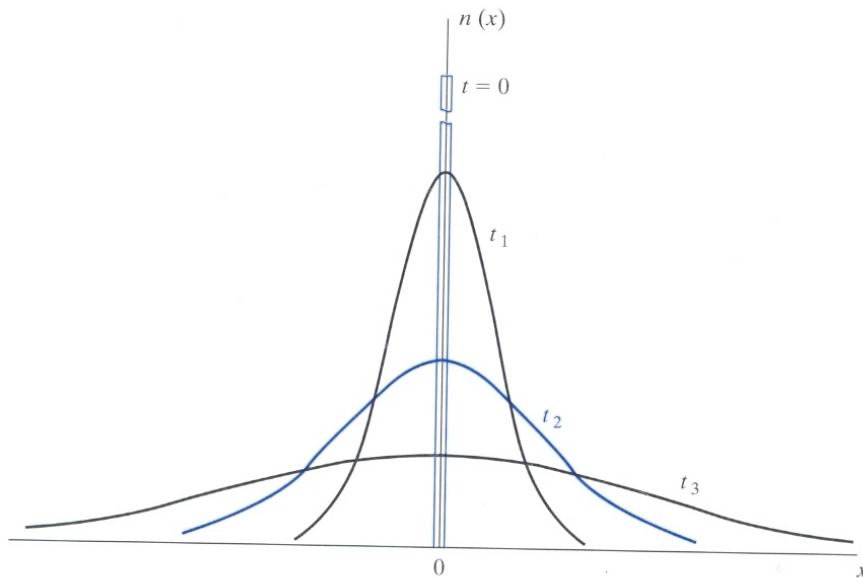
Sources of Current

Three causes of net flow of current

- An electric potential gradient dV/dx (i.e. an E-field)
- An electron density gradient dn/dx
- A temperature gradient dT/dx



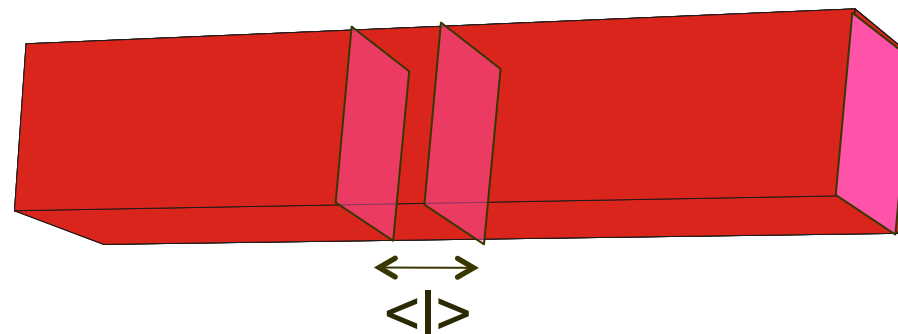
Diffusion - General



- Diffusion has been studied for a long time – salt in liquids, dust particles in air, population dynamics in biology, etc.
- Net flow (flux) of particles from high concentration to low concentration
- Acts to cancel out a non-uniform concentration distribution
- Governed by Fick's Laws

Mean free Path

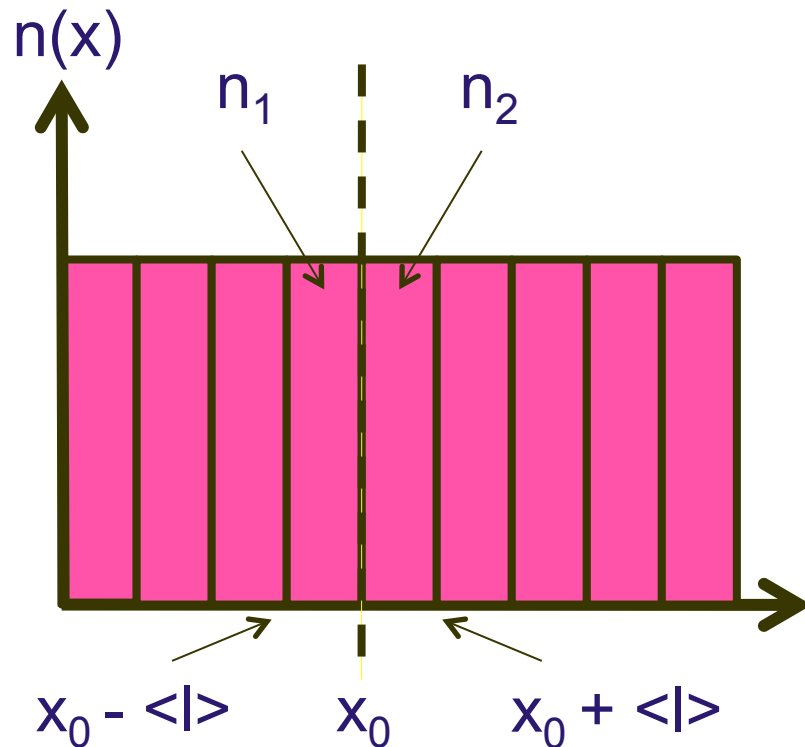
- For a carrier population we have a mean velocity, and a mean scattering time.
- There is a mean distance the carrier travels before scattering (Distance = Velocity x time). Termed the mean free path = $\langle l \rangle$
- Imagine a bar or rod we split into segments $\langle l \rangle$ wide



Area = A



Uniform Carrier Distribution



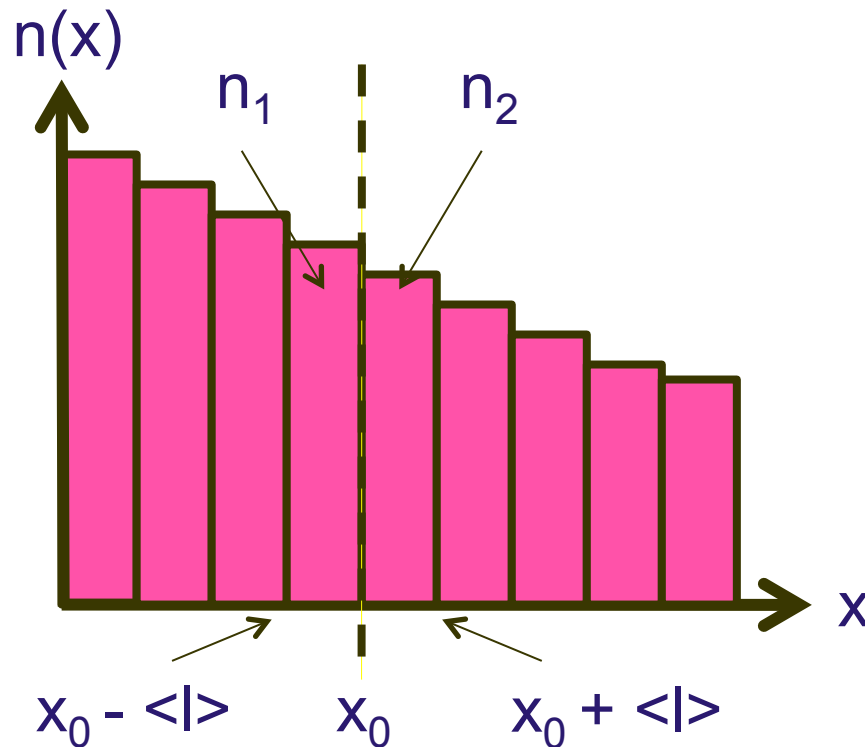
1D - Neighbouring segments of x_0

Concentrations n_1 and n_2

Half of all carriers moving +ve direction, half -ve direction

No net flow of charge – no current
- As many carriers from left to right
as from right to left through x_0

Carrier Distribution Gradient



Carrier flux passing x_0 from left to right (see e.g. Streetman)

$$\phi(x) = -D \frac{dn}{dx}$$

Flux

Diffusion Coefficient

Concentration

Distance

-ve sign as net motion is in direction of decreasing n

Electrons and Holes

- Must consider electrons and holes – electron and hole fluxes per unit area

$$\varphi_e(x) = -D_e \frac{dn}{dx}$$

$$\varphi_h(x) = -D_h \frac{dp}{dx}$$

- Diffusion Current is carrier flux times charge (-q for electrons, +q for holes)

$$J_e = qD_e \frac{dn}{dx}$$

$$J_h = -qD_h \frac{dp}{dx}$$

Diffusion Coefficient

“Einstein relationship”

Diffusion Coefficient or Diffusivity, D is measure of how easily carriers diffuse

$$D_{e,h} = \frac{k_B T \mu_{e,h}}{q}$$

- D increases when T increases – more thermal energy
- D increases when mobility increases – less inhibition to motion

Drift and Diffusion

- E-field *and* carrier concentration gradient

$$J_e^{\text{total}}(x) = J_e^{\text{drift}} + J_e^{\text{diffusion}} = q\mu_e E_x n + qD_e \frac{dn}{dx}$$

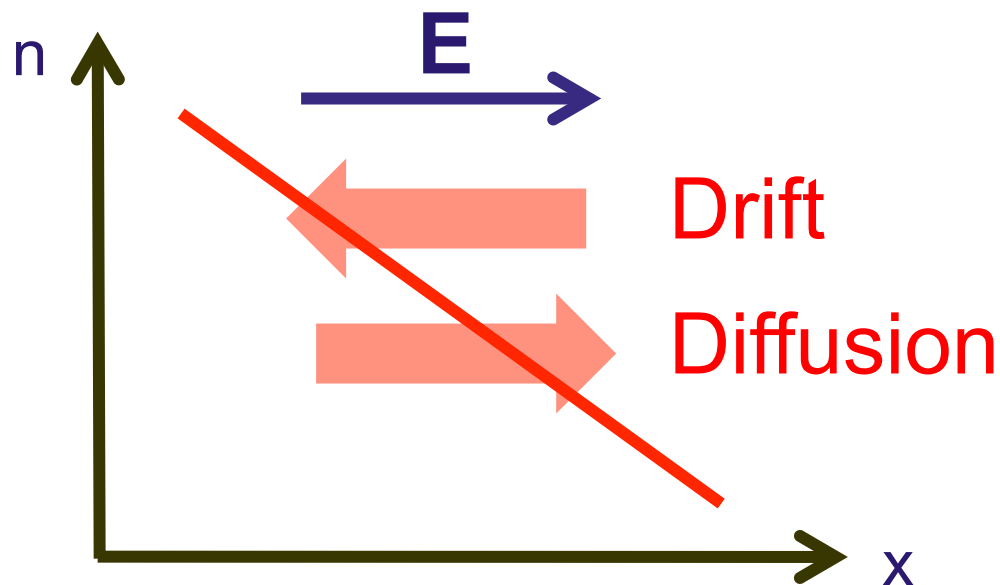
$$J_h^{\text{total}}(x) = J_h^{\text{drift}} + J_h^{\text{diffusion}} = q\mu_h E_x p - qD_h \frac{dp}{dx}$$

Minority Diffusion

- As drift current is proportional to carrier concentration, we know that minority carriers seldom provide much drift current and may often be ignored with little error
- As diffusion current is proportional to the *gradient* of carrier concentration, minority carrier diffusion currents can therefore be large

Drift Vs. Diffusion

- Consider case where there is a composition gradient *and* an E-field



$$J_e = q\mu_e E_x n + qD_e \frac{dn}{dx}$$

There is a case when $J=0$

$$E_x = -\frac{D_e}{n\mu_e} \frac{dn}{dx}$$

Carrier Concentration Gradients At Equilibrium

- Imagine a sample with carrier concentration – e.g. Vary doping in one direction
- At equilibrium there must be no net flow of current
- There is an *internal* field induced to ensure this is the case
- Varying doping concentrations results in built-in E-fields and potentials

Summary

- Discussed polarisation and capacitance
- Explored how a free electron is scattered within the crystal
- Discussed by drift and diffusion currents
 - Drift – E-field driven
 - Diffusion – carrier concentration gradient driven
- Combination of Diffusion and Drift – give rise to internal E-Fields